

Experiments and *ab initio* simulations The case of Amyloid-metal complexes (or the role of metals in the Alzheimer disease)

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dessiné à l'aigu par D. Biondi Bertoni

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SM&FT 2015

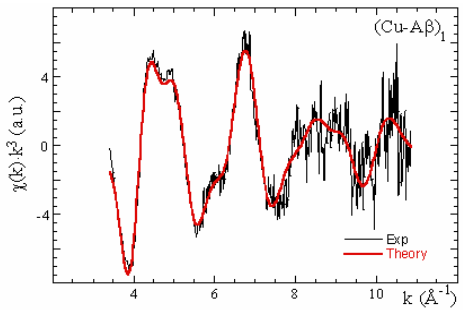
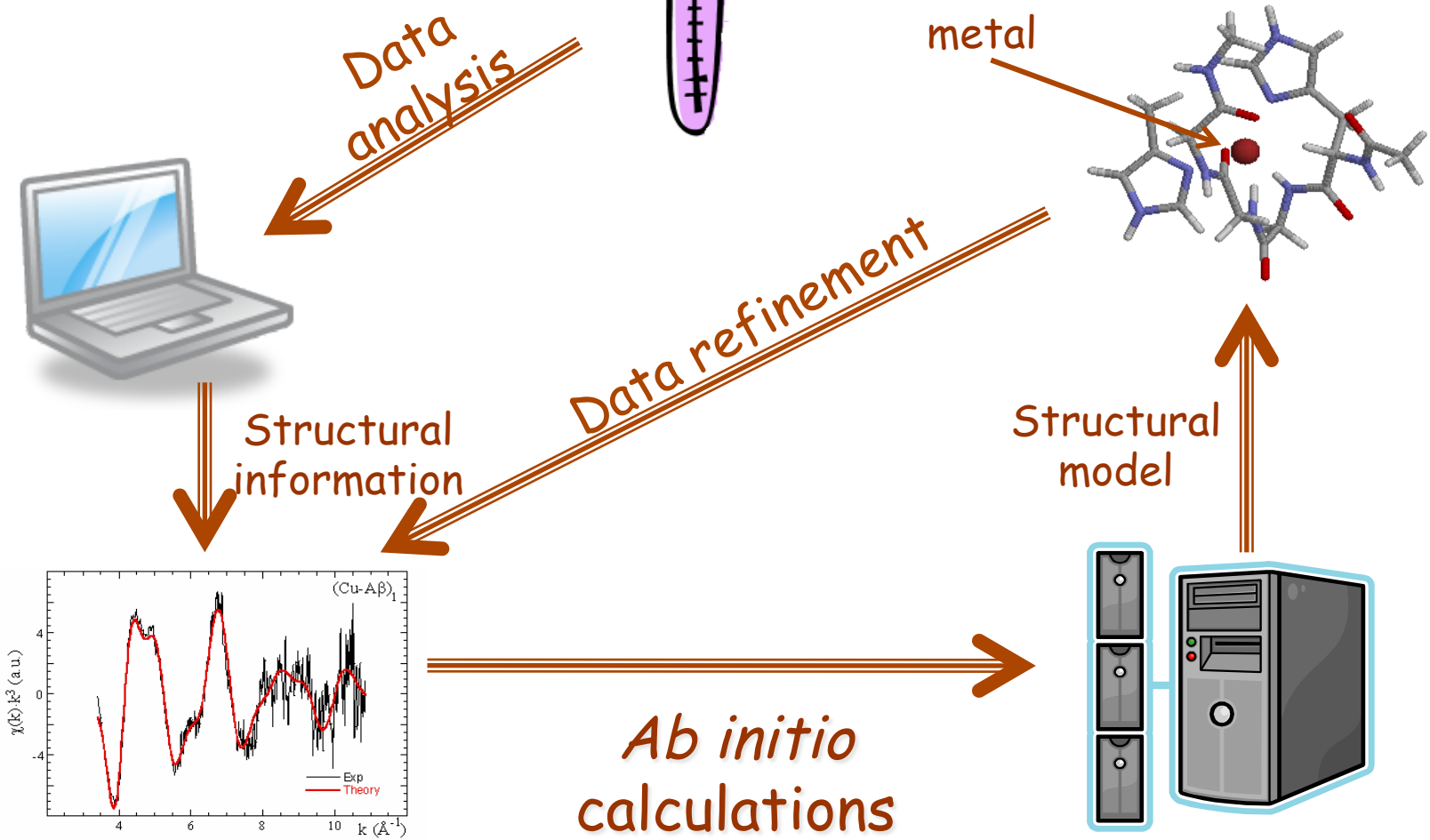
THE XVI WORKSHOP ON STATISTICAL MECHANICS
AND NON PERTURBATIVE FIELD THEORY

Bari (Italy), December 9-11, 2015

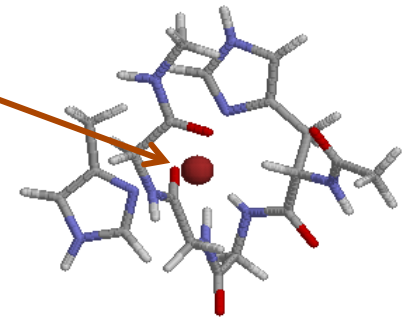
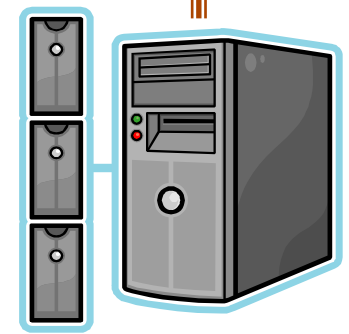
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XAS - EXAFS



Ab initio calculations



Biological challenge

understanding the molecular basis of Protein Conformational Diseases (PCD's) among which the **Alzheimer disease**

- misfolding and aggregation
- the A β -peptide
- metal ions and brain

Scientific instruments

focused experiments + numerical approaches

- XAS spectroscopy
- classical and *ab initio* molecular dynamics

Results

unveiling the (relevant) role of metal ions

Protein Conformational Diseases (PCD's)

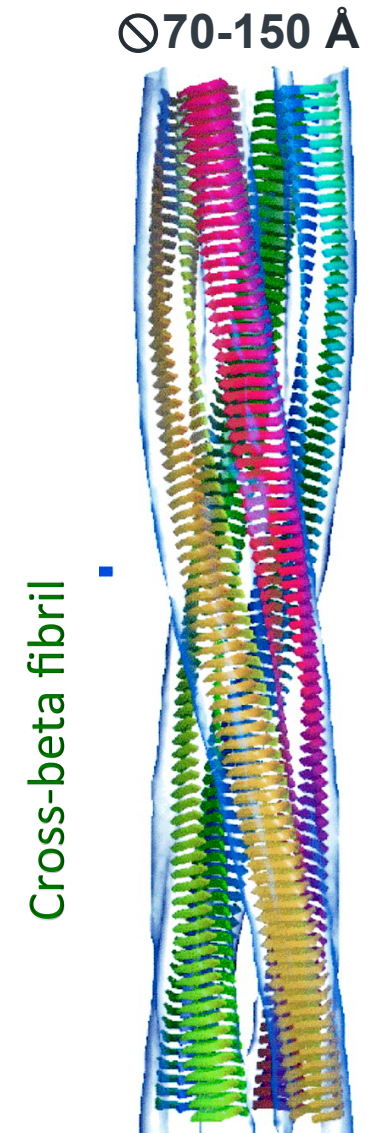
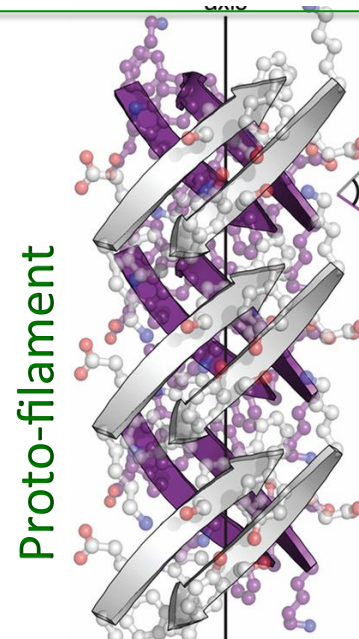
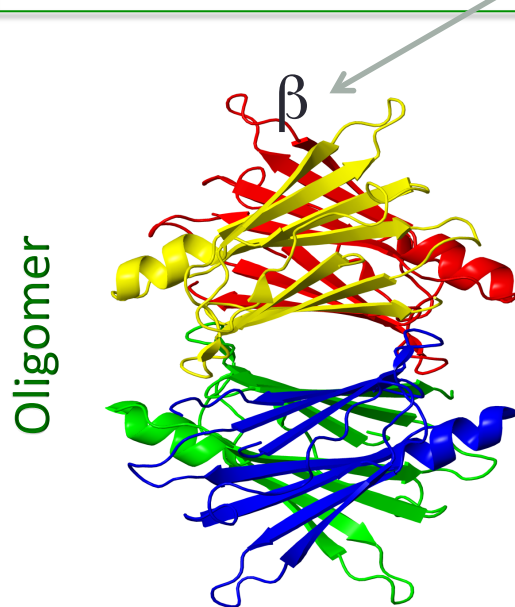
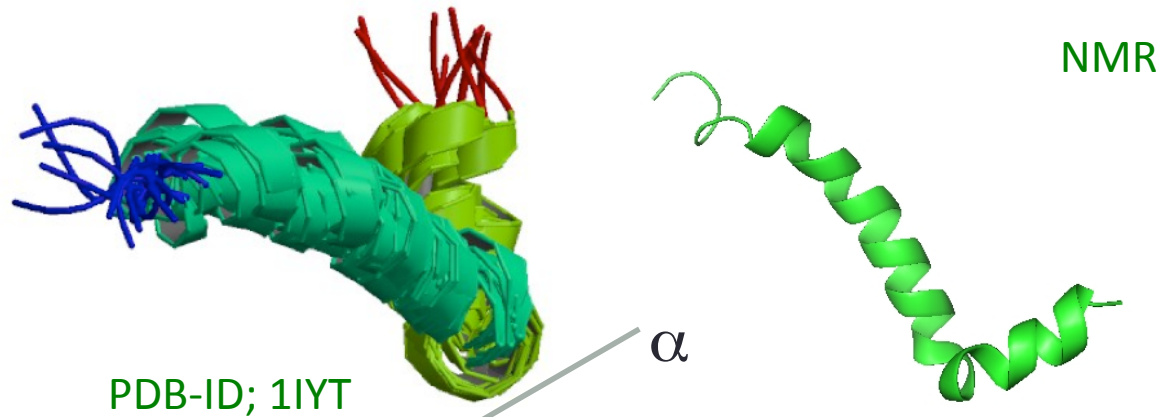


