A condensation transition

at equilibrium and out-of-equilibrium

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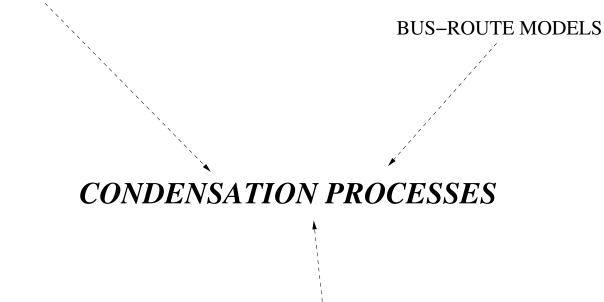
Stefano Iubini

What is a condensation/localization?

It is a process/transition where a single site of the lattice accumulates a finite fraction of mass/energy/number of particles

Here: homogeneous localization in the real space

GRANULAR MATERIALS

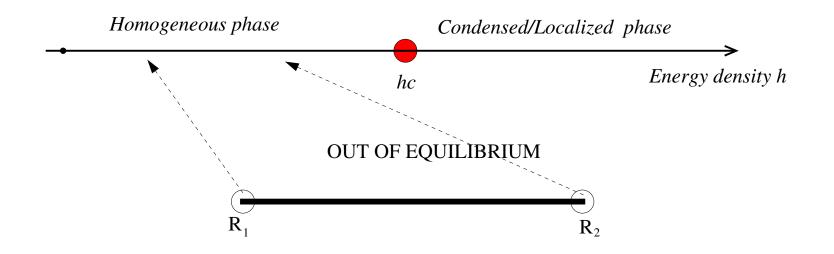


Discrete NonLinear Schroedinger (DNLS) equation

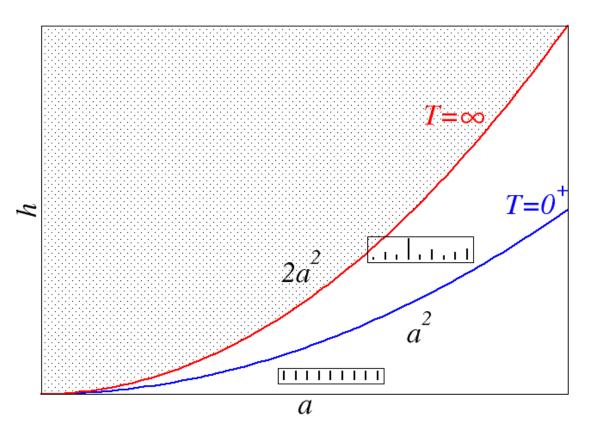
$$H = \sum_{j} (z_{j}^{*} z_{j+1} + z_{j} z_{j+1}^{*}) + \sum_{j} |z_{j}|^{4} \quad i\dot{z}_{n} = -\frac{\partial H}{\partial z_{n}^{*}} \qquad A = \sum_{j} |z_{j}|^{2}$$

What we are talking about: a lattice model with two conserved quantities





DNLS model $(z_i) \Longrightarrow$ C2C model $(c_i = |z_i|^2)$ $A = \sum_i c_i \equiv Na$ $H = \sum_i c_i^2 = \sum_i \epsilon_i \equiv Nh$ condensation $\leftrightarrow h > h_c = 2a^2$

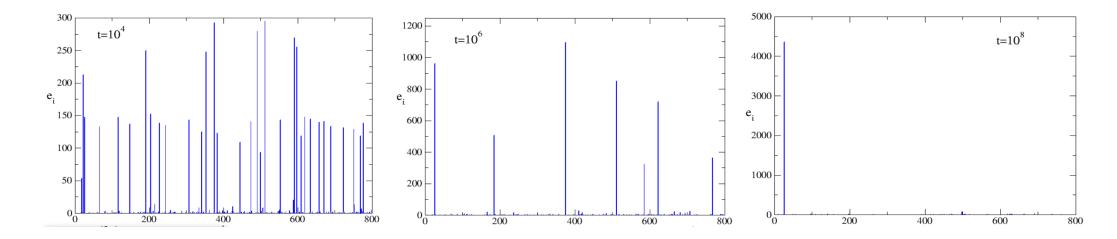


Region above $h = 2a^2$: Gradenigo, Iubini, Livi, Majumdar (2021)

