

Fundamental research and applications with the EuPRAXIA facility at LNF

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Large Hadron (LHC) and Future Circular (FCC) colliders

Physics at EuPRAXIA@LNF



Plasma acceleration



While globally plasma is electrically neutral, if an external electromagnetic field strong enough is applied, the plasma electrons will separate spatially from the massive ions creating a charge imbalance in the perturbed region \rightarrow charge separation field that can accelerate an injected particle beam

One can then perturbe this field to resonantly excites a plasma wake, which accelerates a trailing witness bunch injected at the accelerating phase at a few **GeV/m**.



EuPRAXIA



https://www.eupraxia-project.eu/accelerator-technology.html



Novel and small plasma accelerator compared to the FLASH accelerator at DESY. Credit: Heiner Müller-Elsner/DESY

Experimental area

The EuPRAXIA@ SPARC_Lab complex

Plasma module

1 GeV Linac

5

Photo-injector

Experimental areas

Secondary sources

Laser sources

Physics at EuPRAXIA@LNF

Undulators



EuPRAXIA Advanced Photon Sources "EuAPS"

"Advanced Photon Sources act here as drivers for plasma waves in ultra-high-gradient accelerators or as plasma-based sources of ultra-short pulses of high intensity x-rays. The proposed work will enable the production of unique ultra-short particle and photon beams with applications in ultra-fast science, amongst others highly accurate medical imaging, material characterization and medical treatment."





EuPRAXIA is a leading European project aimed at the **development** of a dedicated, groundbreaking, ultra-compact accelerator research infrastructure based on novel plasma acceleration concepts and laser technology.

LNF will be equipped with a unique combination of:

- an **X-band RF LINAC** generating high-brightness GeV-range electron beams
- o a **0.5 PW class laser system**
- the first fifth-generation free electron laser (FEL) source driven by a plasma-based accelerator (EuPRAXIA)
- Betatron radiation (EuAPS): Wiggler-like radiation emitted by electrons accelerated in plasma wakefields → it will gives rise to brilliant, ultra-short X-ray pulses





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The laser can be used to produce plasma and will allow to perform measurements as

- 1. Fusion processes of astrophysical interest in plasma
- 2. Nuclear decays in plasma





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- 1. Coherent Imaging of Biological Samples
- 2. Time-Resolved X-ray Absorption Spectroscopy in the Water Window
- 3. Time-Resolved Coherent Raman Experiments with X-ray Pulses
- 4. Photo-Fragmentation of Molecules
- 5. Resonant Inelastic X-ray Scattering
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Pulse duration [fs]





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Biology and material science



Betatron and FEL applications



Free-electron laser at EuPRAXIA:

- Photon pulses with high intensity: 10¹² photons/pulse
- \circ *λ* → 4 nm: water window
- Spectra in the *soft X-ray region*
- Pulse energy up to 180 μ J
- Bandwidth will range between 0.4% and 0.9%, according to the machine operation scheme

Betatron at EuAPS:

- Spectra in the hard X-ray region
- Peak at few keV

Both:

 \circ pulse duration of tens of fs





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Incident Energy (eV)

 $\sigma(\lambda) \rightarrow$ Some processes are activated by soft X-rays, and also need a high intensity as with FEL. However, some excitation will be beyond the reach of the FEL spectra (as Al)



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Betatron and FEL applications

Biology aims at single particle imaging!

However, at the moment no atomic resolution structural information from single biological macromolecules is feasible due to FEL intensity, the available detectors, and techniques to introduce the sample into the focused X-ray sampling position



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Water window \rightarrow region of the <u>electromagnetic spectrum</u> in which <u>water</u> is transparent to <u>soft x-rays</u>: it goes from the K-<u>absorption edge</u> of carbon at 282 eV - 68 PHz, 4.40 nm wavelength - to the <u>K-edge</u> of oxygen at 533 eV - 129 PHz, 2.33 nm wavelength.

Water is transparent to these X-rays, while carbon and its <u>organic</u> <u>compounds</u> are absorbing \rightarrow the absorption contrast between the carbon of organelles and the water of both cytoplasm and the liquid surrounding the cell is maximal!



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- Water window and room temperature
- O High degree of transverse coherence (between 80% and 100%) → 2D images of a variety of biological samples, including bacteria, viruses, cells
- Combining diffractive patterns on identical objects may allow 3D reconstruction





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Betatron and FEL applications

Warm dense matter

Article Talk

From Wikipedia, the free encyclopedia

Warm dense matter, abbreviated **WDM**, can refer to either equilibrium or non-equilibrium states of matter in a (loosely defined) regime of temperature and density between condensed matter and hot plasma. It can be defined as the state that is too dense to be described by weakly coupled plasma physics yet too hot to be described by condensed matter physics. In this state, the potential energy of the Coulomb interaction between electrons and ions is on the same order of magnitude (or even significantly exceeds) their thermal energy, while the latter is comparable to the Fermi energy.^[1] Typically, WDM has a density somewhere between 0.01 and 100 g/cm³ and a temperature on the order of several thousand kelvins (somewhere between 1 and 100 eV, in the units favored by practitioners).



