

# Nuclear Physics at EuPRAXIA

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# The National Scientific Committee 3

## 6. APPLICATIONS AND SOCIETAL BENEFITS

FOOT

## 5. FUNDAMENTAL INTERACTIONS

LEA, ALPHA, JEDI, VIP, FAMU

## 1. QUARKS AND HADRON DYNAMICS

KAONNIS, JLAB12, MAMBO, ULYSSES, EIC

## 2. PHASE TRANSITION IN HADRONIC MATTER

ALICE, NA60+

## 3. NUCLEAR STRUCTURE AND REACTION MECHANISMS

FORTE, GAMMA, CHIRONE, NUCL-EX, NUMEN, PRISMA\_FIDES

## 4. NUCLEAR ASTROPHYSICS

ASFIN, ERNA, LUNA, n\_TOF, PANDORA, Future Experiments

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# The periodic table of elements

period	group 1*	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1	1 <b>H</b>																	2 <b>He</b>
2	3 <b>Li</b>	4 <b>Be</b>											5 <b>B</b>	6 <b>C</b>	7 <b>N</b>	8 <b>O</b>	9 <b>F</b>	10 <b>Ne</b>
3	11 <b>Na</b>	12 <b>Mg</b>											13 <b>Al</b>	14 <b>Si</b>	15 <b>P</b>	16 <b>S</b>	17 <b>Cl</b>	18 <b>Ar</b>
4	19 <b>K</b>	20 <b>Ca</b>	21 <b>Sc</b>	22 <b>Ti</b>	23 <b>V</b>	24 <b>Cr</b>	25 <b>Mn</b>	26 <b>Fe</b>	27 <b>Co</b>	28 <b>Ni</b>	29 <b>Cu</b>	30 <b>Zn</b>	31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>	36 <b>Kr</b>
5	37 <b>Rb</b>	38 <b>Sr</b>	39 <b>Y</b>	40 <b>Zr</b>	41 <b>Nb</b>	42 <b>Mo</b>	43 <b>Tc</b>	44 <b>Ru</b>	45 <b>Rh</b>	46 <b>Pd</b>	47 <b>Ag</b>	48 <b>Cd</b>	49 <b>In</b>	50 <b>Sn</b>	51 <b>Sb</b>	52 <b>Te</b>	53 <b>I</b>	54 <b>Xe</b>
6	55 <b>Cs</b>	56 <b>Ba</b>	57 <b>La</b>	72 <b>Hf</b>	73 <b>Ta</b>	74 <b>W</b>	75 <b>Re</b>	76 <b>Os</b>	77 <b>Ir</b>	78 <b>Pt</b>	79 <b>Au</b>	80 <b>Hg</b>	81 <b>Tl</b>	82 <b>Pb</b>	83 <b>Bi</b>	84 <b>Po</b>	85 <b>At</b>	86 <b>Rn</b>
7	87 <b>Fr</b>	88 <b>Ra</b>	89 <b>Ac</b>	104 <b>Rf</b>	105 <b>Db</b>	106 <b>Sg</b>	107 <b>Bh</b>	108 <b>Hs</b>	109 <b>Mt</b>	110 <b>Ds</b>	111 <b>Rg</b>	112 <b>Cn</b>	113 <b>Nh</b>	114 <b>Fl</b>	115 <b>Mc</b>	116 <b>Lv</b>	117 <b>Ts</b>	118 <b>Og</b>
lanthanoid series 6	58 <b>Ce</b>	59 <b>Pr</b>	60 <b>Nd</b>	61 <b>Pm</b>	62 <b>Sm</b>	63 <b>Eu</b>	64 <b>Gd</b>	65 <b>Tb</b>	66 <b>Dy</b>	67 <b>Ho</b>	68 <b>Er</b>	69 <b>Tm</b>	70 <b>Yb</b>	71 <b>Lu</b>				
actinoid series 7	90 <b>Th</b>	91 <b>Pa</b>	92 <b>U</b>	93 <b>Np</b>	94 <b>Pu</b>	95 <b>Am</b>	96 <b>Cm</b>	97 <b>Bk</b>	98 <b>Cf</b>	99 <b>Es</b>	100 <b>Fm</b>	101 <b>Md</b>	102 <b>No</b>	103 <b>Lr</b>				

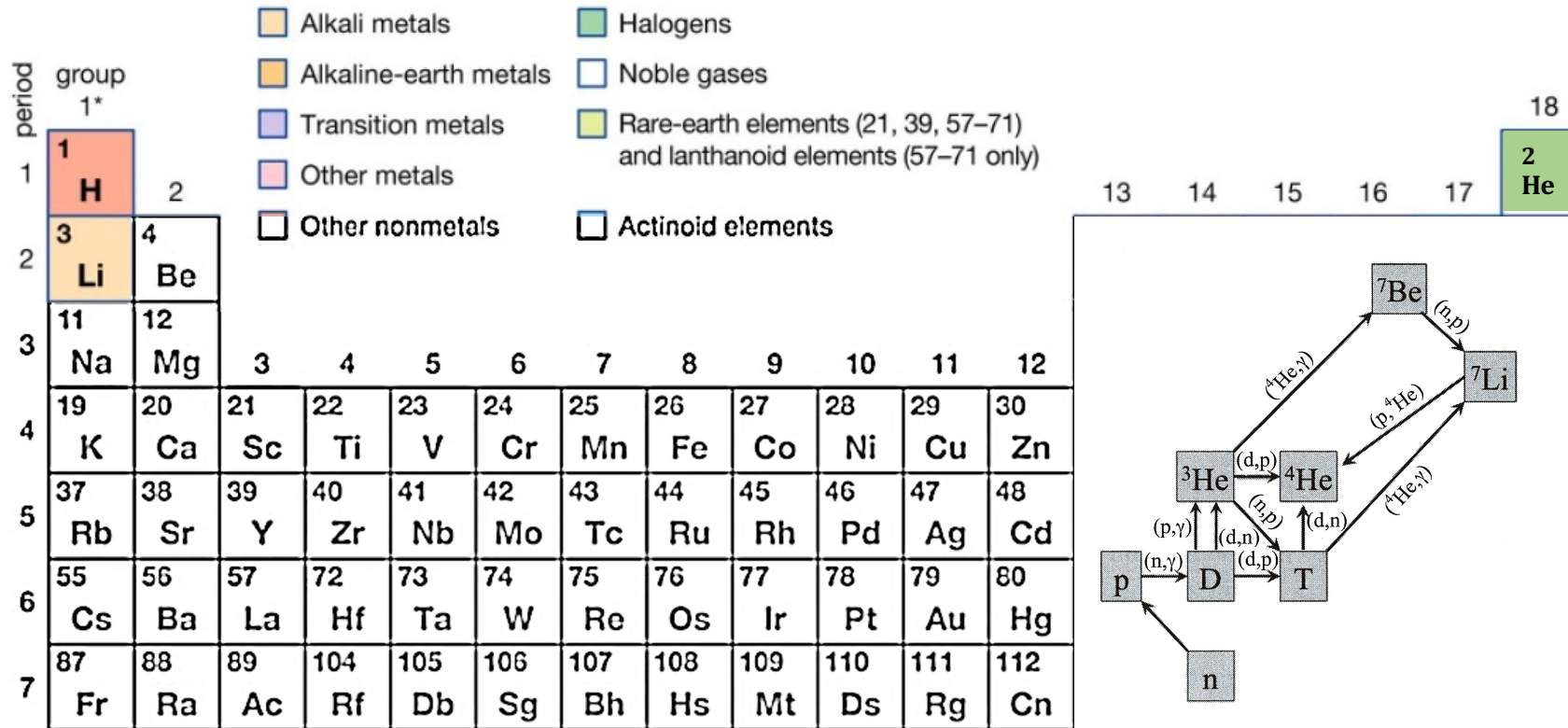


# Big-Bang Nucleosynthesis

period	group 1*	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1	1 H																	2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
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5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og

lanthanoid series 6	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
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# Big-Bang Nucleosynthesis



lanthanoid series 6	58	59	60	61	62	63	64	65	66	67	68	69	70	71
	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
actinoid series 7	90	91	92	93	94	95	96	97	98	99	100	101	102	103
	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

# Stellar Nucleosynthesis

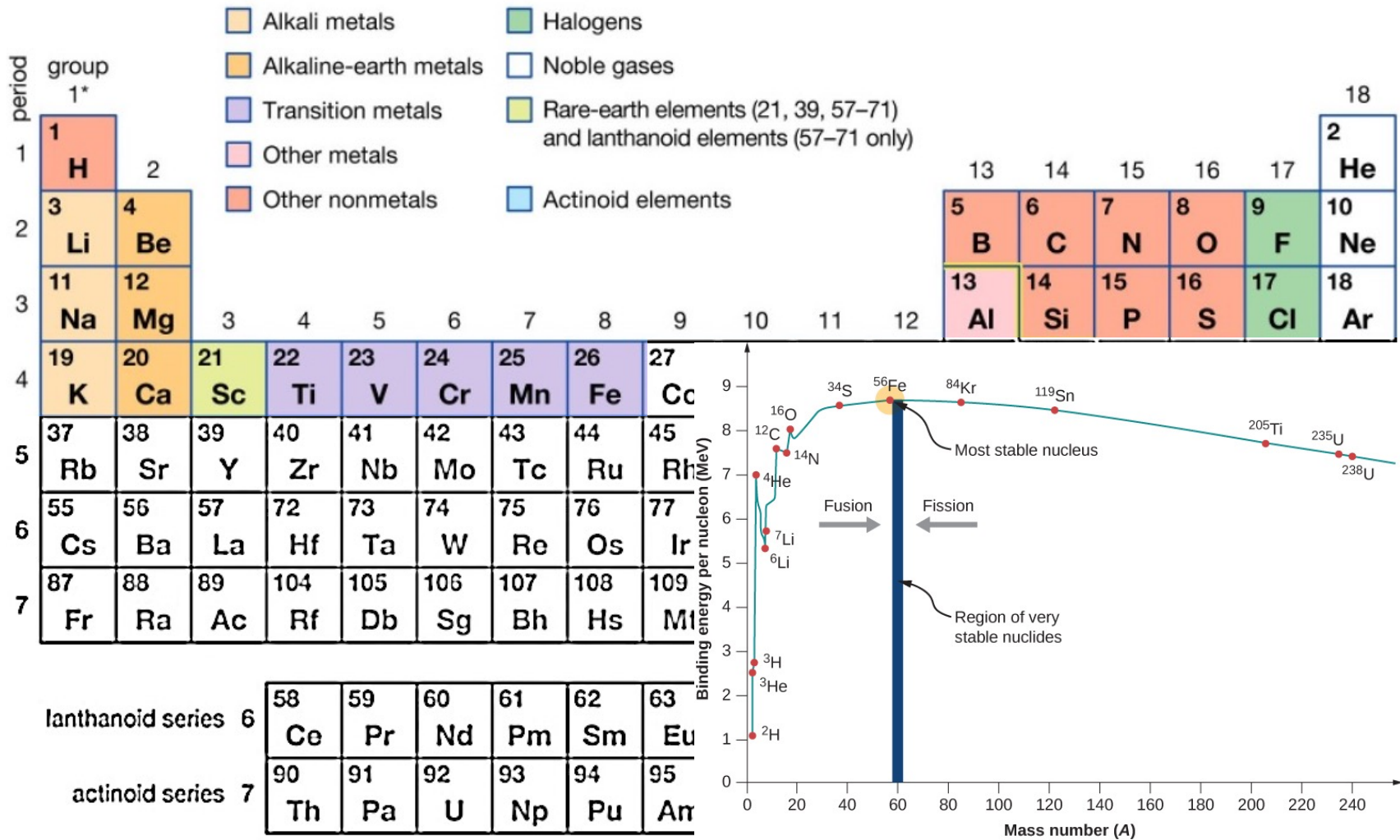
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# Stellar Nucleosynthesis



# S- and r-process Nucleosynthesis

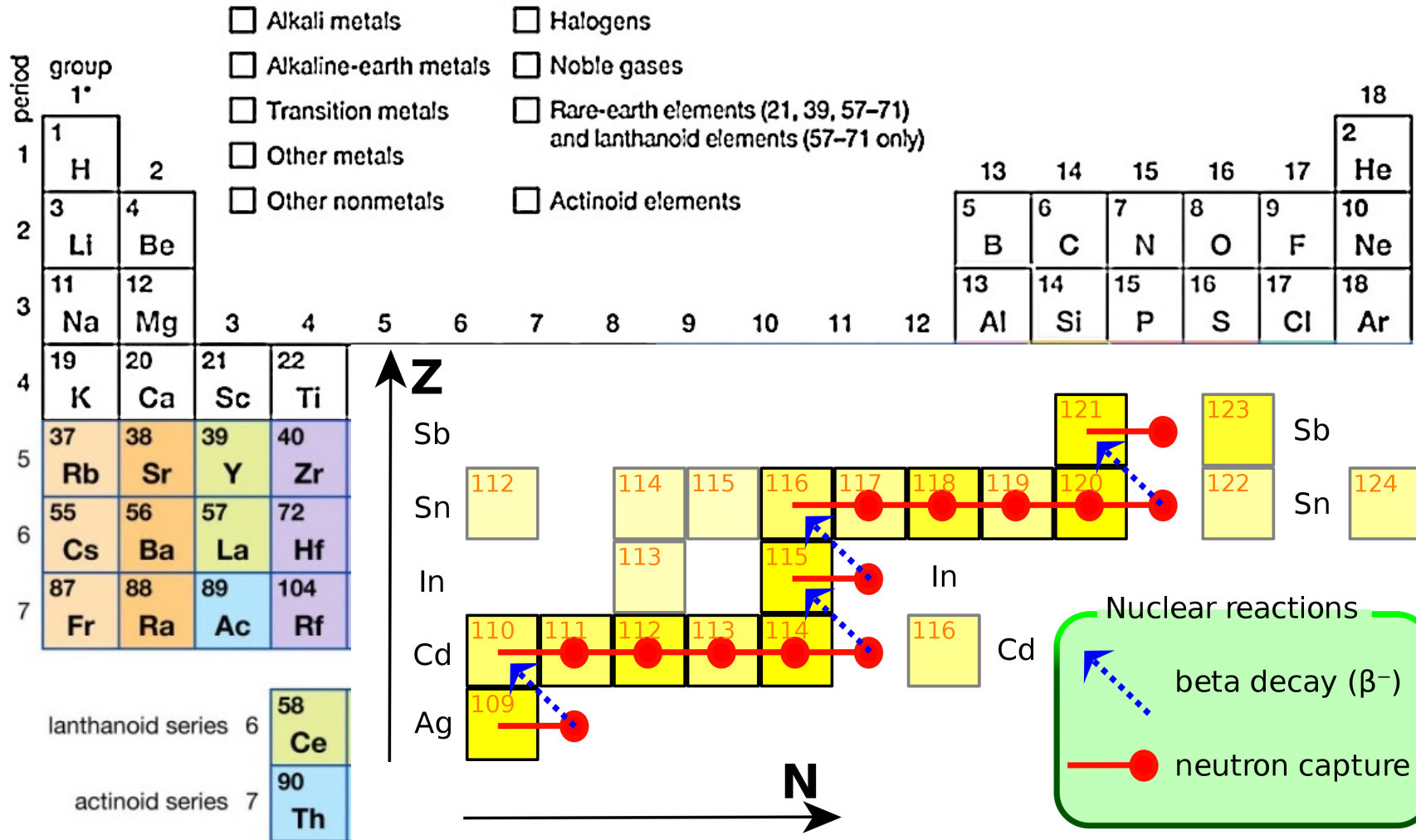
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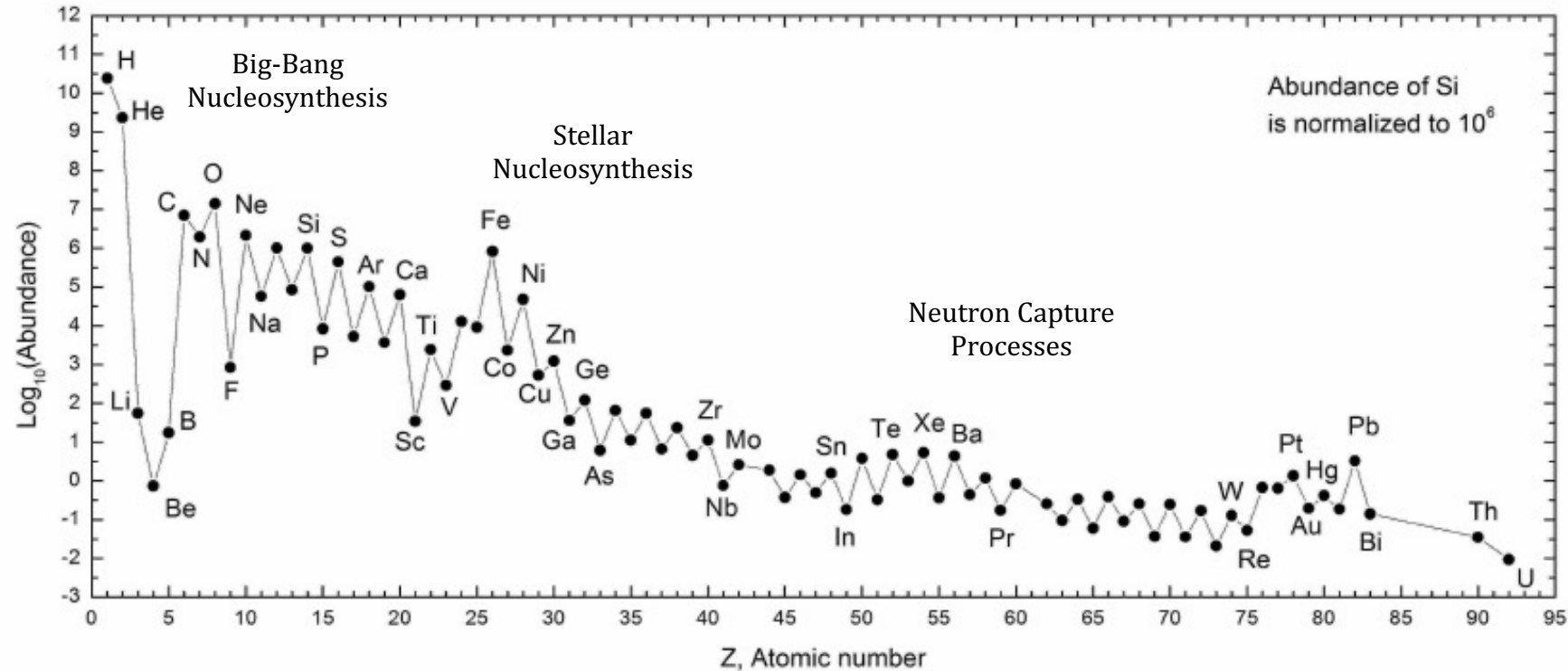
- Alkali metals
- Alkaline-earth metals
- Transition metals
- Other metals
- Other nonmetals
- Halogens
- Noble gases
- Rare-earth elements (21, 39, 57-71) and lanthanoid elements (57-71) only
- Actinoid elements

# S- and r-process Nucleosynthesis



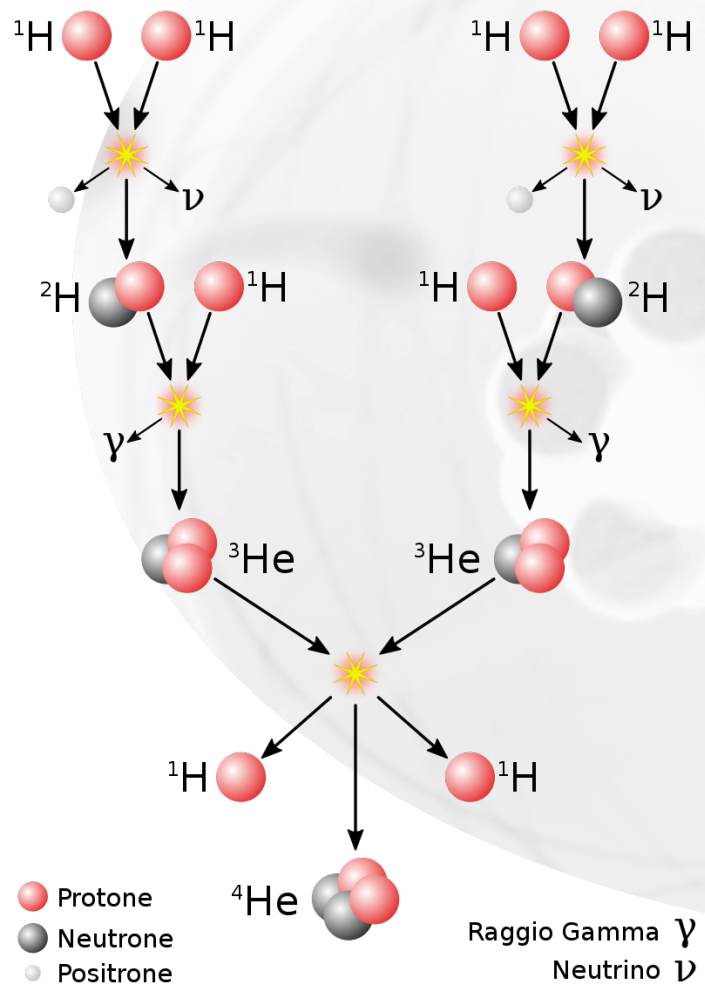


# Solar system abundances

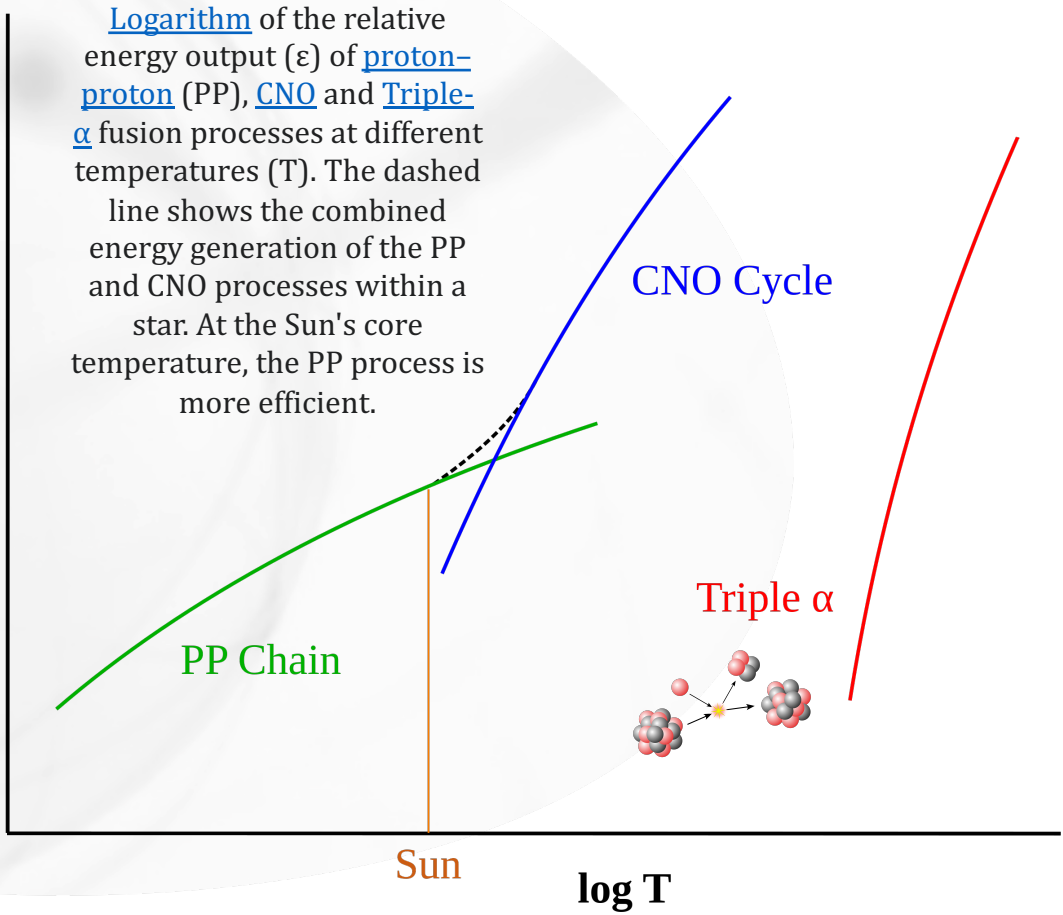
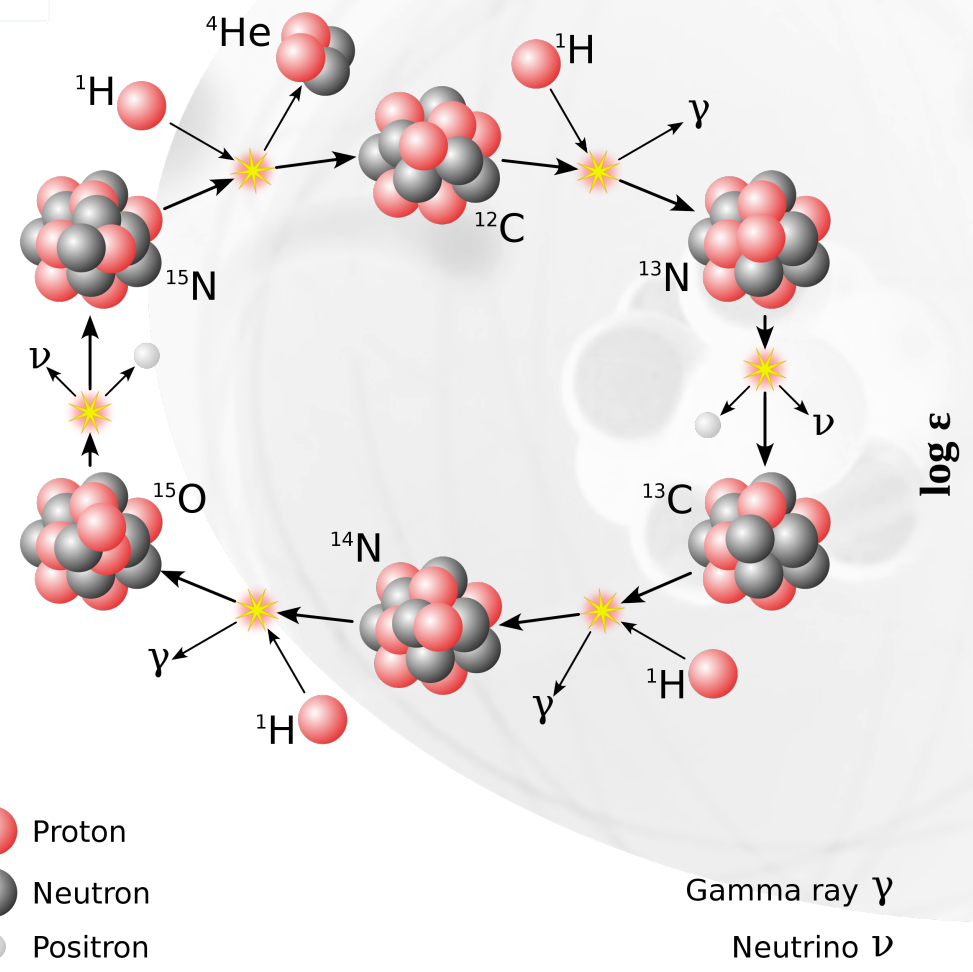


**Figure 1.1:** Solar abundance distribution normalised to Silicon at  $10^6$ , adapted from [Lodders 2003](#). The peaks in the distribution show the signatures of the different processes. The first peak around helium results from the primordial nucleosynthesis. The second peak around iron originates from nuclear statistical equilibrium and the following double peak structures from neutron capture processes.