Disease	Misfolded Protein
Alzheimer's Disease	A β -peptide
Transmissible Spongiform Encephalopathies (TSE)	Prion protein
Parkinson's disease	α -synuclein
Type II diabetes	Amylin
Thyroid carcinoma	Procalcitonin
Atrial amyloidosis	Atrial natriuretic factor
Amyotrophic lateral sclerosis	Superoxide dismutase
Huntington disease	Glutamine
Primary systemic amyloidosis	Ig light chains
Secondary systemic amyloidosis	Serum amyloid A
Senile systemic amyloidosis	Transthyretin (wild tipe)
Familial amyloidotic polyneuropathy I	Transthyretin (mutant)
Familial amyloidotic polyneuropathy II	Apolipoprotein A1
Familial Mediterranean fever	Serum amyloid A
Hemodialysis-related amyloidosis	b2-microglobulin
Finnish hereditary systemic amyloidosis	Gelsolin (mutant)
Lysozyme systemic amyloidosis	Lisozime
Insulin-related amyloidosis	Insulin
Fibrinogen systemic amyloidosis	Fibrinogen α chain

The A β -peptide aggregation

5

Crescenzi et al. (2002) Eur. J. Biochem. 269, 5642–5648



Amyloid-like aggregates are now widely studied also in **non-medical** contexts
food science, pharmacology, nanomaterials ...

Metal ions are essential cell components

They can easily and quickly form complexes with enzymes and other biological molecules

They are involved in a wide range of biological processes

≈ ¼ ÷ ⅓ of all proteins are estimated to require metals to function

Metalloproteins have many different functions in cells

- storage and transport
- enzymatic activity
- signalling and transduction
- ...
- ...

**BUT METALS HIGH REACTIVITY CAN EASILY BECOME
HIGHLY TOXIC FOR THE CELL**

Metals should cross the blood-brain barrier

Central Nervous System



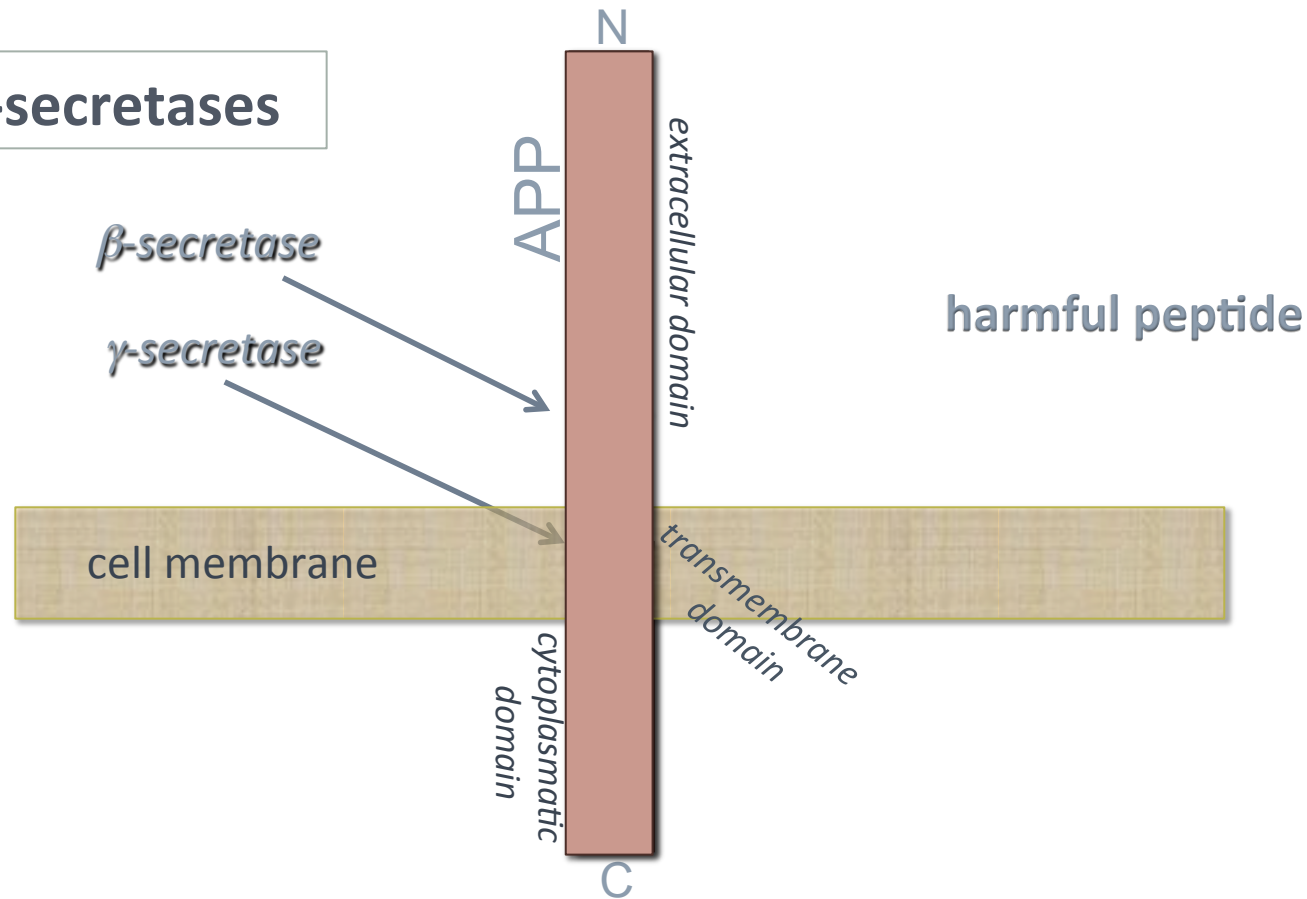
- Metals are **essential** for neurochemical activity
- **Fe, Cu and Zn** are fairly abundant in **grey matter**
- during neuro-transmission processes, high concentrations of Zn ($\sim 300 \mu\text{M}$) and Cu ($\sim 30 \mu\text{M}$) are normally released
- **Fe, Cu and Zn** are continuously **trafficked** within the cell and also into and out of it
- Metals **deficiency** may cause **dysfunctionality** and metal **excess** may lead to **toxicity**
- **Failure of metal trafficking (dyshomeostasis)** is known to occur both in AD and in Parkinson diseases

In presence of dyshomeostasis, metal ions may

- interfere with normal nervous cells activities
- disrupt brain electrochemical balance
- ...
- **promote protein misfolding and aggregation?**