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- $\circ \quad V_{\text{Coulomb}} \geq E_{\text{Thermal}}$
- Typical densities are between 0.01 and 100 g/cm³, typical temperatures between 1 and 100 eV
- Femtosecond lasers can rapidly heat matter, leading to ultrafast solid-liquid-WDM transitions



Betatron and FEL applications



Istituto Nazionale di Fisica Nucleare Laboratori Nazionali di Frascati

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Betatron and FEL applications





1s - 3p line vapor liquid solid K-edge 1.6 1.62 1.58 Energy (keV) Femtosecond lasers can rapidly heat

matter, leading to ultrafast solid-liquid-WDM transitions



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Betatron and FEL applications



Real-space distribution of the molecular charge at six representative times of the phenylalanine aminoacid after illumination with an ionizing XUV 300-as pulse

Physics at EuPRAXIA@LNF



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How organic and biological molecules redistribute the energy of absorbed light?

 → It will help understanding the basic mechanisms of photo protection/damage of amino acids, proteins and DNA/RNA 26





Nuclear physics



The periodic table of elements



Laboratori Nazionali di Frascati





2	1* Transition me				als	Is Rare-earth elements (21, 39, 57–71)												
	H 2			Other metals			and lanthanoid elements (57–71 only)						13	14	15	16	17	
2	3	4	Other nonmetals			Actinoid elements						5	6	7	8	9	10	
	Li	Be										В	С	N	0	F	ľ	
3	11	12											13	14	15	16	17	18
	Na	Mg	3	4	5	6	7	8	9	10	11	12	AI	Si	Р	S	CI	4
1	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35 D.:	36
	N	Ca	SC	10	V	Cr 40	MIN 42	Fe	60	NI 46	Cu	Z n	Ga	Ge	AS	Se	Br	5
5	Ph	Sr	39 V	40 7r	41 Nb	42 Mo	43 To	44 Du	40 Dh	Pd	4/	Cd	49 In	Sn	Sh	JZ To	53	54
	55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
	Cs	Ba	La	Hf	Та	w	Re	Os	Ir	Pt	Au	Ha	TI	Pb	Bi	Po	At	F
	87	88	89	104	105	106	107	108	109	110	111	112	113	114	115	116	117	11
7	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	FI	Мс	Lv	Ts	0
leathenoid eavies 6				59	60	61	62	63	64	65	66	67	68	69	70	71	1	
ianthanolo serie			nes o	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
actinoid series 7				91	92	93	94	95	96	97	98	99	100	101	102	103		
	aoth	1010 30	1105 1	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

 Sb
 121
 123
 Sb

 Sn
 112
 114
 115
 116
 117
 118
 119
 120
 50

 Sn
 112
 114
 115
 116
 117
 118
 119
 120
 50
 Sn
 124

 In
 113
 115
 In
 Nuclear reactions
 Nuclear reactions

 Cd
 110
 112
 113
 116
 Cd
 Muclear reactions

 Ag
 100
 110
 112
 113
 116
 Cd
 Muclear reactions

Slow neutron capture process (s-process) Occurs in very old stars over millions of years. Elements are released into the universe at the end of the star's life.



Rapid neutron capture process (r-process) Occurs in the debris ejected from a neutron star merger. The whole process takes about 1 second.

Intense neutron capture in a short time

Immediately

Why plasma



Physics at EuPRAXIA@LNF



Why plasma: fusion processes



ELECTRON SCREENING AND THERMONUCLEAR REACTIONS E. E. SALPETER 1954



Relatively small enhancements due to electron screening could cause significant errors in the extrapolation to lower energies, if the cross-section curve is forced to follow the trend of the enhanced cross sections without correcting for screening.





Deuterium fusion process in plasma





$d + d \rightarrow {}^{3}He + n (2.45 \text{ MeV})$

It is a nuclear fusion reaction crucial for understanding early phases of **Nucleosynthesis**

It took place right after the hadronization step was over, when there were free p and n that eventually combine to form deuterium.

Indirect measurements of the deuterium burning available (1.5 MeV ÷ 2 keV), also exploiting the so-called Trojan-Horse Method. **However, a full comprehension of possible electron screening effects is crucial.**



Deuterium fusion process in plasma







Model-independent determination of the astrophysical S factor in laser-induced fusion plasmas

D. Lattuada, M. Barbarino, A. Bonasera, W. Bang, H. J. Quevedo, M. Warren, F. Consoli, R. De Angelis, P. Andreoli, S. Kimura, G. Dyer, A. C. Bernstein, K. Hagel, M. Barbui, K. Schmidt, E. Gaul, M. E. Donovan, J. B. Natowitz, and T. Ditmire

Phys. Rev. C 93, 045808 - Published 19 April 2016



Why plasma: β -decays



Stellar nucleosynthesis proceeds in a hot and dense environment which affects the degree of ionization of the atoms involved in the stellar nucleosynthesis.

What happens when atoms are highly ionized?

The beta decay in highly ionized atoms shows important variations compared to neutral species

- 1. Electron Capture becomes impossible in fully ionized atoms.
- 2. Bound state β -decay typically marginal can become important.

https://www.frontiersin.org/research-topics/25146/nuclearphysics-and-astrophysics-in-plasma-traps





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Y. Litvinov and F. Bosh: Rep. Prog. Phys. 74, 016301 (2011)



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Example: Bare ¹⁸⁷**Re**⁷⁵⁺ **ions decay**, due to the boundstate beta decay, becomes 9 orders of magnitude faster than neutral ¹⁸⁷Re atoms with a half-life of 42 Gyr.