Dynamics of a simple microcanonical MonteCarlo

$$a = 1, \quad h = 7.3, \quad N = 800$$

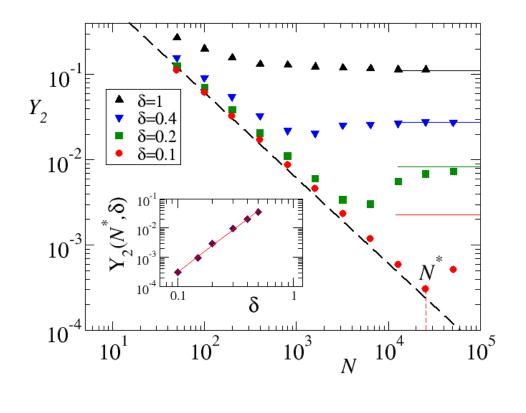
We choose a random triplet and we choose uniformly a new triplet having the same mass and energy

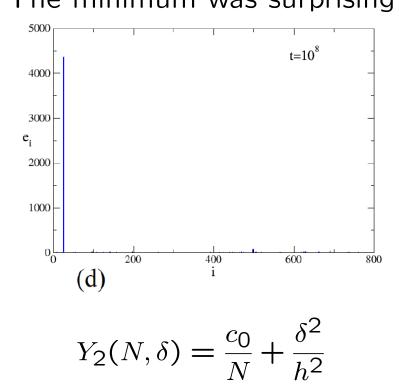


When we approach h_c things are more complicated

Order parameter:
$$Y_2 = \frac{\langle \epsilon^2 \rangle}{N \langle \epsilon \rangle^2} \equiv \frac{\langle \epsilon^2 \rangle}{Nh^2}$$

Control parameter: $\delta \equiv h - h_c$

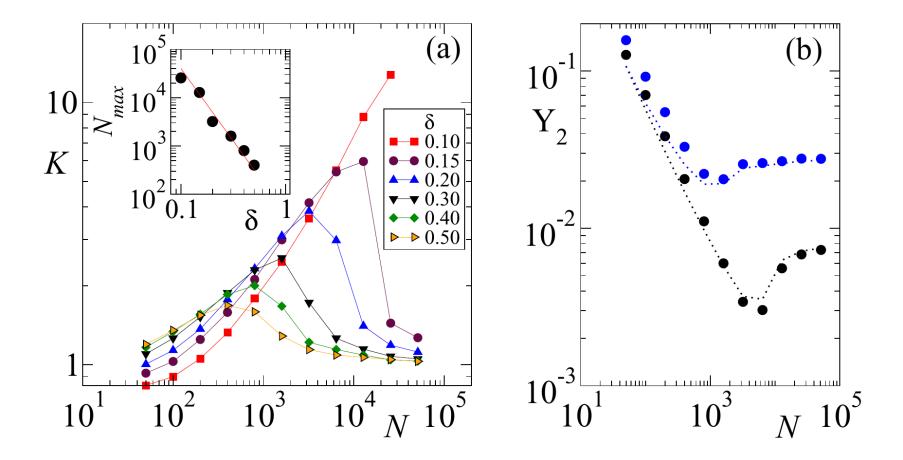




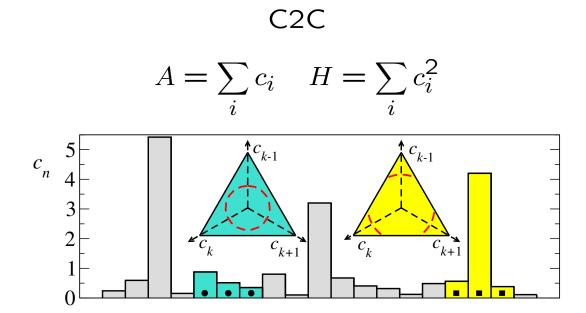
 $Y_2^{min} \simeq \delta^{\gamma}, \qquad \gamma = 3$

