

r-process nucleosynthesis and Kilonovae in the PANDORA project

S. Cristallo

in collaboration with M. Bezmalinovich, D. Vescovi, E. Loffredo



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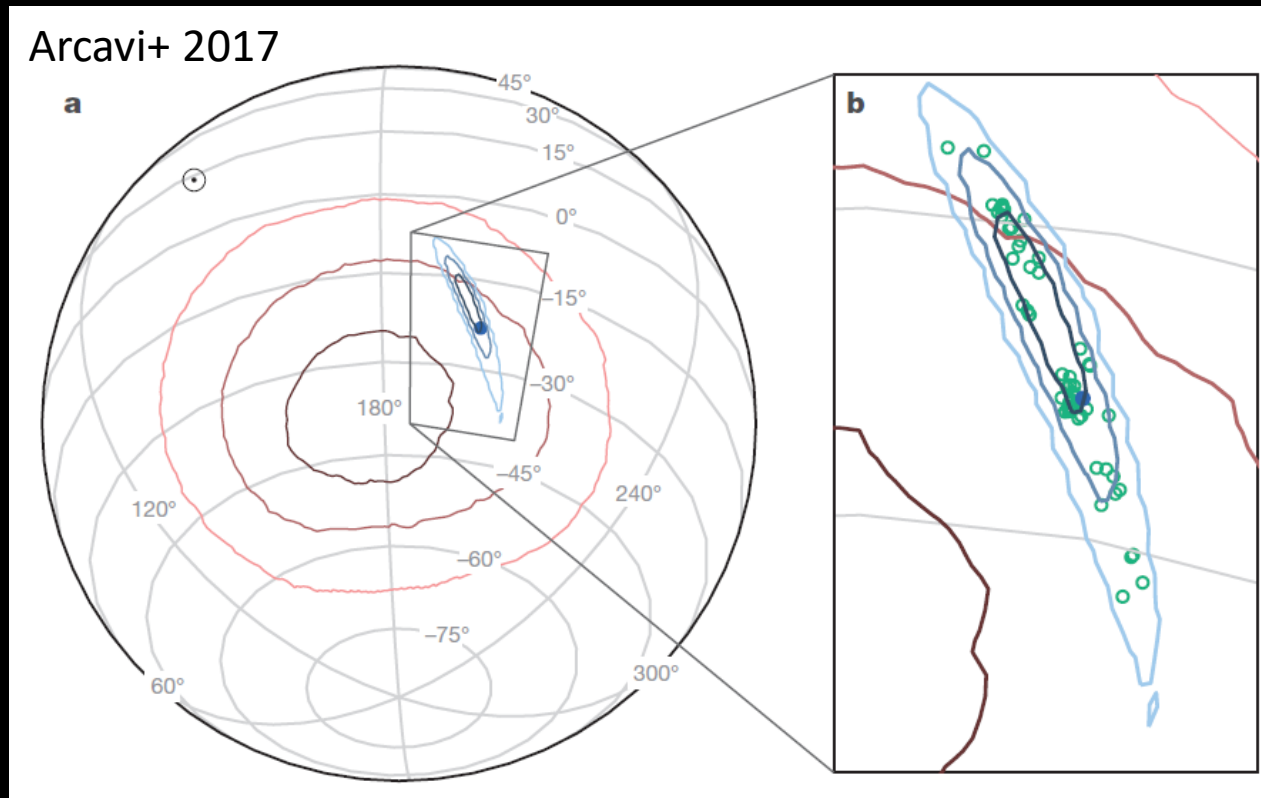
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GW170817: «THE» gravitational signal

A revolution similar to Jupiter observations by Galileo...



Hosting galaxy: NGC 4993 (distance ≈ 44 Mpc)



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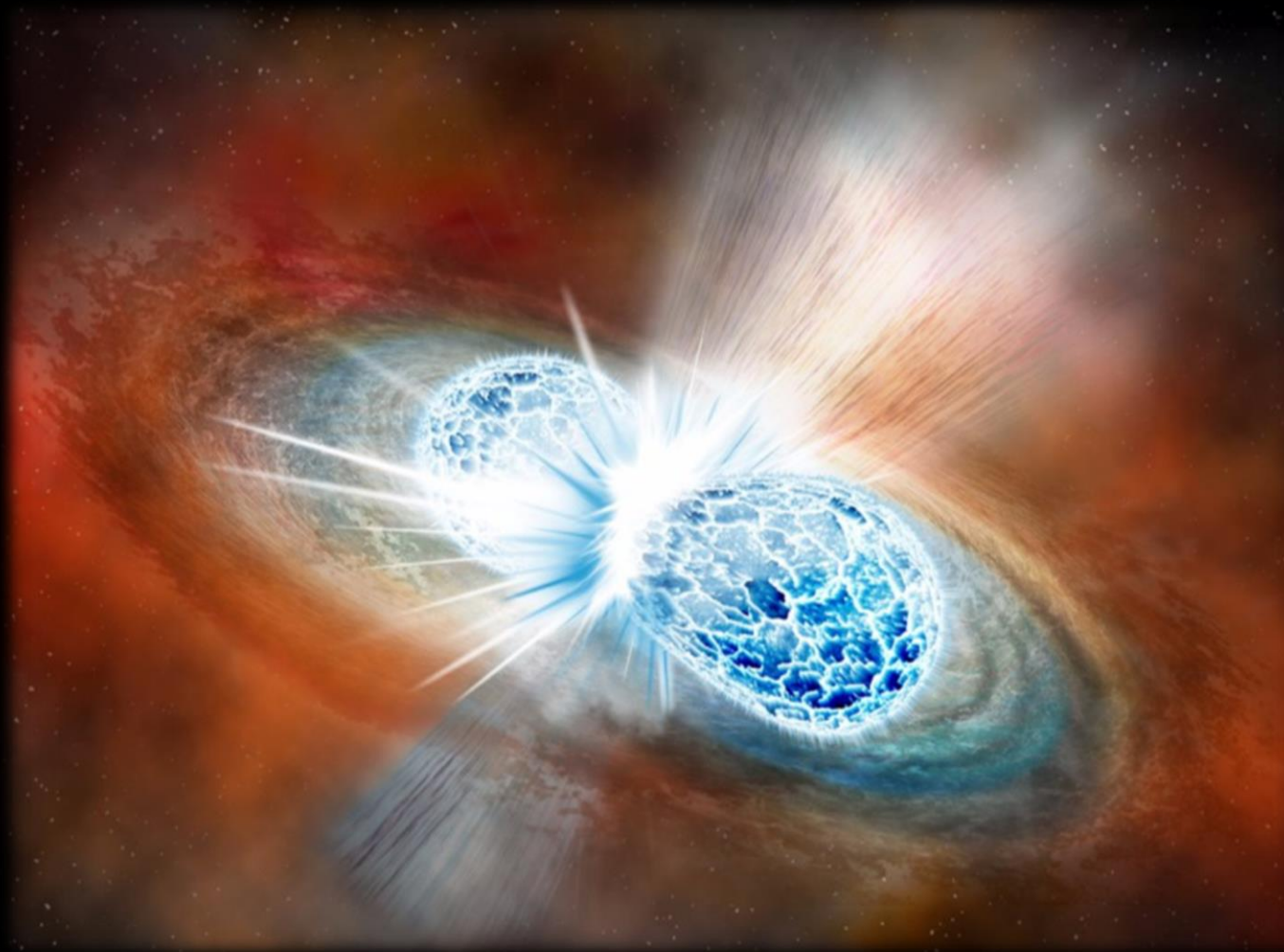


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Neutron Star Mergers (NSMs)





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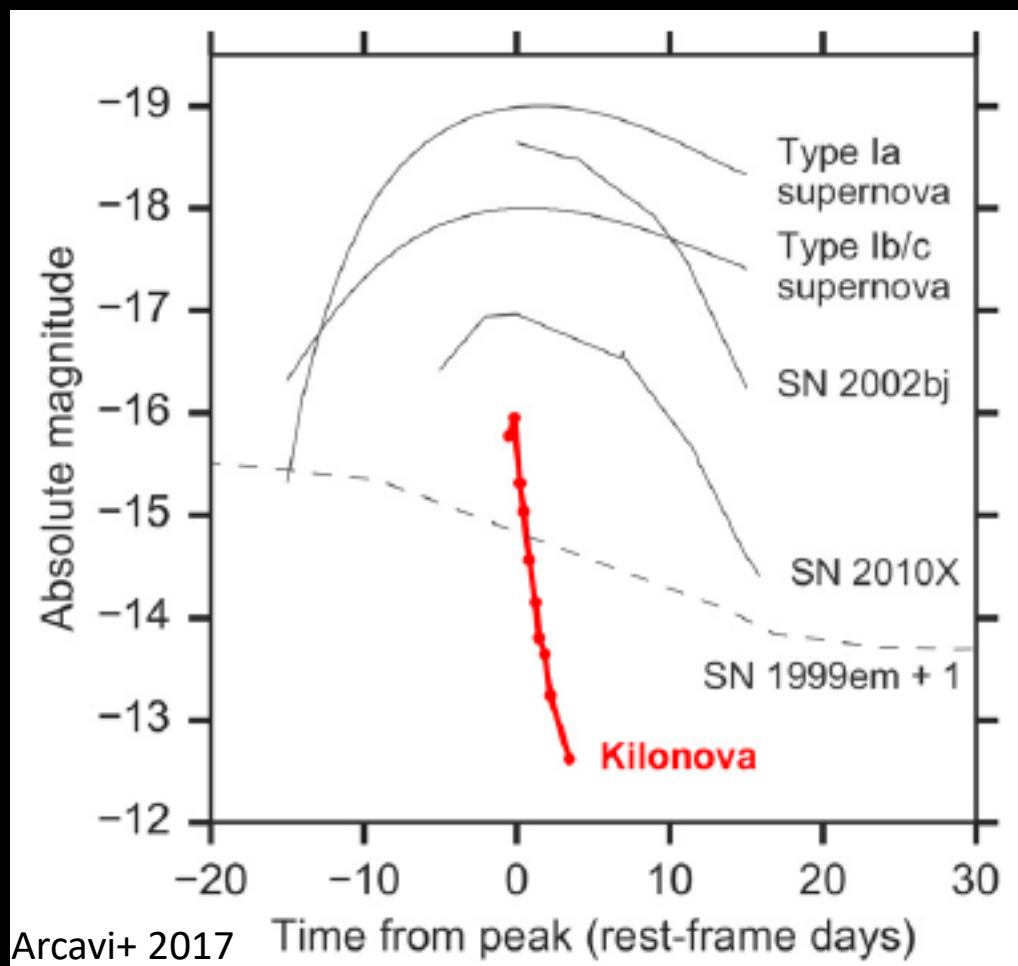
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The corresponding Lightcurve

$$M_{\text{SUN}} \sim +4.5$$
$$M_{\text{VENUS}} \sim -4.5$$



Optical filter (r)



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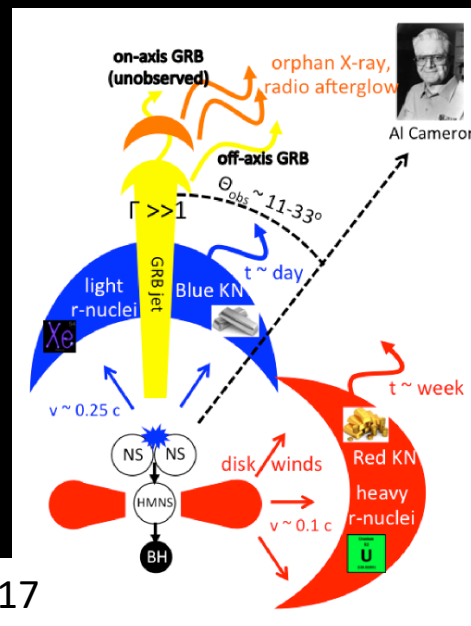


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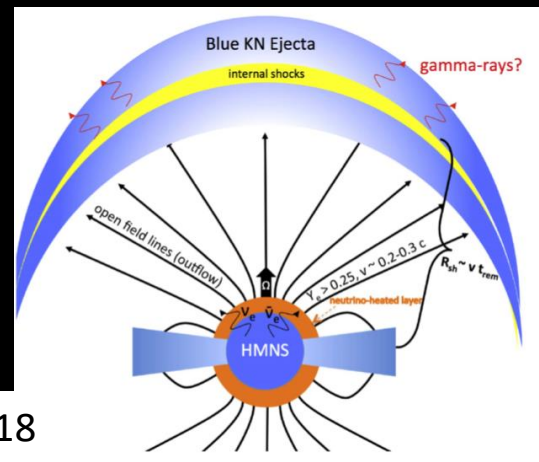
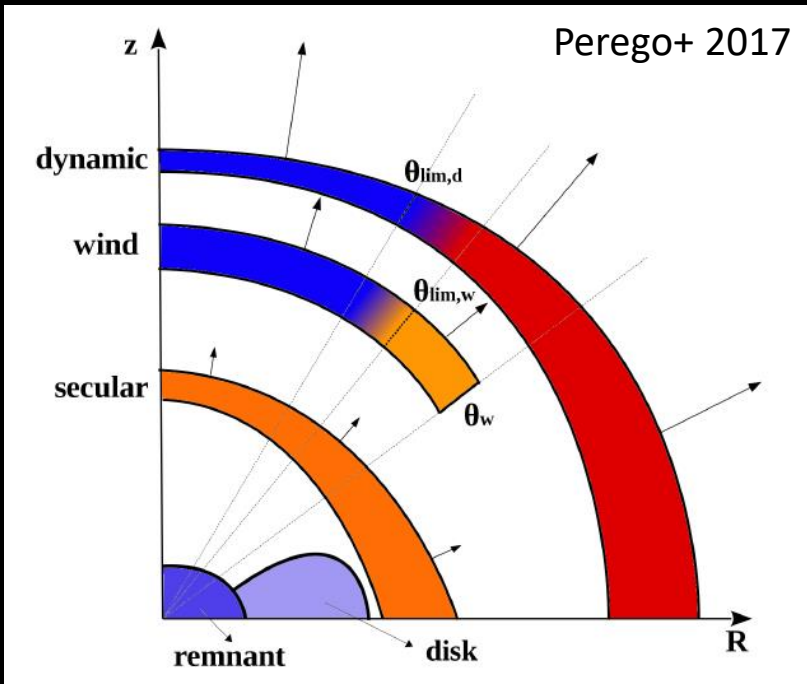


