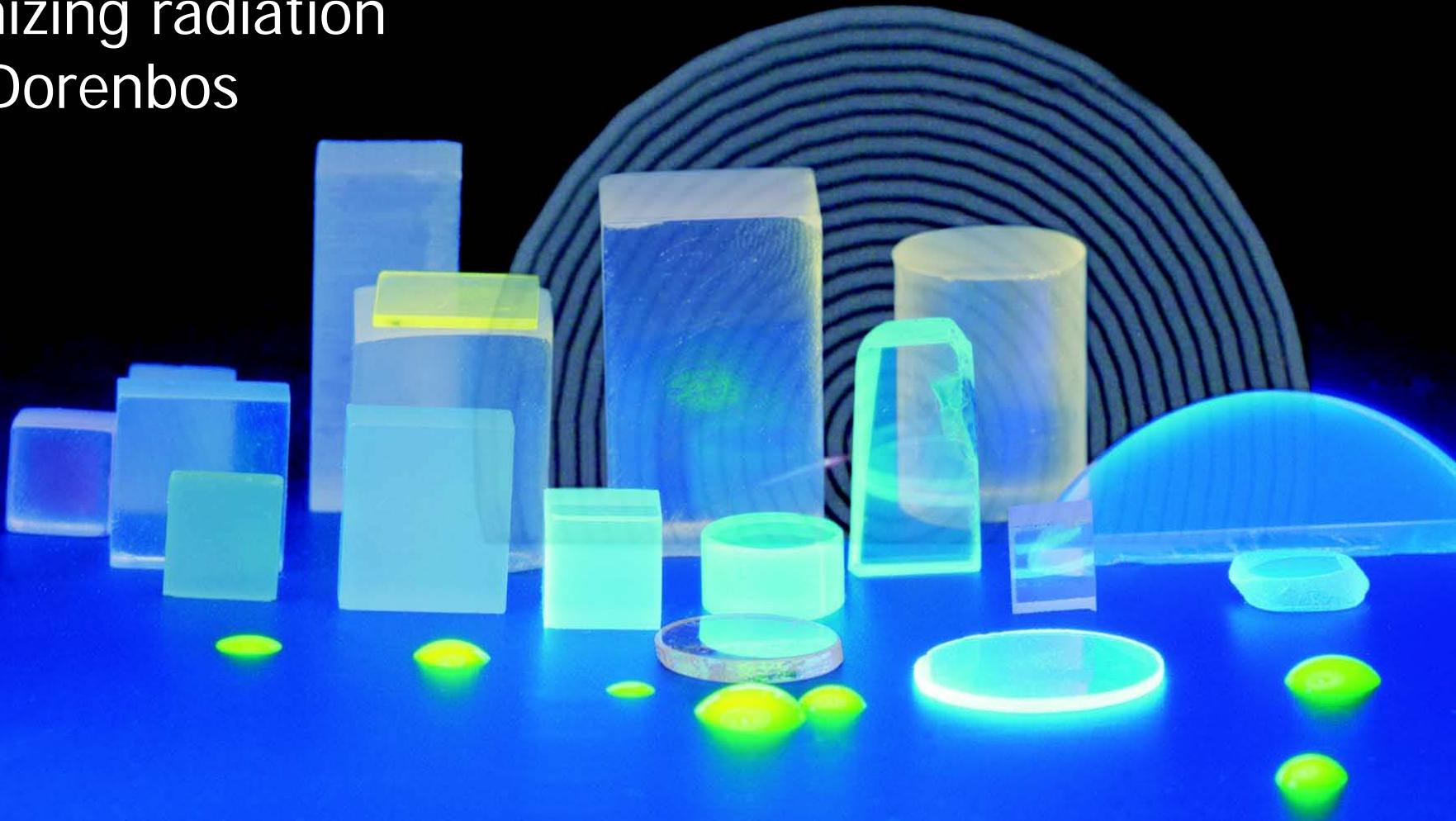


# Inorganic Scintillators for the detection of ionizing radiation

P. Dorenbos

# Inorganic Scintillators for the detection of ionizing radiation

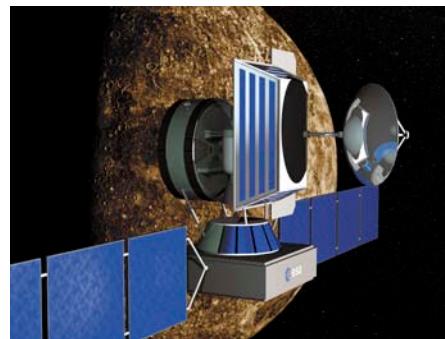
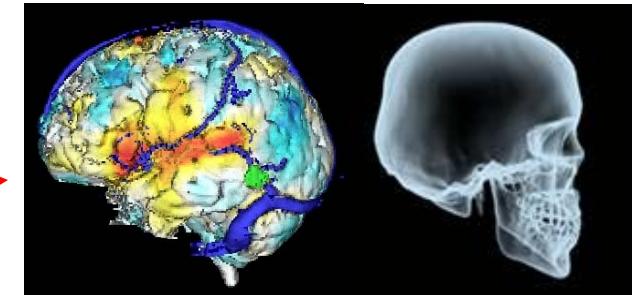
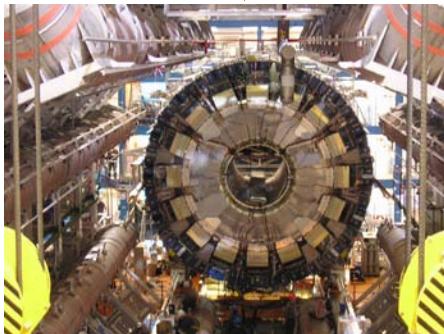
P. Dorenbos





## Scintillator applications

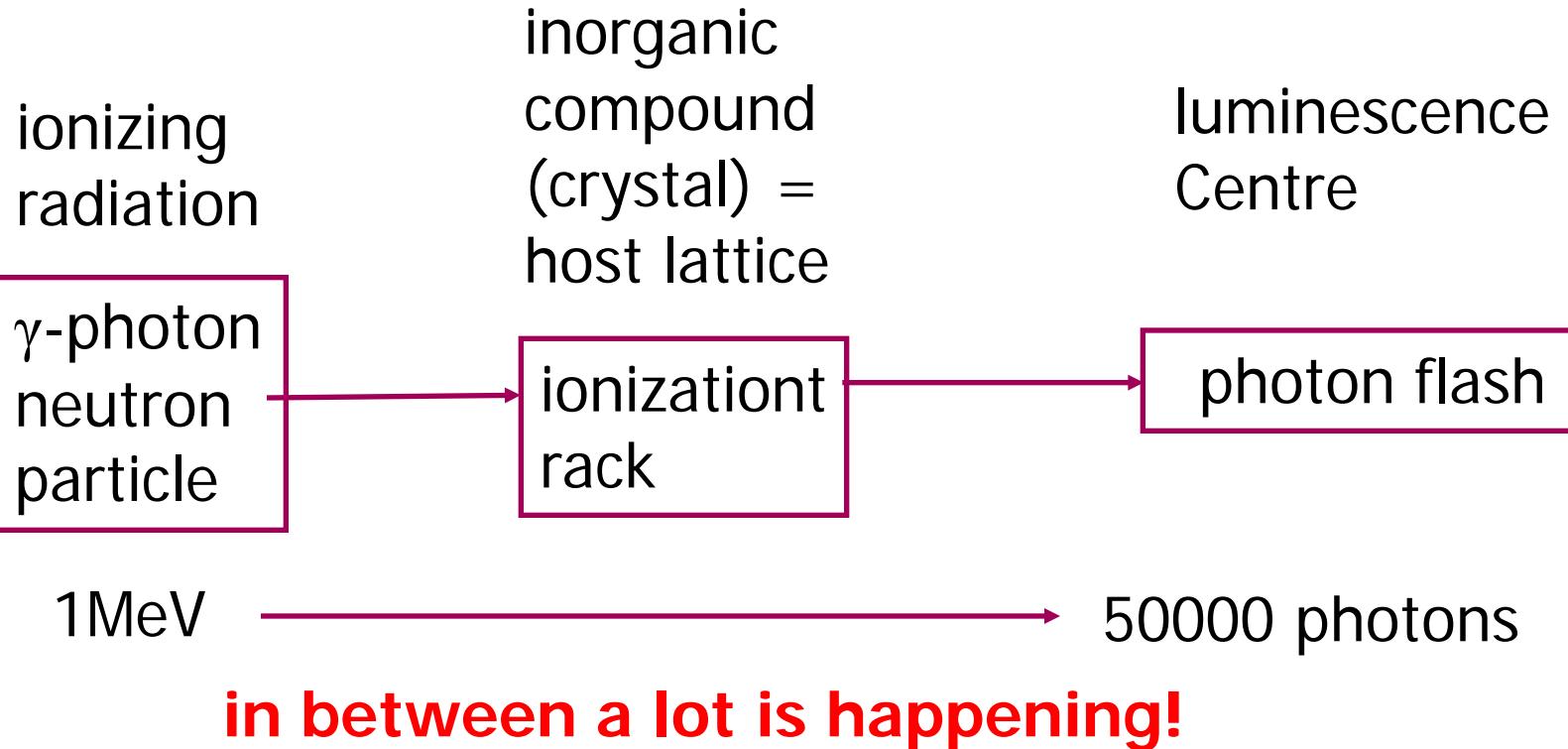
- security
- medical diagnostics
- industry + science
- space explorations
- high energy physics



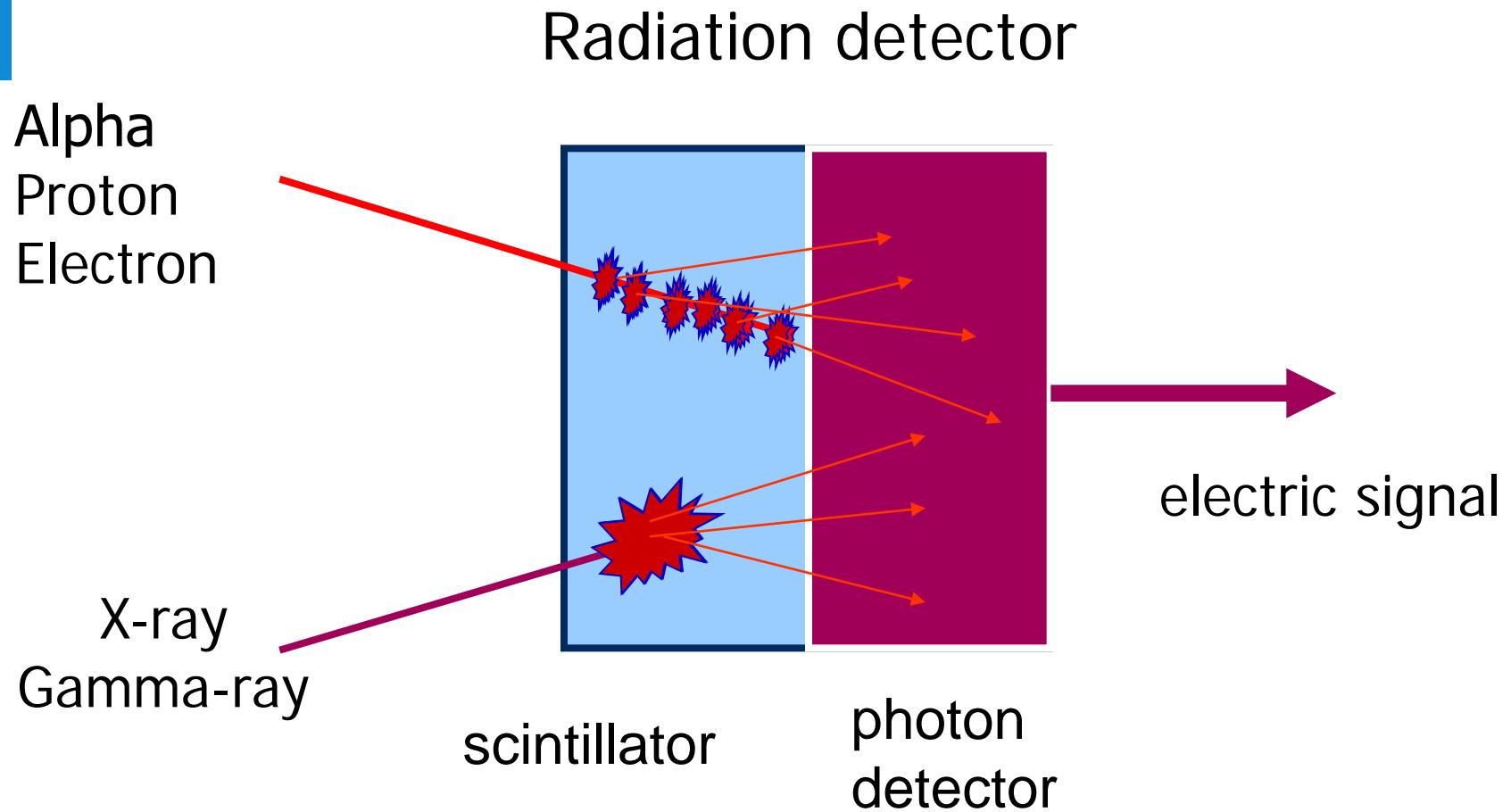
# Topics

- How it works?
    - The light output
    - The luminescence
    - The scintillation mechanism
  - How it performs?
    - Energy resolution
    - Time resolution
  - How it is used?
    - Current developments
- How many photons are emitted?  
What is color and speed of emission?  
What happens on an atomic scale?
- What is the status and are the limits?  
What is the status and are the limits?
- What are the hot topics today?

# The scintillation process



# Schematic of a radiation detector



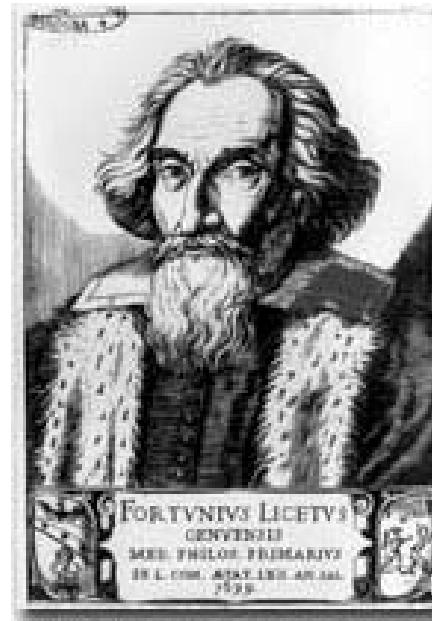
## Scintillators

inorganic crystals/powder  
organic plastics/crystals  
glass  
liquid  
gas

## Light sensors

Human eye  
Photomultiplier tubes  
silicon photo diodes  
Si-PMTs  
CCDs  
gas-filled detectors

# The Bolognian stone



## LITHEOSPHORVS. DE LAPIDE BONONIensi

Acum inde excerptam ab ambiente clara  
HOC in similem esse constituer  
CEPTE.

## FORTVNII LICETI GENVENSIS

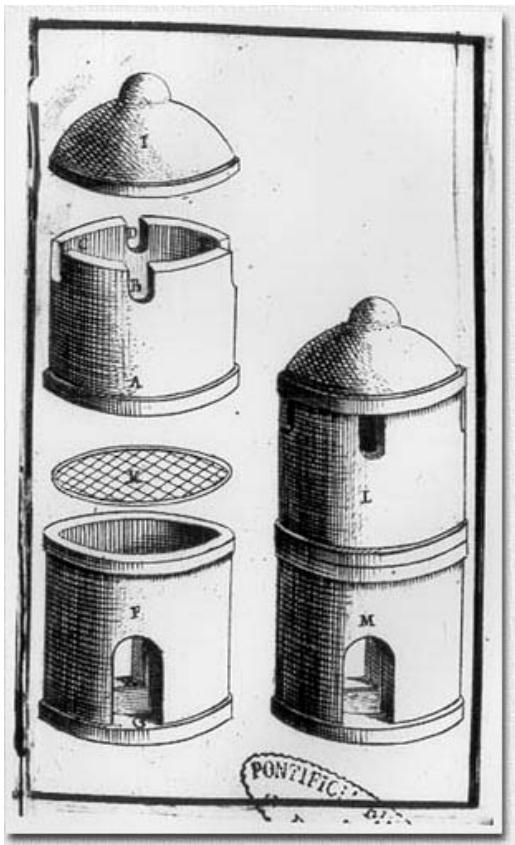
Pridem in Pisanis, super in Patavinis, natus in Bononiis  
et trahigimus Philosopha Universitate

Emendatio Etiamq; D.D.

## ALOYSIO CARDINALI CAPPONIO RAVENNE ARCHIEPISCOPO DIGAVIT.



Vnde, Ex Typographia Nicolai Schiraldi, & deo  
EMENDATIO ETIAMQ; D.D.

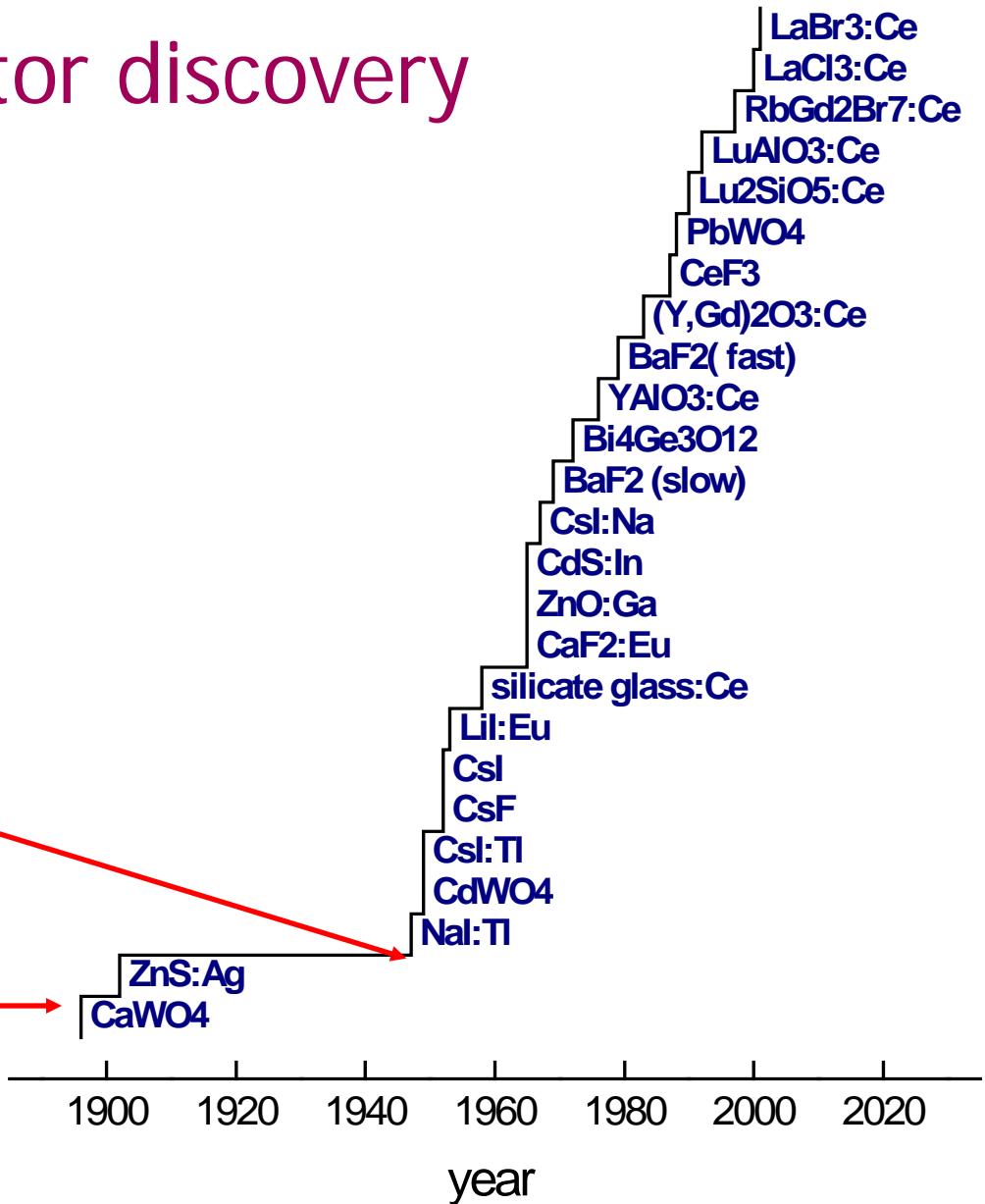


# History of scintillator discovery



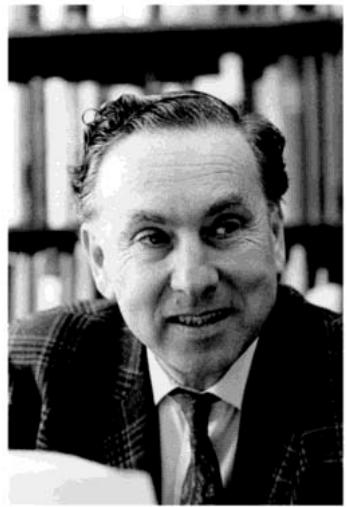
Invention of  
Photomultiplier  
tube

Human eye

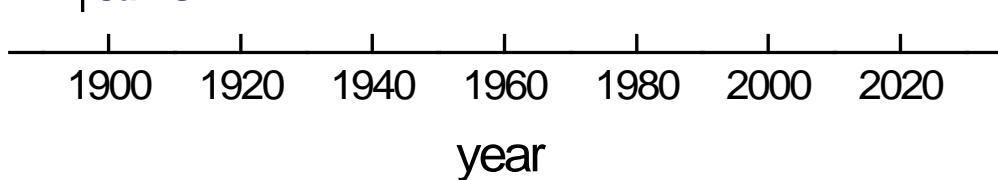


# History of scintillator discovery

TU-Delft discoveries



R. Hofstadter



## Main drivers of scintillation research

# Important gamma ray scintillator parameters

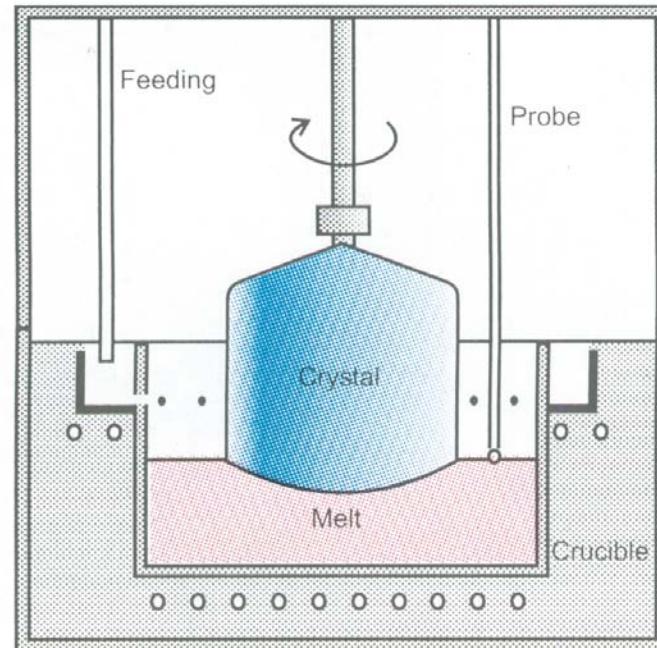
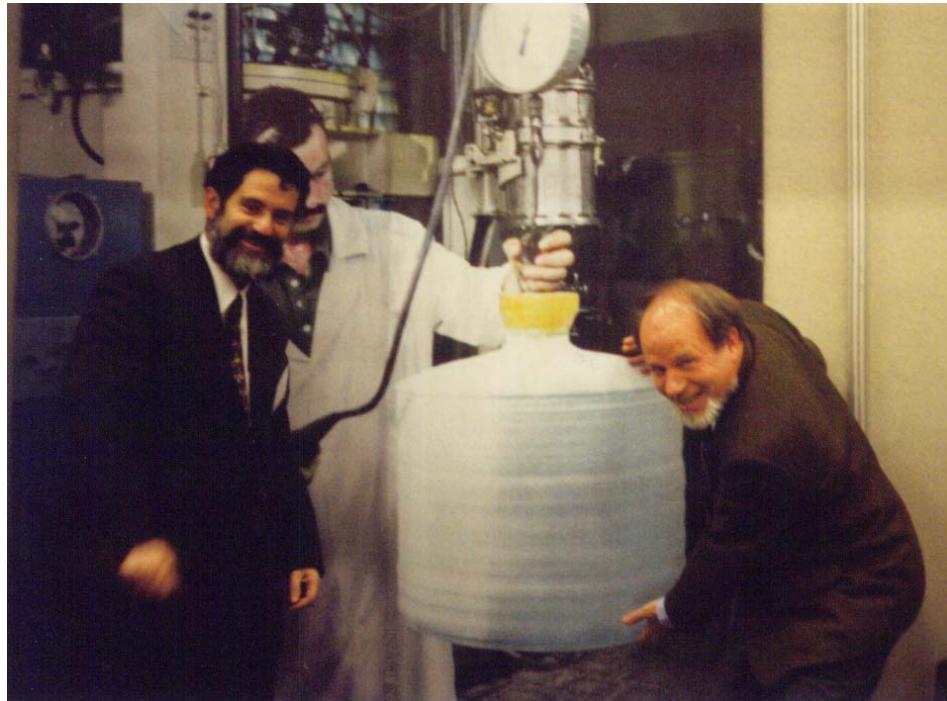
- High light output  $Y$  (photons/MeV)
- fast scintillation speed  $\tau_s$  (ns)
- low energy resolution  $R_{FWHM}$  (%)
- high density for  $\gamma$  detection  $\rho$  (g/cm<sup>3</sup>)
- large size of crystal 10-100-1000 cm<sup>3</sup>
- low cost per cm<sup>3</sup>
- low afterglow (low phosphorescence)
- low background count rate (low intrinsic activity)
  - absence of radioactive isotopes