The amyloid β -peptide

α -, β - and γ -secretases



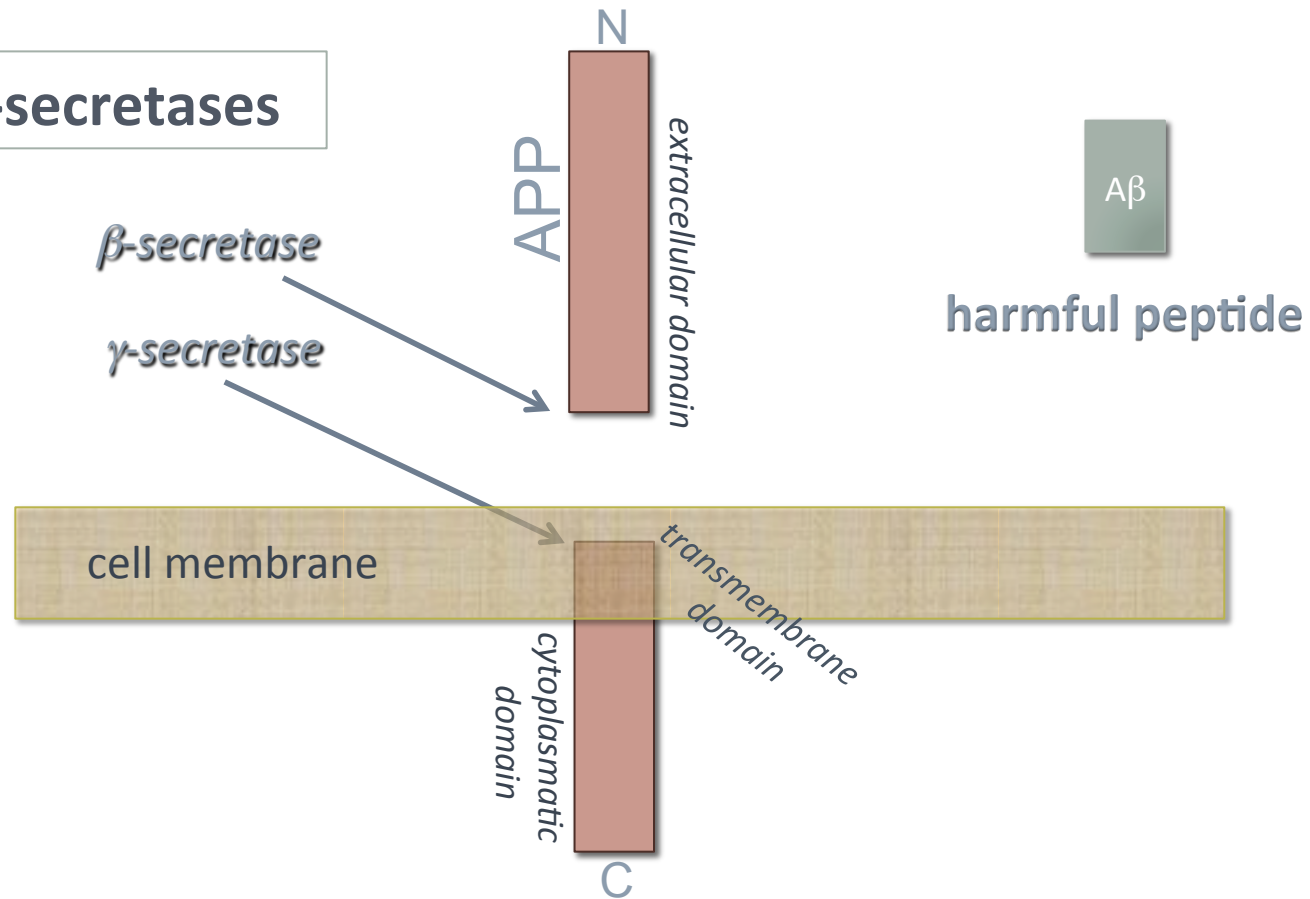
APP: transmembrane 770 a.a.'s protein



The amyloid β -peptide

10

α -, β - and γ -secretases



672

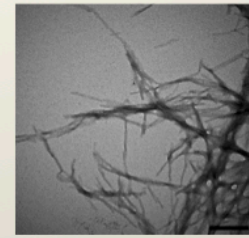
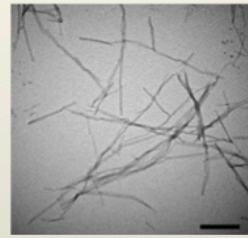
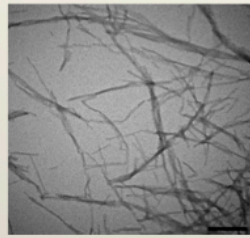
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711

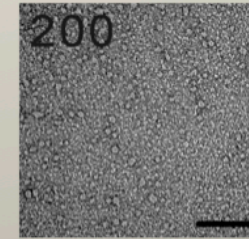
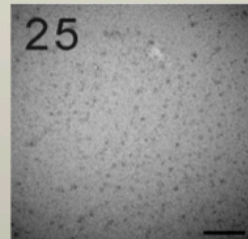
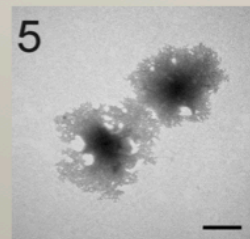
A β ₁₋₄₀

Metals affect A β -peptide aggregation

A β
(25 μ M)

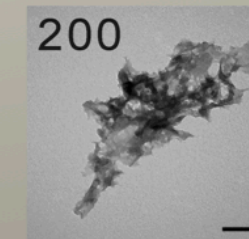
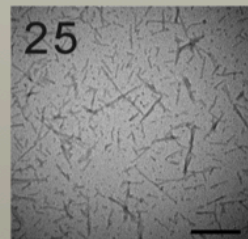
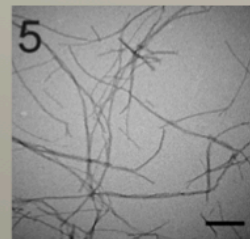


A β +Zn²⁺



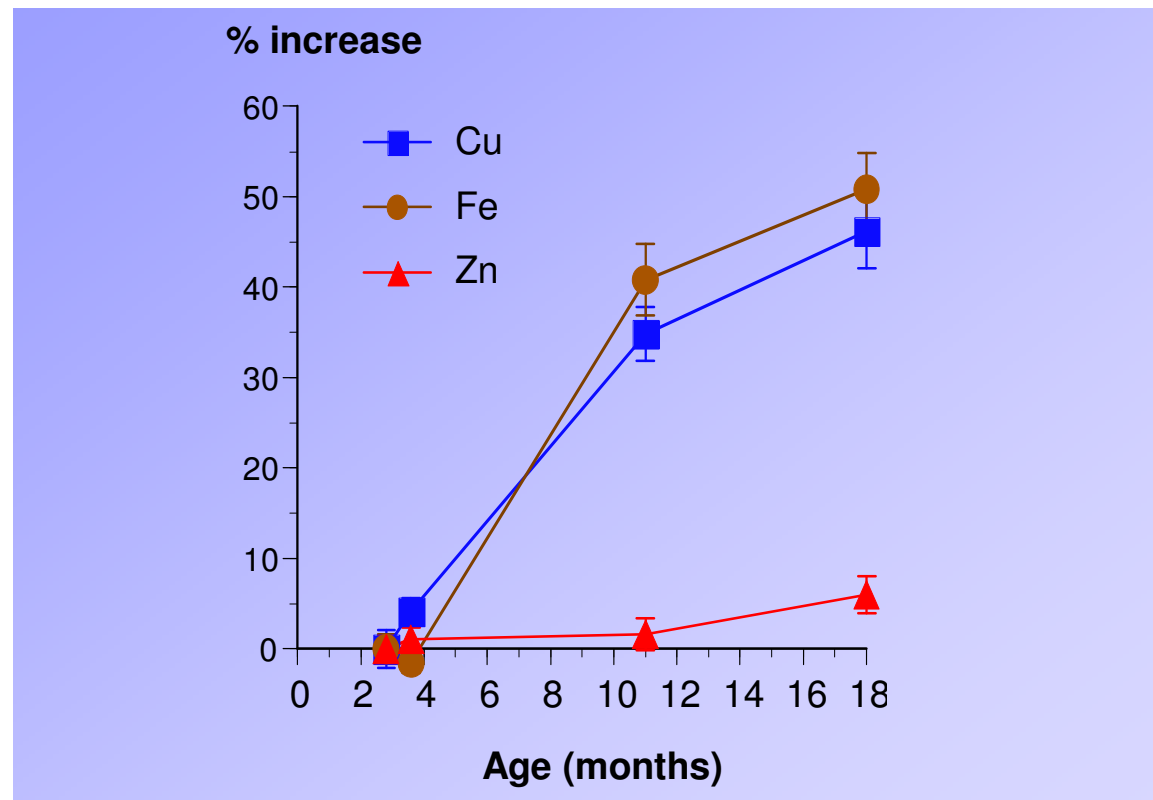
— 200 nm

A β +Cu²⁺



Normal aging and increased brain metal level

Maynard et al. (2002) JBC, 277: 44670



The triggering event is possibly metal dyshomeostasis

- The number of people aged 60 and over has doubled since 1980
- The number of people aged 80 years or older will be almost quadrupled to 395 millions between 2000 and 2050
- The risk of dementia (like Alzheimer disease) rises sharply with age with an estimated 25-30% of people aged 85 or older having some degree of cognitive decline.

