



Dipartimento di Fisica
G. Occhialini



Sez. Milano-Bicocca

Status of LAB radiopurity measurements

Italian JUNO Meeting - PoliMi
5-6th May, 2022

Radiopurity LAB measurements

Juno baseline request for LAB:

$$^{238}\text{U}, ^{232}\text{Th}, ^{40}\text{K} < 1 \cdot 10^{-15} \text{g/g}$$

Two procedures have been tested in order to identify natural contamination in LAB samples

High sensitivity measurements



NAA + Low background detector

In recent months we have **validated a measurement procedures** suitable to achieve the required sensitivity



Uranium and Thorium

Potassium

- Concentrate the sample
- Remove interferences

Avoiding to introduce external contamination

Main steps for ^{238}U and ^{232}Th measurements

Cleaning protocol
(Pre-Irradiation)



Any manipulation or treatment of the sample could introduce contaminations before irradiation

Chemical/Radiochemical
Treatments
(Pre-Irradiation)



Allows to **remove interferences** and **concentrate** the sample

Sample irradiation



Allows to transform long life nuclide $^{238}\text{U}/^{232}\text{Th}$ into the radioactive short life $^{239}\text{Np}/^{233}\text{Pa}$ nuclide. Sensitivity <1ppt

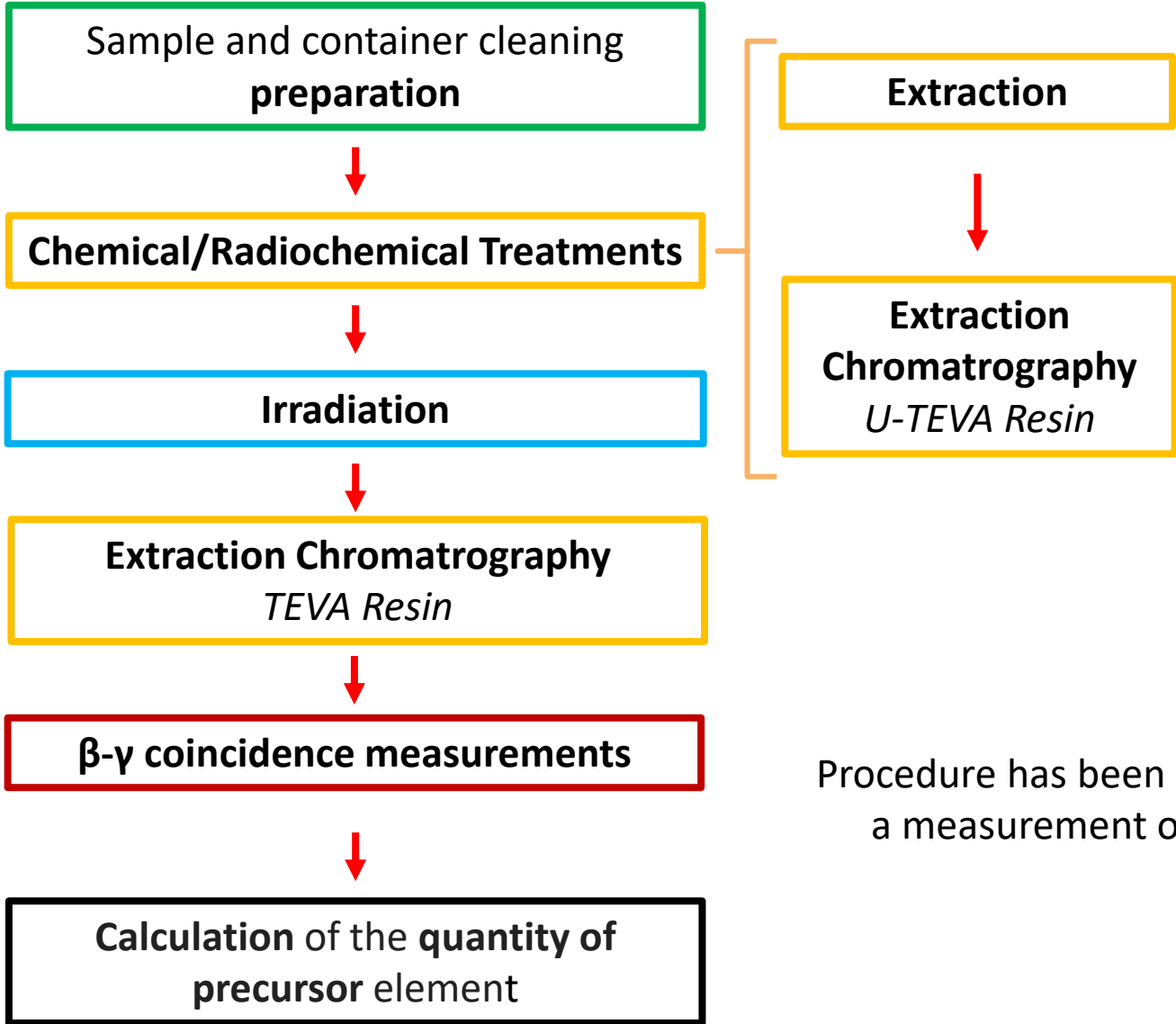


γ measurements



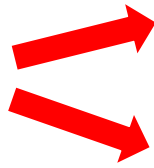
We developed a new detector suitable to **$\beta - \gamma$ coincidence** measurements on liquid irradiated samples

^{238}U - ^{232}Th procedure validation



Procedure has been **validated** performing a measurement on a **blank** sample

**Cleaning protocol
(Pre-irradiation)**



Tools washing

Reagents/Containers validation

All operations are carried out in a
clean room

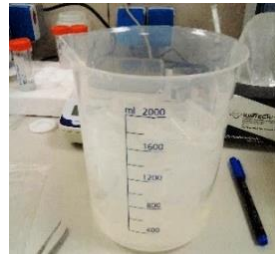


All tools are cleaned prior to sample
handling with a specific protocol



4 days of immersion in slightly nitric
acid water for all the tools

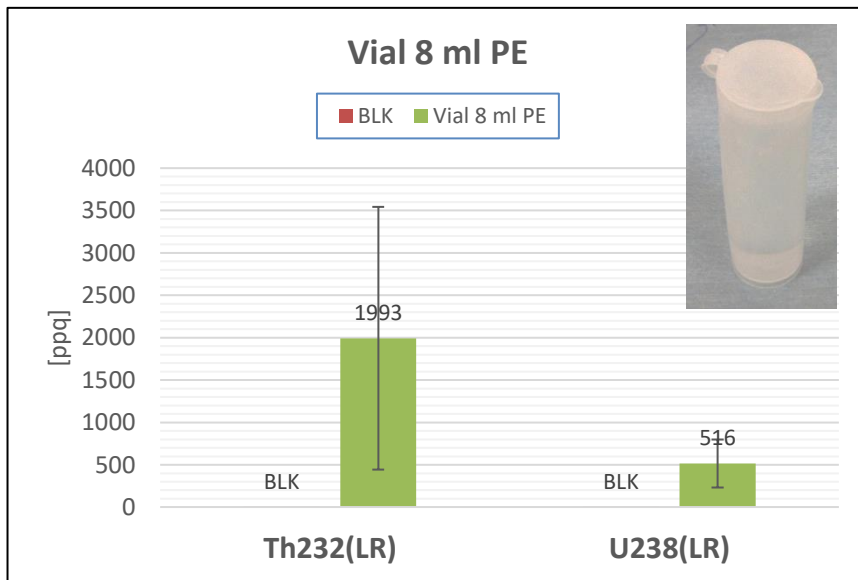
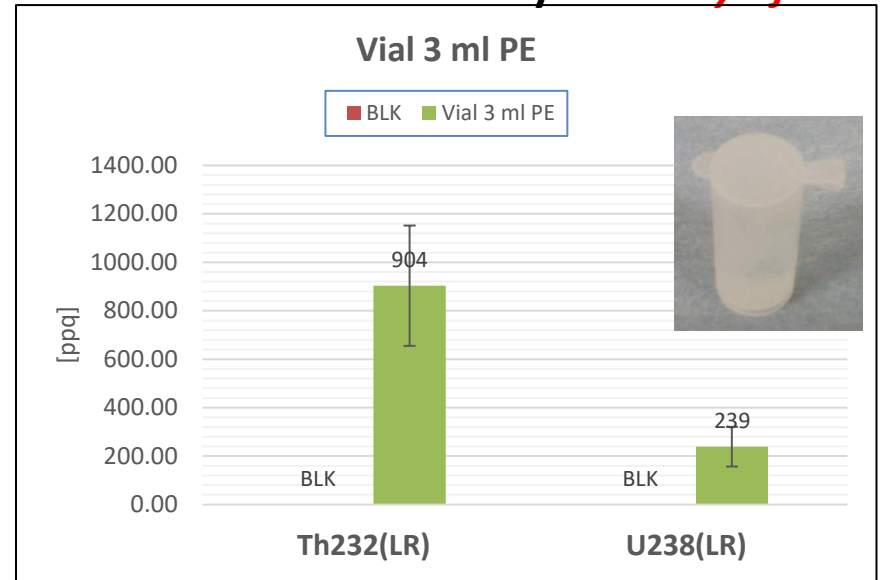
Vials are filled with nitric acid and water
for several days