Importance depends on application

# Properties of inorganic scintillators

scintillator	$\rho$	$n$	$\tau$	$\lambda_{\text{max}}$	$N_{\text{phot}}/\text{MeV}$
NaI (at 80 K)	3.67	1.75	60	303	76000
NaI(Tl)	3.67	1.75	230	415	38000
CsI(Tl)	4.51	1.75	3340	540	65000
BaF <sub>2</sub>	4.89	1.5	630	310	9500
(valence e <sup>-</sup> to core)			0.6	220	1400
Bi <sub>4</sub> Ge <sub>3</sub> O <sub>12</sub> (BGO)	7.13	2.15	300	480	8200
PbWO <sub>4</sub>	8.28	2.20	10	470	100
Lu <sub>2</sub> SiO <sub>5</sub> :Ce (LSO)	7.4	1.8	47	420	25000
YAlO <sub>3</sub> :Ce (YAP)	5.37	1.95	27	370	18000
LaCl <sub>3</sub> :Ce	3.7	1.8	35	350	50000
LaBr <sub>3</sub> :Ce	5.1	2.1	17	380	70000

# The growth of NaI:Tl scintillation crystals



dr. A. Gekhtin and prof. C.W.E. van Eijk with a NaI:Tl ingot

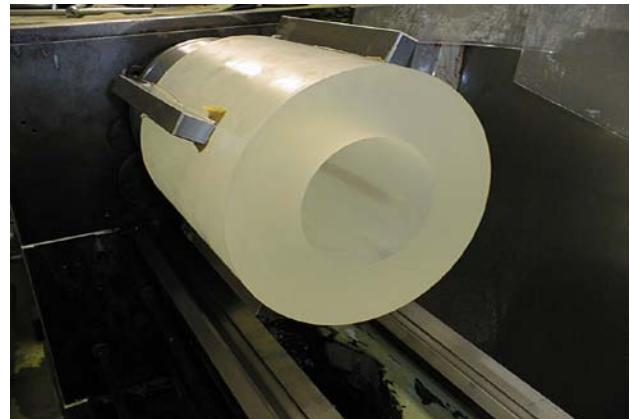
# Examples of bare scintillation crystals



**CsI(Tl)**



**BGO**



**NaI(Tl)**

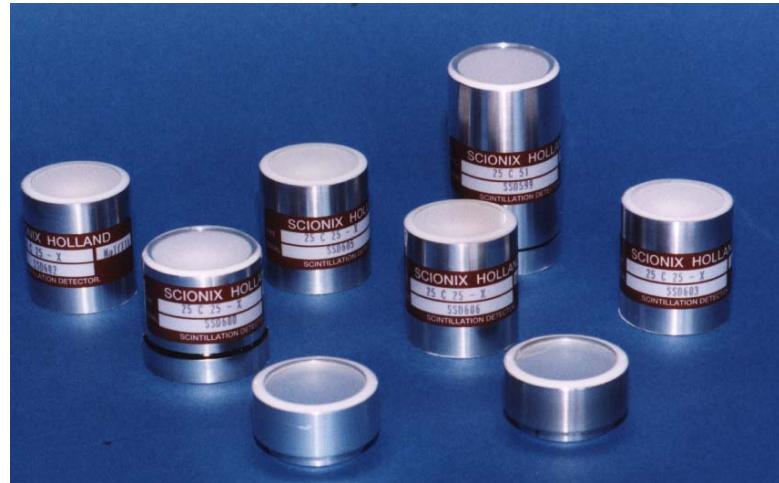


**CdWO<sub>4</sub>**

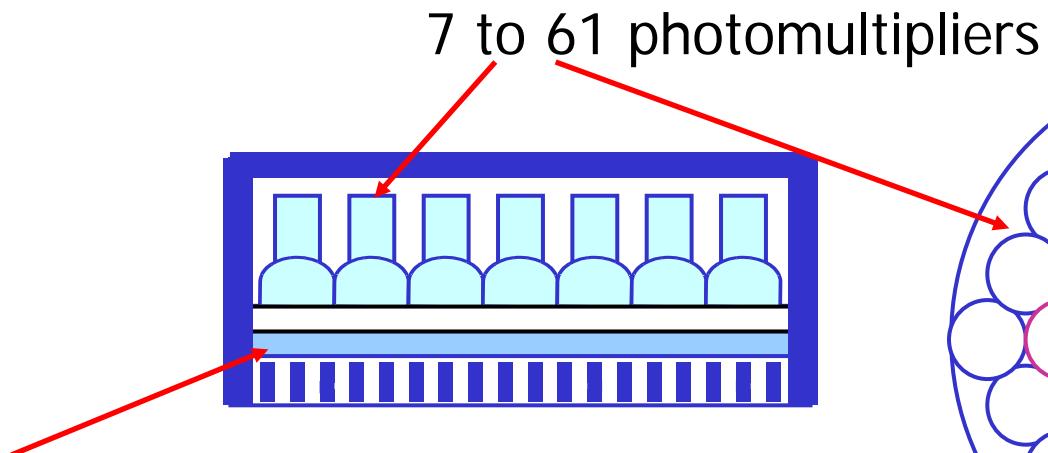
# Examples of packed crystals



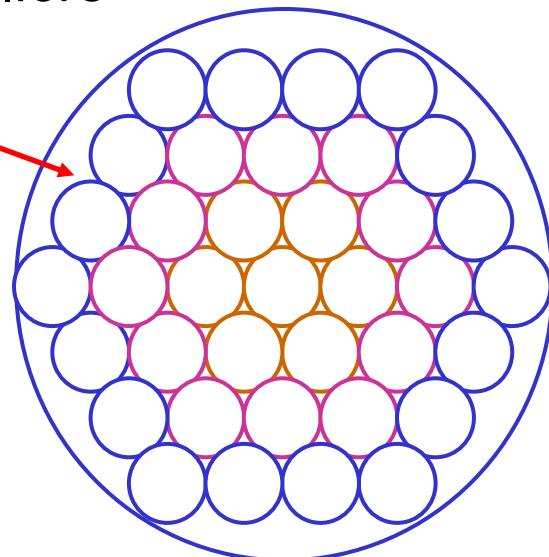
Gamma camera crystal



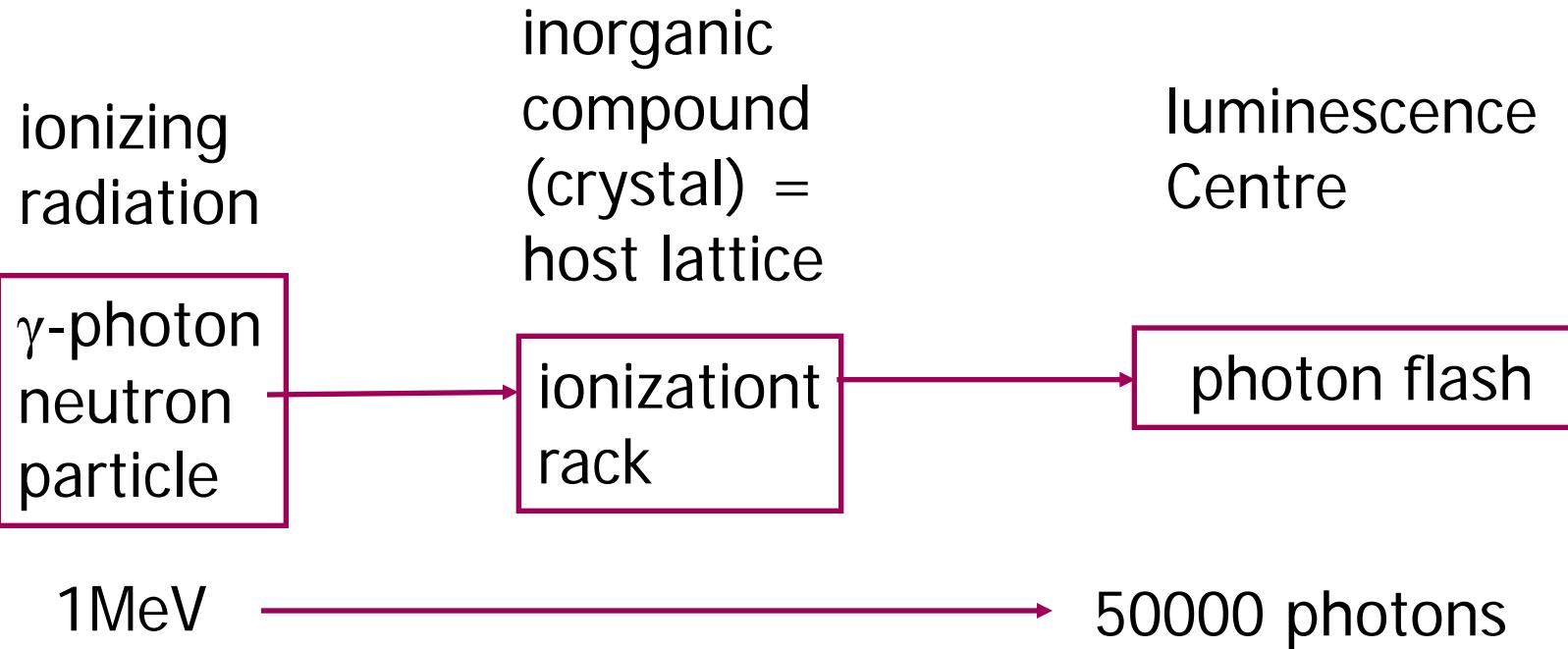
NaI(Tl) crystals



NaI:Ti<sup>+</sup> crystal 20-60 cm diameter and  
2-25 mm thick



# The scintillation process



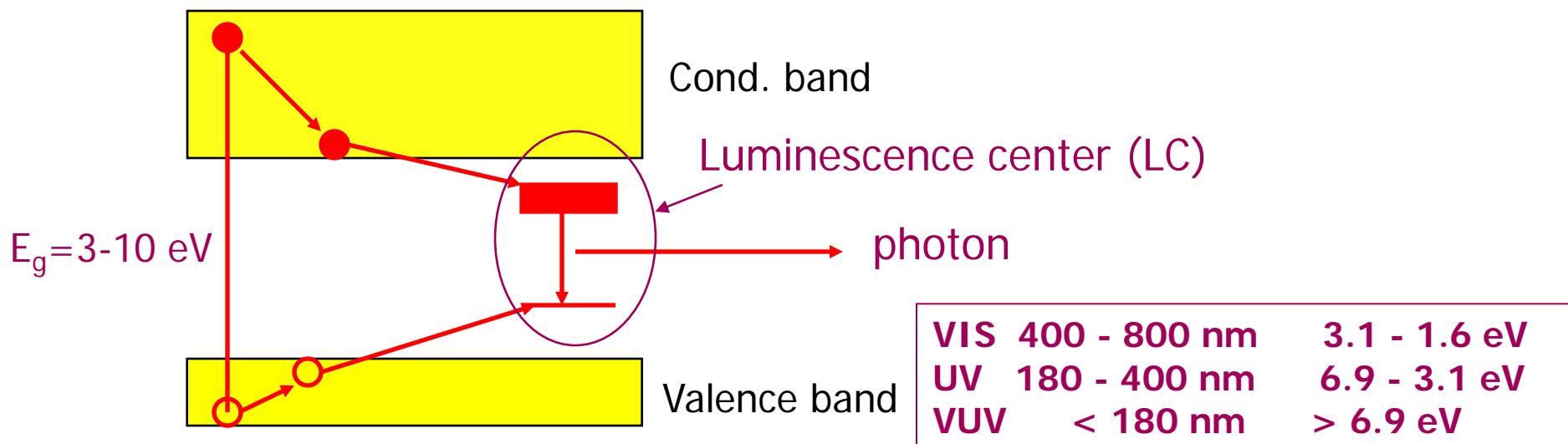
What is happening in between?

A lot!

# The scintillation process

## Three phases

1. The interaction phase + thermalization phase (ps)
2. The charge carrier and energy migration phase (ps-ns-ms)
3. The luminescence phase (ns- $\mu$ s)

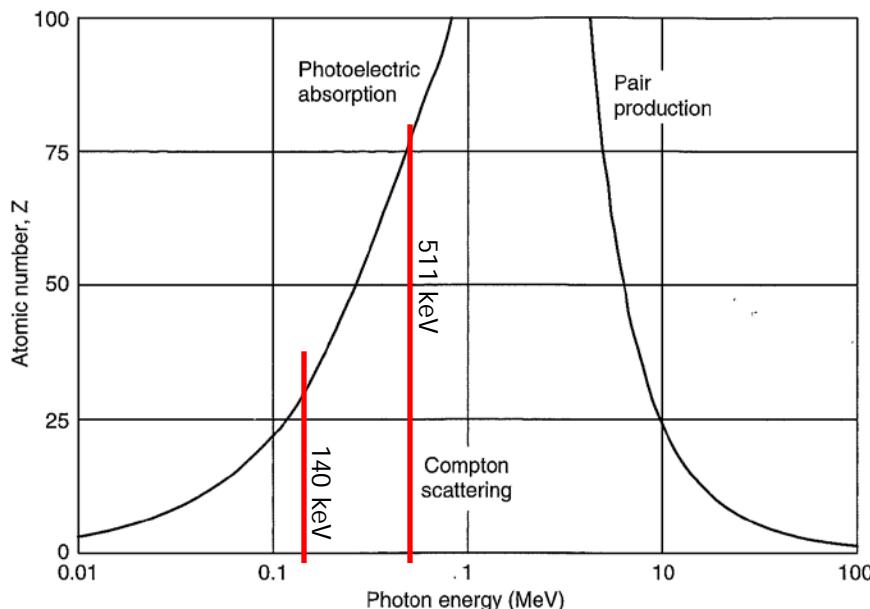


# Phase I: The interaction phase

- Gamma ray interaction → energetic electron

- The photo-electric effect
- The Compton scattering
- Pair production

$Z_{\text{eff}}^{3-4}$  (dominant < 100 keV)  
 $\rho$  (dominant around 1 MeV)  
 $E_\gamma > 1.02 \text{ MeV}$



$Z(I)$	= 53
$Z(Cs)$	= 55
$Z(La)$	= 57
$Z(Gd)$	= 64
$Z(Lu)$	= 71
$Z(Pb)$	= 82
$Z(Bi)$	= 83

# The ionization track in scintillators

- 10 keV – 10 MeV X-ray or gamma-rays produce fast primary electrons

## The Bethe formula

$$-\frac{dE}{dx} = \frac{4\pi e^4 z^2}{m_0 v^2} N Z \left[ \ln \frac{2m_0 v^2}{I} - \ln \left( 1 - \frac{v^2}{c^2} \right) - \frac{v^2}{c^2} \right]$$

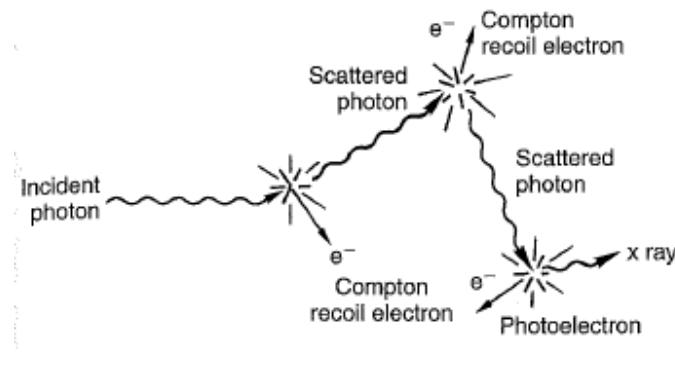
$ez$  = particle charge

$v$  = particle velocity

$N, Z$  = number density and atomic number atoms

$I$  = average ionization potential atoms

Smaller energy → smaller  $v$  → higher  $dE/dx$  → more ionizations/volume



**Figure 6–15.** Multiple interactions of a photon passing through matter. Energy is transferred to electrons in a sequence of photon-energy degrading interactions.

- Multiple interactions
  - One scintillation event can produce separate ionization tracks in the scintillator
  - Everything takes place in the 1-10 ps time scale
  - Separate flashes add to one scintillation flash
- High energy physics GeV-TeV → shower of events

# Energy needed for electron-hole pair generation

Semiconductor	$\rho$ (g/cm <sup>3</sup> )	Z	$E_g$ (eV)	$E_{ion}$ (eV)
Si	2.33	14	1.12	3.6
Ge	5.33	32	0.67	2.9
GaSe	4.55	31, 34	2.03	4.5
InP	4.78	49, 15	1.30	4.2
CdS	4.84	48, 16	2.60	7.3
GaAs	5.32	31, 33	1.43	4.3
InSb	5.77	49, 51	0.20	0.6
CdSe	5.80	48, 34	1.73	5.5
CdTe	6.20	48, 52	1.44	4.7
PbI <sub>2</sub>	6.20	82, 53	2.55	7.7
HgI <sub>2</sub>	6.40	80, 53	2.13	4.2
TlBr	7.56	81, 35	2.68	6.5

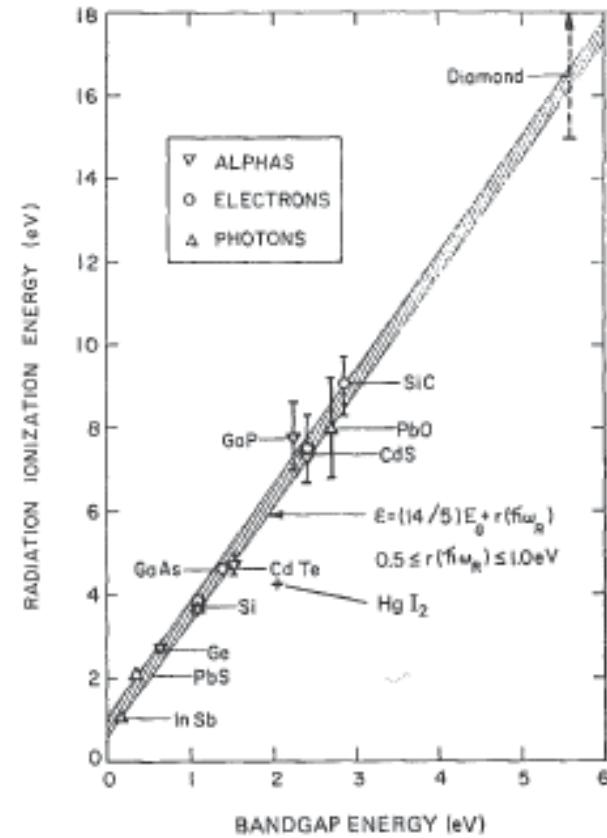
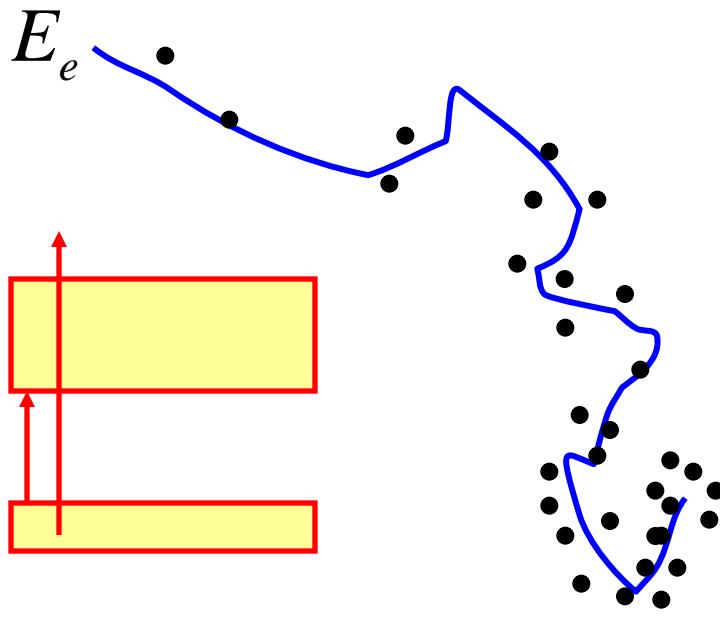


Figure 13.21 The average energy required to form one electron-hole pair ( $E$ ) versus bandgap energy for a number of semiconductor materials. (From Klein.<sup>45</sup>)

Linear relationship between  $E_{ion}$  and  $E_g$  (In figure  $\beta=2.8$ )

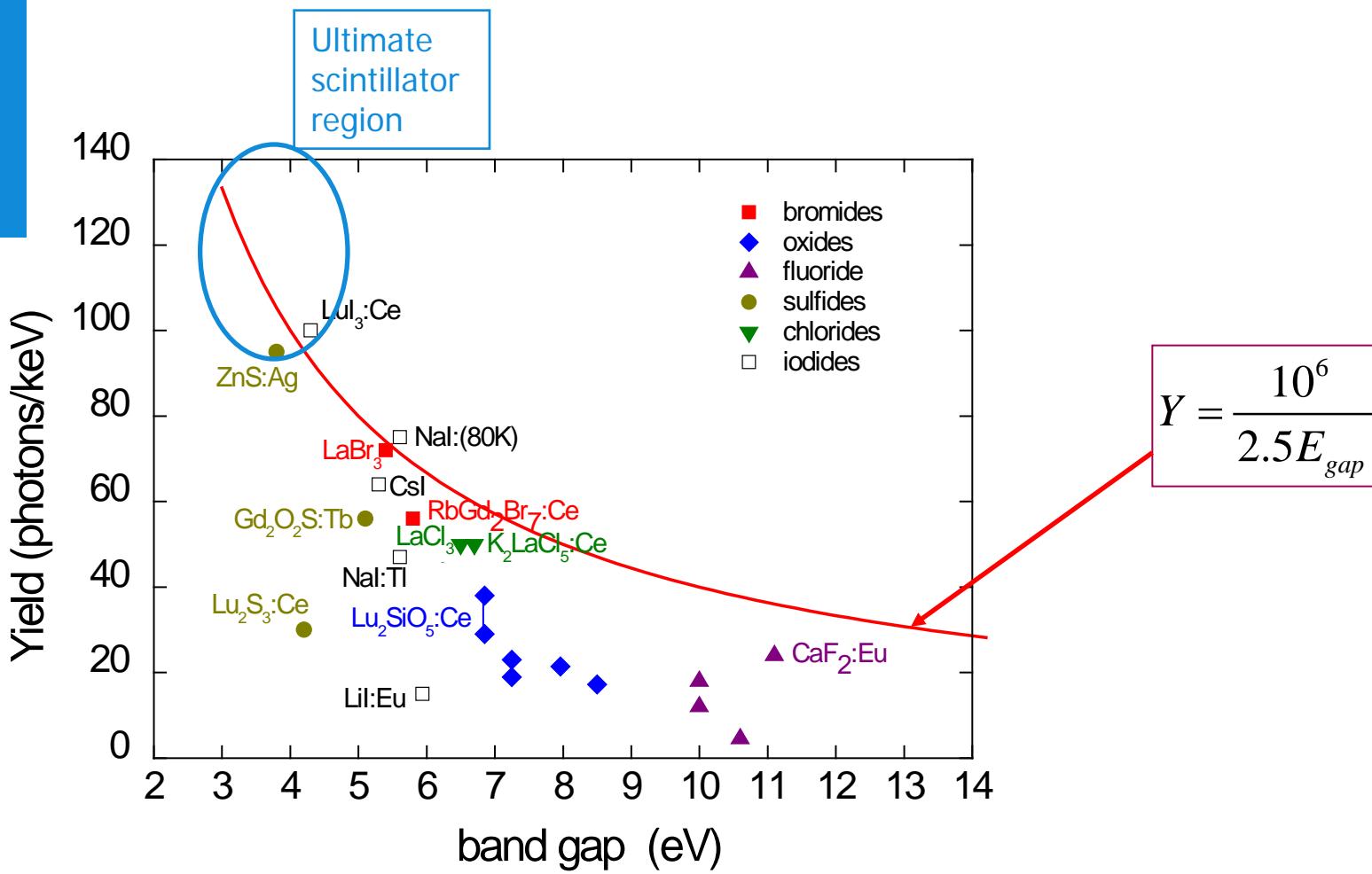
# The ionization track and electron-hole creation



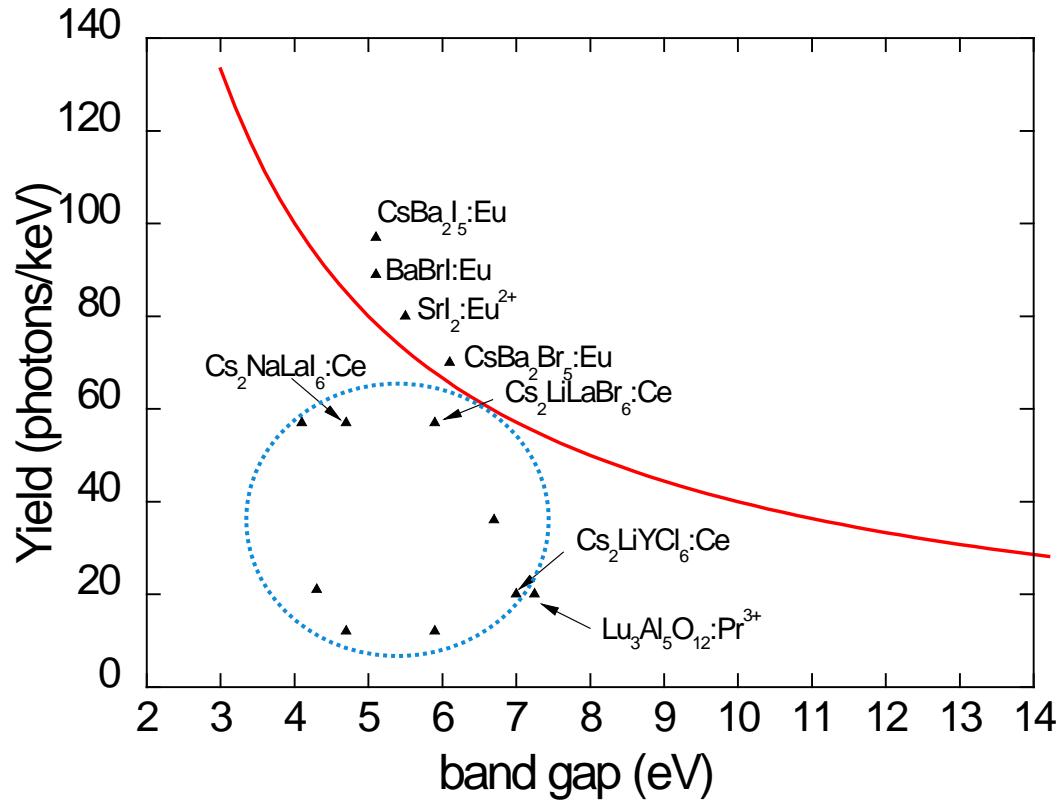
$$N_{eh} = \frac{E_\gamma}{E_{ion}} = \frac{E_\gamma}{\beta E_g}$$
$$\left. \begin{array}{l} \beta \approx 2.5 \\ E_g = 8 \text{ eV} \end{array} \right\} \Rightarrow N_{eh} = 50.000 / \text{MeV}$$
$$N_{ph} = S Q N_{eh}$$