# Fusion processes: light elements



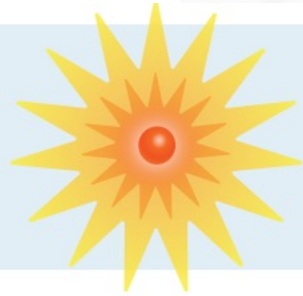
# Fusion processes: elements up to Fe





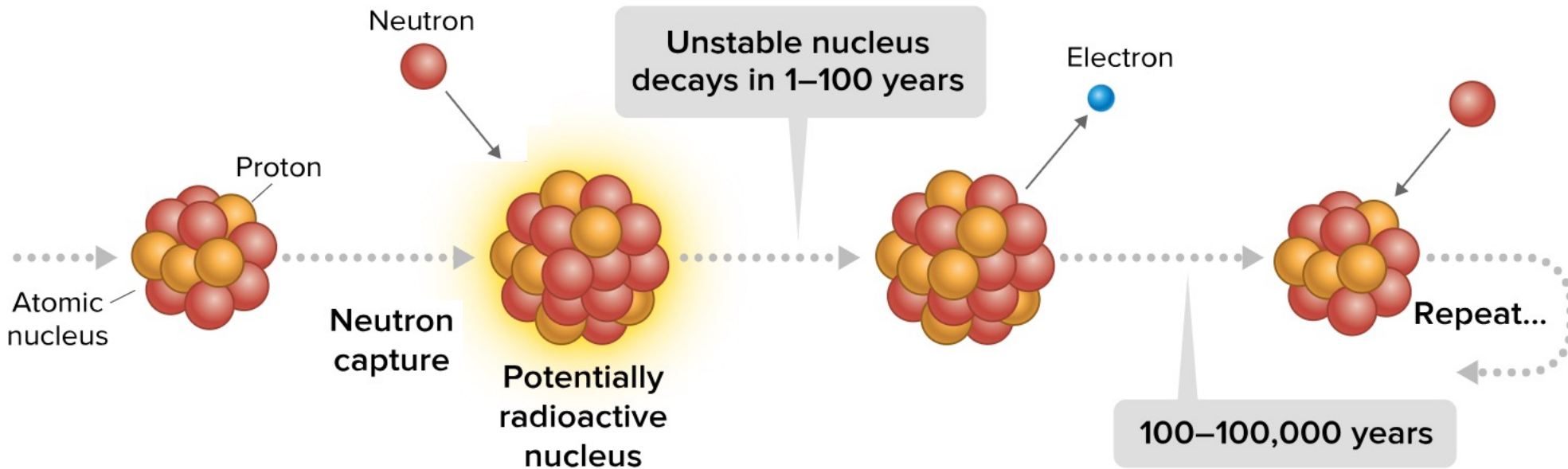


# Fusion processes: heavy elements



## Slow neutron capture process (s-process)

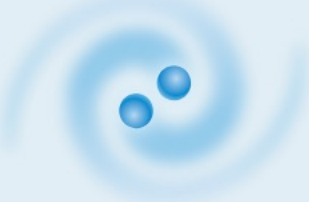
Occurs in very old stars over millions of years. Elements are released into the universe at the end of the star's life.



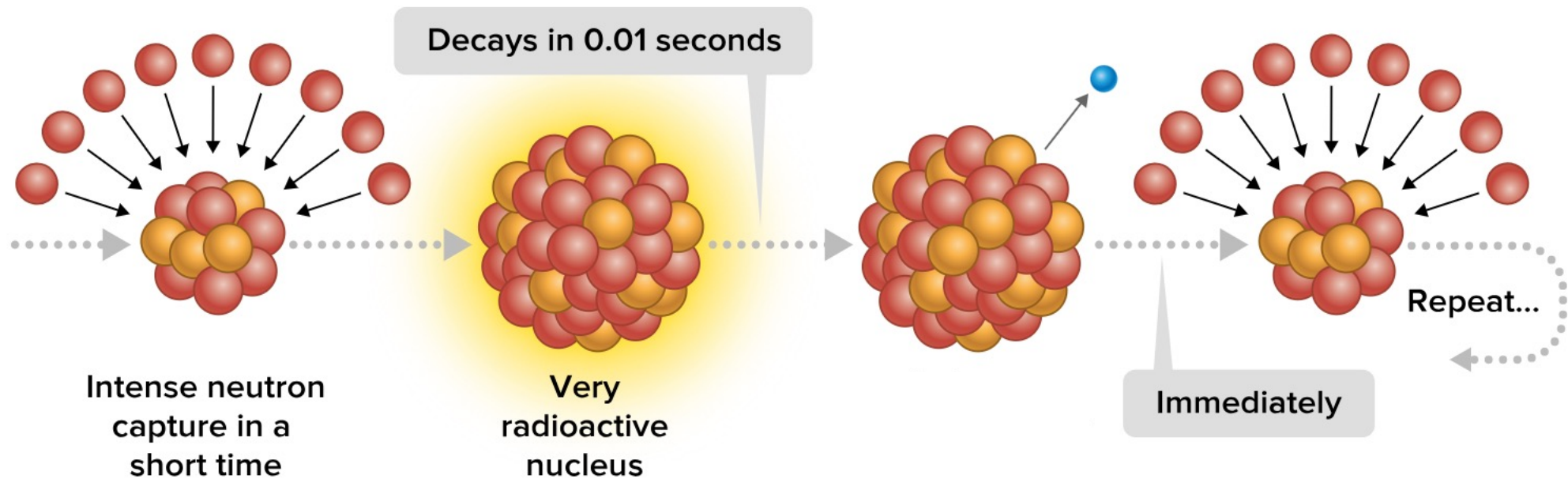
Picture from <https://knowablemagazine.org/article/physical-world/2018/crash-stars-reveals-origins-heavy-elements>



# Fusion processes: heavy elements



**Rapid neutron capture process (r-process)**  
Occurs in the debris ejected from a neutron star merger.  
The whole process takes about 1 second.



Picture from <https://knowablemagazine.org/article/physical-world/2018/crash-stars-reveals-origins-heavy-elements>



# Why plasma



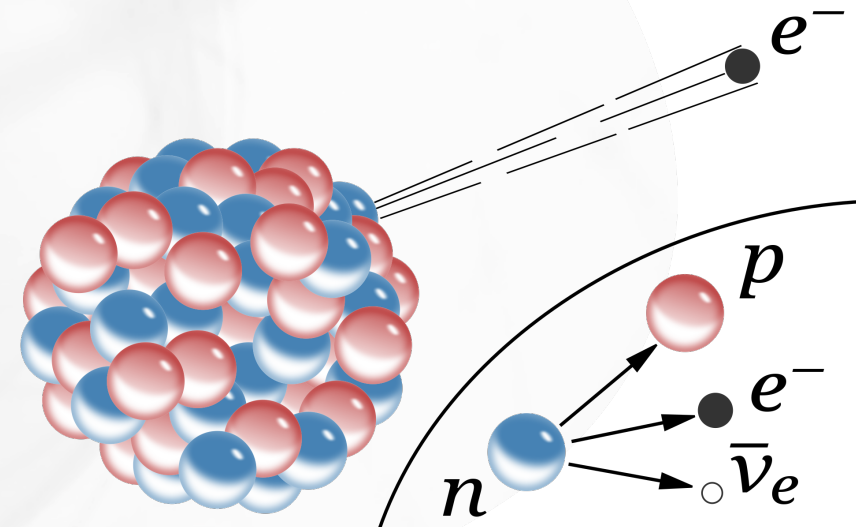
# Why plasma: $\beta$ -decays

Stellar nucleosynthesis proceeds in a hot and dense environment which affects the degree of ionization of the atoms involved in the stellar nucleosynthesis.

**What happens when atoms are highly ionized?**

**The beta decay in highly ionized atoms shows important variations compared to neutral species**

1. Electron Capture becomes impossible in fully ionized atoms.
2. Bound state  $\beta$ -decay typically marginal can become important.



<https://www.frontiersin.org/research-topics/25146/nuclear-physics-and-astronomy-in-plasma-traps>



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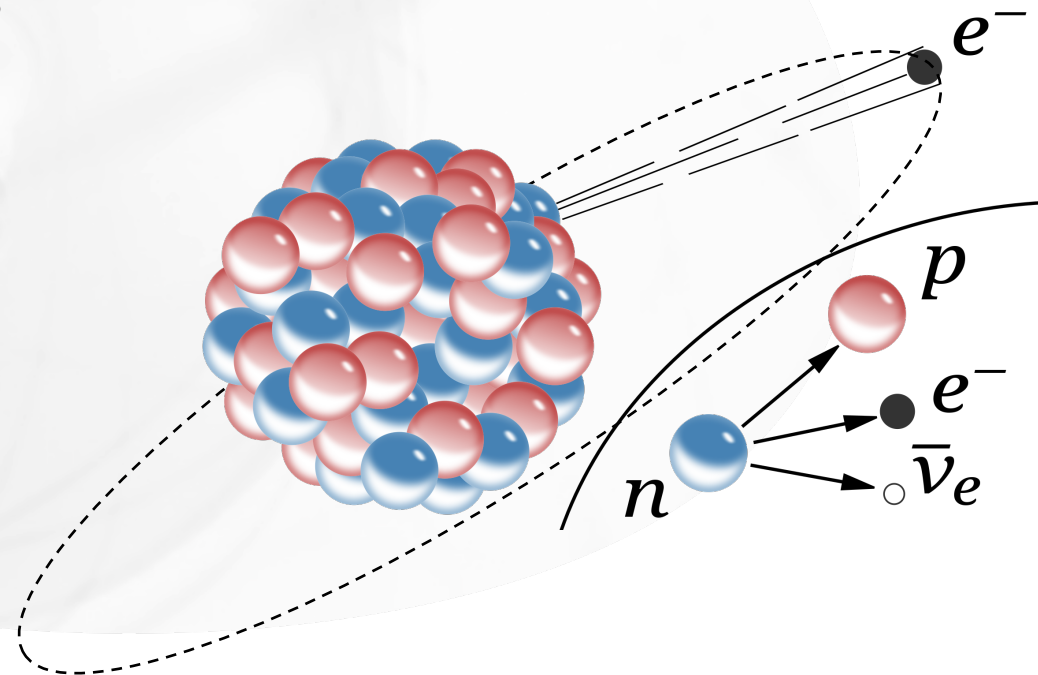
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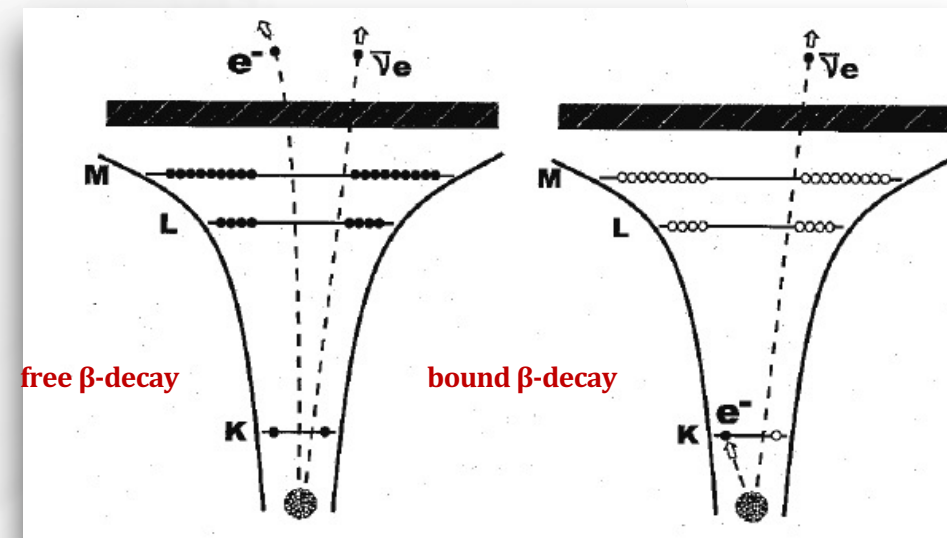
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Y. Litvinov and F. Bosh: *Rep. Prog. Phys.* 74, 016301 (2011)

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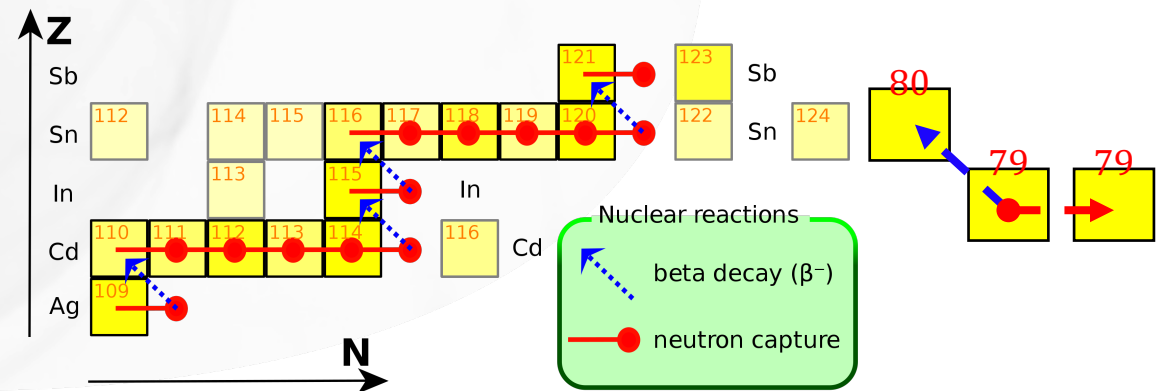
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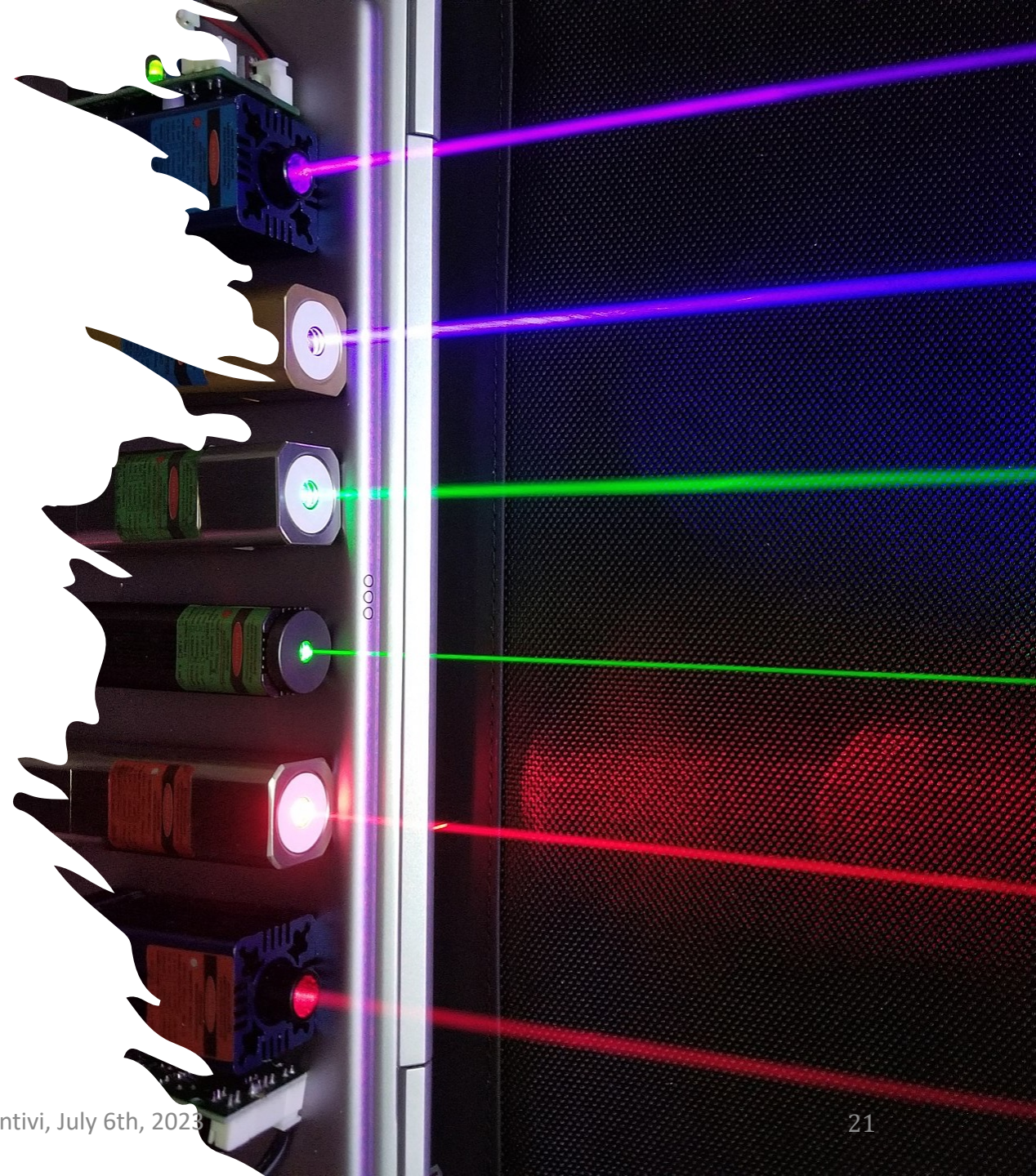
**Bound-state  $\beta$ -decay** is a nuclear  $\beta$ -decay process in which an electron is created in a previously unoccupied atomic orbital rather than in the continuum.

**Example: Bare  $^{187}\text{Re}^{75+}$  ions decay, due to the bound-state beta decay, becomes 9 orders of magnitude faster than neutral  $^{187}\text{Re}$  atoms with a half-life of 42 Gyr.**





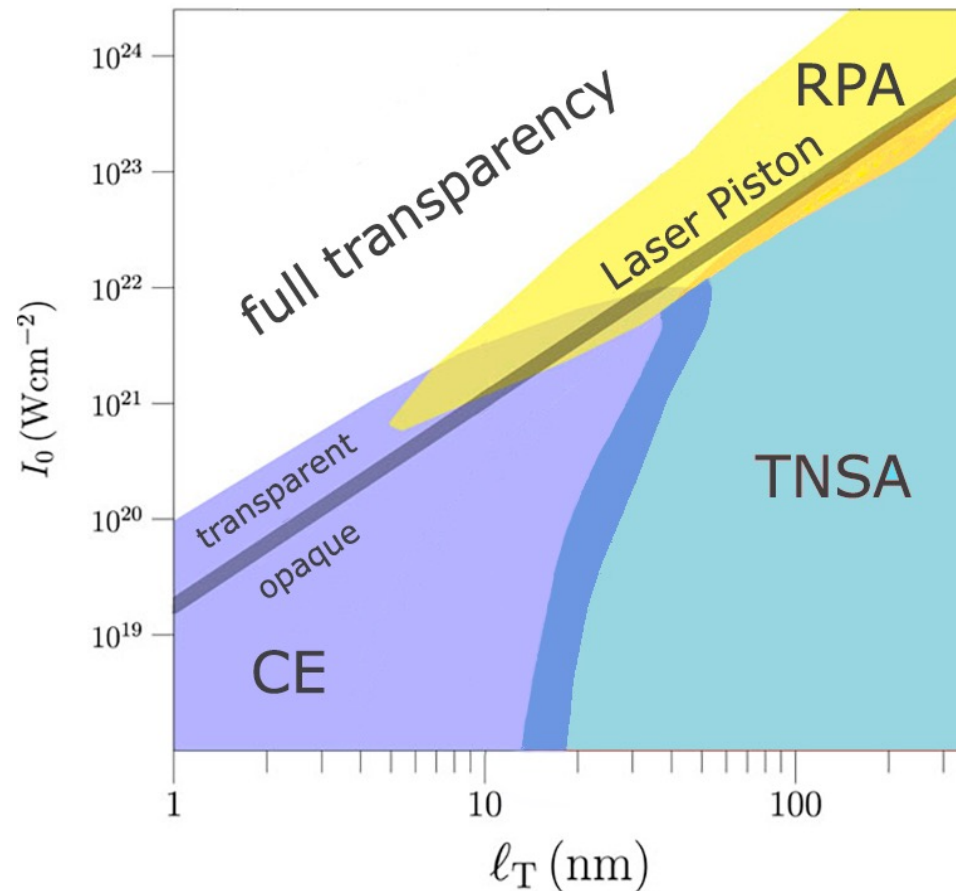
# Why laser







# Laser-matter interaction

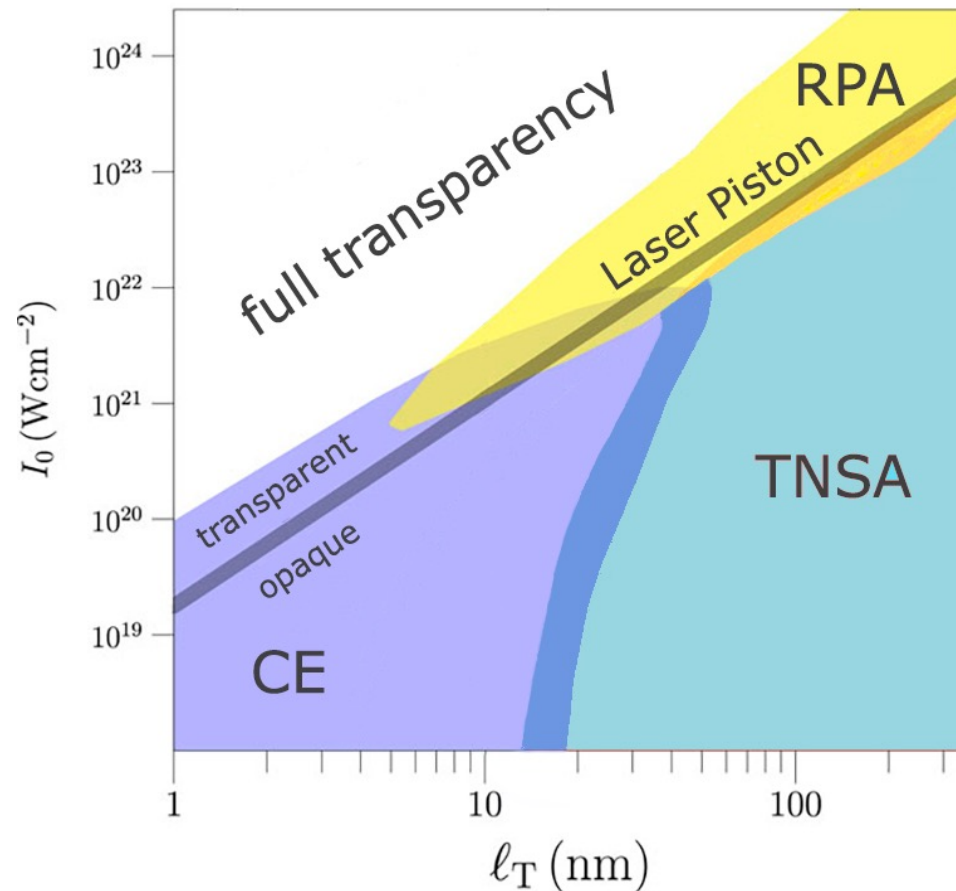


When a high intensity laser pulse (above  $10^{18} \text{ W/cm}^2$ ) is focused in a spot of the order of a few microns on a target placed in vacuum, a plasma consisting of electrons and ions is created almost instantaneously.

- **Target Normal Sheath Acceleration (TNSA):** effective in accelerating protons and light ions  $\rightarrow$  a short laser pulse interacting with the target front surface produces a plasma made of ions and fast electrons.
- **Coulomb Explosion (CE):** optimized for clustered gaseous targets, intensities in the range  $10^{18} \div 10^{20} \text{ W/cm}^2$  and  $\tau < 200 \text{ fs}$   $\rightarrow$  an explosion may occur due to the intense laser field that, extricating several electrons from the molecule cluster, induces a high level of ionization. Possible also for thin (1-10 nm), solid targets or nano-structured targets



# Laser-matter interaction

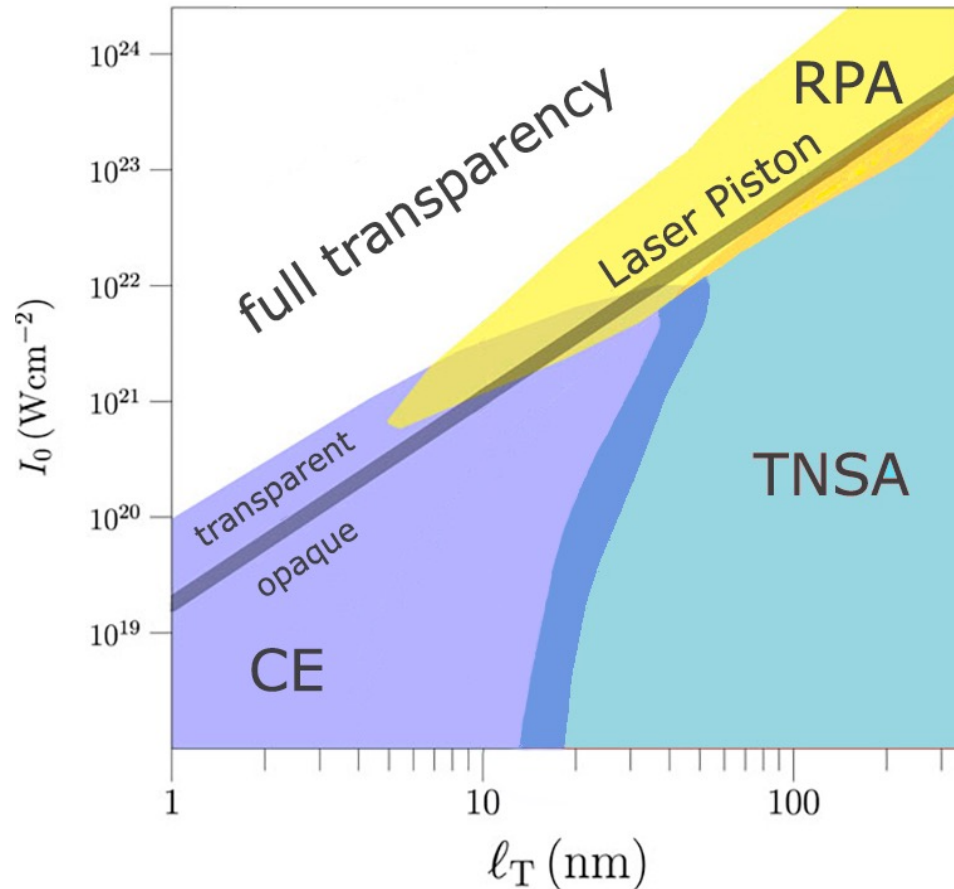


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- **Radiation Pressure Acceleration (RPA), or Laser Piston regime:** based on the action of the radiation pressure induced in the interaction of a short laser pulse, of extremely high intensity (above  $10^{20} \div 10^{21} \text{ W/cm}^2$ ), with a thin and dense pre-plasma layer created, in front of a target, by the laser-pulse leading edge. The plasma electrons are locally separated from the plasma ions creating a strong accelerating field which efficiently accelerates the ions in the irradiated target area.