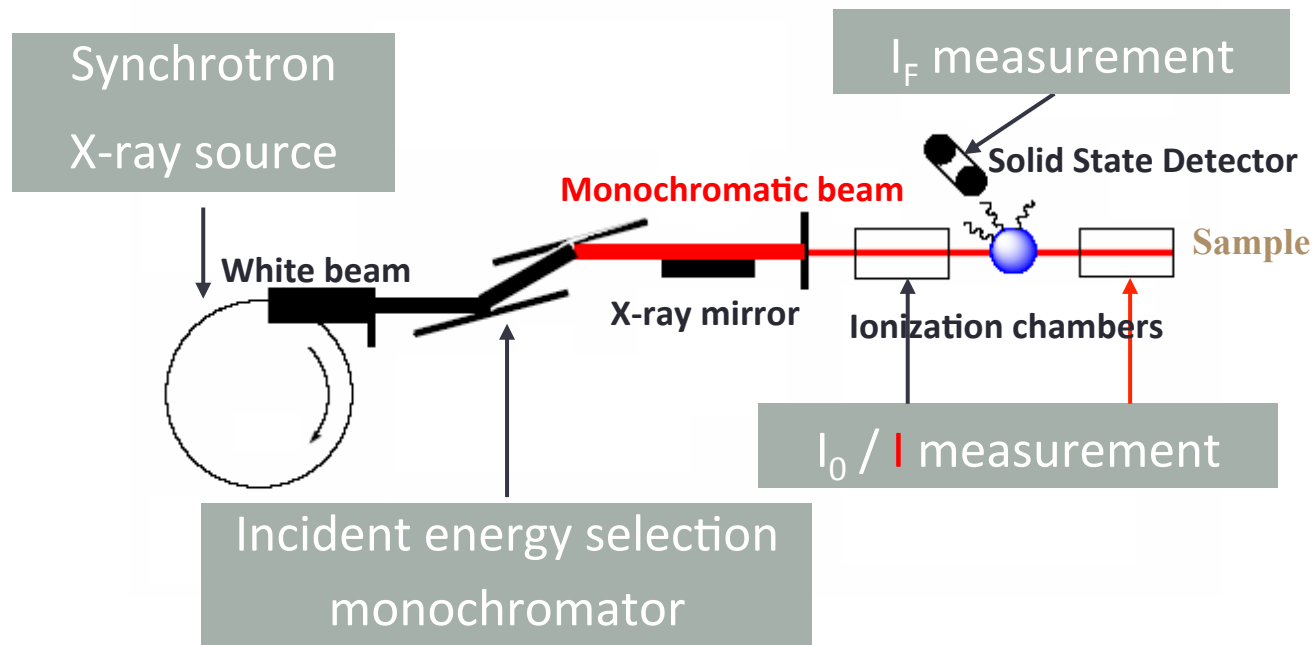
Horizon 2020: improve healthy ageing

The challenge is:
understand, control and (possibly) stop
PCD's evolution



X-ray Absorption Spectroscopy - XAS₁₄

Photons are produced by synchrotrons



X-ray Absorption Spectroscopy - XAS₁₅

- It may be used to study samples in any aggregation state (not necessarily crystallized, frozen or transparent)

proteins can be studied in physiological or controlled solution condition

- There are no selection rules for absorption

a XAS spectrum can always be measured

- Spectra registered in a relatively short time (few min's to few msec's)

limited radiation damage

- High chemical selectivity -

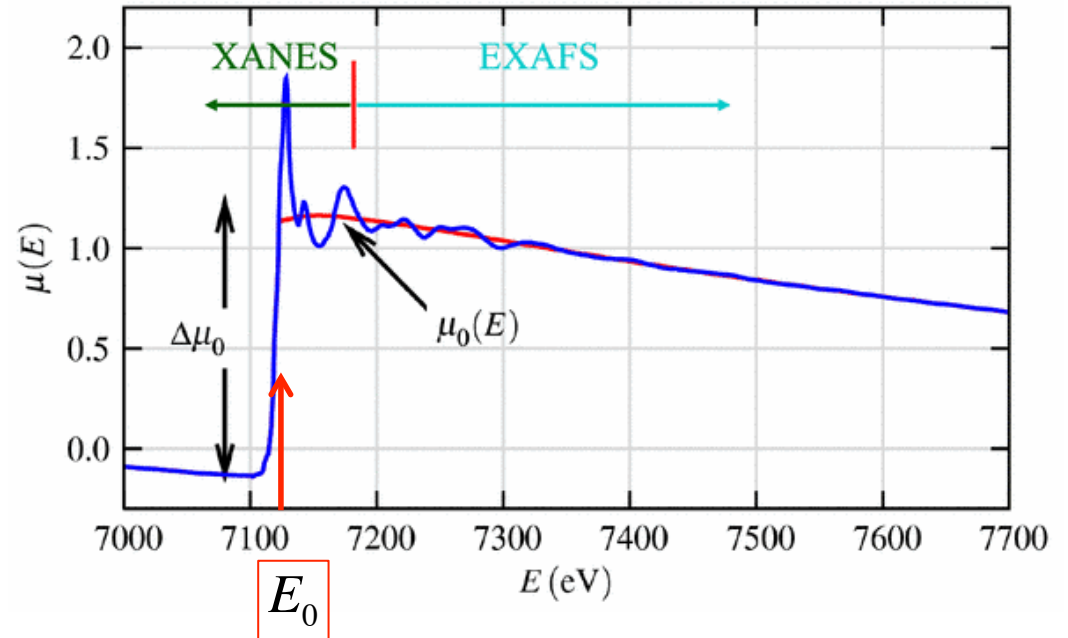
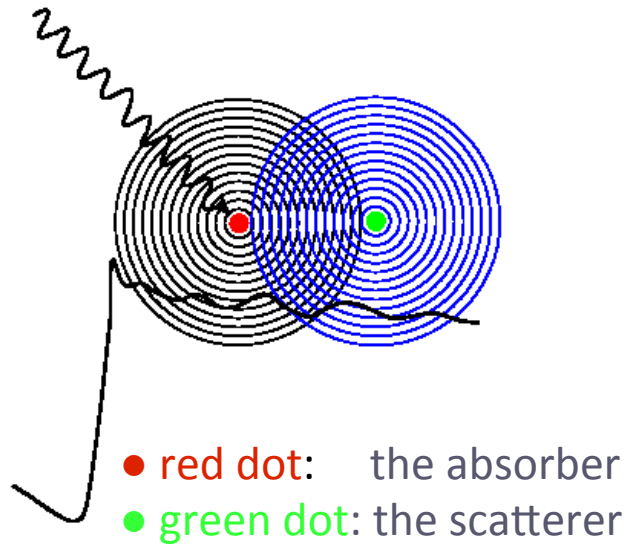
local atomic environment around a selected absorber

1s ionization energy (K-edge)

${}_{26}\text{Fe} \approx 7112 \text{ eV}$; ${}_{29}\text{Cu} \approx 8993 \text{ eV}$; ${}_{30}\text{Zn} \approx 9673 \text{ eV}$

XAS of a multi atomic system

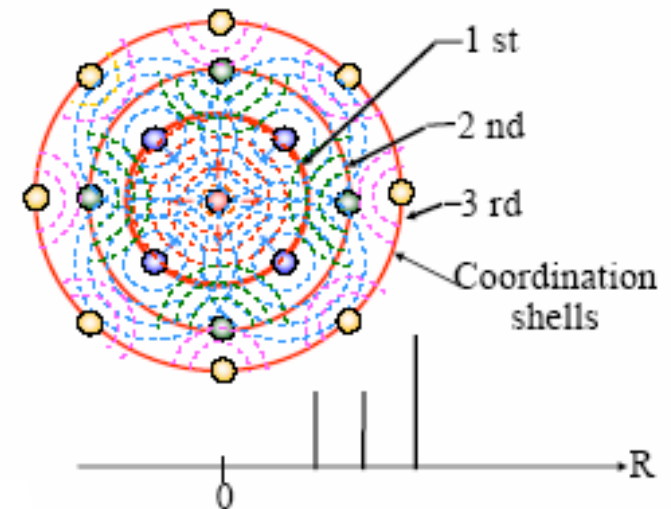
Absorption coefficient, μ , behaviour as a function of emitted photon energy



$\hbar\omega \gg E_0$ Single Scattering approximation

$$\chi(k) = \frac{\sigma_a - \sigma_0}{\sigma_0}$$

$$\chi(k) = S_0^2 \sum_i \frac{N_i}{kR_i^2} |f_i(k, \pi)| e^{-2k^2\sigma_i^2} e^{-2R_i/\lambda(k)} \sin[2kR_i + \Phi_i(k)]$$



Car Parrinello Molecular Dynamics based on *Density Functional Theory* and *Classical MD*
 CPMD DFT

Density Functional Theory

The electronic energy is a functional of the electronic density, $n(\vec{r})$

$$E_{DFT} = \Gamma_{HK}[n(\vec{r})] + \int n(\vec{r})V(\vec{r})d\vec{r}$$

that is at its minimum when $n(r)$ is the ground-state electronic density.