- S is the transfer efficiency from track to luminescence center
- Q is quantum efficiency of luminescence center

# The ideal scintillator has S=Q=1



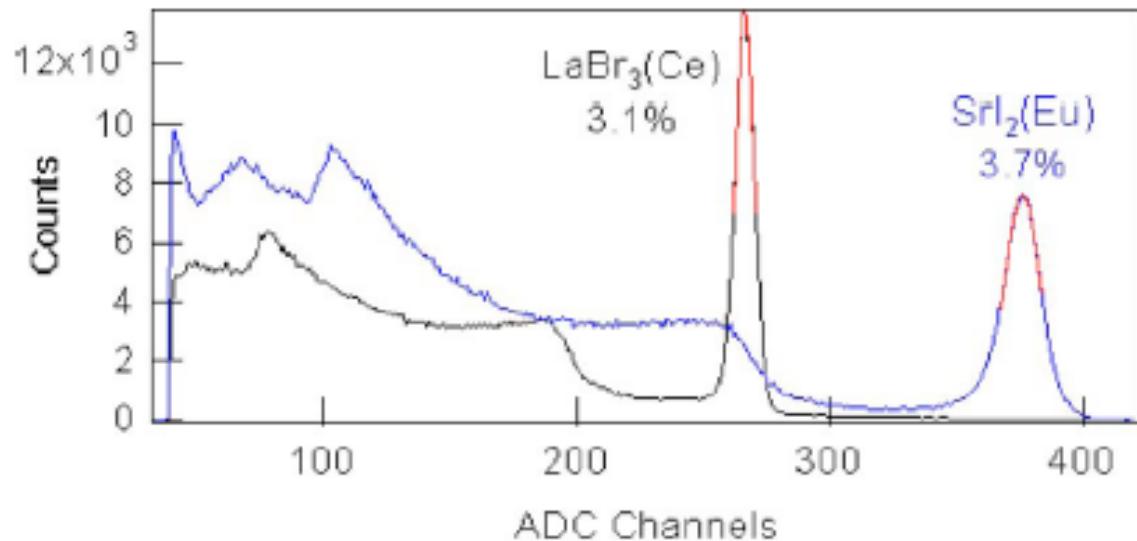
# Latest developments in scintillation research



## Latest materials developments

- Elpasolites (Radiation Monitoring Devices, dr. K. Shah)
- $\text{Eu}^{2+}$  doped halides (University of Berkeley, prof. Derenzo, prof. Moses)
- $\text{Lu}_3\text{Al}_5\text{O}_{12}:\text{Pr}^{3+}$  (Tohoku Univ, prof. Yoshikawa, Inst. of Phys. Prague, dr. M. Nikl)

# Current developments SrI<sub>2</sub>:Eu



Cherepy et al. Appl. Phys. Lett. 92 (2008) 083508

# Current developments Ba<sub>2</sub>CsI<sub>5</sub>:Eu<sup>2+</sup>

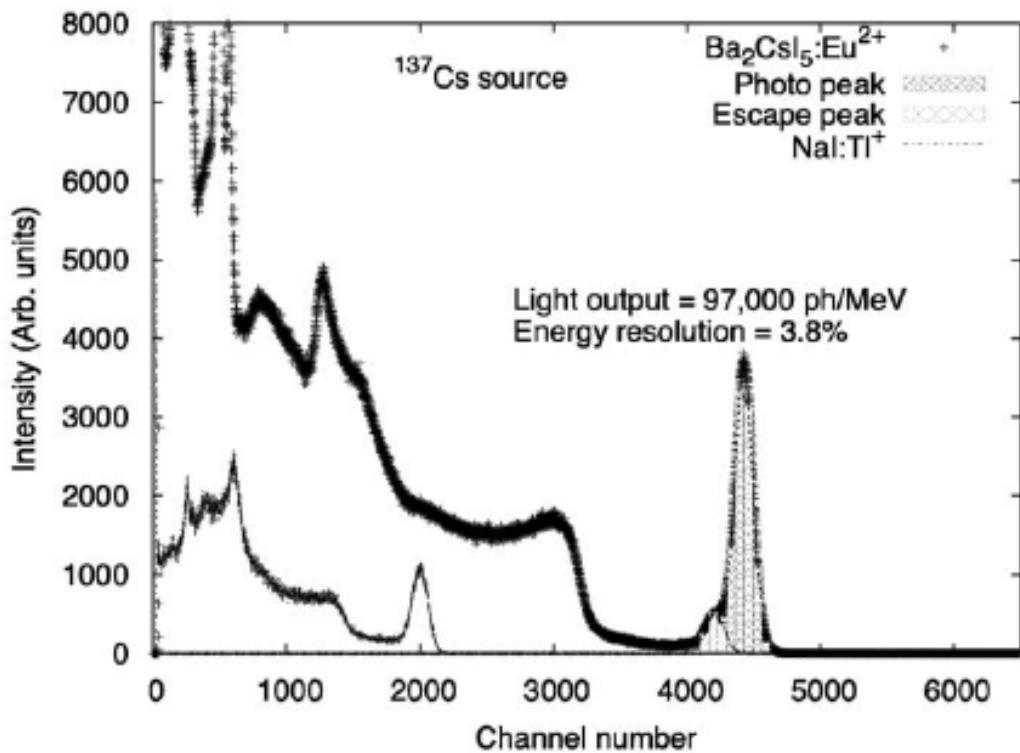


Fig. 3. Pulse height spectra of Ba<sub>2</sub>CsI<sub>5</sub>:Eu<sup>2+</sup> and NaI:Tl measured under <sup>137</sup>Cs gamma-ray excitation.



Fig. 1. A 1 cm in diameter Ba<sub>2</sub>CsI<sub>5</sub>:Eu<sup>2+</sup> shown as-grown in a sealed quartz ampoule.

Bourret-Courchesne et al. Nucl. Instr. Meth. A612 (2009) 138

# Current developments BaBrI<sub>3</sub>:Eu<sup>2+</sup>

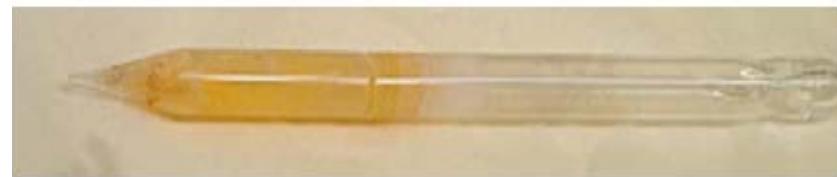
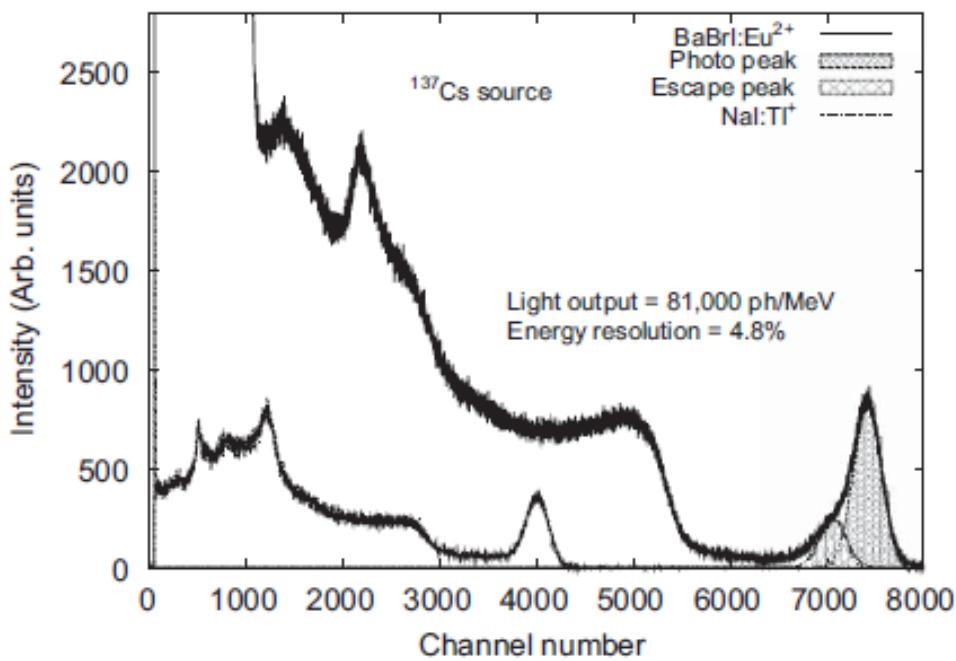


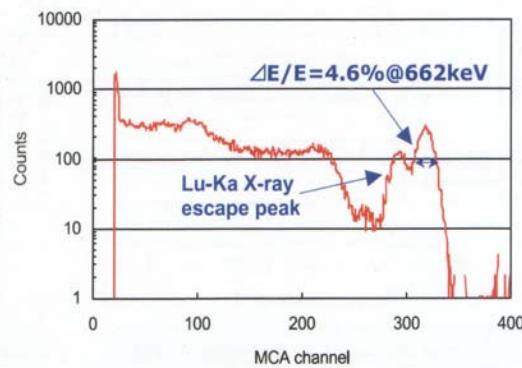
Fig. 1. 1 cm in diameter BaBrI<sub>3</sub>:Eu<sup>2+</sup> shown as-grown in a sealed quartz ampoule.

Fig. 2. Pulse height spectra of BaBrI<sub>3</sub>:Eu<sup>2+</sup> and NaI:Ti measured under <sup>137</sup>Cs gamma-ray excitation.

Bourret-Courchesne et al. Nucl. Instr. Meth. A613 (2010) 95

Recently more materials were discovered. All have their pros and cons.

Scintillators	Pr:LuAG (Lu <sub>3</sub> Al <sub>5</sub> O <sub>12</sub> )
Density (g/cm <sup>3</sup> )	6.7
Light Yield (BGO=100)	330
Decay Time (ns)	<25
Peak emission (nm)	310
Energy Resolution (%@662keV)	5
Hygroscopicity	No
Cleavage	No
Melting Point (°C)	1970



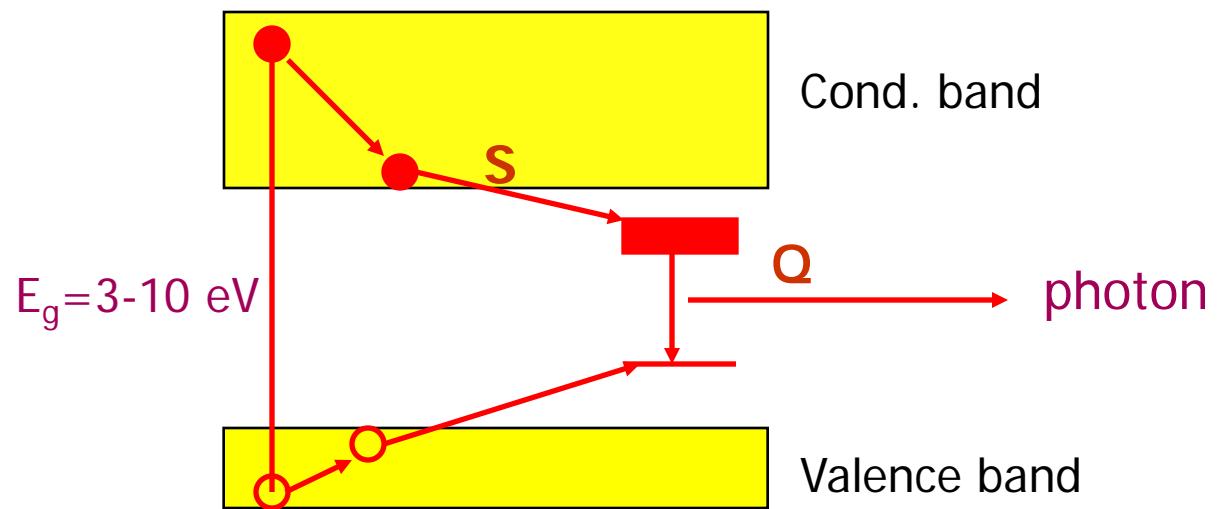
Fast and good energy resolution

But high intrinsic background (Lu)

# The scintillation process; phase III

Three phases

1. The interaction phase + thermalization (ps)
2. The charge carrier and energy migration phase (ns-ms)
3. The luminescence phase (ns- $\mu$ s)

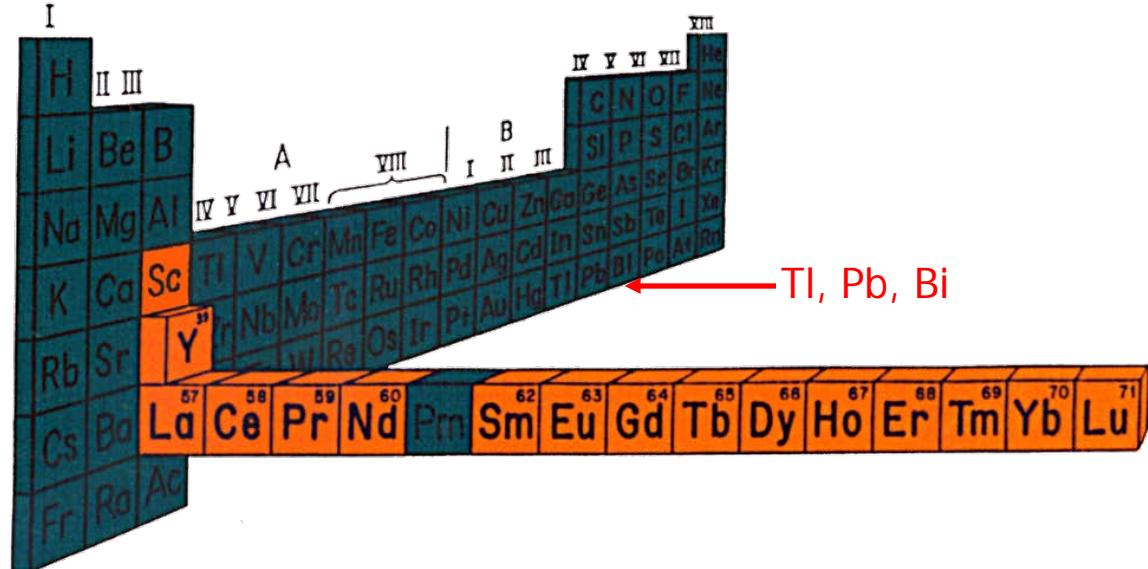


Ideal scintillator  
 $S=Q=1$

# The luminescence phase

The most popular activator ions

- The  $6s^2$ -ions  $Tl^+$ ,  $Pb^{2+}$ ,  $Bi^{3+}$ 
  - $NaI:Tl$ ,  $CsI:Tl$ ,  $Bi_4Ge_3O_{12}$ ,  $PbWO_4$
- The lanthanide ions  $Ce^{3+}$ ,  $Pr^{3+}$ ,  $Eu^{2+}$ 
  - $YAlO_3:Ce$ ,  $Y_3Al_5O_{12}:Ce$ ,  $Lu_2SiO_5:Ce$ ,  $LuAlO_3:Ce$ ,  $LaCl_3:Ce$ ,  $LaBr_3:Ce$ ,  $LiI:Eu$ ,  $Lu_3Al_5O_{12}:Pr^{3+}$



# $6s^2$ -ions $\text{TI}^+$ , $\text{Pb}^{2+}$ , $\text{Bi}^{3+}$

ground state configuration  
excited state configuration

$6s^2$   
 $6s6p$

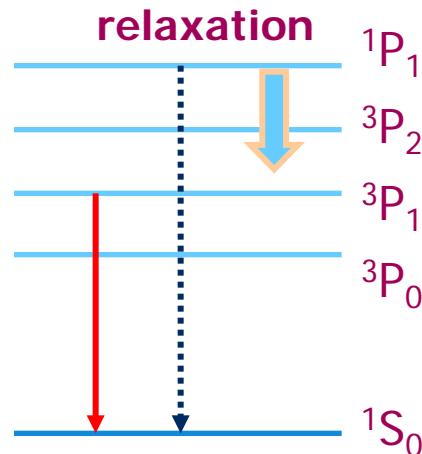
$2S+1L_J$

$^1S_0$	singlet
$^1P_1$	
$^3P_2$	
$^3P_1$	triplet
$^3P_0$	

## Selection rules    Allowed transitions

$\Delta J = 0, \pm 1$  (not  $J = 0 \rightarrow J' = 0$ ),     $\Delta S = 0$

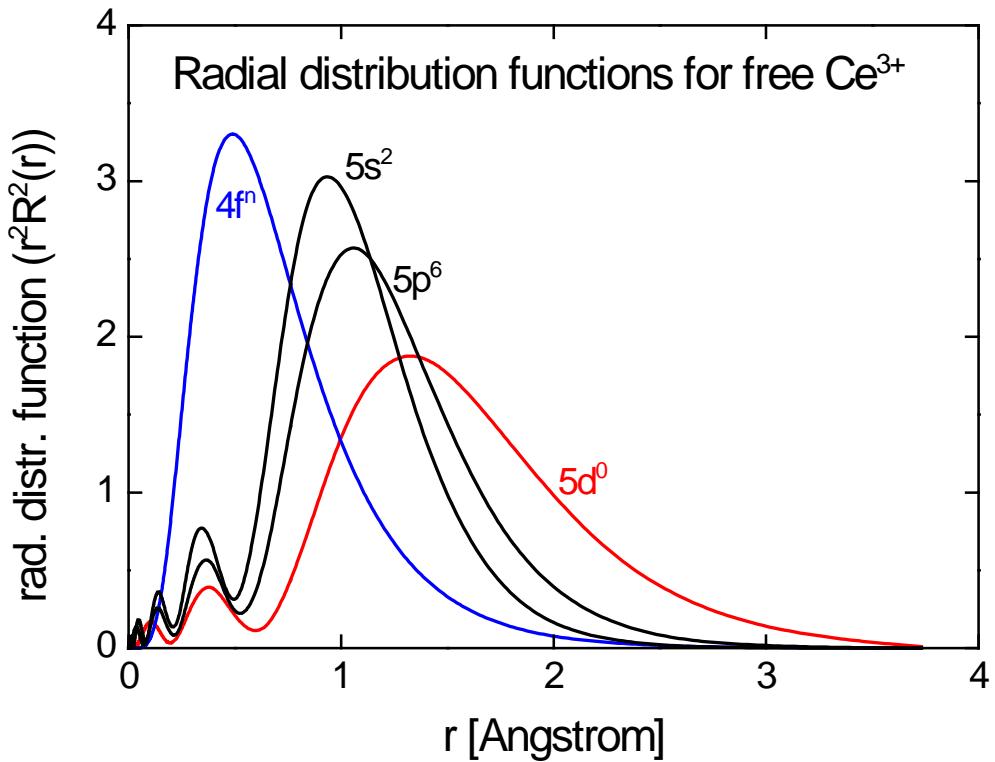
Singlet  $\rightarrow$  singlet      allowed, fast  
Triplet  $\rightarrow$  singlet      forbidden, slow



NaI:Ti     $\tau=230 \text{ ns}$   
CsI:Ti     $\tau=3.3 \mu\text{s}$

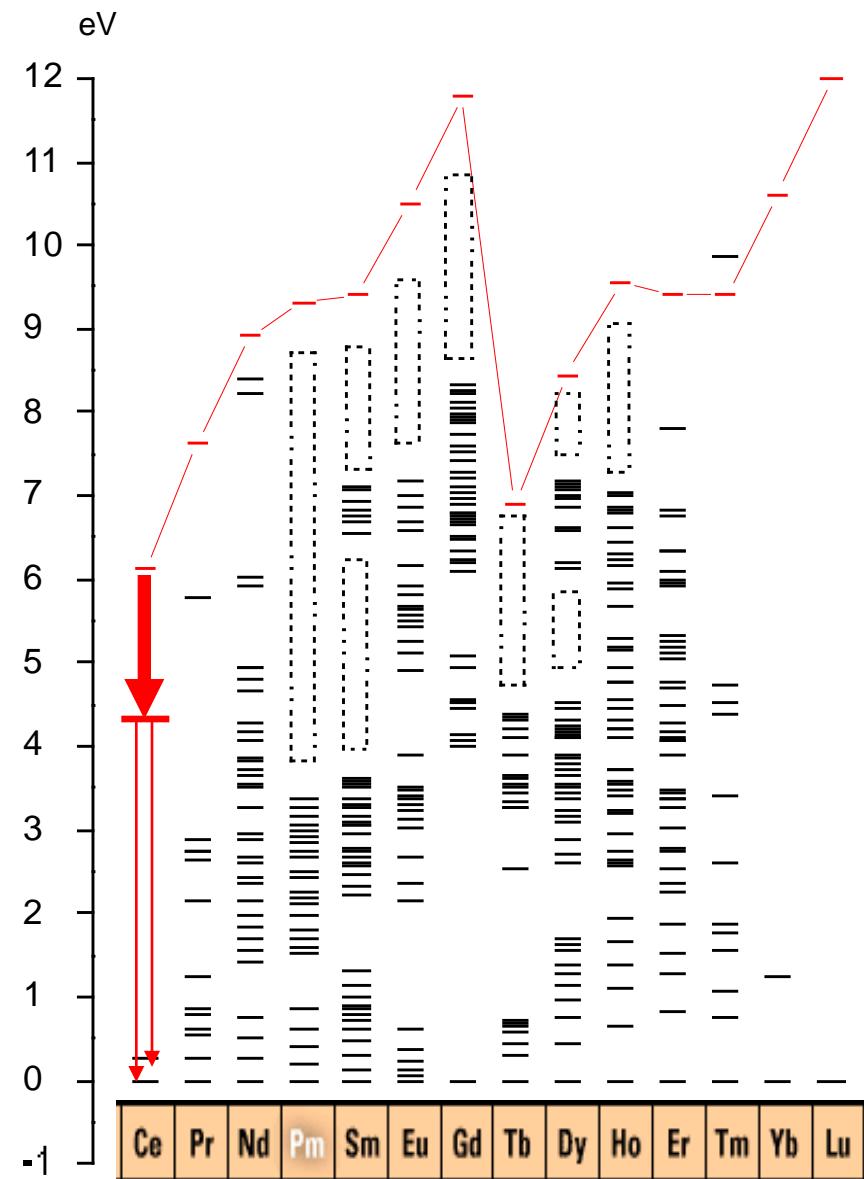
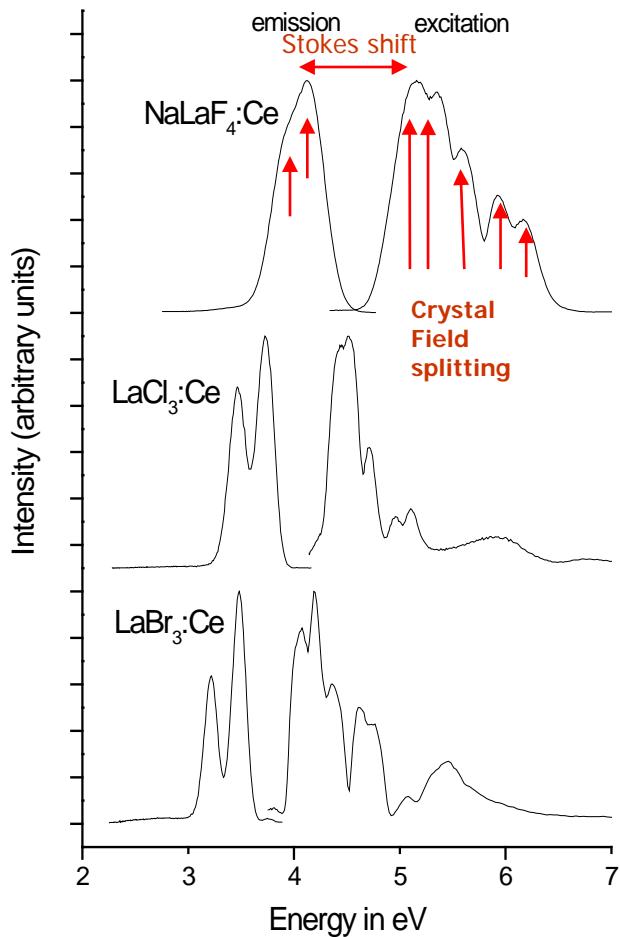
# Lanthanide luminescence

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
----	----	----	----	----	----	----	----	----	----	----	----	----	----	----



- electron configuration  $[\text{Xe}]4\text{f}^n5\text{d}^0$
- 4f electrons are shielded from the crystalline field. Level energies independent on host lattice
- 5d electron strong interaction with crystal field. Level energies depend on host crystal