Containers validation

ICP-MS measurements – Gran Sasso Laboratory *Courtesy of S. Nisi*

- The containers in Polyethylene (PE) could release uranium and thorium from the walls
- The containers in with pressure cap are unsuitable for ultra-low background: the insertion of the cap has high probability to introducing contamination in the sample



Screw cap

It's **mandatory** to use tools and container in **PFA/PTFA material with screw cap** suitable for trace elements analysis

Reagents validation

Water

ICP-MS measurements – Gran Sasso Laboratory

| | ^{238}U | ^{232}Th |
|---------------------------------|------------------------------------|------------------------------------|
| H ₂ O MilliQ | $< 0.7 \cdot 10^{-15} \text{ g/g}$ | $< 0.8 \cdot 10^{-15} \text{ g/g}$ |
| H ₂ O MilliQ Element | $< 0.7 \cdot 10^{-15} \text{ g/g}$ | $< 0.8 \cdot 10^{-15} \text{ g/g}$ |

Courtesy of S. Nisi

Nitric Acid

ICP-MS measurements – Gran Sasso Laboratory

| | ^{238}U | ^{232}Th |
|---------------------------|----------------------------------|----------------------------------|
| HNO ₃ Iper-pur | $< 3 \cdot 10^{-14} \text{ g/g}$ | $< 3 \cdot 10^{-14} \text{ g/g}$ |



Chemical/Radiochemical Treatments (Pre-irradiation)



Allow to **remove interferences** and **concentrate** the sample



Liquid-Liquid Extraction

Consist in the transfer of the contaminations of U and Th from **LAB sample** into a liquid solution(**HNO₃ + water**)

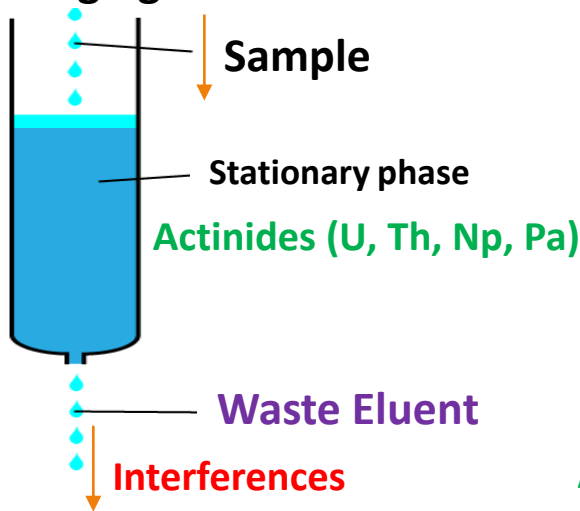
Proportion: HNO₃+Water:LAB =1:5



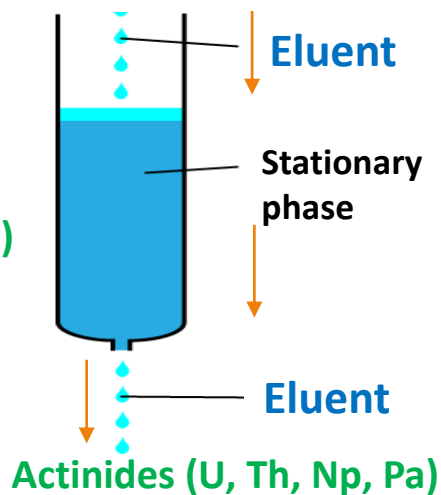
Extraction Chromatography

Ideally the column chromatography **selectively absorbs actinide activities** (U, Th, Pa ,Np) while allowing interferences pass through

