How a small accelerator can be useful for interdisciplinary applications Part II: cultural heritage studies

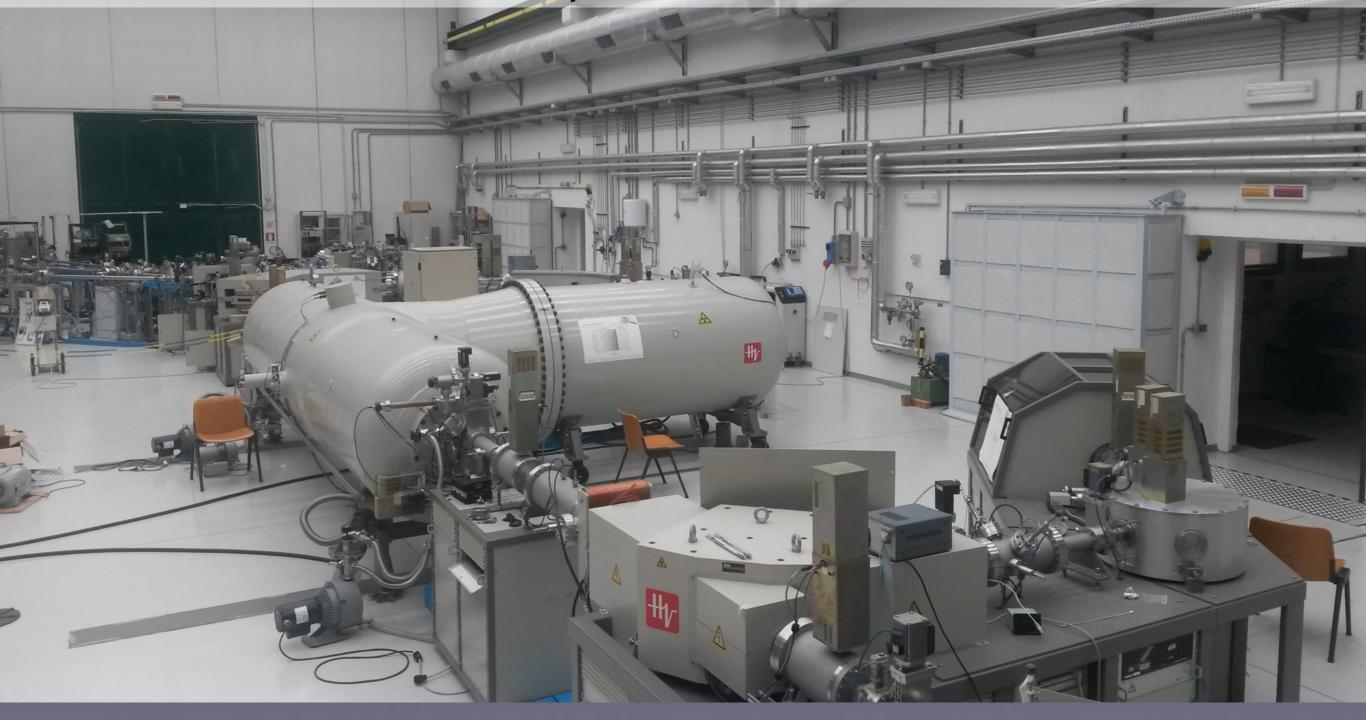
Mariaelena Fedi

CHNet (Cultural Heritage Network)
INFN Sezione di Firenze



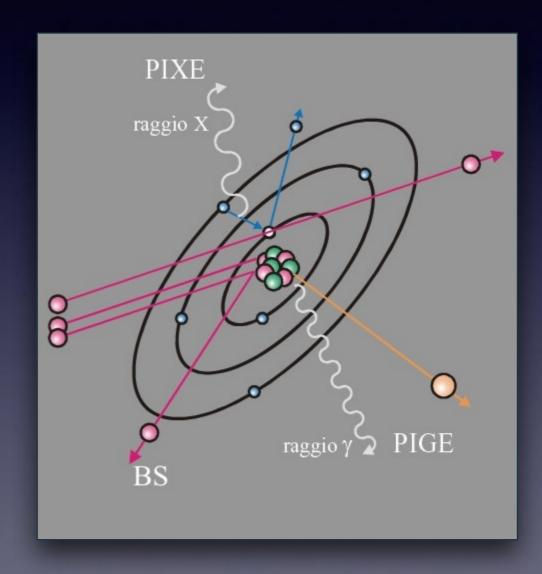
An example of small accelerator

Laboratorio di tecniche nucleari per l'Ambiente e i Beni Culturali - Firenze



What you have already learnt

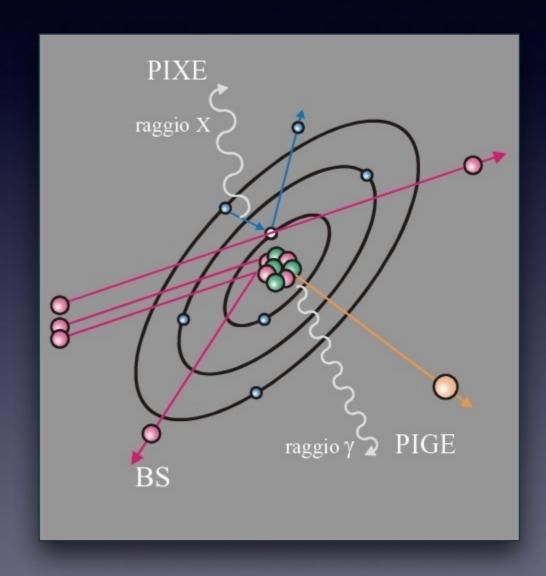
- IBA (Ion Beam Analysis) measurements can allow us to study the composition of a sample
 - PIXE (Particle Induced X-ray Emission) -> information on Z
 - PIGE (Particle Induced γ-ray Emission) -> information on the isotope
 - BS (Back Scattering) -> information on A



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But... what about these nuclear techniques and Cultural Heritage?

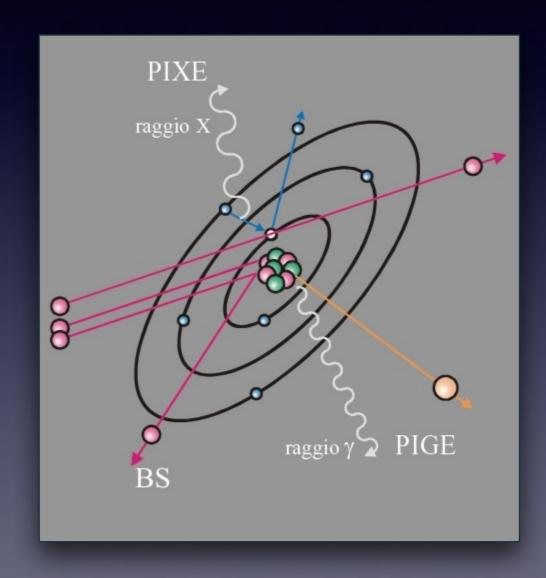


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But... what about these nuclear techniques and Cultural Heritage?

or... more in general... why mixing nuclear techniques, material analysis and CH?



The use of science and modern technology in archaeology to examine and interpret archaeological remains

(Collins dictionary def.)

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Materials studies

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Materials studies

- to study the preservation state of artworks and findings
- to support restoration (to find the most appropriate materials which are compatible with the original ones)
- to study the artistic techniques
- to date (although in an indirect way)

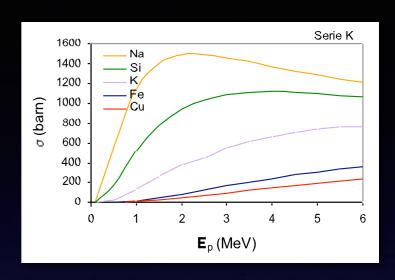
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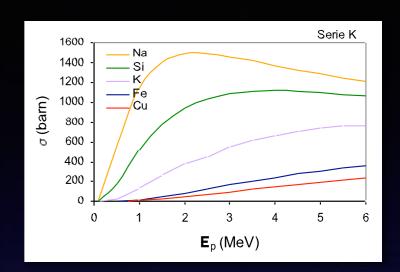
- Materials studies
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- Dating



 X-ray production cross sections are pretty high -> we can keep the incident ions current as low as possible, still allowing for fast acquiring times



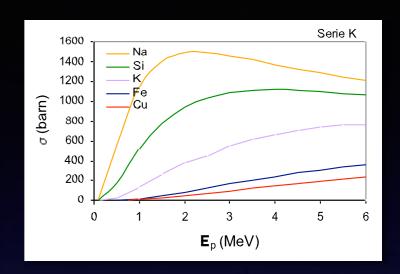
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 Incident ions can be extracted into atmosphere
 no sampling is needed



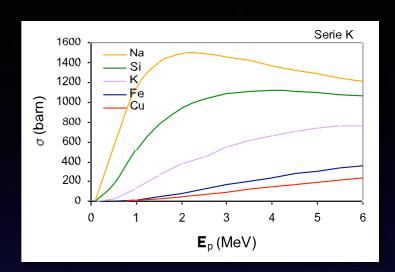
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Thus, PIXE measurements (IBA in general) are: non invasive and non destructive quick



 In external beam set-ups, X-rays with energies below 1 keV are not collected -> no information on elements with Z<11



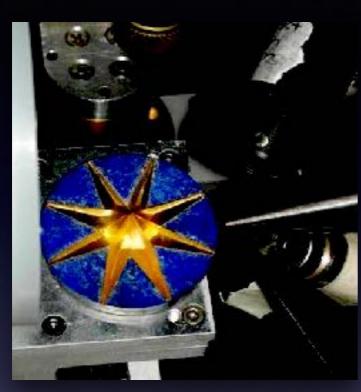
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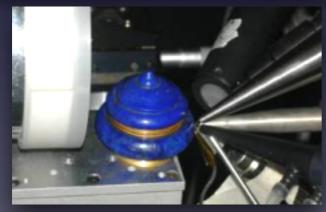


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 - -> information on the element and not on possible molecular structure and compounds
- In-situ measurements are not typically possible



Some examples of artworks to be analyzed





"Collezione Medicea di Pietre Ornamentali", Museum of Natural History, University of Florence