$K(N, \delta) \equiv$ effective number of peaks

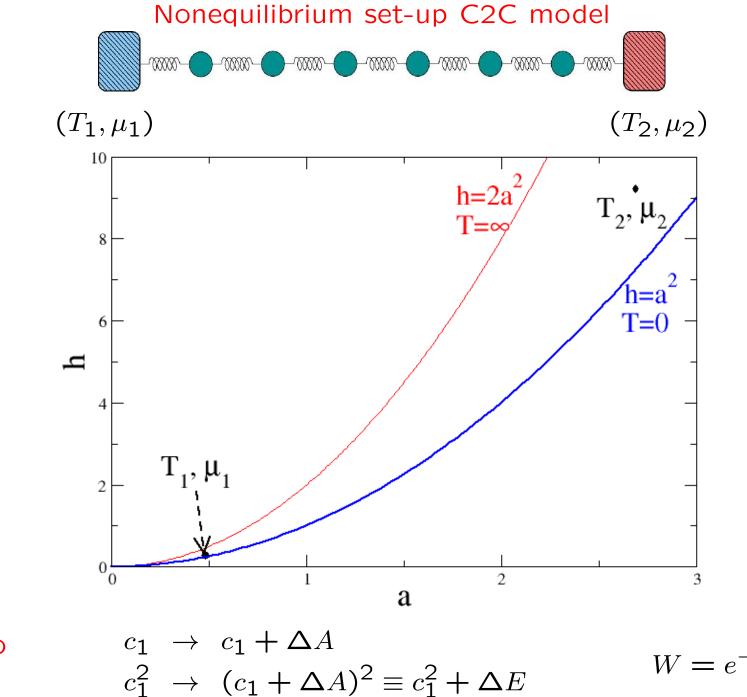
$$Y_2(N,\delta) = \frac{\mu_2}{Nh^2} + \frac{\delta^2}{K(N,\delta)h^2}$$



Equilibrium \implies Nonequilibrium

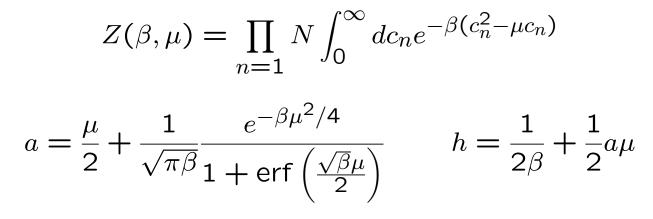


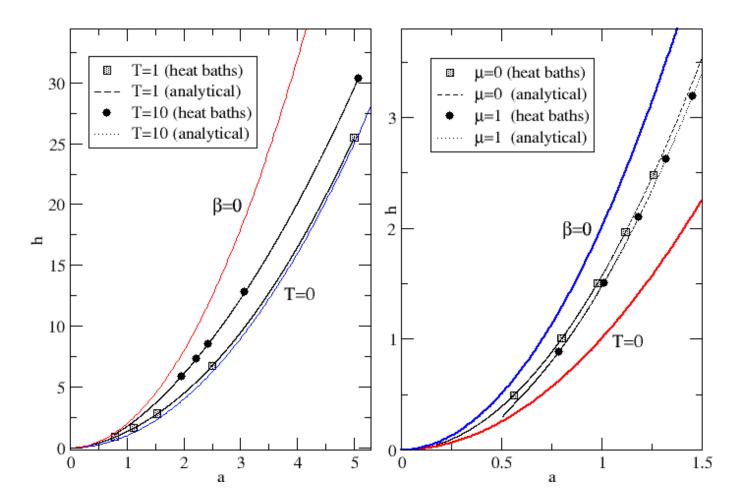
- No pinning (choice in any of the three arcs)
 - Pinning (choice in the same arc)



 $W = e^{-\beta(\Delta E - \mu \Delta A)}$

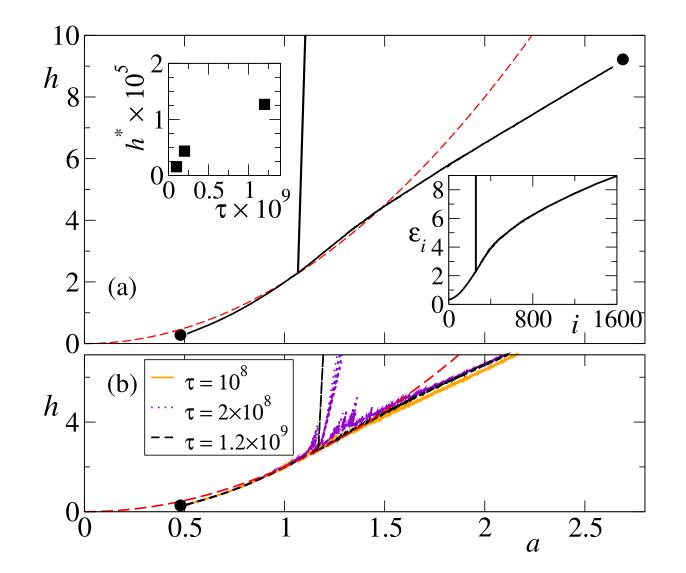
MonteCarlo heat baths Heat baths work well