Charging



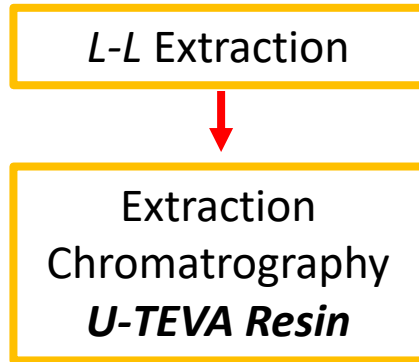
Washing (Acid solution)



Pre-Irradiation: U-TEVA Resin
Uranium and Thorium

Efficiency of pre irradiation steps

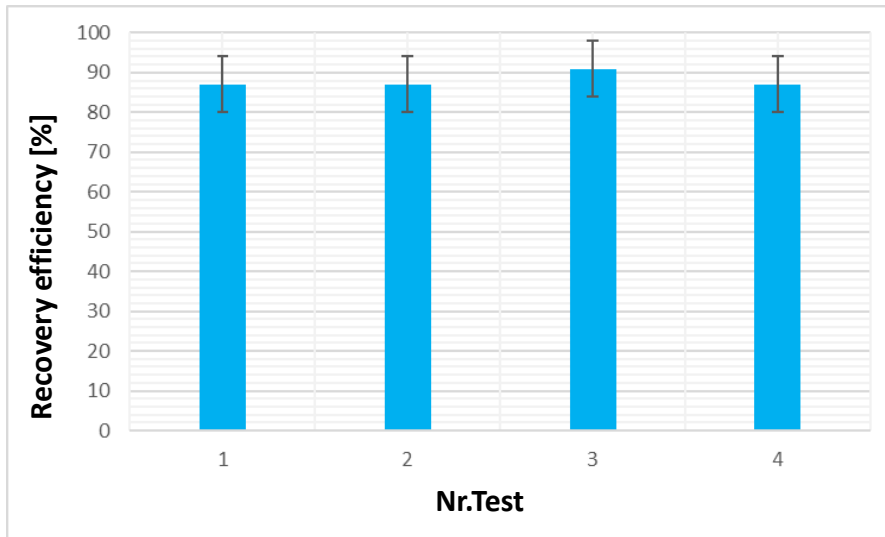
The effectiveness of the chemistry and radiochemistry treatments has been studied considering **spiked LAB samples**



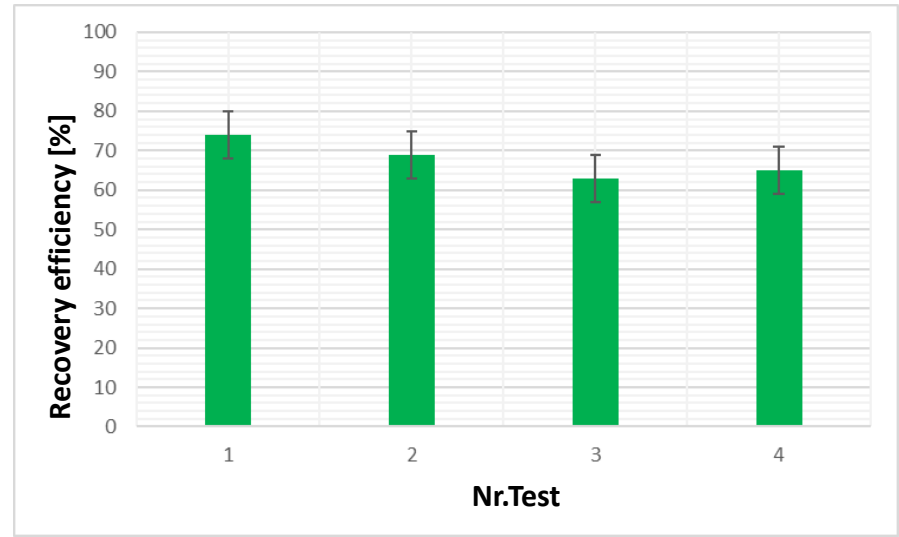
STDs containing a known amount of the element are used as **tracers**



