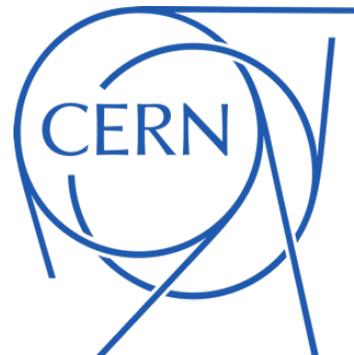




Campioni per Astrofisica



Meeting nTOF Italia. 29 Aprile 2025

Possibili campioni per misure di Astrofisica?

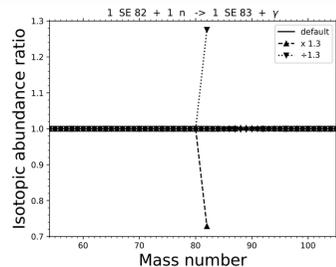
Tenuto conto delle indicazioni riportate da Diego nel corso del meeting di Novembre 2024, delle interlocuzioni con Emilio Maugeri e delle ricerche di mercato di Pino, ho ristretto la lista dei possibili campioni di interesse al seguente elenco:

- ^{82}Se
- ^{86}Sr
- ^{99}Tc
- ^{41}Ca
- ^{107}Pd
- ^{182}Hf
- ^{205}pb

⁸²Se

- ⁸²Se is a stable nucleus
- Its production during the s-process largely depends on the branching at ⁷⁹Se and ⁸¹Se
- Its solar value is mostly of r-process origin (88,9%; see Prantzos+ 20)
- It can be produced via the i-process
- Its abundance is mostly determined by the ⁸²Se(n,γ) (Pignatari+ 23, Martinet+ 24)

Pignatari+ 23, EPJA



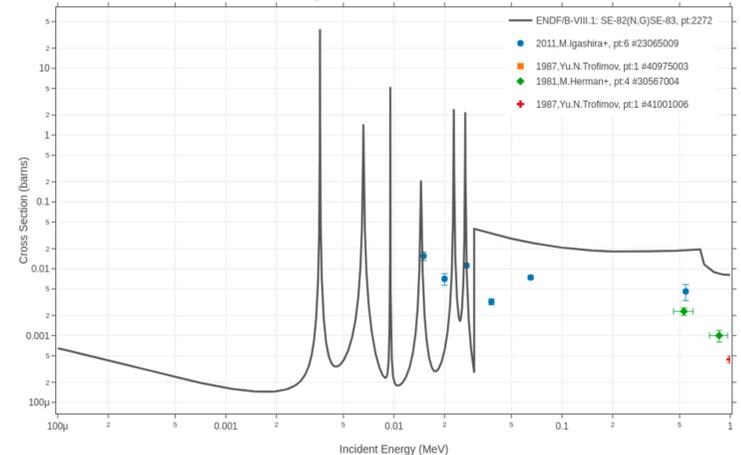
Br 79 50.69	Br 80 17.68 m	Br 81 49.31	Br 82 35.282 h	Br 83 2.374 h	Br 84 31.76 m	Br 85 2.90 m
Se 78 23.77	Se 79 327 ky	Se 80 49.61	Se 81 18.45 m	Se 82 8.73	Se 83 22.25 m	Se 84 3.26 m
As 77 38.79 h	As 78 90.7 m	As 79 9.01 m	As 80 15.2 s	As 81 33.3 s	As 82 19.1	As 83 13.4 s
Ge 76 7.73	Ge 77 11.211 h	Ge 78 88.0 m	Ge 79 18.98 s	Ge 80 29.5 s	Ge 81 8 s	Ge 82 4.56 s

Martinet+ 24, A&A

 Mild Impact
 Medium Impact
 High Impact

⁸²Se

- ⁸²Se MACS adopted in the astrophysical codes is theoretical



Isotopic composition (atomic fraction, %)

- Se-74: 0.10
- Se-76: 0.05
- Se-77: 0.04
- Se-78: 0.18
- Se-80: 1.32
- Se-82: 98.31

Da: ISOFLEX/RU <info@isoflex.com.ru>

Data: 16 dicembre 2024 alle ore 15:41:21 CET

A: giuseppe.tagliente@ba.infn.it

Oggetto: Re: Se-82, Sr-86 quotation

Dear Giuseppe,

Thank you very much for your new inquiry.

We can offer the following materials:

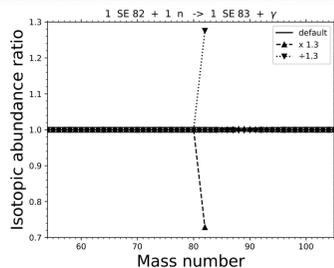
1000 mg quantity of Se-82 in elemental powder form with I.E.>98% at price:

US\$6.70 per mg. Delivery time: 7-9 weeks ARO

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- Its production during the s-process largely depends on the branching at ⁷⁹Se and ⁸¹Se
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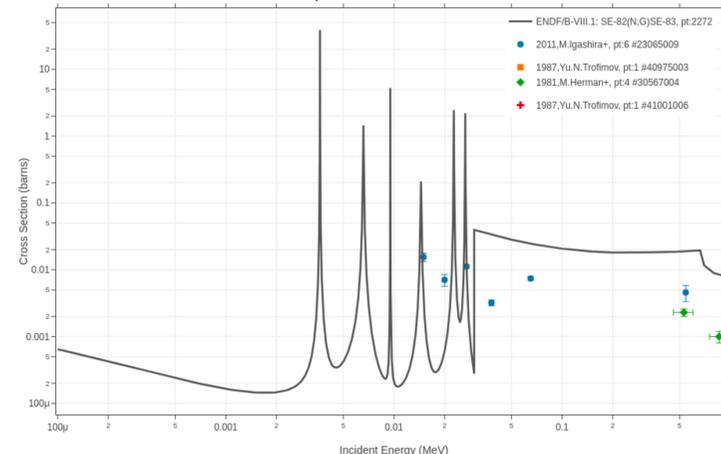
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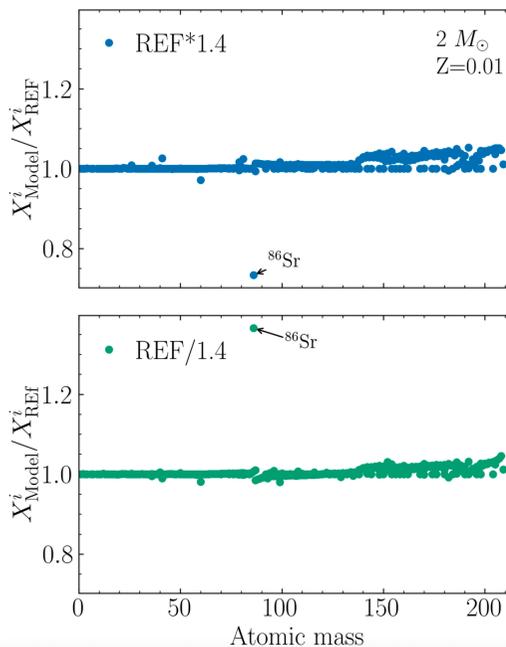
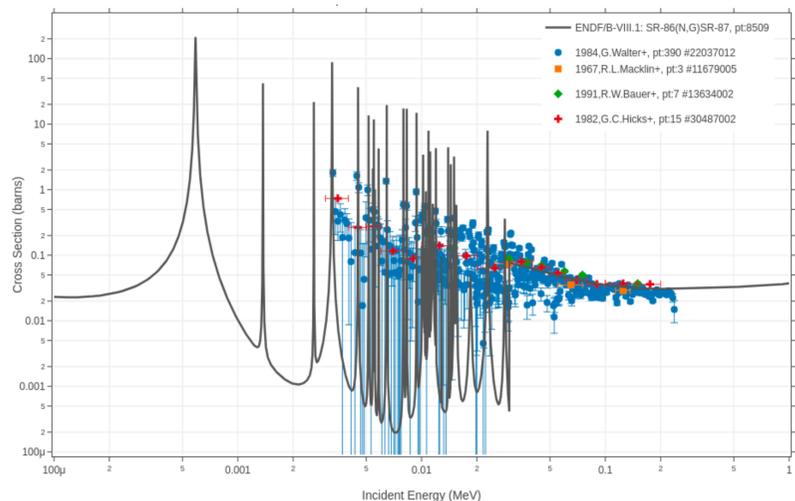




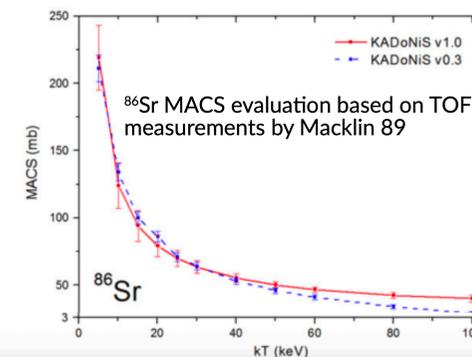
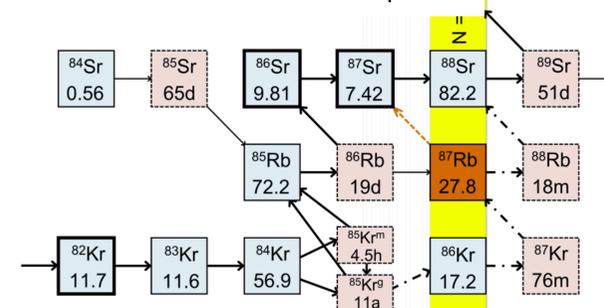
^{86}Sr



Tests varying the ^{86}Sr MACS by $\pm 40\%$



^{86}Sr is a reference term for Sr isotopic ratio



Item	Description	Quantity	Price per Unit (USD)	Value
0010	SR-86 Strontium 86 Chemical Form: Strontium carbonate Assay: 95.3000% Batch: 234201	500.000 MG	24.59	12,295.00
0020	ZZ-S01 EM Stable Isotope Packaging Fee	1.000 LOT	4,180.00	4,180.00
Quotation Total in Currency (USD)				16,475.00

National Isotope Development Center
Oak Ridge National Laboratory
P.O. Box 2008
Oak Ridge, TN 37831-6158

Customer Reference:

Quotation For: 999000
ISTITUTO NAZIONALE DI FISICA NUCLEARE
VIA ORABONA 4
70216 BARI BA BA
ITALY

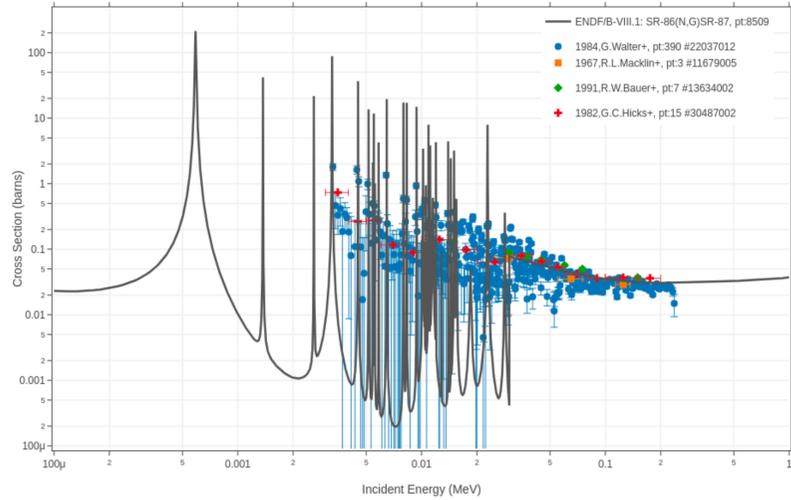
Isotope Quotation

Quotation	3011464
Date	02/13/2025
Quotation Period	02/13/2025 to 03/13/2025
Terms of Payment	Advance Payment Request
Terms of Delivery	FCA DOE Facility

