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Microdosimetry and radiolytic species production in UHDR proton beam using GATE and Geant4-DNA

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FLASHMOD

Proton FLASH irradiation at ARRONAX

ARRONAX cyclotron facility:

IBA Cyclone[®]70 Protons, deuterium and alpha particles **Protons energy range: 30-70 MeV** (ranges in water: 8-38 mm)

Max dose rate > 7500 Gy/s

RT CONV $\sim 0,3Gy/s$





Radiobilogy Zebra fish irradiation Human cells irradiation

Radiochemistry: H_2O_2 measurements

Beam line simulation





H₂O₂ G values - physicslists



CHEM6 example



H₂O₂ G values along time, using 3 physics lists

Water radiolysis beyond the microsecond for UHDR protons



Chemical stage beyond the microsecond:

- Homogeneity within each voxel
- Diffusion of species between voxels
- Reaction between species within voxels
- Voxel size increases over time (heterogeneous -> homogeneous)

Simulated conditions:

- Aerated water radiolysis
- NO₂ scavenger
- Fricke dosimeter

Physical stage

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What is a cutoff dose?
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Protons are generated till a cutoff dose is reached

Tran, H.N.; Chappuis, F.; Incerti, S.; Bochud, F.; Desorgher, L.

Geant4-DNA Modeling of Water Radiolysis beyond the Microsecond: An On-Lattice Stochastic Approach.

Int. J. Mol. Sci. 2021, 22, 6023. https://doi.org/10.3390/ ijms22116023





Results – Aerated water



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Experimental measurements



Concentrations of H₂O₂ have been determined after irradiation about 15 minutes post irradiation.

The concentration of H_2O_2 was obtained indirectly by measurement of I_3^- absorbance using a spectrophotometer.

The radiolytic yield (G) is defined as the number of species formed or consumed per unit of deposit energy.

Blain,G. et al. **Proton Irradiation at Ultra-High Dose Rate vs. Conventional Dose Rate: Strong Impact on Hydrogen Peroxide Yield.** *Radiation Research*, **2022**, 198, 318-324.

Simulations vs experimental measurements



Cutoff dose (toward higher dose rates)



Hypothesis:

- pH variation
- Longer time

NO2 scavenger validation





 $OH + NO_2^- \rightarrow NO_2 + OH^$ k=8e9 M⁻¹s⁻¹

Proton 57 MeV

	0.0125 mM	0.125 mM	1.25 mM	12.5 mM	1.25 M
mol/J (10 ⁻⁸)	7.90	7.52	7.03	5.85	1.85

Hydrogen peroxide yields in water radiolysis by high-energy ion beams at constant LET Wasselin-Trupin⁻ Baldacchino. Bouffard. Hickel Radiation Physics and Chemistry Volume 65, Issue 1, August 2002, Pages 53-61

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Fricke dosimeter

This dosimetry technique depends

(Fe²⁺) to ferric ions (Fe³⁺) by ionizing

spectrophotometrically at 304 nm. The Fricke dosimeter is 96% water by weight; therefore, its dosimetric

properties are very similar to those

on the oxidation of ferrous ions

The increased concentration of

ferric ions is measured

16 14 experimental measure =1.48*10⁻⁶ mol/J 12 $= 1.43*10^{+1}$ molecules / 100 eV eV) 4 2 • 0.01 Gy 0 1,E-03 1,E+05 1,E-02 1,E-01 1,E+00 1,E+01 1,E+02 1,E+03 1,E+04 1,E+06 1,E+07 1,E+08 1,E+09 1,E+10 Time (ns)

Added reactions

radiation.

of water.

 Fe^{2+} + $^{\circ}OH \rightarrow Fe^{3+}$ + $^{\circ}OH^{-}$ $Fe^{2+} + HO_2 \rightarrow Fe^{3+} + HO_2^ Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + ^{\circ}OH + OH^-$

Added scavengers:

/chem/env/Scavenger 02 19 % /chem/env/Scavenger Fepp 10 mM $Fe^{3+} \rightarrow$ yields are comparable with experimental measurements





- Validated simulation of the beam line using GATE
- A new version of the Geant4-DNA chemistry module for water radiolysis simulation in UHDR conditions (with scavengers)
- H_2O_2 G-values \neq experimental measurements -> WORK IN PROGRESS
- NO₂ scavenging capacity validated with literature
- First results for Fricke dosimeter in conventional dose rate show good agreement with experimental measurements