Galileo Galilei, hand-written manuscript, Biblioteca Nazionale, Florence

Some examples of artworks to be analyzed

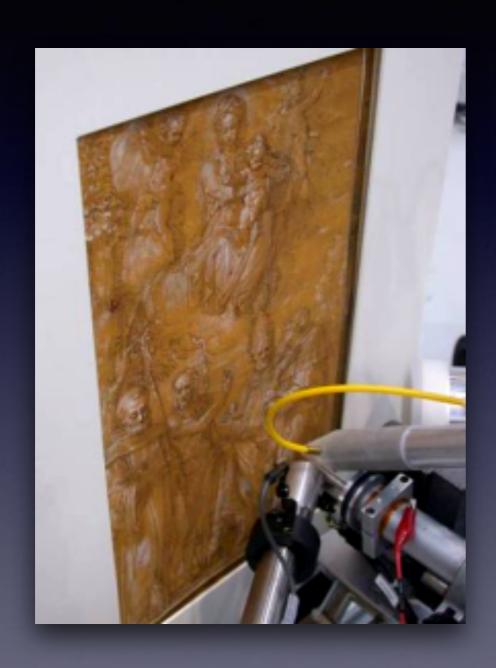


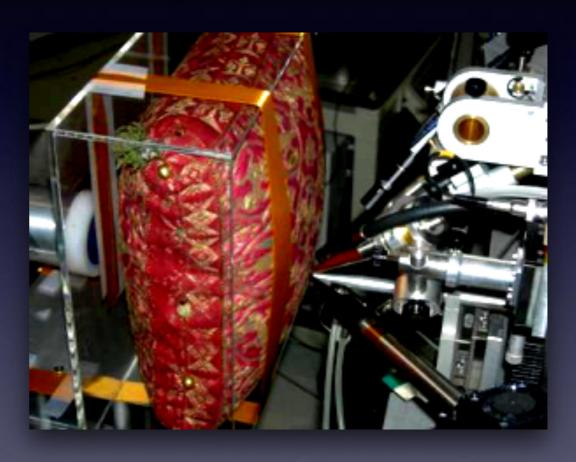
Leonardo da Vinci, Madonna dei Fusi (private collection)

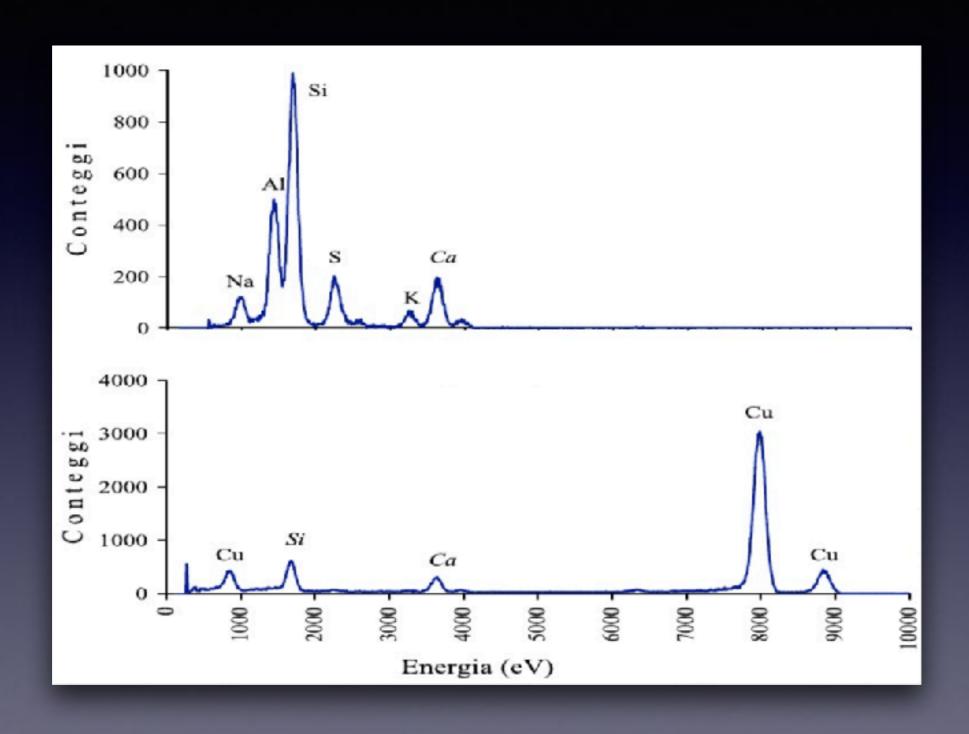


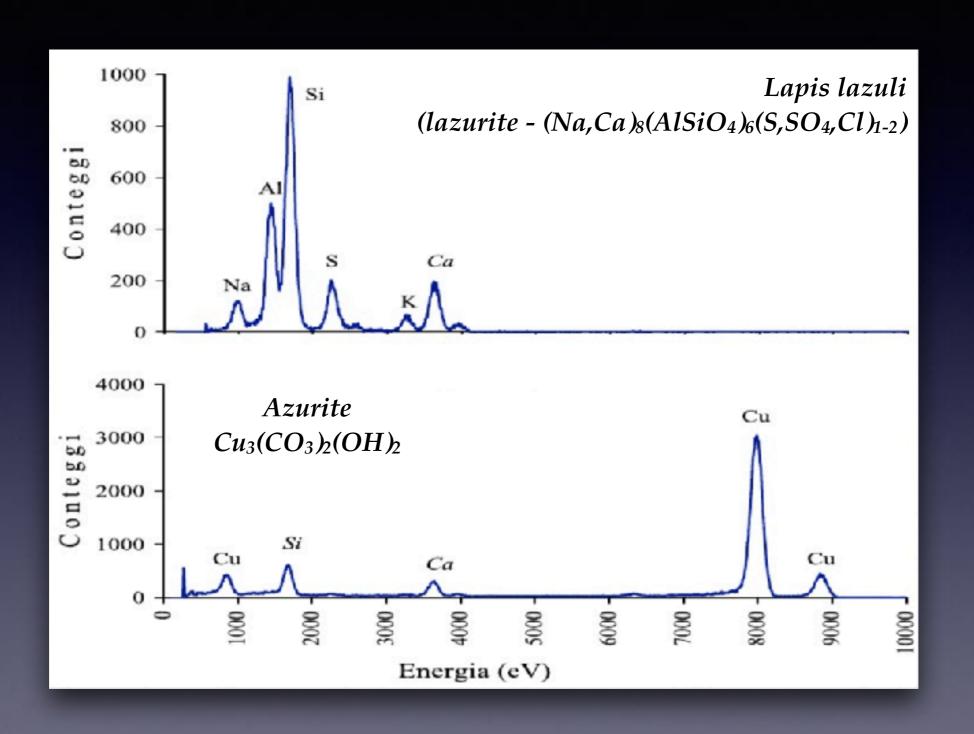
Antonello da Messina, Ritratto Trivulzio, Palazzo Madama, Turin

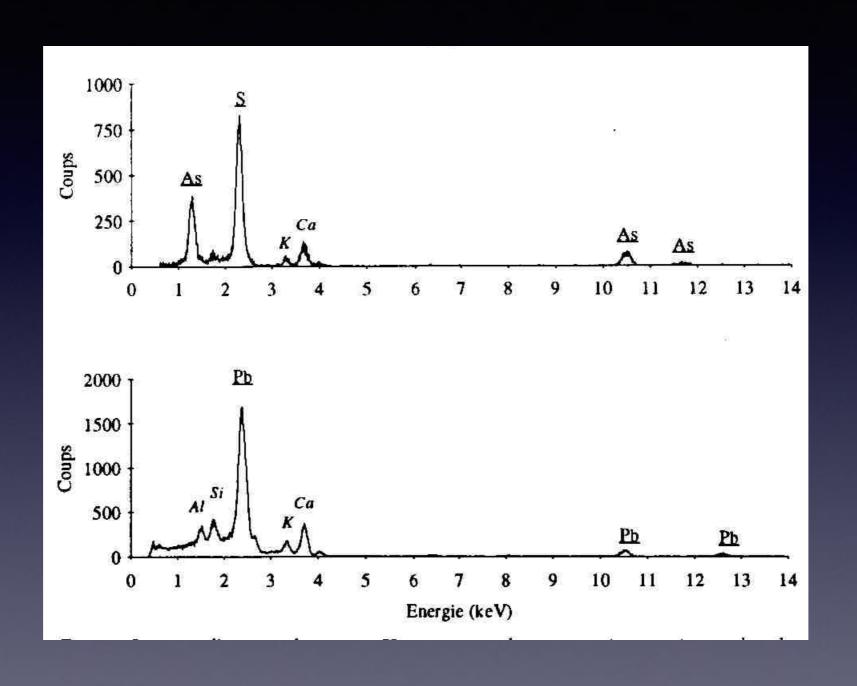
Some examples of artworks to be analyzed