# Laser-matter interaction

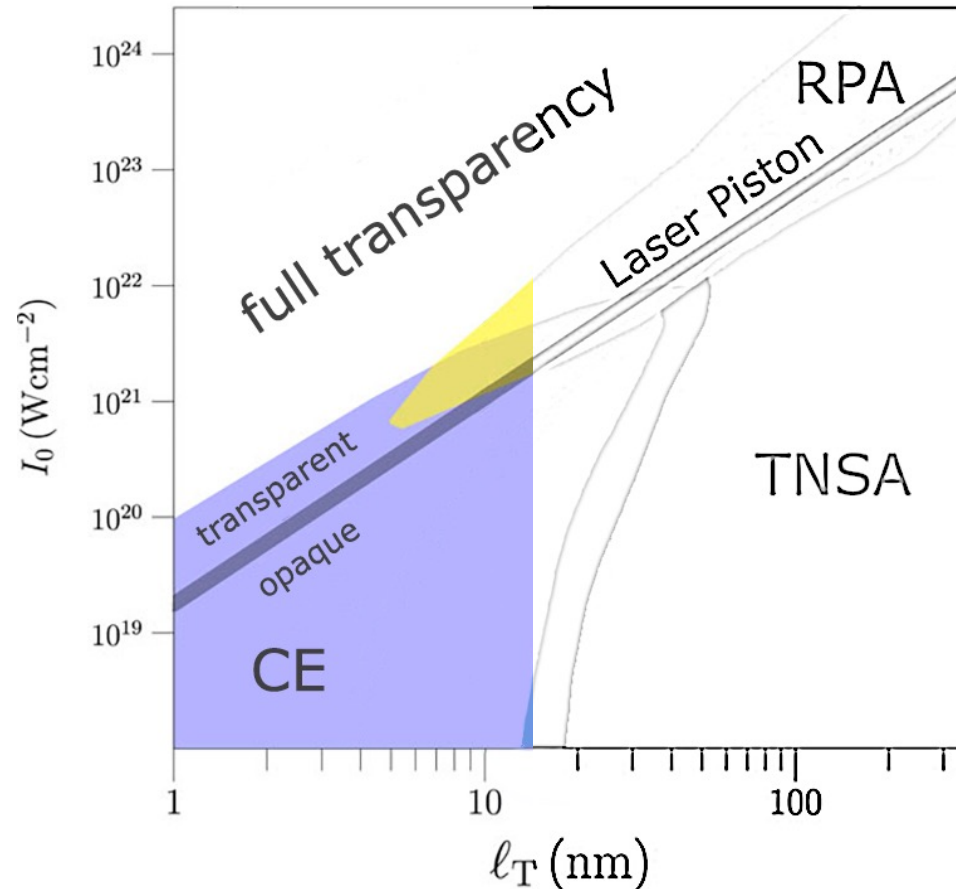


**A precise control of the experimental conditions is challenging with high-power lasers** → variations observed between experiments performed in conditions which would seem similar at a first glance.

The scaling of the most important characteristics (such as the energy per particle) with laser and target parameters is still unclear to a large extent, despite the large number of investigations performed.



# Experiments at the $10^2$ TW regime

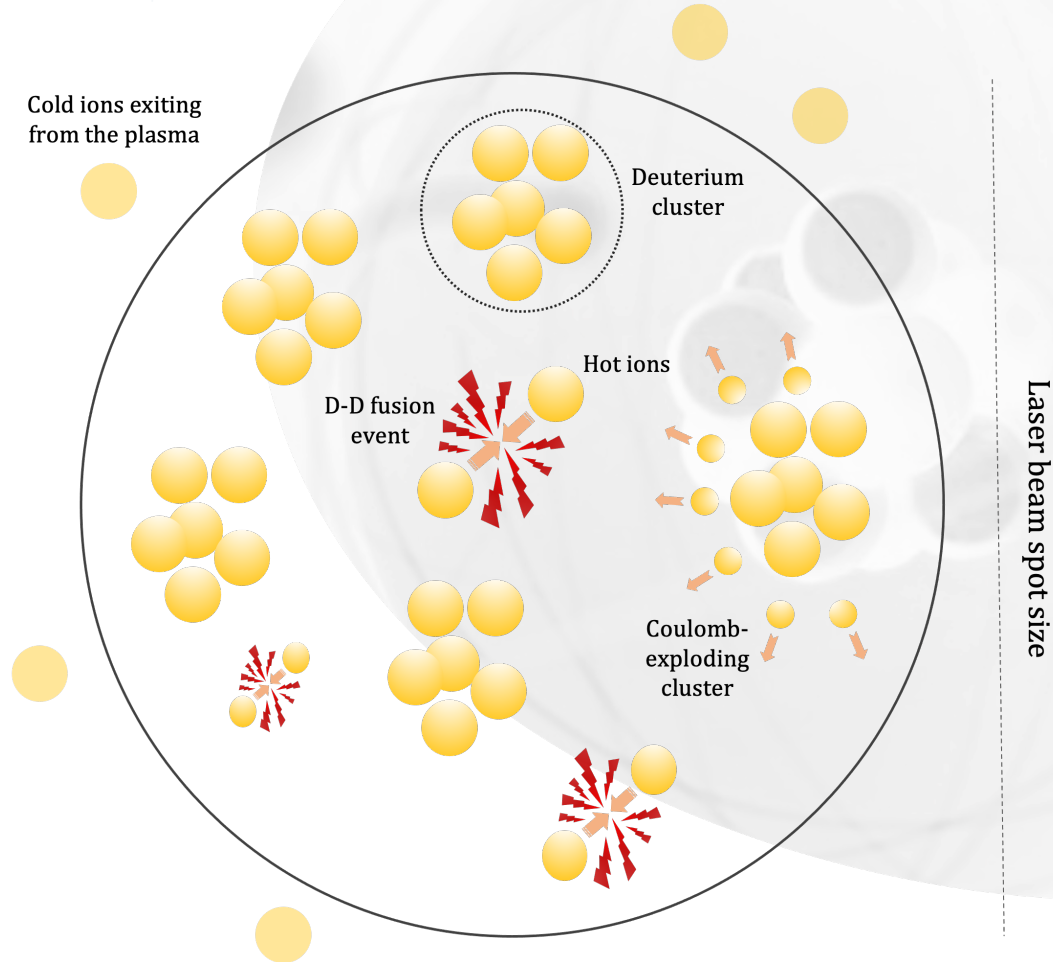
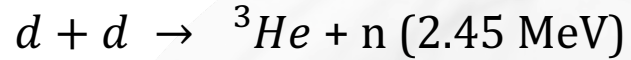


**Low-density target** → one of the most effective way for transferring energy from lasers to a gas target occurs **when the molecules in the gas are organized in clusters**

If the electromagnetic field is strong enough the cluster atoms are ionized, and a Coulomb Explosion can take place.



# Deuterium fusion process in plasma



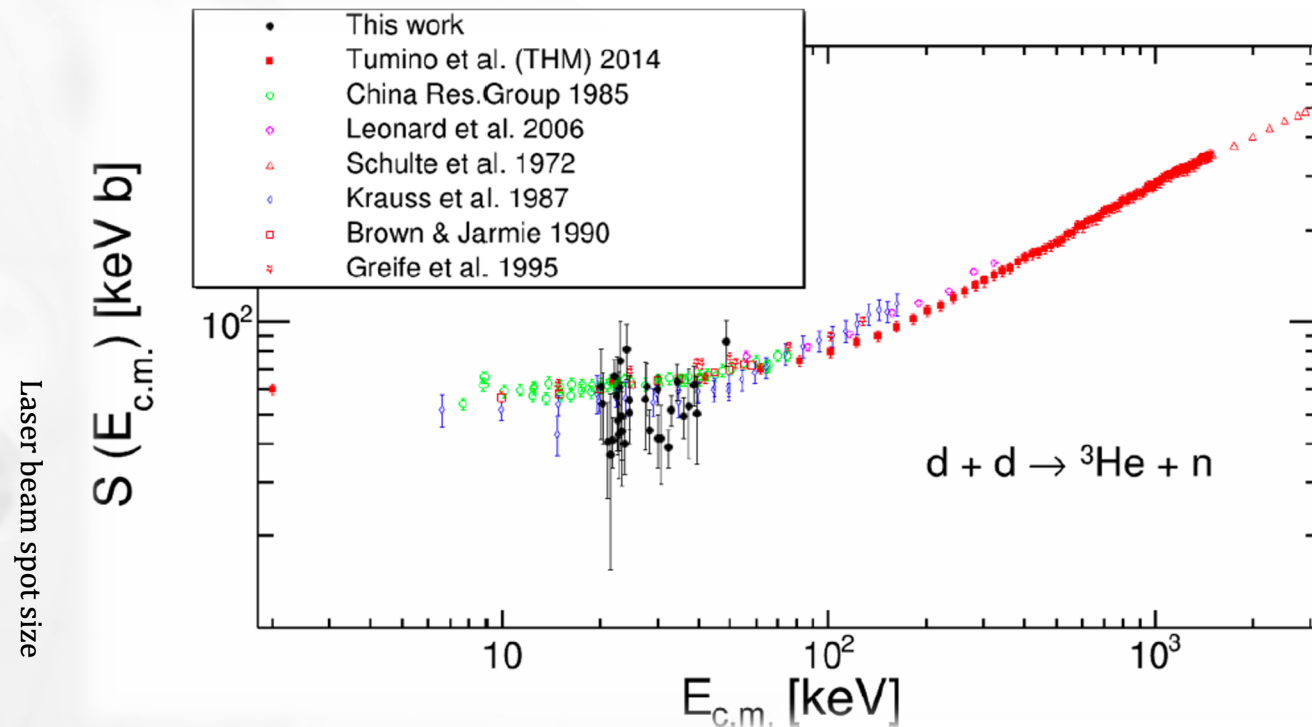
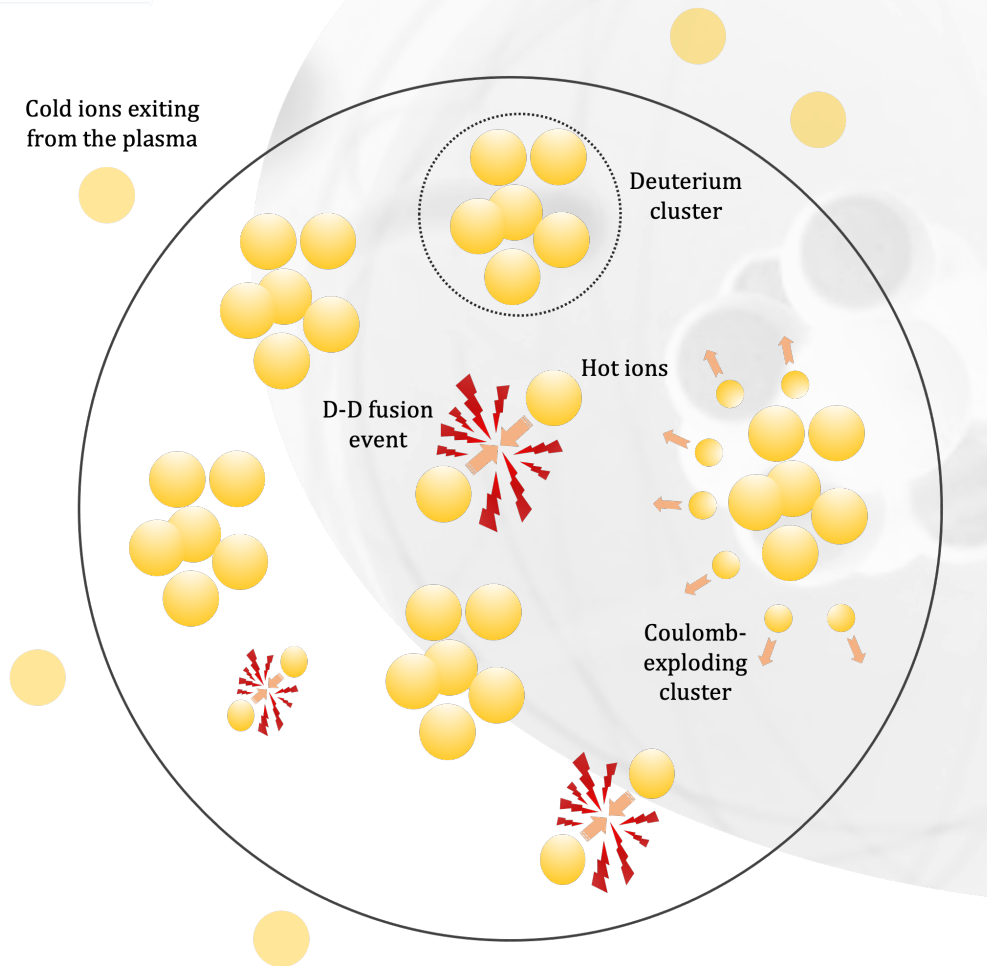
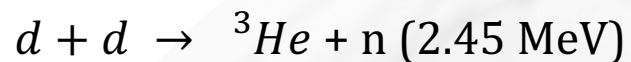
It is a nuclear fusion reaction occurred **in the Big-Bang Nucleosynthesis** → a *deuterium nucleus* formed from a *proton and a neutron* fuses with another *proton* to form a *helium-3 nucleus*.

It took place right after the hadronization step was over, when there were free p and n that eventually combine to form deuterium.

Indirect measurements of the deuterium burning available (1.5 MeV ÷ 2 keV), also exploiting the so-called Trojan-Horse Method. **However, a full comprehension of possible electron screening effects is crucial.**



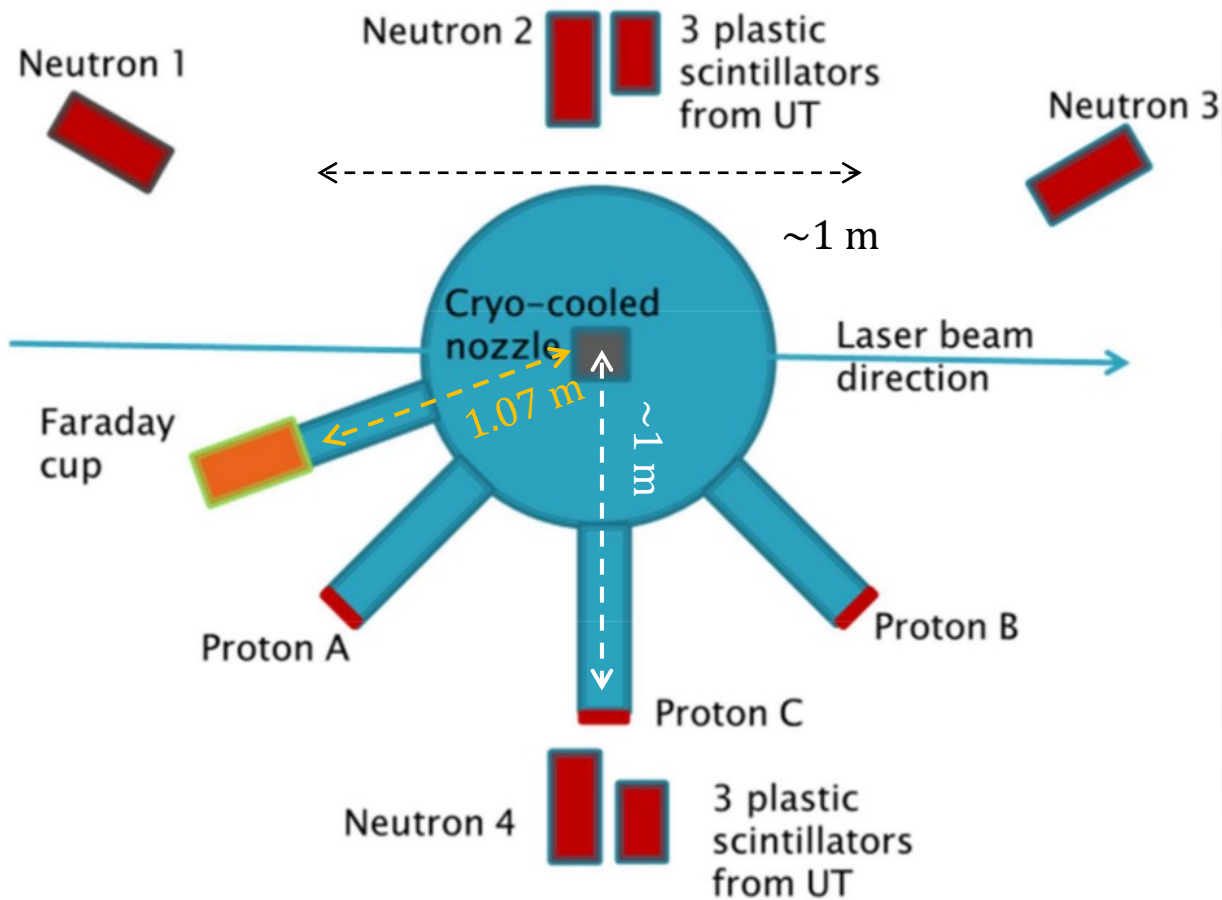
# Deuterium fusion process in plasma



Model-independent determination of the astrophysical S factor in laser-induced fusion plasmas

D. Lattuada, M. Barbarino, A. Bonasera, W. Bang, H. J. Quevedo, M. Warren, F. Consoli, R. De Angelis, P. Andreoli, S. Kimura, G. Dyer, A. C. Bernstein, K. Hagel, M. Barbui, K. Schmidt, E. Gaul, M. E. Donovan, J. B. Natowitz, and T. Ditmire  
Phys. Rev. C **93**, 045808 – Published 19 April 2016

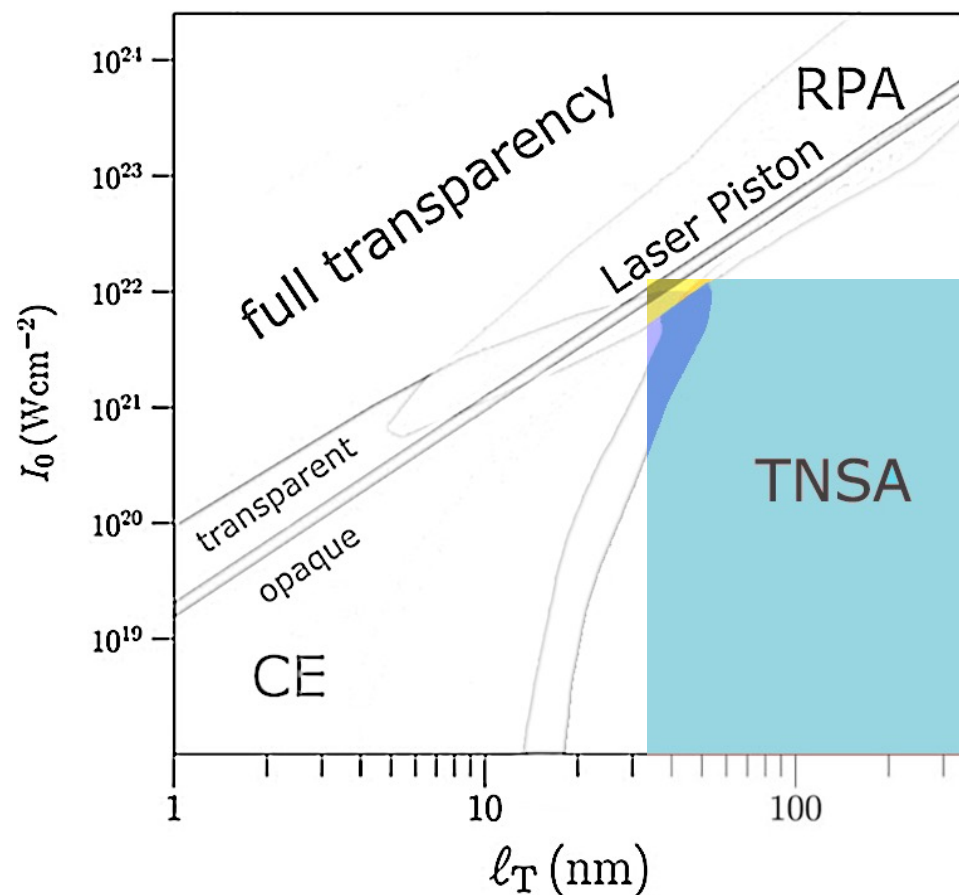
# Experimental area: an example



- Proper *time-of-flight paths* must be foreseen for a reliable particle identification
- Neutron detectors must be kept displaced enough from the walls
- Cablings, signal transportation



# Experiments at the PW regime



**High-density target** → solid, Lu target

Target choice to be optimized in view of the thermalization goal.

<https://www.frontiersin.org/articles/10.3389/fphy.2022.727718/full>



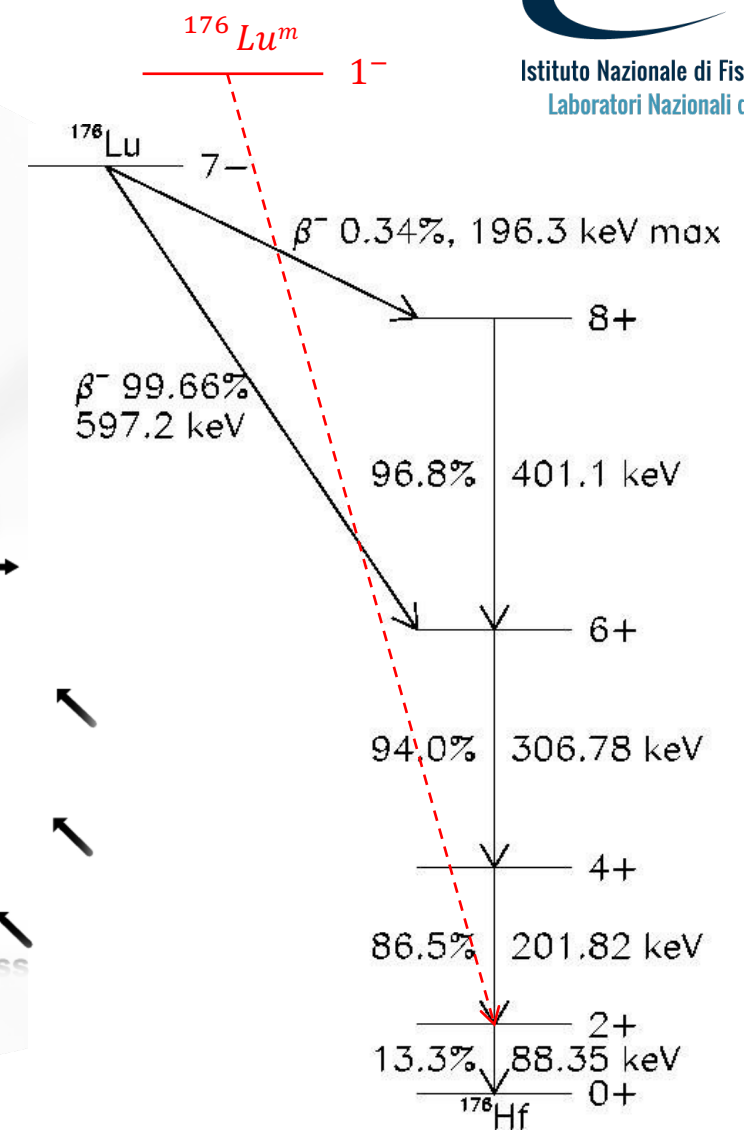
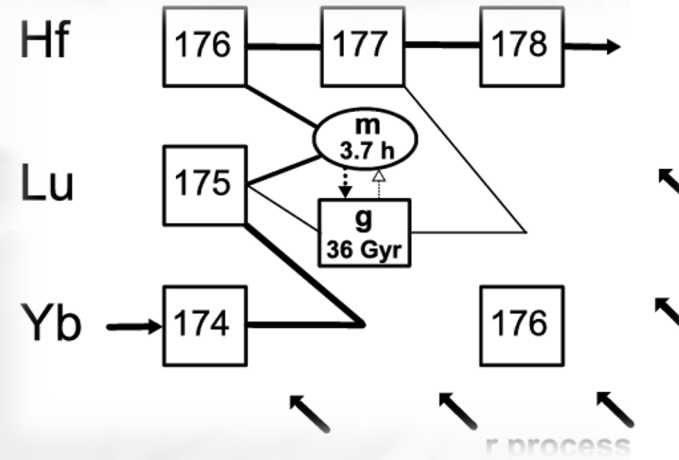


# $^{176}\text{Lu}$ physics case

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$^{176}\text{Lu}$  is a very long-lived in laboratory conditions and **in principle might act as a cosmo-chronometer**

- the s-process branching point at  $^{176}\text{Lu}$  is among the most important ones for the understanding of slow neutron captures in the Asymptotic Giant Branch (AGB) phases of low and intermediate mass stars;
- it determines the abundance of  $^{176}\text{Hf}$ , an “s-only” nucleus
- Scenario is complex due to the presence of an isomeric state placed at 122.45 keV with a very short lifetime