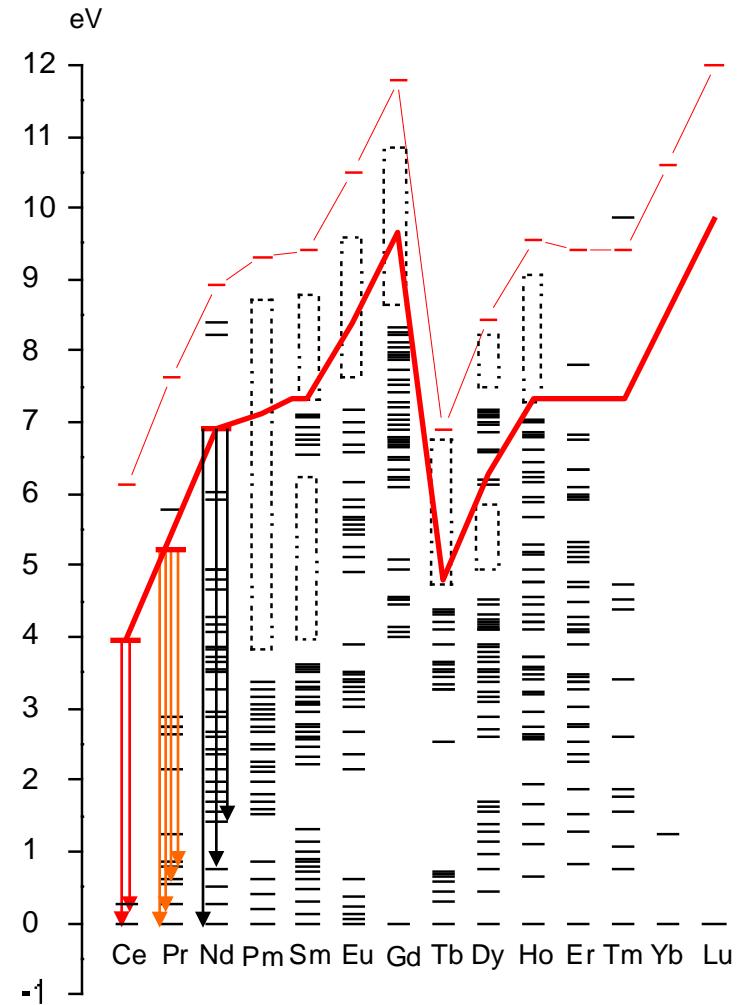
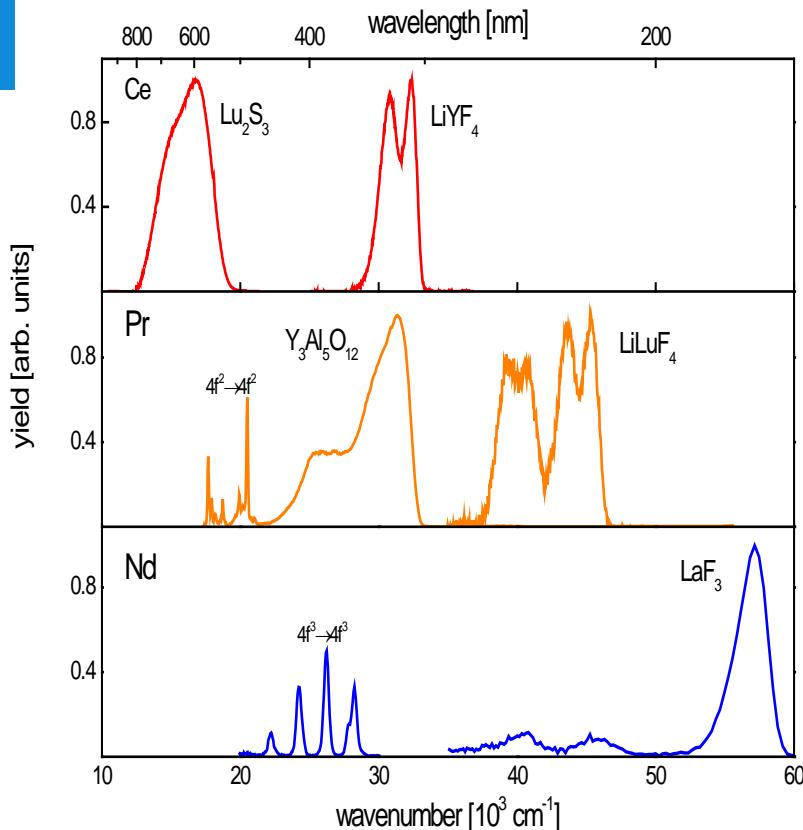
# The Dieke diagram of free ion trivalent lanthanide $4f^n$ and $5d$ levels



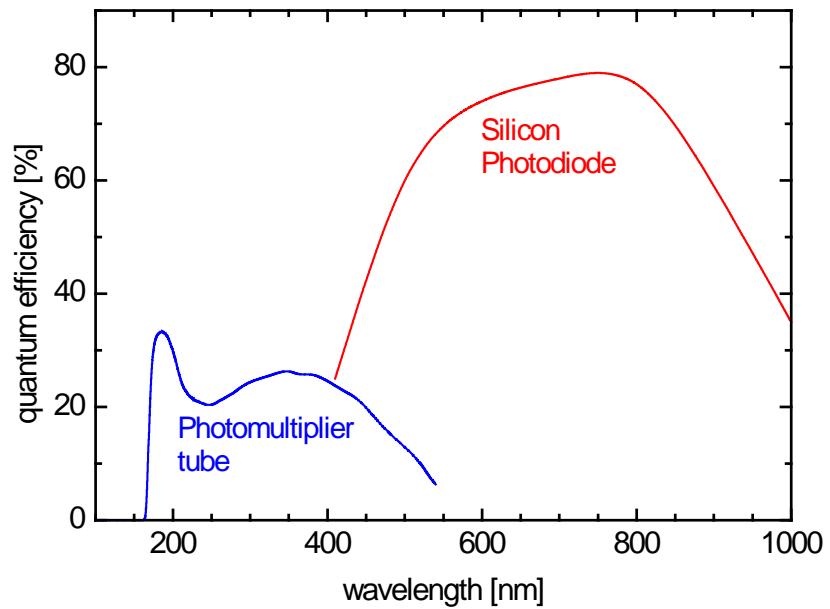
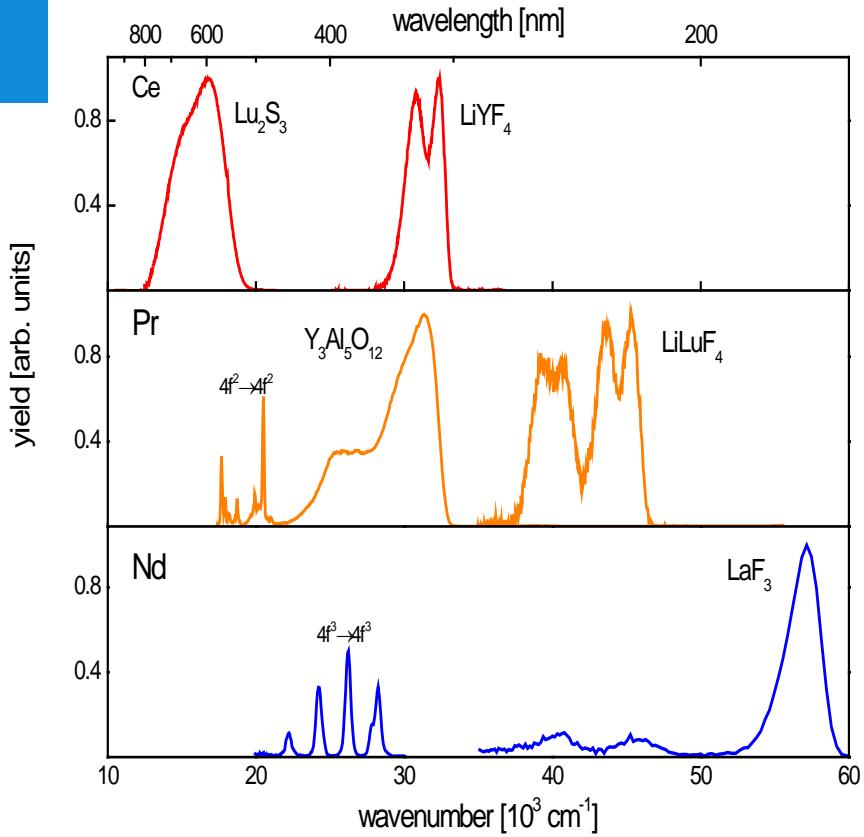
# The emission wavelength

- Electrons either in the ground state or excited state of a luminescence center interact with the surrounding atoms in a compound
  - 4f electrons have negligible interaction
  - 5d electrons (also 6s and 6p electrons of 6s<sup>2</sup>-ions) have strong interaction
- Interaction depends on
  - Crystal structure,
    - coordination number, bond lengths etc.
  - The chemical properties of the anions
    - F<sup>-</sup>, Cl<sup>-</sup>, Br<sup>-</sup>, I<sup>-</sup> and O<sup>2-</sup>, S<sup>2-</sup>
- The emission of Ce<sup>3+</sup> may vary from 300 nm in fluoride compounds up to 600 nm in sulfide compounds

# Characteristic emission spectra Ce, Pr, Nd



# Emission and quantum efficiency



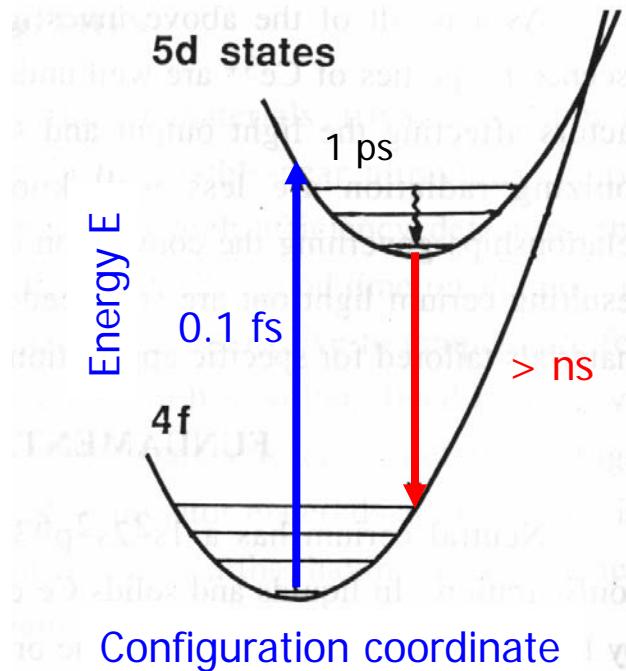
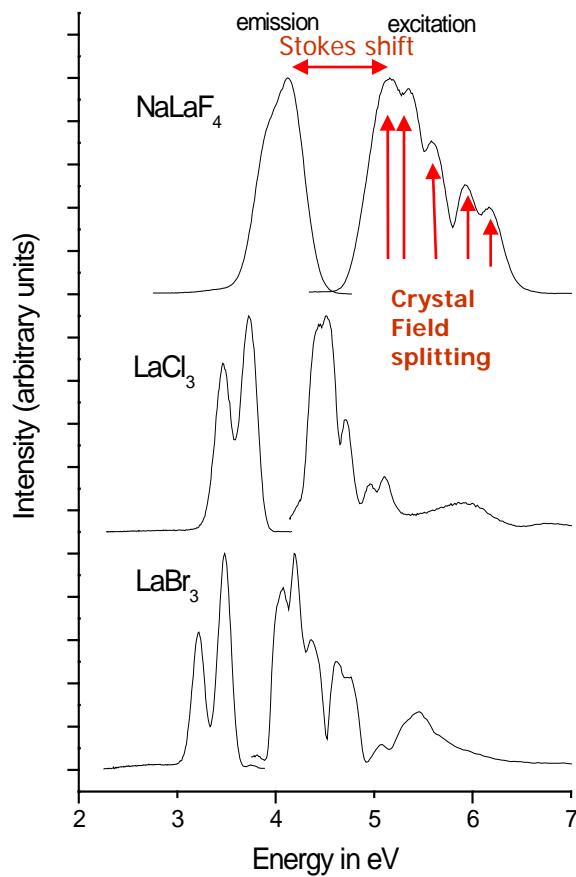
# Ce<sup>3+</sup> doped lanthanide compounds are ideal for gamma ray scintillators

- Fast dipole and spin allowed 5d-4f emission (15-60 ns)
- Emission wavelength matches sensitivity of photomultiplier tubes
- High thermal stability of emission
- Absence of slow 4f-4f emission
- Ce<sup>3+</sup> is a good hole trap which is regarded as a first important step in the scintillation process
- Lanthanide (La, Gd, Lu) based host lattices are dense providing efficient gamma ray detection materials
  - LaBr<sub>3</sub>:Ce, Gd<sub>2</sub>SiO<sub>5</sub>:Ce, Lu<sub>2</sub>SiO<sub>5</sub>:Ce

# Stokes shift and self-absorption

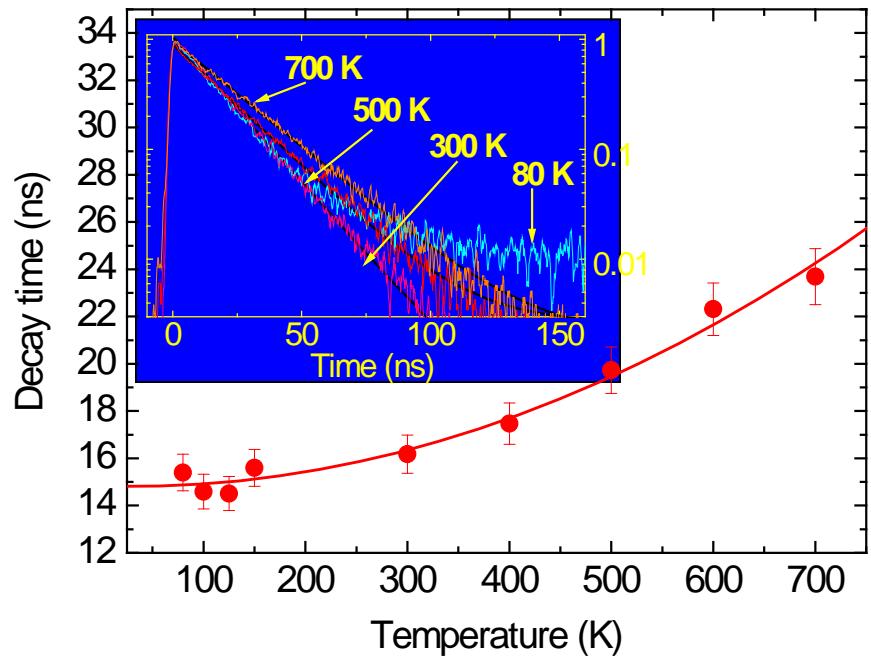
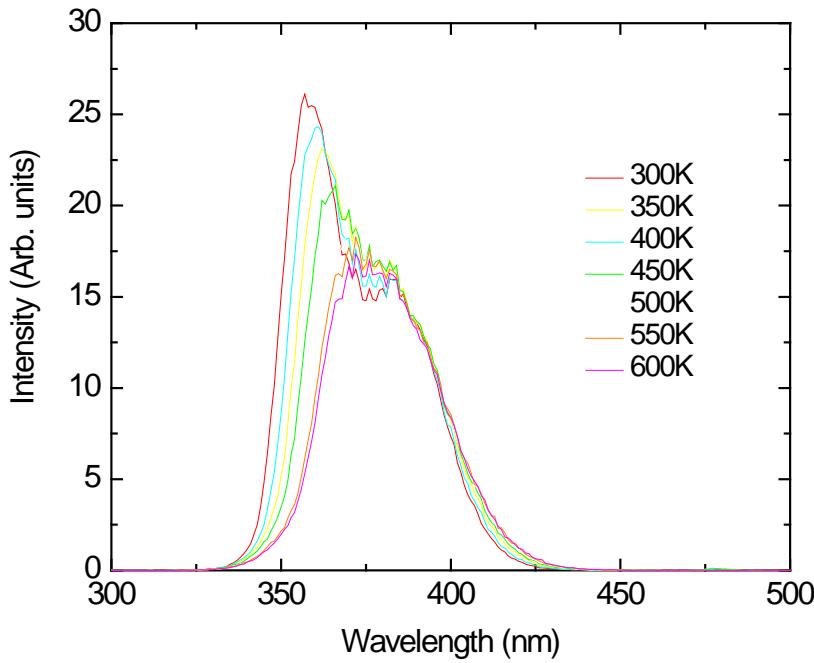
- A scintillator crystal needs to be transparent to its own scintillation light
- A photon emitted by a luminescence center can be re-absorbed by another luminescence center of the same type
- Probability increases with
  - Larger activator concentration
  - Higher temperature (broadening of emission and absorption bands)
  - Larger crystal size
  - Smaller value for the Stokes shift
    - Stokes shift = energy difference between emission and first absorption band of luminescence center

# Stokes shift and self-absorption



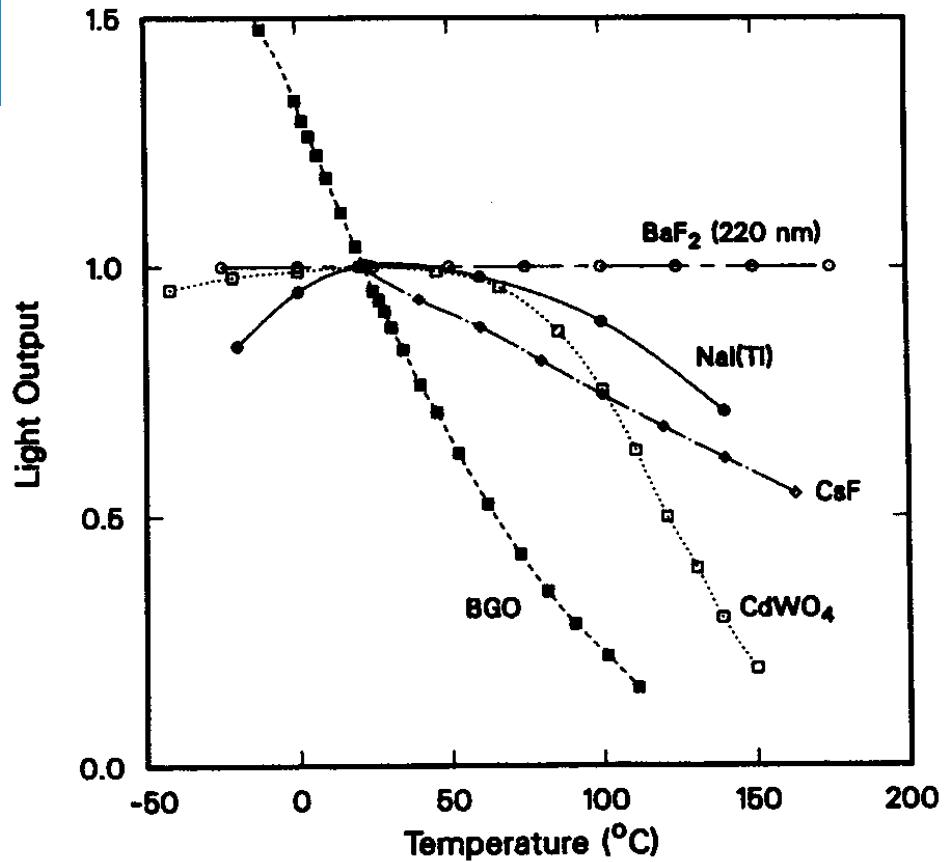
Stokes shift is the energy lost due to lattice relaxation

# Aspects of self-absorption in $\text{LaBr}_3:5\% \text{ Ce}$



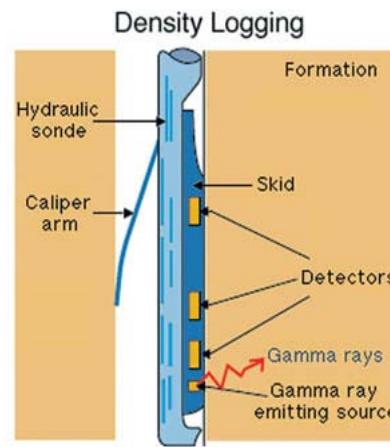
- High energy part of emission re-absorbed by  $\text{Ce}^{3+}$
- Re-emission by  $\text{Ce}^{3+}$ 
  - emission profile changes
  - scintillation decay time lengthening
  - scintillation light losses (because  $Q < 1$ )

# Thermal quenching

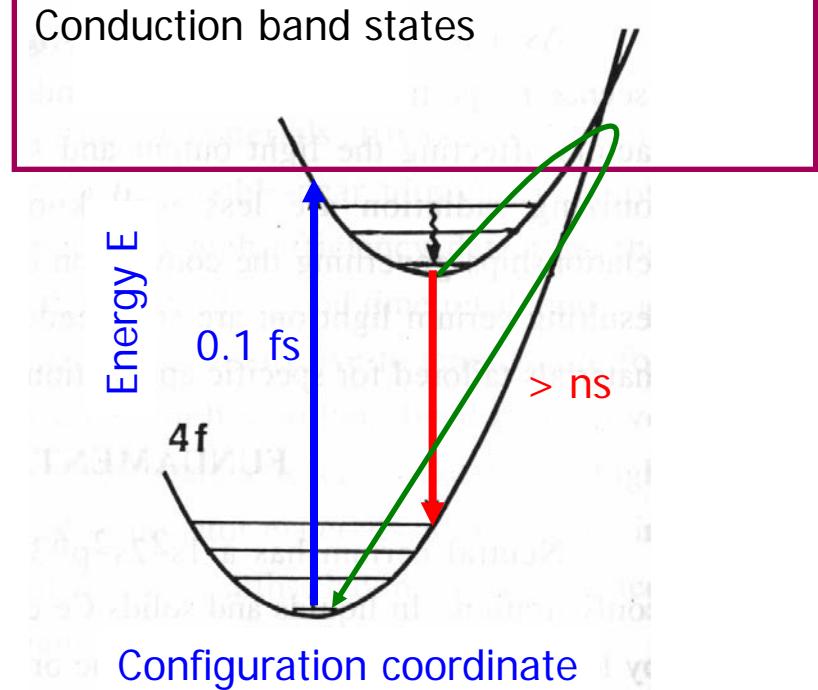
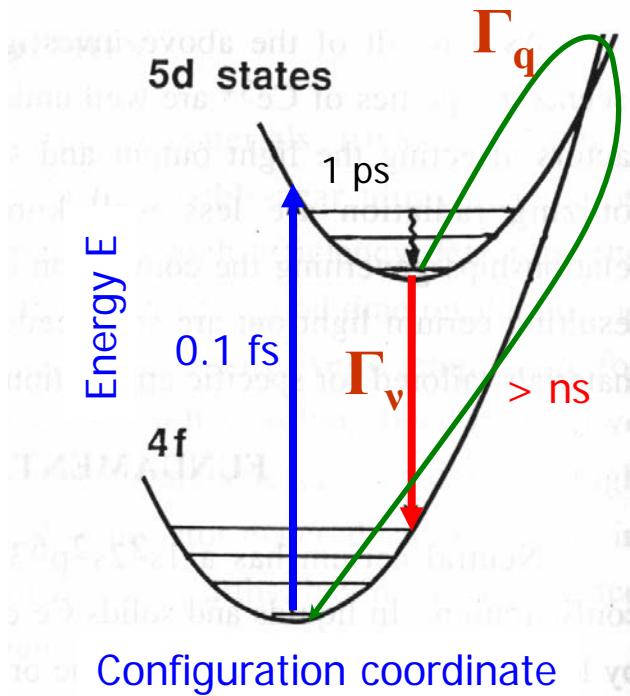


Important for:

- luminescence efficiency Q
- stable scintillator operation against temperature fluctuations
- high temperature applications e.g. oil-well logging



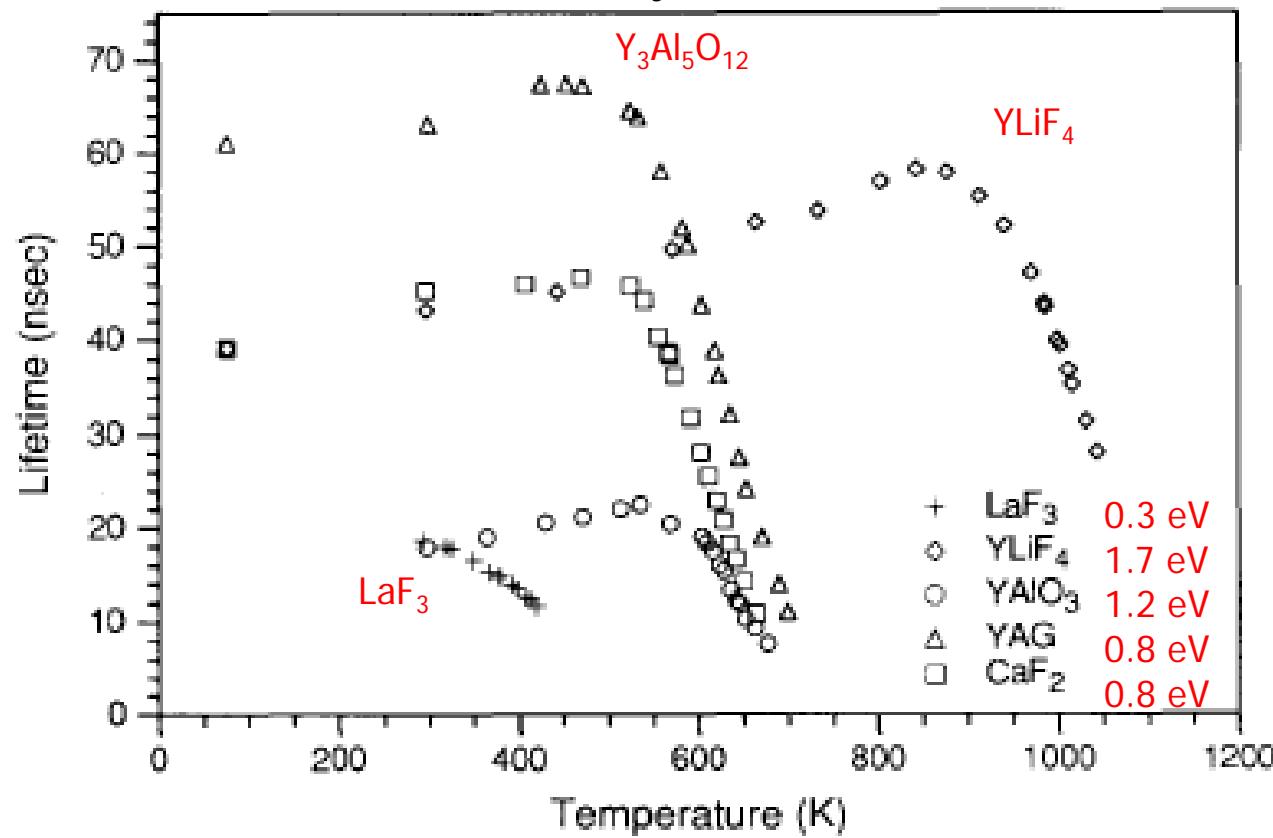
# Thermal quenching of luminescence



$$Q(T) = \frac{\Gamma_v}{\Gamma_v + \Gamma_q} = \frac{\tau(T)}{\tau_v} = \frac{\tau_v^{-1}}{\tau_v^{-1} + s_0 \exp \frac{-\Delta E_q}{k_B T}}$$

