

Decay spectroscopy in the region of

neutron-rich lead isotopes

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Contents



The Z=82 and beyond N=126 region

Presence of isomers involving high-j orbitals $vg_{9/2}$, $vi_{11/2}$, $vj_{15/2}$. Taking advantage of these isomers we want to study the development of nuclear structure from ²¹²Pb up to ²¹⁸Pb and nearby nuclei



Experimental setup

FRS-Rising at GSI: stopped beam campaign



Nuclei populated in the fragmentation





A/Z

^{212,214,216}Pb: 8⁺ isomer



The shell model space



Experimental level schemes



Effective three-body forces

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New Isomers in the Full Seniority Scheme of Neutron-Rich Lead Isotopes: The Role of Effective Three-Body Forces

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Effective 3-body interaction: results



The explicit coupling to the core restores a seniority-like behaviour (midshell symmetry) and also justifies $e_v = 0.8e$ from the standard $e_v = 0.5e$

Mercury isotopes (I)



Mercury isotopes (II): ²¹⁰Hg



Lifetimes : a normal seniority scheme?



YES, but something more...



0.6(1)/0.11(4)

Isomeric ratio R

And what about the theory?

Shell model calculations already proved effective in the Pb isotopes and ²⁰⁸Hg (apart from the 3-body effect)



1- Very good agreement for the seniority scheme $8^+ \rightarrow 6^+ \rightarrow 4^+ \rightarrow 2^+ \rightarrow 0^+$ 2- B_{exp}(E2; $8^+ \rightarrow 6^+$) ~ $8 \cdot B_{theor}$ (E2; $8^+ \rightarrow 6^+$): effective 3-body forces as in ²¹²Pb 3- 3⁻ energy by particle-vibration coupling model (impossible by shell-model)

Conclusions

- Experiment with radioactive beam, with the in-flight technique. Several experimental challenges overcome. State-of-the-art experimental devices.
- The neutron-rich region along Z = 82 was populated, enabling to study the nuclear structure in this region up to now unknown due to experimental difficulties
- The observed shell structure seems to follow a seniority scheme. However, a closer look reveals that the B(E2) values have an unexpected behaviour. The mechanism of effective 3-body forces is general, and could be relevant also for other parts of the nuclide chart (Sn?, Ni?, Cd?).
- ²¹⁰Hg seems to present an unexpected deviation from standard theoretical predictions.

neutrons

Collaboration

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and ^{211,213}TI as well...



A glimpse of the other results



Bismuth isotopes (I)



Bismuth isotopes (II)



²¹³Pb: a strange case



$\gamma\gamma$ coincidences



Charge state identification

Formation of many charge states owing to interactions with materials

→Isotope identification is more complicated

➔ Need to disentangle nuclei that change their charge state after S2 deg.



²³⁸U charge-state problem



Nuclei populated in the fragmentation





FRS setting



Monochromatic setting for the S2 degrader:

Same velocity for all the ions with a certain Z



Some old good γ spectroscpy

A state with a µs lifetime decaying to 0⁺ must be 1⁺,2⁺,1⁻,3⁻

But in the systematics there are very well known limits for the various multipolarities

Energy	E,Mλ	$B(E,M\lambda)$ (W.u)	
663 keV	E1	$5(1) \cdot 10^{-10}$	
663 keV	E2	$2.7(7) \cdot 10^{-5}$	
663 keV	E3	2.7(7)	
663 keV	M1	$2.2(5) \cdot 10^{-3}$	
20 keV	M1	$7(2) \cdot 10^{-5}$	
20 keV	E1	$1.4(4) \cdot 10^{-6}$	



3⁻ seems the only resonable assumption...

The octupole states in the Pb region



²¹⁰Hg: a summary

An isomeric state must exist above the 2⁺, very close in energy: the 663-keV line may be this level. From all the experimental evidences it looks like it is 3⁻ state



From a theoretical point of view, no state is predicted so close to the first 2⁺ for a spherical nucleus

Isomeric ratios and other possibilities

Isomeric ratio R



Spin cut off-model: $R \downarrow exp3 - /R \downarrow exp8 \uparrow$ + =1.7

Maybe this 1306-keV is a 3⁻ or 5⁻ (at the right energy) BUT... (8+ 1306 1366 6 170 $B(E,M\lambda)$ (W.u) Energy E,M*λ* 1196 2 4+ ංං $5(1) \cdot 10^{-10}$ 663 keV E1 E2 $2.7(7) \cdot 10^{-5}$ 663 keV 553 Maybe we missed the E3 663 keV 2.7(7)643 (663) coincidence? There are 2^{+} 663 keV M1 $2.2(5) \cdot 10^{-3}$ few events... 643 Exp. W. 663 State Decays (keV) branch, ratio branch, ratio 0+. $\frac{E1(663)}{E1(110)} > 4$ $\frac{E1(663)}{E1(110)} = 191$ 3-663 keV to 2+ ²¹⁰Hg-Exp. 1306 keV 110 keV to 4⁺ $\frac{E3(663)}{E1(110)} > 4$ $\frac{E3(663)}{E1(110)} = 2 \cdot 10^{-4}$ 51 663 keV to 2⁺ 1306 keV 110 keV to 4⁺