²³⁸U Recovery Efficiency: $(88 \pm 5)\%$



²³²Th Recovery Efficiency: $(68 \pm 7)\%$

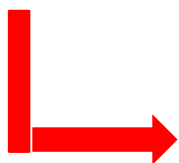
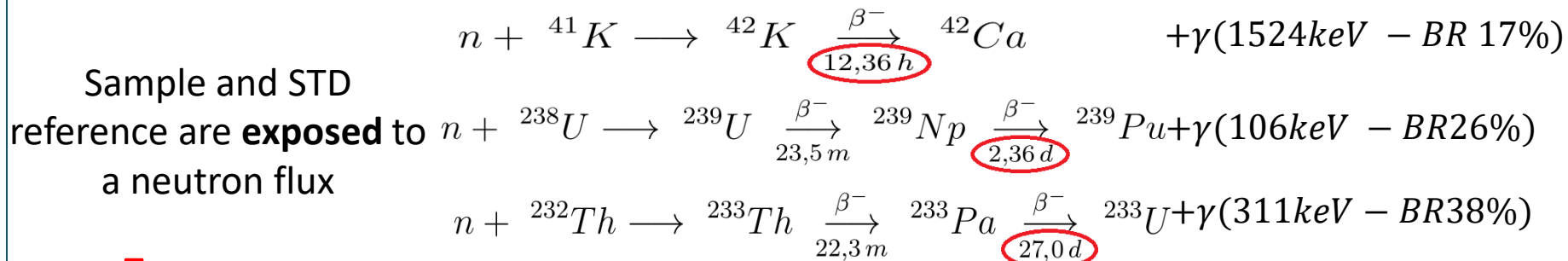


Neutron Activation Analysis (NAA)

Sample irradiation

The neutron activation process consists in the production of unstable isotopes through neutrons absorption by the nuclei present in the sample

The NAA technique consists of several steps:



Extraction of the irradiated sample and **measurement** of induced γ radioactivity



Calculation of the quantity of precursor element (A_ZX)



TRIGA Mark II
Research reactor
(250 kW) - Pavia, Italy

LAZY SUSAN facility:
 Flux of neutrons: $\approx 10^{12} \text{ cm}^{-2} \text{ s}^{-1}$
Irradiation Time: 6 hours

^{238}U - ^{232}Th steps post irradiation

Irradiation



Extraction Chromatography
TEVA Resin



β - γ coincidence measurements



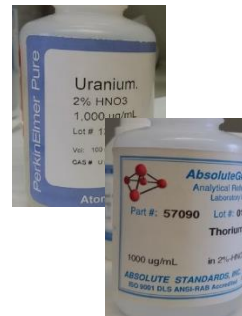
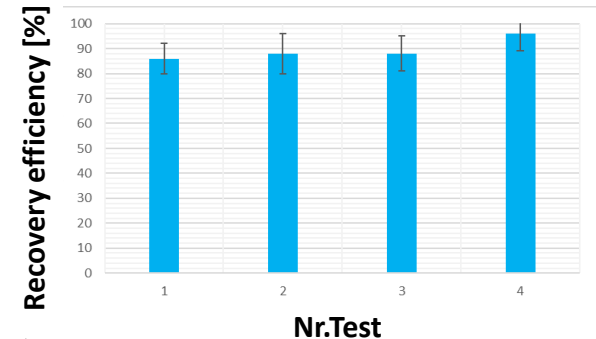
Calculation of the quantity of precursor element