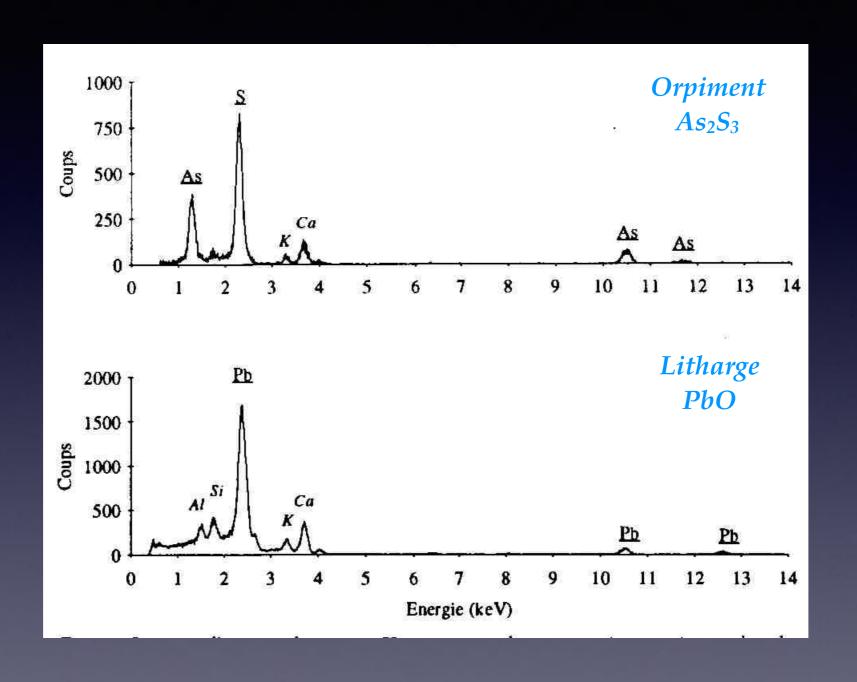








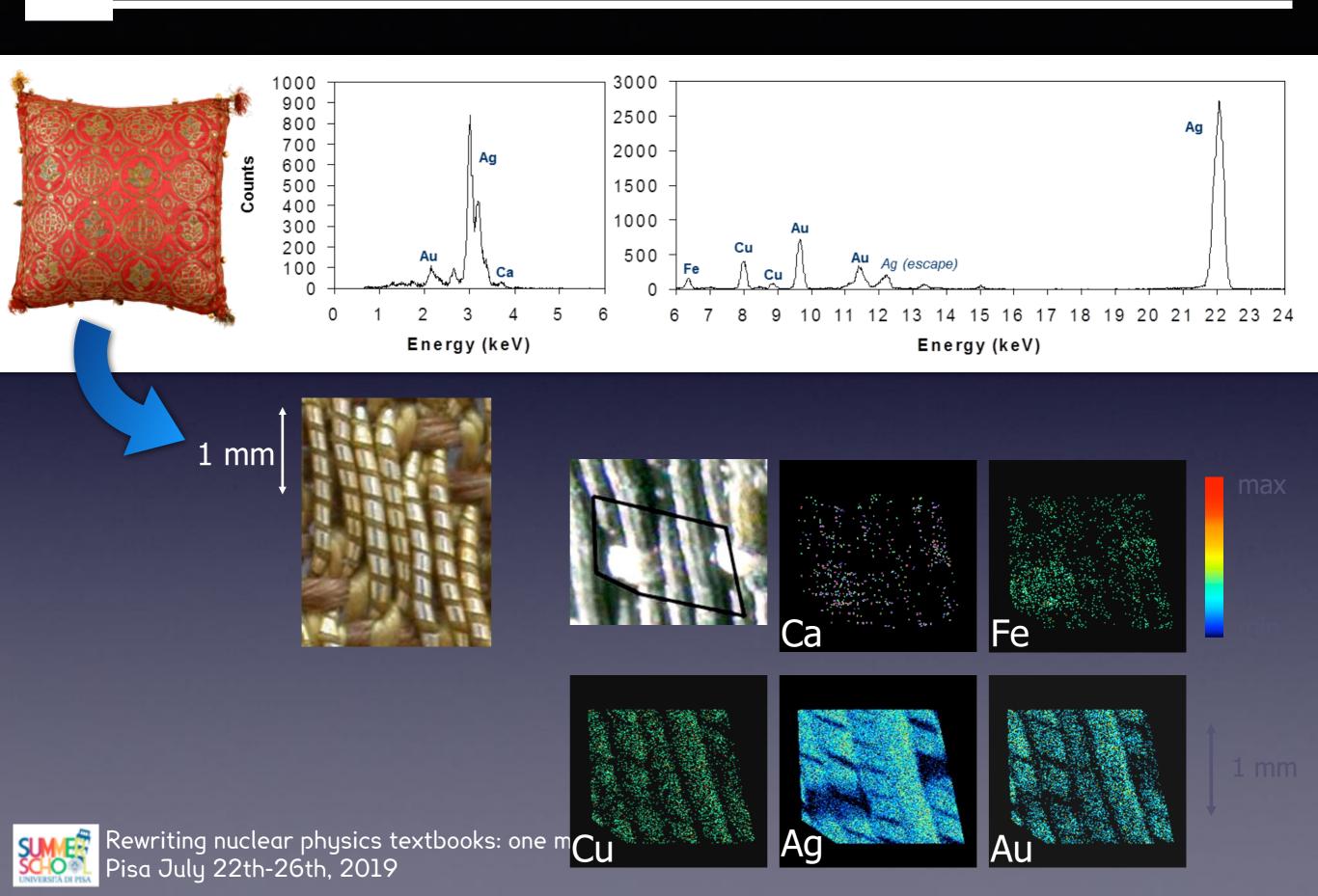




Scanning PIXE -> imaging

- We can exploit very focused ion beams through strong focusing by magnetic lenses \rightarrow about 10 μm FWHM on target (as order of magnitude)
 - Scan of the target in front of the beam or
 - Scan of the beam in front of the target (in this case, the scanning area is limited to the exit beam window dimensions)
 - ✓ For each of the analyzed point we can acquire both the X-rays energies and the point coordinates
 - \checkmark (x_i , y_i , E_z) to represent how the elements are distributed over the scanned area: elemental maps

Golden embroideries



"Madonna con il Bambino" by Mantegna



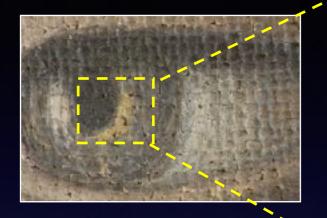
Accademia Carrara, Bergamo

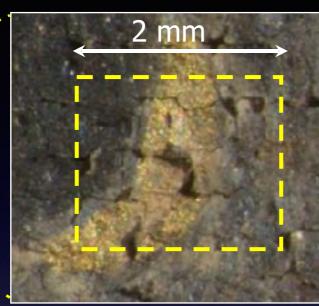
"Madonna con il Bambino" by Mantegnaino

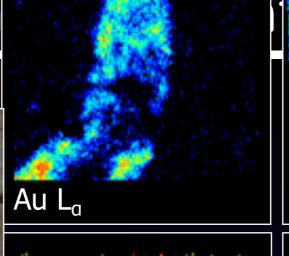


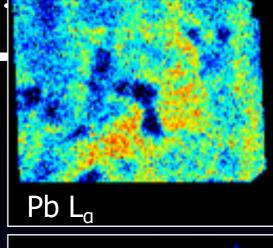


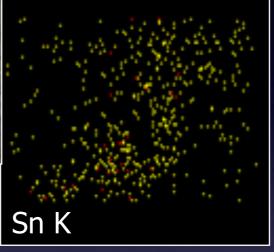
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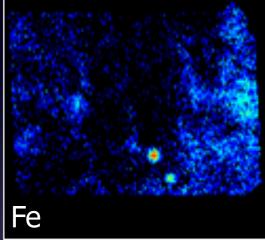




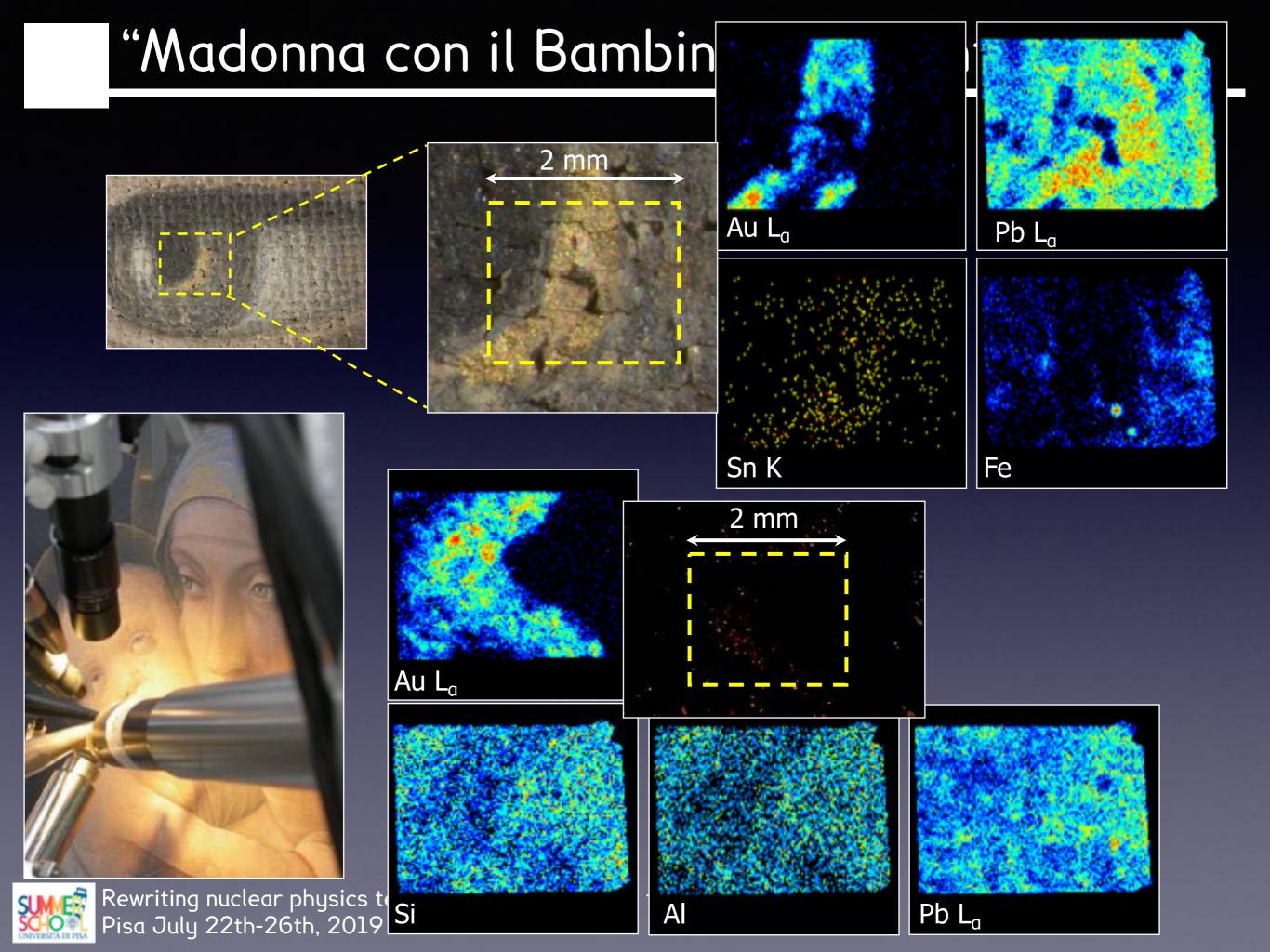














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 - Incident ions energy
 - Sample composition
 - Z of the analyzed element (i.e. collected x-ray energy)

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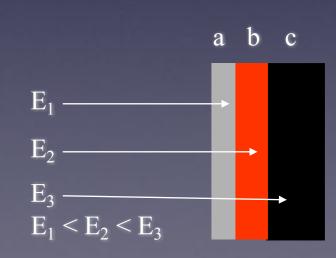
The information we get is integrated over the whole analyzed thickness

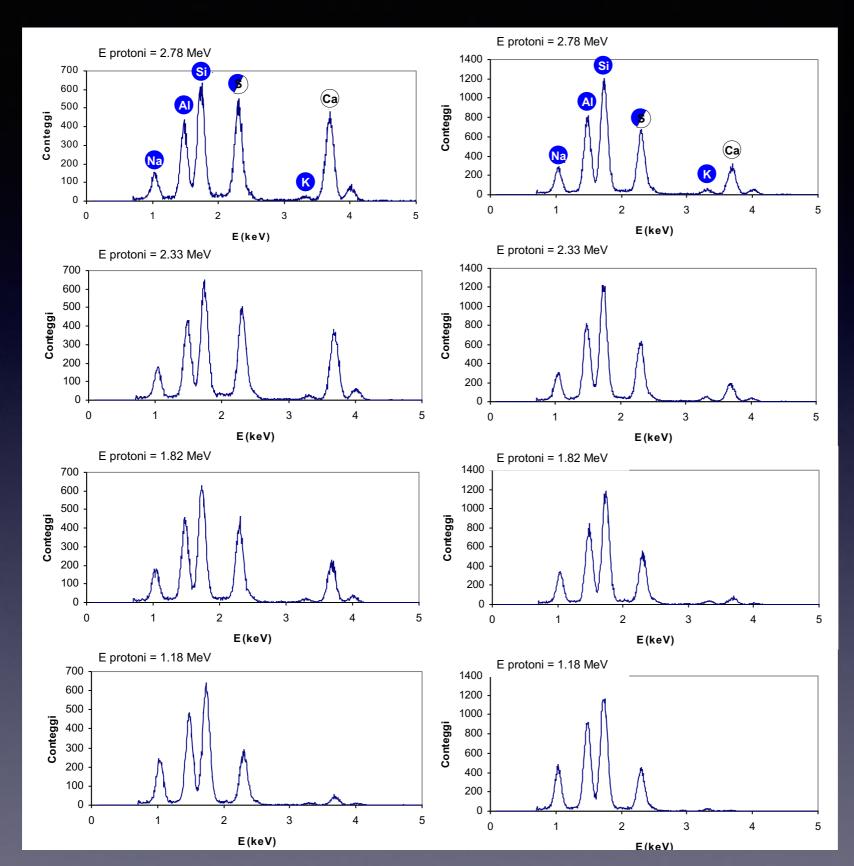
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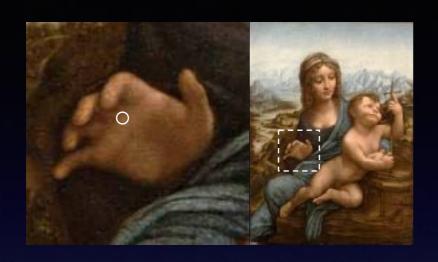
- If we change the incident proton energy:
 - Protons range inside the target changes
 - X-ray production cross section changes







