SPES @ LNL (... and Laser @ SPES)

Aggiornamento per docenti di scuola secondaria di II grado LNL-INFN

Daniele Scarpa

- Tortona (AL) 1979
- I.T.I.S. G. Marconi => Perito Elettronica e Telecomunicazioni
- Università degli studi di Pavia => Ing. Elettronica (2004)

Dott. Elettronica (2007)

Assegno di Ric. (2008)

• INFN

=> Pavia (2009) AdR LNL (2013)

Topics

- SPES Project
- Exotic species and radioactivity
- SPES α,β,γ,δ
- SPES and Lasers
- Laser labs @ LNL



What? Exotic Species => Radioactive Atoms

















Where? Nature and its laws 🔗

How take a picture of a world record?



Easy Way?



SPES @ Laboratori Nazionali di Legnaro



Why? Demand & Supply



Easy Way?

Get fruit salad & separate each fruit





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SPES-a: The cyclotron and related infrastructure.



High Power Cyclofron:

Several application at LNL

✓ Fundamental Research:

• SPES and its phases

• ...

✓ Applied Research

Medical application

• • •



Main Parameter

Accelerator type	Cyclotron AVF with 4 sectors, Resistive Magnet
Particle	Protons (H ⁻ accelerated)
Energy range	35-70 MeV
Max Current Intensity	700 μA (variable within the range 1μΑ-700μΑ)
Extraction	Dual stripping extraction
Max Magnetic Field	1.6 T (Bo = 1 T)
RF System	nr. 2 delta cavities; harmonic mode=4; f _{RF} =56 MHz; 70 kV peak voltage; 50 kW RF power (2 RF amplifiers)
lon Source	Multi-cusp volume H ⁻ source; I _{ext} =8mA; V _{ext} =40 kV; axial injection
Dimensions	Φ=4.5 m, h=2 m, W=190 tons

SPES-B: The ISOL facility and the acceleration of neutron-rich unstable nuclei.



SPES-β: The ISOL facility and the acceleration of neutron-rich unstable nuclei. SPES Zoom IN: Inside SPES Target (>10 Years of R&D)









SPES- γ : The production of radionuclides for applications.



Direct production of Medical-Radio-Isotope using the ISOL technique



MED

Direct production of Medical-Radio-Isotope using the Cyclotron

SPES-6: The multidisciplinary neutron sources.

Neutrons: Several ideas in the future



- NEPIR: (NEutron and Proton IRradiation facility) project
- QMN: Quasi Mono-energetic Neutron source with a controllable energy peak in the 20-70Mev energy range
- ANEM: Atmospheric Neutron Emulator to get a fast neutrons (E>1 MeV) with a continuous energy distribution similar to that of neutrons found at flight-altitudes and sea-level for SEE testing
- SLOWNE: a high intensity slow neutron (E < MeV) flux for applications

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Light

From Wikipedia, the free encyclopedia

"Visible light" redirects here. For light that cannot be seen with human eye, see Electromagnetic radiation. For other uses, see Light (disambiguation) and Visible light (disambiguation). "Lightsource" redirects here. For the solar energy developer named Lightsource, see Lightsource Renewable Energy.

Light is electromagnetic radiation within a certain portion of the electromagnetic spectrum. The word usually refers to **visible light**, which is the visible spectrum that is visible to the human eye and is responsible for the sense of sight.^[1] Visible light is usually defined as having wavelengths in the range of 400–700 nanometres (nm), or 4.00×10^{-7} to 7.00×10^{-7} m, between the infrared (with longer wavelengths) and the ultraviolet (with shorter wavelengths).^{[2][3]} This wavelength means a frequency range of roughly 430–750 terahertz (THz).

The main source of light on Earth is the Sun. Sunlight provides the energy that green plants use to create sugars mostly in the form of starches, which release energy into the living things that digest them. This process of photosynthesis provides virtually all the energy used by living things. Historically, another important source of light for humans has been fire, from ancient campfires to modern kerosene lamps. With the development of electric lights and power systems, electric lighting has effectively replaced firelight. Some species of animals generate their own light, a process called bioluminescence. For example, fireflies use light to locate mates, and vampire squids use it to hide themselves from prey.