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The various components of a NSM



Metzger 2017



Metzger+ 2018



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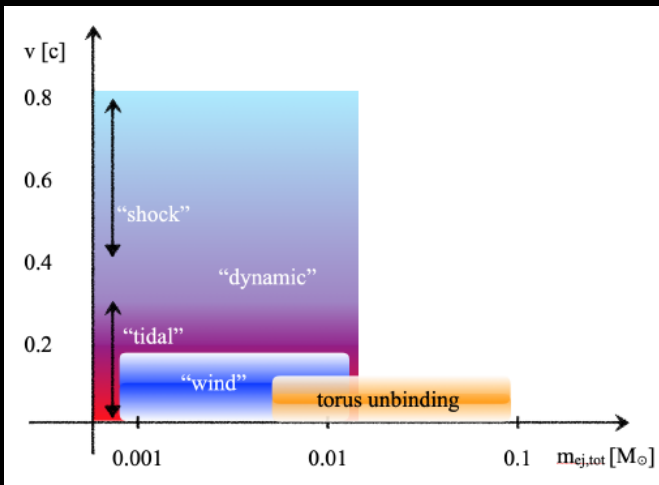


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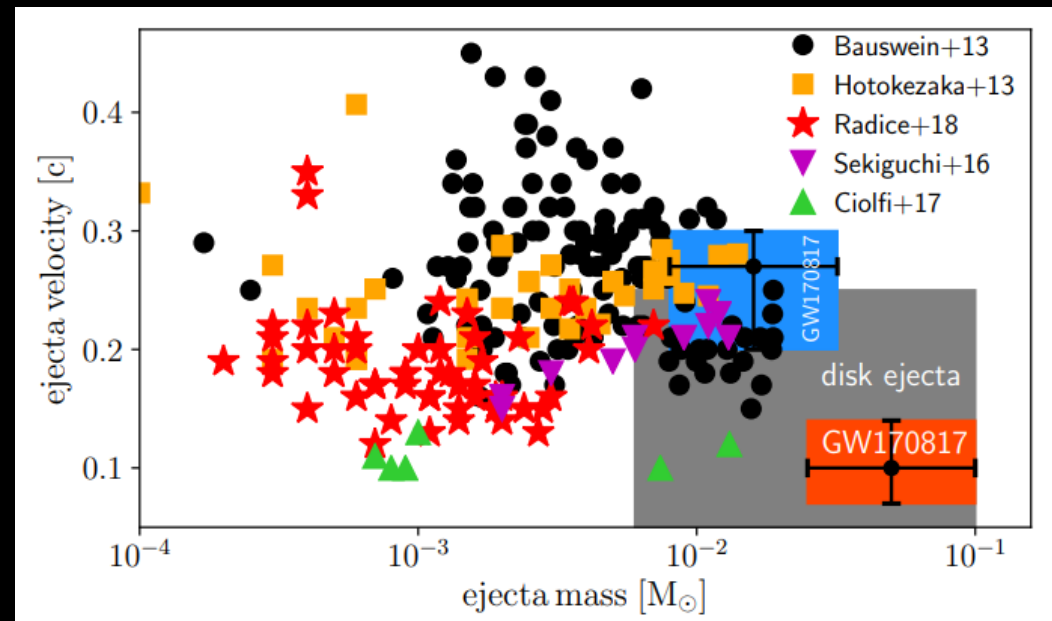
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Rosswog & Korobkin 2022



The NSM ejecta

Metzger 2019





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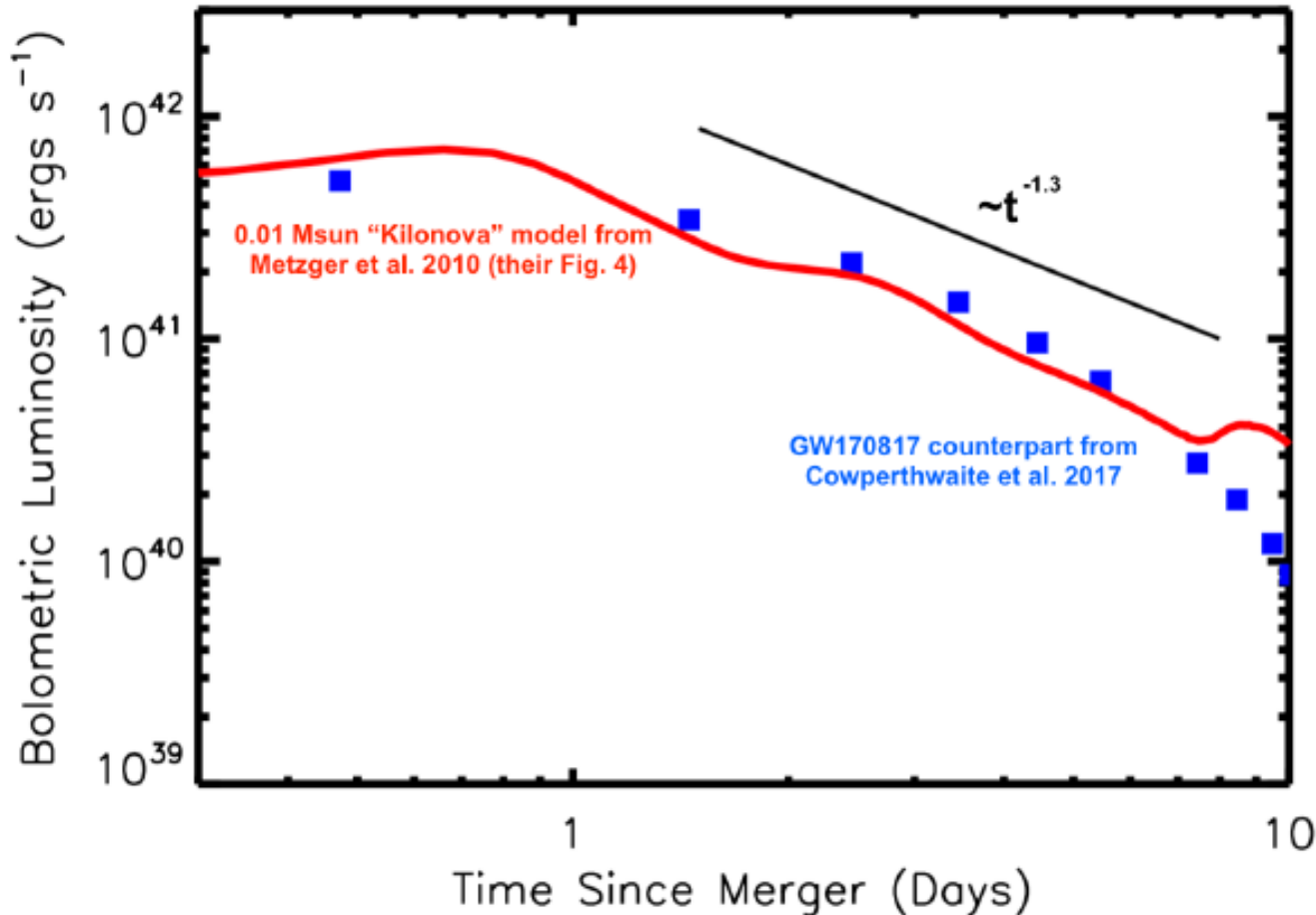


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Bolometric curve of GW170817



Chemical abundances
come from network
calculations.
But the question is:
how do they interact
with radiation?

It accounts for the whole energy leaving the «surface» at all wavelengths



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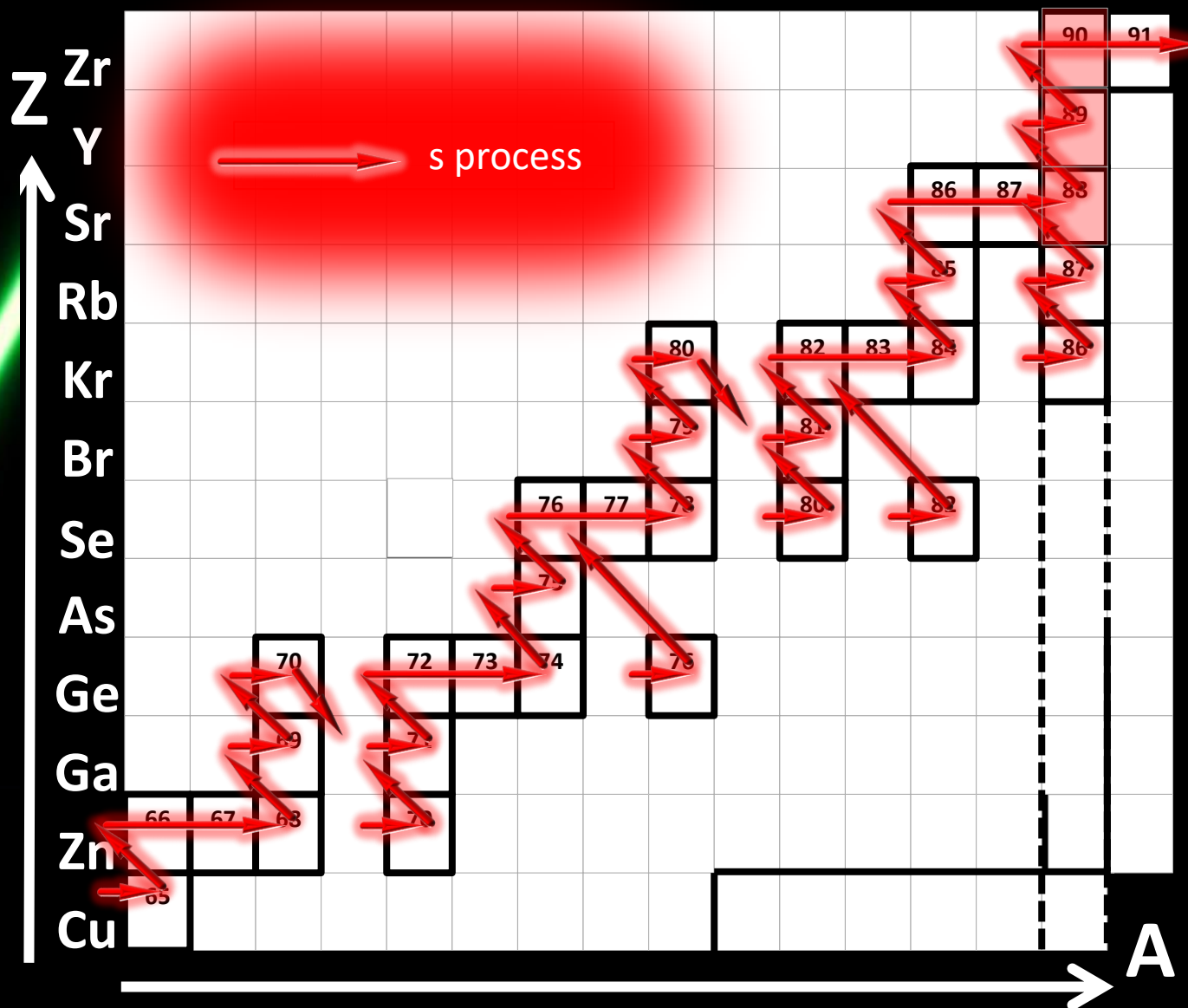


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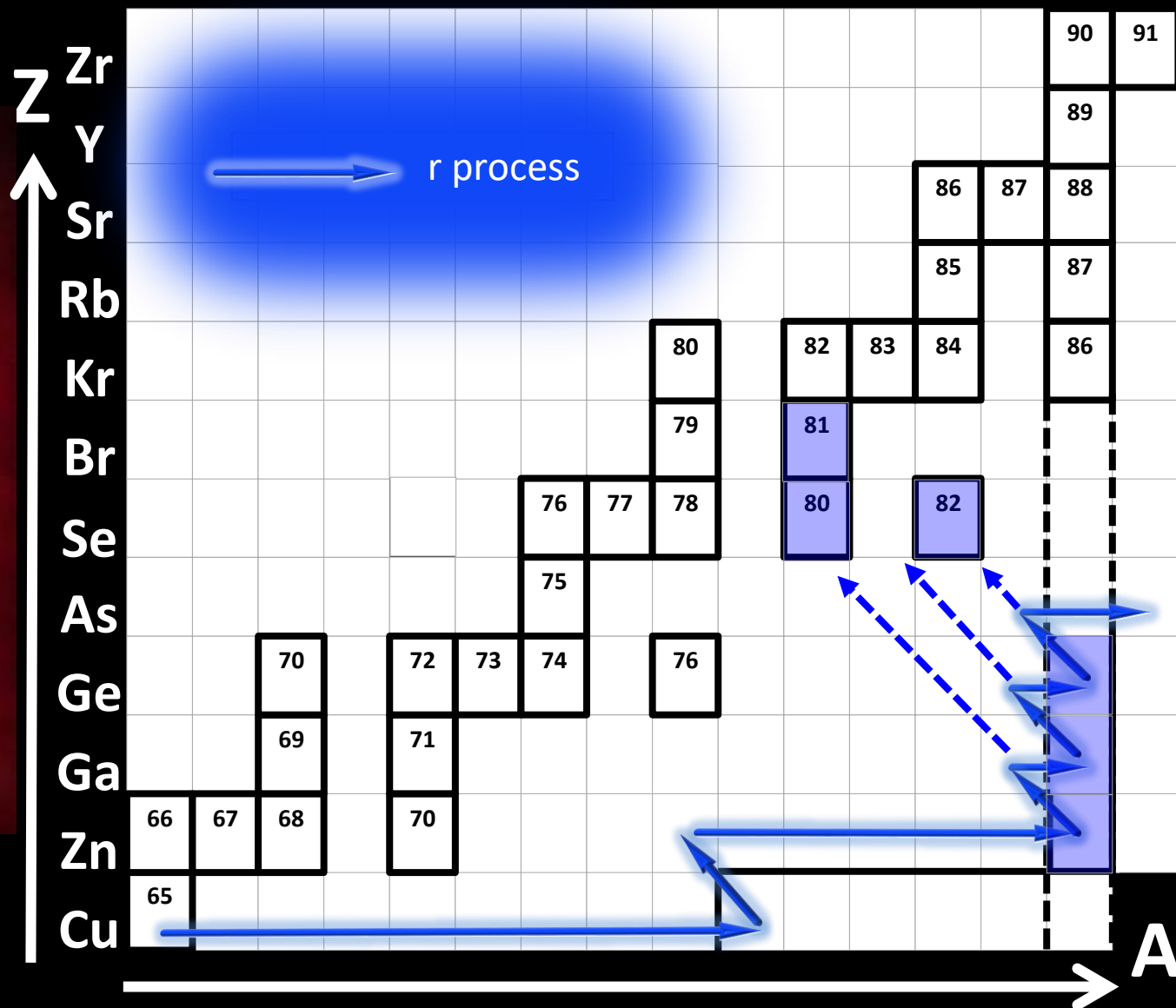