An example from Leonardo



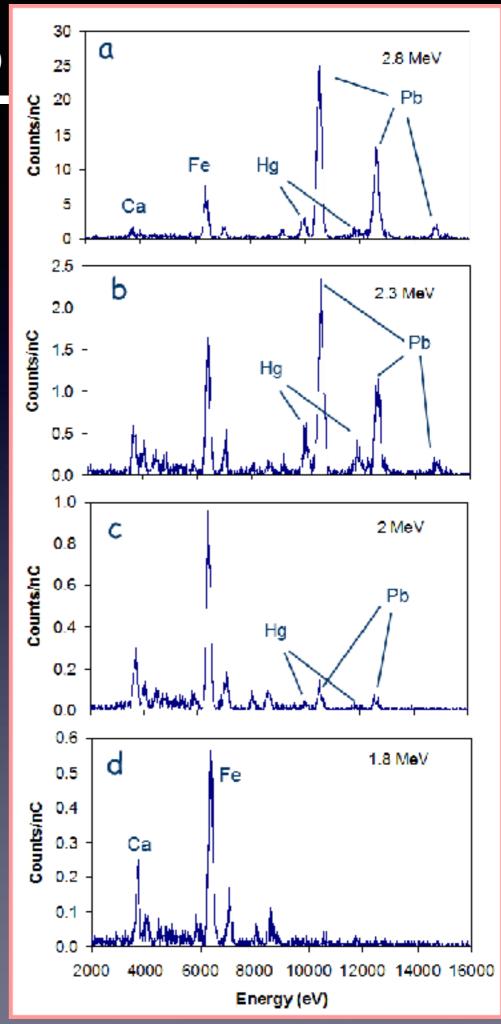
- Hg -> cinnabar (HgS)
- Pb -> white lead (2PbCO₃·Pb(OH)₂)

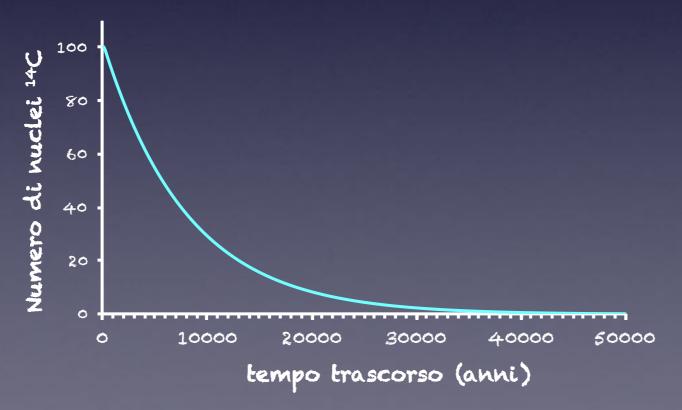
In b) and c), **Hg/Pb** increases with respect what happens in a): the preparation layer underneath is not hit by incident protons anymore -> no contribution by this layer to the total Pb counts.

In d), neither Hg or Pb are evident: protons stop in the superficial varnish layer

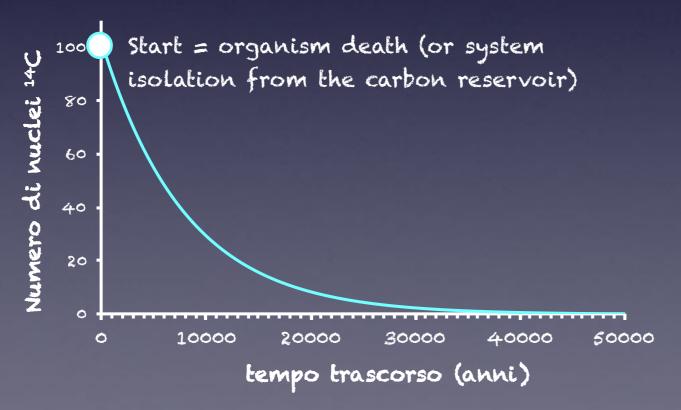
Fe, Ca -> varnish



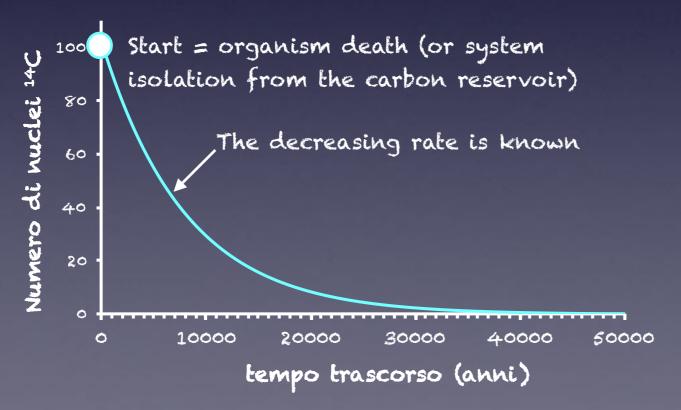






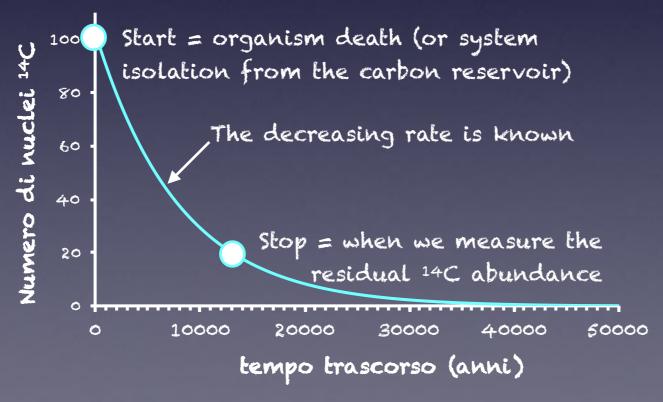






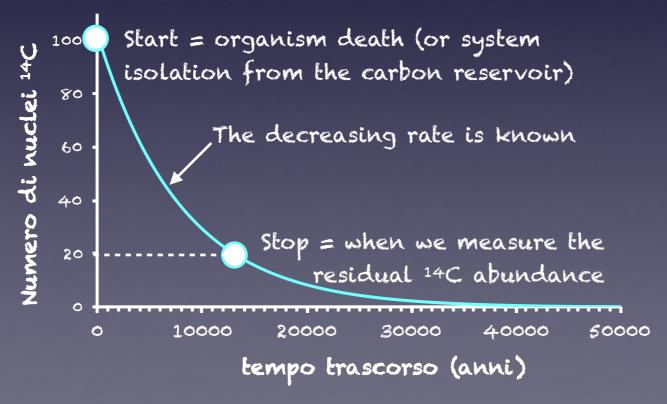


• 14C can be used as a natural chronometer to date organic materials, or - in more generally speaking - those systems that have been directly (or indirectly) in equilibrium with atmosphere, acquiring 14C from it, and that have ceased that equilibrium condition in a well defined moment



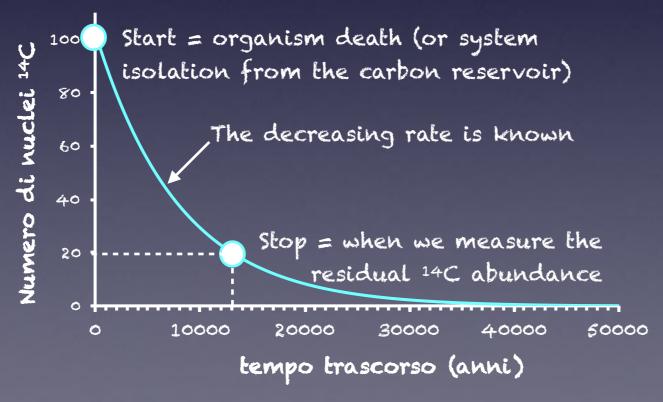


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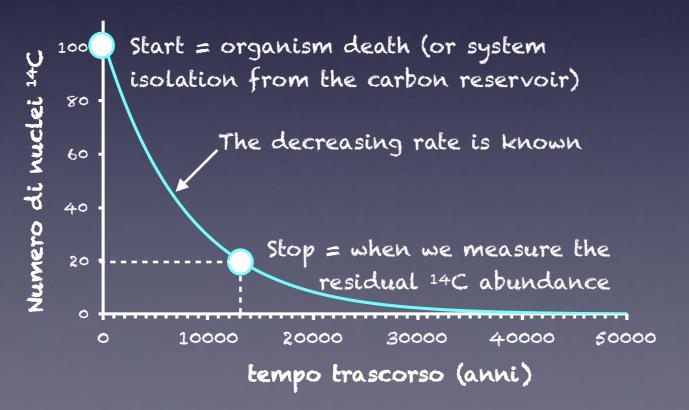


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condition in a well defined moment

Thanks to Nicolas Alamanos
for having explained the
principles of radiocarbon
dating in his introductory
lecture



Few pills of radiocarbon dating

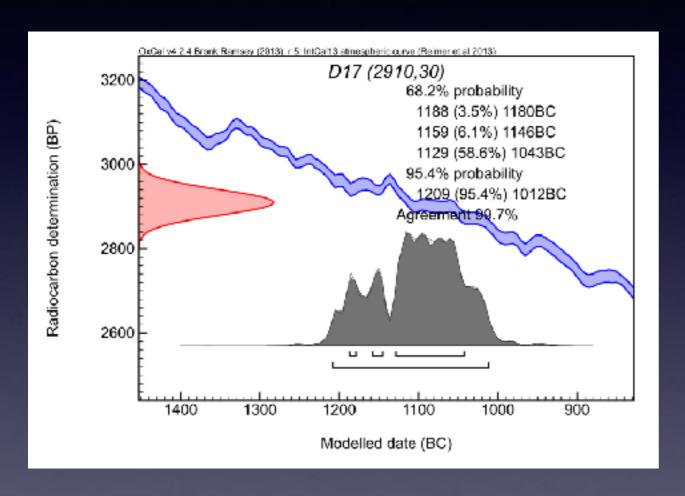
• From the measurement we obtain the conventional radiocarbon age (measured in years BP):