Q-value modifications: in a stellar plasma, ions are embedded in a cloud of charges, both positive and negative. These charges create EM fields which act as perturbation to the atomic/ionic levels leading to corrections of Q-values which affects the decay rates.

F. Bosch at al., Observation of Bound-State β⁻ Decay of Fully Ionized ¹⁸⁷Re: ¹⁸⁷Re-¹⁸⁷Os Cosmochronometry, Phys. Rev. Lett. 77, 1996


¹⁷⁶Lu: is a very long-lived in laboratory conditions and in principle might act as a cosmo-chronometer

Ηf

Lu

Yb

176

175

174

- the s-process branching point Ο at ¹⁷⁶Lu is among the most important ones for the understanding of slow neutron captures in the Asymptotic Giant Branch (AGB) phases of low and intermediate mass stars;
- it determines the abundance 0 of ¹⁷⁶Hf, an "s-only" nucleus
- Scenario is complex due to the presence of an isomeric state placed at 122.45 keV with a very short lifetime

See talk and poster by B. Mishra





How to measure ¹⁷⁶Lu $t_{1/2}$ in plasma?



The PANDORA experiment

Build a plasma trap where ion species are confined in a magnetic field and a plasma is created with:

- Electron density: $10^{12} \div 10^{14} \ cm^{-3}$
- \circ Electron temperature: 0.1 \div 100 keV
- Ion density: $10^{11} cm^{-3} \rightarrow$ relies on the radiactive isotope concentration in plasma
- $\circ \quad \mbox{Ion temperature:} \sim 1 \ eV \rightarrow \mbox{Ions are cold: no} \\ access to the excited states$





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¹⁷⁶Lu: lifetime vs. T – theoretical predictions



PANDORA: only ground state will be studied \rightarrow T too low to investigate variation on the isomeric state





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How can we populate the 1⁻ isomeric level?

Using a laser-plasma as a source of polychromatic high energy X-ray photon flux one could simultaneously investigate isomeric photoactivation as well as in-plasma decay rate modification of ground and isomer levels

Thermalization between the ground and isomer levels may occur

See poster by B. Mishra

The laser-plasma can be expected to produce X-ray spectra similar to the stellar interior, which can answer the question of equilibration more accurately than previous experiments on this topic!





Magnetic confinement

PRO:

- Long-living plasma (order of weeks)
- Steady state dynamical equilibrium for density and temperature (by compensating ion losses)
- Hence, over days/weeks constant values for charge state distribution of in-plasma ions
- Online monitoring of plasma density, temperature, volume, at any energy domain in nLTE conditions

CONS:

- Low density/high temperature plasma: nLTE conditions
- Difficult "plasmization" of solid/metallic isotopes
- No access to nuclear excited state studies (too low T)

Laser-induced plasma

PRO:

- High density plasma, reaching LTE
- Fully thermodynamical equilibrium allows, in principle, to estimate the population of nuclear excited states

CONS:

- Difficult to implement diagnostics following on-time the fast time-variation of plasma parameters
- Short living plasma, with duration much shorter than typical lifetimes of isotopes involved in stellar nucleosynthesis



Biophysics:

- Francesco Stellato[§]
- Antonella Balerna*

Nuclear physics:

- Giorgio Finocchiaro°
- Paola Gianotti^{*}
- Dario Lattuada^{+°}
- David Mascali°
- Alberto Mengoni[^]
- o Bharat Mishra°
- Eugenia Naselli°
- Angelo Pidatella[°]
- o Silvia Pisano*
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The environment ma «Tor



Istituto Nazionale di Fisica Nucleare Laboratori Nazionali di Frascati





backup





Biology and applications



EuPRAXIA@SPARC_LAB

Betatron and FEL applications



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EuPRAXIA@SPARC_LAB

Betatron and FEL applications

X-ray pulses can be exploited as pump pulse for stimulating chemical reactions or for generating coherent excitations, and, on the other hand, they can be used as selective probe to monitor the evolution from reactant to photoproduct





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Monitoring transient atomic motions that govern physical, chemical and biological phenomena, measuring structural molecular changes of reacting species over few Ångstrom lengths on sub-picosecond timescales \rightarrow pump-probe scheme



EuPRAXIA@SPARC_LAB Betatron: Warm Dense Matter





- 1. Cores of large planets
- 2. Systems initially solid and evolving in a plasma
- 3. X-ray driven inertial fusion implosion (aspects of indirect-drive inertial fusion)

Femtosecond lasers can rapidly heat matter, leading to ultrafast solidliquid-WDM transitions, followed by a more complex multiphase expansion at a picosecond time scale. Highly nonequilibrium states of matter are expected, due to the finite rate of energy transfer from the excited electrons to the lattice.

As the atomic structure modification is supposed to be driven by the photoexcited electrons, it is of primary importance to determine the respective time scales of the evolution of both electron and atomic structures.



EuPRAXIA@SPARC_LAB Betatron: Warm Dense Matter



Schematic of the aluminum solid-liquid-vapor transition dynamics at the atomic level



Solid Al: in the solid case, the conduction band is partially occupied by valence electrons up to the Fermi energy EF, leading to a sharp x-ray absorption K edge.