The latter is determined by solving the Kohn-Sham equations

$$\left[-\frac{\hbar^2}{2m} \nabla^2 + V(\vec{r}) + V_H(\vec{r}) + V_{xc}(\vec{r}) \right] \psi_i(\vec{r}) = \varepsilon_i \psi_i(\vec{r})$$

$$V_H(\vec{r}) = e^2 \int \frac{n(\vec{r}')}{|\vec{r} - \vec{r}'|} d\vec{r}' \quad \text{Hartree potential}$$

$$V_{xc}(\vec{r}) \quad \text{Exchange-correlation potential}$$

$$n(\vec{r}) = (2) \sum_i |\psi_i(\vec{r})|^2 \quad \langle \psi_i | \psi_j \rangle = \delta_{ij}$$

The simulation strategy



Quantum-ESPRESSO package
www.quantum-espresso.org

- 1 “prepare” the initial state of the system (classical MD)
 - type of atoms: number of electrons and bands → pseudopotentials
 - atomic coordinates
 - dimension of system cell
 - cut-offs → plane wave expansion
 - FFT grid spacing
- 2 electronic minimization (Steepest Descent, SD) with fixed ions → ground state
- 3 ionic positions minimization (SD or damped dynamics) → to relax the system
- 4 start the Car-Parrinello dynamics

NOTE: in order to reach the desired temperature
ions are coupled to Nosè-Hoover thermostat

**QM key feature: bonds not imposed (like in MD)
but derived from solving the Schroedinger equation**

1. Zn-ion-induced aggregation pattern 19

XAS allows to identify the Zn^{2+} -binding site sequence in $\text{A}\beta_{1-40}$

¹DAEFR⁶HDSGYEV¹³¹⁴HHQKLVFFAEDVGSNKGAIIGLMVGGV⁴⁰V

Besides the whole peptide, we measured the XAS spectra of the following samples

Zn^{2+} - $\text{A}\beta_{1-16}$



minimal fragment containing His₆, His₁₃, and His₁₄, suggested to be involved in metal binding

Zn^{2+} - $\text{A}\beta_{17-40}$



complementary to the previous sequence: no His is present

Zn^{2+} - $\text{A}\beta_{1-28}$



besides the three His's, it contains a long hydrophobic region believed to be relevant in the aggregation process

V. Minicozzi ... S.M. et al., *J Biol Chem* (2008) **283**: 10784

F. Stellato ... S.M. et al., *Eur Biophys J* (2006) **35**: 340

1° Zn-ion-induced aggregation pattern ²⁰

Synopsis of XAS results: comparing spectra

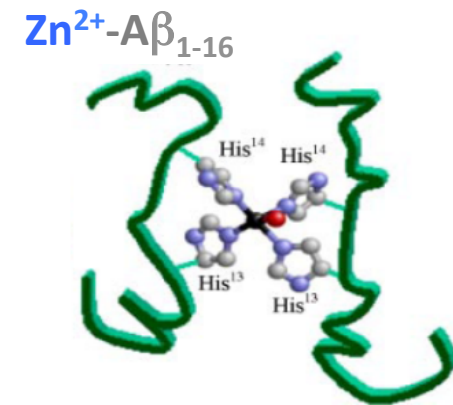
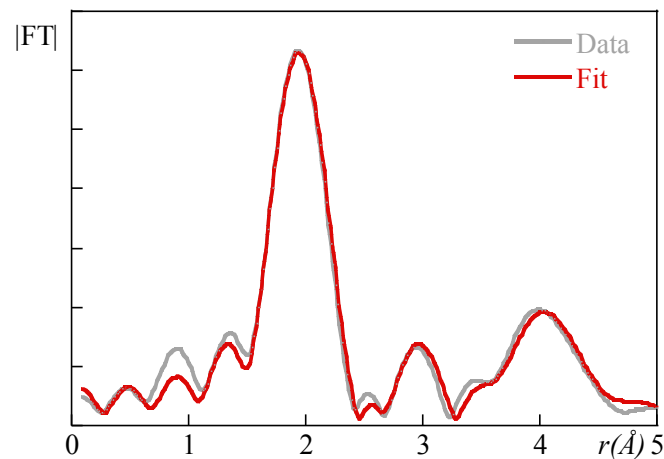
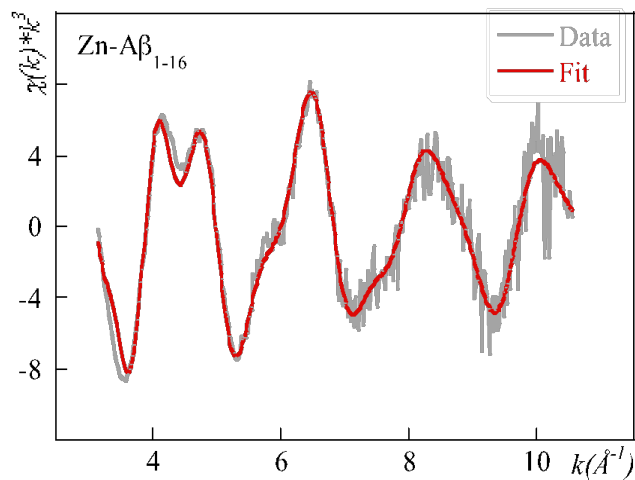
Zn^{2+} - $\text{A}\beta_{17-40}$ = Zn^{2+} - buffer \rightarrow Zn^{2+} is unable to bind $\text{A}\beta_{17-40}$

Zn^{2+} - $\text{A}\beta_{1-16}$ = Zn^{2+} - $\text{A}\beta_{1-28}$ = Zn^{2+} - $\text{A}\beta_{1-40}$ \rightarrow Zn^{2+} always bind to the first 16 a.a. fragment

1 DAEFR⁶HDSGYEV¹³¹⁴HHQKLVFFAEDVGSNKGAIIGLMVGGV⁴⁰V

Zn^{2+} always binds with the same (*very crowded*) binding structure to the first 16 a.a.'s

Zn^{2+} - $\text{A}\beta_{1-16}$



- 4 histidines
- 1 water

1° Zn-ions-induced aggregation pattern

Models and *ab-initio* calculations

Question:

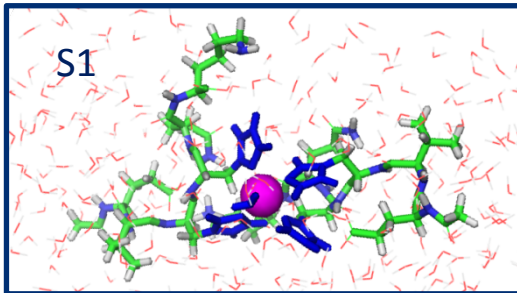
which structure stabilizes this (very crowded), 4 His's - Zn^{2+} coordination?

- XAS: structure information only within 5-6 Å from the absorber
- *ab-initio* simulations: structural information up to a longer distance

Model building

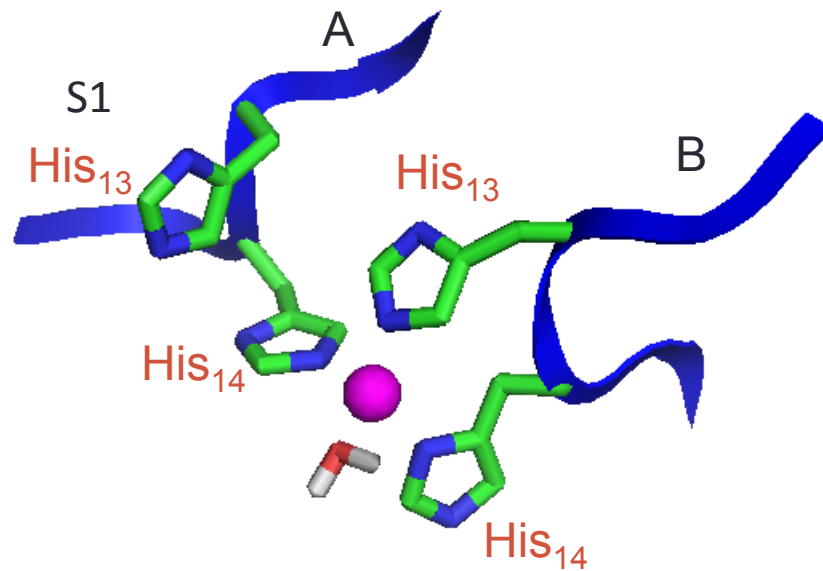
1. XAS \rightarrow 4 His's coordination \rightarrow two $\text{A}\beta_{1-16}$ -peptides (denoted A and B) with a single Zn atom bridging $\text{N}_\delta(\text{His}_{14}(\text{A}))$ and $\text{N}_\delta(\text{His}_{14}(\text{B}))$
2. minim + Monte Carlo Random Walk (MCRW) ("bad" contact discarded) \rightarrow only configurations that fulfill structural constraints of point 1. are retained
3. $\text{A}\beta_{1-16}$ truncated to $\text{A}\beta_{11-16}$ and put in a cell with periodic conditions filled with TIP3P water molecules
4. Four trial binding geometries are chosen for *ab initio* computations