The primary properties of visible light are intensity, propagation direction, frequency or wavelength spectrum, and polarization, while its speed in a vacuum, 299,792,458 metres per second, is one of the fundamental constants of nature. Visible light, as with all types of electromagnetic radiation (EMR), is experimentally found to always move at this speed in a vacuum.^[4]

In physics, the term *light* sometimes refers to electromagnetic radiation of any wavelength, whether visible or not.^{[5][6]} In this sense, gamma rays, X-rays, microwaves and radio waves are also light. Like all types of EM radiation, visible light propagates as waves. However, the energy imparted by the waves is absorbed at single locations the way particles are absorbed. The absorbed energy of the EM waves is called a photon, and represents the quanta of light. When a wave of light is transformed and absorbed as a photon, the energy of the wave instantly collapses to a single location, and this location is where the photon "arrives." This is what is called the wave function collapse. This dual wave-like and particle-like nature of light is known as the wave-particle duality. The study of light, known as optics, is an important research area in modern physics.

- Contents [hide]
- 1 Electromagnetic spectrum and visible light
- 2 Speed of light
- 3 Optics
 - 3.1 Refraction



A triangular prism dispersing a beam of \square white light. The longer wavelengths (red) and the shorter wavelengths (blue) are separated.



Light (Laser)

- Energy
- Wavelength
- Propagation
- Wave.... Particle....





 $E = \hbar \omega = h\nu = \frac{hc}{\lambda}$

 $\boldsymbol{p}=\hbar\boldsymbol{k},$

Light Properties: Color



Light emission..... Absorption firts

• Light absorption



$$E = \hbar \omega = h \nu = rac{hc}{\lambda}$$
 $m{p} = \hbar m{k},$

Emission..... Spontaneous



- LED => Electrical power
- Fluorescence => Light!!! => absorption



Emission..... Stimulated (1905)





Laser (1960)





• light amplification by stimulated emission of radiation



Selecting color...







Color?

• Who drive the laser color? => Active medium & Losses



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Flame test

• «Each element has its own favorite color»



• Emission/Absorbtion



Ionization Potential







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LNL-INFN SPES Offline Laser Laborator



Lasers:

3 Dye lasers pumped by 2 Nd:YAG 10Hz rep rate lasers; Nd:YAG for ablation

Measurement systems:

Hollow Cathode Lamps and Time of Flight Mass Spectrometer

Hollow cathode lamps

Designed to provide spectral emission of different elements.

- Commercial for Opto Galvanic application
- Inexpensive
- Available for almost the whole periodic table
- Electrical noise is very small, comparable with the level of shot noise
- Easy setup

The OG effect arises from the interaction of resonant radiation with atoms present in a discharge.

Lawler, 1980, observed that the OG effect can be considered proportional to the number of photons absorbed [1]



[1]Beniamino Barbieri, Nicolò Beverini, Antonio Sasso: "Optogalvanic spectroscopy", Reviews of Modern Physics, Vol. 62, No.3, July 1990

Optogalvanic Signals

SLOW Opto-Galvanic Signal:

The absorption of laser radiation in the discharge results in a change in the steady-state population of bound atomic or molecular levels. Since different levels will have different ionization cross-sections, a perturbation to the steady-state situation results in a net change in the discharge current or equivalently a change in the discharge impedance. The electric signal detected is the slow signal, negative and lasting μ s.





FAST Opto-Galvanic Signal:

It is a direct ionization process during laser pulse. The laser radiation brings the selectively excited atoms directly to ionization. Electrons are immediately available as carriers. This effect produces a fast electric signal. It was found (Broglia et al 1983 [3]) that this fast signal follows the laser pulse temporal behavior (ns).

[3] M.Broglia, F.Catoni, P.Zampetti: "Temporal behaviour of the optogalvanic signal in a hollow cathode lamp", Journal de Physique, Colloque C7, supplement au n° 11, Tome 44, novembre 1983

Laser resonant ionization





Time of flight mass spectrometer Measure Sequence:





- Ablation
 Plume Expansion
 Photoionization
 Flight
- 5) Collection

Time of flight mass spectrometer



Time of flight mass spectrometer

Simion® simulation VS ToF acquisition & ToF mass resolution







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