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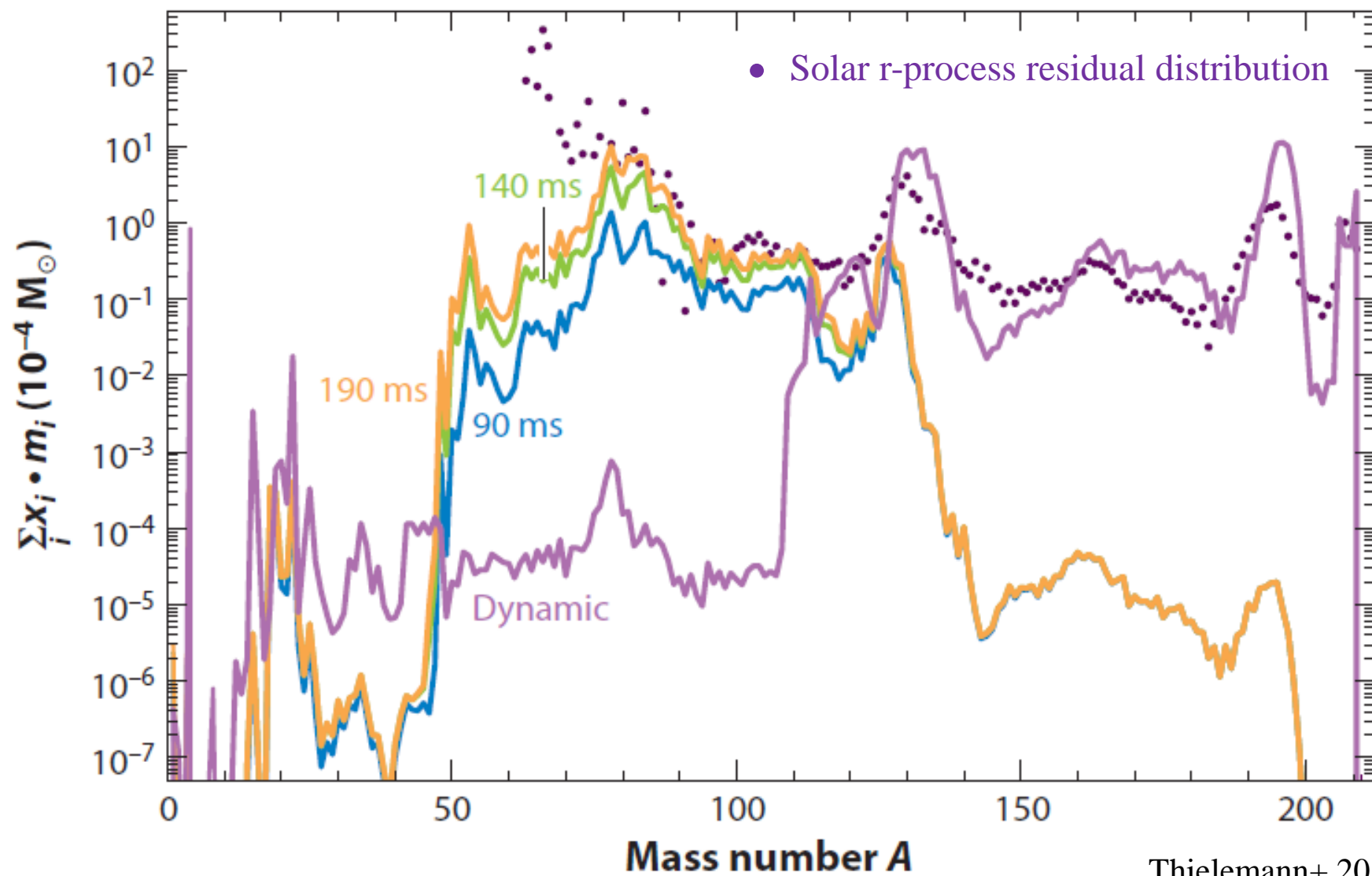


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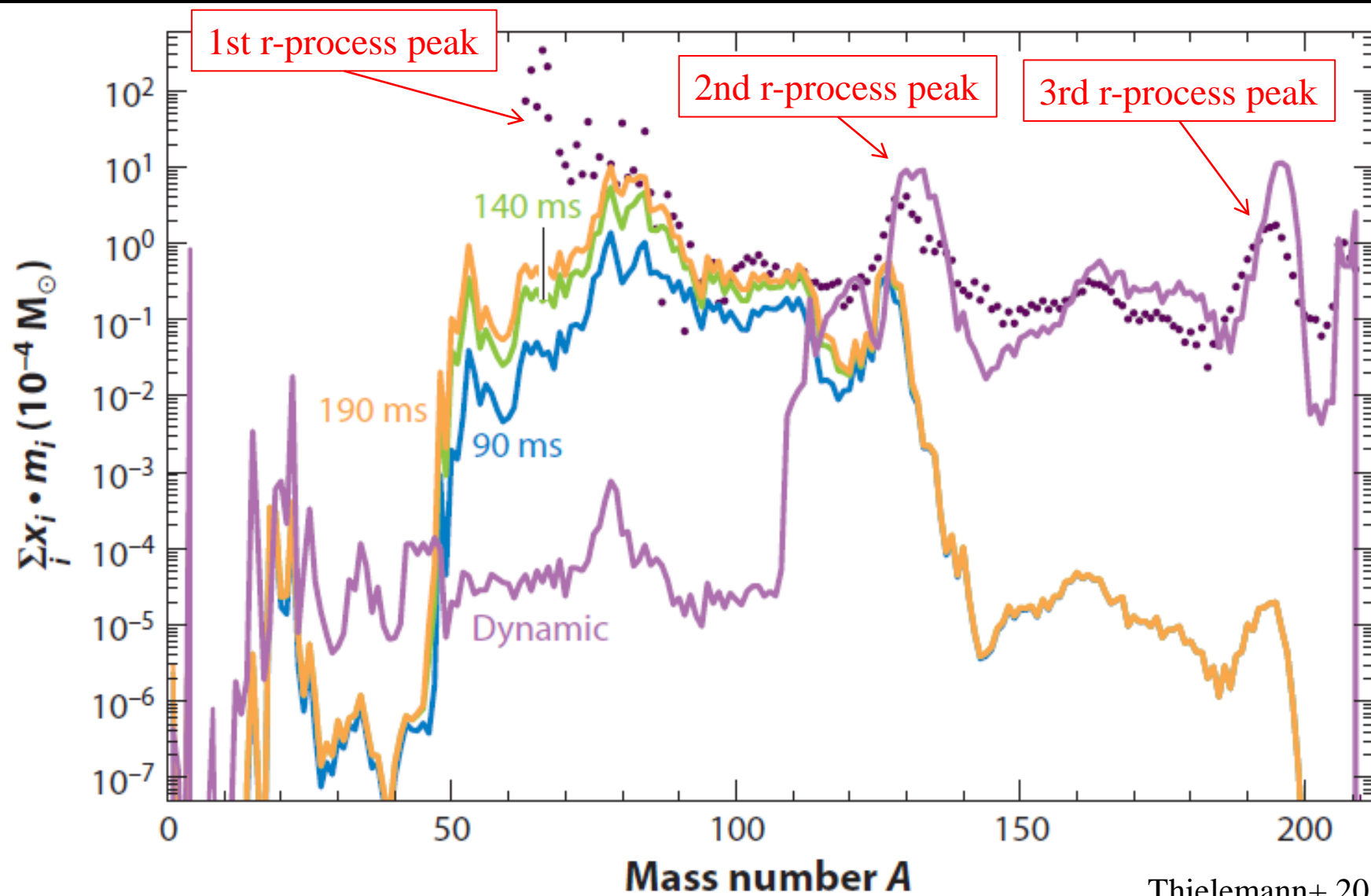
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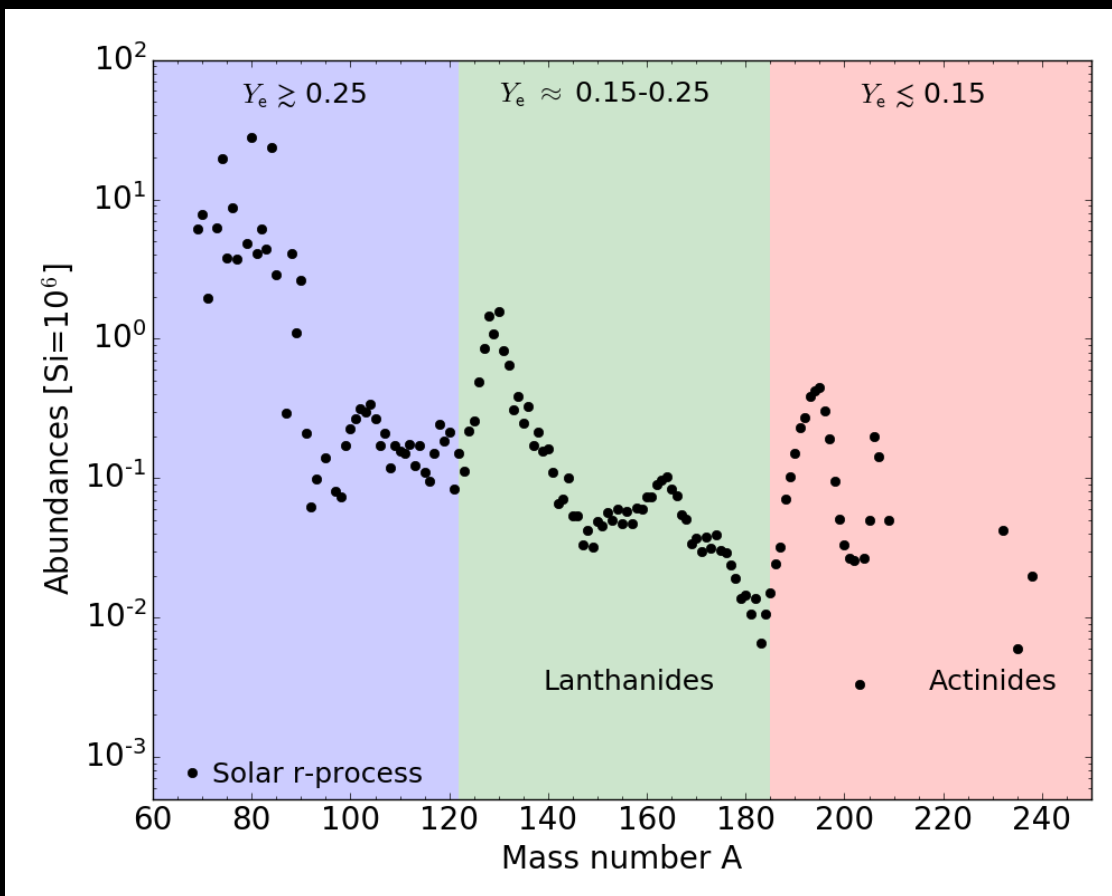
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In low entropy environment ($s \sim$ a few tens of k_b /baryon) Y_e is the dominant parameter

- $Y_e \gtrsim 0.25$: 1st peak up to 2nd r -process peak
- $0.15 \lesssim Y_e \lesssim 0.25$: 2nd and 3rd r -process peaks, but not the 1st
- $Y_e \lesssim 0.15$: robust r -process, due to several fission cycles