1° Zn-ion-induced aggregation pattern²²



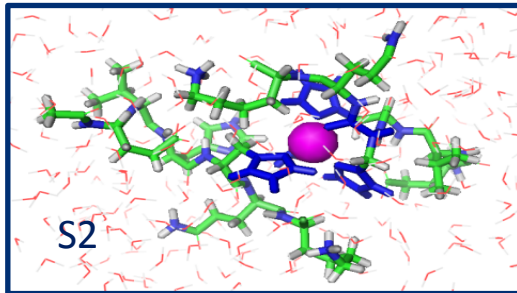
Besides the two His₁₄, also the two His₁₃ happen to be within 3 Å from Zn

Name	Starting configuration
S1	Zn: 4N _{His} + 10O _W



Name	Final configuration
S1	Zn: 3N _{His} + 10O _W

1° Zn-ion-induced aggregation pattern²³



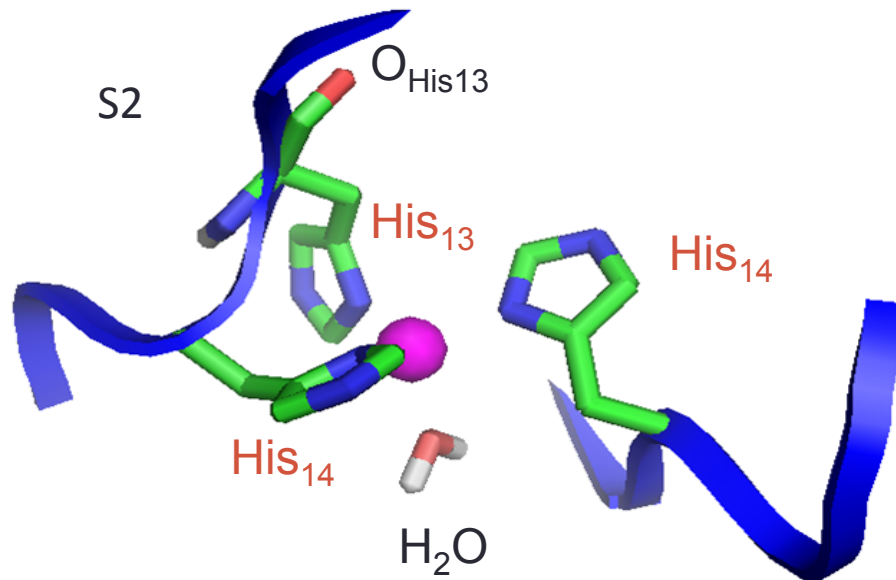
Only three His's. One of the His's is also through a main chain oxygen

Name

Starting configuration

S2

Zn: $3N_{\text{His}} + 1O_{\text{His}} + 1O_{\text{W}}$



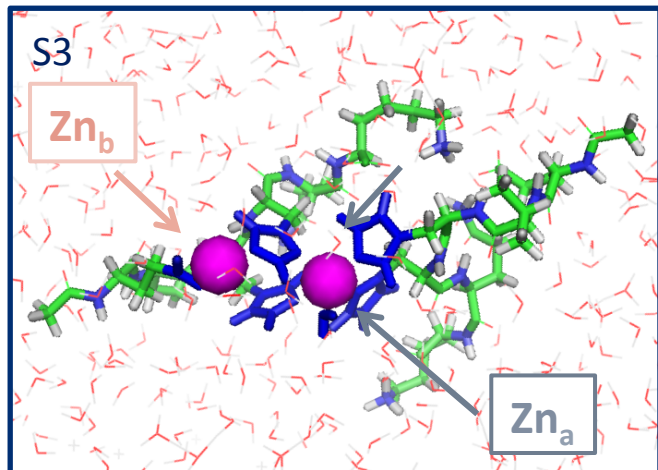
Name

Final configuration

S2

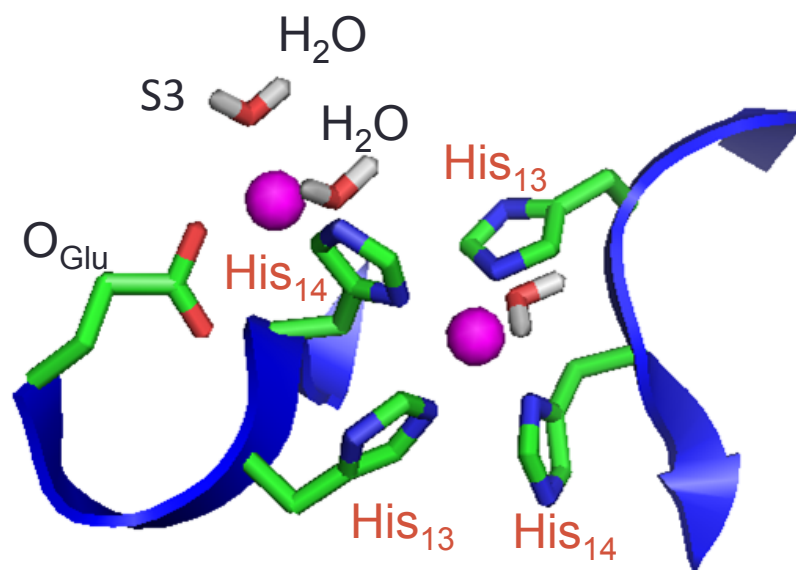
Zn: $3N_{\text{His}} + 1O_{\text{W}}$

1° Zn-ion-induced aggregation pattern²⁴



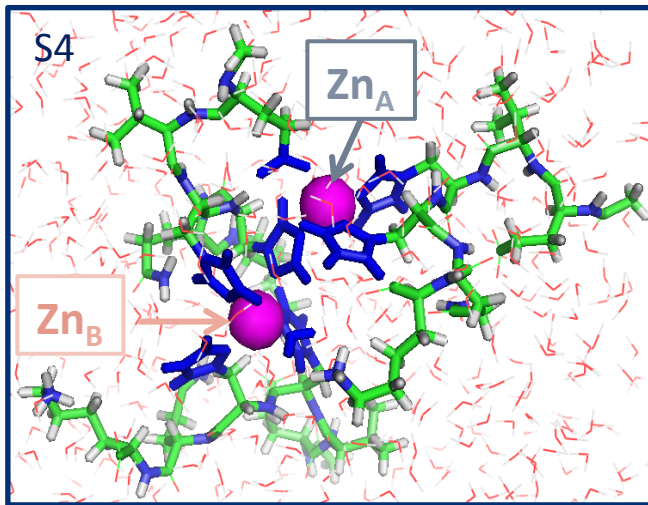
modified S1: the H ϵ of a His₁₄ is substituted by a second Zn

Name	Starting configuration
S3	$Zn_a: 4N_{His} + 1O_W$ $Zn_b: 1N_{His} + 2O_{Glu}$



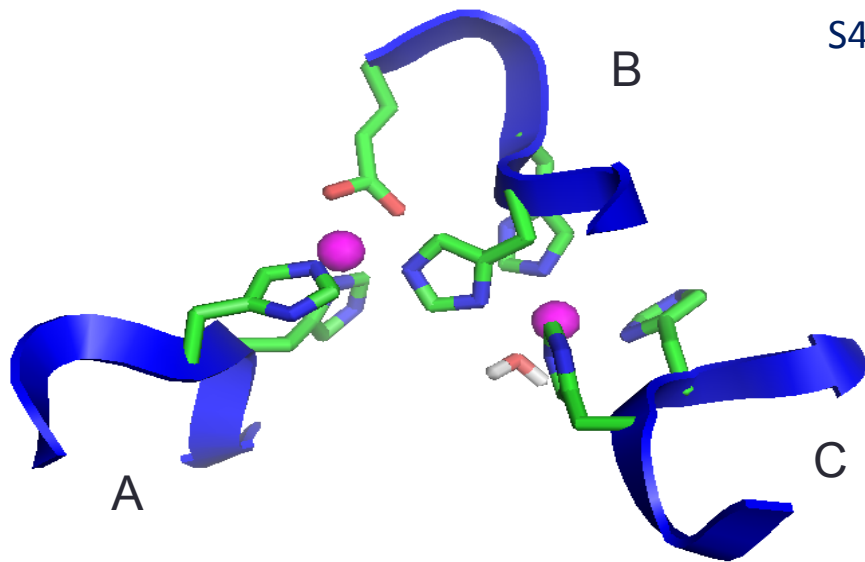
Name	Final configuration
S3	$Zn_a: 4N_{His} + 1O_W$ $Zn_b: 1N_{His} + 1O_{Glu} + 2O_W$