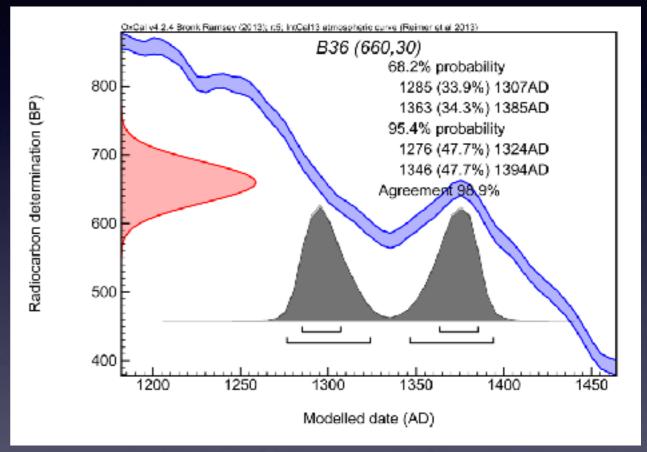
$$t_{rc} = \tau \cdot ln \frac{^{14}R}{^{14}R_0}$$

- T = 8033 years (Libby half life)
- According to the international agreement, $^{14}R_0$ is basically chosen as the radiocarbon concentration in 1950
- Pay attention to correct for isotopic fractionation
- Pay attention to calibrate the measured conventional radiocarbon age

We can date organic materials (or systems once in equilibrium with the atmosphere that have become close systems)
We always date the moment of the "death" of the organism

How a radiocarbon result looks like



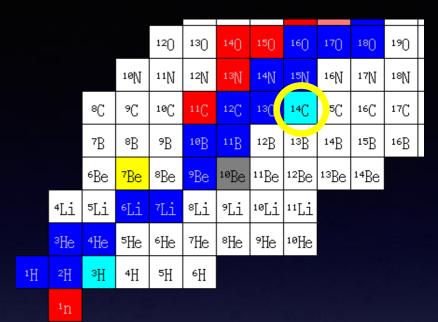


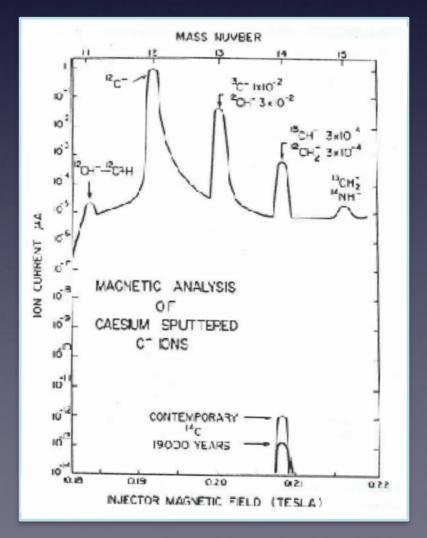
Why the accelerator

- ¹⁴C is a rare isotope
 - Low activity -> counting emitted beta is not favored
 - Low natural abundance (~10⁻¹² in the atmosphere) -> in principle measurable by mass spectrometric techniques
 - Presence of highly abundant isobars

	¹⁴ N	¹² CH ₂	¹³ CH	
Δm/m	0.000012	0.0009	0.0003	

Radiocarbon is a sort of "needle in a stack" We need a highly sensitive technique even when the sample is very small (few milligrams)

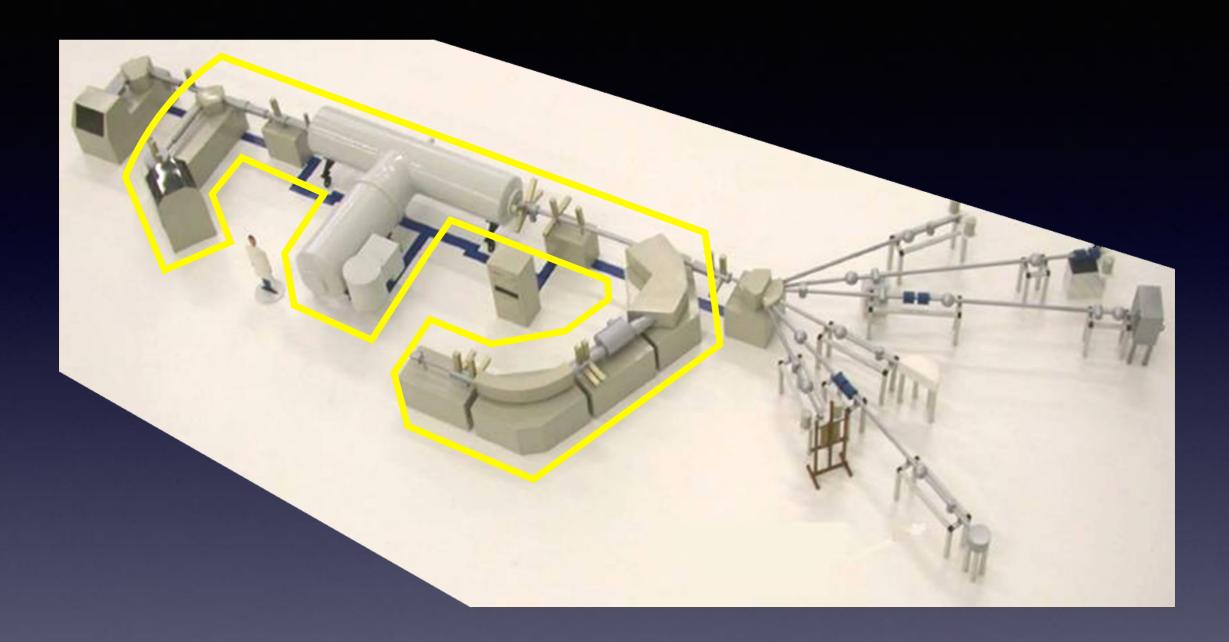




Accelerator Mass Spectrometry

• The key: using a tandem electrostatic accelerator to suppress the isobars abundance (both elemental and molecular interferences)

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The importance of being a tandem

Negative Ion Source:

-> some elements do not form stable negative ions -> suppression of elemental isobars

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MEASURE OF PHYSIQUE ARPLESS FOR

1000 12, OCTUBER 1977, PAGE 1487

AN ATTEMPT TO DETECT STABLE Nº IONS FROM A SPUTTER ION SOURCE AND SOME IMPLICATIONS OF THE RESULTS FOR THE DESIGN OF TANDEMS FOR ULTRA-SENSITIVE CARBON ANALYSIS

K. H. PURSER and R. B. LIEBERT General Jones Corporation Ipswich, MA 01933, U.S.A.

A. E. LITHERLAND and R. P. BBUKENS

University of Toronto(*)
Toronto, Ont., Canada

H. E. GOVE, C. L. BENNETT, M. R. CLOVER and W. E. SONDHEIM

Nuclear Structure Research Laboratory(**)
University of Reichester, Rochester, NY 14627, U.S.A.

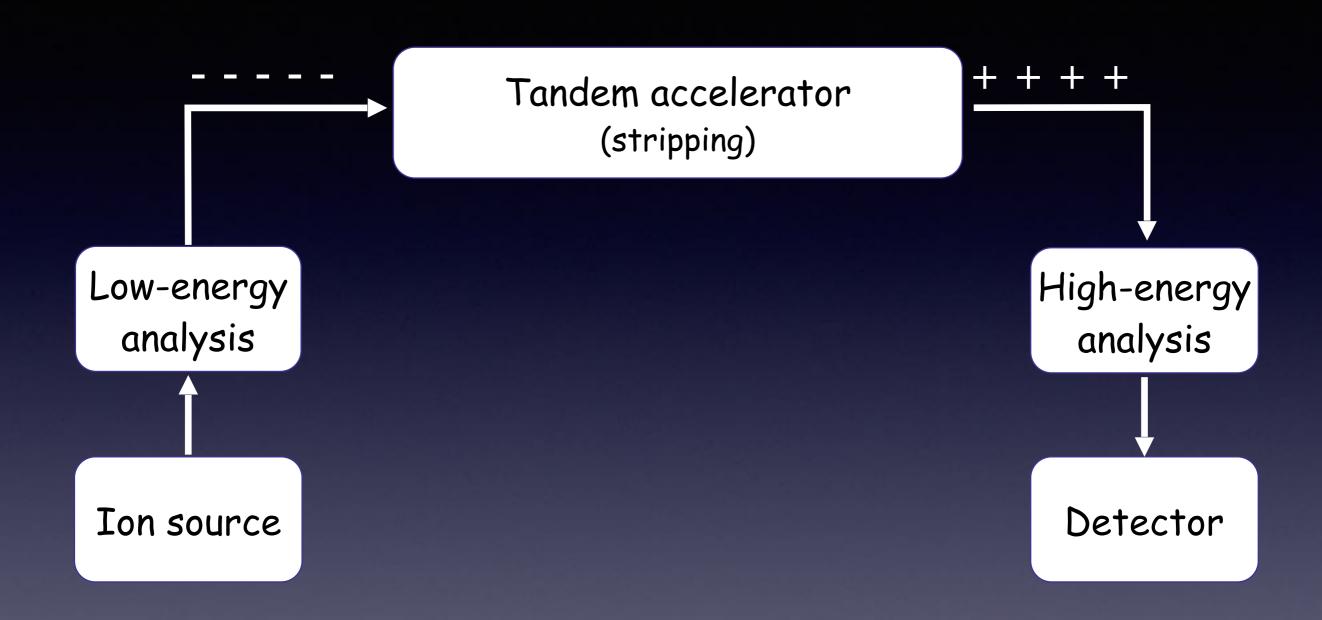
Béauné. — Dans acrès communication en diseate d'une captir error ni l'en fait une recherche d'iero "N" en présence d'un faisceau intense de carbone. On utilise une source d'un faisceau de paravre, Malgré un obamp accidénateur de 1.5 MV un nous n'avect par trouve d'iros autre negatifs avec des intendide désactables. La publication indique autre les conditions que deivent semplir l'accidénateur si on veut bénéficier pleinement de ce résultat négatif afin de faire des meastes avec des faisceaux très faibles de 140.

Abstract.— The possent paper discusses an experiment, using a Middleton spatter ton source, where a sensitive search was made for "N" into an the presence of an intense outloop beam. It was found that regarder through least, capable of without onling acceleration fields of "1.5 MVy", were not produced in detectable intensities. The paper also points out some design requirements of tandem facilities that must be satisfied if full advantage is to be taken of this argainer result to permit the measurement of very work "4" hours.