# Thermal quenching of $\text{Ce}^{3+}$ emission

Luy *et al.* J. Lumin. 48/49 (1991) 251

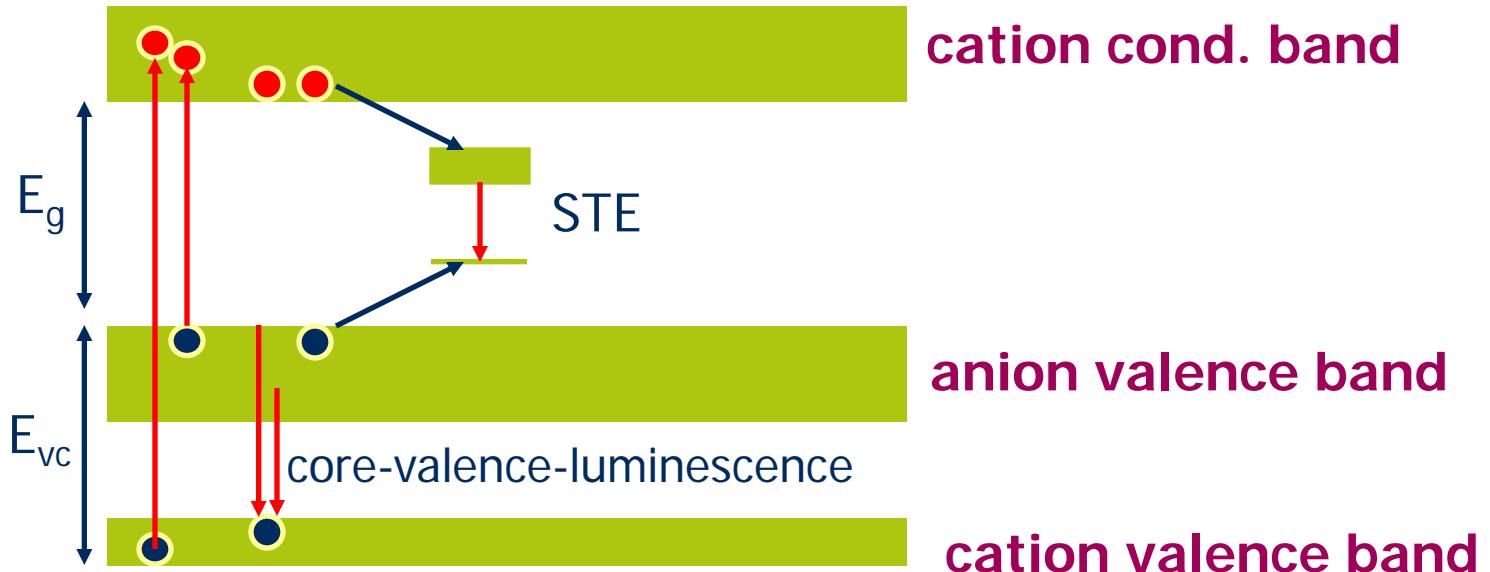


If quenching via CB-states than 5d-level location important

## So far we have treated:

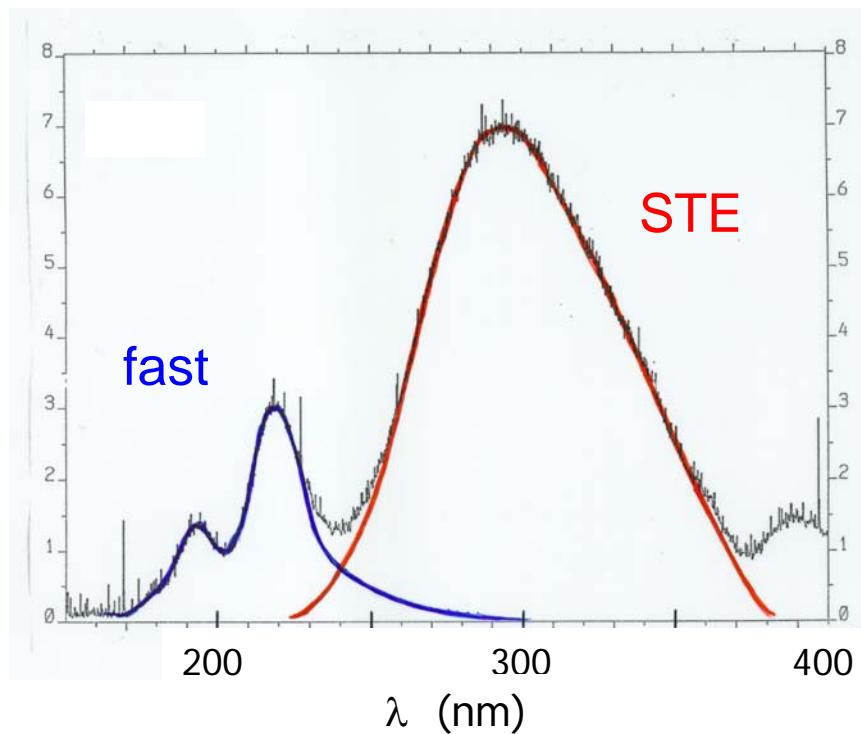
- Phase I: Interaction of a gamma photon with scintillators
  - Creation of the ionization track
  - Relation between number of ionizations and band gap
- Phase III: The luminescence centers
  - s<sup>2</sup>-elements (Tl, Pb, Bi) and lanthanides (Ce<sup>3+</sup>, Pr<sup>3+</sup>)
  - Wavelength of emission, self-absorption, Stokes shift
  - Thermal stability and thermal quenching
- Next, Phase II: scintillation mechanisms
  - Intrinsic scintillators
    - Self trapped exciton and core valence luminescence
  - Activated scintillators
    - energy migration from ionization track to luminescence centers
    - LaCl<sub>3</sub>:Ce and LaBr<sub>3</sub>:Ce as an example

# Core valence luminescence



**condition for CVL:**  $E_{vc} < E_g$

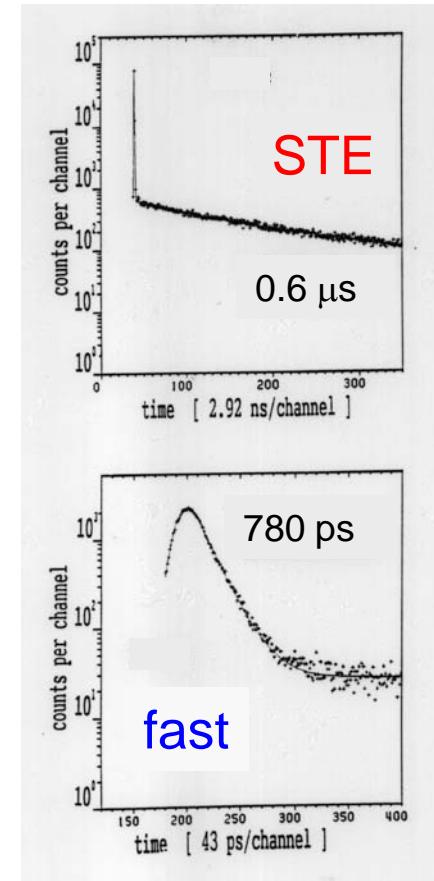
# Self trapped exciton and Core valence luminescence in BaF<sub>2</sub>



CVL: 1400 ph/MeV 800 ps

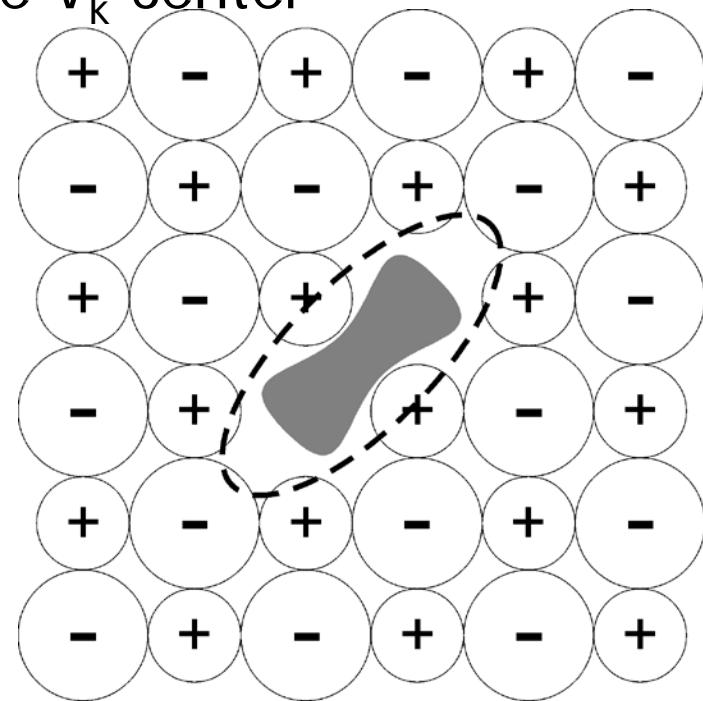
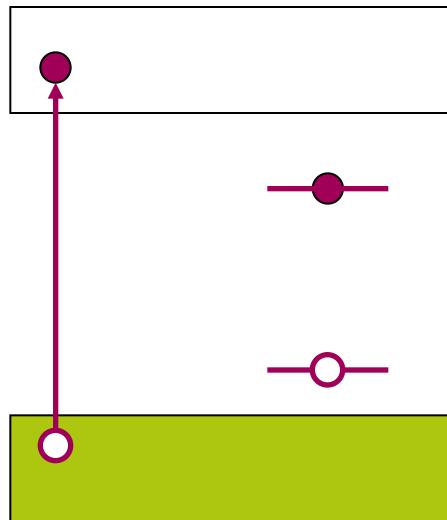
STE: 10<sup>4</sup> ph/MeV 600 ns

N.N. Ershov, N.G. Zakharov, P.A. Rodnyi  
Opt. Spektrosk. 53(1982)89-93

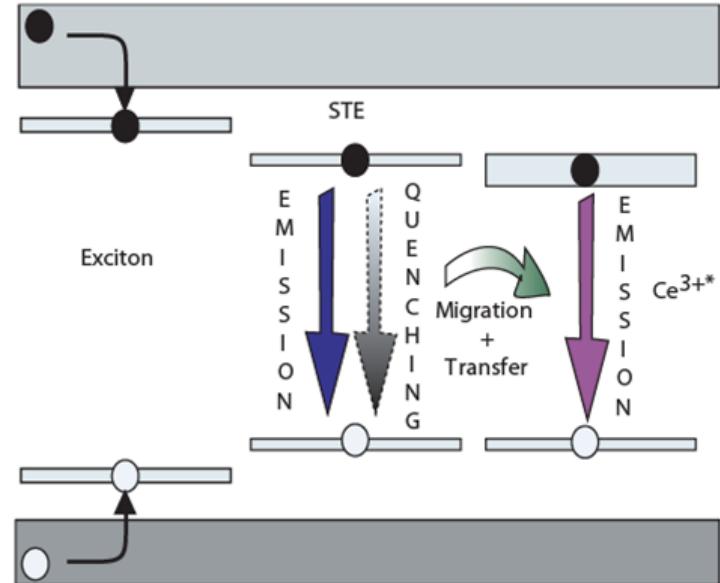
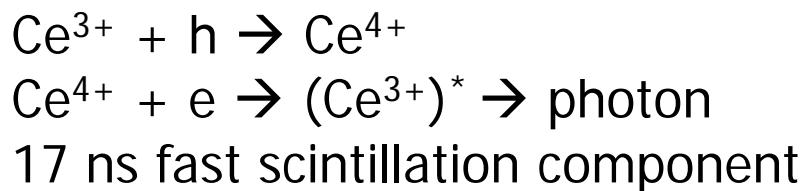
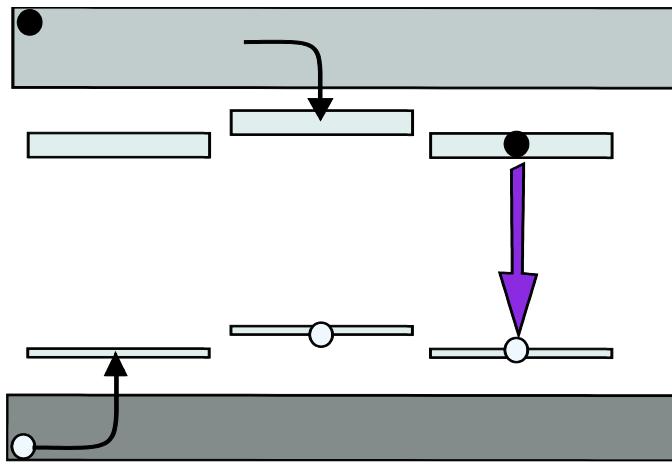


# What is a self trapped exciton?

- hole in valence band is shared between two anions forming an  $X_2^+$  molecule like defect ( $V_k$  center)
- electron in an orbit around the  $V_k$  center



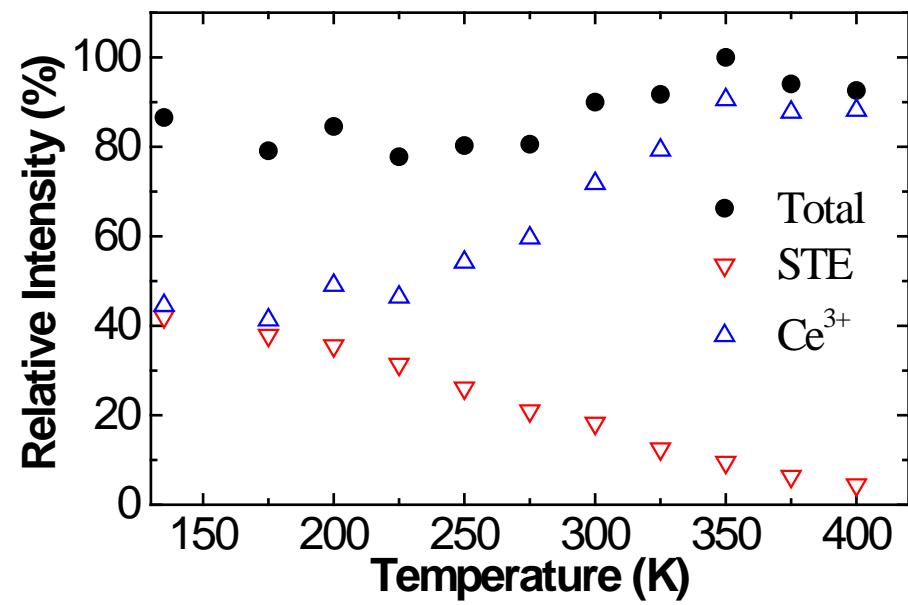
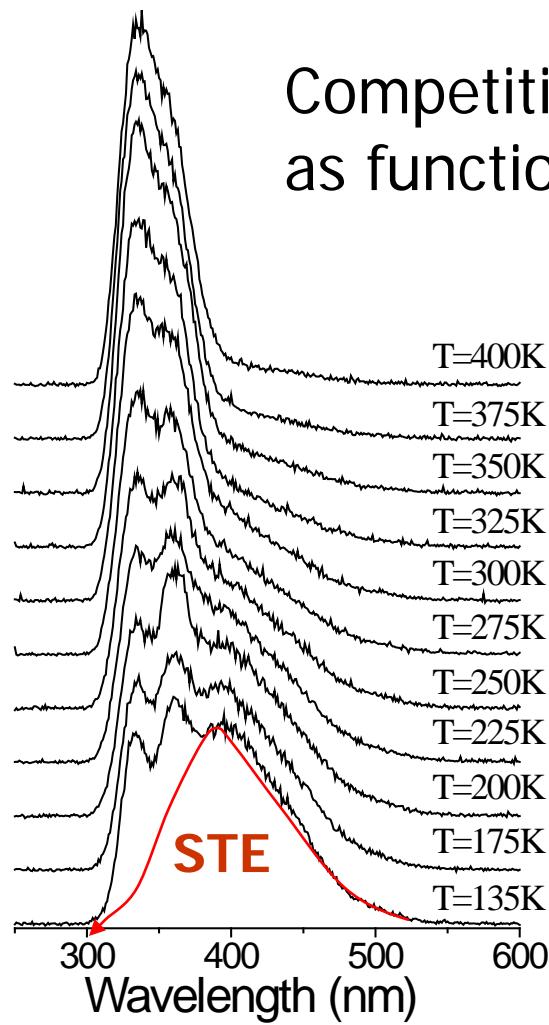
# Scintillation mechanism in $\text{LaCl}_3$ and $\text{LaBr}_3$



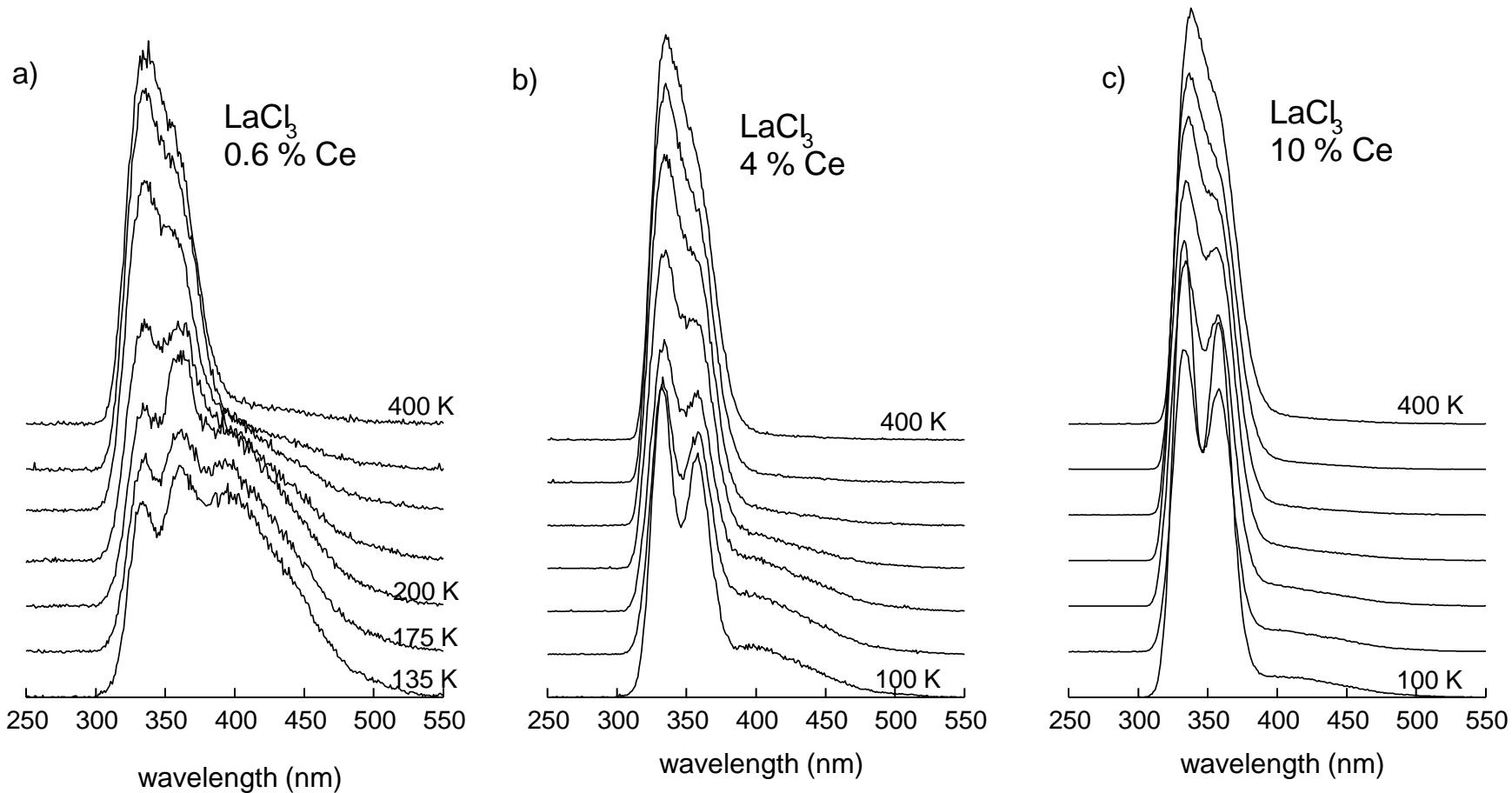
$e + h \rightarrow \text{exciton} \rightarrow \text{STE}$   
 $\text{STE} \rightarrow \text{STE emission (or loss)}$   
μs slow scintillation component  
 $\text{STE} \rightarrow \text{Ce transfer}$

# X-ray excited luminescence $\text{LaCl}_3:0.6\% \text{Ce}^{3+}$

Competition STE emission and Ce emission as function of temperature



# Competition STE emission and Ce emission as function of temperature and concentration



# Summary scintillation processes in La-halides

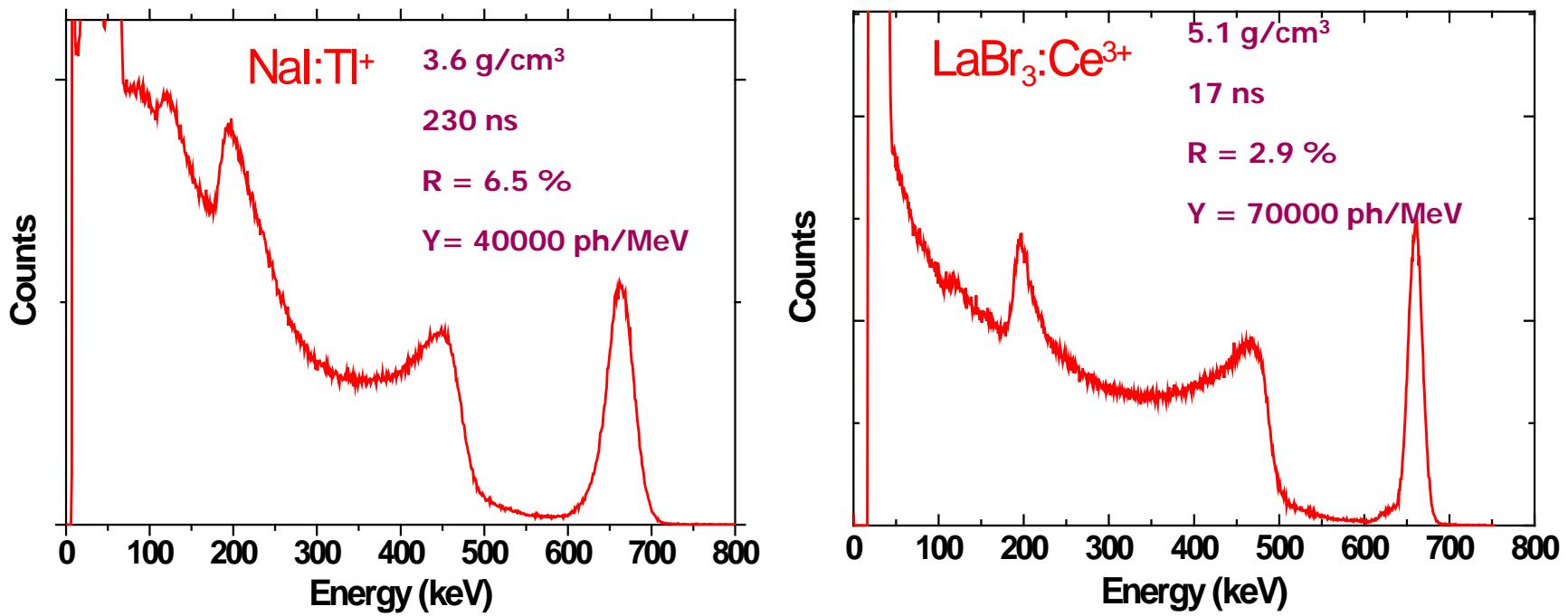
- Direct capture of electron and holes
  - Fast  $\text{Ce}^{3+}$  emission decay component of 16 ns
- Creation of self trapped excitons
  - Slow STE emission
  - Thermally activated STE migration
- Energy transfer from STE to  $\text{Ce}^{3+}$ 
  - Slow  $\text{Ce}^{3+}$  scintillation components

Different competing scintillation processes depending on

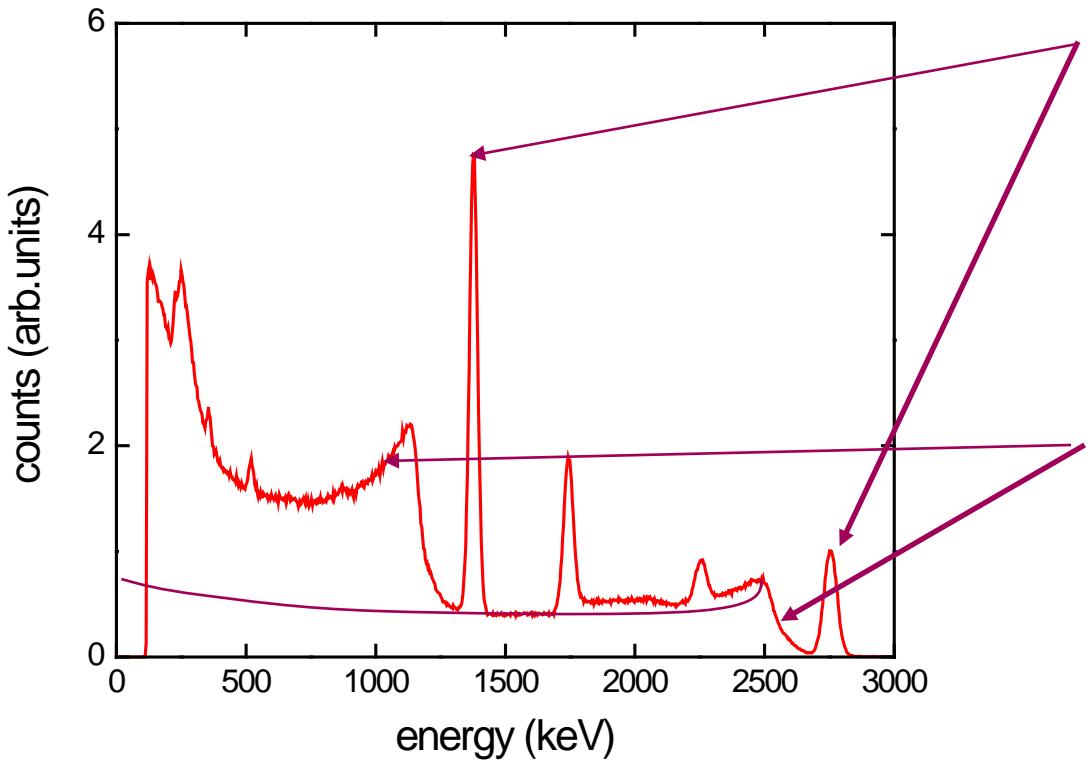
- Ce concentration
- on temperatures

Further reading: G. Bizarri, P. Dorenbos, Phys. Rev. B 75, 184302 2000

# $\text{LaBr}_3:\text{Ce}^{3+}$ ; record low energy resolution at 662 keV



# Anatomy of a pulse height spectrum from $^{24}\text{Na}$ -source measured with $\text{LaBr}_3:5\%\text{Ce}$



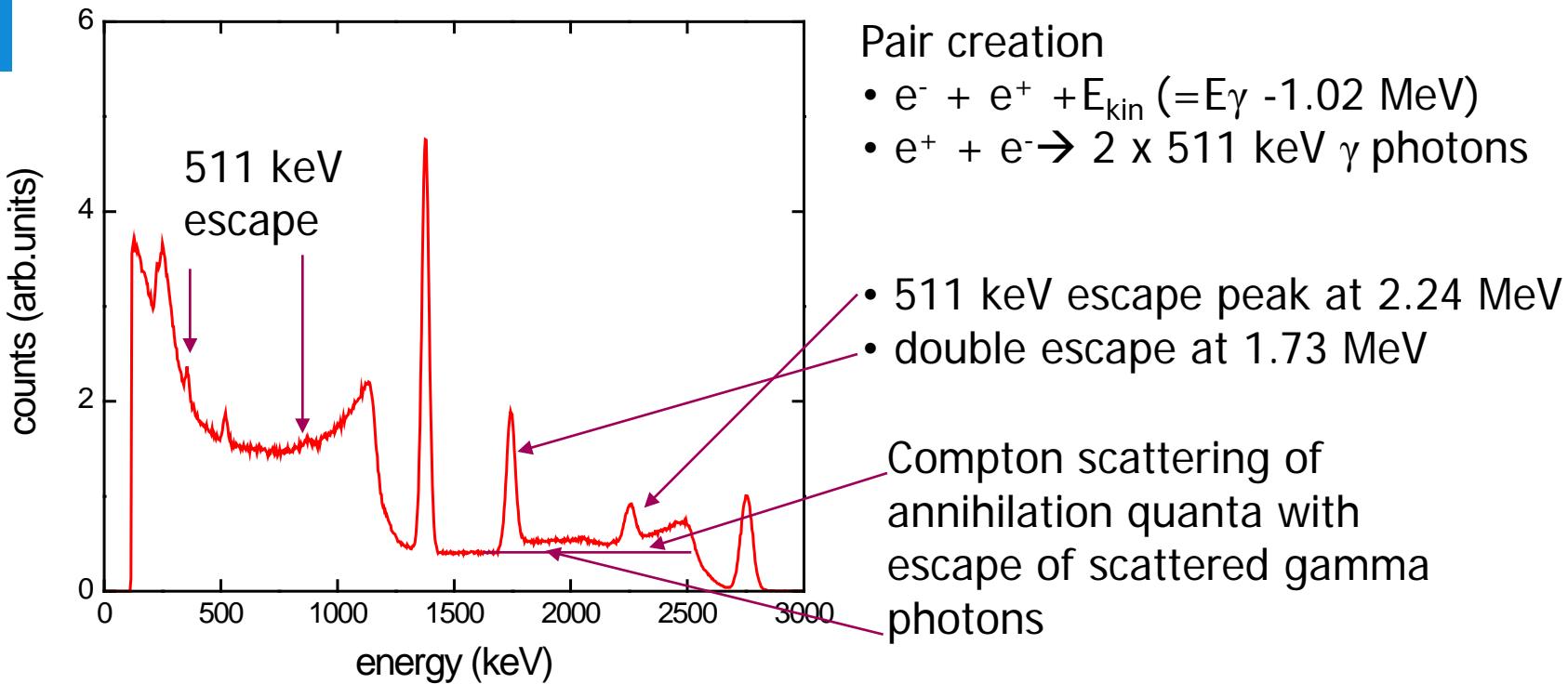
Photoelectric interaction  
→ Total absorption peaks