Liquid Al: the laser deposits energy in the valence electrons and induces a thermal broadening of the band occupation, leading to the K-edge broadening. The energy is transferred from electrons to the lattice, breaking the crystalline order

Atomic Al: due to high thermal pressure, hydrodynamic expansion occurs driving the transition to atomic vapor. In isolated atoms, the electrons are localized on the atomic orbitals. As the 3p orbital is partially unoccupied, the 1s-3p line can be observed in the x-ray absorption spectrum.



Differences among radiation sources





Physics at EuPRAXIA@LNF



Coherent Imaging of Biological Samples



Istituto Nazionale di Fisica Nucleare Laboratori Nazionali di Frascati

Figure 3. Simulated data for coherent imaging experiments at the EuPRAXIA@SPARC_LAB FEL. (a) A simulated diffraction pattern and the reconstruction of the electron density of a 2 μ m long spheroid, with a shape similar to that of an elongated bacterium. (b) A simulated diffraction pattern from a 600 nm diameter icosahedral virus. Simulations were performed using the software Condor [24] assuming a Gaussian-shaped beam with a diameter of 3 μ m, a wavelength of 2.87 nm and a pulse intensity of 100 μ J, which is, according to simulations, the expected pulse energy delivered on the sample by an EuPRAXIA@SPARC_LAB pulse.





Nuclear physics: fusion processes

Physics at EuPRAXIA@LNF



Deuterium fusion process in plasma



- 2. The adiabatic expansion through a supersonic nozzle in the reaction chamber induces the clusterization of the D molecules, which are then irradiated by a laser pulse.
- 3. Most of the pulse energy is absorbed by the clusters, causing the escape of the electrons and the formation of a plasma.
- 4. The high level of electrostatic fields reached in it produces the so-called Coulomb Explosion → emission of hot deuterium ions (with kinetic energy in the range tens-hundreds keV) that can fuse with ions coming from the explosion of other clusters.
- 5. High laser repetition rate and coarse granularity for the PID arrays to identify the fusion reaction products

 $d + d \rightarrow {}^{3}He + n (2.45 \text{ MeV})$

uclear tusion from laser-cluster interaction



MeV neutrons

Istituto Nazionale di Fisica Nucleare

Laboratori Nazionali di Frascati





Nuclear physics: beta decays in plasma



How to measure ¹⁷⁶Lu $t_{1/2}$ in plasma?



Scaling results to stellar environment

Build a plasma trap where ion species are confined in a magnetic field and a plasma is created with:

- Electron density: $10^{12} \div 10^{14} \ cm^{-3}$
- Electron temperature: 0.1 ÷ 100 keV
- Ion density: $10^{11} cm^{-3} \rightarrow$ relies on the radiactive isotope concentration in plasma

$$\frac{dN}{dt} = \lambda n_i V \to \int_0^{T_{meas}} dN = \int_0^{T_{meas}} \lambda n_i V \, dN$$

$$N(T_{meas}) = \lambda n_i V T_{meas}$$

 $n_i V$: density and plasma volume, constant \rightarrow to be measured using multiple diagnostic tools



T_e = 0.1-100 keV in a lab. Magnetoplasma

Istituto Nazionale di Fisica Nucleare

Laboratori Nazionali di Frascati

Variation with T_e stronger than with $\rho_e \rightarrow$ "stellar effect" can be modelled by ECR (*Electron Cyclotron Resonance*) plasma





How can we populate the 1⁻ isomeric level?

The intermixing depends on photoactivation rate λ^c of the nucleus through a bath of high energy X-ray photons obeying a Planck distribution in the thermal equilibrium stellar plasma.

Using a laser-plasma as a source of polychromatic high energy X-ray photon flux one could simultaneously investigate isomeric photoactivation as well as in-plasma decay rate modification of ground and isomer levels The experimental methodology revolves around the measurement of two quantities:

photoactivation rate λ^c (n_e,n_i, T,s)
decay rates λ^d (n_e,n_i, T,s) from g.s. and isomeric states

Thermalization between the ground and isomer levels occurs when:

 $\lambda^{c}(n_{e'}n_{i'}T,s) > = \lambda_{m}^{d}(n_{e'}n_{i'}T,s)$ (onset of equilibrium between the levels)

See poster by B. Mishra

The laser-plasma can be expected to produce X-ray spectra similar to the stellar interior, which can answer the question of equilibration more accurately than previous experiments on this topic!





Build a plasma trap where ion species are confined a magnet c field and a plasma is created with:

- Electron density: $10^{12} \div 10^{14}$ cm
- Electron temperature: 0.1 ÷ 100
- Ion density: $10^{11} cm^{-3} \rightarrow re^{-1}$ on the radiactive isotope concentration in plane 1

$$\frac{dN}{dt} = \lambda n_i V \rightarrow \int_0^{T_{meas}} dN = \int_0^{T_{meas}} \lambda n_i V \, dN$$

$$N(T_{meas}) = \lambda n_i V T_{meas}$$

Simulations by B. Mishra et al.

Exploring the onset of a (Full) Local Thermal Equilibrium:

- \circ Typical lifetime of nuclear excited states $\sim 10^{-15}~{\rm s}$
- Assuming an excited state for, *e.g.*, ¹⁷⁶Lu*, around 122.45 keV
- Considering n_e=<q>n_i= 10²⁷ m⁻³ (a typical stars interior density), at T_e=T_i=6.68 keV, the excited level lifetime is already exactly the same of the excitation rate, meaning that this level can be populated and it is in thermal equilibrium in the assumed laser-induced plasma lifetime (order of ps or tens of ps)
- Calculation also rescaled to a more realistic expected density of a real laser-induced plasma scenario (n=e=n_i= 10²⁵ m⁻³) → the required plasma temperature to get the thermal equilibrium goes to around 37.5 keV. This value seems to be however absolutely achievable in the foreseen laboratory scenario, confirming that the decay from excited states is in principle feasible.