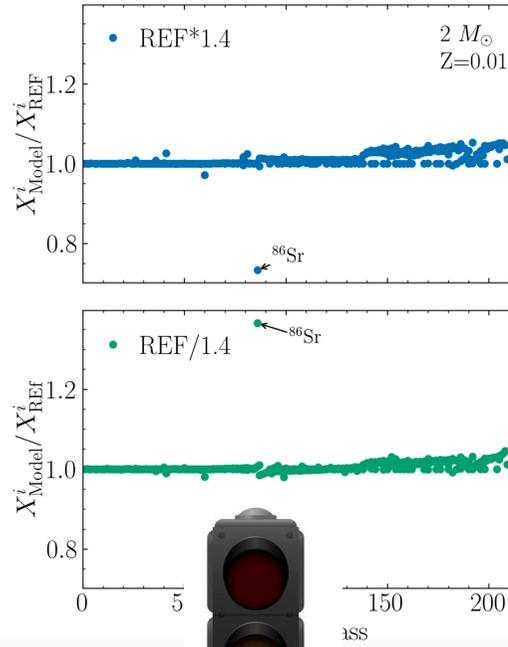
Refer Questions To:

Kim Bryant
Phone: 865-341-2916
Email: bryantka@ornl.gov

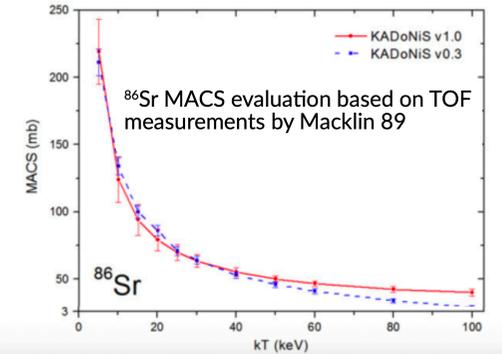
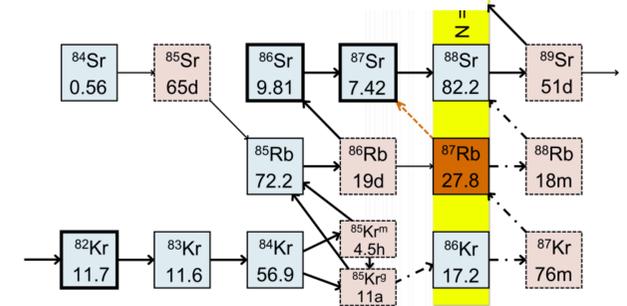
Venduto come SrCO_3
Emilio può convertirlo in SrO (o anche solfuro o cloruro)
Lievemente igroscopico, Emilio ha studiato una modalità di packaging, messa a punto per l'esperimento di Frank



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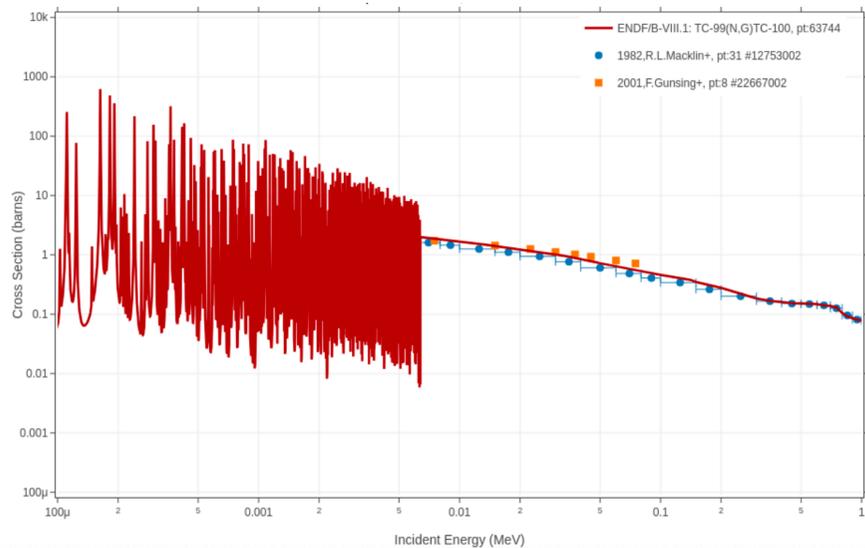
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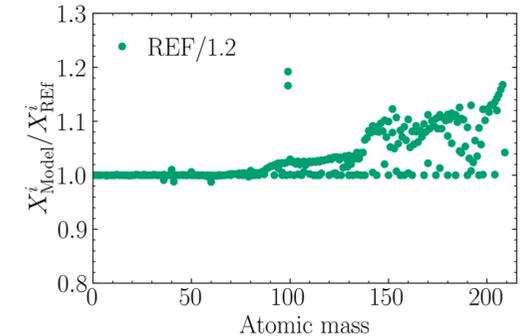
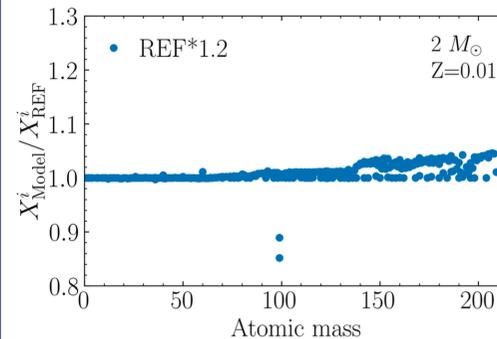
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▼ List of all available values