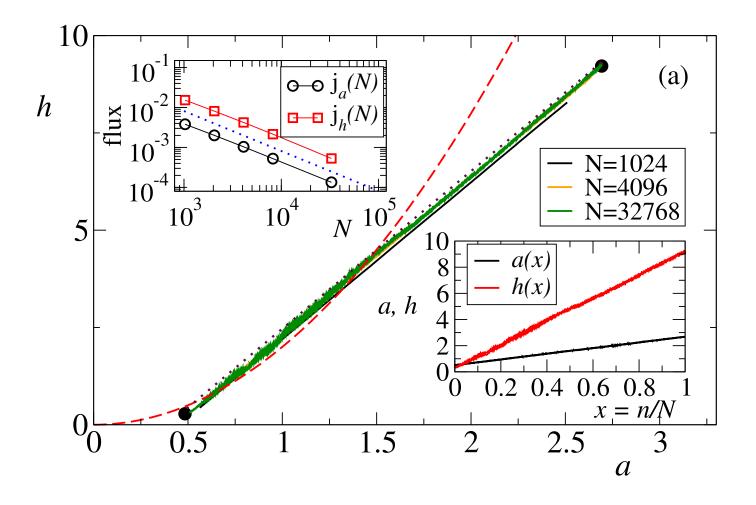


Out-of-equilibrium setup (C2C pinned)

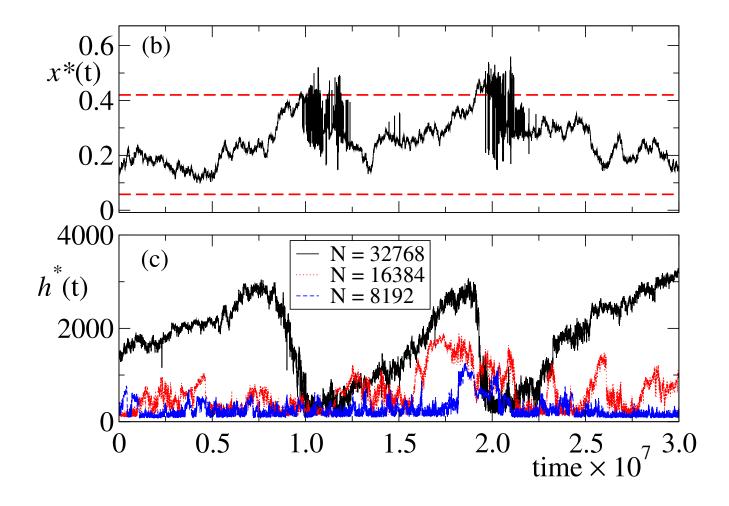
For each site *i* we evaluate the time averages $a_i = \overline{c_i}$, $h_i = \overline{c_i^2}$, then we plot $h_i(a_i)$



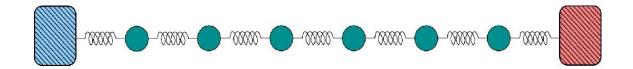
C2C unpinned: condensed steady state...



... because breathers diffuse



Why does nonequilibrium setup favour the condensation process?



$$(c_1, b, c_2) \longrightarrow (c_1 + x, b + \delta b, c_2 + y)$$

If $c_i = \bar{c}_i + \delta_i$ and $b \gg c_{1,2}$, if we perform an average over $\delta_{1,2}$,

$$\delta b \simeq -Q + \frac{1}{2b} \Delta (y - x), \quad Q > 0, \quad \Delta \equiv \overline{c}_1 - \overline{c}_2$$

Open questions

- We need to understand better the process of creation, diffusion and death of breathers in the unpinned case
- What does it happen to the Onsager coefficients when we cross the critical curve?
- What about the same setup for the DNLS equation?

Equilibrium

Finite-size localization scenarios in condensation transitions Phys. Rev. E 103, 052133 (2021)

Nonequilibrium

Condensation induced by coupled transport processes Phys. Rev. E 106, 054158 (2021)