**Important to investigate the *in-plasma* variations of the half-life**

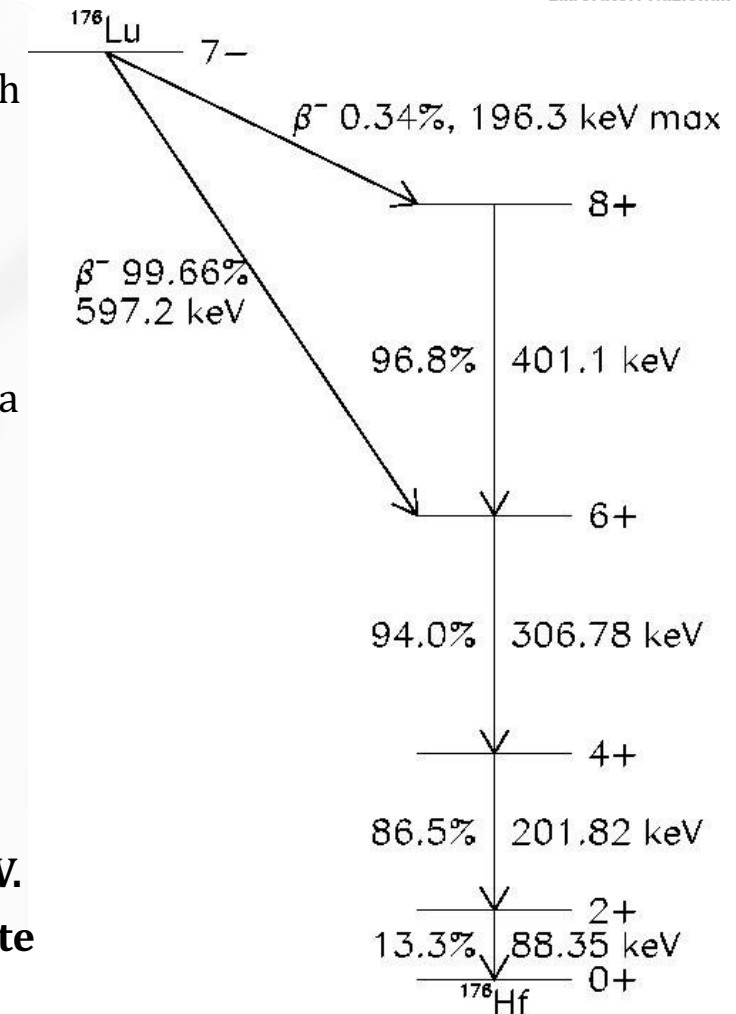
# Measurement strategy

1. Once the solid  $^{176}\text{Lu}$  target is hit by a laser pulse with an intensity as high as  $10^{21} \text{ W/cm}^2$ , the ionization and the subsequent ion emission takes place
2. Lu ions travelling at a velocity of the order of hundreds of keV
3. Given the high energy administered by the laser in a short time interval, a local thermal equilibrium can be reached not only by the electrons, but also by the ion clouds, that can reach temperature as high as  $10^8 \text{ K}$
4. At this temperature, the nuclei may be excited, and the Lu isomeric state  $^{176,m}\text{Lu}$  can be populated
5.  $^{176}\text{Lu}$  decays to the Hf  $6^+$  excited states, whose de-excitation proceeds through three different steps, leading to the subsequent emission of photons with energies equal to  $E_\gamma = 307, 202$  and  $88 \text{ keV}$ .  $^{176,m}\text{Lu}$ , on the other hand, directly decays to the first Hf excited state  $\rightarrow$  only the emission of a photon with  $E_\gamma = 88 \text{ keV}$  is observed



# Measurement strategy

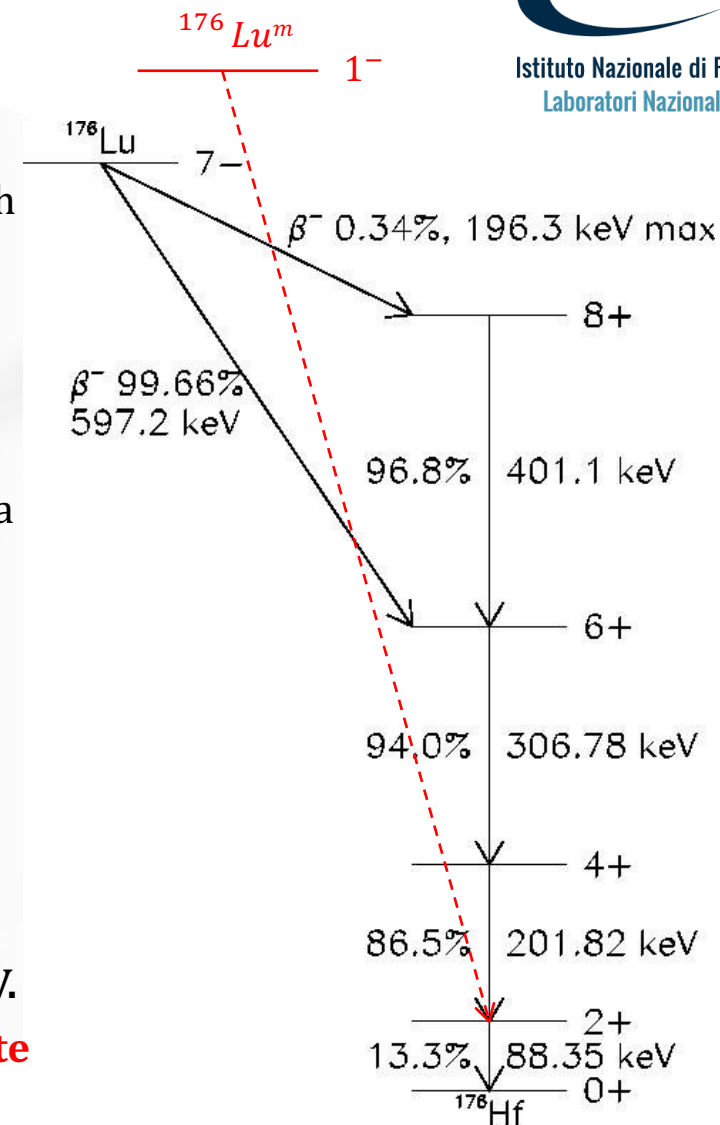
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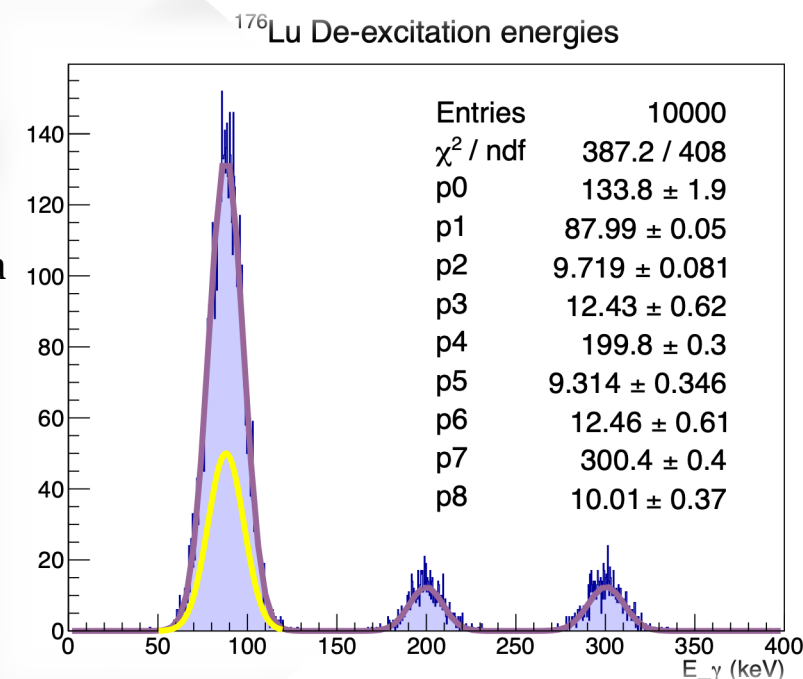
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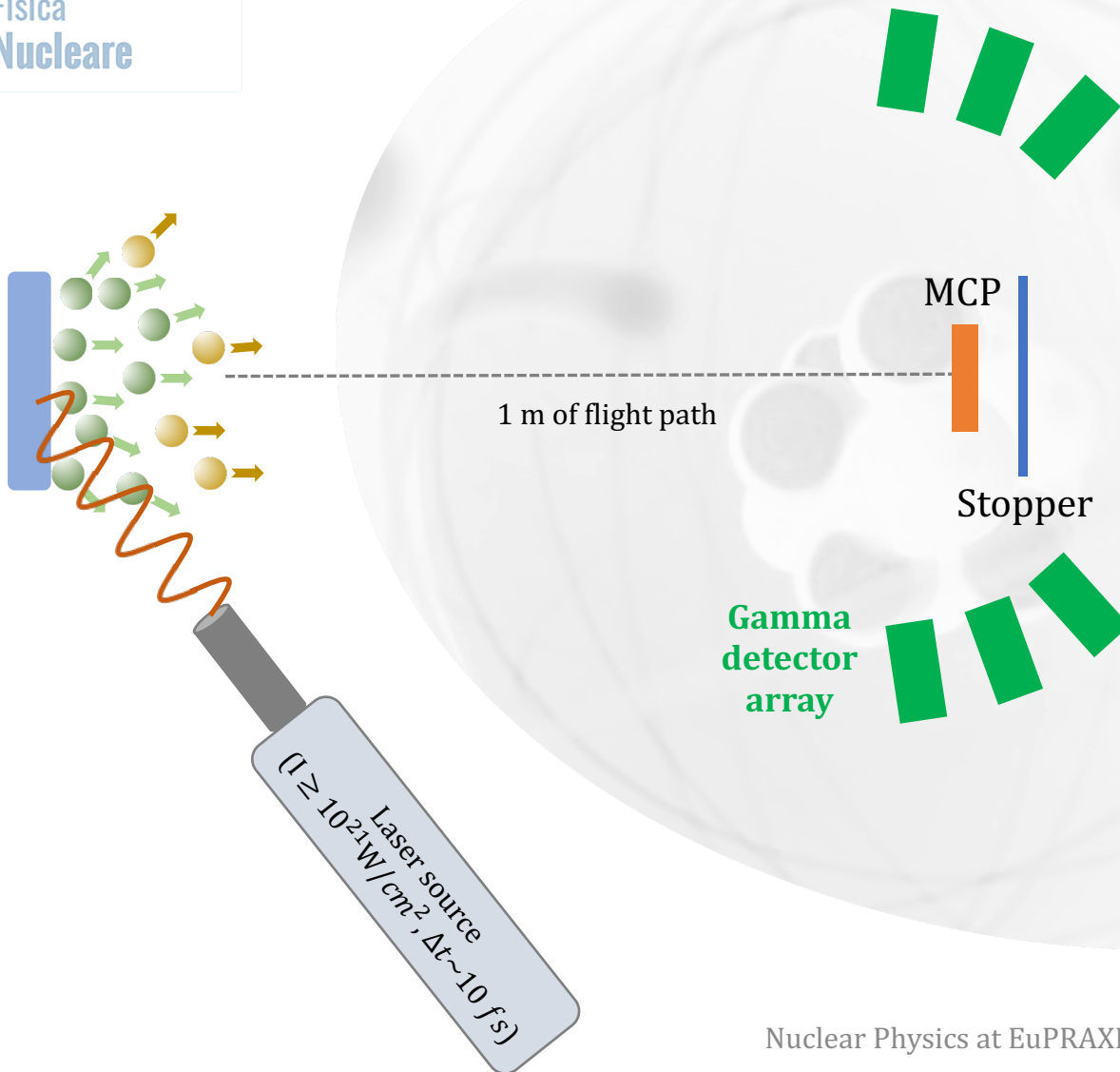
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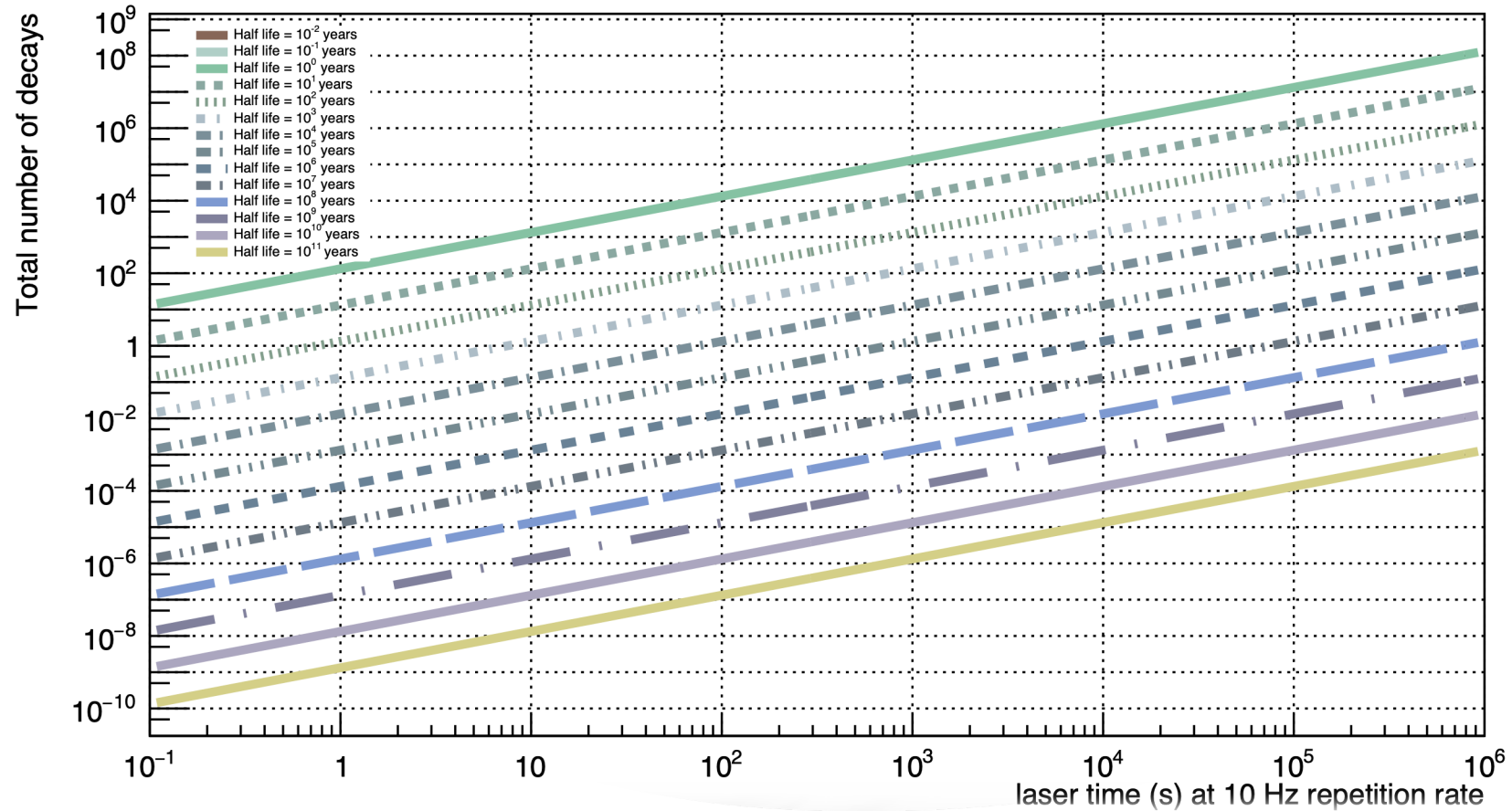
# Possible experimental setup for $\beta$ -decay



1. A PW laser pulse is sent to a solid target containing the radio-isotope under investigation.
2. The plasma is created and a forward emission of the thermalized excited nuclei takes place.
3. The nuclei travel and eventually decay in flight, populating daughter nuclei in excited states.
4. The flight path, and then the distance between the target and a suitable stopper, must be optimized in order to guarantee a proper time window for the decay measurement ( $\sim 1 \mu\text{s}$ ).
5. This poses limits on the half-life range that can be explored.
6. The gamma emitted in the decay process may be detected through a dedicated detection system.

# Projections for a 10 Hz repetition rate

Number of decays as a function of laser time



# Conclusions

1. **Possible physics program with hundreds of TW lasers:** study of fusion processes in plasma, only barely explored at the moment in stellar-like conditions → few seminal measurements available (*e.g.*, *Lattuada et al.*), to be confirmed with higher statistics.
2. **Possible physics program with hundreds of TW lasers:** also other fusion processes can be explored (*e.g.*,  $^{12}\text{C}$  burning, crucial for the field of nuclear astrophysics)
3. **Possible physics program with PW lasers:** first-time *in-plasma* measurement of the  $^{176}\text{Lu}$   $t_{1/2}$  → implications for the understanding of the heavy-element production through *s*-processes
4. Natural evolution of PANDORA@CSN3 physics program
5. Possible dedicated detector R&D program (*e.g.*, for timing)

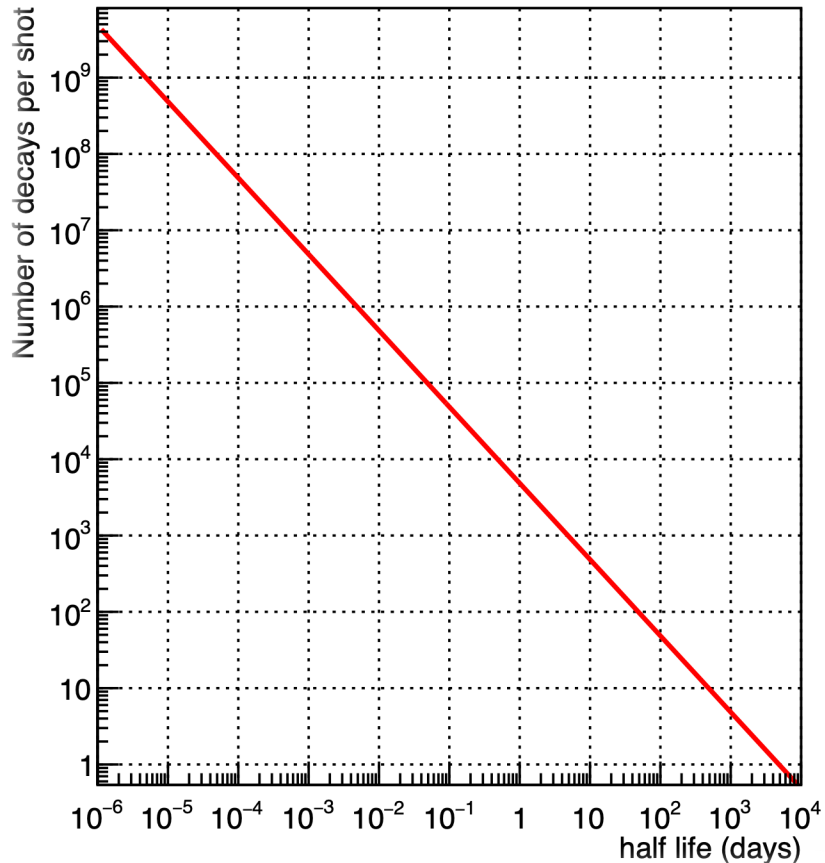




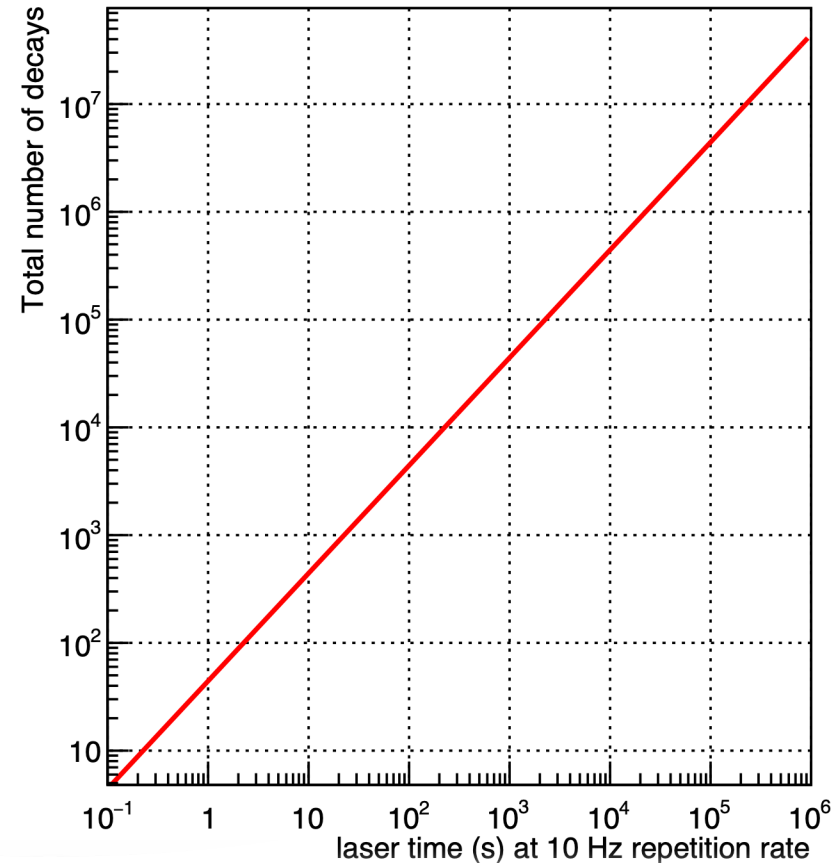
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# Projections for a 10 Hz repetition rate

Number of decays as a function of half lives

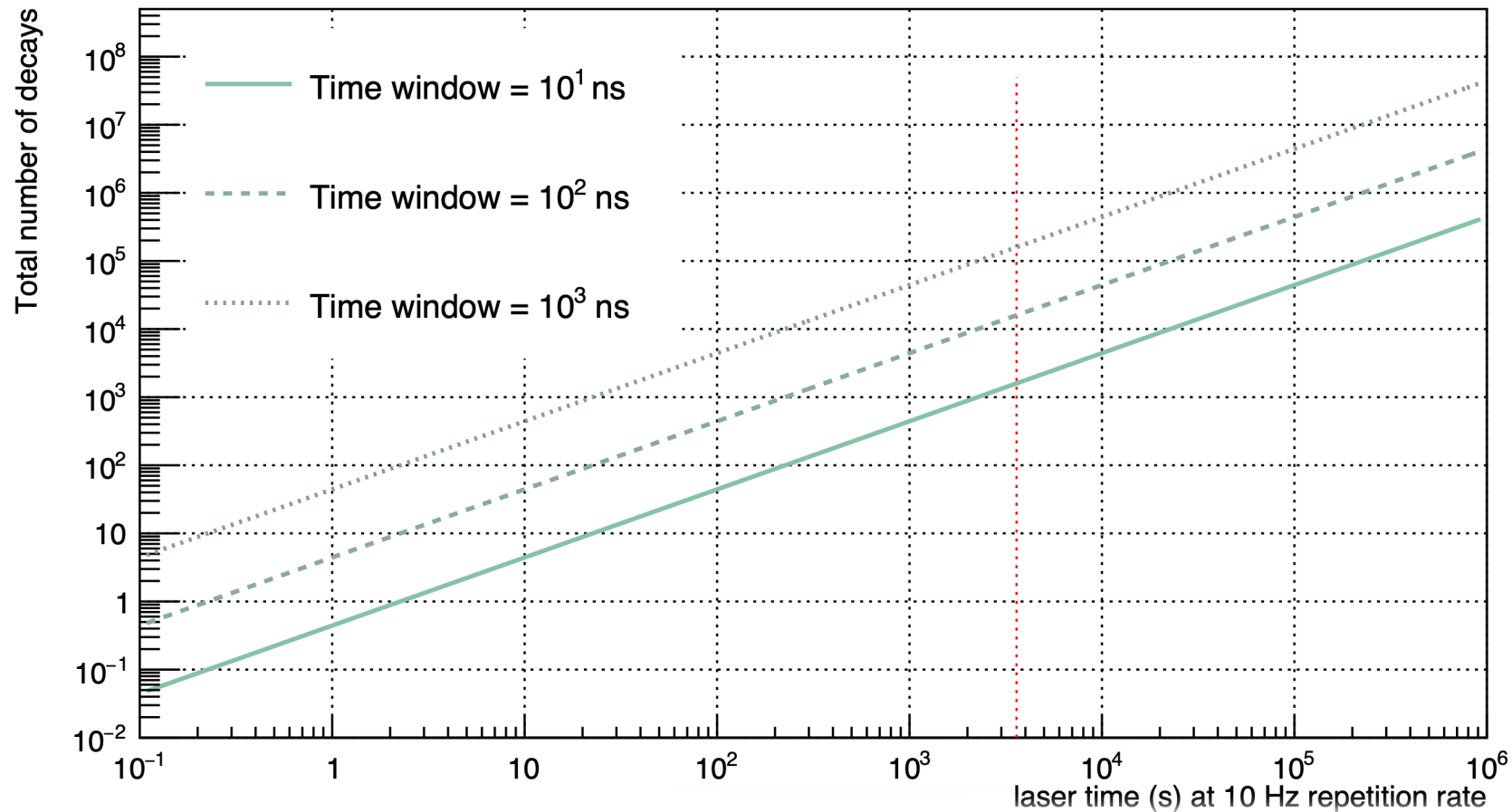


Number of decays as a function of laser time (for  $\tau = 3$  years)