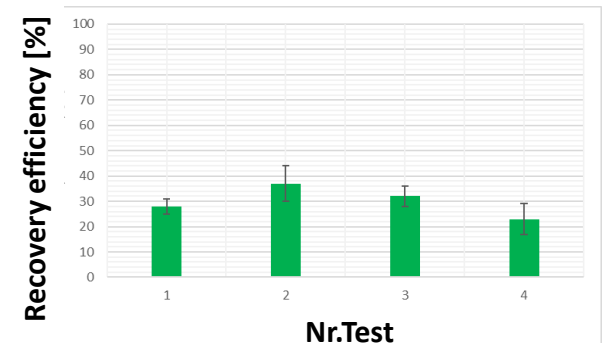
Removes any remaining interferences

$\epsilon_{\text{Removal}} \sim 99\%$

^{238}U Recovery
Efficiency: $(89 \pm 6)\%$



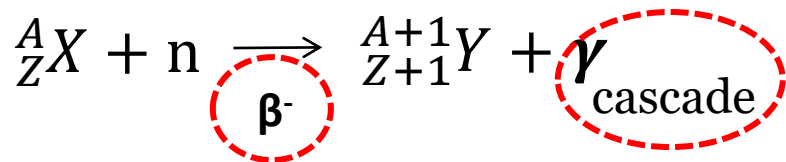
^{232}Th Recovery
Efficiency: $(30 \pm 8)\%$



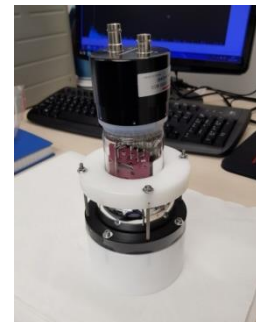
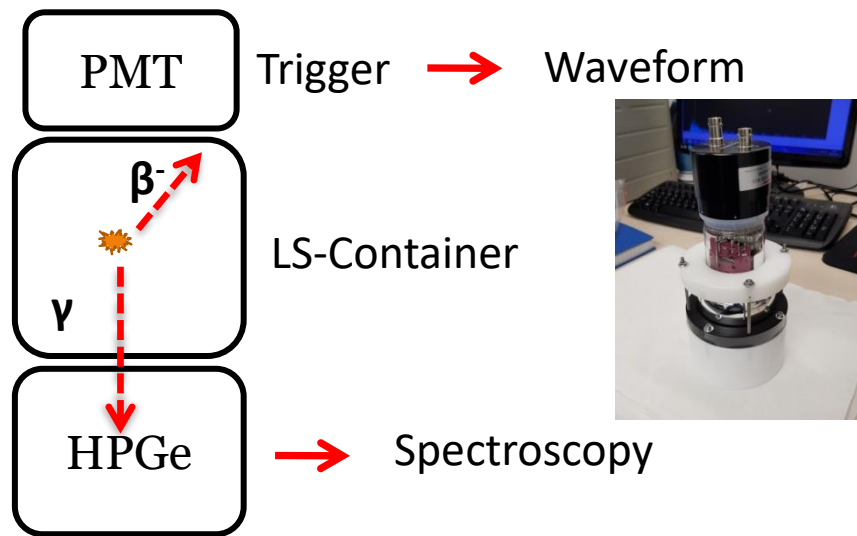
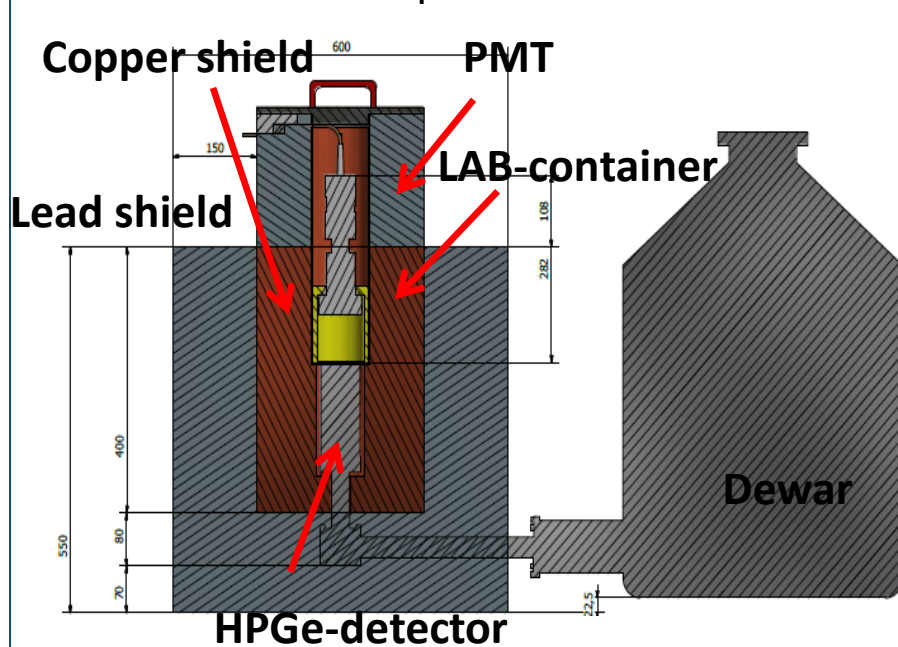
β-γ coincidence detector