Compton scattering  
→ Compton edges  
→ Compton continua

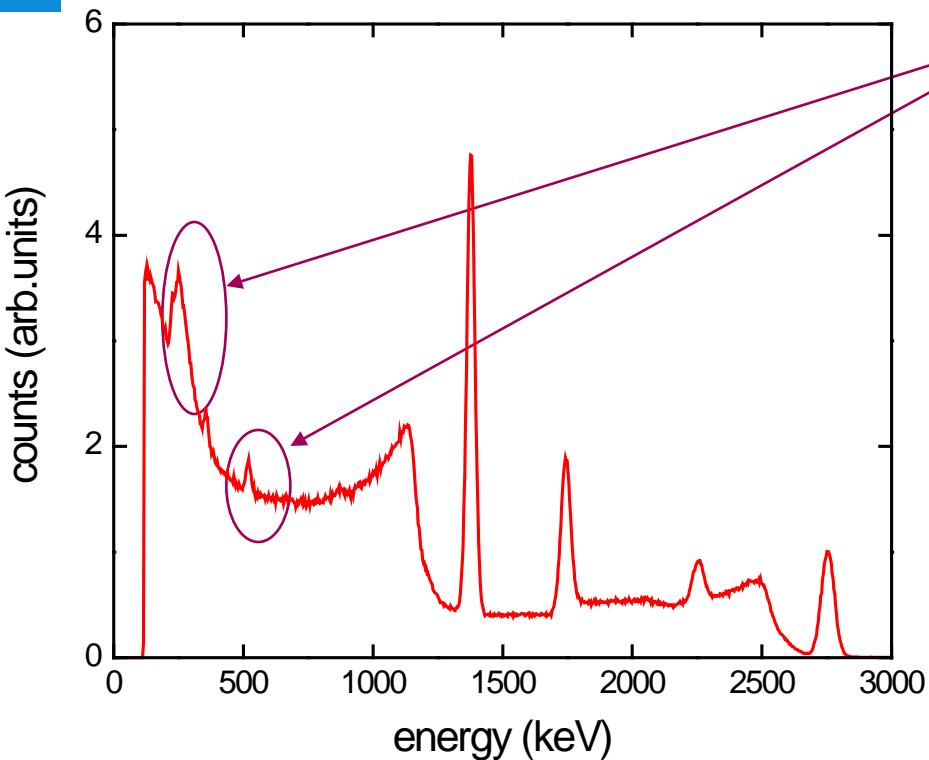
$$E_C = \frac{2E_\gamma^2}{511 + 2E_\gamma} = 2.52 \text{ and } 1.16 \text{ MeV}$$

spectrum due to 1.37 MeV + 2.75 MeV gamma source

# Anatomy of a pulse height spectrum from $^{24}\text{Na}$ -source measured with $\text{LaBr}_3:5\%\text{Ce}$



# Anatomy of a pulse height spectrum from $^{24}\text{Na}$ -source measured with LaBr<sub>3</sub>:5%Ce



Two remaining features at 511 keV and around 250 keV

Back scatter peaks due to gamma rays scattered from materials outside the scintillator

- Compton scattered gamma's
- annihilation gamma 511 keV

# BrilLanCe™ trademark of La-halide scintillators



Discovered by TU-Delft



# $\text{LaCl}_3:\text{Ce}$ crystal growth progress

## From labsize to 4"x6" scintillators crystals

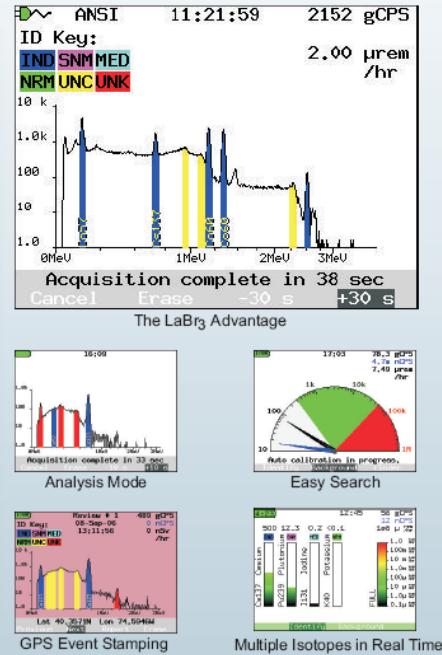


Scintillation Material

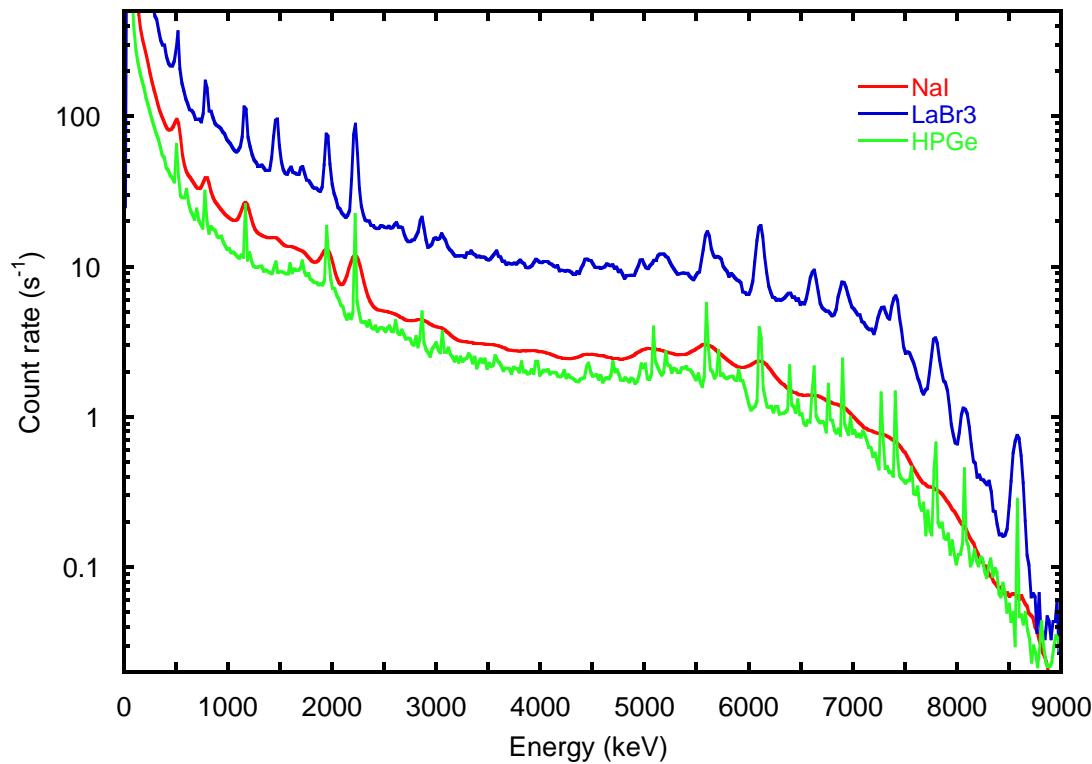


# LaBr<sub>3</sub>:Ce for nuclide identification

## SAM Revealer



# Comparison NaI:Tl, LaBr<sub>3</sub>:Ce, HP-Ge



Real energy [keV]	Element
511	annihilation peak
518	Cl
583.1	208Tl
	Ge
	Ge
788	Cl
911.1	228Ac
1165	Cl
1201	H d
1460	40K decay
1600	Cl f
1686	Nas
1712	H s
1778	Al f
1953	Cl d
1957	Cl f
2223	H
2611	208Tl
2622	Na d
2676	Cl f
2843	Si d
2864	Cl f
2977	F s
2997	Cl f
3017	F f
3062	Cl f

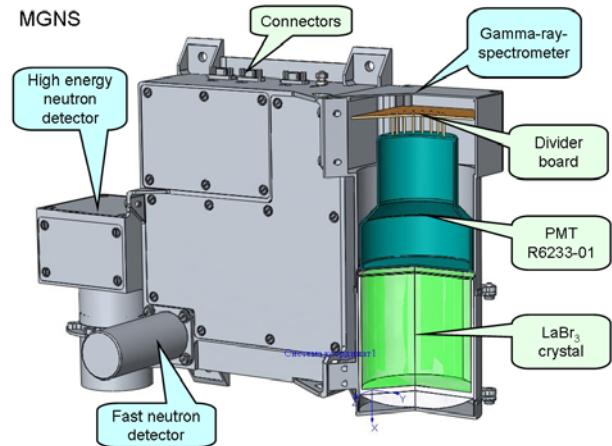
# $\text{LaBr}_3:\text{Ce}$ in 2011 on mission to Phobos (Mars) and in 2014 on mission to Mercury



$\text{LaBr}_3$  detector  
for the  
Phobos-Grunt  
mission



$\text{LaBr}_3$  detector for the  
BepiColombo mission



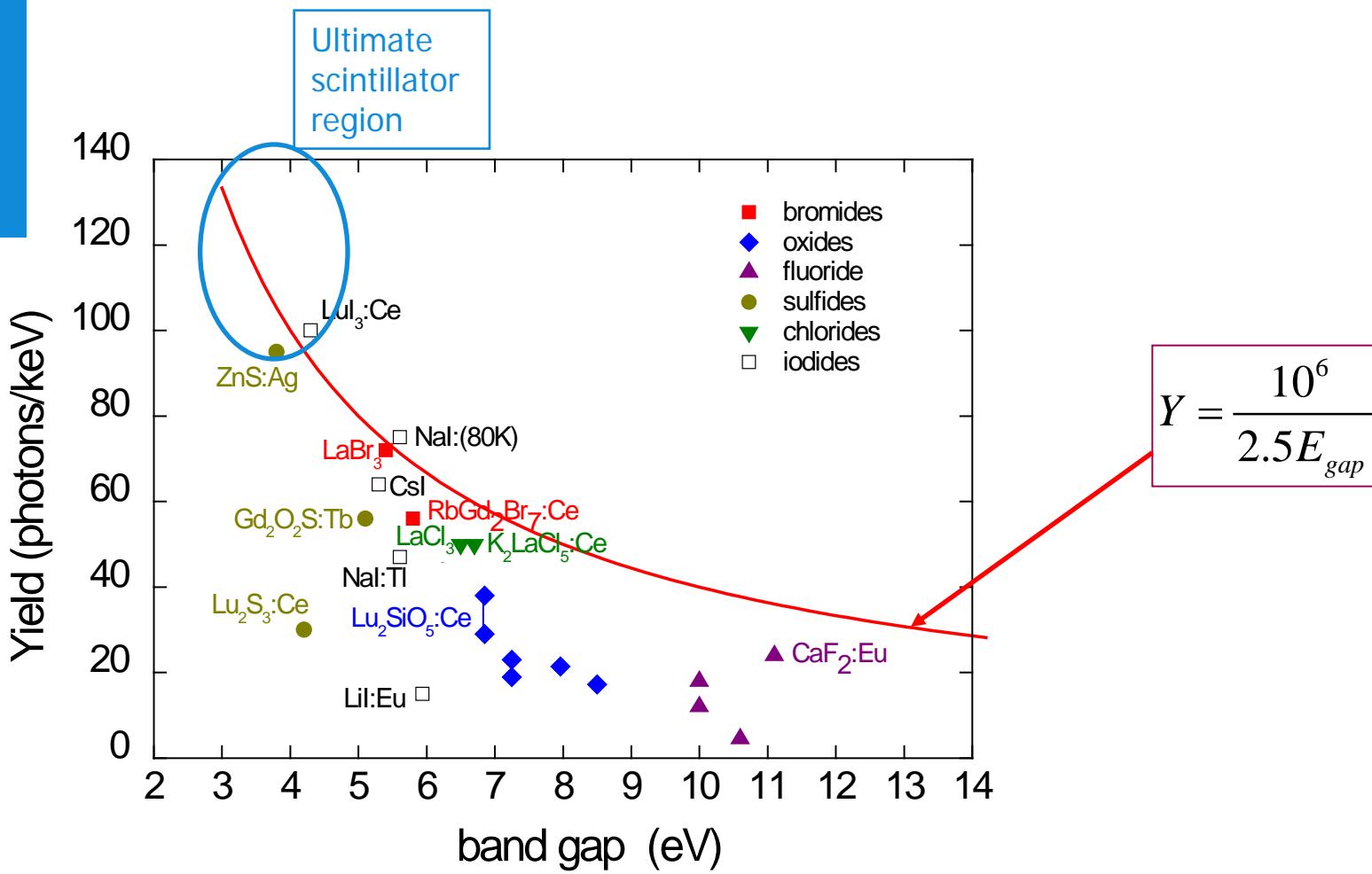
# Aspects that determine resolution

$$R^2 = R_{\text{stat}}^2 + R_{\text{inhom}}^2 + R_{\text{nonprop}}^2$$

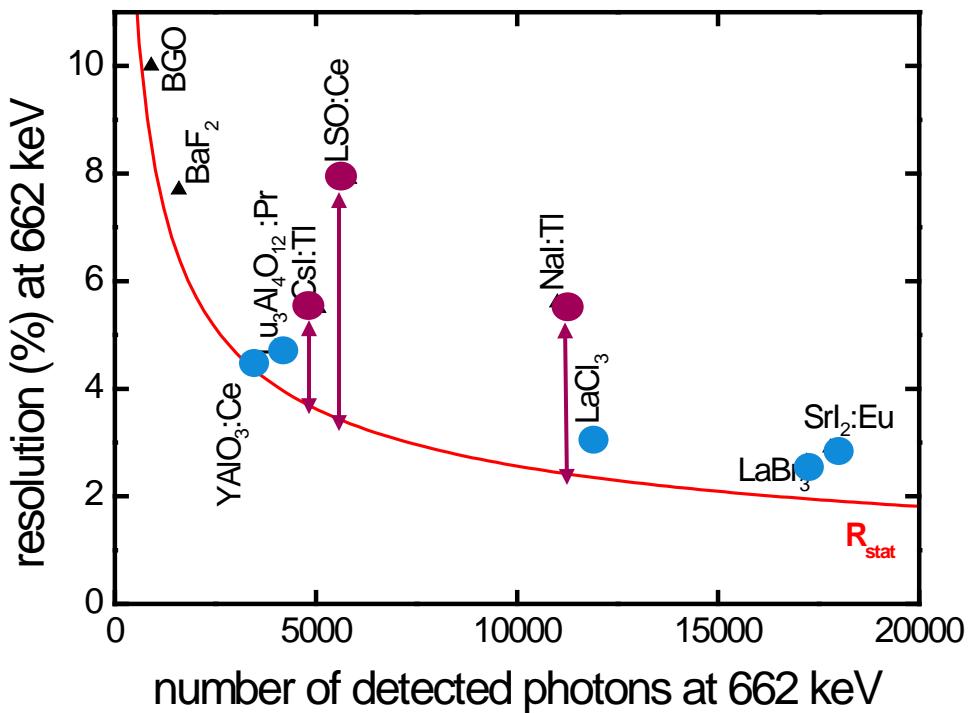
$$R_{\text{stat}}^{FWHM} (\%) = 235 \sqrt{\frac{1 + v_{PMT}}{N_{dph}}} \quad v_{PMT} \approx 0.2$$

- the fundamental limit is governed by statistics
- you need to detect as many photons as possible
  - bright scintillator
  - efficient low noise photon detector

# The ideal scintillator has S=Q=1



# Energy resolution @ 662 keV



$$R_{\text{stat}} = 2.35 \sqrt{\frac{1 + 0.2}{N_{dph}}}$$

- YAlO<sub>3</sub>:Ce, Lu<sub>3</sub>Al<sub>4</sub>O<sub>12</sub>:Pr, LaCl<sub>3</sub>:Ce, LaBr<sub>3</sub>:Ce, SrI<sub>2</sub>:Eu are reasonably close to fundamental limit.
- Lu<sub>2</sub>SiO<sub>5</sub>, NaI:Tl, CsI:Tl, strong deviation

What causes the strong deviation?

# Aspects that determine resolution

$$R^2 = R_{\text{stat}}^2 + R_{\text{inhom}}^2 + R_{\text{nonprop}}^2$$

$$R_{\text{stat}}^{\text{FWHM}} (\%) = 235 \sqrt{\frac{1 + v_{PMT}}{N_{dph}}} \quad v_{PMT} \approx 0.2$$

- when the number of created photons is not proportional with the energy of the primary electron
  - additional contribution to resolution ( $R_{\text{nonprop}}$ )
- topic of current interest
  - for study, we need a tunable  $\beta$ -source inside scintillator!

# The Compton Coincidence Technique

- Developed by Valentine and Rooney
  - Nucl. Instr. and Meth A353 (1994) 37; IEEE Trans.Nucl.Sci.43(1996)1271
- Provides electron response curves

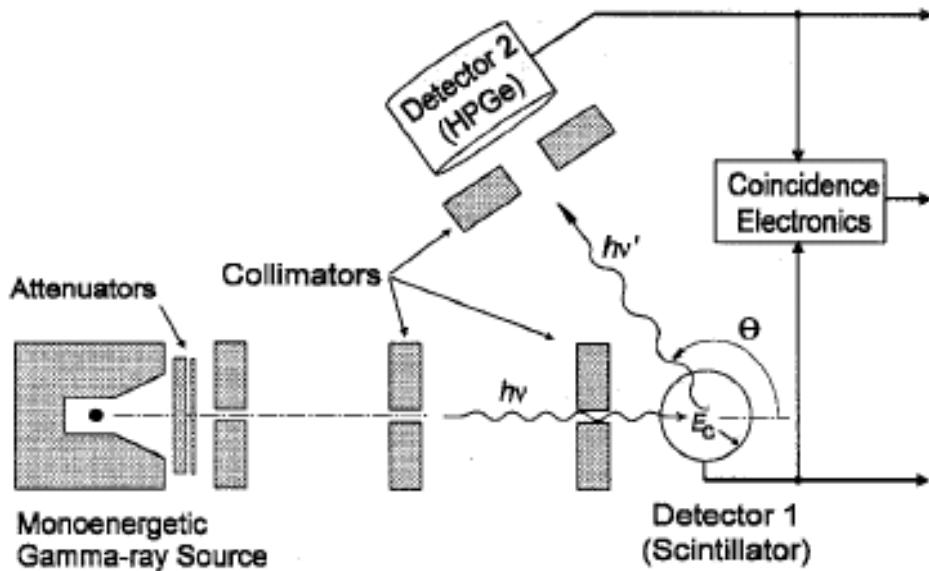


Fig. 1. Block diagram of Compton Coincidence Technique (CCT).

$$h\nu' = \frac{h\nu}{1 + (h\nu/m_0 c^2)(1 - \cos \theta)}$$
$$E_C = h\nu - h\nu'$$