• Stripping:

-> after stripping at the high voltage terminal, molecules are not stable (if we consider charge >1+) -> suppression of molecular isobars

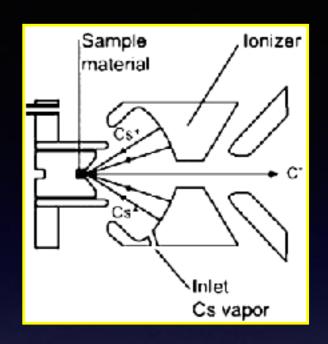
AMS: the general scheme



• Sensitivity on concentration measurement down to ~ 10^{-15} —> dating samples as older as 40000 years

About the negative ion source

- It is based on sputtering by Cs+ ions
- Negative ions (elemental, molecular, cluster...) are extracted through a voltage of few tens of keV



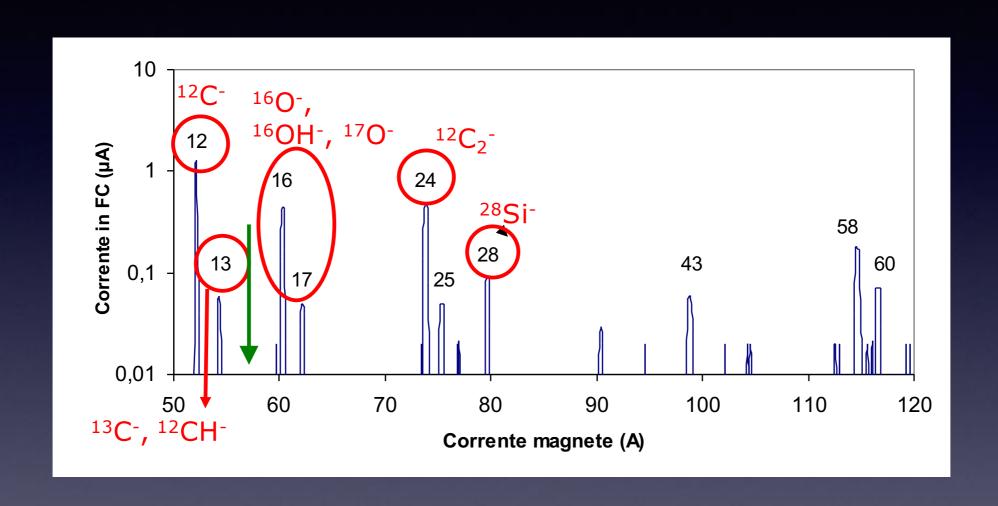
Not all the elements can be extracted as negative ions

Ionization Potentials and Electron Affinities of the Elements									
IV	IIA	IIIA	IVA	VA	VIA	VIIA	VIIIA		
₁ H 13.59 0.754			₂ He 24.48 0.078*						
зLi	4Be	5 B	6C	7N	80	9F	10Ne		
5.39	9.32	8.30	11.26	14.53	13.61	17.42	21.56		
0.618	0.195	0.277	1.263	-0.07	1.461	3.399	< 0		
11Na	₁₂ Mg	₁₃ Al	14Si	15 P	168	17 CI	18 A F		
5.14	7.64	5.98	8.15	10.48	10.36	13.01	15.76		
0.548	< 0	0.441	1.385	0.747	2.077	3.617	< 0		
19K	₂₀ Ca	31 Ga	₃₂ Ge	33A s	34Se	35Br	36 K r		
4.34	6.11	6.00	7.90	9.81	9.75	11.81	14.00		
0.501	0.043	0.30	1.2	0.81	2.021	3.365	< 0		
37 R b	39 Sr	49In	50Sn	51Sb	52Te	53 L	54Xe		
4.18	5.70	5.79	7.34	8.64	9.01	10.45	12.13		
0.486	< 0	0.3	1.2	1.07	1.971	3.059	< 0		
55 C/s	₅Ba	81TI	82Ph	83Bi	s ₄ Po	85A t	s₅l₹n		
3.89	5.21	6.11	7.42	7.29	8.42	9.5	10.75		
0.472	< 0	0.2	0.364	0.946	1.9	2.8	< 0		
*Metastable									

¹⁴N is suppressed

R. Middleton, A negative-ion cookbook (www.pelletron.com/cookbook.pdf)

Which ions inside the beam?



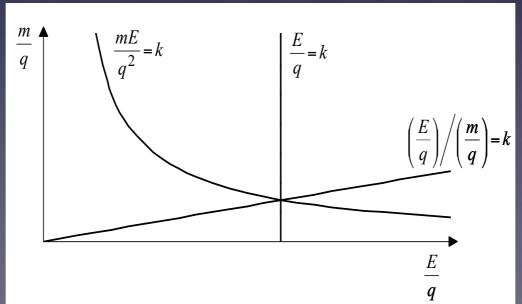
 We need some elements to discriminate the particles according to their mass, energy and charge

Beam analysis

• Electrostatic analysis:
$$m \frac{v^2}{r} = q \mathcal{E}$$
 \longrightarrow $\mathcal{E}r = 2 \frac{E}{q}$

• Magnetic analysis:
$$m \frac{v^2}{r} = qvB \longrightarrow Br = \frac{\sqrt{2mE}}{q}$$

• Wien filter:
$$q\mathscr{E} = qvB \longrightarrow \left(\frac{\mathscr{E}}{B}\right)^2 = 2\frac{E}{m}$$



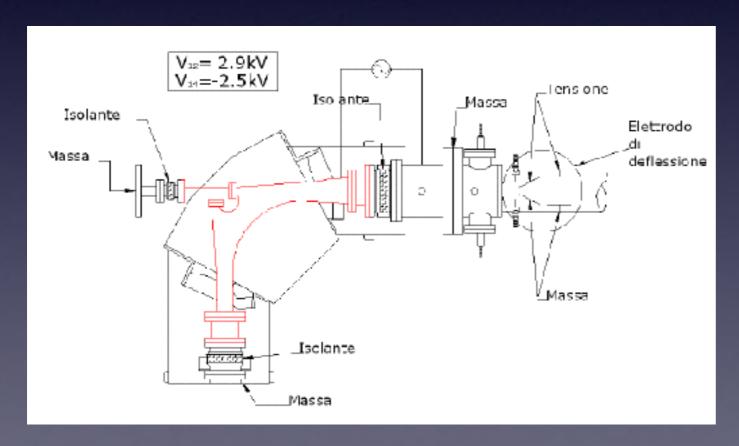
Each of the analysis identifies a locus of points

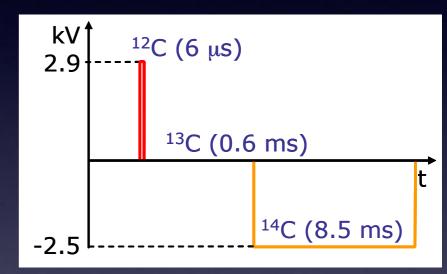
Bouncing injection

Magnet kept fixed to transmit mass 13

• Other masses injected thanks to different beam energies inside

the magnet chamber





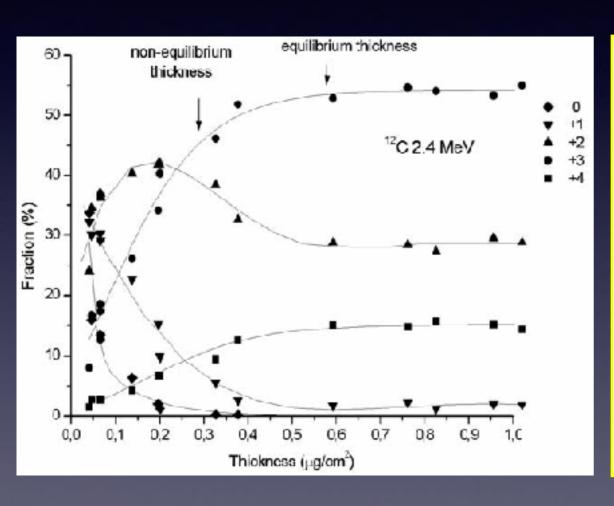
Injection times adjustable according to the average currents we would like to measure on the high energy side

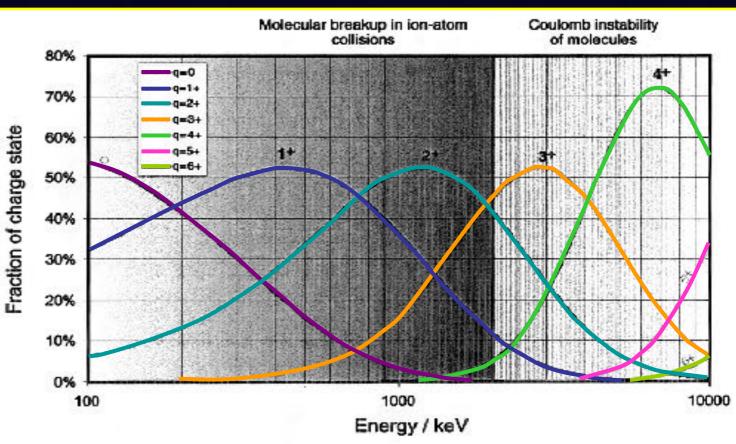
Molecular isobars are injected as well



The stripping process

 Distribution of charge state after stripping depends on stripper thickness and incident ion velocity



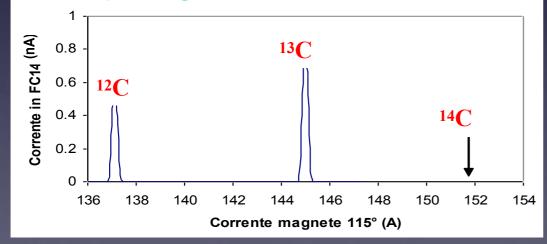


$$E_{fin} = \left(E_{inj} + eV_t\right) \frac{m_i}{m_{tot}} + qeV_t$$

 After stripping, the probability to find molecules characterized by a high charge state is low (Coulomb explosion) -> the capability to suppress isobars is due to the combination of terminal voltage, stripper thickness and charge state, which is analyzed on the high energy side

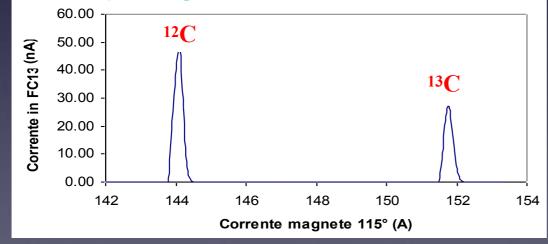
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When injecting mass 14

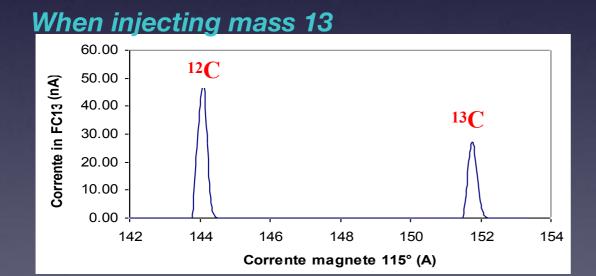


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When working at 2.5 MV, the charge state which is analyzed at the high energy side is the most probable i.e. 3+
Working at lower voltages, we should choose lower charge state → decreased capability of isobars suppression (increase of the equivalent stripper thickness)