NSM nucleosynthesis

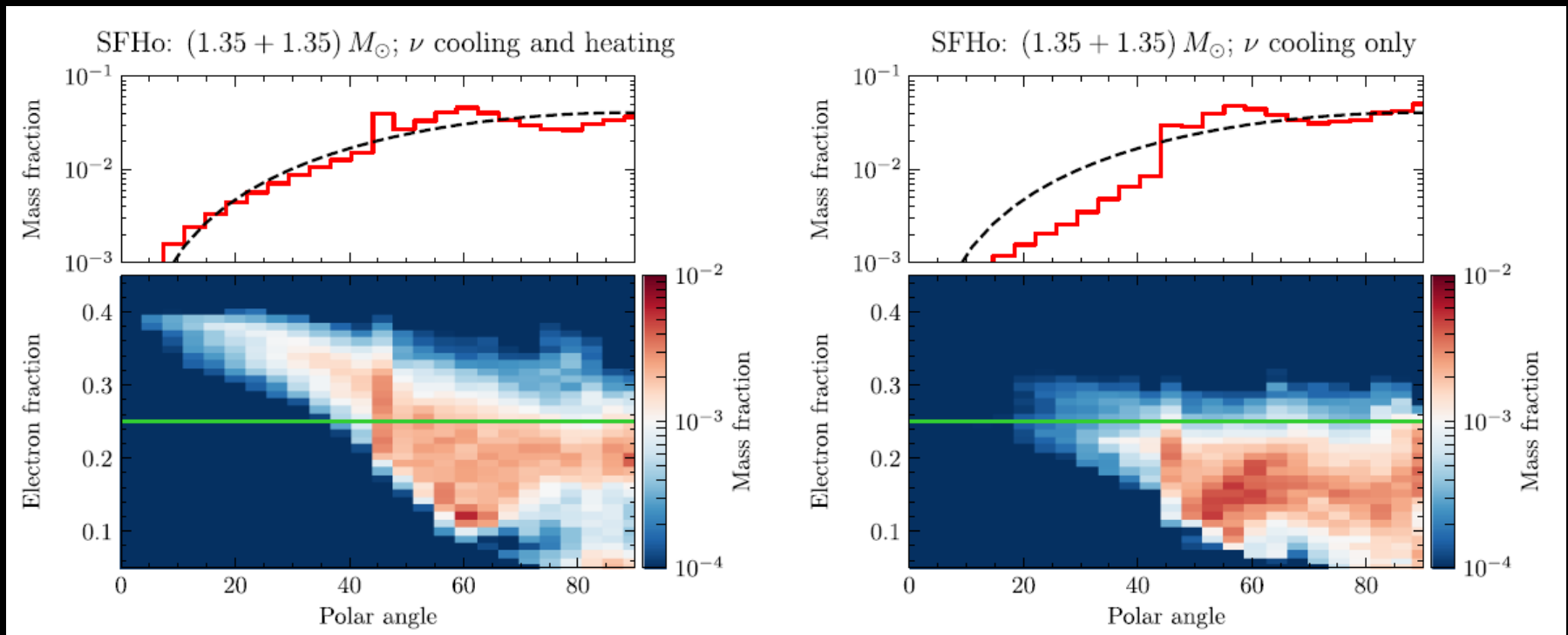
Electron fraction Y_e

$$Y_e \equiv \frac{n_{e^-} - n_{e^+}}{n_b} = (1 + n_n/n_p)^{-1}$$

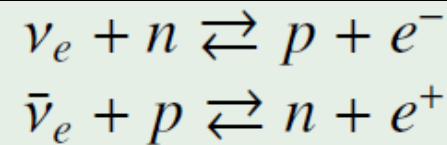


The role of neutrinos

The presence of neutrinos increases Y_e in the polar direction



Perego+ 2017





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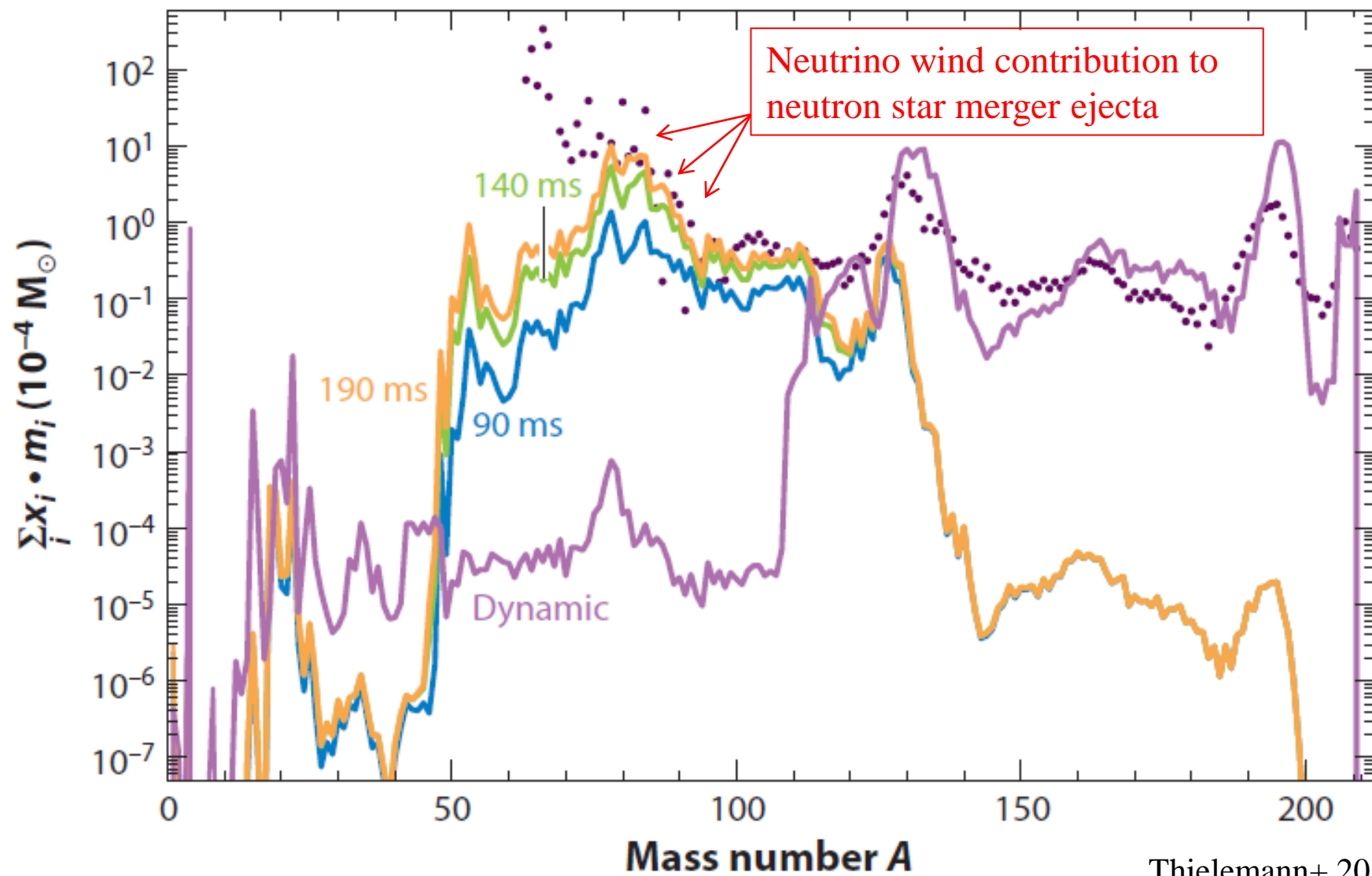
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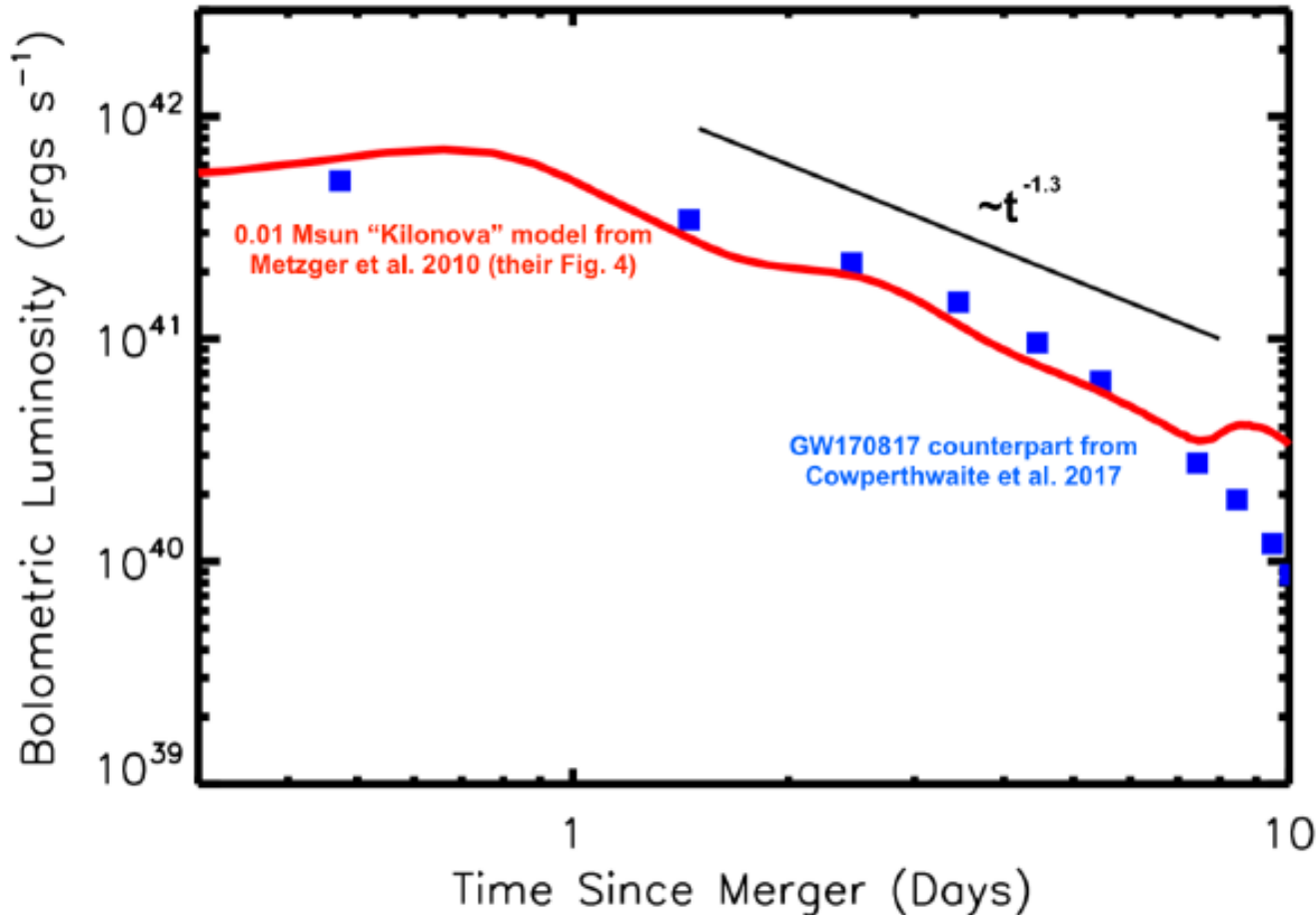
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Bolometric curve of GW170817



How do photons
interact with gas?

It accounts for the whole energy leaving the «surface» at all wavelengths



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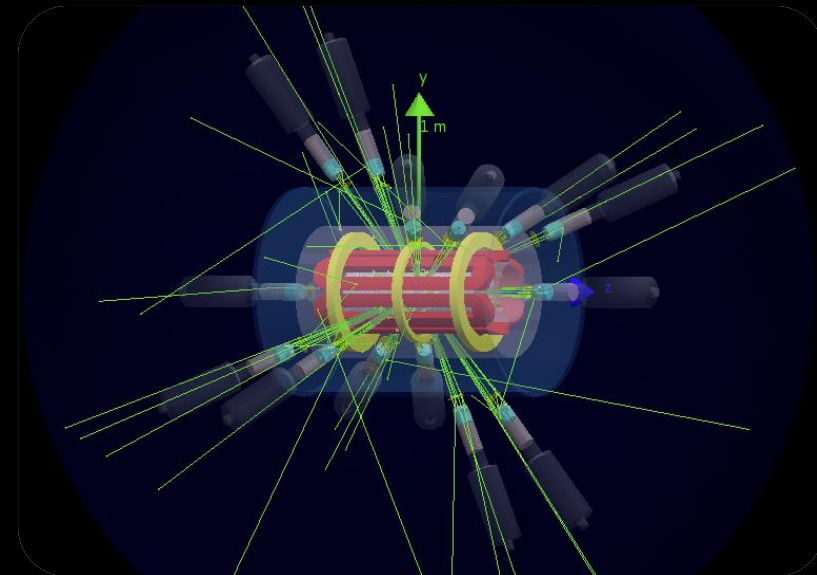
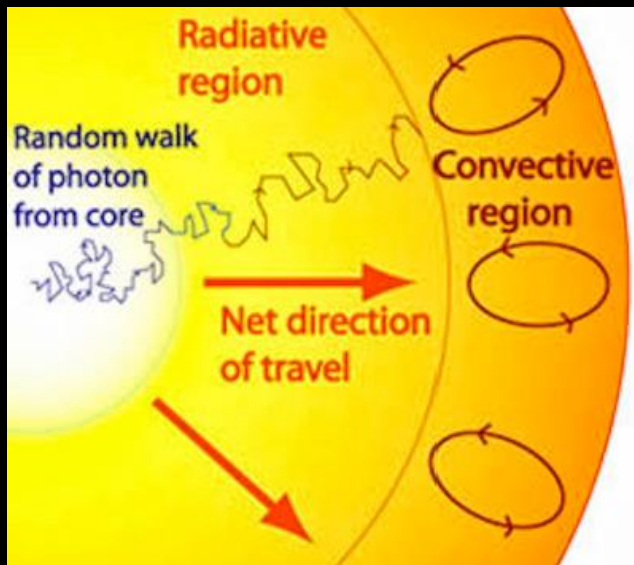
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Atomic opacities

Opacity (κ_ν), which is proportional to the plasma atomic level population and to radiative process cross sections, regulates the energy exchange between radiation and plasma, via multiple absorption-scattering processes through the radiative transport, and arises from the blending of millions of atomic line transitions.