# Projections for a 10 Hz repetition rate

Number of decays as a function of laser time





# How to measure $^{176}\text{Lu}$ $t_{1/2}$ in plasma?

## The PANDORA experiment

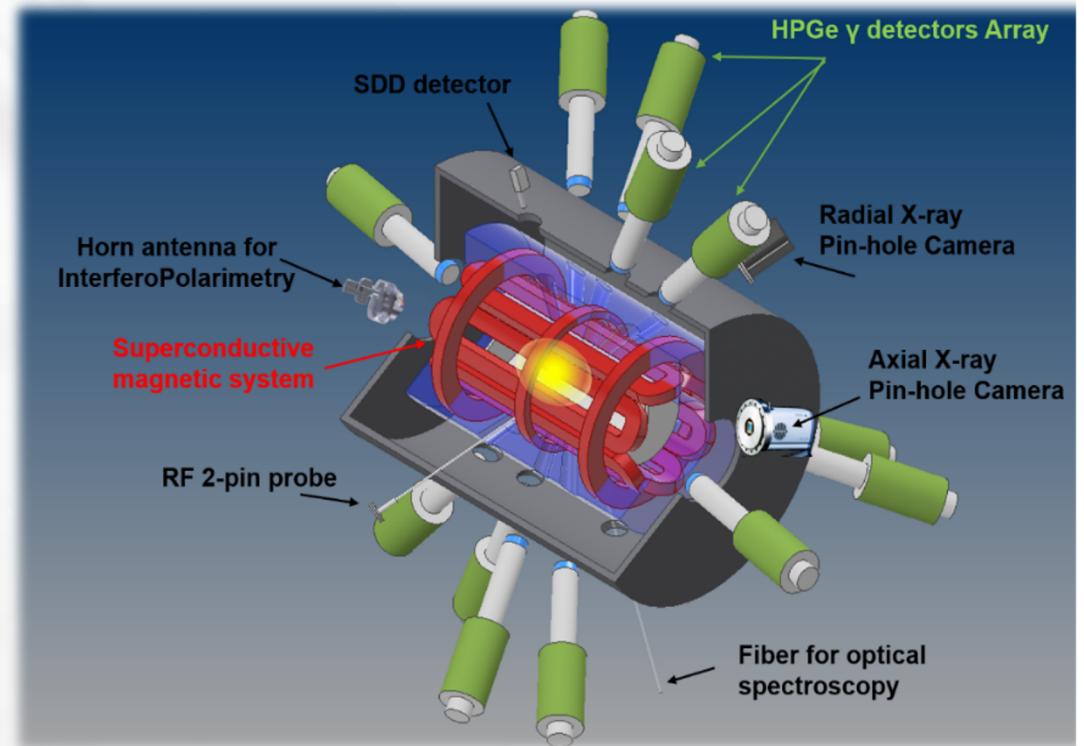
Build a plasma trap where ion species are confined in a magnetic field and a plasma is created with:

- **Electron density:**  $10^{12} \div 10^{14} \text{ cm}^{-3}$
- **Electron temperature:**  $0.1 \div 100 \text{ keV}$
- **Ion density:**  $10^{11} \text{ cm}^{-3}$  → relies on the radioactive isotope concentration in plasma
- **Ion temperature:**  $\sim 1 \text{ e}$  → Ions are cold: no access to the excited states

$$\frac{dN}{dt} = \lambda n_i V \rightarrow \int_0^{T_{meas}} dN = \int_0^{T_{meas}} \lambda n_i V dN$$

$$N(T_{meas}) = \lambda n_i V T_{meas}$$

$n_i V$ : density and plasma volume, constant → to be measured using multiple diagnostic tools



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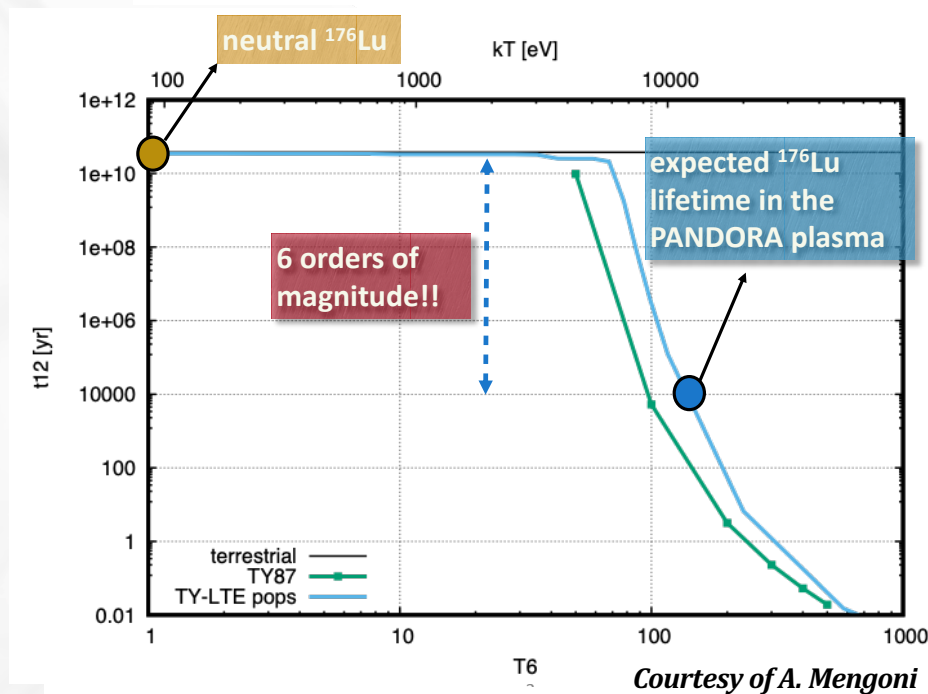
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### $^{176}\text{Lu}$ : lifetime vs. T - theoretical predictions



Takahashi et al. 1987, Phys Rev C 36, 1522

PANDORA: only ground state will be studied → T too low to investigate variation on the isomeric state



# How to measure $^{176}\text{Lu}$ $t_{1/2}$ in plasma?

## Scaling results to stellar environment

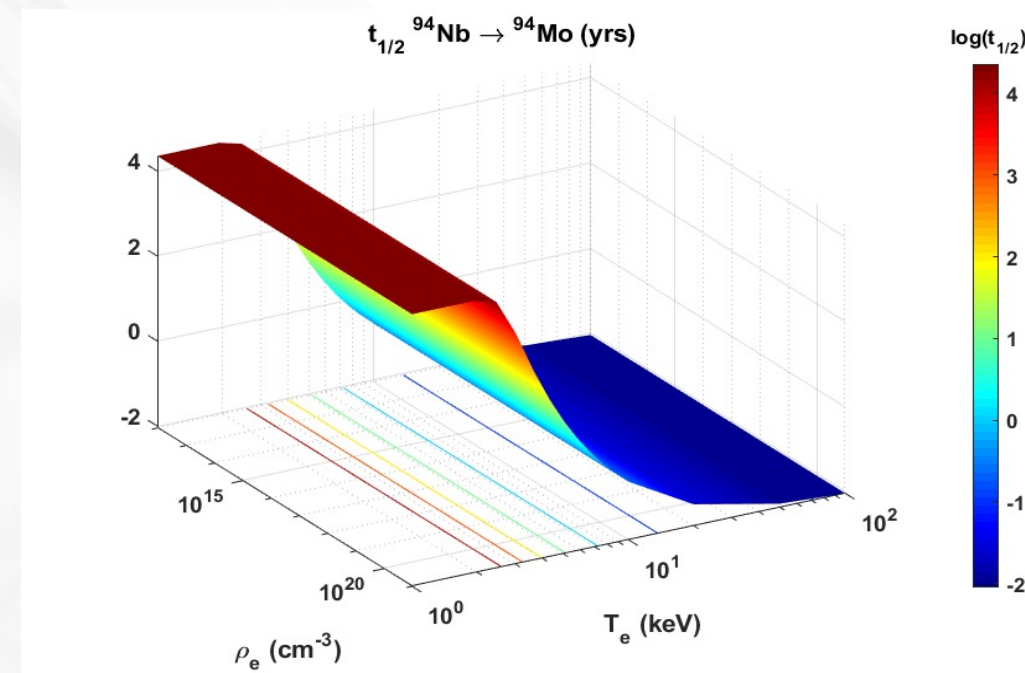
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$T_e = 0.1\text{-}100 \text{ keV}$  in a lab. Magnetoplasma

Variation with  $T_e$  stronger than with  $\rho_e$  → “stellar effect” can be modelled by ECR (*Electron Cyclotron Resonance*) plasma

# Why to use laser-induced plasma

Build a plasma trap where ion species are confined in a magnetic field and a plasma is created with:

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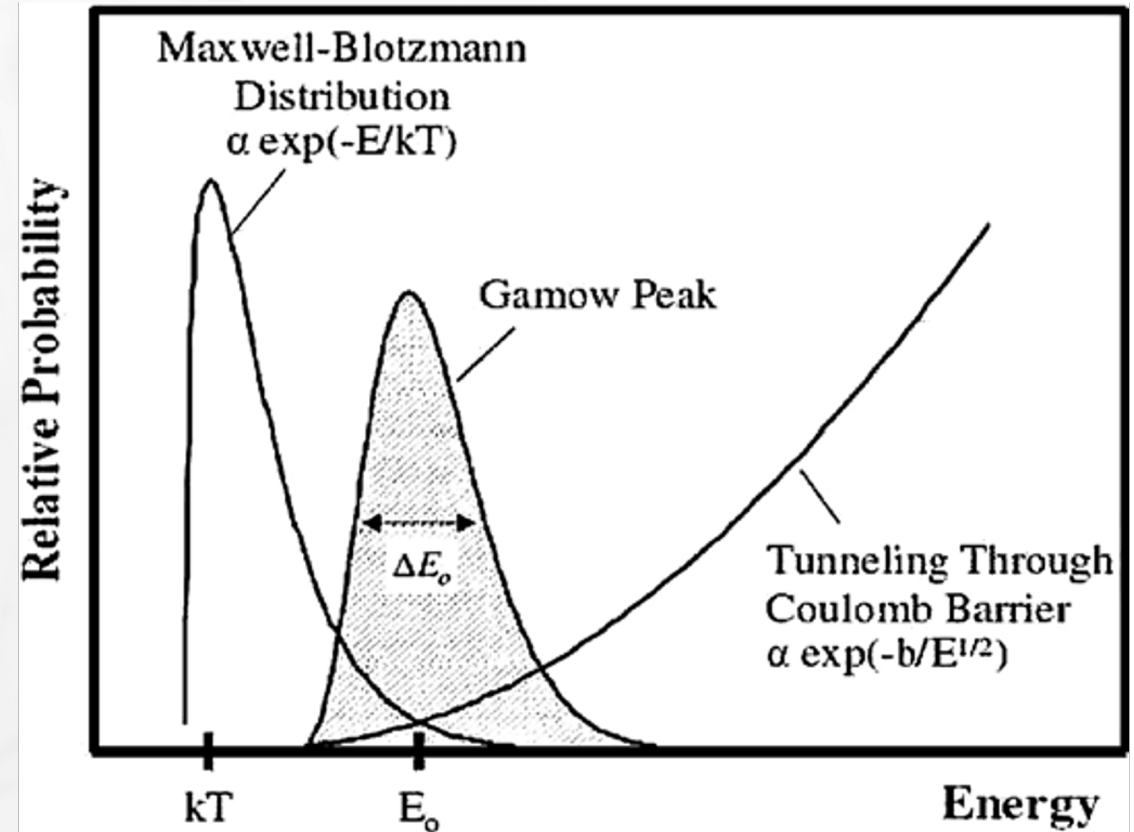
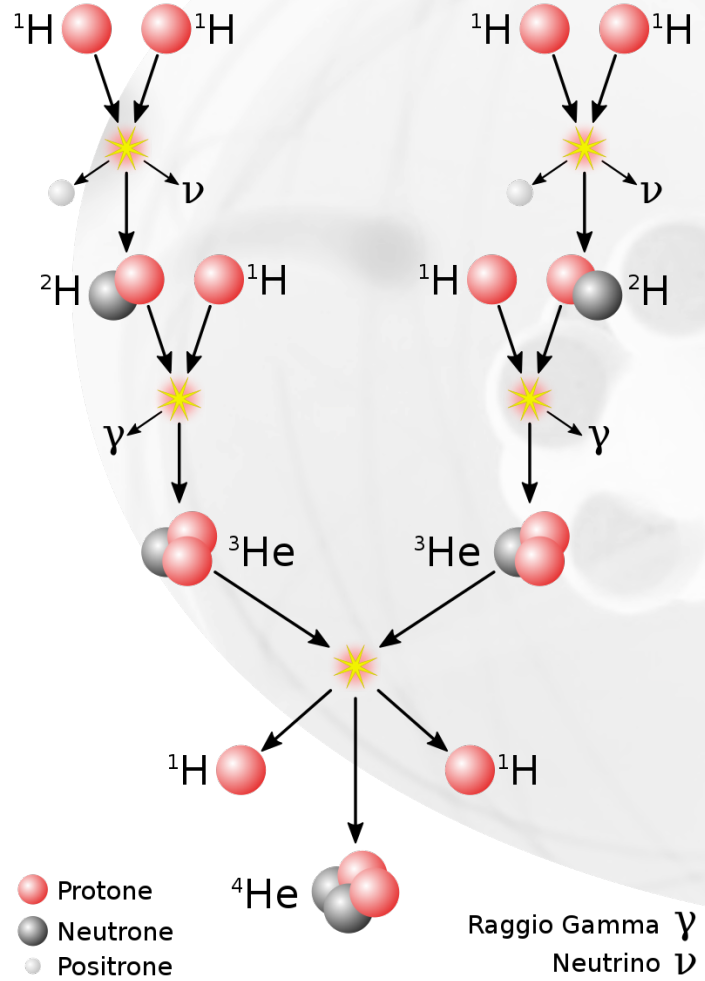
$$N(T_{meas}) = \lambda n_i V T_{meas}$$

## Exploring the onset of a (Full) Local Thermal Equilibrium:

- Typical lifetime of nuclear excited states  $\sim 10^{-15} \text{ s}$
- Assuming an excited state for, e.g.,  $^{176}\text{Lu}^*$ , around 122.45 keV
- Considering  $n_e = n_i = 10^{27} \text{ m}^{-3}$  (a typical stars interior density), at  $T_e = T_i = 6.68 \text{ keV}$ , the excited level lifetime is already exactly the same of the excitation rate, meaning **that this level can be populated and it is in thermal equilibrium in the assumed laser-induced plasma lifetime** (order of ps or tens of ps)
- Calculation also rescaled to a more realistic expected density of a real laser-induced plasma scenario ( $n_e = n_i = 10^{25} \text{ m}^{-3}$ ) → the required plasma temperature to get the thermal equilibrium goes to around 37.5 keV. **This value seems to be however absolutely achievable in the foreseen laboratory scenario, confirming that the decay from excited states is in principle feasible.**

*Simulations by B. Mishra et al.: thanks, Bharat!*

# Fusion processes: light elements



# s-process endpoint

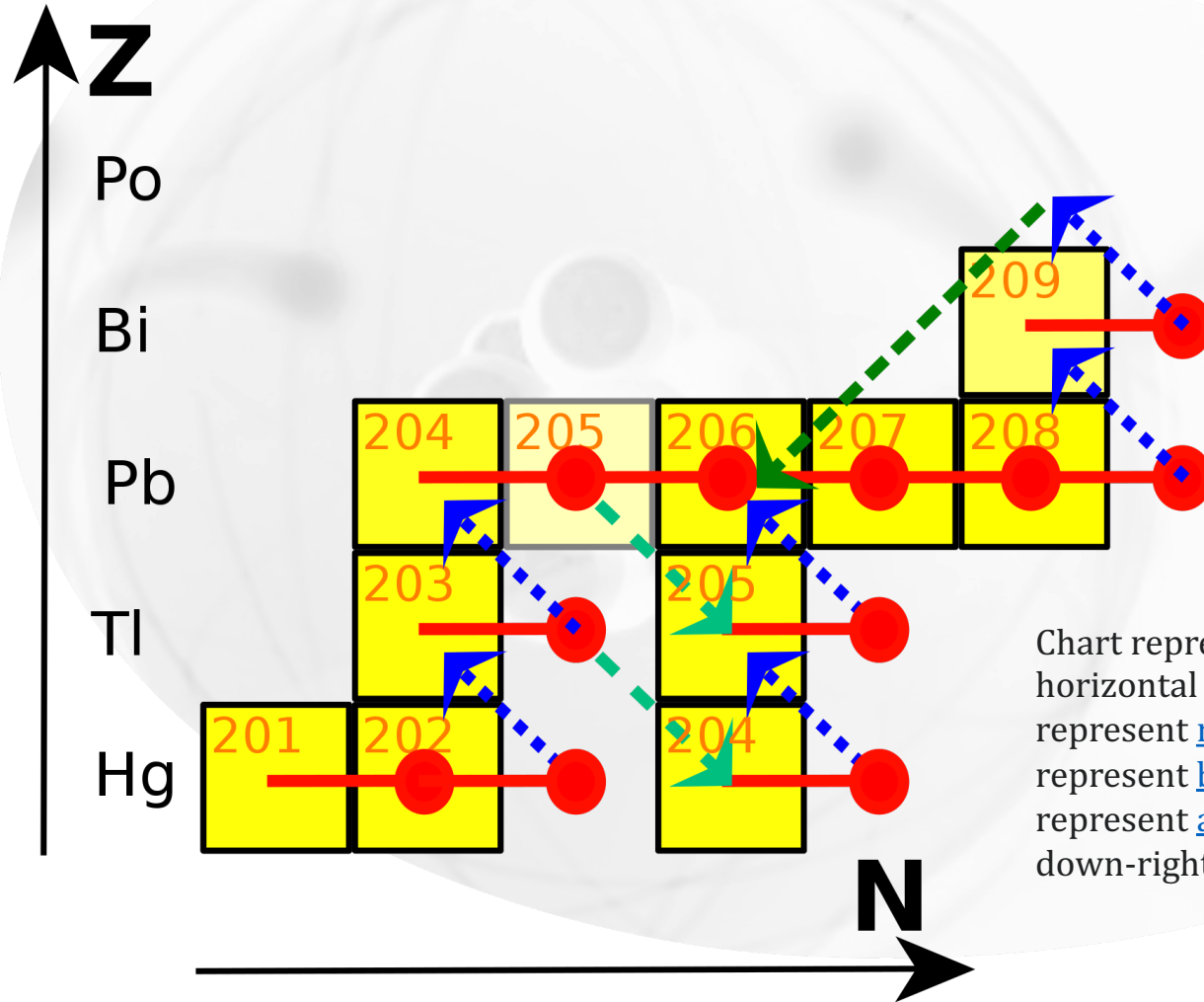
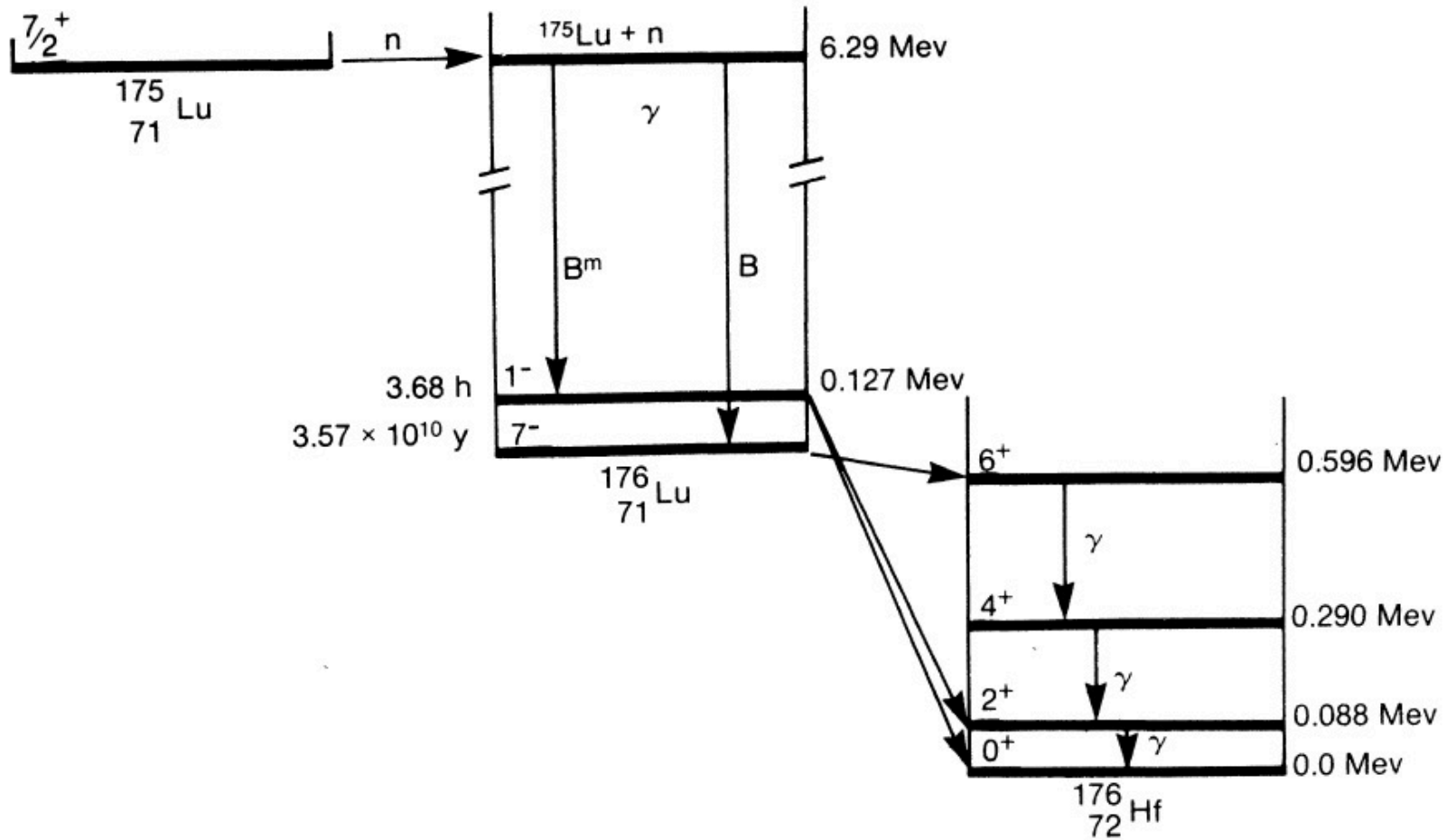


Chart representing the final part of the *s*-process. Red horizontal lines with a circle in their right ends represent [neutron captures](#); blue arrows pointing up-left represent [beta decays](#); green arrows pointing down-left represent [alpha decays](#); cyan/light-green arrows pointing down-right represent [electron captures](#).