original	renorm.	year	type	Comment	Ref
933 ± 47		2000	c	Linac, TOF	GLM00, GLM01
782 ± 50		1987	c	Linac, TOF, ⁶ Li, Au:Sat.	WiM87
779 ± 40		1982	b	Linac, TOF, ⁶ Li, Au:Sat.	Mac82b



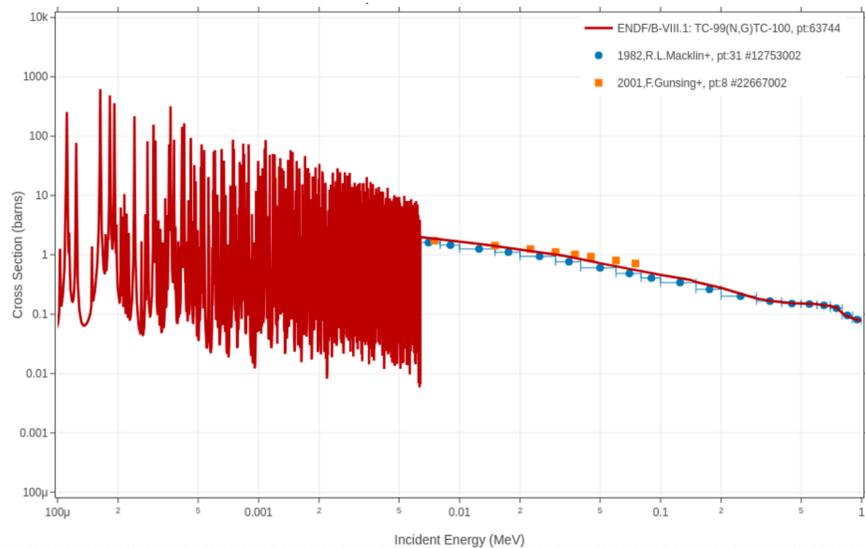
Tests varying the ⁹⁹Tc MACS by ±20%



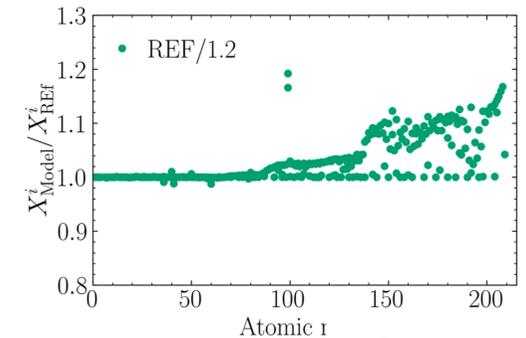
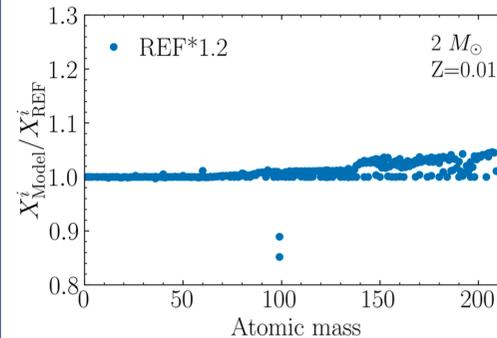
Produrlo sarebbe relativamente complesso. Pare che un Prof. di Zurigo ne abbia qualche mg già separato dal metastabile. In ogni caso la preparazione del sample richiede lavorazione con resine e complessivamente ci vorrebbero circa 10 kCHF per la produzione. Se ci convince, si può provare a contattare il Prof mediante Emilio (lui la manderebbe, ma la lettera la scriviamo noi). Se necessario potremmo coinvolgerlo nell'esperimento. Emilio suggerisce di provare a recuperarlo in ogni caso anche se non immediatamente interessati, perchè il prof andrà in pensione a breve.

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$^{41}\text{Ca}(n,p)$ & $^{41}\text{Ca}(n,\alpha)$

- ^{41}Ca is long-lived radioactive nuclei lighter than iron
- Important also for $^{41}\text{K}/^{39}\text{K}$ excesses measured in presolar grains from supernovae (Amari+ 96)
- Can be made by neutron captures on ^{40}Ca (half life of 0.1 Myr), which is stable and with relatively high solar abundance
- Neutron captures also destroy ^{41}Ca via different channels, the predominant being $^{41}\text{Ca}(n,\alpha)^{38}\text{Ar}$
- Experimental estimates for the neutron-capture cross section is available (e.g. de Smet et al. 2006),
- Electron-capture rate of ^{41}Ca is expected to vary significantly for different temperatures and densities relevant to stellar conditions
- Theoretical computations by Fuller+ 82