About the rare isotope detector

- Possibility to measure the energy of the ions
- Possibility to use the detector just as a simple counter

Ionization chamber



Silicon diode



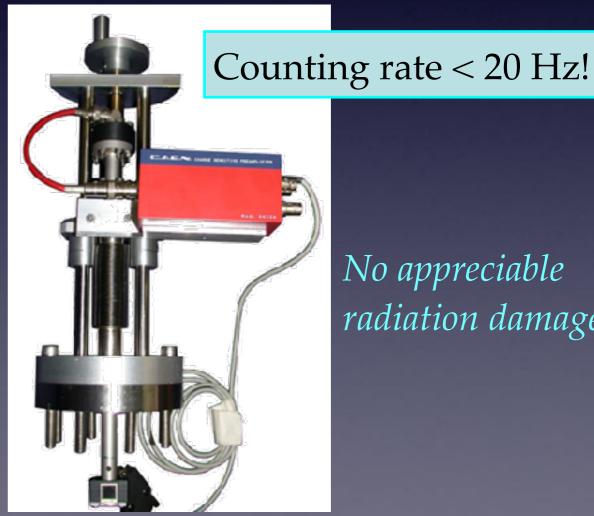
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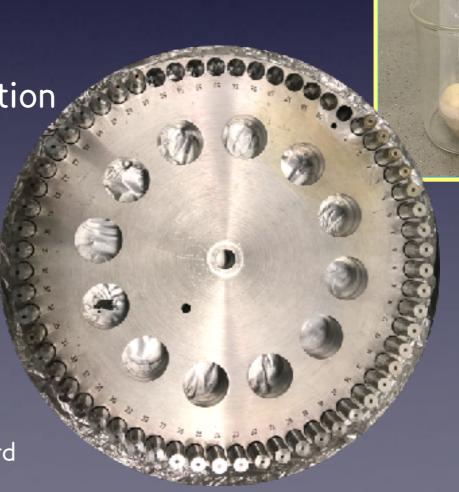


No appreciable radiation damage!

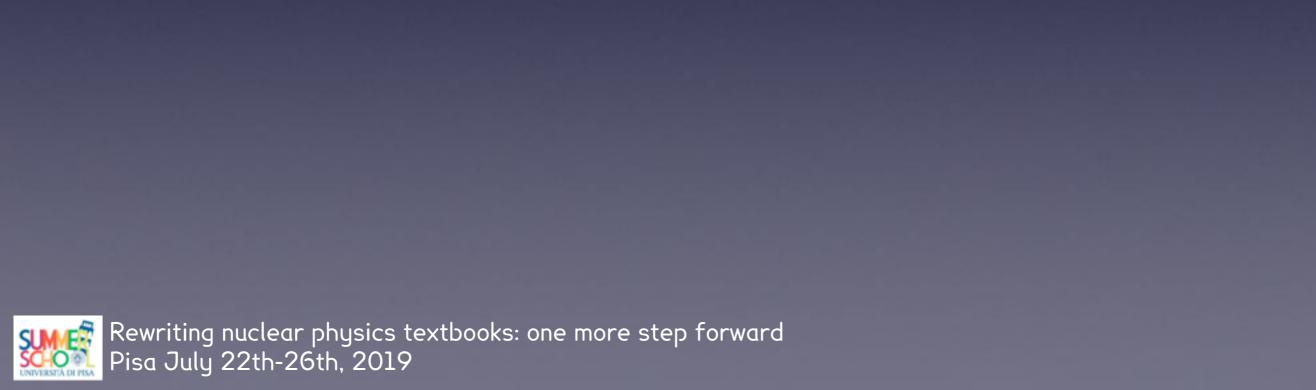
How to run a dating measurement

- Sample selection:
 - · Identification of the material
 - · Identification of the finding context (e.g. the archaeological excavation)
 - · Information on its preservation state
- Sample preparation
 - · To remove any possible contamination
 - To convert the cleaned material to the most appropriate chemical form (e.g. graphite)
- AMS measurement and data analysis





Archaeological sites



Archaeological sites







Baratti, Toscana – Etruscan and Roman-imperial periods





Erimi Laonin Tou Porakou, Cipro – Bronze Age and Hellenistic Age





Trittico di Badia a Rofeno - frame, Ambrogio Lorenzetti



F. Léger (attr.), Contraste des Formes



Trittico di Badia a Rofeno - frame, Ambrogio Lorenzetti



F. Léger (attr.), Contraste des Formes



Trittico di Badia a Rofeno - frame, Ambrogio Lorenzetti



U. Boccioni, L'Idolo Moderno



Rewriting nuclear physics textbooks: one mo Pisa July 22th-26th, 2019



Contraste de formes
Peggy Guggenheim Collection
(never shown to the public)



Contraste de formes Peggy Guggenheim Collection (never shown to the public)

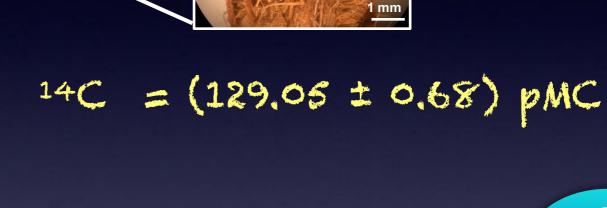




14C = (129.05 ± 0.68) pMC

Contraste de formes Peggy Guggenheim Collection (never shown to the public)

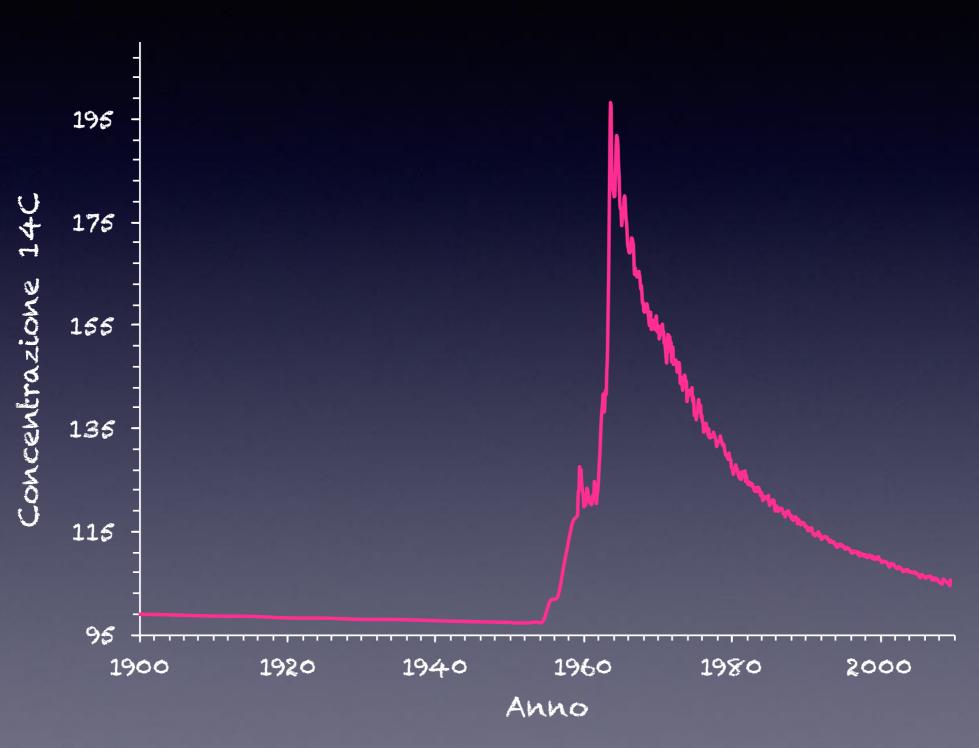




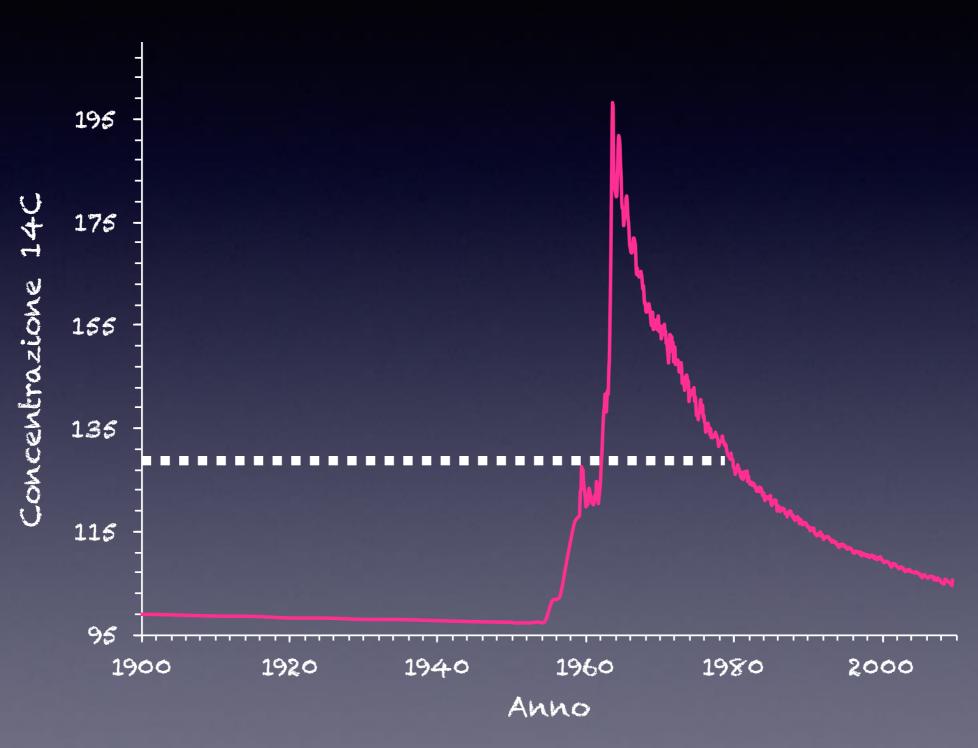


Ma per un campione moderno $^{14}C \sim 100 \ pMC...$

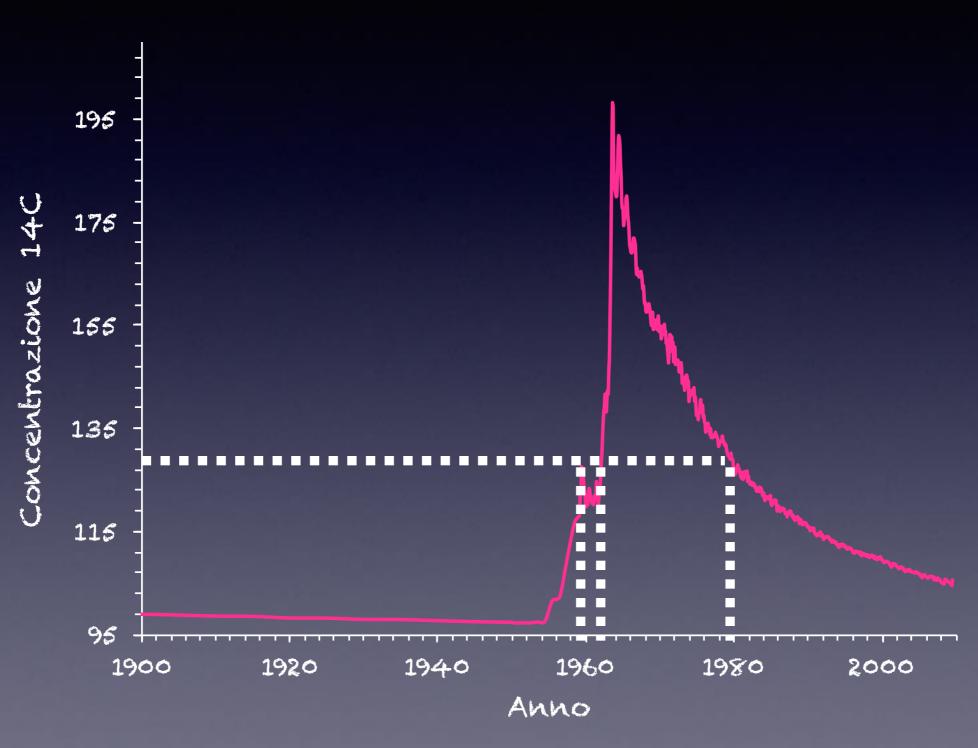
Contraste de formes Peggy Guggenheim Collection (never shown to the public)