Why plasma: β -decays



Stellar plasma environment (ρ , T) can play a major role in modifying the rates at the branching point in s-process nucleosynthesis. *Temperature dependent variations evaluated in the seminal work of Takahashi and Yoko*i.





https://doi-org.ezproxy.cern.ch/10.1103/PhysRevC.36.1522

Original predictions of modifications in β -decay rates in plasma by Takahashi and Yokoi



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T8 https://doi-org.ezproxy.cern.ch/10.1103/PhysRevC.36.1522 Original predictions of modifications in β-decay rates in plasma by Takahashi and Yokoi

kT [MeV] 0.01

NETGEN ne=1.e+26

ne=3.e+26

ne=9.5e+26

ne=30e+26 TaY87

TaY87

TaY87

TaY87

10



Fusion processes: heavy elements





Slow neutron capture process (s-process)

Occurs in very old stars over millions of years. Elements are released into the universe at the end of the star's life.



world/2018/crash-stars-reveals-origins-heavy-elements



Fusion processes: heavy elements



Rapid neutron capture process (r-process)

Occurs in the debris ejected from a neutron star merger. The whole process takes about 1 second.



Picture from *https://knowablemagazine.org/article/physical-world/2018/crash-stars-reveals-origins-heavy-elements*



s-process endpoint



Po Bi Pb TI Chart representing the final part of the *s*-process. Red horizontal lines with a circle in their right ends represent <u>neutron captures</u>; blue arrows pointing up-left Hg represent <u>beta decays</u>; green arrows pointing down-left represent <u>alpha decays</u>; cyan/light-green arrows pointing down-right represent electron captures. N

Physics at EuPRAXIA@LNF



Decay scheme for lutetium







Big-Bang Nucleosynthesis



				Alkali n	netals		📃 Ha	Halogens												
po	group			Alkaline	e-earth	metals		Noble gases												
Deri	1*			Transiti	ion met	als	Ra	Rare-earth elements (21, 39, 57–71)												
1	1			Other n	netals		an	nd lantha	anoid el	ements	(57-71	only)						2		
	н	2			iletais							13	14	15	16	17	Не			
2	3	4	ים ן	Other r	ionmeta	lis		ctinoid	elemen	IS		5	6	7	8	9	10			
2	Li	Be	B C N O F											F	Ne					
~	11	12											13	14	15	16	17	18		
3	Na	Mg	3	4	5	6	7	8	9	10	11	12	AI	Si	P	S	CI	Ar		
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36		
	K	Ca	Sc	Ti	V V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr		
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54		
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	1	Xe		
~	55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86		
0	Cs	Ba	La	Hf	Та	w	Re	Os	lr	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn		
-	87	88	89	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118		
7	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	FI	Mc	Lv	Ts	Og		
	lantha	a a i di a a	rice C	58	59	60	61	62	63	64	65	66	67	68	69	70	71]		
	iantha	ianthanolo series 6			Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu			
	o oti	nold co	rion 7	90	91	92	93	94	95	96	97	98	99	100	101	102	103]		
	acti	noid se	nes 7	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr			



Big-Bang Nucleosynthesis





actinoid series 7

Pa

Th

U

Np

Pu

Am

Bk

Cf

Es

Fm

Md

No

Lr

Cm



Stellar Nucleosynthesis



	Alkali metals						Halogens														
period	group	Alkaline-earth metals Noble gases																			
	1*			Transiti	ion met	als	Rare-earth elements (21, 39, 57–71)														
	1			0.1			an	and lanthanoid elements (57–71 only) 2													
	н	2		Other n	netals							13	14	15	16	17	He				
	3 4			Other r	nonmeta	als	Ac	ctinoid	elemen	ts			5	6	7	8	9	10			
2	Li	Be											в	С	N	0	F	Ne			
	11	12	1										13	14	15	16	17	18			
3	Na	Mg	3	4	5	6	7	8	9	10	11	12	AI	Si	Р	S	CI	Ar			
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36			
	к	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr			
-	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54			
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	1	Xe			
~	55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86			
b	Cs	Ba	La	Hf	Та	w	Re	Os	lr	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn			
-	87	88	89	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118			
7	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	FI	Mc	Lv	Ts	Og			
				58	59	60	61	62	63	64	65	66	67	68	69	70	71	1			
	lantha	Ianthanoid series 6			Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu				
		a a la ca	7	90	91	92	93	94	95	96	97	98	99	100	101	102	103	1			
	acti	noid se	nes 7	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr				