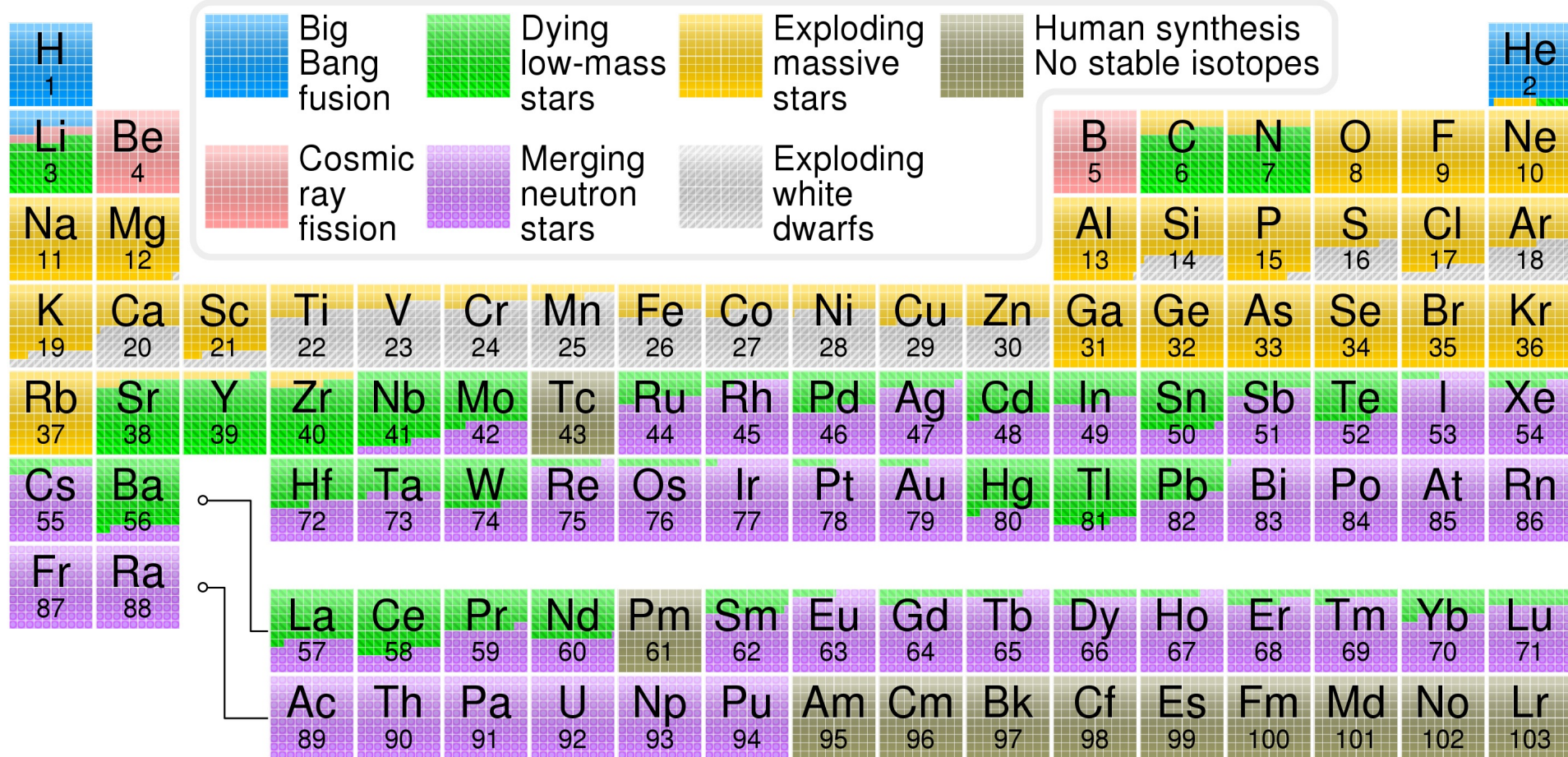
# Decay scheme for lutetium



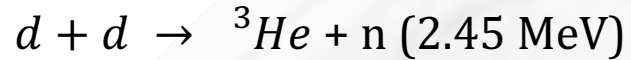




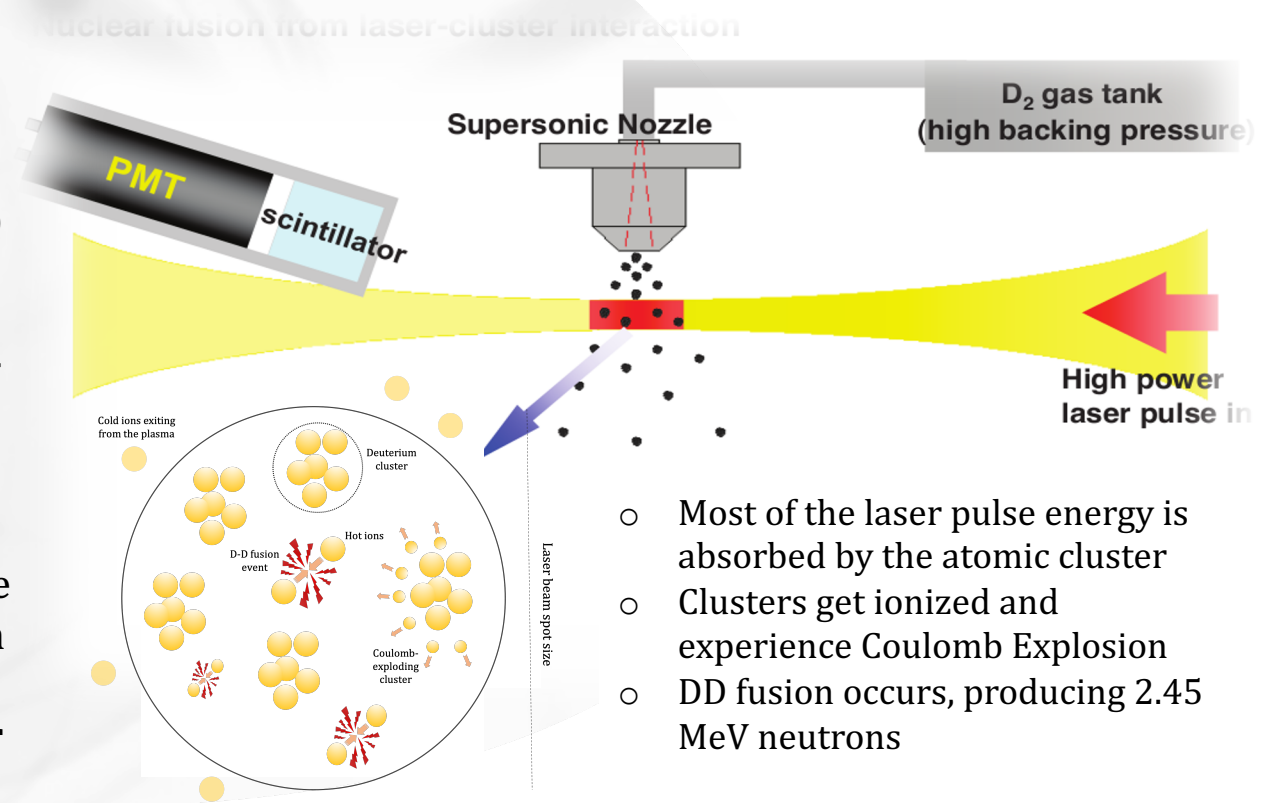
# Cosmogenic origin of elements



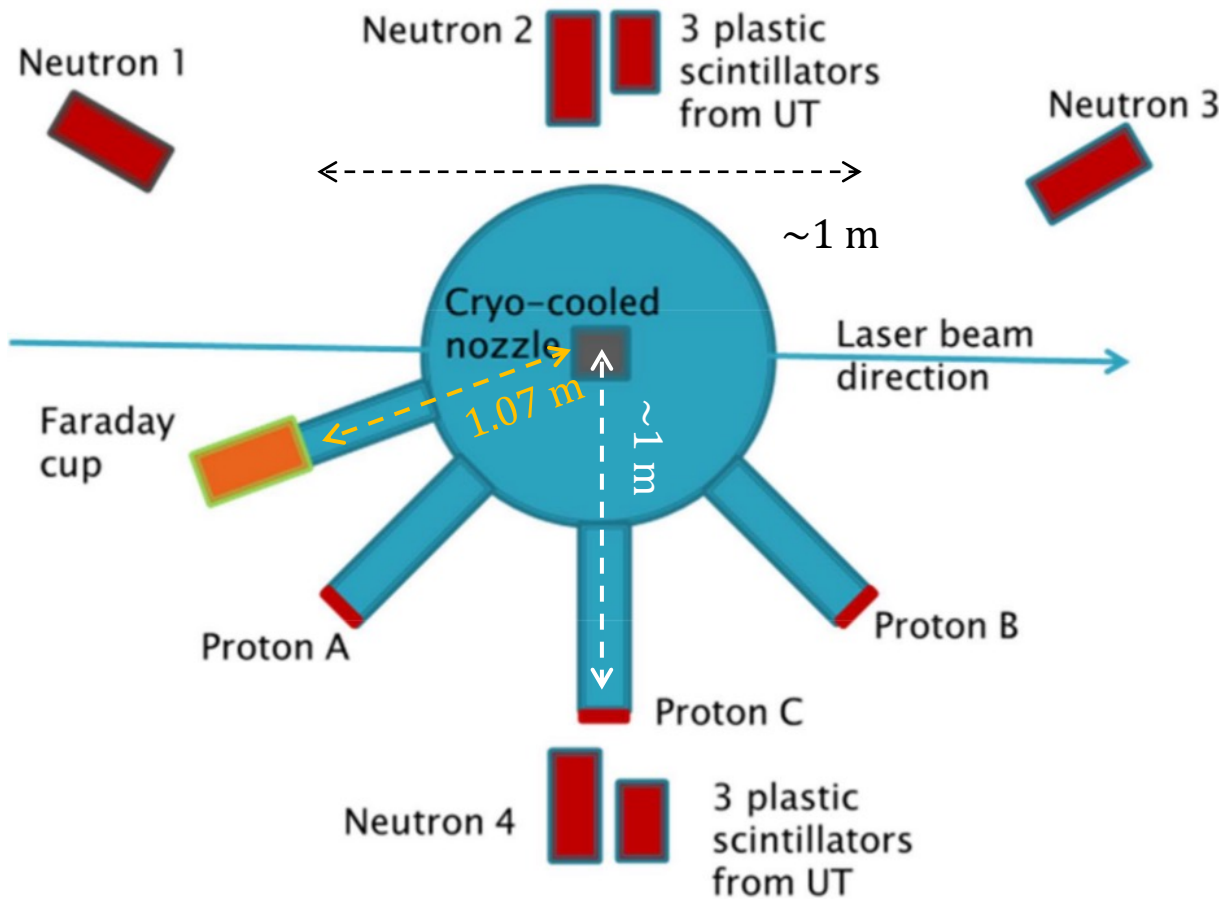
# Deuterium fusion process in plasma



1. The deuterium gas is kept at a low temperature, close to the critical temperature **where gas and liquid phase coexist**.
2. The adiabatic expansion through a supersonic nozzle in the reaction chamber induces the clusterization of the D molecules, which are then irradiated by a laser pulse.
3. Most of the pulse energy is absorbed by the clusters, causing the escape of the electrons and the formation of a plasma.
4. The high level of electrostatic fields reached in it produces the so-called Coulomb Explosion  $\rightarrow$  **emission of hot deuterium ions** (with kinetic energy in the range tens-hundreds keV) that can fuse with ions coming from the explosion of other clusters.
5. **High laser repetition rate and coarse granularity for the PID arrays** to identify the fusion reaction products



# Experimental area: an example



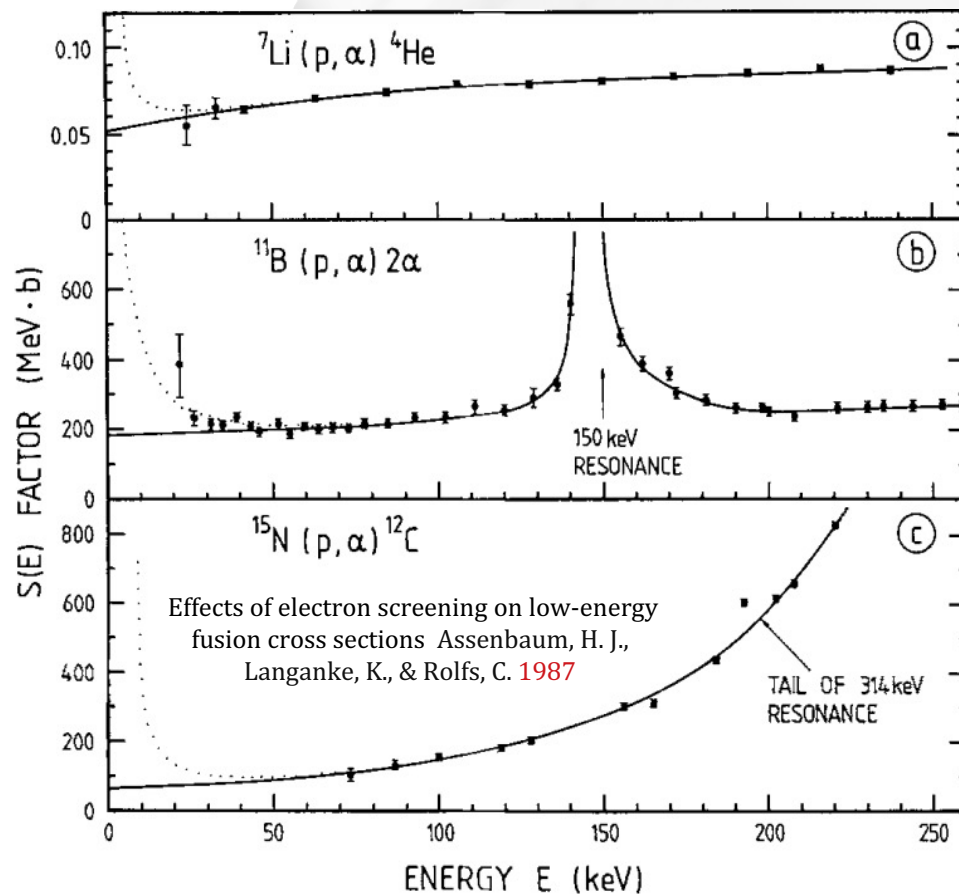
- Proper *time-of-flight paths* must be foreseen for a reliable particle identification
- Neutron detectors must be kept displaced enough from the walls
- Cablings, signal transportation



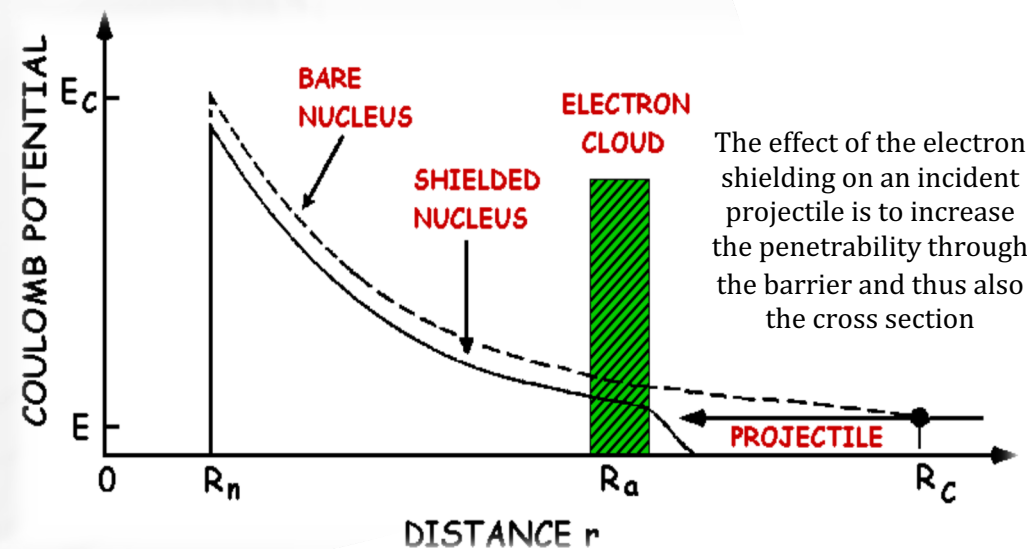


# Why plasma: fusion processes

ELECTRON SCREENING AND THERMONUCLEAR REACTIONS E. E. SALPETER 1954

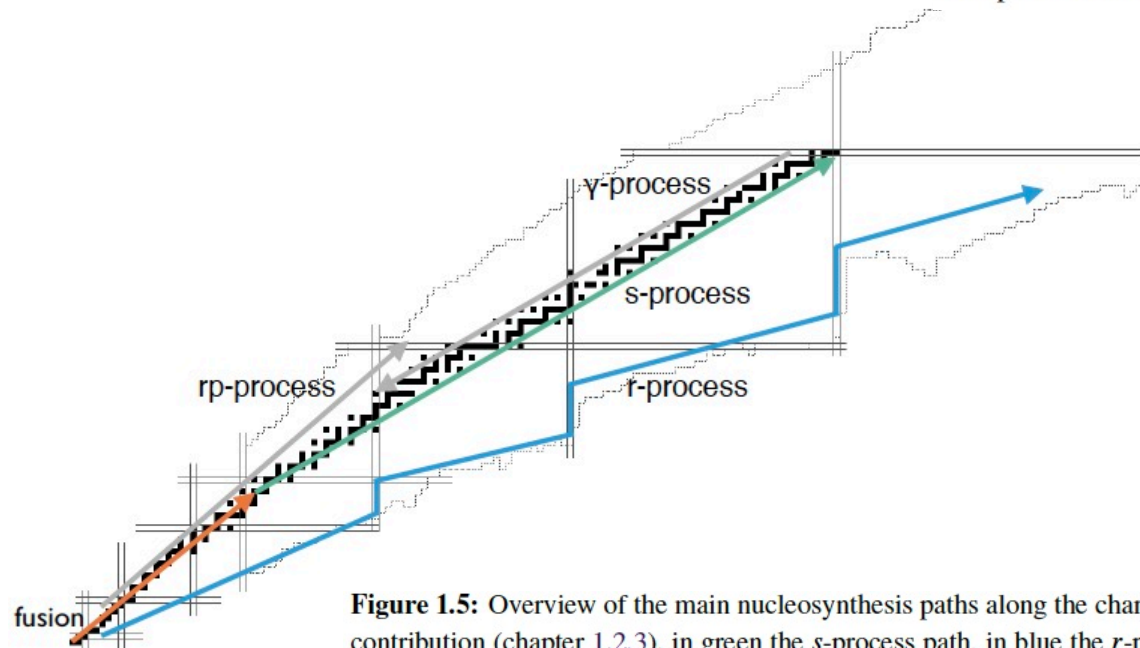


Relatively small enhancements due to electron screening could cause significant errors in the extrapolation to lower energies, if the cross-section curve is forced to follow the trend of the enhanced cross sections without correcting for screening.



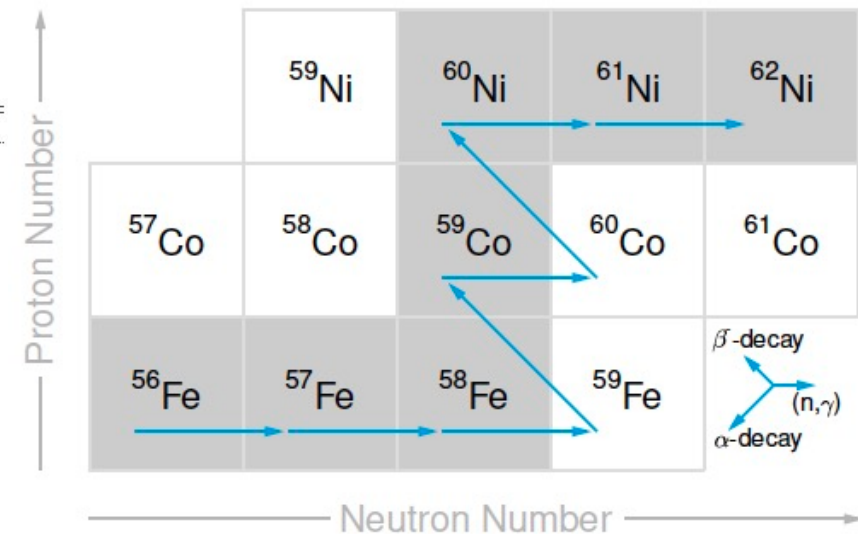


# Main nucleosynthesis path



**Figure 1.5:** Overview of the main nucleosynthesis paths along the chart of nuclides. In orange the fusion contribution (chapter 1.2.3), in green the *s*-process path, in blue the *r*-process path (page 24) and in grey the *rp*- and  $\gamma$  processes (page 24). The *s* process and the  $\gamma$  process are secondary processes, which depend on certain seed isotopes, whereas fusion, the *r* process and the *rp* process are primary processes, which are not dependent on prior nucleosynthesis. Adapted from [Glorius 2013](#).

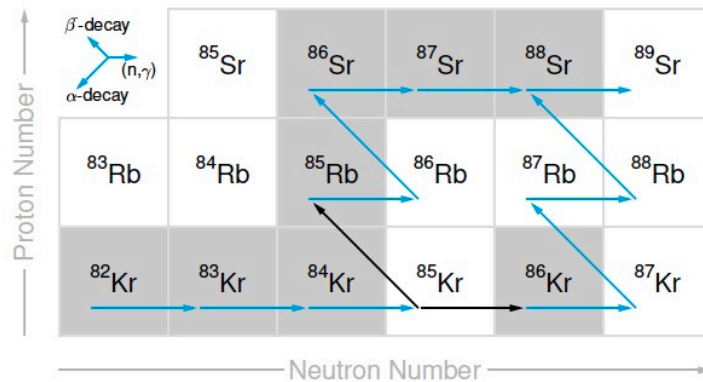
**Figure 1.6:** The *s*-process path. If a beta unstable isotope is reached, it decays back to the next stable isotope. Stable isotopes are marked grey, and unstable isotopes white.



# Branching points and s-only isotopes

## BRANCHING POINTS

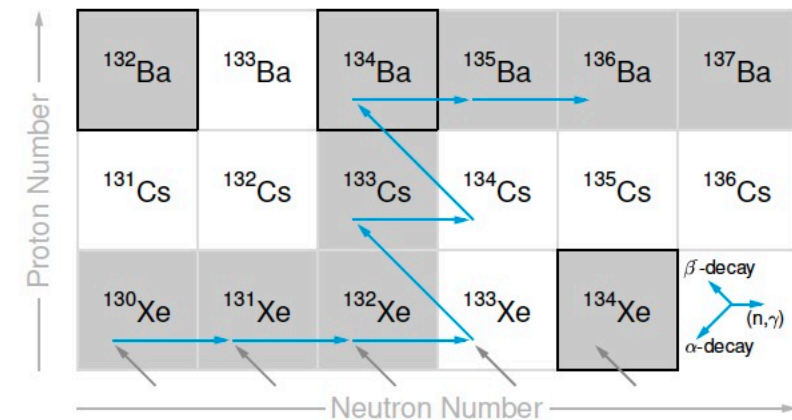
Branching points are isotopes or isomers on the  $s$ -process path, which undergo a  $\beta$ -decay on the same timescale as a neutron capture. This leads to a branching in the  $s$ -process path where some of the mass flow follows the  $\beta$ -decay branch and the rest of the mass flow the neutron capture path (figure 1.10).



**Figure 1.10:** Branching point  $^{85}\text{Kr}$  along the  $s$ -process path.  $^{85}\text{Kr}$  has a half-life of about 10 years, which would make it an excellent probe for  $s$ -process conditions, if all reaction channels would be known to good precision. Stable isotopes are marked grey, and unstable isotopes white.

## THE $s$ -ONLY ISOTOPES

The  $s$ -only isotopes are isotopes, which are created almost solely by the  $s$  process and are shielded from other processes. These isotopes are often considered when comparing observations to simulations (figure 1.11).





# $^{176}\text{Lu}$ physics case

## The Decay Scheme of Natural Lutetium 176

JAMES R. ARNOLD AND THOMAS SUGIHARA\*

*Institute for Nuclear Studies, University of Chicago, Chicago, Illinois*

(Received February 26, 1953)

THE nuclide lutetium 176 is of particular interest for two related reasons; first, that it is the central member of one of the four known triads of naturally occurring adjacent isobars, and second, that its spin of at least 7 units<sup>1</sup> is the highest known. Flammersfeld has reported a decay scheme for this nuclide,<sup>2</sup> in which both  $K$  capture and  $\beta$ -decay appear, the ratio of the branches being  $K/\beta^- = 2$ . The observed gamma-ray was placed in the  $K$  branch, and its energy fixed at 0.260 Mev.

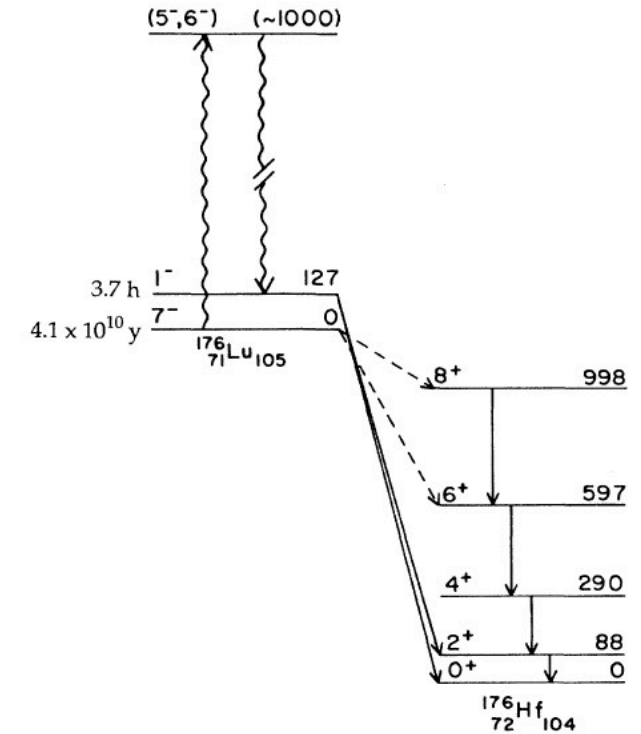
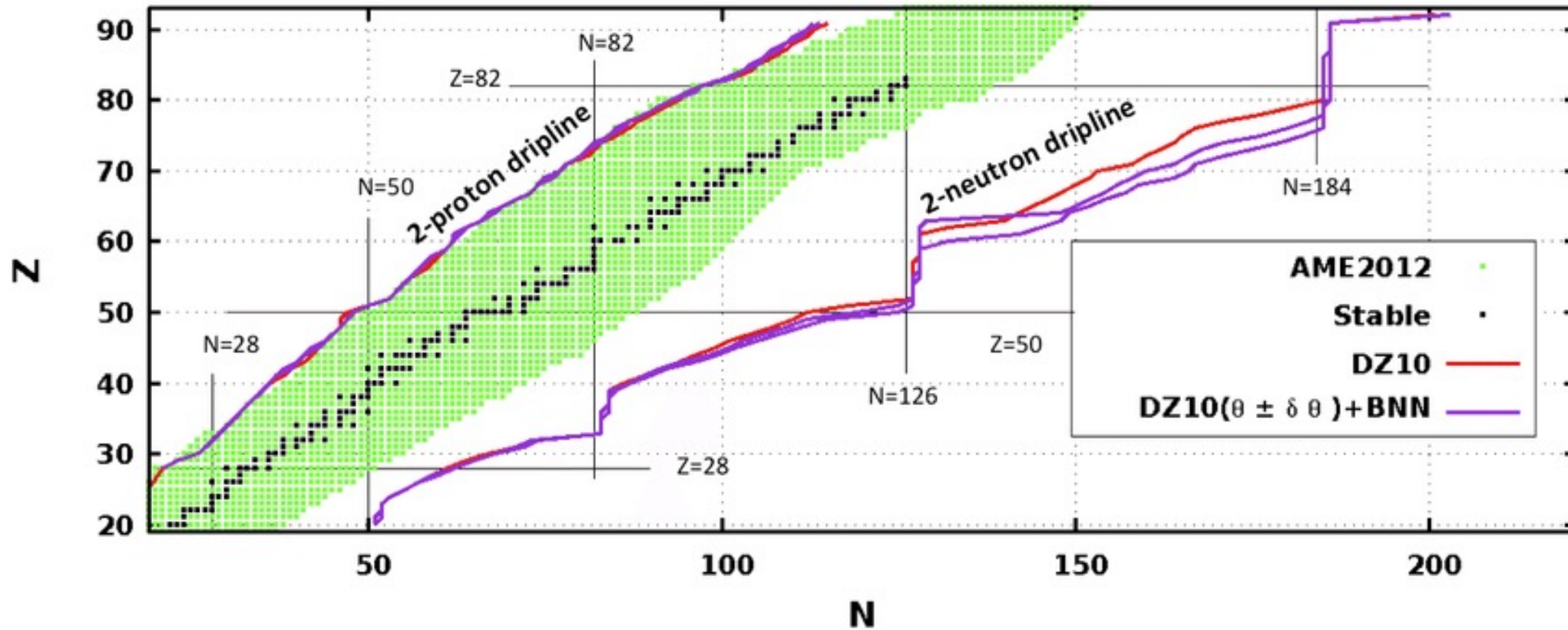


FIG. 3. A partial level scheme of  $^{176}\text{Lu}$ , showing the positions and decays of the ground state and isomer at 122.9 keV. The equilibration of these two levels could be achieved by way of a level of intermediate spin, as illustrated in the figure.