β-γ measurements

Neutron Activation Reaction



Irradiated sample is mixed with not irradiated liquid scintillator



β-γ coincidence detector allows a strong **background reduction**

Sensitivity for ^{238}U - ^{232}Th

The blanks are critical for determining the contamination introduced during pre-treatment and are essential for evaluate the sensitivity

The blank went through all processing steps **just without LAB**

Nitric acid water
Blank mass: 228g



*It is representative of the measure of
1 kg of Liquid scintillator (LAB)*

Proportion: LAB:HNO₃+Water = 1: 5

| Blank | ^{238}U [g/g] | ^{232}Th [g/g] |
|------------------|------------------------------|-------------------------|
| Mass sample 228g | $(9,5\pm 2,4)\cdot 10^{-15}$ | $<7,7\cdot 10^{-14}$ |

limits @ 90% C.L.

In the hypothesis that we got a LAB sample of 1 kg without contaminations we could achieve a sensitivity of: $2\cdot 10^{-15}\text{g/g}$ for ^{238}U - $1,5\cdot 10^{-14}\text{g/g}$ for ^{232}Th

Radiopurity LAB – ^{40}K

Sample and container cleaning
preparation



LAB + STD reference has irradiated



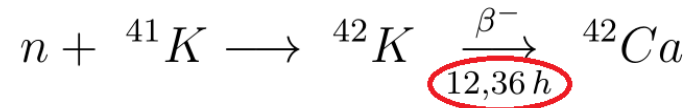
γ measurements (HPGe)



Calculation of the quantity of
precursor element

To study potassium contamination we
irradiated a LAB sample

Sample mass: **19g**



Isotopic abundance K :

$K^{39} \rightarrow \sim 93\%$

$K^{40} \rightarrow \sim 0,01\%$

$K^{41} \rightarrow \sim 7\%$

| γ -ray(keV) | BR(%) |
|--------------------|-------|
| 312.6 | 0.34 |
| 1524.6 | 17.64 |

^{40}K [g/g]

$< 8,3 \cdot 10^{-15}$

limits @ 90% C.L.

Summary

^{238}U and ^{232}Th measurements

- The cleaning protocol has been defined Processes are reliable and consistent
- Chemistry procedure has been tested Recovery efficiency has been determined



For **Uranium** and **Thorium** a representative **blank sample** of the whole procedure has been measured

| Blank | ^{238}U [g/g] | ^{232}Th [g/g] |
|------------------|--------------------------------|-------------------------|
| Mass sample 228g | $(9,5 \pm 2,4) \cdot 10^{-15}$ | $< 7,7 \cdot 10^{-14}$ |

Rescaling these results for a mass of 1 kg we could achieve a **sensitivity**: $2 \cdot 10^{-15}$ g/g for ^{238}U and $1,5 \cdot 10^{-14}$ g/g for ^{232}Th

Radiopurity LAB – Potassium

| | | |
|---------|-----------------|----------------------------|
| LAB 19g | ^{40}K | $< 8,3 \cdot 10^{-15}$ g/g |
|---------|-----------------|----------------------------|

Future plan:

- We plan to perform measurements on new blank samples
- We will try out to increase sensitivity