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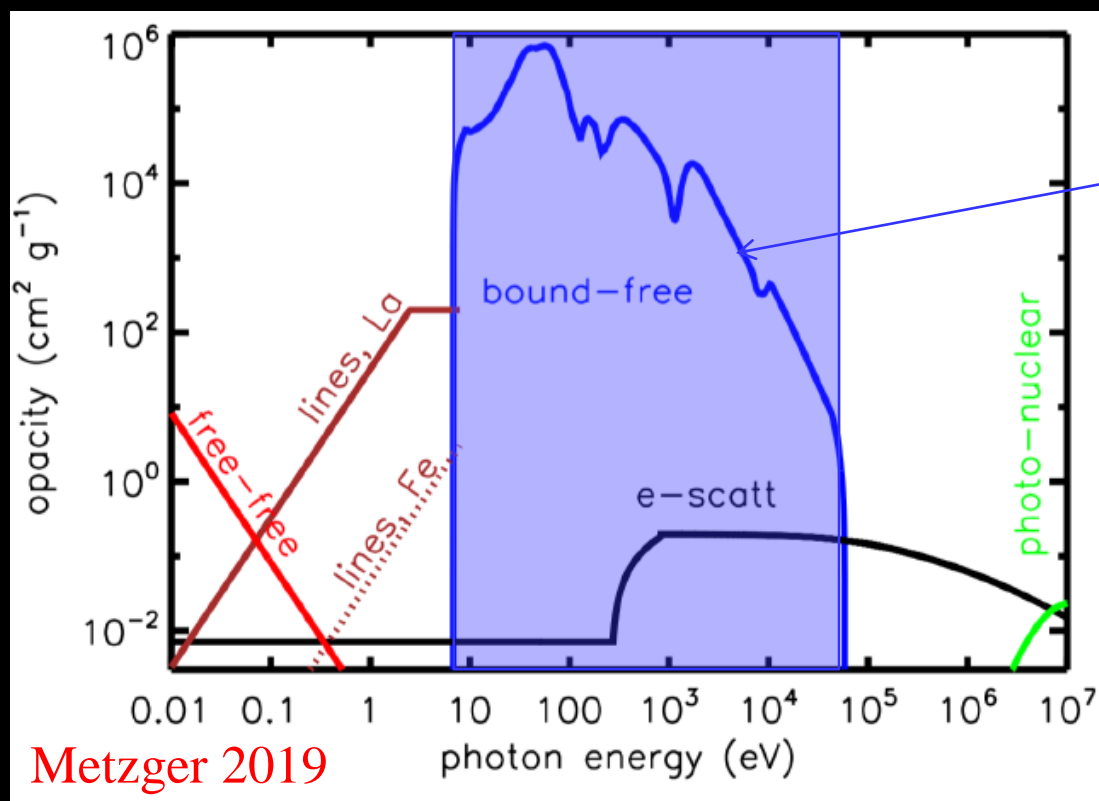
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Atomic opacities

Kilonova emission is centered in the optical/IR band, as this is the first spectral window through which the expanding merger ejecta becomes transparent.

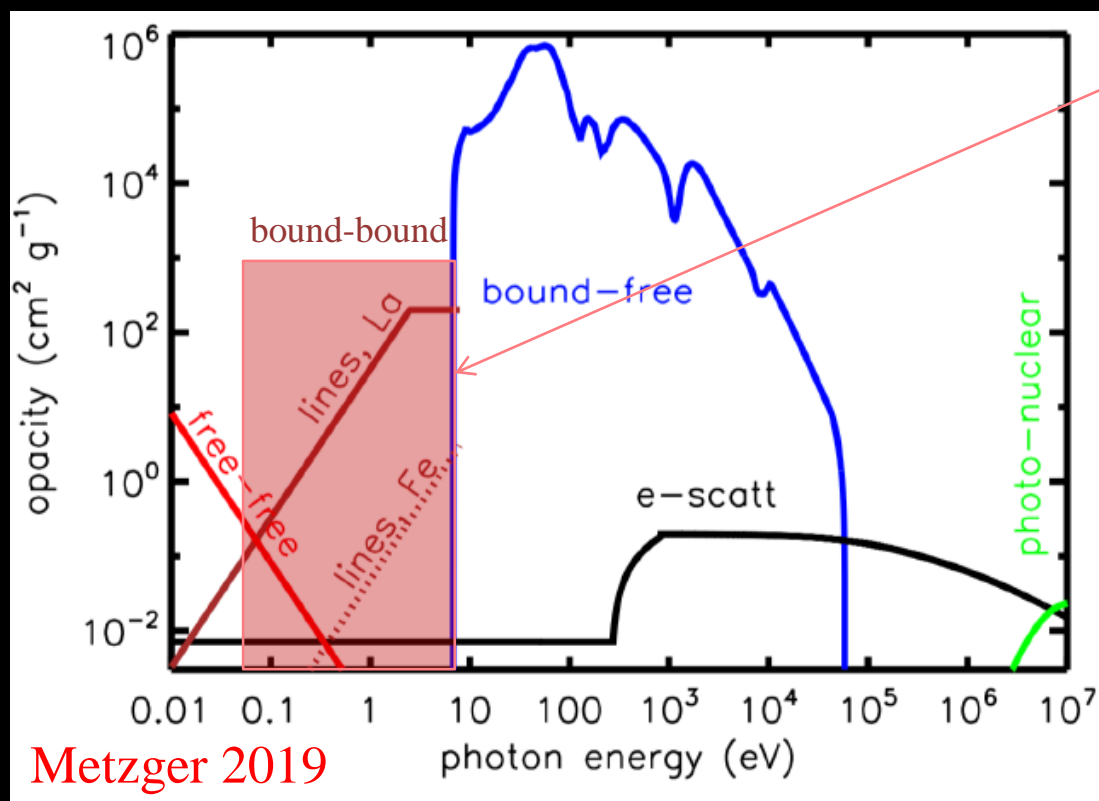


Throughout the far UV and X-ray bands, bound-free transitions of the ejecta dominate the opacity (blue line). This prevents radiation from escaping the ejecta at these frequencies.

VERY EARLY TIME KILONOVA

Atomic opacities

Kilonova emission is centered in the optical/IR band, as this is the first spectral window through which the expanding merger ejecta becomes transparent.



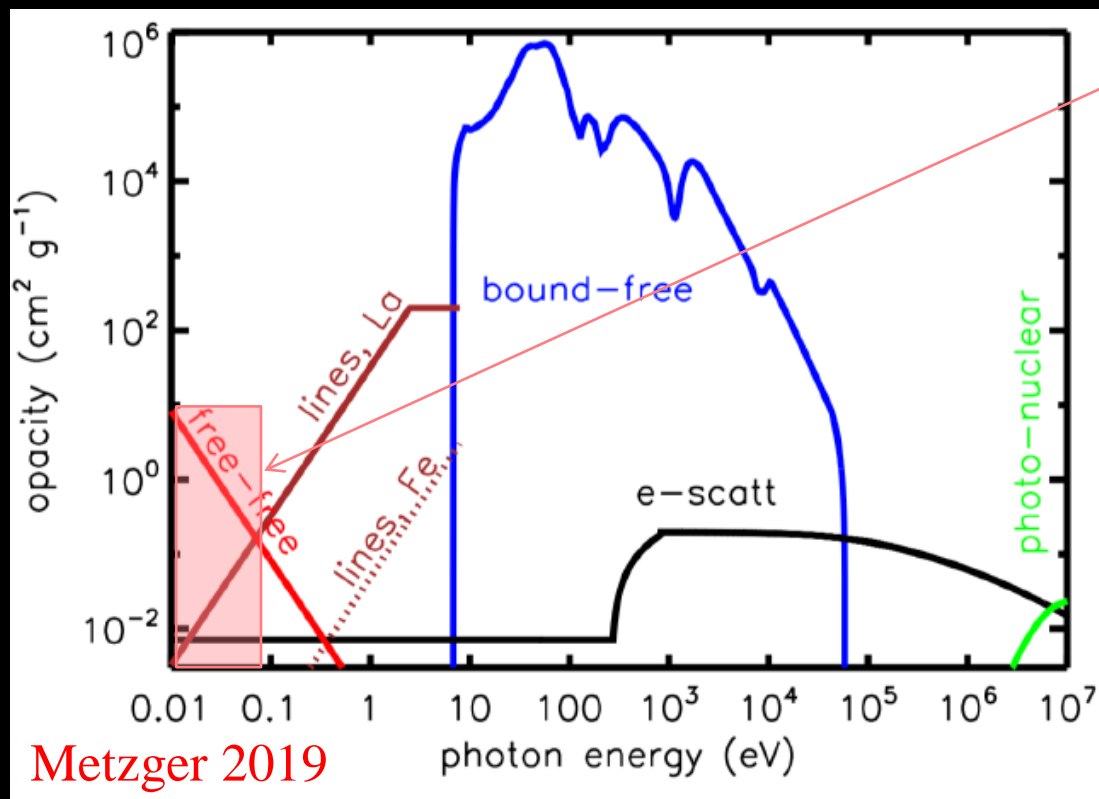
At optical/near-IR frequencies (brown line), the dominant source of opacity is a dense forest of line (bound-bound) transitions. The magnitude of this opacity is determined by the strengths and wavelength density of the lines, which in turn depend sensitively on the ejecta composition.

0.5d - 5d KILONOVA



Atomic opacities

Kilonova emission is centered in the optical/IR band, as this is the first spectral window through which the expanding merger ejecta becomes transparent.



At the lowest frequencies (radio and far-IR), free-free absorption from ionized gas dominates (red line). As the ejecta expands, the free-free opacity will decrease rapidly due to the decreasing density $\rho \propto t^{-3}$ and the fewer number of free electrons as the ejecta cools and recombines.

LATE TIME KILONOVA



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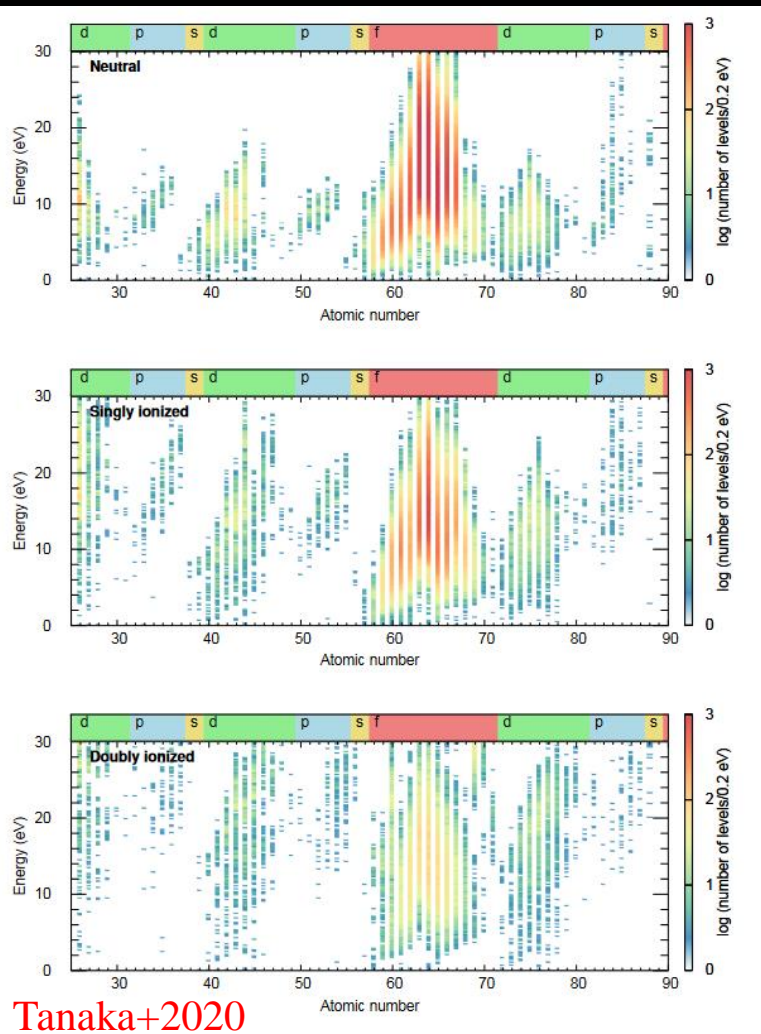


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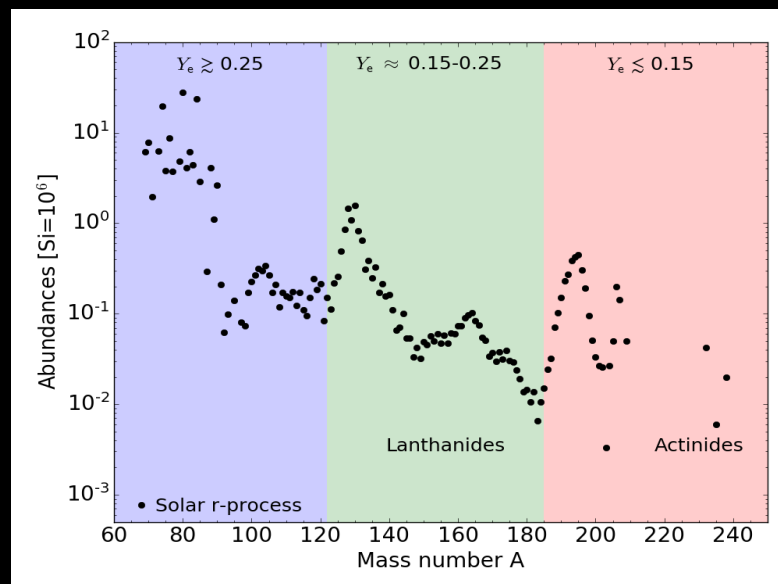
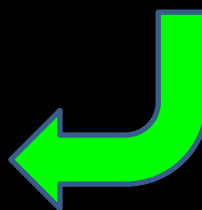