^{41}Ca



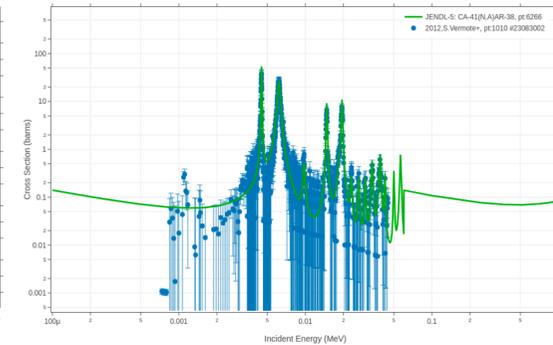
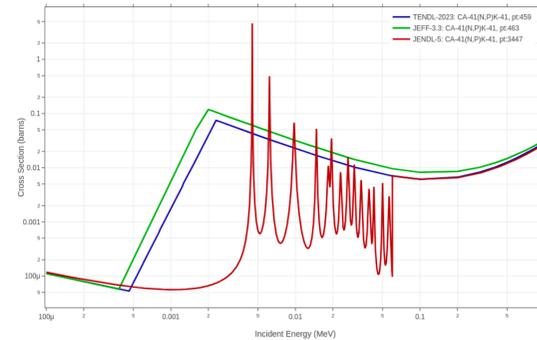
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- Thermal n-capture cross sections from Wagemans+ 98
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- Studied via inverse kinematics \rightarrow $^{38}\text{Ar}(\alpha,n)^{41}\text{Ca}$ and $^{38}\text{Ar}(\alpha,p)^{41}\text{K}$ (Talwar+ 18)

Lugaro+ 18, PrPN

^{40}Ca STABLE 96.94%	^{41}Ca 1.02E+5 Y ϵ : 100.00%	^{42}Ca STABLE 0.647%
^{39}K STABLE 93.2581%	^{40}K 1.248E+9 Y β^- : 89.28% ϵ : 10.72%	^{41}K STABLE 6.7302%
^{38}Ar STABLE 0.0632%	^{39}Ar 269 Y β^- : 100.00%	^{40}Ar STABLE 99.6003%

Reactions shown: n,γ (40Ca to 41Ca), n,α (41Ca to 38Ar), n,p (41Ca to 41K).



Definito da Emilio 'cammuriuso', come il ^{205}Pb . Sono entrambi 'difficult to measure' perche' EC senza gamma, e pertanto potrebbe essere un problema quantificarne l'ammontare. Si potrebbe procedere con cattura del ^{40}Ca , per quanto gli isotopi puri siano molto costosi. In tal caso il sample finale sarebbe $^{40}\text{Ca} + ^{41}\text{Ca}$, ossia 'no carrier free'. Si potrebbe anche procedere con il potassio 41 via (p,n), irraggiando per mesi. In tal caso risulta dispendioso fare la separazione, ma è l'unico modo per produrlo carrier free. Rimane comunque sempre il problema della misura. La realizzazione richiede comunque un progetto di 1.5 – 2 anni. E costi importanti dell'ordine di circa 50kCHF (da verificare).



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^{41}Ca

$^{41}\text{Ca}(n,p)$ & $^{41}\text{Ca}(n,\alpha)$

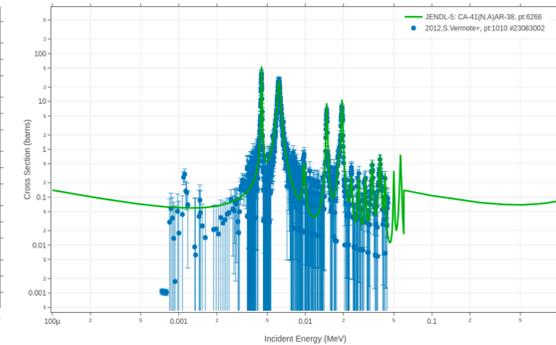
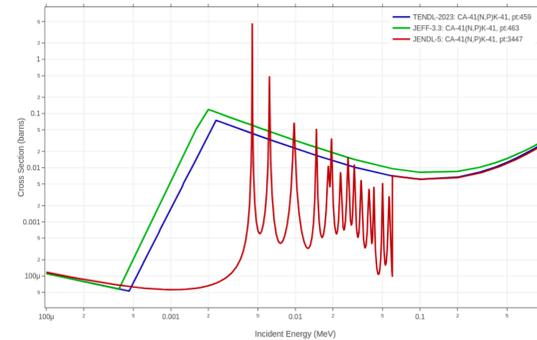


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Disponibile a Geel...





¹⁰⁷Pd

- Too long-living ($T_{1/2} = 6.5$ Myr, down to 700 years at 300 MK) to act as a branching point during the s-process
- Behaves as a stable nucleus
- Experimentally determined neutron-capture cross section (Macklin 1985)
- Its radiogenic decay is responsible for production of ¹⁰⁷Ag