# Drip lines



# $^{176}\text{Lu}$ physics case

PHYSICAL REVIEW C

VOLUME 44, NUMBER 6

DECEMBER 1991

## $^{176}\text{Lu}$ : An unreliable *s*-process chronometer

K. T. Lesko, E. B. Norman, R-M. Larimer, and B. Sur

*Nuclear Science Division, Lawrence Berkeley Laboratory, 1 Cyclotron Road, Berkeley, California 94720  
and Center for Particle Astrophysics, University of California, Berkeley, California 94720*

C. B. Beausang\*

*Nuclear Science Division, Lawrence Berkeley Laboratory, 1 Cyclotron Road, Berkeley, California 94720*

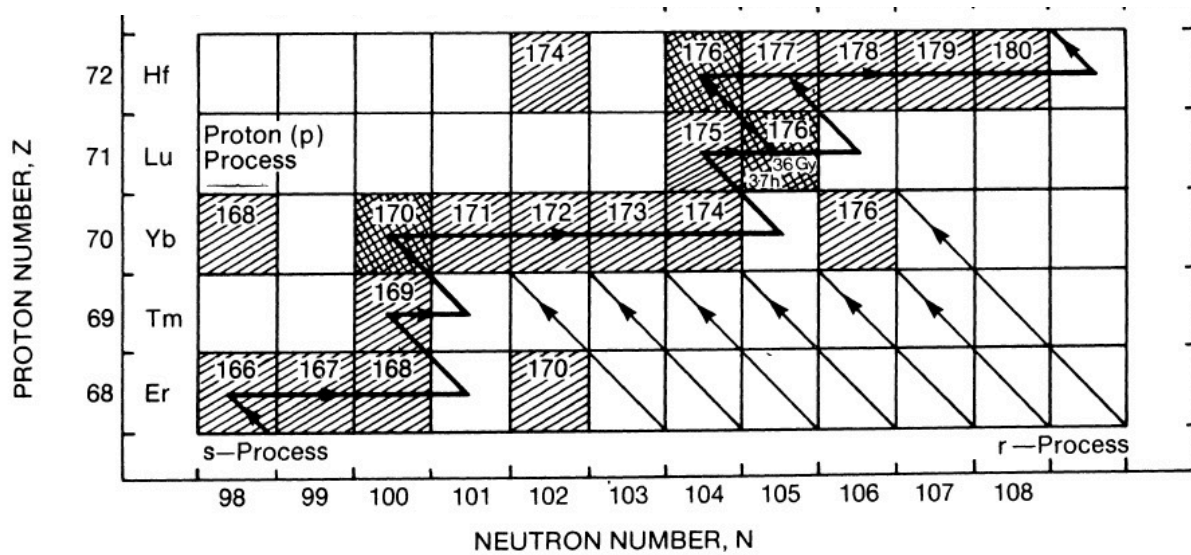
(Received 17 October 1990)

A level scheme of  $^{176}\text{Lu}$  up to  $\sim 1400$  keV excitation energy is deduced from a  $\gamma$ - $\gamma$  coincidence experiment and previously published particle transfer data. 170  $\gamma$ -ray transitions are placed between 85 levels. We identify 27 previously unknown levels and 131 previously unknown transitions in  $^{176}\text{Lu}$ . With this  $\gamma$ -ray data we place the energy of the isomer at 122.9 keV. A level at 838.5 keV ( $J^\pi = 5^-$ ,  $t_{1/2} < 10$  ns) is found to decay with substantial strength to both the ground state ( $7^-$ ,  $4.08 \times 10^{10}$  yr) and the 122.9 keV isomer ( $1^-$ , 3.7 hr). The presence of this level guarantees the thermal equilibrium of  $^{176}\text{Lu}^{g,m}$  for  $T \geq 3 \times 10^8$  K and therefore during *s*-process nucleosynthesis. The resulting temperature sensitivity of its effective half-life rules out the use of  $^{176}\text{Lu}$  as an *s*-process chronometer. The use of  $^{176}\text{Lu}$  to determine *s*-process temperatures is discussed.

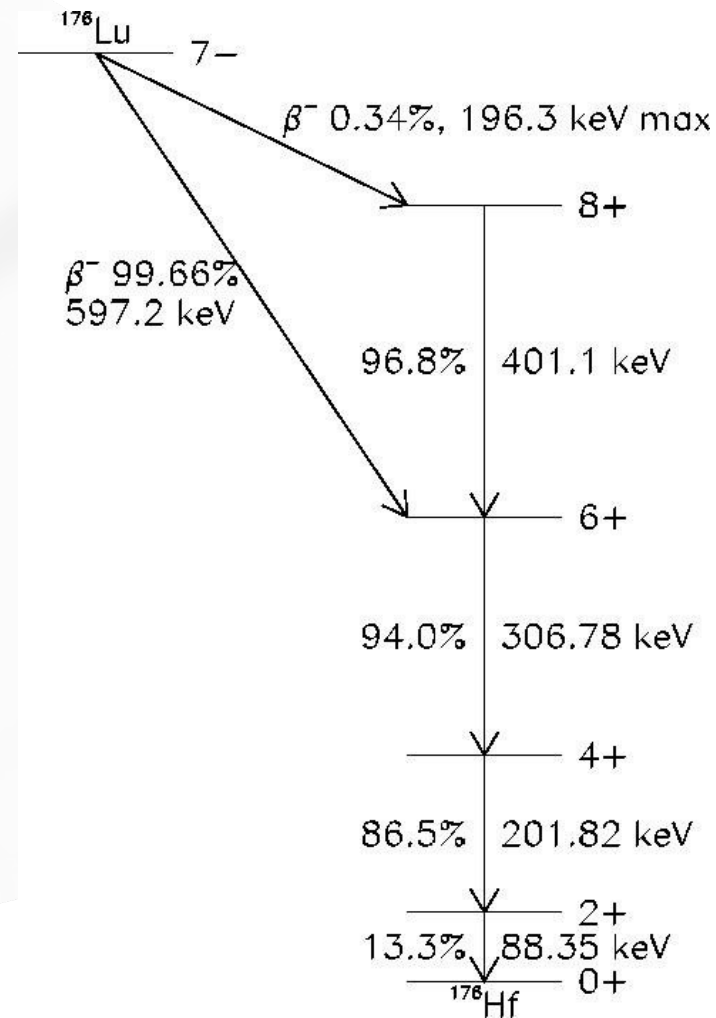


# $^{176}\text{Lu}$ physics case

$^{176}\text{Lu}$  is one of the few naturally occurring radio nuclides that have survived from the era of nucleosynthesis. Its present isotopic abundance [1] is 2.6% and its half-life is  $4.08 \times 10^{10}$  yr [2].

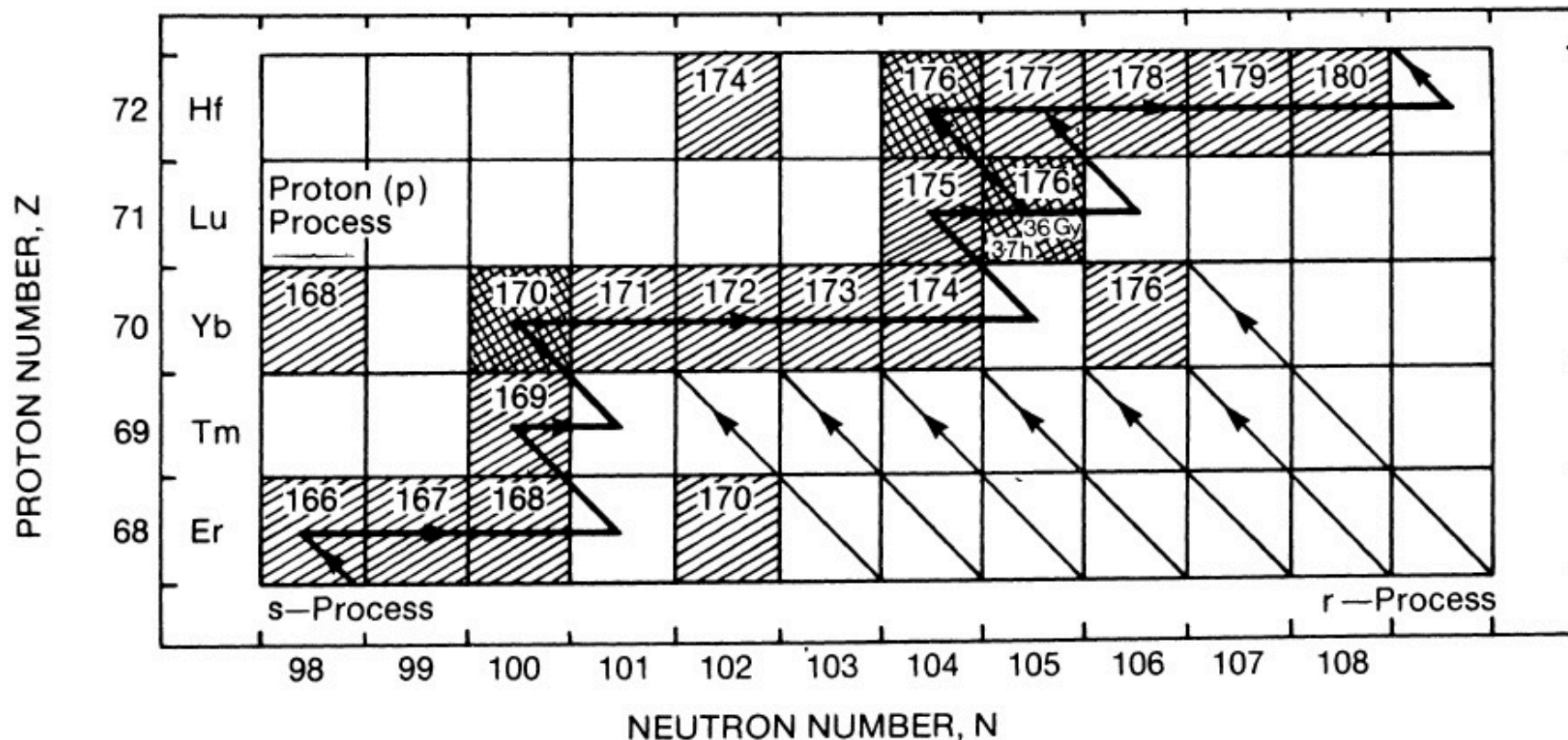


**Figure 1.** s-process path in the rare earth element mass region. s-only process nuclides  $^{170}\text{Yb}$ ,  $^{176}\text{Lu}$  and  $^{176}\text{Hf}$  are shielded from r-process contributions by  $^{170}\text{Er}$  and  $^{176}\text{Yb}$  respectively. The s-process branches at  $^{176}\text{Lu}$  if a significant population of the 3.68 h isomeric state occurs.





# $^{176}\text{Lu}$ branch in the s-process

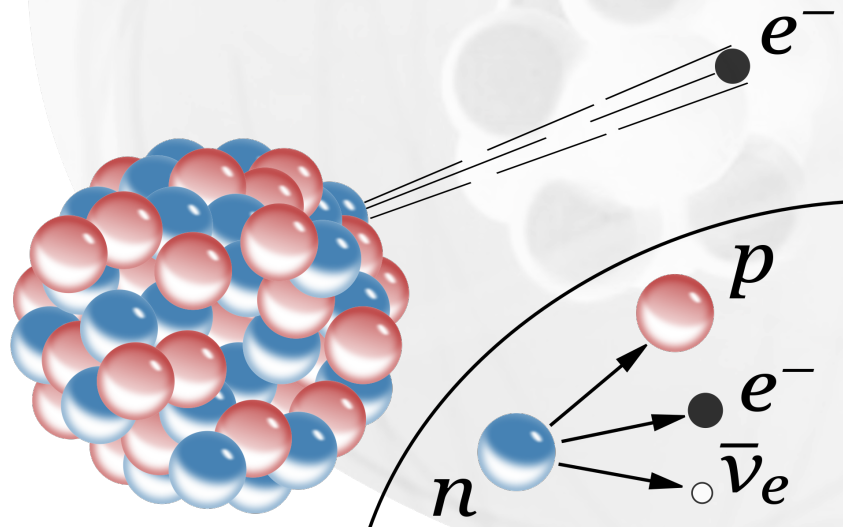


**Figure 1.** s-process path in the rare earth element mass region. s-only process nuclides  $^{170}\text{Yb}$ ,  $^{176}\text{Lu}$  and  $^{176}\text{Hf}$  are shielded from r-process contributions by  $^{170}\text{Er}$  and  $^{176}\text{Yb}$  respectively. The s-process branches at  $^{176}\text{Lu}$  if a significant population of the 3.68 h isomeric state occurs.



# Why plasma: $\beta$ -decays

**$\beta$ -decay in stellar environment:** in a stellar plasma, ions are embedded in a cloud of charges, both positive and negative. These charges create EM fields which act as perturbation to the atomic/ionic levels leading to corrections of Q values which affects the decay rates.



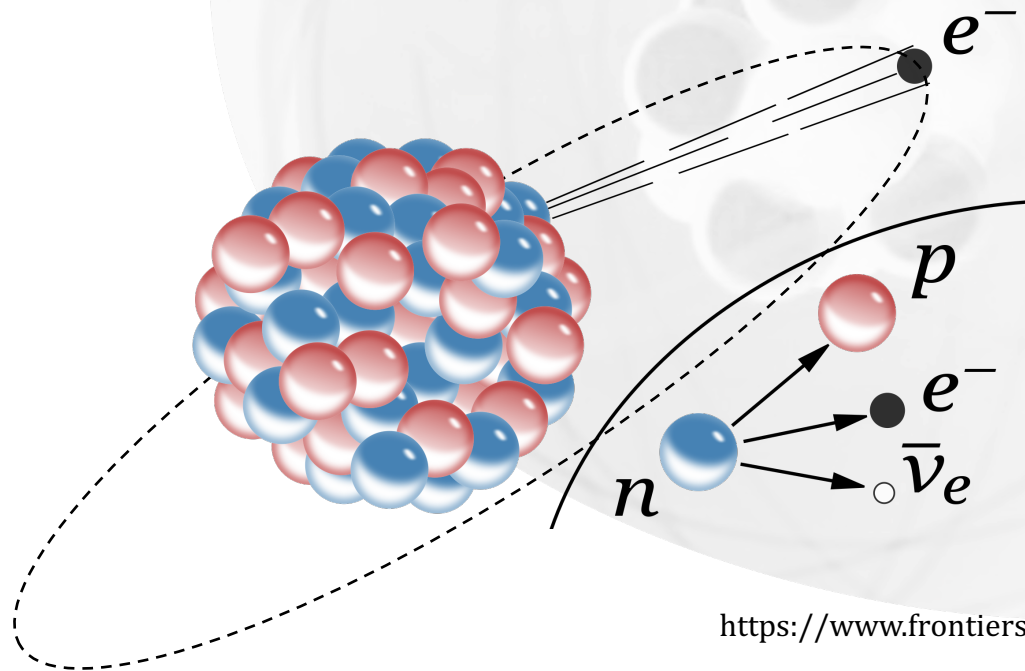
**Bound-state  $\beta$ -decay** is a nuclear  $\beta$ -decay process in which an electron is created in a previously unoccupied atomic orbital rather than in the continuum.

For fully ionized atoms (bare nuclei), it is possible for electrons to fail to escape the atom, and to be emitted from the nucleus into low-lying atomic bound states (orbitals). This cannot occur for neutral atoms with low-lying bound states which are already filled by electrons.

<https://www.frontiersin.org/research-topics/25146/nuclear-physics-and-astrophysics-in-plasma-traps>

# Why plasma: $\beta$ -decays

**$\beta$ -decay in stellar environment:** in a stellar plasma, ions are embedded in a cloud of charges, both positive and negative. These charges create EM fields which act as perturbation to the atomic/ionic levels leading to corrections of Q values which affects the decay rates.



**Bound-state  $\beta$ -decay** is a nuclear  $\beta$ -decay process in which an electron is created in a previously unoccupied atomic orbital rather than in the continuum.

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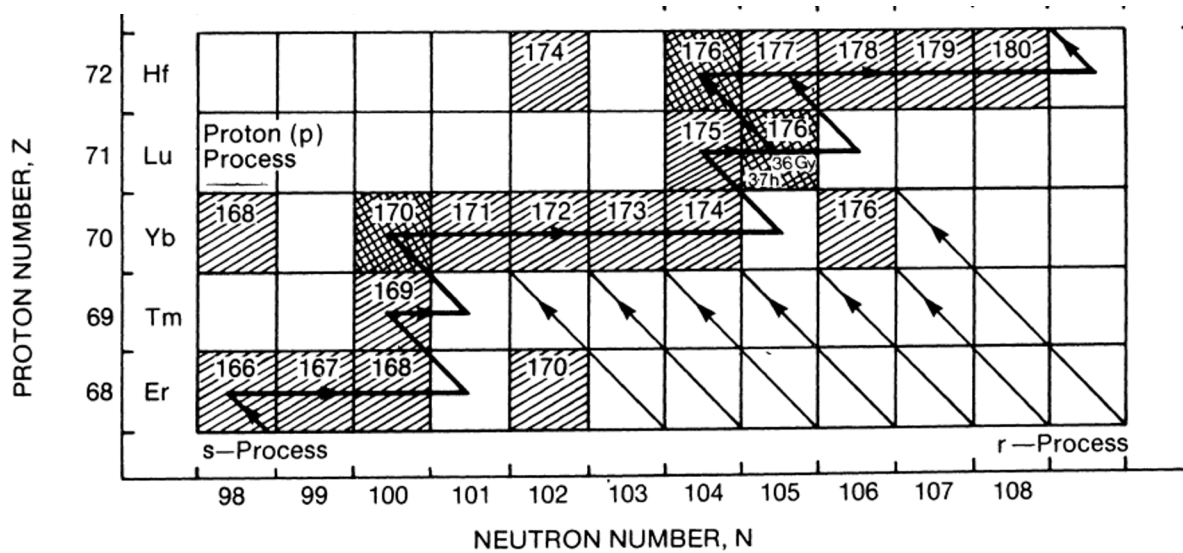
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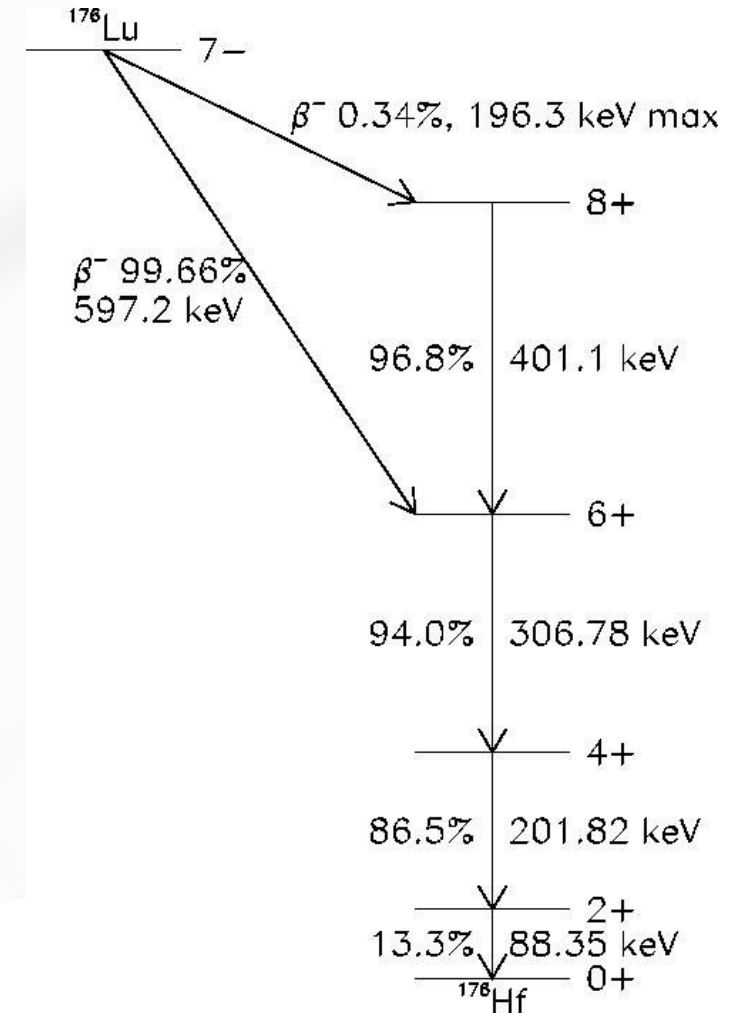
# $^{176}\text{Lu}$ physics case

## Cosmo-chronometer or stellar thermometer?

$^{176}\text{Lu}$  is one of the few naturally occurring radio nuclides that have survived from the era of nucleosynthesis. Its present isotopic abundance [1] is 2.6% and its half-life is  $4.08 \times 10^{10}$  yr [2].



**Figure 1.** s-process path in the rare earth element mass region. s-only process nuclides  $^{170}\text{Yb}$ ,  $^{176}\text{Lu}$  and  $^{176}\text{Hf}$  are shielded from r-process contributions by  $^{170}\text{Er}$  and  $^{176}\text{Yb}$  respectively. The s-process branches at  $^{176}\text{Lu}$  if a significant population of the 3.68 h isomeric state occurs.

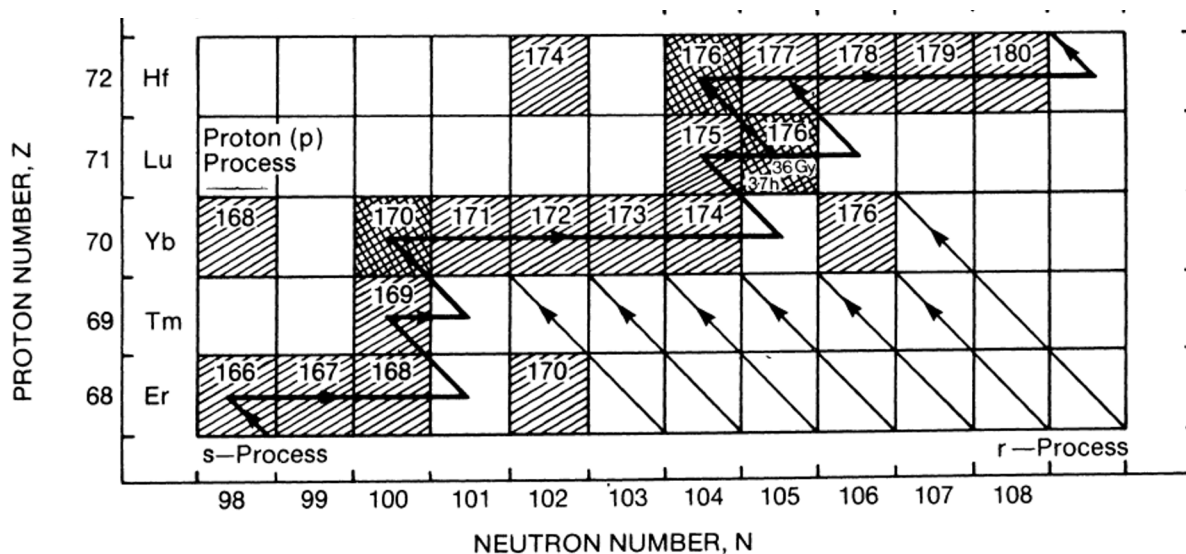




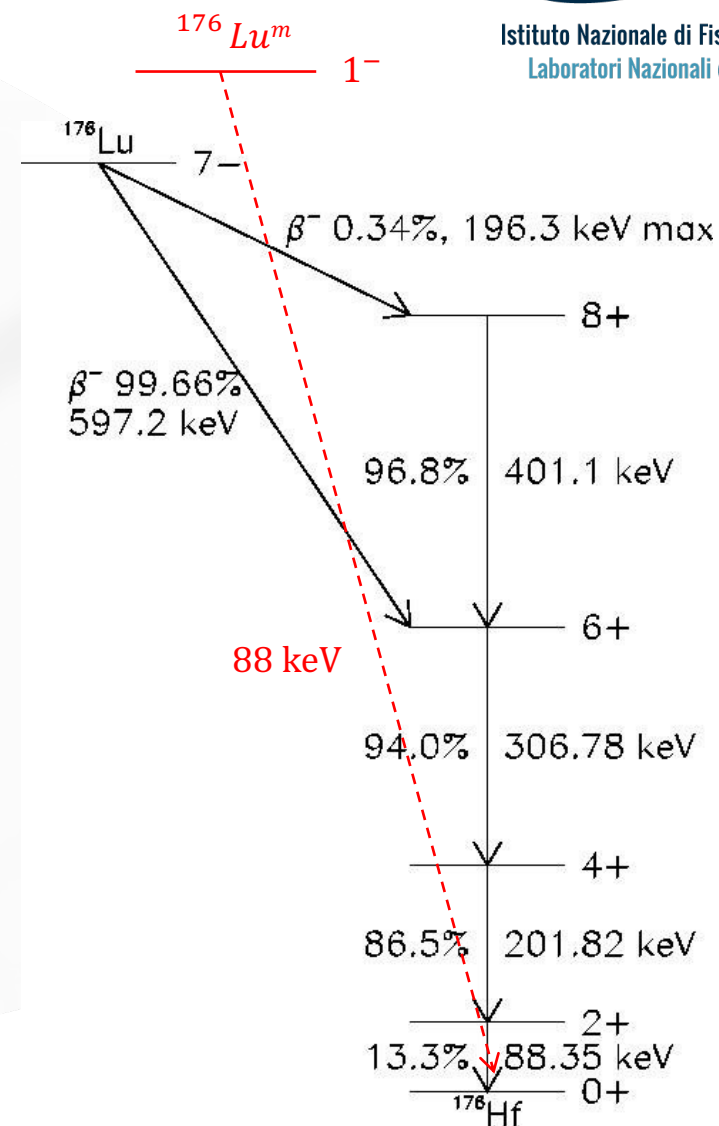
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### $^{176}\text{Lu}$ : An unreliable *s*-process chronometer

K. T. Lesko, E. B. Norman, R-M. Larimer, and B. Sur

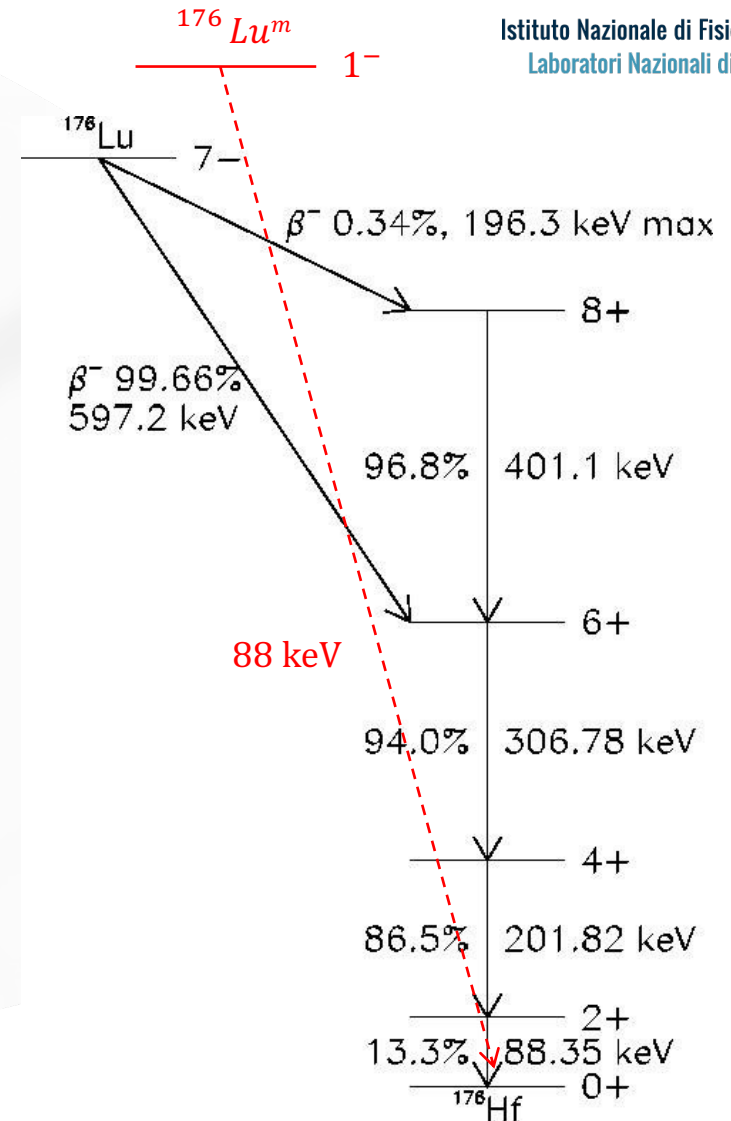
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(Received 17 October 1990)

A level scheme of  $^{176}\text{Lu}$  up to  $\sim 1400$  keV excitation energy is deduced from a  $\gamma$ - $\gamma$  coincidence experiment and previously published particle transfer data. 170  $\gamma$ -ray transitions are placed between 85 levels. We identify 27 previously unknown levels and 131 previously unknown transitions in  $^{176}\text{Lu}$ . With this  $\gamma$ -ray data we place the energy of the isomer at 122.9 keV. A level at 838.5 keV ( $J^\pi = 5^-, \tau_{1/2} < 10$  ns) is found to decay with substantial strength to both the ground state ( $7^-, 4.08 \times 10^{10}$  yr) and the 122.9 keV isomer ( $1^-, 3.7$  hr). The presence of this level guarantees the thermal equilibrium of  $^{176}\text{Lu}^{s,m}$  for  $T \geq 3 \times 10^8$  K and therefore during *s*-process nucleosynthesis. The resulting temperature sensitivity of its effective half-life rules out the use of  $^{176}\text{Lu}$  as an *s*-process chronometer. The use of  $^{176}\text{Lu}$  to determine *s*-process temperatures is discussed.