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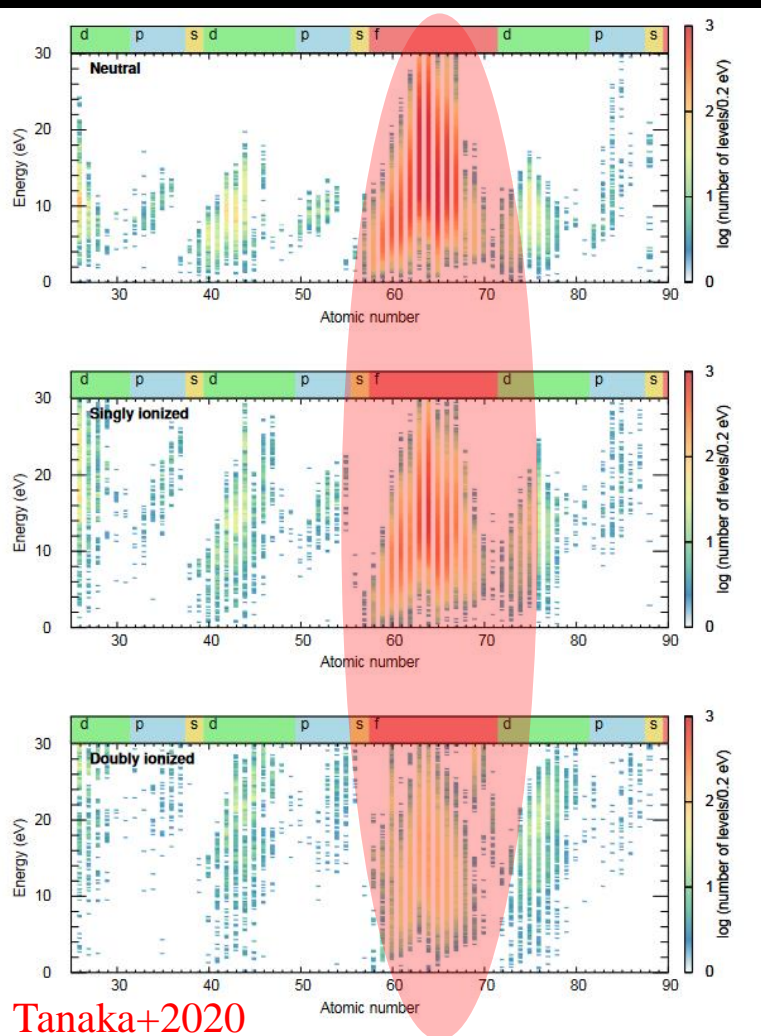
Atomic opacities



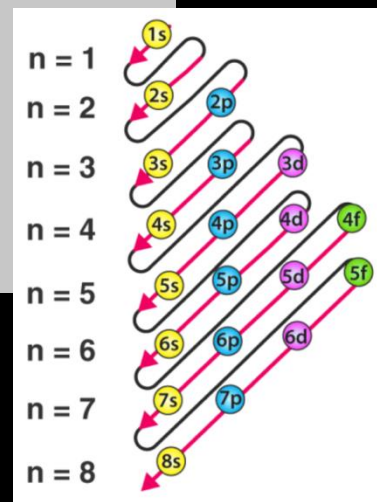
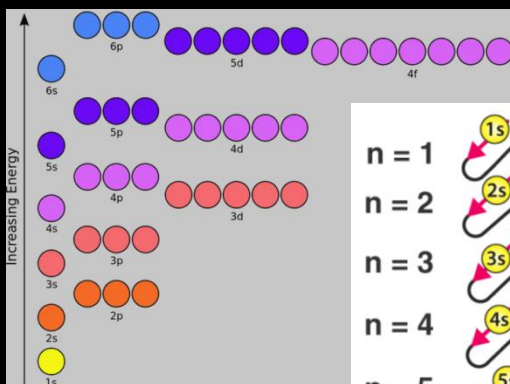
Chemical elements contribute to the global opacity with very different contributions, basing on their **electronic configuration** and their **abundance**.



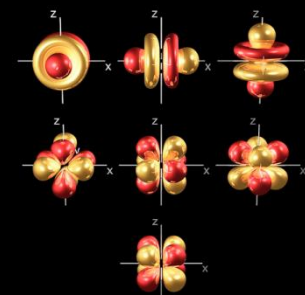
Atomic opacities



Chemical elements contribute to the global opacity with very different contributions, basing on their **electronic configuration** and their **abundance**. In particular, open f-shell elements (lanthanides) have larger opacities than the elements with other outermost electron shells.



f-shell orbitals





The Periodic Table of Elements

1	1 IA																										2	18 VIIIA
1	H																	2	He									
2	3	4	ATOMIC NUMBER - 1										5	6	7	8	9	10										
	Li	Be	SYMBOL - H										B	C	N	O	F	Ne										
			NAME - Hydrogen																									
3	11	12	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18										
	Na	Mg	III B	IV B	V B	VI B	VII B	VIII B	VIII B	VIII B	IB	II B	Al	Si	P	S	Cl	Ar										
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36										
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr										
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54										
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe										
6	55	56	57-71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86										
	Cs	Ba	La-Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn										
7	87	88	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118										
	Fr	Ra	Ac-Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og										
	LANTHANIDES		57	58	59	60	61	62	63	64	65	66	67	68	69	70	71											
			La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu											
	ACTINIDES		89	90	91	92	93	94	95	96	97	98	99	100	101	102	103											
			Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr											



The Periodic Table of Elements

1 IA																	18 VIIIA											
1 H	2 IIA												3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIIIB	9 VIIIB	10 VIIIB	11 IB	12 IIB	13 IIIB	14 IVB	15 VB	16 VIB	17 VIIB	18 VIIIA
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne											
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar											
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr											
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe											
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn											
87 Fr	88 Ra	89-103 Actinides																										

LETTER

Identification of strontium in the merger of two neutron stars

Darach Watson^{1,2}, Camilla J. Hansen^{3,*}, Jonatan Selsing^{1,2,*}, Andreas Koch⁴, Daniele B. Malesani^{1,2,5}, Anja C. Andersen¹, Johan P. U. Fynbo^{1,2}, Almudena Arcones^{6,7}, Andreas Bauswein^{7,8}, Stefano Covino⁹, Aniello Grado¹⁰, Kasper E. Heintz^{1,2,11}, Leslie Hunt¹², Chryssa Kouveliotou^{13,14}, Giorgos Leloudas^{1,5}, Andrew Levan^{15,16}, Paolo Mazzali^{17,18}, Elena Pian¹⁹ [See end for affiliations]



The Periodic Table of Elements

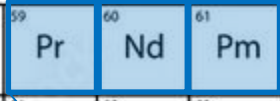
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1	H																	2	He		
2	3	4	ATOMIC NUMBER - 1														13	14	15	16	17
	Li	Be	SYMBOL - H																		
			NAME - Hydrogen																		
3	11	12	3														4	5	6	7	
	Na	Mg	IB														IVB	V	VI	VII	
4	19	20	21	22	23																
	K	Ca	Sc	Ti	V																
5	37	38	39	40	41																
	Rb	Sr	Y	Zr	Nb																
6	55	56	57-71	72	73																
	Cs	Ba	La-Lu	Hf	Ta																
7	87	88	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118			
	Fr	Ra	Ac-Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og			
LANTHANIDES		57	58	59	60	61	62	63	64	65	66	67	68	69	70	71					
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu					
ACTINIDES		89	90	91	92	93	94	95	96	97	98	99	100	101	102	103					
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Mn	Sg	Bo					

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MNRAS 518, 332–352 (2023)
Advance Access publication 2022 November 4
<https://doi.org/10.1093/mnras/stac3129>

Opacity calculations in four to nine times ionized Pr, Nd, and Pm atoms for the spectral analysis of kilonovae

NH. Carvajal Gallego,¹ J. Deprince,^{1,2} J. C. Berengut,³ P. Palmeri¹ and P. Quinet^{1,4*}

- ¹Physique Atomique et Astrophysique, Université de Mons, B-7000 Mons, Belgium
- ²Institut d'Astronomie et d'Astrophysique, Université Libre de Bruxelles, B-1050 Brussels, Belgium
- ³School of Physics, University of New South Wales, Sydney NSW 2052, Australia
- ⁴IPNAS, Université de Liège, Sart Tilman, B-4000 Liège, Belgium



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MNRAS 517, 281–293 (2022)
Advance Access publication 2022 September 1
<https://doi.org/10.1093/mnras/stac2401>

Theoretical investigation of energy levels and transitions for Pr IV

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- ²Astronomical Institute, Tohoku University, Sendai 980-8578, Japan
- ³Division for the Establishment of Frontier Sciences, Organization for Advanced Studies, Tohoku University, Sendai 980-8577, Japan
- ⁴National Institute for Fusion Science, 322-6 Oroshi-cho, Toki 509-5292, Japan
- ⁵Interdisciplinary Graduate School of Engineering Sciences, Kyushu University, Kasuga, Fukuoka 816-8580, Japan



The Periodic Table of Elements

1 1A	2 2A	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										18 VIII					
1 H		3 Li	4 Be														2 He
11 Na	12 Mg			3 III	4 IV	5 VB											
19 K	20 Ca	21 Sc	22 Ti	23 V													
37 Rb	36 Sr	39 Y	40 Zr	41 Nb													
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
LANTHANIDES	57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu		
ACTINIDES	89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr		



atoms



Article

Relativistic Atomic Structure of Au IV and the Os Isoelectronic Sequence: Opacity Data for Kilonova Ejecta

Zahra Sadat Taghadomi¹, Yier Wan¹, Alicia Flowers¹, Phillip Stancil¹, Brendan McLaughlin¹, Steven Bromley², Joan Marler³, Chad Sosolik³ and Stuart Loch^{2,*}



The Periodic Table of Elements

1 IA																	18 VIIIA	
1 H																	2 He	
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og	
																69 Tm	70 Yb	71 Lu
																101 Md	102 No	103 Lr

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Tungsten versus Selenium as a potential source of kilonova nebular emission observed by Spitzer

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⁷Institute of Theoretical Physics and Astronomy, Vilnius University, Saulėtekio Ave. 3, Vilnius, Lithuania



The Periodic Table of Elements

1	2											18						
1 IA																	2 VIIIA	
1 H																		2 He
3 Li	4 Be																	
11 Na	12 Mg																	
19 K	20 Ca	21 Sc	22 Ti															
37 Rb	38 Sr	39 Y	40 Zr															
55 Cs	56 Ba	57-71 La-Lu	72 Hf															
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og	
LANTHANIDES		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu		
ACTINIDES		89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr		

Monthly Notices
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ROYAL ASTRONOMICAL SOCIETY

MNRAS 513, 2302–2325 (2022)
Advance Access publication 2022 April 27

<https://doi.org/10.1093/mnras/stac1063>

Atomic data and opacity calculations in La V–X ions for the investigation of kilonova emission spectra

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²School of Physics, University of New South Wales, Sydney, NSW 2052, Australia
³IPNAS, Université de Liège, Sart Tilman, B-4000 Liège, Belgium



The Periodic Table of Elements

1 IA	2 IIA																	18 VIIIA	
1 H																		2 He	
3 Li	4 Be																	10 Ne	
11 Na	12 Mg																	18 Ar	
19 K	20 Ca	21																	36 Kr
37 Rb	38 Sr	39																	54 Xe
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn		
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og		
LANTHANIDES		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
ACTINIDES		89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			









atoms



Article

Structure Calculations in Nd III and U III Relevant for Kilonovae Modelling

Ricardo F. Silva ^{1,2,*} , Jorge M. Sampaio ^{1,2} , Pedro Amaro ³ , Andreas Flörs ⁴ , Gabriel Martínez-Pinedo ^{4,5,6} 
and José P. Marques ^{1,2} 



The Periodic Table of Elements

1 IA	1 H																	2 VIIA	2 He		
2 3	Li																	16 VA	9 F	17 VIA	10 Ne
3 11	Na																	17 S	18 Cl	18 Ar	
4 19	K																	35 Se	36 Br	36 Kr	
5 37	Rb																	53 Te	54 I	54 Xe	
6 55	Cs	Ba	La-Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn			
7 87	Fr	Ra	Ac-Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og			
LANTHANIDES		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu					
ACTINIDES		89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr					

Monthly Notices

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ROYAL ASTRONOMICAL SOCIETY

MNRAS **509**, 6138–6154 (2022)

Advance Access publication 2021 November 26

<https://doi.org/10.1093/mnras/stab3423>

Large-scale atomic data calculations in Ce V – X ions for application to early kilonova emission from neutron star mergers

H. Carvajal Gallego,¹ J. C. Berengut,² P. Palmeri¹ and P. Quinet^{1,3*}

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²School of Physics, University of New South Wales, Sydney NSW 2052, Australia

³IPNAS, Université de Liège, Sart Tilman, B-4000 Liège, Belgium



The Periodic Table of Elements

1	1 IA																2	18 VIIIA															
1	H																	2	He														
3	Li	4	Be	ATOMIC NUMBER - 1																9	F	10	Ne										
11	Na	12	Mg																	17	Cl	18	Ar										
19	K	20	Ca	21																	35	Br	36	Kr									
37	Rb	38	Sr	39																	53	I	54	Xe									
55	Cs	56	Ba	57-71	La-Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn													
87	Fr	88	Ra	89-103	Ac-Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og													
LANTHANIDES				57	La	58	Ce	59	Pr	60	Nd	61	Pm	62	Sm	63	Eu	64	Gd	65	Tb	66	Dy	67	Ho	68	Er	69	Tm	70	Yb	71	Lu
ACTINIDES				89	Ac	90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Md	102	No	103	Lr

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<https://doi.org/10.3847/1538-4365/ab8312>



Extended Calculations of Energy Levels and Transition Rates for Singly Ionized Lanthanide Elements. I. Pr–Gd

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 Received 2019 December 17; revised 2020 March 20; accepted 2020 March 20; published 2020 May 7



The Periodic Table of Elements

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<https://doi.org/10.3847/1538-4365/ac1ad2>



Extended Calculations of Energy Levels and Transition Rates for Singly Ionized Lanthanide Elements. II. Tb–Yb