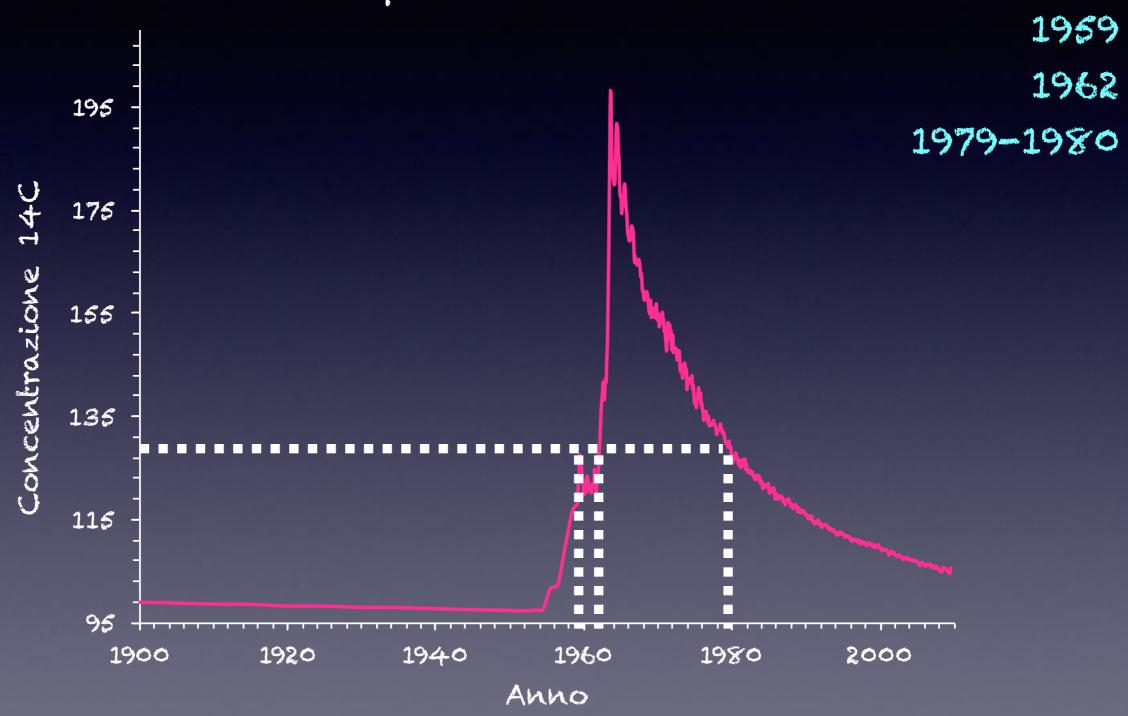




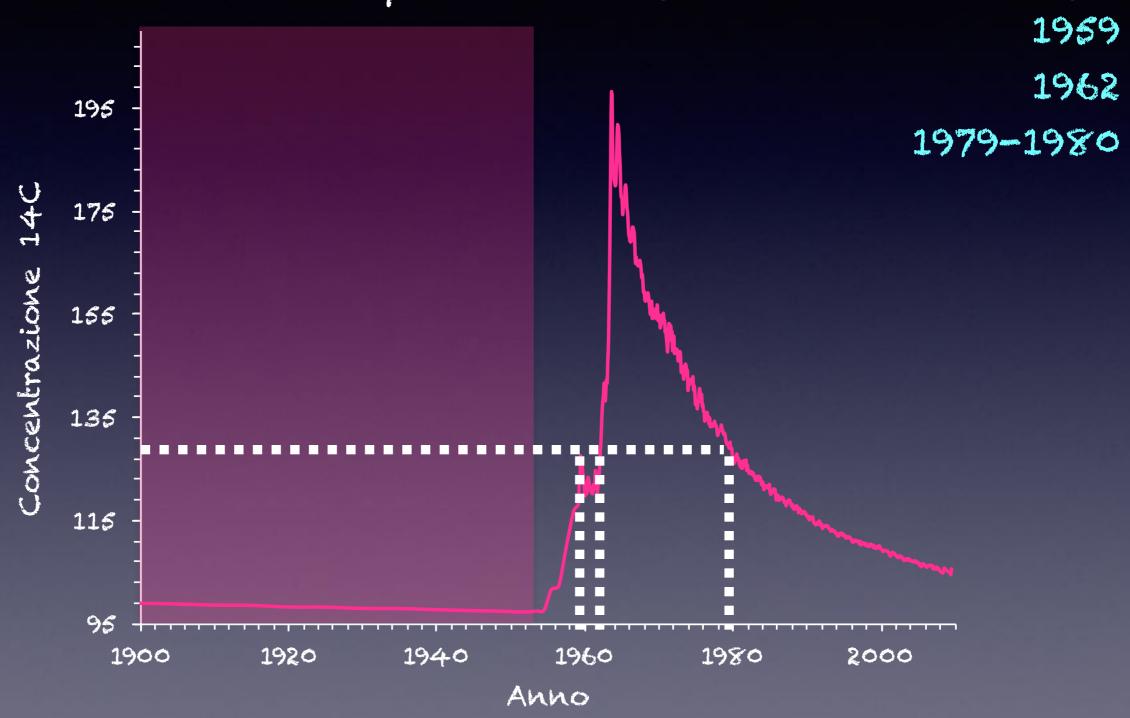




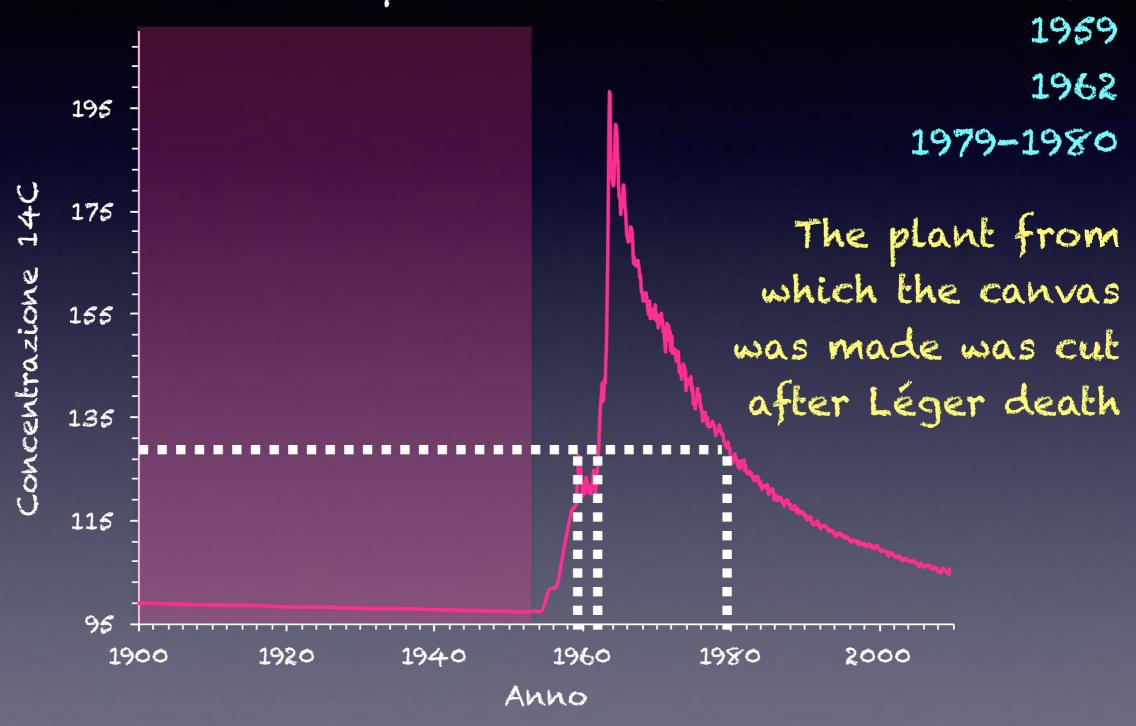
Three probable ranges of calibrated age:



Three probable ranges of calibrated age:



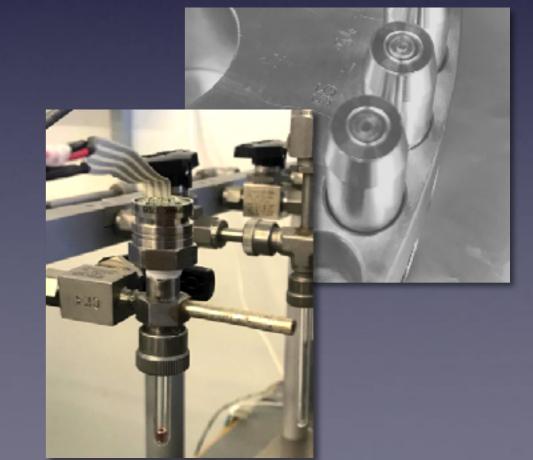
Three probable ranges of calibrated age:



Where the research is going to?

- IBA and PIXE in particular:
 - Small and compact accelerators to be transportable (also to be installed in national restoration facilities)
 - -> MACHINA project (INFN & CERN): acceleration is achieved through radio-frequency quadrupole technology (<u>qiuntini@fi.infn.it</u> - <u>ftaccetti@fi.infn.it</u>)
- 14C-AMS:
 - Smaller terminal voltages
 - Single stage AMS
 - Smaller samples -> graphite samples down to few tens of µg
 - Handling of big data -> Ariadne+







Thanks to all colleagues from LABEC and CHNet

(in particular Lucia Liccioli e Serena Barone)

...and many thanks to you!