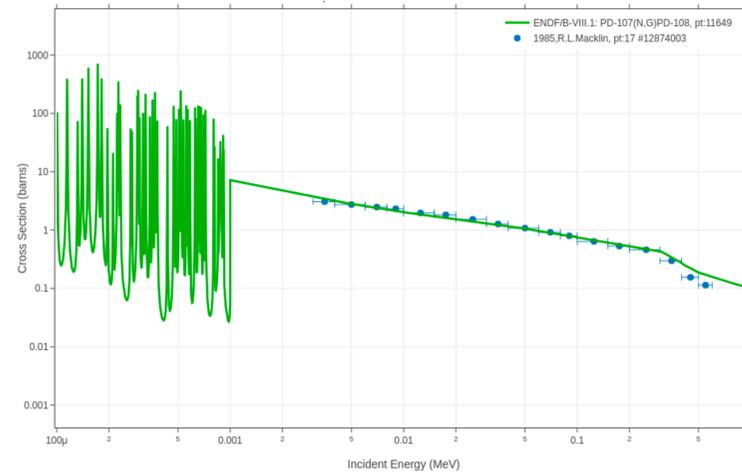
Lugaro+ 18, PrPNP

¹⁰⁷ Ag STABLE 51.839%	¹⁰⁸ Ag 2.37 M β^- : 97.15% ϵ : 2.85%	¹⁰⁹ Ag STABLE 48.161%	¹¹⁰ Ag 24.6 S β^- : 99.70% ϵ : 0.30%
¹⁰⁶ Pd STABLE 27.33%	¹⁰⁷ Pd 6.5E+6 Y β^- : 100.00%	¹⁰⁸ Pd STABLE 26.46%	¹⁰⁹ Pd 13.7012 H β^- : 100.00%
¹⁰⁵ Rh 35.36 H β^- : 100.00%	¹⁰⁶ Rh 29.80 S β^- : 100.00%	¹⁰⁷ Rh 4.99 S β^- : 100.00%	¹⁰⁸ Rh 16.8 S β^- : 100.00%

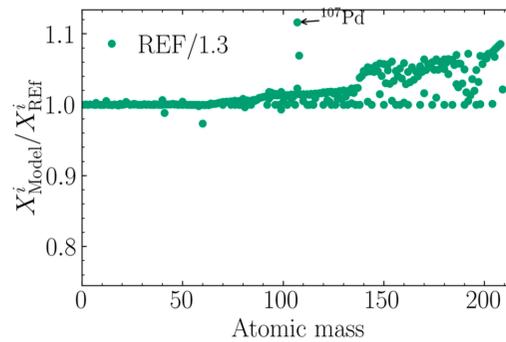
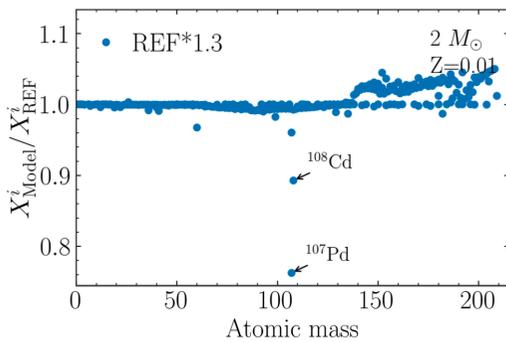
Annotations:
 - White arrow: ¹⁰⁶Pd → ¹⁰⁷Pd (n,γ)
 - Red arrow: ¹⁰⁷Pd → ¹⁰⁸Pd (n,γ)
 - Green arrow: ¹⁰⁷Pd → ¹⁰⁷Ag (β decay following the s-process)

¹⁰⁷Pd

Experimentally determined neutron-capture cross section (Macklin 1985)



Tests varying the ¹⁰⁷Pd MACS by ±30%



Si potrebbe produrre tramite cattura dal ¹⁰⁶Pd, con irraggiamento a ILL che richiederebbe decine di giorni facendo uso di un campione con purezza superiore al 99%. A titolo di esempio, se si irraggiano dentro il reattore 3g di ¹⁰⁶Pd naturale (27%), con flussi fino a 10¹⁵ n/sec, dopo 55gg di irraggiamento si raggiungono circa 10¹⁹ atomi.



¹⁰⁷Pd

- Too long-living ($T_{1/2} = 6.5$ Myr, down to 700 years at 300 MK) to act as a branching point during the s-process
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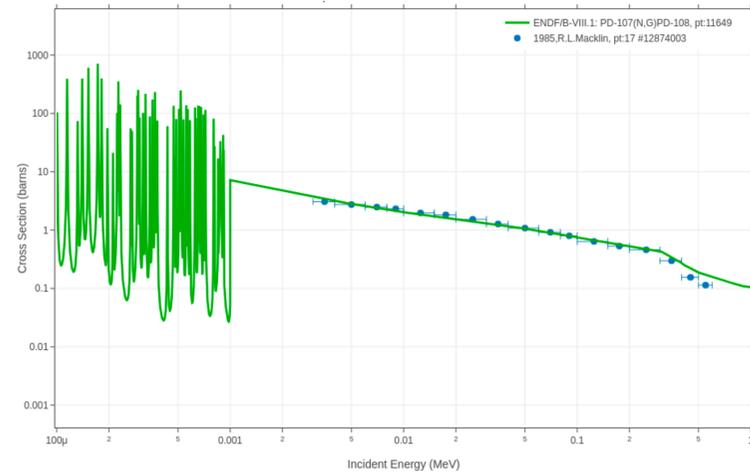
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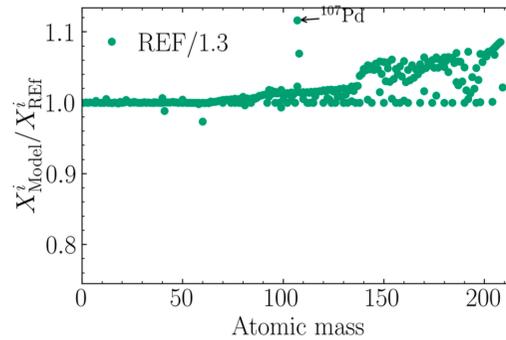
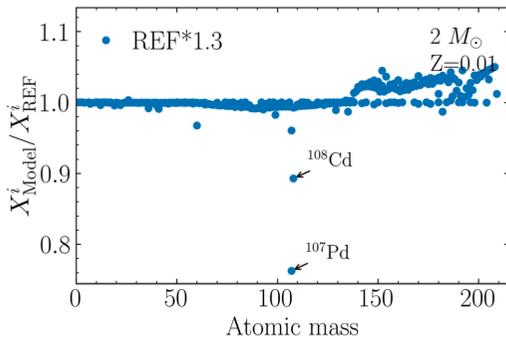
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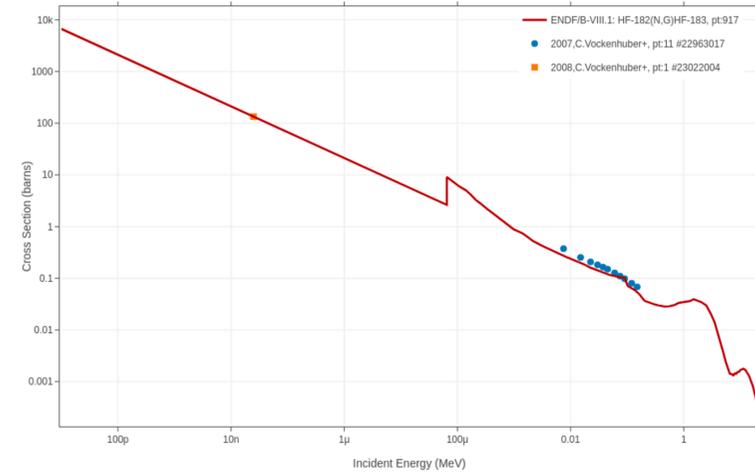