Laima Radžiūtė¹, Gediminas Gaigalas¹, Daiji Kato^{2,3}, Pavel Rynkun¹, and Masaomi Tanaka⁴

¹Institute of Theoretical Physics and Astronomy, Vilnius University, Saulėtekio Ave. 3, Lithuania; Laima.Radziute@tfai.vu.lt

²National Institute for Fusion Science, 322-6 Oroshi-cho, Toki 509-5292, Japan

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⁴Astronomical Institute, Tohoku University, Aoba, Sendai 980-8578, Japan

Received 2021 June 11; revised 2021 July 15; accepted 2021 August 2; published 2021 November 12

1	2																	18
1	2																	2
3	4																	10
11	12																	18
19	20																	36
37	38																	54
55	56	57-71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	
87	88	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	

LANTHANIDES	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71
	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
ACTINIDES	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103
	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr



The Periodic Table of Elements

1 IA																	18 VIIIA	
1 H																	2 He	
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar											
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og	
																69 Tm	70 Yb	71 Lu
																101 Md	102 No	103 Lr

Monthly Notices

of the

ROYAL ASTRONOMICAL SOCIETY

MNRAS **506**, 3560–3577 (2021)

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<https://doi.org/10.1093/mnras/stab1861>

Constraints on the presence of platinum and gold in the spectra of the kilonova AT2017gfo

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The Periodic Table of Elements

1 IA																	18 VIIIA
1 H																	2 He
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar										
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og

LANTHANIDES

57
La

ACTINIDES

89
Ac

Monthly Notices

of the
ROYAL ASTRONOMICAL SOCIETY

MNRAS **526**, L155–L159 (2023)

Advance Access publication 2023 September 28

<https://doi.org/10.1093/mnras/stad128>

Tellurium emission line in kilonova AT 2017gfo

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²Astronomical Institute, Tohoku University, Aoba, Sendai 980-8578, Japan

³National Institute for Fusion Science, 322-6 Oroshi-cho, Toki 509-5292, Japan

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⁵Institute of Theoretical Physics and Astronomy, Vilnius University, Saulėtekio Ave. 3, Vilnius 10222, Lithuania



The Periodic Table of Elements

1 IA																	18 VIIIA
1 H																	2 He
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar										
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
LANTHANIDES		57 La	58 Ce														
ACTINIDES		89 Ac	90 Th														

A&A, 678, A67 (2023)
<https://doi.org/10.1051/0004-6361/202346198>
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**Astronomy
&
Astrophysics**

Atomic data and expansion opacity calculations in two representative 4d transition elements, niobium and silver, of interest for kilonovae studies[★]

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e-mail: sirine.bennasr@umons.ac.be; sirinamarouan@gmail.com; pascal.quinet@umons.ac.be
² Institut d'Astronomie et d'Astrophysique, Université Libre de Bruxelles, CP 226, 1050 Brussels, Belgium
³ IPNAS, Université de Liège, Sart Tilman, 4000 Liège, Belgium



The Periodic Table of Elements

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













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<https://doi.org/10.3847/1538-4357/acdc95>



CrossMark

Cerium Features in Kilonova Near-infrared Spectra: Implication from a Chemically Peculiar Star

Masaomi Tanaka^{1,2} , Nanae Domoto¹ , Wako Aoki^{3,4} , Miho N. Ishigaki³ , Shinya Wanajo⁵ , Kenta Hotokezaka^{6,7} ,
Kyohei Kawaguchi^{5,8,9} , Daiji Kato^{10,11} , Jae-Joon Lee¹² , Ho-Gyu Lee^{12,13} , Teruyuki Hirano^{3,4,14} ,
Takayuki Kotani^{3,4,14} , Masayuki Kuzuhara^{3,14} , Jun Nishikawa^{3,4,14} , Masashi Omiya^{3,14} , Motohide Tamura^{3,14,15} , and
Akitoshi Ueda^{3,4,14} 

1	1A	1	H	2	2A	2	He	18	VIIIA
3	3A	3	Li	4	4A	4	Be	10	VIIIA
11	1A	11	Na	12	2A	12	Mg	18	VIIIA
19	1A	19	K	20	2A	20	Ca	36	VIIIA
37	1A	37	Rb	38	2A	38	Sr	54	VIIIA
55	1A	55	Cs	56	2A	56	Ba	72	VIIIA
87	1A	87	Fr	88	2A	88	Ra	104	VIIIA
73	3B	73	Ta	74	4B	74	W	80	4B
75	4B	75	Re	76	4B	76	Os	82	4B
77	4B	77	Ir	78	4B	78	Pt	84	4B
79	4B	79	Au	80	4B	80	Hg	86	4B
105	6B	105	Db	106	6B	106	Sg	112	6B
107	6B	107	Bh	108	6B	108	Hs	114	6B
109	6B	109	Mt	110	6B	110	Ds	116	6B
111	6B	111	Rg	112	6B	112	Cn	118	6B
57	LANTHANIDES	57	La	58	58	58	Ce	59	59
59	LANTHANIDES	59	Pr	60	60	60	Nd	61	61
61	LANTHANIDES	61	Pm	62	62	62	Sm	63	63
63	LANTHANIDES	63	Eu	64	64	64	Gd	65	65
65	LANTHANIDES	65	Tb	66	66	66	Dy	67	67
67	LANTHANIDES	67	Ho	68	68	68	Er	69	69
69	LANTHANIDES	69	Tm	70	70	70	Yb	71	71
71	LANTHANIDES	71	Lu	89	89	89	Ac	90	90
89	ACTINIDES	89	Ac	90	90	90	Th	91	91
91	ACTINIDES	91	Pa	92	92	92	U	93	93
93	ACTINIDES	93	Np	94	94	94	Pu	95	95
95	ACTINIDES	95	Am	96	96	96	Cm	97	97
97	ACTINIDES	97	Bk	98	98	98	Cf	99	99
99	ACTINIDES	99	Es	100	100	100	Fm	101	101
101	ACTINIDES	101	Md	102	102	102	No	103	103
103	ACTINIDES	103	Lr						



The Periodic Table of Elements

1 IA																	18 VIIIA
1 H																	2 He
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
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55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn


Eur. Phys. J. D (2023) 77:126
<https://doi.org/10.1140/epjd/s10053-023-00695-5>

THE EUROPEAN
PHYSICAL JOURNAL D



Regular Article – Atomic Physics

Calculations of multipole transitions in Sn II for kilonova analysis

A. I. Bondarev^{1,2,a} , J. H. Gillanders³, C. Cheung⁴, M. S. Safronova^{4,5}, and S. Fritzsche^{1,2,6}



The Periodic Table of Elements

1 IA																	18 VIIIA
1 H																	2 He
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne
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37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
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87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og

A&A 675, A194 (2023)

<https://doi.org/10.1051/0004-6361/202346421>

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**Astronomy
&
Astrophysics**

Discovery of a 760 nm P Cygni line in AT2017gfo: Identification of yttrium in the kilonova photosphere

Albert Sneppen^{1,2} and Darach Watson^{1,2}



Finanziato dall'Unione europea
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PIANO NAZIONALE DI RIPRESA E RESILIENZA



INAF

ISTITUTO NAZIONALE DI ASTRONOMIA

Osservatorio Astronomico d'Abruzzo

The Periodic Table of Elements

1 IA	1 H	<p>Monthly Notices of the ROYAL ASTRONOMICAL SOCIETY</p> <p>MNRAS 493, 4143–4171 (2020) Advance Access publication 2020 February 26</p> <p>doi:10.1093/mnras/staa485</p> <p>A line-binned treatment of opacities for the spectra and light curves from neutron star mergers</p> <p>C. J. Fontes ¹, ¹★ C. L. Fryer, ^{1,2,3} A. L. Hungerford, ¹ R. T. Wollaeger ¹ and O. Korobkin ¹</p> <p>¹Los Alamos National Laboratory, Los Alamos, NM 87545, USA ²Physics Department, University of Arizona, Tucson, AZ 85721, USA ³Physics and Astronomy Department, University of New Mexico, Albuquerque, NM 87131, USA</p>																2 VIIA	2 He
2	3 Li																	10 Ne	
3	11 Na																	18 Ar	
4	19 K																	36 Kr	
5	37 Rb																	54 Xe	
6	55 Cs																	86 Rn	
7	87 Fr																	118 Og	
LANTHANIDES		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
ACTINIDES		89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			



The Periodic Table of Elements

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1 H																	2 He
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57-71 La-Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh											118 Og
LANTHANIDES		57 La	58 Ce	59 Pr	60 Nd	61 Pm											
ACTINIDES		89 Ac	90 Th	91 Pa	92 U	93 Np											

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Letter | [Published: 16 October 2017](#)

A kilonova as the electromagnetic counterpart to a gravitational-wave source

[S. J. Smartt](#) , [T.-W. Chen](#), [A. Jerkstrand](#), [M. Coughlin](#), [E. Kankare](#), [S. A. Sim](#), [M. Fraser](#), [C. Inerrera](#), [K. Maguire](#), [K. C. Chambers](#), [M. E. Huber](#), [T. Krühler](#), [G. Leloudas](#), [M. Magee](#), [L. J. Shingles](#), [K. W. Smith](#), [D. R. Young](#), [J. Tonyr](#), [R. Kotak](#), [A. Gal-Yam](#), [J. D. Lyman](#), [D. S. Homan](#), [C. Agliozzo](#), [J. P. Anderson](#), ... [O. Yaron](#)

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The Periodic Table of Elements

1 1A																	18 VIIIA	
1 H																	2 He	
3 Li	4 Be	ATOMIC NUMBER - 1 SYMBOL - H NAME - Hydrogen										5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
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87 Fr	88 Ra	89-103 Ac-Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og	
LANTHANIDES		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu		
ACTINIDES		89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr		

Japan-Lithuania Opacity Database for Kilonova (version 1.1)

Daiji Kato and Izumi Murakami (National Institute for Fusion Science, Japan)

Masaomi Tanaka and Smaranika Banerjee (Tohoku University, Japan)

Gediminas Gaigalas, Laima Kitovienė, and Pavel Rynkun (Vilnius University, Lithuania)



Last input: nuclear heating rates

$$\dot{Q}_{r-process} = \sum_{i \in \text{reactions}} Q_i \lambda_i$$

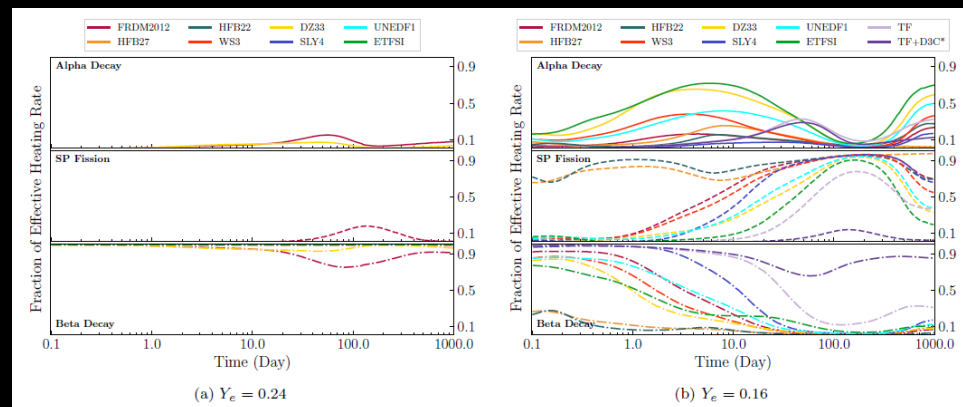
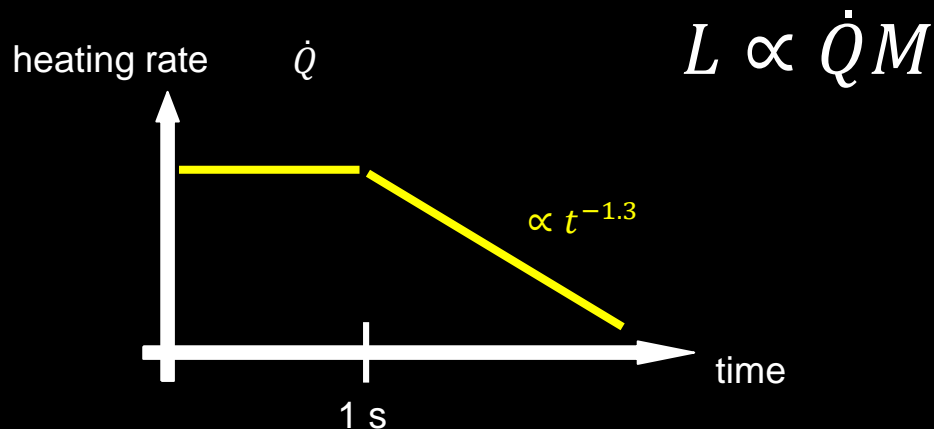
$$Q = M_{\text{initial}} - M_{\text{final}}$$

$$\lambda = \text{decay rate}$$

Heating efficiencies

$$\frac{d\varepsilon}{dt} = \dot{\varepsilon}_0 \left(\frac{1}{2} - \frac{1}{\pi} \arctan \left[\frac{t-t_0}{\sigma} \right] \right)^\alpha \left(\frac{1}{2} + \frac{1}{\pi} \arctan \left[\frac{t-t_1}{\sigma_1} \right] \right)^{\alpha_1} + C_1 e^{-t/\tau_1} + C_2 e^{-t/\tau_2} + C_3 e^{-t/\tau_3} \quad (2)$$

...13 free parameters...