# The onset of the LTE (Local Thermal Equilibrium)

- È necessario dissipare l'energia per avere termalizzazione
- *We are assuming a typical lifetime of nuclear excited states to be in the order of fs (a few  $10^{-15}$  sec). We are using a typical reaction rate estimate according to the well-known formula  $R = I_X N_t \sigma$ , where  $I_X$  is the in-plasma total photon flux in  $s^{-1}$ ,  $N_t$  is a surface "target" density term, that in a plasma represents the radius averaged density in an assumed spherical plasma plume, and  $\sigma$  is the interaction cross-section. Assuming an excited state for, e.g.,  $^{176}\text{Lu}^*$ , around 122.45 keV, and considering  $n_e = n_i = 10^{27} \text{ m}^{-3}$  (a typical stars interior density), at  $T_e = T_i = 6.68 \text{ keV}$  the excited level lifetime is already exactly the same of the excitation rate, meaning that this level can be populated and it is in thermal equilibrium in the assumed laser-induced plasma lifetime (order of ps or tens of ps). This calculation has been also rescaled to a more realistic expected density of a real laser-induced plasma scenario, assuming  $n_e = n_i = 10^{25} \text{ m}^{-3}$ . In this case, the required plasma temperature to get the thermal equilibrium goes to around 37.5 keV. This value seems to be however absolutely achievable in the foreseen laboratory scenario, confirming that the decay from excited states is in principle feasible.*

# Why plasma: $\beta$ -decays

Stellar nucleosynthesis proceeds in a hot and dense environment which affects the degree of ionization of the atoms involved in the stellar nucleosynthesis.

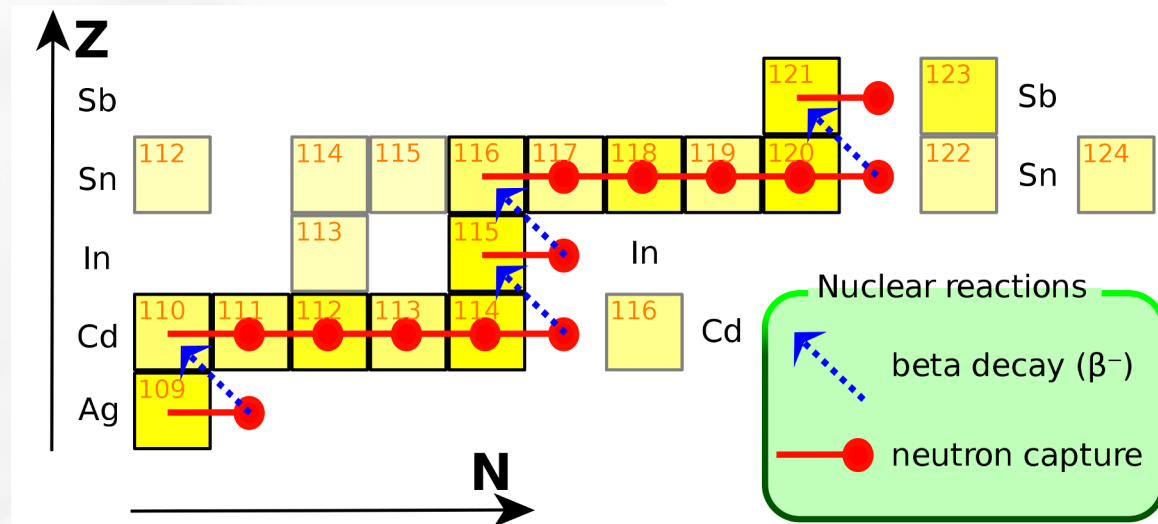
**What happens when atoms are highly ionized?**

**The beta decay in highly ionized atoms shows important variations compared to neutral species**

1. Electron Capture becomes impossible in fully ionized atoms.
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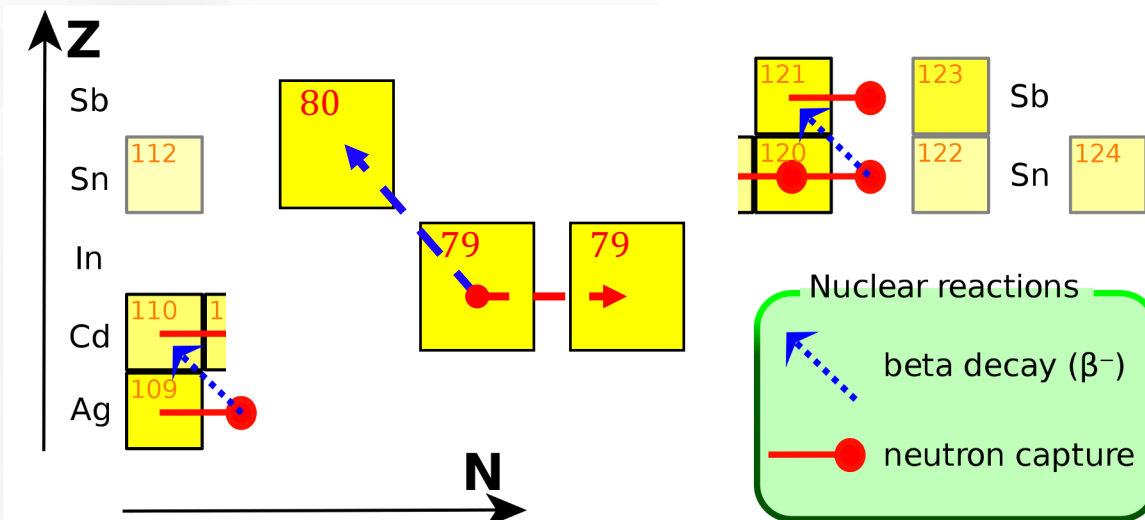
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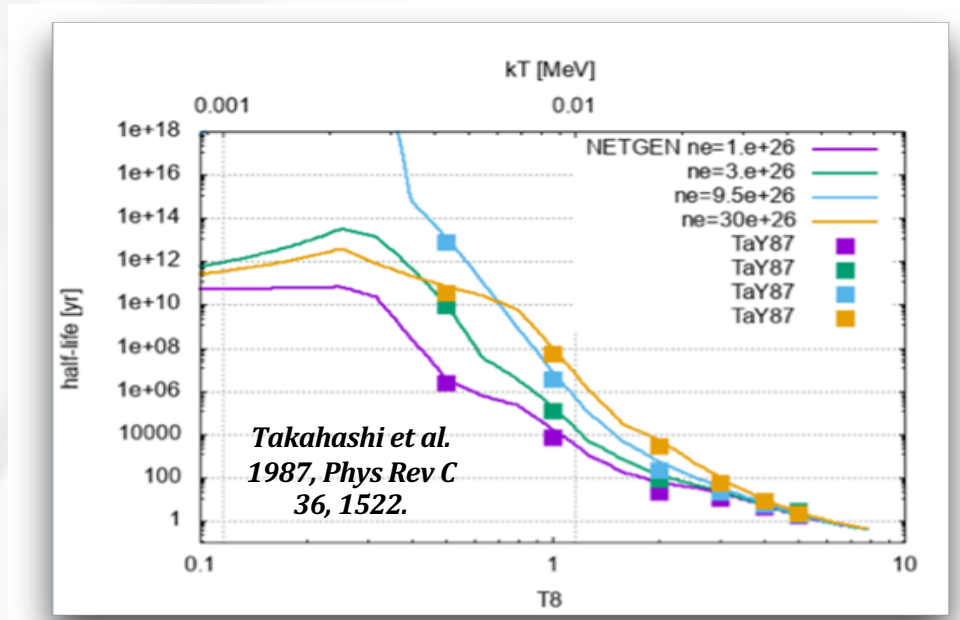
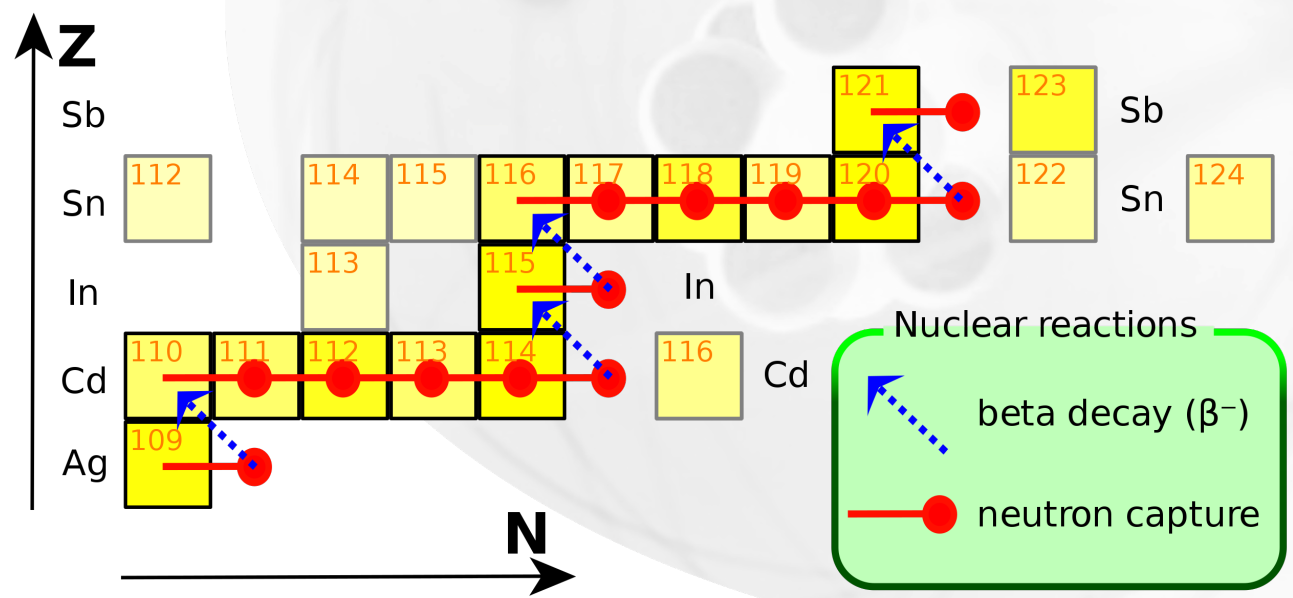
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# Why plasma: $\beta$ -decays

Stellar plasma environment ( $\rho, T$ ) can play a major role in modifying the rates at the branching point in s-process nucleosynthesis. *Temperature dependent variations evaluated in the seminal work of Takahashi and Yokoi.*

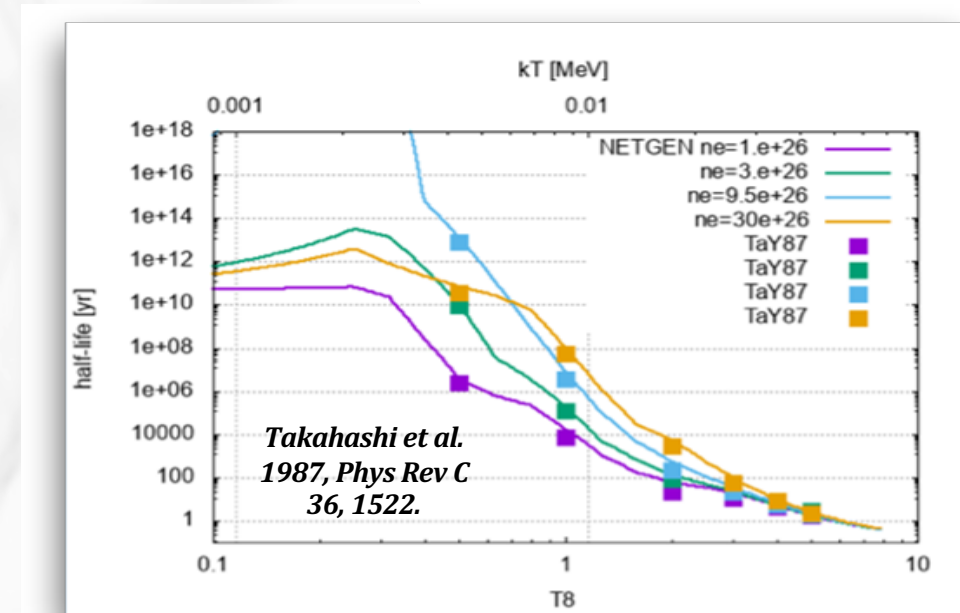
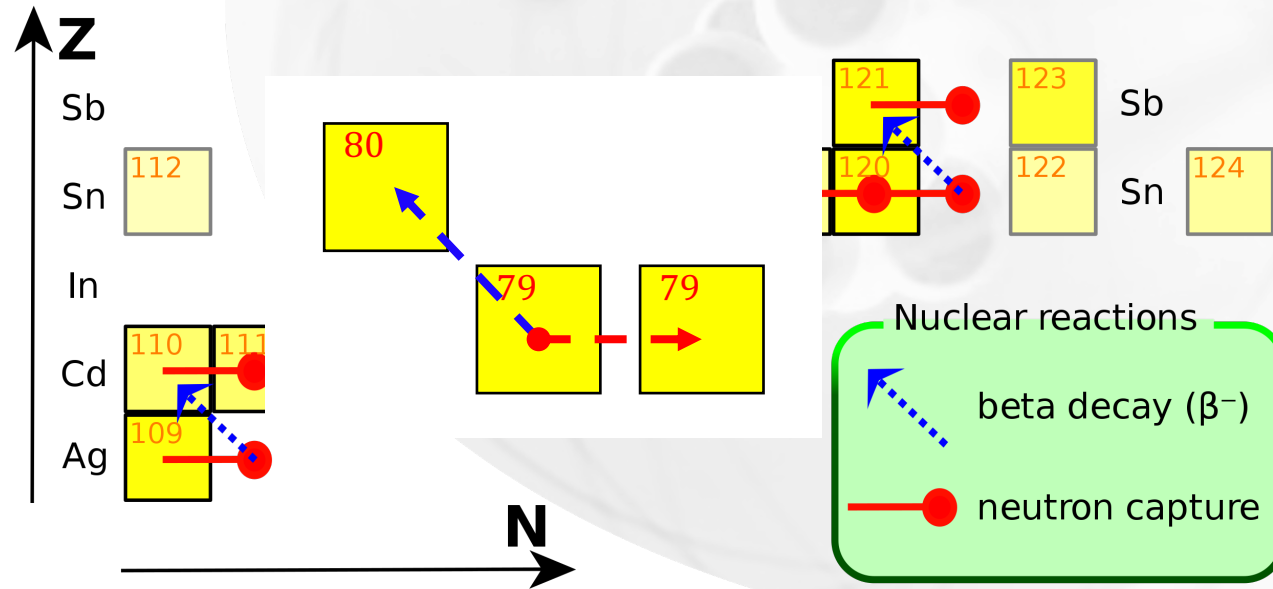


<https://doi-org.ezproxy.cern.ch/10.1103/PhysRevC.36.1522>

*Original predictions of modifications in  $\beta$ -decay rates in plasma by Takahashi and Yokoi*

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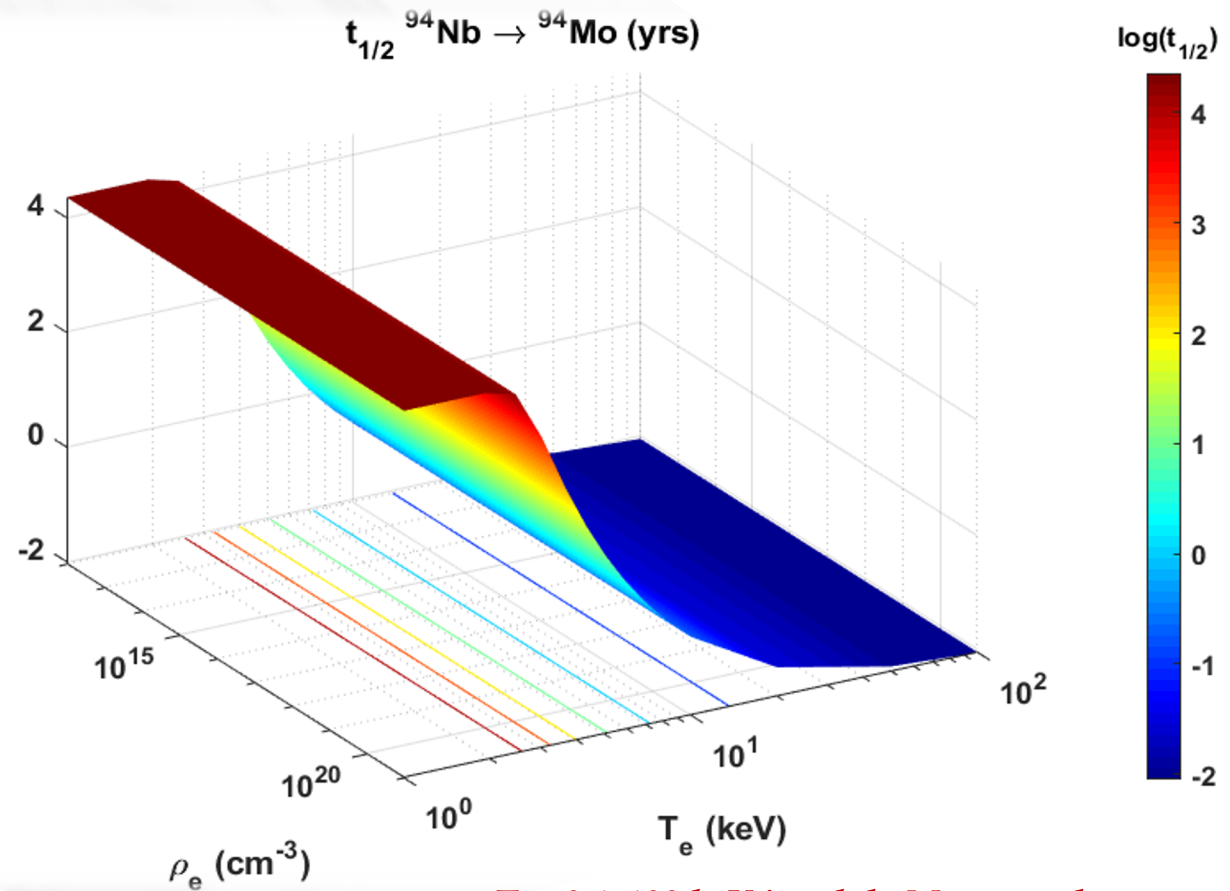
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# Scaling results to stellar environment

Once measured the variation, it can be compared to Takahashi-Yokoi theory (without LTE hypothesis)

Adapt to stellar atmosphere assuming  $n_e$  and  $T_e$  leading to ions in LTE population

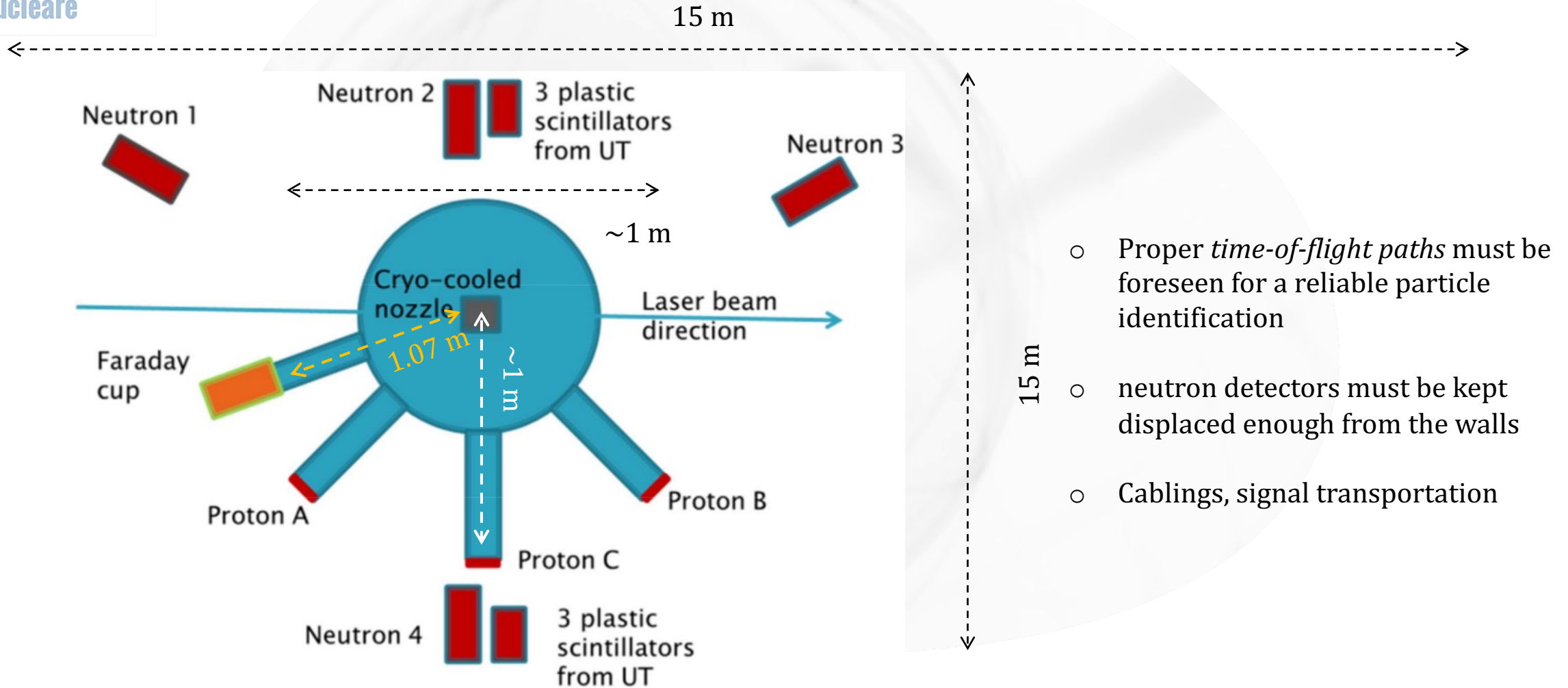
Variation with  $T_e$  stronger than with  $\rho_e$  so “stellar effect” can be modelled in ECR plasmas



*$T_e = 0.1-100$  keV in a lab. Magnetoplasma*



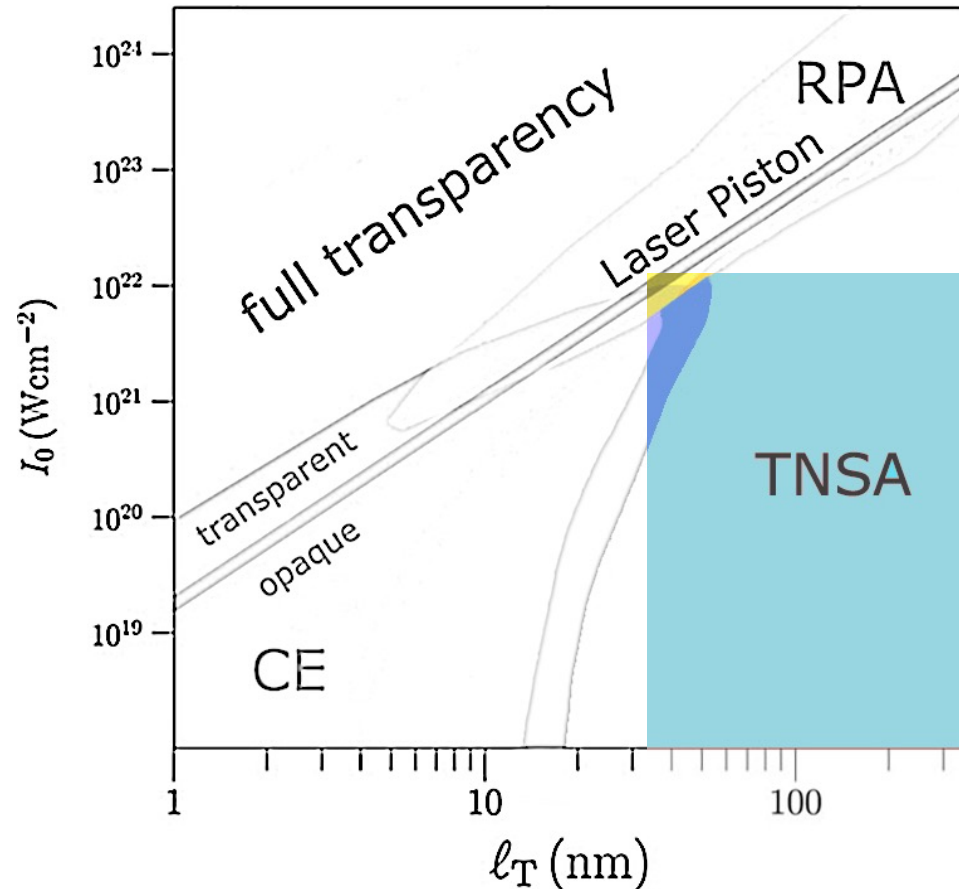
# Experimental area: an example







# Experiments at the PW regime



**High-density target** → solid, Lu target

Target choice to be optimized in view of the thermalization goal

Possible choice: non-isotopic cuboids of metallic bulk material (*e.g.*, natural Lu) coated with a layer of at least 200-300  $\mu\text{m}$  of isotopic enriched material

<https://www.frontiersin.org/articles/10.3389/fphy.2022.727718/full>



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