Stellar Nucleosynthesis







S- and r-process Nucleosynthesis



				Alkali n	netals		Halogens																
Deriod	group			Alkaline	e-earth	metals		Noble gases															
	1.			Transiti	ion met	als		Rare-earth elements (21, 39, 57–71)															
1	1							d lanth	anoid el	ements	(57-71	only)	2										
	H	2								<u>.</u>			13	14	15	16	17	He					
2	3	4 Other nonmetals						ctinoid	elemen	ts			5	6	7	8	9	10					
2	Li	Be	B C N O											F	Ne								
3	11	12											13	14	15	16	17	18					
	Na	Mg	3	4	5	6	7	8	9	10	11	12	AI	Si	Р	S	CI	Ar					
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36					
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr					
-	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54					
5	Rb	Sr	Y	Zr	Nb	Мо	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Те		Xe					
e	55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86					
0	Cs	Ba	La	Hf	Та	w	Re	Os	lr	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn					
7	87	88	89	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118					
	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	FI	Мс	Lv	Ts	Og					
			12																				
	lantha	noid on	rice 6	58	59	60	61	62	63	64	65	66	67	68	69	70	71]					
	anula	noiu se	nes o	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu						
	acti	noid so	rios 7	90	91	92	93	94	95	96	97	98	99	100	101	102	103	1					
	actil	iolu se	1162 /		-			-					-										

Th

Pa

U

Np

Pu Am Cm Bk Cf Es Fm Md No Lr



Cosmo-chronometer or stellar thermometer?

¹⁷⁶Lu is one of the few naturally occurring radio nucleos that have survived from the era of nucleosynthesis. Its present isotopic abundance [1] is 2.6% and its half-life is 4.08×10^{10} yr [2].











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PHYSICAL REVIEW C

VOLUME 44, NUMBER 6

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¹⁷⁶Lu: An unreliable *s*-process chronometer

K. T. Lesko, E. B. Norman, R-M. Larimer, and B. Sur Nuclear Science Division, Lawrence Berkeley Laboratory, 1 Cyclotron Road, Berkeley, California 94720 and Center for Particle Astrophysics, University of California, Berkeley, California 94720

C. B. Beausang*

Nuclear Science Division, Lawrence Berkeley Laboratory, 1 Cyclotron Road, Berkeley, California 94720 (Received 17 October 1990)

A level scheme of 176 Lu up to ~ 1400 keV excitation energy is deduced from a γ - γ coincidence experiment and previously published particle transfer data. 170 γ -ray transitions are placed between 85 levels. We identify 27 previously unknown levels and 131 previously unknown transitions in 176 Lu. With this γ -ray data we place the energy of the isomer at 122.9 keV. A level at 838.5 keV (J^{π} =5⁻, $t_{1/2}$ < 10 ns) is found to decay with substantial strength to both the ground state (7⁻, 4.08 × 10¹⁰ yr) and the 122.9 keV isomer (1⁻, 3.7 hr). The presence of this level guarantees the thermal equilibrium of 176 Lu^{8.m} for $T \ge 3 \times 10^8$ K and therefore during s-process nucleosynthesis. The resulting temperature sensitivity of its effective half-life rules out the use of 176 Lu as an s-process chronometer. The use of 176 Lu to determine s-process temperatures is discussed.








- 1. Once the solid ${}^{176}Lu$ target is hit by a laser pulse with an intensity as high as $10^{21} W/cm^2$, the ionization and the subsequent ion emission takes place
- 2. Lu ions travelling at a velocity of the order of hundreds of keV
- 3. Given the high energy administered by the laser in a short time interval, a local thermal equilibrium can be reached not only by the electrons, but also by the ion clouds, that can reach temperature as high as 10⁸K
- 4. At this temperature, the nuclei may be excited, and the Lu isomeric state ${}^{176,m}Lu$ can be populated
- 5. ¹⁷⁶Lu decays to the Hf 6⁺excited states, whose de-excitation proceeds through three different steps, leading to the subsequent emission of photons with energies equal to E_{γ} = 307, 202 and 88 keV. ^{176,m}Lu, on the other hand, directly decays to the first Hf excited state \rightarrow only the emission of a photon with E_{γ} = 88 keV is observed





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- 2. Lu ions travelling at a velocity of the order of hundreds of keV
- 3. Given the high energy administered by the laser in a short time interval, a $_{100}$ local thermal equilibrium can be reached not only by the electrons, but also by the ion clouds, that can reach temperature as high as 10^8 K⁸⁰
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Possible experimental setup for β -decay

Stopper



MCP 1 m of flight path Gamma detector array AN 102114 Course

- 1. A PW laser pulse is sent to a solid target containing the radio-isotope under investigation.
- 2. The plasma is created and a forward emission of the thermalized excited nuclei takes place.
- 3. The nuclei travel and eventually decay in flight, populating daughter nuclei in excited states.
- 4. The flight path, and then the distance between the target and a suitable stopper, must be optimized in order to guarantee a proper time window for the decay measurement ($\sim 1\mu s$).
- 5. This poses limits on the half-life range that can be explored.
- 6. The gamma emitted in the decay process may be detected through a dedicated detection system.



Projections for a 10 Hz repetition rate





Number of decays as a function of half lives

Number of decays as a function of laser time (for $\tau = 3$ years)



Projections for a 10 Hz repetition rate



Number of decays as a function of laser time





Projections for a 10 Hz repetition rate



Total number of decays 10⁸ Time window = 10^1 ns 10⁷ 10⁶ Time window = 10^2 ns 10⁵ Time window = 10^3 ns 10⁴ 10³ 10² 10 10⁻¹ 10⁻² 10² 10^{3} 10⁵ 10⁴ 10^{6} 10 10 laser time (s) at 10 Hz repetition rate

Number of decays as a function of laser time





Nuclear physics: general information



S- and r-process Nucleosynthesis





Physics at EuPRAXIA@LNF



Solar system abundances





Figure 1.1: Solar abundance distribution normalised to Silicon at 10^6 , adapted from Lodders 2003. The peaks in the distribution show the signatures of the different processes. The first peak around helium results from the primordial nucleosynthesis. The second peak around iron originates from nuclear statistical equilibrium and the following double peak structures from neutron capture processes.