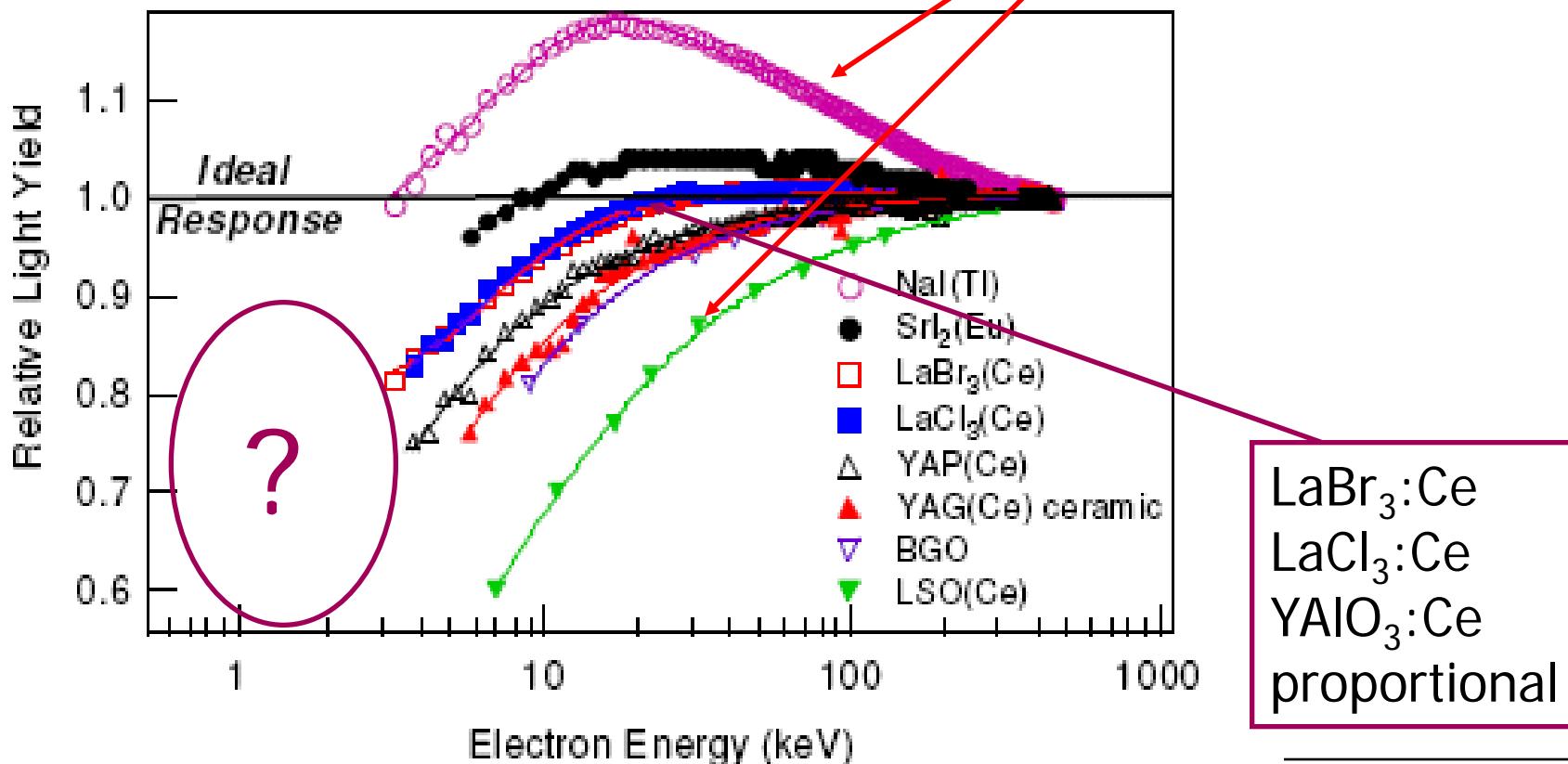
Advantage

- intrinsic tunable  $\beta$ -source
- energy range 6-400 keV

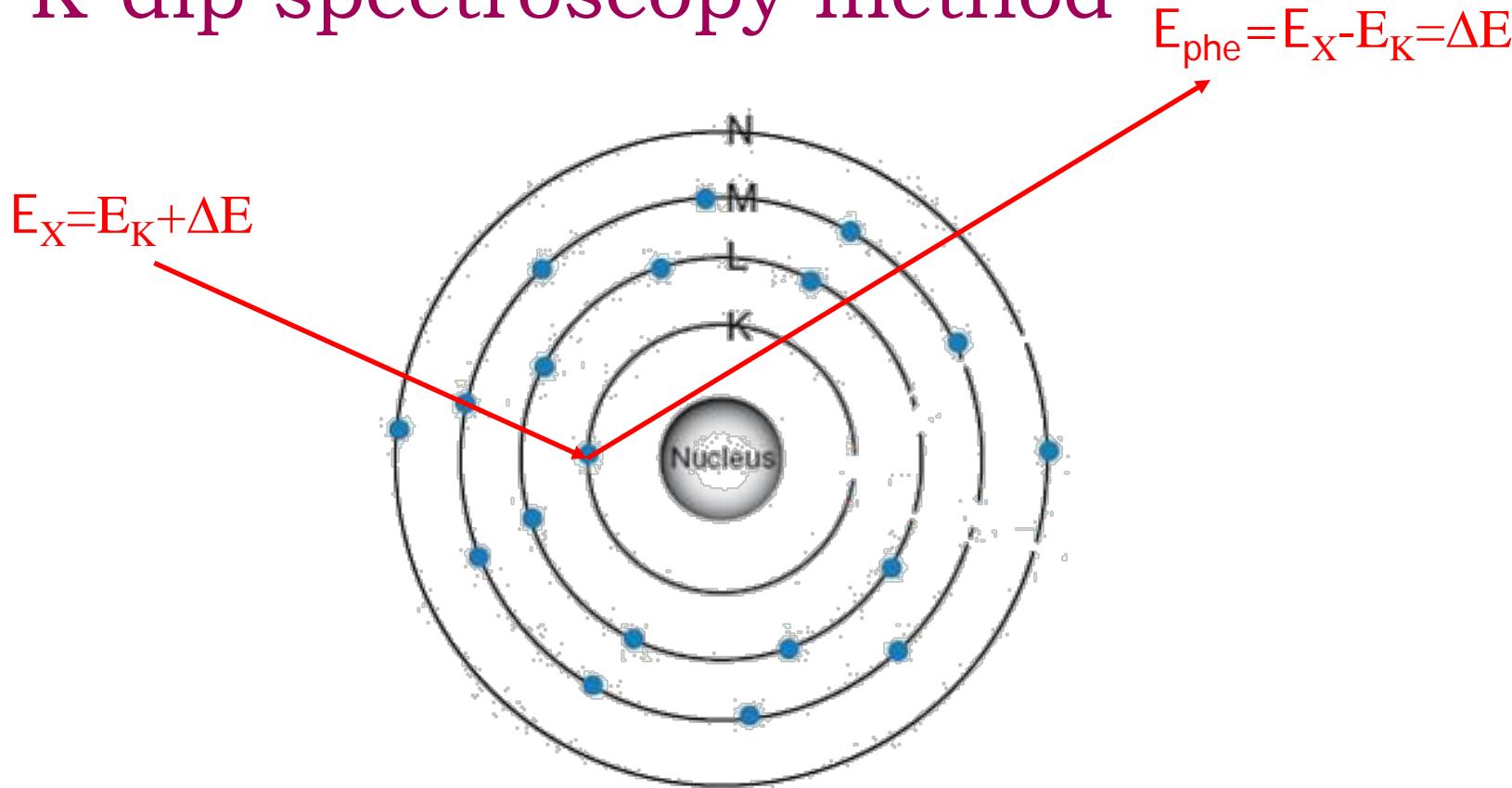
# Scintillator response to electrons

NaI:Ti, LSO:Ce  
Strongly  
Non-proportional

Cherepy et al. IEEE Trans. Nucl. Sci. 56(3) (2009) 873



# K-dip spectroscopy method



Energy photo-electron =  $E_x - E$  keV

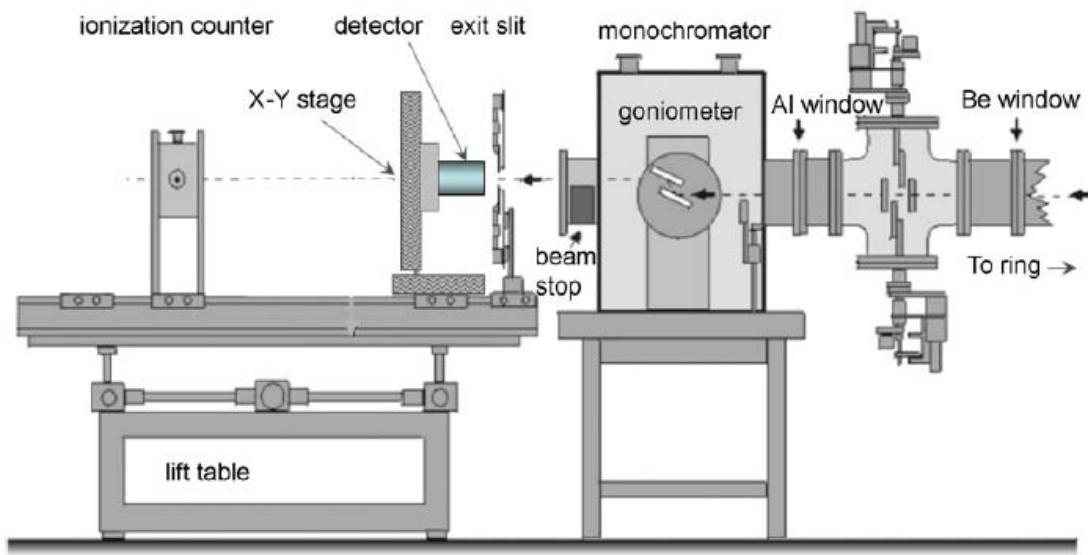
Yield (photo-electron) = total yield – yield@ $E_K$  keV

→ We have created a tunable internal β-source

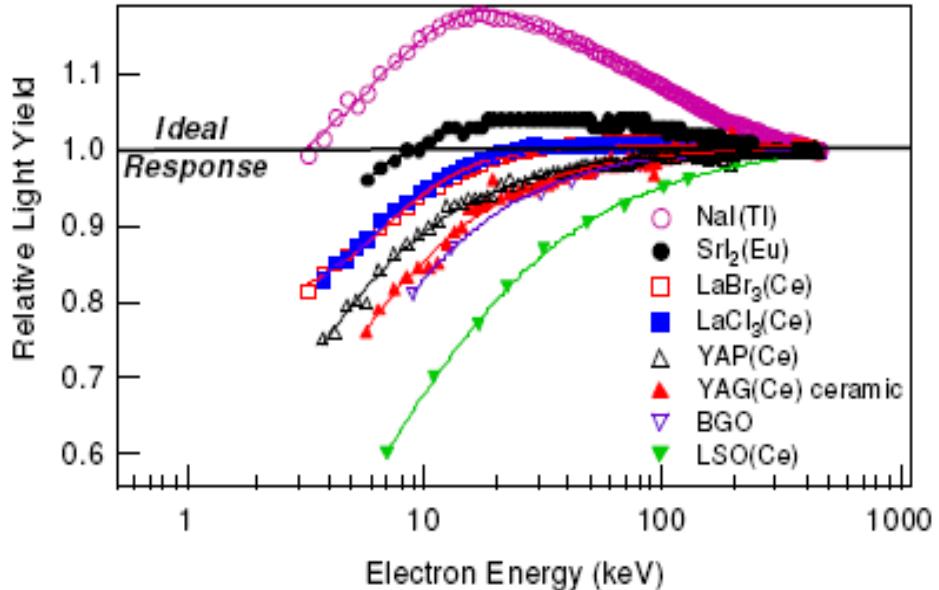
→ Electron response from ~0.1 to  $(100 - E_K)$  keV

# Tunable synchrotron X-ray source

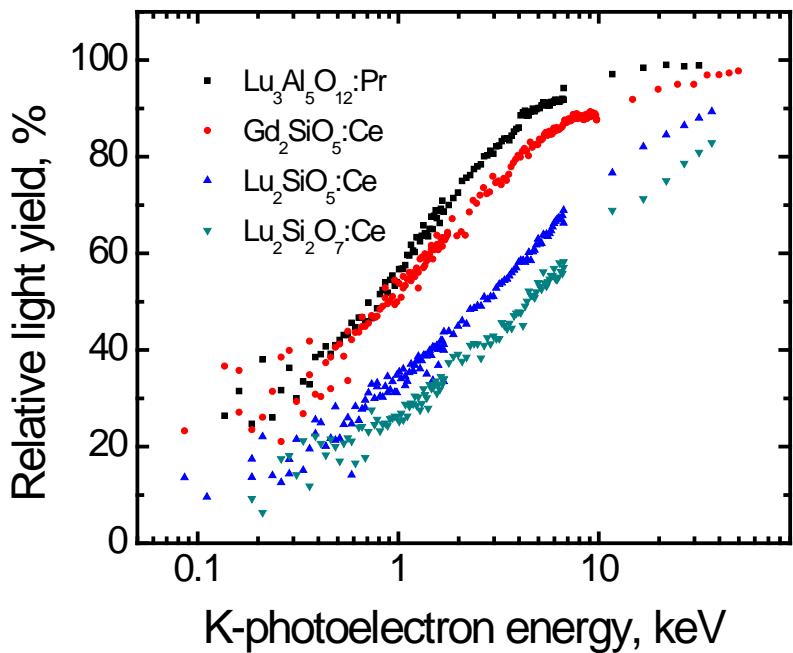
- Synchrotron X-ray beamline HASYLAB, DESY, Hamburg
- 9-100 keV monochromatic and tunable
  - 5-15 eV energy resolution
- Pencil beam
  - → we always excite the same volume element of the scintillator
  - → minimum contribution to resolution from scintillator inhomogeneity



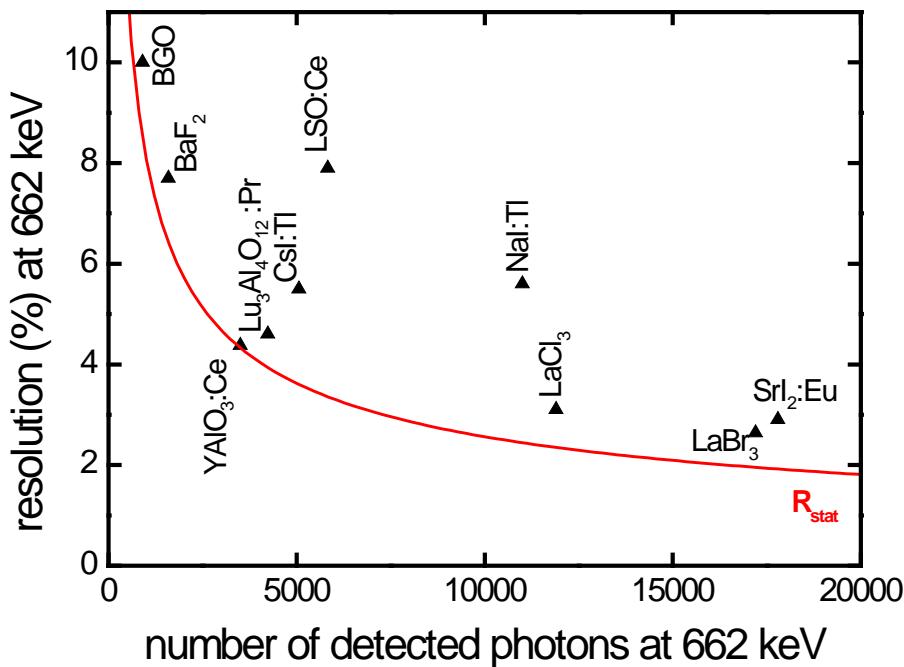
## Compton-electron spectroscopy



## K-electron spectroscopy



Joining K-dip with SLYNCI data  
→ 100eV – 400keV  
→ Electron response curves  
→ 20 different scintillators



- Can we go below 2% energy resolution at 662 keV?
  - We need bright scintillators combined with low noise photon detectors
  - We need proportional scintillators

# Time resolution

Time resolution = accuracy with which the moment of interaction can be determined

It depends:

- scintillation speed (decay time and rise time)
  - more ph/MeV
- photon detector time resolution
- electronics

# Timing resolution

BaF<sub>2</sub>

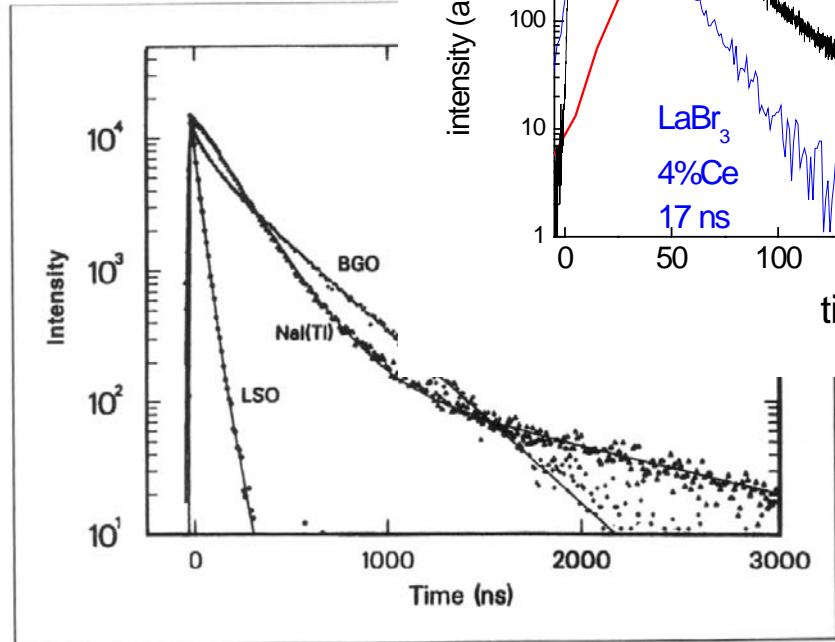
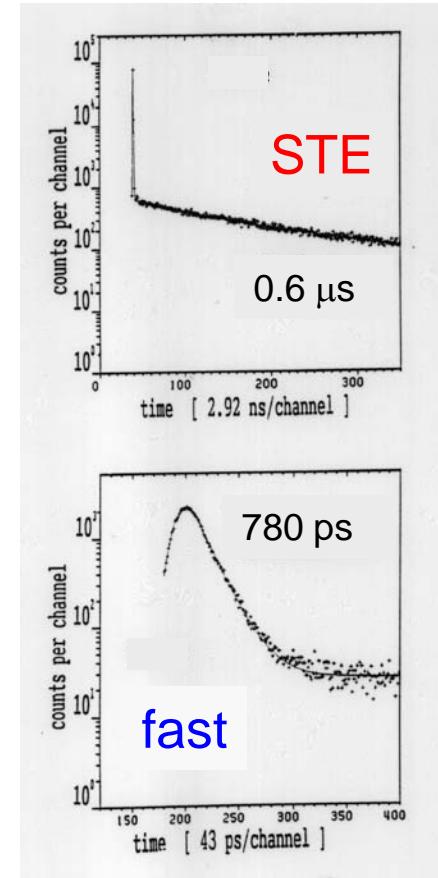
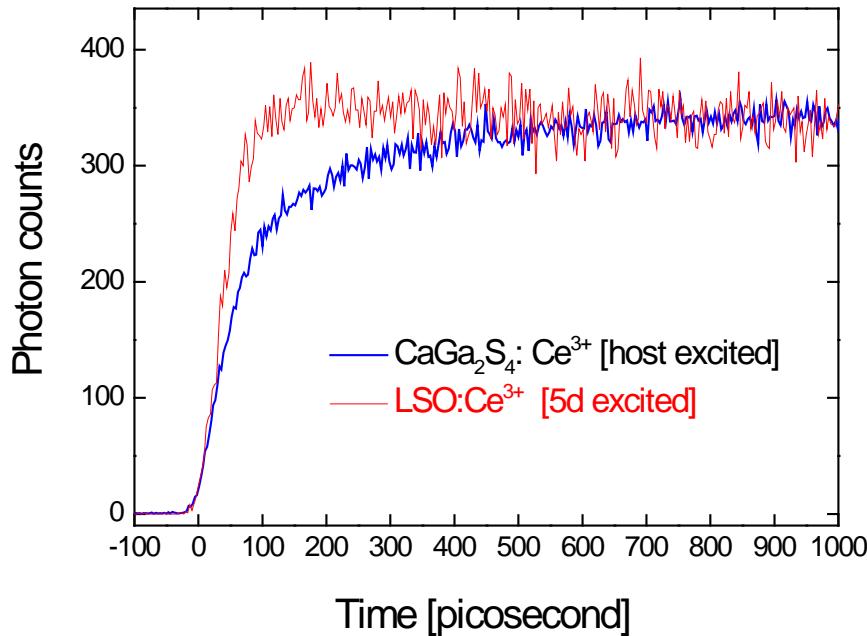


FIGURE 3. Decay of the scintillation emission of NaI(Tl), BGO, and LSO(Ce), after excitation by  $\gamma$ -rays.

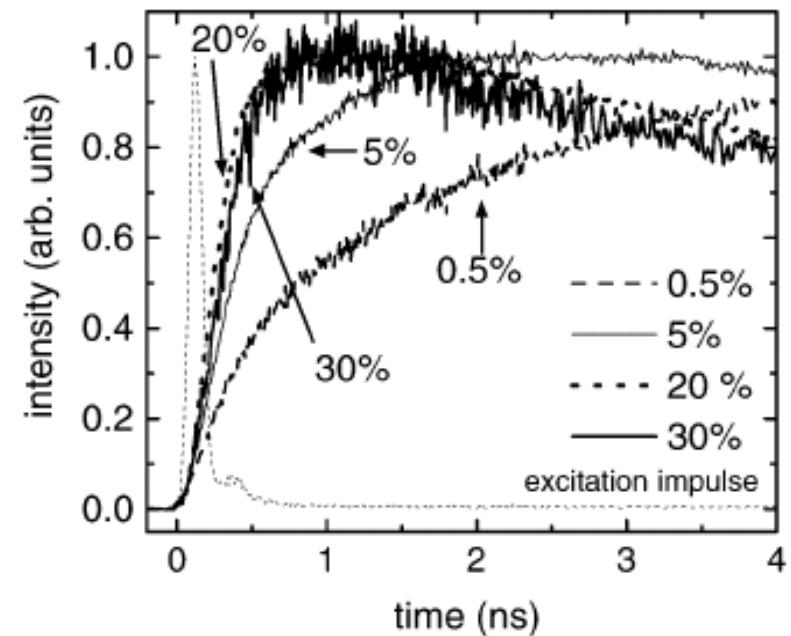


The steepness of the rising pulse  
is important

# Scintillator risetime studies



fs laser excitation



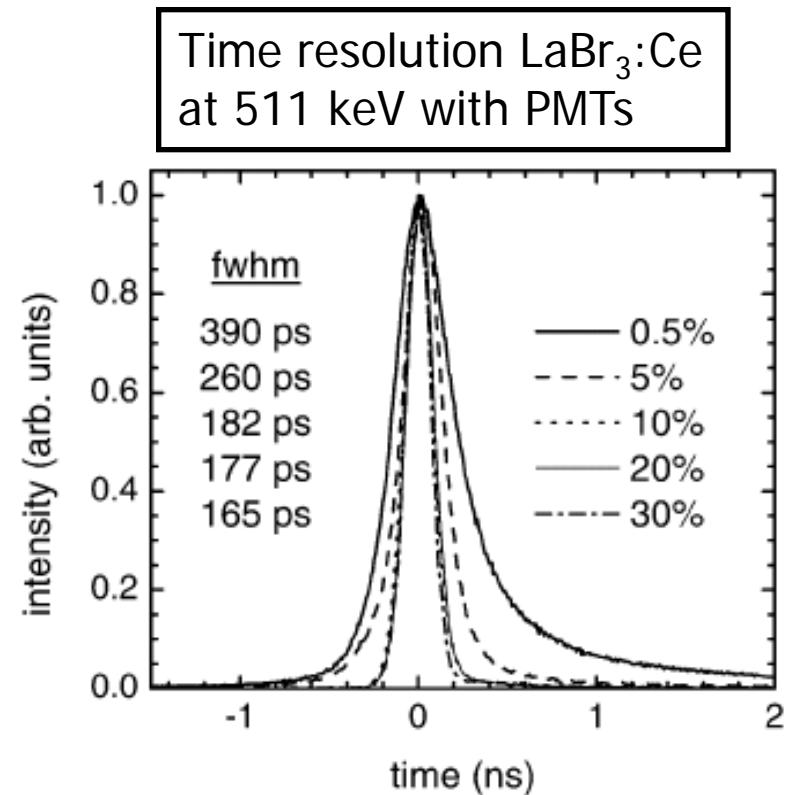
Pulsed X-ray excitation  
LaBr<sub>3</sub>:Ce

# Fast timing is important for

- positron emission tomography
- positron life time studies
- time of flight applications

Typical numbers for time resolution with PMTs and 1 MeV gamma's :

Nal(Tl)	:	1 ns
BGO	:	2-4 ns
BaF <sub>2</sub>	:	200 ps
Plastics	:	150 ps

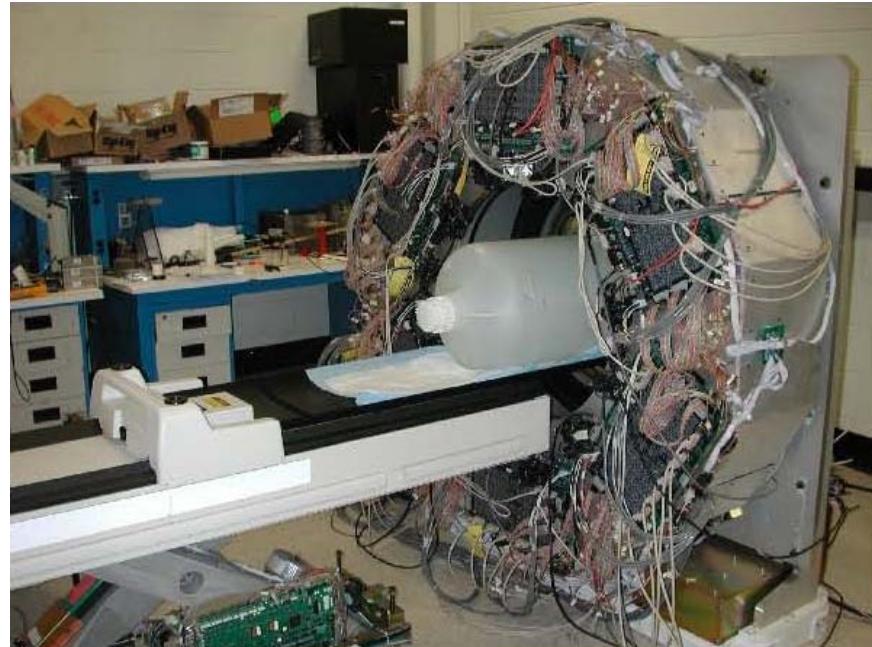


# $\text{LaBr}_3$ TOF-PET scanner

Time-of flight positron  
Emission tomography scanner

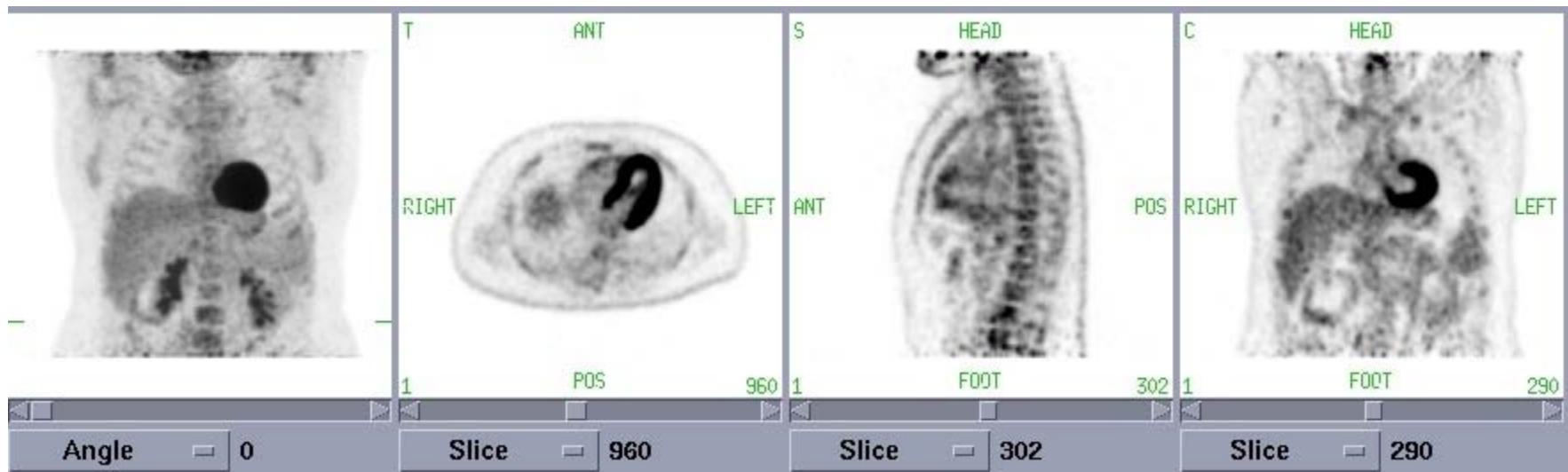
- 511 keV annihilation photons
- coincident detection
- time resolution 200 ps