Last input: nuclear heating rates

$$\dot{Q}_{r-process} = \sum_{i \in \text{reactions}} Q_i \lambda_i$$

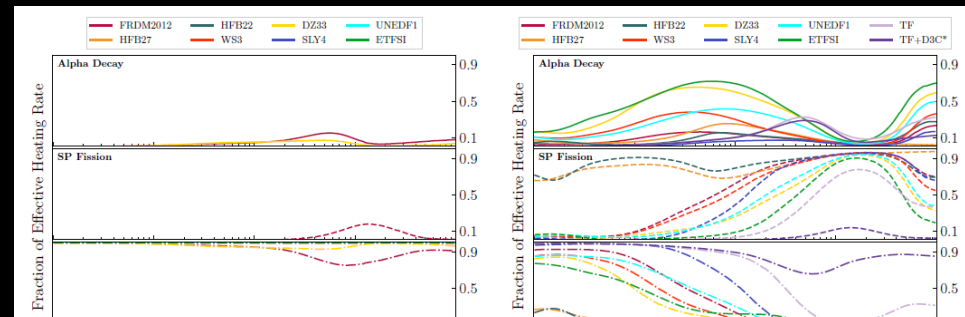
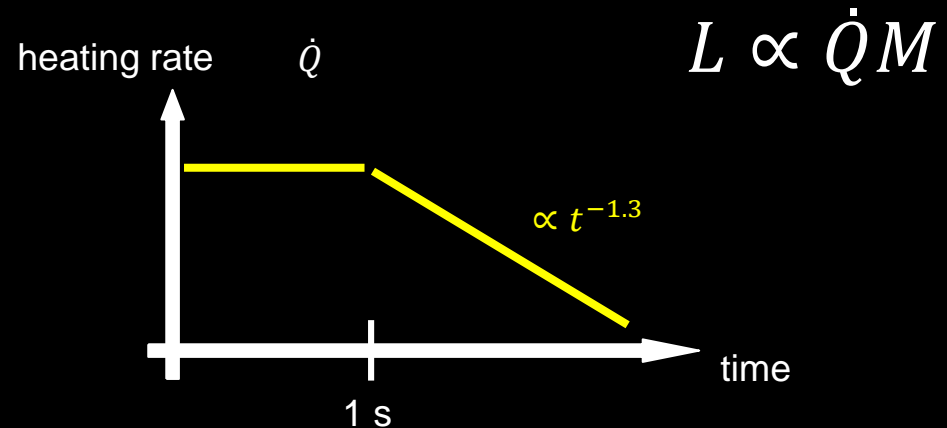
$$Q = M_{\text{initial}} - M_{\text{final}}$$

$\lambda = \text{decay rate}$

Heating efficiencies

$$\frac{d\varepsilon}{dt} = \dot{\varepsilon}_0 \left(\frac{1}{2} - \frac{1}{\pi} \arctan \left[\frac{t-t_0}{\sigma} \right] \right)^\alpha \left(\frac{1}{2} + \frac{1}{\pi} \arctan \left[\frac{t-t_1}{\sigma_1} \right] \right)^{\alpha_1} + C_1 e^{-t/\tau_1} + C_2 e^{-t/\tau_2} + C_3 e^{-t/\tau_3} \quad (2)$$

...13 free parameters...



With four parameters I can fit an elephant, and with five I can make him wiggle his trunk. [J. VON NEUMANN]



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...and finally...





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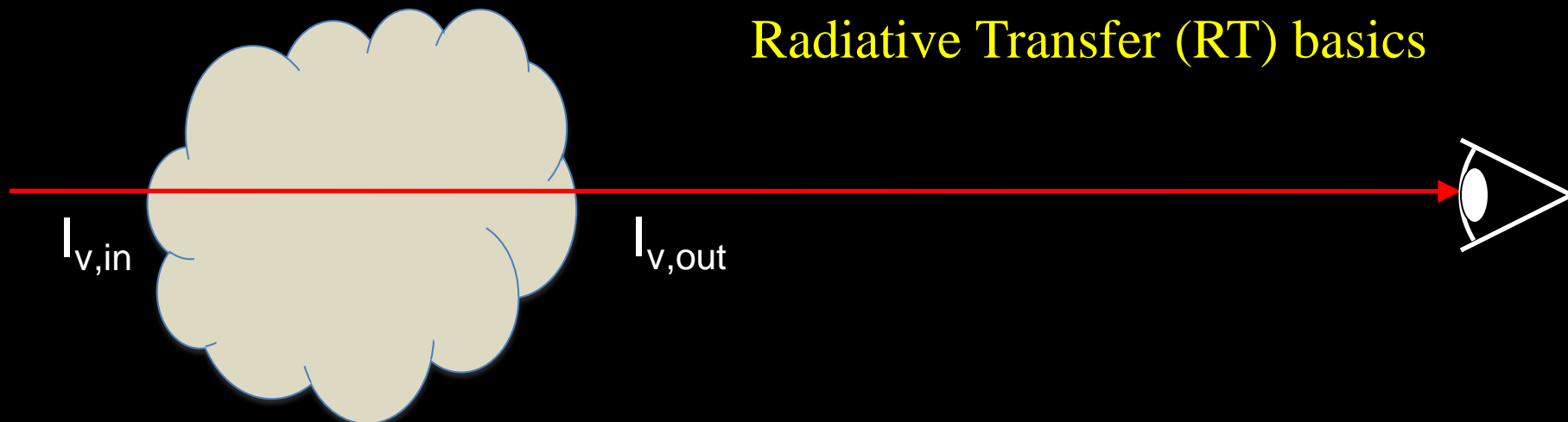


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Radiative Transfer (RT) basics



Basic quantity: *intensity*

$$I(\Omega, \nu) = \frac{\text{erg}}{\text{scm}^2 \text{Hzster}}$$

Definition of *mean intensity*:

$$J(\nu) = \frac{1}{4\pi} \oint_{4\pi} I(\Omega, \nu) d\Omega = \frac{\text{erg}}{\text{s cm}^2 \text{Hz ster}}$$

Definition of *flux*:

$$\vec{F}(\nu) = \oint_{4\pi} I(\Omega, \nu) \vec{\Omega} d\Omega = \frac{\text{erg}}{\text{scm}^2 \text{Hz}}$$



RT equation

$$\frac{dI_\nu}{d\tau_\nu} = S_\nu - I_\nu$$

Optical depth

Optical depth is a measure of the extinction coefficient or absorptivity up to a specific stellar layer.

$$\tau_\nu(D) = \int_0^D \alpha_\nu(s) ds$$

$$\alpha_\nu = \kappa_\nu \rho$$

Source function

(emissivity to absorption ratio)

κ_ν is the frequency-dependent opacity of the medium

$$l_{\text{free},\nu} = \frac{1}{\rho \kappa_\nu}$$

Photon mean free path



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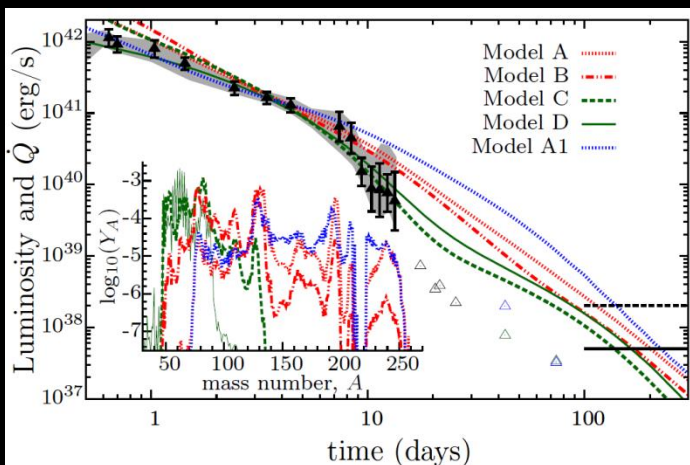


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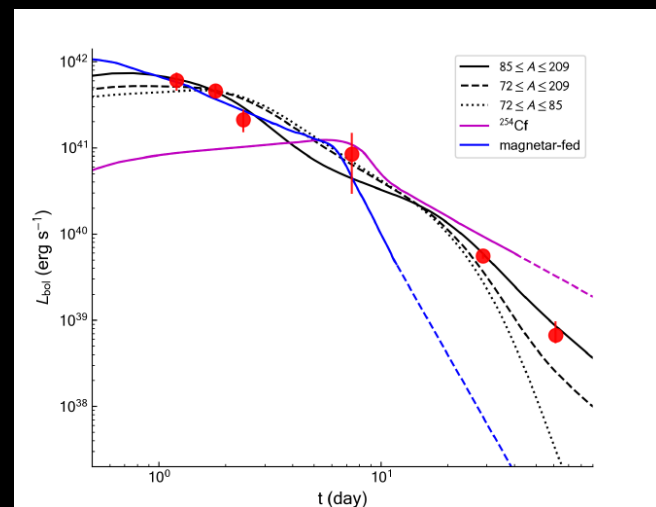
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A plethora of KN lightcurves

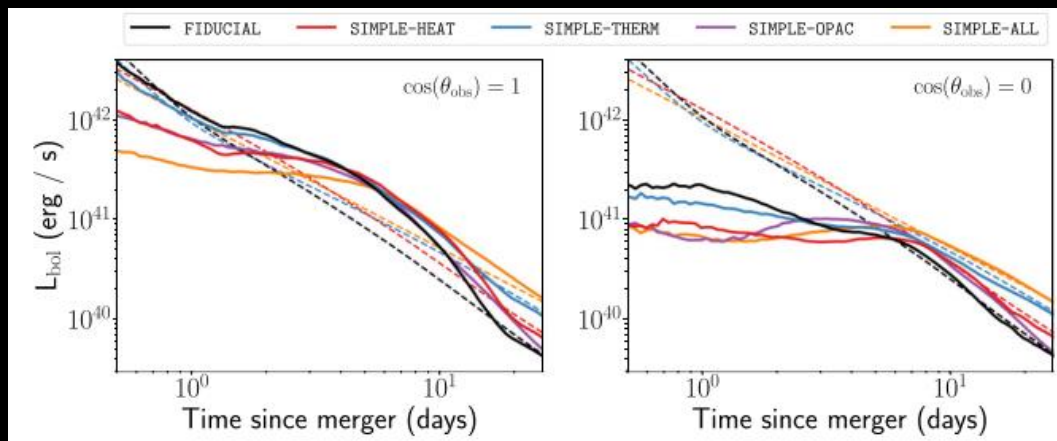


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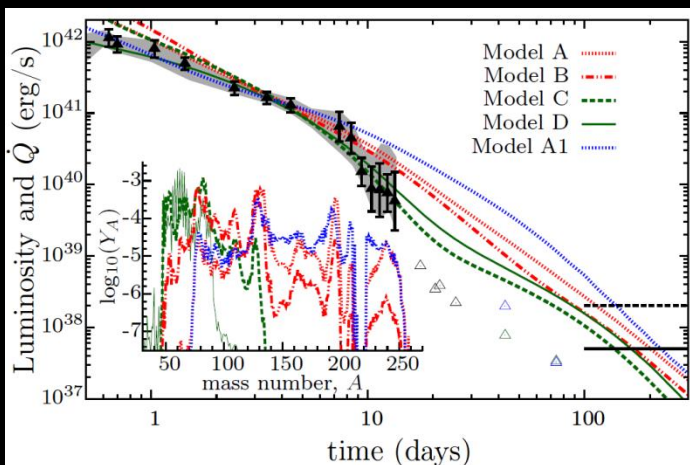


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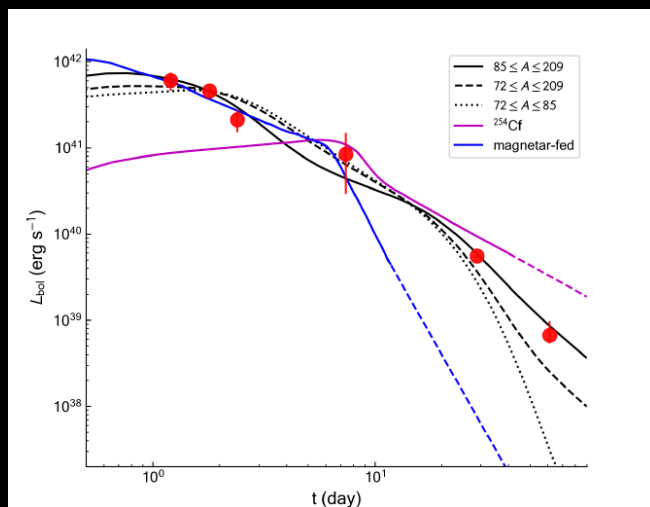
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A plethora of KN lightcurves

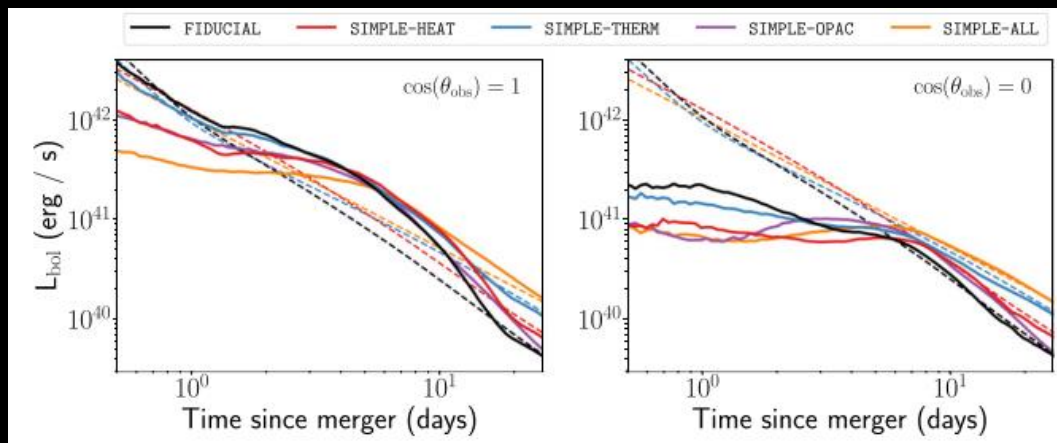


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Thanks for the attention