Fusion processes: light elements







Fusion processes: light elements







Fusion processes: elements up to Fe







Cosmogenic origin of elements



H			Big Bang fusion Cosmic			Dying low-mass stars		Exploding massive stars Exploding				Human synthesis No stable isotopes					
Li 3	Be 4					Merging						B 5	C 6	N 7	O 8	F 9	Ne 10
Na 11	Mg 12		ray fissi	on		stars		dwarfs				Al 13	Si 14	P 15	S 16	CI 17	Ar 18
K 19	Ca 20	Sc 21	Ti 22	V 23	Cr 24	Mn 25	Fe 26	Co 27	Ni 28	Cu 29	Zn 30	Ga ³¹	Ge 32	As 33	Se 34	Br 35	Kr 36
Rb 37	Sr 38	Y 39	Zr 40	Nb 41	Mo 42	TC 43	Ru 44	Rh 45	Pd 46	Ag 47	Cd 48	In 49	Sn 50	Sb 51	Te 52	 53	Xe 54
Cs 55	Ba	°	Hf 72	Ta 73	W 74	Re ₇₅	Os 76	lr 77	Pt 78	Au 79	Hg 80	TI 81	Pb 82	Bi 83	Po 84	At 85	Rn 86
Fr 87	Ra	~		0	Dr	Nd	Dm	Sm		Cd					Tm	Vh	
	888888888888888888888888888888888888888		Ld 57	58	59	60	61	62	EU 63	64	65	Бу 66	ПО 67	6 8	69	70	LU 71
			Ac 89	Th 90	Pa 91	U 92	Np 93	Pu 94	Am 95	Cm 96	Bk 97	Cf 98	Es 99	Fm 100	Md 101	No 102	Lr 103



Main nucleosynthesis path





the *rp*- and γ processes (page 24). The *s* process and the γ process are secondary processes, which depend on certain seed isotopes, whereas fusion, the *r* process and the *rp* process are primary processes, which are not dependent on prior nucleosynthesis. Adapted from Glorius 2013.



Branching points and s-only isotopes



BRANCHING POINTS

Branching points are isotopes or isomers on the *s*-process path, which undergo a β -decay on the same timescale as a neutron capture. This leads to a branching in the *s*-process path where some of the mass flow follows the β -decay branch and the rest of the mass flow the neutron capture path (figure 1.10).



Figure 1.10: Branching point ⁸⁵Kr along the *s*-process path. ⁸⁵Kr has a half-life of about 10 years, which would make it an excellent probe for *s*-process conditions, if all reaction channels would be known to good precision. Stable isotopes are marked grey, and unstable isotopes white.

THE S-ONLY ISOTOPES

The *s*-only isotopes are isotopes, which are created almost solely by the *s* process and are shielded from other processes. These isotopes are often considered when comparing observations to simulations (figure 1.11).





7-.7-

¹⁷⁶Lu level scheme





FIG. 6. Partial level scheme of ¹⁷⁶Lu. Spin and parity assignments of the $\frac{3}{2}^{*}[411]$, $\frac{7}{2}^{-}[523]$, $\frac{5}{2}^{*}[402]$, and $\frac{9}{2}^{-}[514]$ proton orbitals coupled to the $\frac{7}{2}^{-}[514]$ neutron orbital in ¹⁷⁶Lu. The assignments for the $K^{\P} = 0^{+} \frac{7}{2}^{-}[523] - \frac{7}{2}^{-}[514]$ band are considered as tentative. Assignments for a $K^{\P} = 2^{-}$ vibrational band are included and the unassigned levels populated in the (t, α) reaction are shown on the extreme right. Gamma transitions assigned were taken from unassigned transitions of Ref. 1. All levels shown are populated in the (t, α) reaction.



¹⁷⁶Lu physics case



The Decay Scheme of Natural Lutetium 176

JAMES R. ARNOLD AND THOMAS SUGIHARA* Institute for Nuclear Studies, University of Chicago, Chicago, Illinois (Received February 26, 1953)

THE nuclide lutetium 176 is of particular interest for two related reasons; first, that it is the central member of one of the four known triads of naturally occurring adjacent isobars, and second, that its spin of at least 7 units¹ is the highest known. Flammersfeld has reported a decay scheme for this nuclide,² in which both K capture and β -decay appear, the ratio of the branches being $K/\beta^-=2$. The observed gamma-ray was placed in the K branch, and its energy fixed at 0.260 Mev.



FIG. 3. A partial level scheme of ¹⁷⁶Lu, showing the positions and decays of the ground state and isomer at 122.9 keV. The equilibration of these two levels could be achieved by way of a level of intermediate spin, as illustrated in the figure.