- Production of ^{182}Hf via activation of the branching point at ^{181}Hf (Lugaro+ 18, PrPNP)
- The half life of ^{181}Hf , is believed to strongly decrease from 42 days to ~ 2 days in stellar conditions, mostly via population of an excited state at 68 keV (TY 87)
- However, the more recent, detailed experiments of Bondarenko+ 02 on the nuclear structure of ^{181}Hf suggested that this energy level does not exist
- Possible large production of ^{182}Hf to the s-process in AGB stars
- This may resolve the discrepancy between the abundances of ^{129}I and ^{182}Hf in the early solar system and allow to time the latest r- and s-process events that contributed to the build-up of solar system matter before the formation of the Sun (see discussion in Lugaro+ 14 and Vescovi+ 18)
- Its decay into ^{182}W is of importance also for determining the solar r-process residual of this isotope

^{182}W STABLE 26.50%	^{183}W STABLE 14.31%	^{184}W STABLE 30.64%	^{185}W 75.1 D β^- : 100.00%
^{181}Ta STABLE 99.988%	^{182}Ta 14.43 D β^- : 100.00%	^{183}Ta 5.1 D β^- : 100.00%	^{184}Ta 8.7 H β^- : 100.00%
^{180}Hf STABLE 35.08%	^{181}Hf 42.59 D β^- : 100.00%	^{182}Hf $3.90\text{E}+6$ Y β^- : 100.00%	^{183}Hf 1.067 H β^- : 100.00%

β^- decay following the r process
 n, γ

Experimentally determined n-capture cross section via activation technique (Vockenhuber+ 07)



Partendo da ^{180}Hf mediante $2n$ capture. Con 2.5g ci vorrebbe circa 1 mese di irraggiamento a ILL, usando un campione arricchito

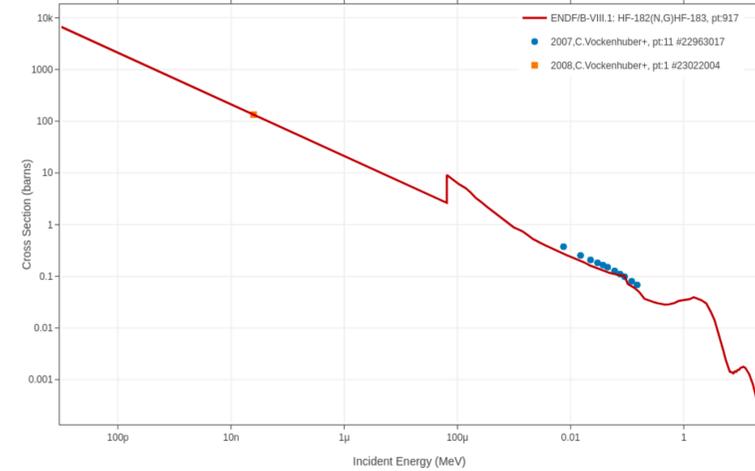


- Production of ^{182}Hf via activation of the branching point at ^{181}Hf (Lugaro+ 18, PrPNP)
- The half life of ^{181}Hf , is believed to strongly decrease from 42 days to ~ 2 days in stellar conditions, mostly via population of an excited state at 68 keV (TY 87)
- However, the more recent, detailed experiments of Bondarenko+ 02 on the nuclear structure of ^{181}Hf suggested that this energy level does not exist
- Possible large production of ^{182}Hf to the s-process in AGB stars
- This may resolve the discrepancy between the abundances of ^{129}I and ^{182}Hf in the early solar system and allow to time the latest r- and s-process events that contributed to the build-up of solar system matter before the formation of the Sun (see discussion in Lugaro+ 14 and Vescovi+ 18)
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Arrows indicate transitions: $^{180}\text{Hf} \xrightarrow{n,\gamma} ^{181}\text{Hf} \xrightarrow{\beta^-} ^{182}\text{W}$ and $^{181}\text{Hf} \xrightarrow{n,\gamma} ^{182}\text{Hf} \xrightarrow{\beta^-} ^{182}\text{W}$.