1° Zn-ion-induced aggregation pattern²⁵



modified S3: a third peptide is bound to the second Zn

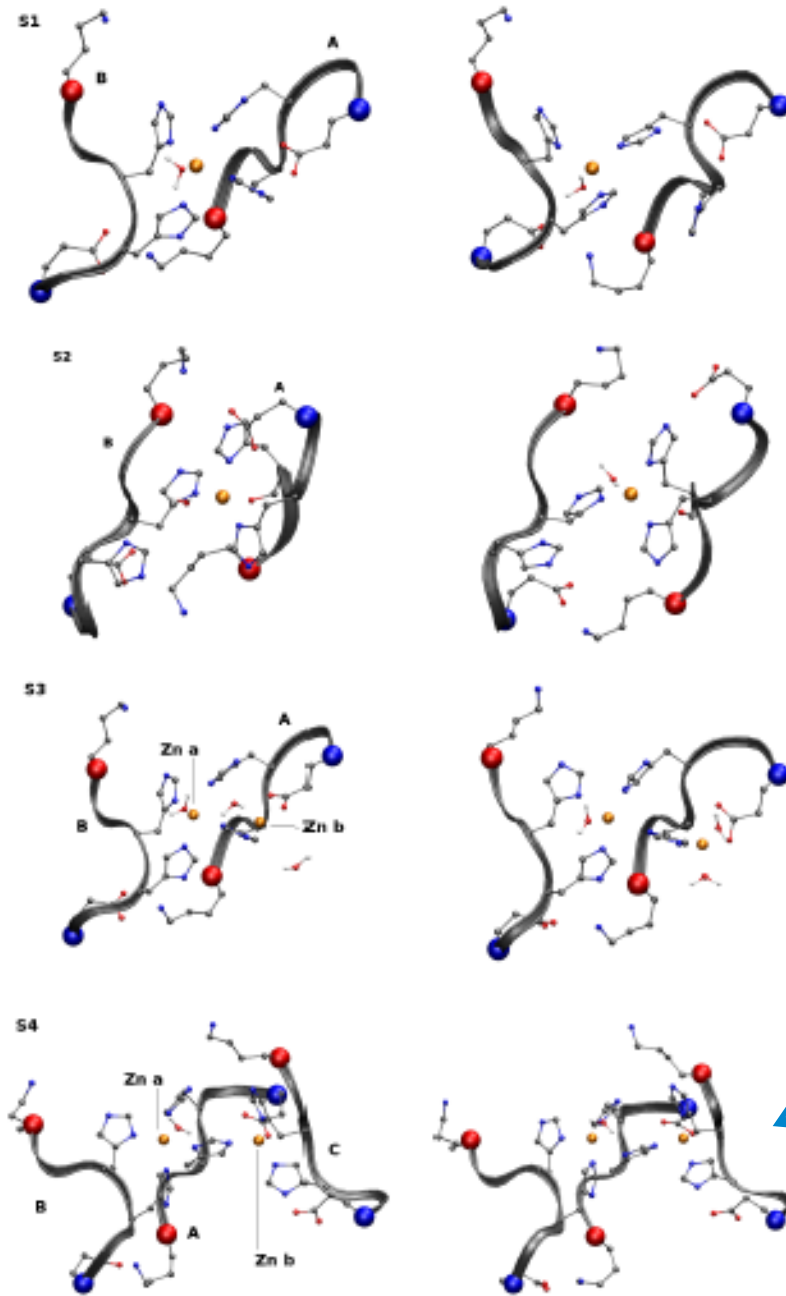
Name	Starting configuration
S4	$Zn_a: 4N_{His} + 1O_W$ $Zn_b: 3N_{His} + 1O_{Glu}$



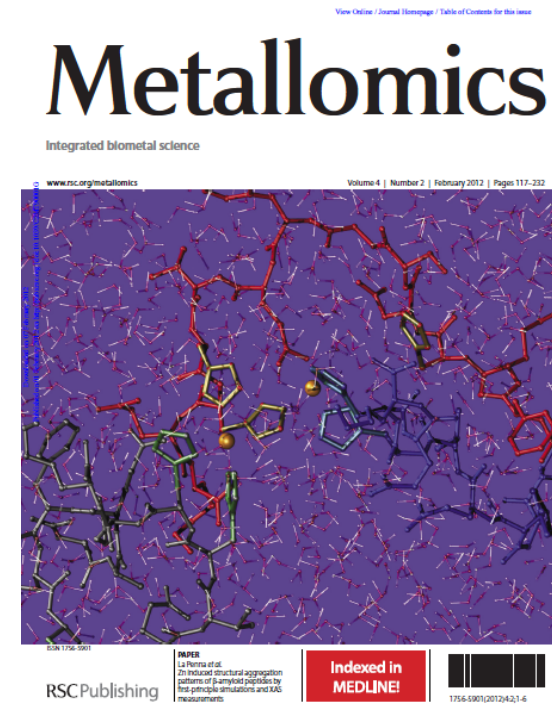
Name	Final configuration
S4	$Zn_a: 4N_{His} + 1O_W$ $Zn_b: 3N_{His} + 1O_{Glu}$