Drip lines







¹⁷⁶Lu physics case



PHYSICAL REVIEW C

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A level scheme of ¹⁷⁶Lu up to ~ 1400 keV excitation energy is deduced from a $\gamma - \gamma$ coincidence experiment and previously published particle transfer data. 170 γ -ray transitions are placed between 85 levels. We identify 27 previously unknown levels and 131 previously unknown transitions in ¹⁷⁶Lu. With this γ -ray data we place the energy of the isomer at 122.9 keV. A level at 838.5 keV ($J^{\pi}=5^{-}$, $t_{1/2} < 10$ ns) is found to decay with substantial strength to both the ground state (7⁻, 4.08×10¹⁰ yr) and the 122.9 keV isomer (1⁻, 3.7 hr). The presence of this level guarantees the thermal equilibrium of ¹⁷⁶Lu^{g,m} for $T \ge 3 \times 10^8$ K and therefore during s-process nucleosynthesis. The resulting temperature sensitivity of its effective half-life rules out the use of ¹⁷⁶Lu as an s-process chronometer. The use of ¹⁷⁶Lu to determine s-process temperatures is discussed.



¹⁷⁶Lu physics case



¹⁷⁶Lu is one of the few naturally occurring radio nuclides that have survived from the era of nucleosynthesis. Its present isotopic abundance [1] is 2.6% and its half-life is 4.08×10^{10} yr [2].





Figure 1. s-process path in the rare earth element mass region. s-only process nuclides 170 Yb, 176 Lu and 176 Hf are shielded from r-process contributions by 170 Er and 176 Yb respectively. The s-process branches at 176 Lu if a significant population of the 3.68 h isomeric state occurs.



¹⁷⁶Lu branch in the s-process





Figure 1. s-process path in the rare earth element mass region. s-only process nuclides 170 Yb, 176 Lu and 176 Hf are shielded from r-process contributions by 170 Er and 176 Yb respectively. The s-process branches at 176 Lu if a significant population of the 3.68 h isomeric state occurs.



Experiments at the PW regime





High-density target \rightarrow solid, Lu target

Target choice to be optimized in view of the thermalization goal

Possible choice: non-isotopic cuboids of metallic bulk material (*e.g.*, natural Lu) coated with a layer of at least 200-300 µm of isotopic enriched material

https://www.frontiersin.org/articles/10.3389/fphy.2022.72 7718/full



Why laser

Physics at EuPRAXIA@LNF



Laser-matter interaction





When a high intensity laser pulse (above 10^{18} W/cm²) is focused in a spot of the order of a few microns on a target placed in vacuum, a plasma consisting of electrons and ions is created almost instantaneously.

- Target Normal Sheath Acceleration (TNSA): effective in accelerating protons and light ions → a short laser pulse interacting with the target front surface produces a plasma made of ions and fast electrons.
- **Coulomb Explosion (CE):** optimized for clustered gaseous targets, intensities in the range $10^{18} \div 10^{20}$ W/cm² and $\tau < 200$ fs \rightarrow an explosion may occur due to the intense laser field that, extricating several electrons from the molecule cluster, induces a high level of ionization. Possible also for thin (1-10 nm), solid targets or nano-structured targets



Laser-matter interaction





When a high intensity laser pulse (above 10^{18} W/cm²) is focused in a spot of the order of a few microns on a target placed in vacuum, a plasma consisting of electrons and ions is created almost instantaneously.

• Radiation Pressure Acceleration (RPA), or Laser Piston regime: based on the action of the radiation pressure induced in the interaction of a short laser pulse, of extremely high intensity (above $10^{20} \div 10^{21}$ W/cm^2), with a thin and dense pre-plasma layer created, in front of a target, by the laser-pulse leading edge. The plasma electrons are locally separated from the plasma ions creating a strong accelerating field which efficiently accelerates the ions in the irradiated target area.



Laser-matter interaction





A precise control of the experimental conditions is challenging with high-power lasers \rightarrow variations observed between experiments performed in conditions which would seem similar at a first glance.

The scaling of the most important characteristics (such as the energy per particle) with laser and target parameters is still unclear to a large extent, despite the large number of investigations performed.



Experiments at the 10² TW regime





Low-density target → one of the most effective way for transferring energy from lasers to a gas target occurs when the molecules in the gas are organized in clusters

If the electromagnetic field is strong enough the cluster atoms are ionized, and a Coulomb Explosion can take place.





- 1. Once the solid ${}^{176}Lu$ target is hit by a laser pulse with an intensity as high as $10^{21} W/cm^2$, the ionization and the subsequent ion emission takes place
- 2. Lu ions travelling at a velocity of the order of hundreds of keV
- 3. Given the high energy administered by the laser in a short time interval, a local thermal equilibrium can be reached not only by the electrons, but also by the ion clouds, that can reach temperature as high as 10⁸K
- 4. At this temperature, the nuclei may be excited, and the Lu isomeric state ${}^{176,m}Lu$ can be populated
- 5. ¹⁷⁶Lu decays to the 16.6⁺ excited states, whose de-excitation proceeds through three different scope, leading to the subsequent emission of photons with energies equal to E_{γ} = 307, 202 and 88 keV. ^{176,m}Lu, on the other hand, directly decays to the first Hf excited state \rightarrow only the emission of a photon with E_{γ} = 88 keV is observed



Experiments at the PW regime





High-density target \rightarrow solid, Lu target

Target choice to be optimized in view of the thermalization goal.

https://www.frontiersin.org/articles/10.3389/fphy.2022.72 7718/full