Contains 38880  
 $\text{LaBr}_3$  crystals



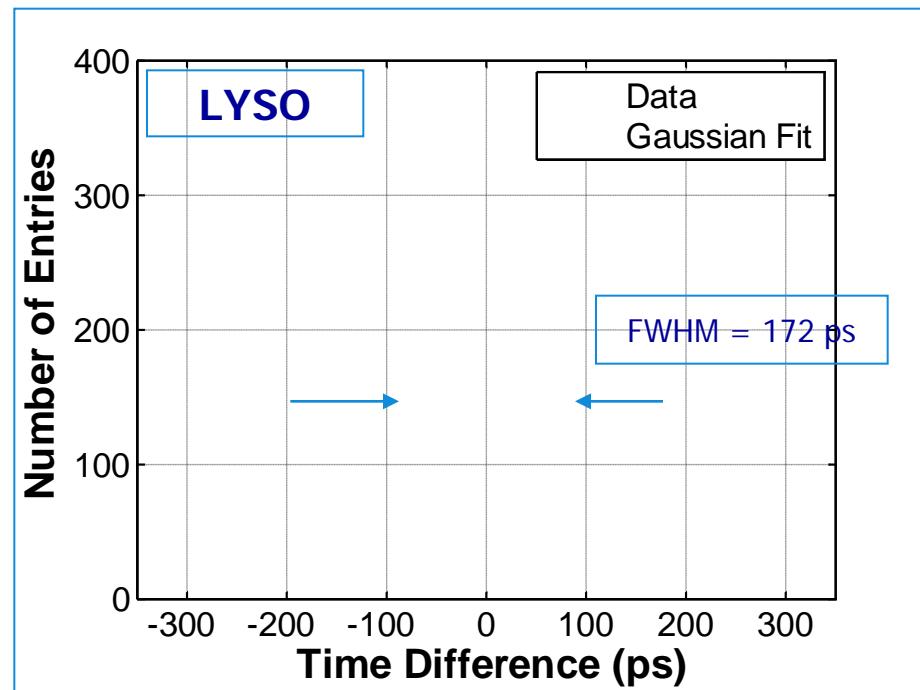
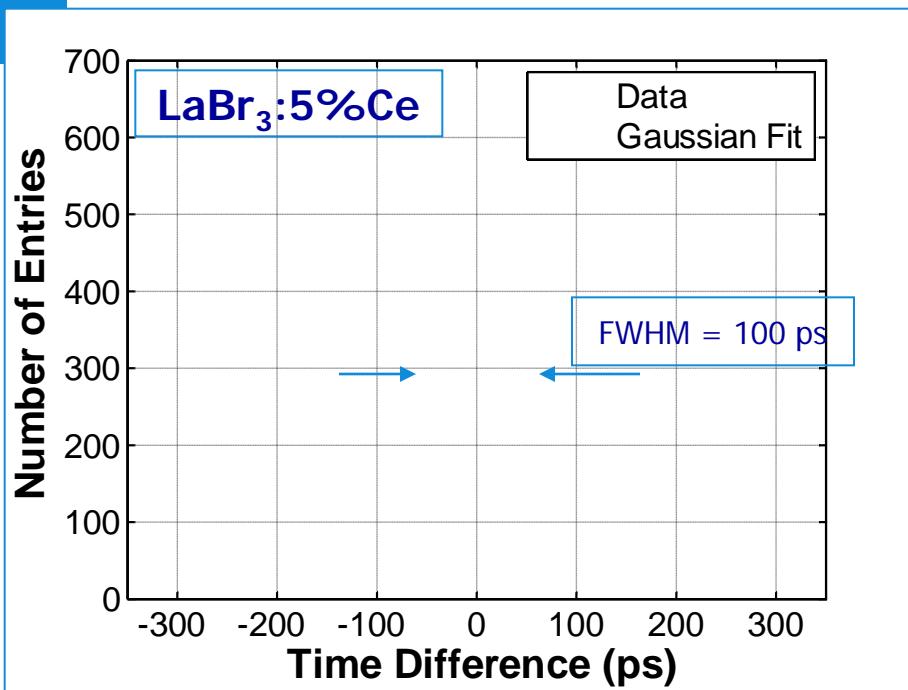
First prototype  $\text{LaBr}_3$   
TOF-PET-scanner (2008-2009)  
(dr. J. Karp Pennsylvanian State University)  
(and Philips Medical Systems)

# First TOF PET-image with LaBr<sub>3</sub>:Ce (October 2009)



Advantage TOF-PET strong false signal reduction especially important  
For fat patients

# Timing Spectra with SiPM – CRT at Optimum Threshold Settings



# Intrinsic activity of Lu<sub>2</sub>SiO<sub>5</sub>:Ce

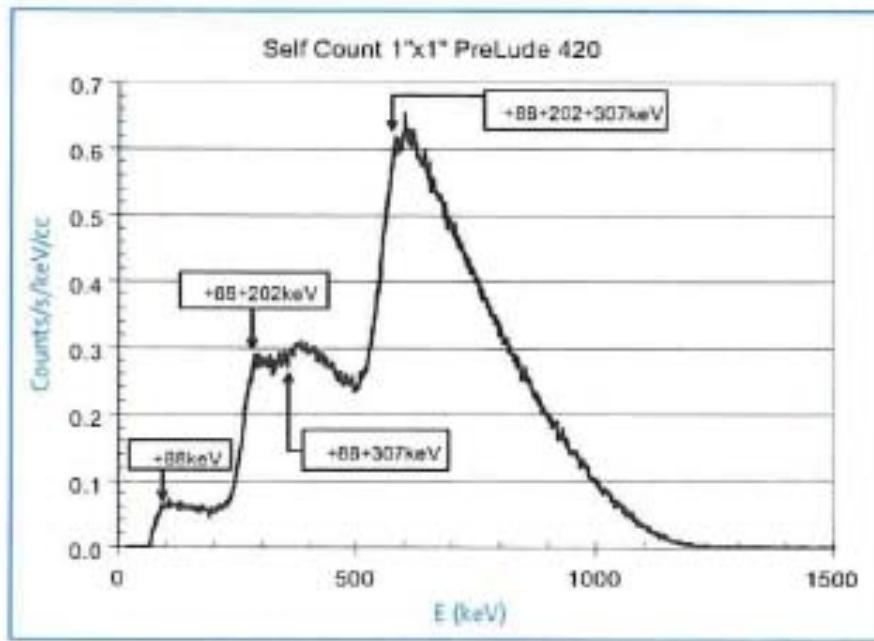


Figure 4. Beta distributions

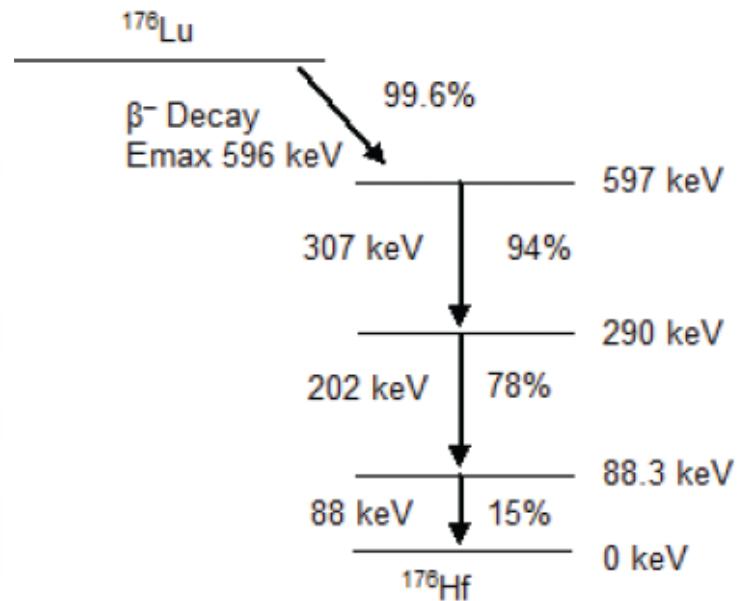
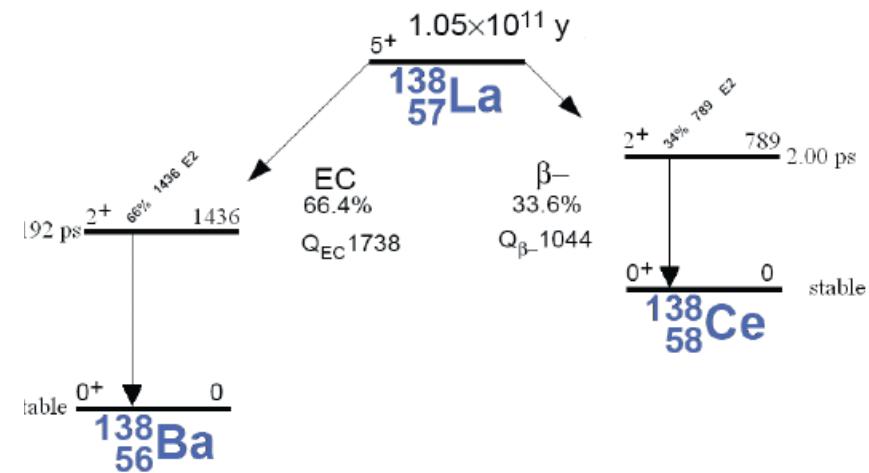
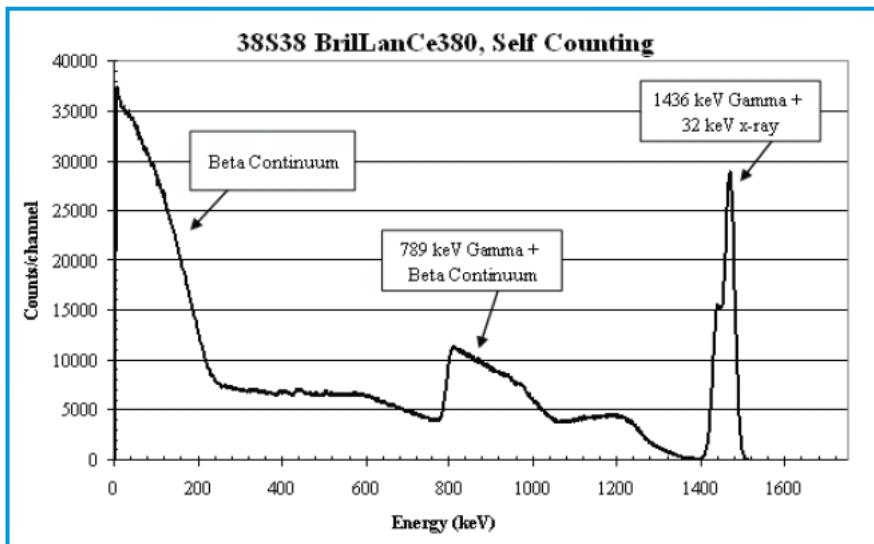


Fig.1 <sup>176</sup>Lu Decay Scheme

With abundance of 2.6% <sup>176</sup>Lu with half life of  $3.8 \times 10^{10}$  years an intrinsic activity of 290 cnts/cm<sup>3</sup> is obtained

<sup>176</sup>Lu isotopes emit 89, 202, plus 307 keV  $\gamma$ -photons plus emission of a  $\beta^-$  with maximum energy of 596 keV (plus a neutrino).

# Intrinsic activity of LaBr<sub>3</sub>:Ce



With abundance of 0.089% <sup>138</sup>La with half life of 1.0x10<sup>11</sup> years an intrinsic activity of 1.4 cnts/cm<sup>3</sup> is obtained

<sup>138</sup>La emits either a 1.436 MeV gamma plus an 32 keV X-ray from Ba or it emits a  $\beta^-$  with  $E_{max}=255$ keV plus a 789 keV gamma.

# Final remarks

- Ideal scintillator does not exist
  - compromise based on the application
- High resolution for spectroscopy (isotope identification)
  - proportional scintillators
- extremely fast rise time for timing and TOF applications
- low intrinsic count rate for space explorations
- hight density at low cost
  - for calorimeters
- current topics
  - understanding non-proportionality
  - Eu<sup>2+</sup> and Pr<sup>3+</sup> activated scintillators
  - scintillator pulse rise time studies

# Acknowledgements

- Some of the materials for these slides were provided by
  - Dr. P. Schotanus, company Scionix, The Netherlands
  - Dr. E. Mattmann, company Saint Gobain crystals, France
  - Dr. A. Owens and F. Quarati, European Space Agency, Netherlands
  - Dr. J Karp, Pennsylvanian State University, USA

# References

- Glodo et al. IEEE TNS, 52 (2005) 1805.
- Radiation Detection and Measurements, G. Knoll, John Wiley&Sons, Inc
- Saint Gobain brochures on scintillators, <http://www.detectors.saint-gobain.com/>
- N. Cherepy, S.A. Payne, et al. IEEE Trans. Nucl. Sci. 56(3) (2009) 873
- I.V. Khodyuk, J.T.M. de Haas, P. Dorenbos, IEEE Trans. Nucl. Science 57(2010) 1175-1181
- G. Bizarri, P. Dorenbos, Phys. Rev. B. 75 (2007) 184302.
- P. Dorenbos, IEEE Trans. Nucl. Sci. 57 (2010) 1162-1167.
- P. Dorenbos, J.T.M. de Haas, C.W.E. van Eijk, IEEE Trans. Nucl. Sci. 51(3) (2004) 1289-1296.
- M.J. Weber, J. Lumin. 100 (2002) 35.

# Thermal neutron scintillators

- Efficient thermal neutron detection
  - high capture cross section
    - suitable isotopes and isotope enrichment
  - absence of competing capture processes
- gamma ray back ground rejection
  - thin layers of low density
  - pulse height discrimination
  - pulse shape discrimination

## Thermal neutron capture reactions

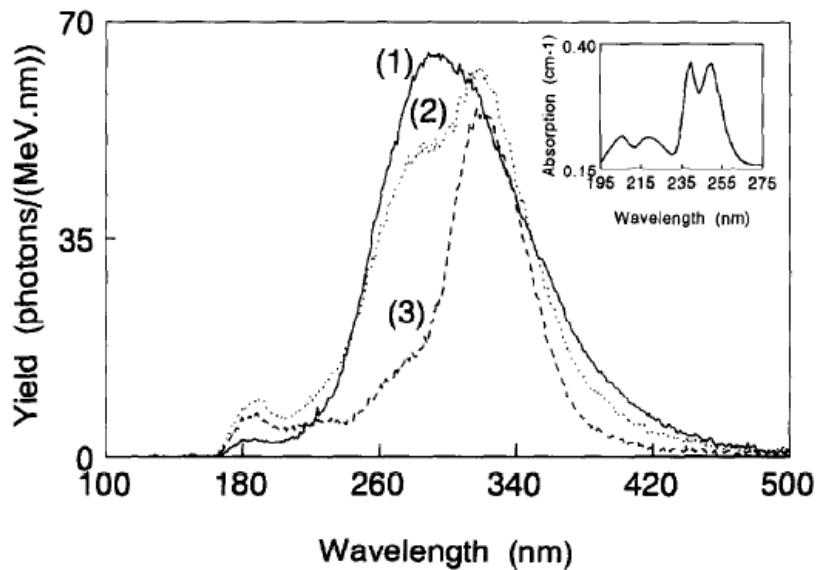
- $n + {}^6\text{Li} \rightarrow {}^3\text{H} + \alpha + 4.79 \text{ MeV}$
- $n + {}^{10}\text{B} \rightarrow {}^{11}\text{B} \rightarrow {}^7\text{Li} + \alpha + 2.78 \text{ MeV} \quad (7\%)$   
 ${}^7\text{Li}^* + \alpha + 2.30 \text{ MeV} \quad (93\%)$

# inorganic thermal-neutron scintillators

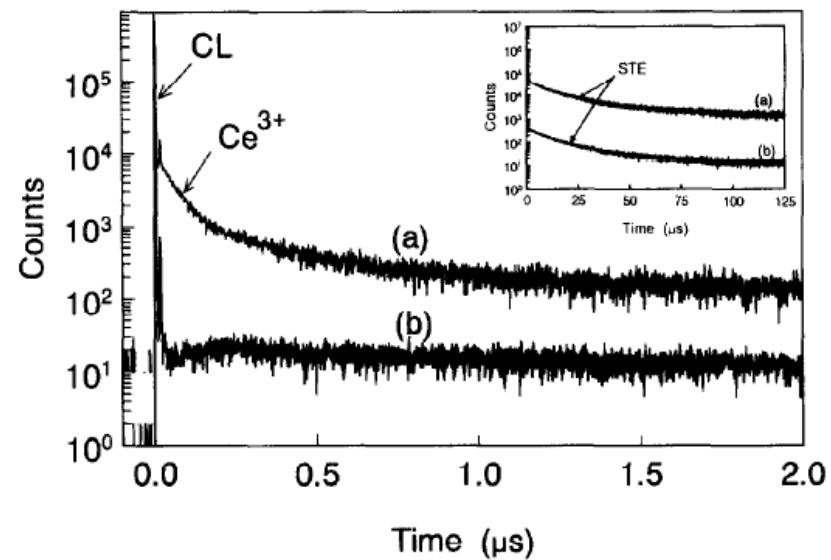
Scintill. host	Density g/cm <sup>3</sup>	Abs. Length at 1.8 Å mm	$\rho Z_{eff}^4$ (x 10 <sup>6</sup> )	Dopant/ conc-mol%	$\lambda_{em}$ nm	Light yield		$\tau$ ns
						neutron	photons per MeV gamma	
<sup>6</sup> Li-glass	2.5	0.52		Ce	395	6.000	4.000	75
<sup>6</sup> LiI	4.1	0.54	31	Eu	470	50,000	12,000	$1.4 \times 10^3$
<sup>6</sup> LiF/ZnS	2.6	0.8	1.2	Ag	450	160,000	75,000	$10 \times 10^3$
<sup>6</sup> Li <sub>6</sub> Gd(BO <sub>3</sub> ) <sub>3</sub>	3.5	0.35	25	Ce	385/415	40,000	20,000	200/800

# Pulse shape discrimination due to core-valence luminescence in $\text{LiBaF}_3:\text{Ce}^{3+}$

C.M. Combes et al. / Journal of Luminescence 72–74 (1997) 753–755



C.M. Combes et al. / Journal of Luminescence 72–74 (1997) 753–755

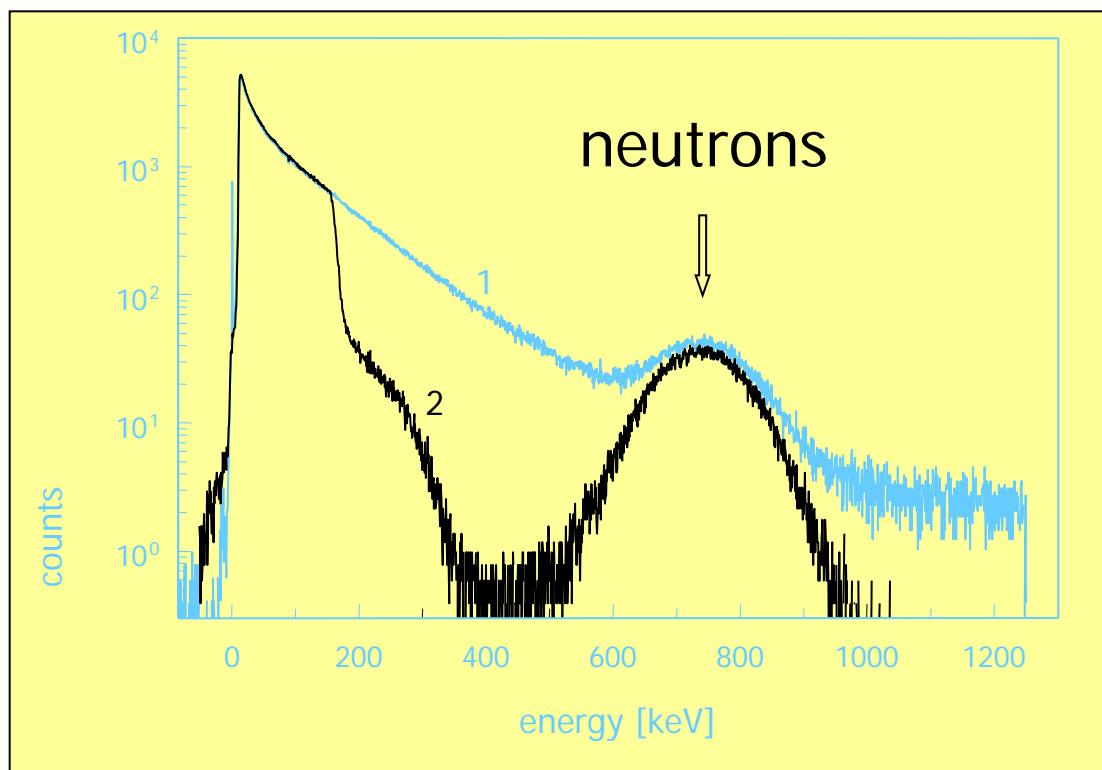


# $\text{LiBaF}_3:\text{Ce}$ scintillator

C.M. Combes *et al*, NIM A416(1998)364

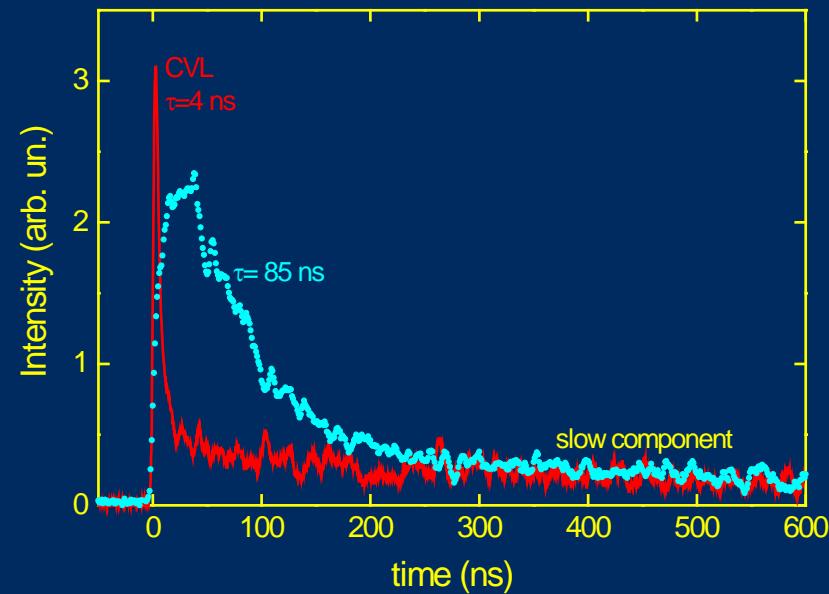
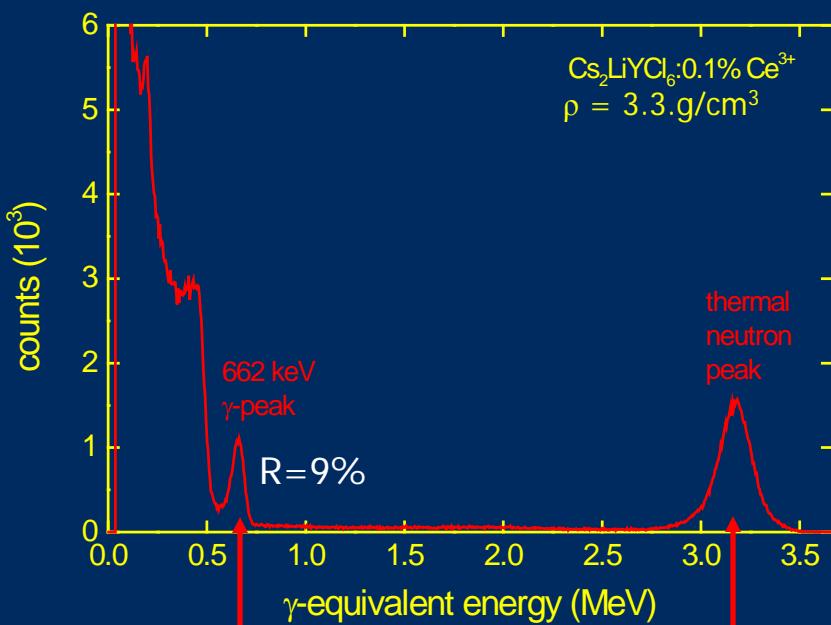
Spectra of radiation emitted by a Pu-Be source shielded with paraffine and 6 cm of lead

CVL does not respond to  
 ${}^3\text{H}$  (2.75 MeV) +  ${}^4\text{He}$  (2.05 MeV)



# 1 is without discrimination, # 2 with discrimination.  
The neutron peak at 750 keV gamma-equivalent energy  
is almost free from background

# Thermal neutron detection with $\text{Cs}_2\text{LiYCl}_6:0.1\% \text{Ce}^{3+}$



22000 ph/MeV

70000 ph/neutron

$\alpha/\beta = 0.66$

pulse shape plus  
pulse height  
discrimination

# Thermal neutron scintillators

Selection of the scintillator based on thermal neutron detection efficiency

$\sigma$ (barn)	6Li - nat /% $\Sigma\sigma$	6Li - 95enr /% $\Sigma\sigma$	Absorption length 0.18 nm - 95enr (mm)
$\text{Cs}_2\text{LiYBr}_6$	71/41	895/90	3.7
$\text{Rb}_2\text{LiYBr}_6$	71/62	895/95	3.5
$\text{Li}_2\text{NaYBr}_6$	142/77	1790/98	1.7
$\text{K}_2\text{LiYBr}_6$	71/60	895/95	3.5
$\text{Rb}_2\text{LiLaBr}_6$	71/58	895/95	3.5
$\text{Cs}_2\text{LiLuI}_6$	71/29	895/84	3.9
$\text{Rb}_2\text{LiYI}_6$	71/65	895/96	3.4

# Thermal neutron scintillation light yield and peak resolution $\text{Rb}_2\text{LiYBr}_6:\text{Ce}^{3+}$

Compounds	Photons/neutron	$\alpha/\beta$ ratio	Peak resolution %
$\text{Rb}_2\text{LiYBr}_6:$ 0.1% $\text{Ce}^{3+}$	$59,000 \pm 5,900$	0.74	3.6
$\text{Rb}_2\text{LiYBr}_6:$ 0.5% $\text{Ce}^{3+}$	$83,000 \pm 8,300$	0.75	5.4
$\text{Rb}_2\text{LiYBr}_6:$ 1% $\text{Ce}^{3+}$	$49,000 \pm 4,900$	0.79	8.5
	$51,000 \pm 5,100$	0.82	4.2
$\text{Rb}_2\text{LiYBr}_6:$ 5% $\text{Ce}^{3+}$	$39,000 \pm 3,900$	0.74	5.8
	$42,000 \pm 4,200$	0.80	7.8

Scintill. host	Density $\rho$ g/cm <sup>3</sup>	Abs. Length at 1.8 Å mm	$\lambda_{em}$ nm	Light yield		$\alpha/\beta$ ratio	T ns	Peak Resolution (%)			
				photons per							
				neutron	MeV gamma						
<sup>6</sup> Li Glass:Ce	2.5	0.52	395	~6,000	~4,000	0.31	75	13-22 % 3.9 % -			
Cs <sub>2</sub> LiYBr <sub>6</sub> :Ce <sup>3+</sup>	4.1	3.7	389	73,000	20,000	0.76	89, 2.5 x 10 <sup>3</sup>	4.6 %			
Rb <sub>2</sub> LiYBr <sub>6</sub> :Ce <sup>3+</sup>	4.8	3.5	385	83,000	23,000	0.74	42,140 1.6 x 10 <sup>3</sup>	3.6 %			
Li <sub>2</sub> NaYBr <sub>6</sub> :Ce <sup>3+</sup>	2.9	1.7	380	39,000	12,400	0.66	30,560 2.6 x 10 <sup>3</sup>	-			