Experimentally determined n-capture cross section via activation technique (Vockenhuber+ 07)



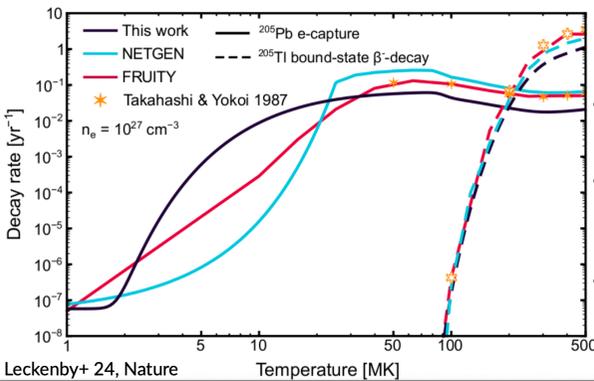
Partendo da ^{180}Hf mediante 2n capture. Con 2.5g ci vorrebbe circa 1 mese di irraggiamento a ILL, usando un campione arricchito



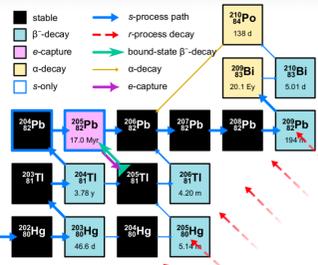


²⁰⁵Pb

- Despite its long terrestrial half-life ($T_{1/2} = 17$ Myr) of ²⁰⁵Pb acts as a **branching point** because of the strong dependence on temperature and electron density
- ²⁰⁵Tl becomes unstable during TPs and its β^- decay is competing with the β^+ decay of ²⁰⁵Pb



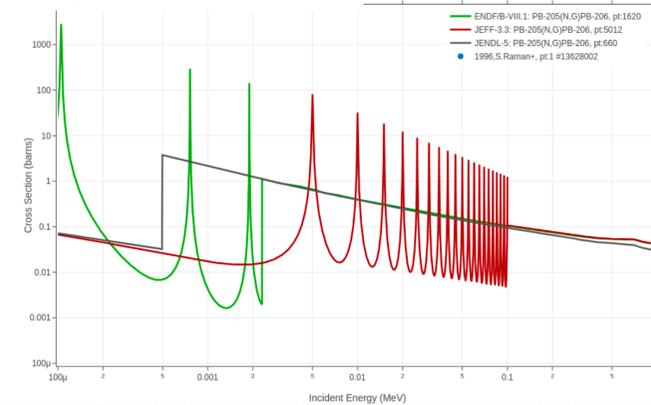
Leckenby+ 24, Nature



- Measured for the first time the bound-state β^- decay of ²⁰⁵Tl
- The measured half-life is **4.7 times larger** than the previous theoretical estimate (291 days vs. 58 days)
- **Diverging behavior at low temperatures** due to the different extrapolation to the terrestrial value (log versus linear)

original	renorm.	year	type	Comment	Ref
54 ± 12		1976	s		MaW76
102		2000	t		RaT99
83		1981	t		Har81
58		1976	t		HWF76
65.8		2002	t	MOST 2002	Gor02
81.0		2005	t	MOST 2005	Gor05

- Theoretical (n, γ) cross section

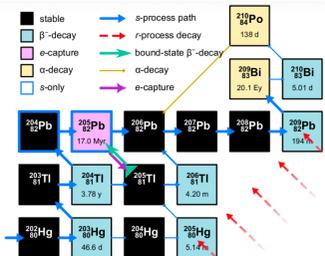


Al PSI vi è in embrione un progetto per produrre ²⁰⁵Pb. Si potrebbe trovare una forma di collaborazione mediata da Emilio, per quanto non sia semplice. Si può produrre partendo dal ²⁰⁴Pb arricchito, irraggiandolo per due mesi, e per via dei contaminanti occorre poi separarlo. Costi elevati solo per la produzione. In ogni caso il sample finale sarebbe 'no carrier free', ossia ²⁰⁴Pb + ²⁰⁵Pb. Ci si potrebbe anche arrivare dal tallio non arricchito (²⁰³+²⁰⁵) con la (p,n) dopo lungo irraggiamento e aspettare due settimane per far decadere i prodotti. La separazione è un problema perché no gamma emettitore. Alla fine comunque dovrebbe essere separato con uno spettrometro di massa. Come per il ⁴¹Ca, occorrono oltre 50kCHF e ci vuole un progetto.

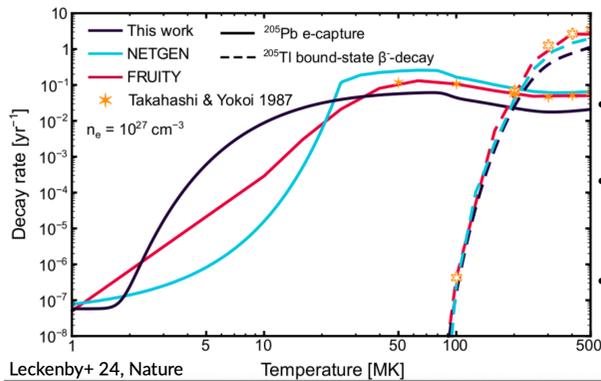


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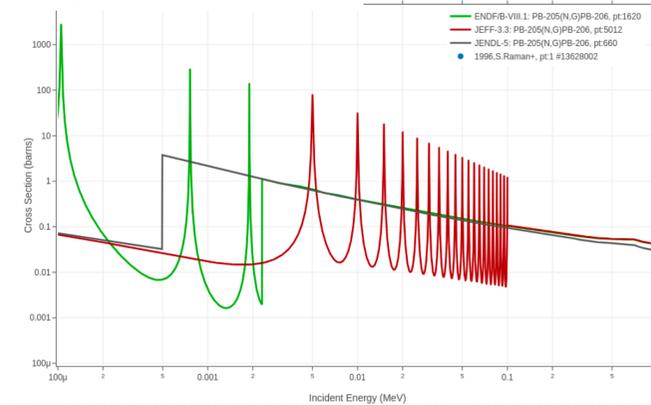


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