1° Zn-ion-induced aggregation pattern²⁶

P. Giannozzi, K. Jansen, G. La Penna, V. Minicazzi, S. M., G.C. Rossi, F. Stellato.
Metalomics (2012)



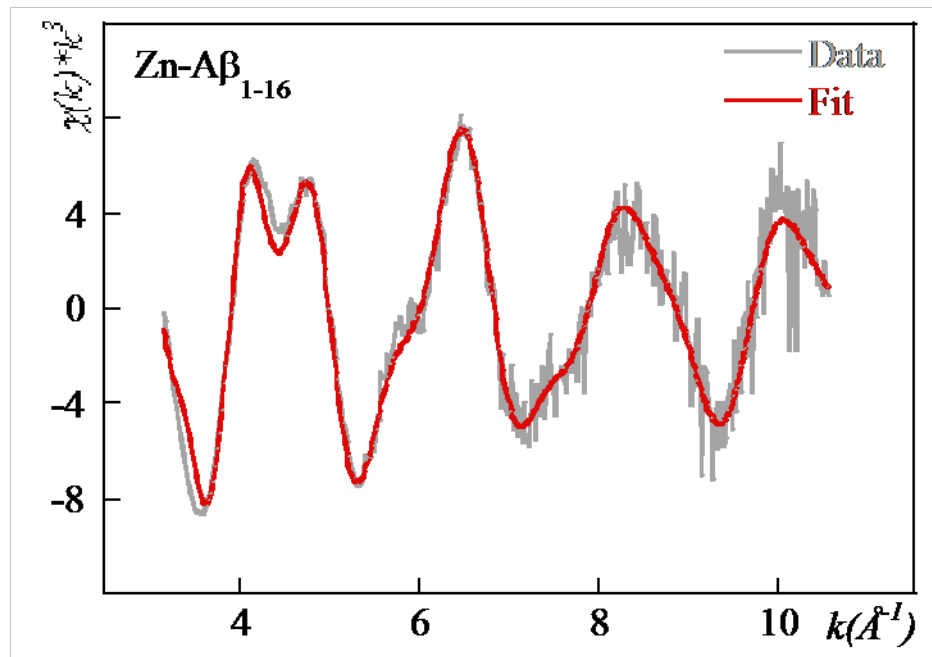
Zn site structures in the four models



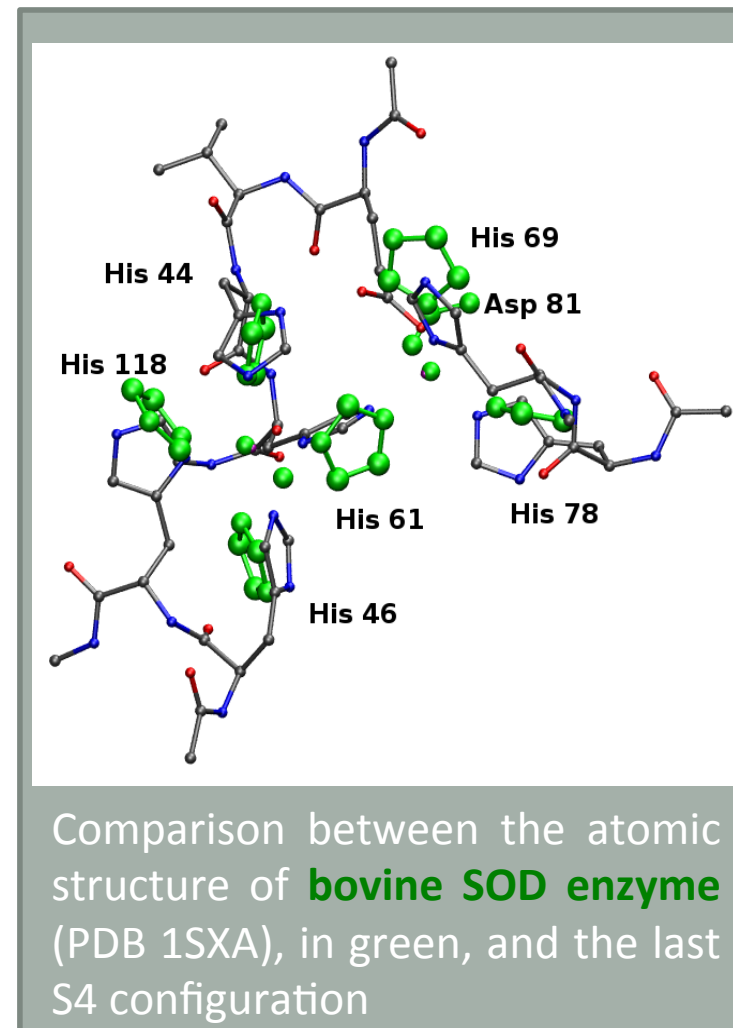
Our favorite model



1° Zn-ion-induced aggregation pattern²⁷



Comparing the experimental XAS spectrum with the theoretical one of the S4 model

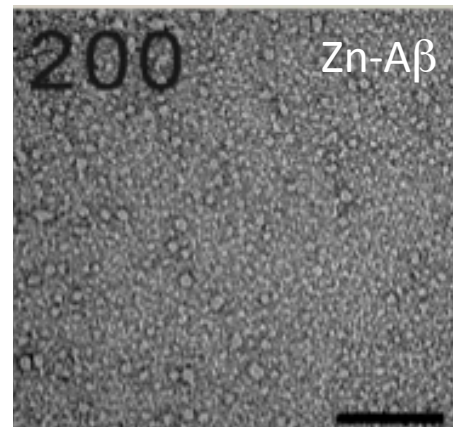
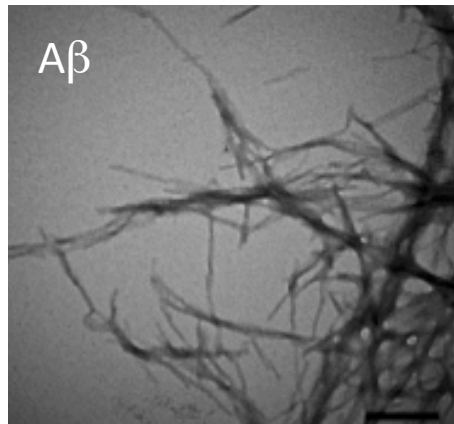


His's “crowding” in SOD enzymes is helped by the presence of two metal ions bridged by the His imidazole (in the unusual form of an imidazolate anion)

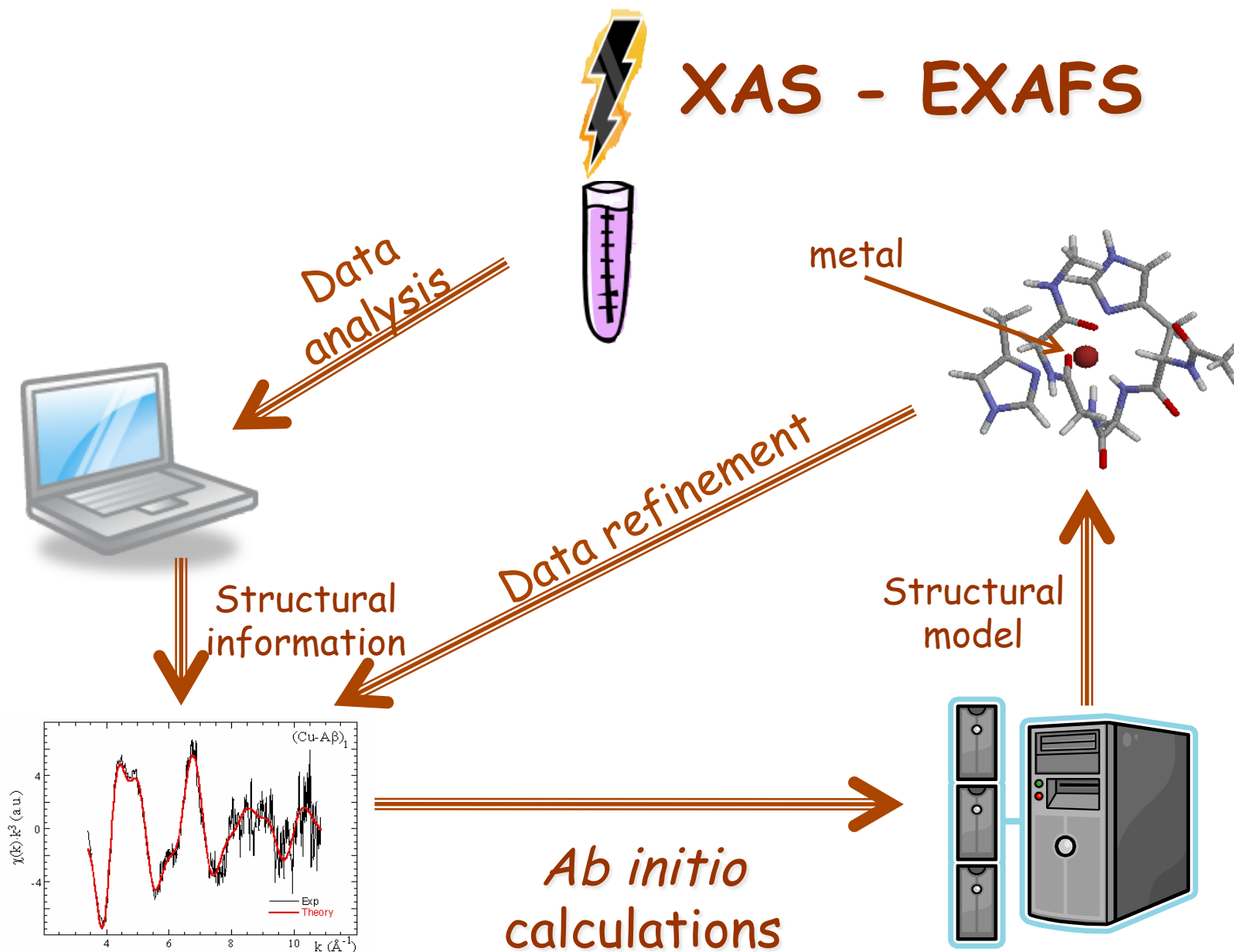
1° Zn-ion-induced aggregation pattern²⁸

Conclusions

- Zn promotes oligopeptides formation



- XAS cannot discriminate among different structural models, but
- empirical and first-principle simulations → stable Zn-His-Zn structure (\approx SOD)



2° - Cu²⁺-Zn²⁺ cross-modulation in A β binding

XAS measurements: A β + Zn and/or Cu \rightarrow they compete for the binding

De Santis, ... S.M. *J. Phys. Chem. B* (2015) accepted

It would be of the utmost importance to know what happens to the whole peptide structure as a result of metal binding competition ...

... we are planning to investigate this problem via MD and CPMD simulations

3° - Beta-sheet-breakers BSB's: targeting A β fibrils

MD of A β + BSB's that inhibit or delay A β conformational transition \rightarrow
we devised a new, more efficient, BSB

It would be of great interest to study how BSB's affect metal-A β interaction ...

... we are planning XAS of A β + BSB's in the presence of Cu and/or Zn

V. Minicozzi, ...S. M. *J. Biol. Chem.* (2014) **289**, 11242-11252

Acknowledgements

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biophys.roma2.infn.it

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Collaborators

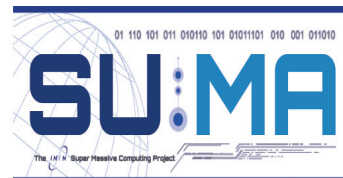
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Sunil Saxena (Uni Pittsburgh)

Ann Spevacek (UCSC)



Thank you