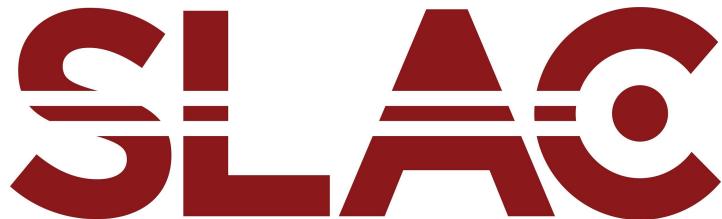


Lessons from the beamline: The TES detector at beamline 10-1

Charles J. Titus



SLAC

Sang-Jun Lee
Dennis Nordlund
Hsiao-Mei Cho
Dale Li



Stanford

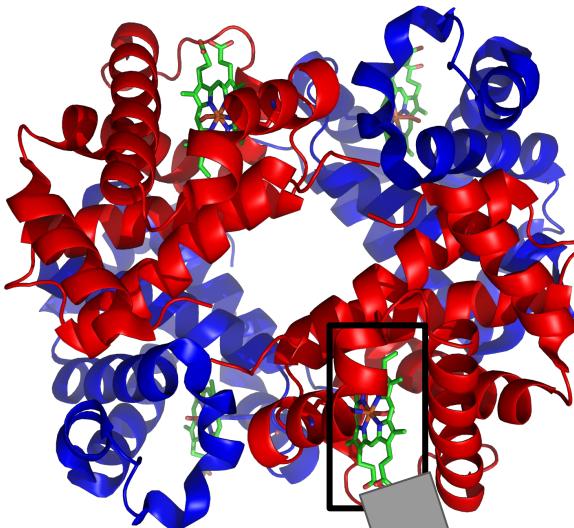
Charles Titus
Kent Irwin



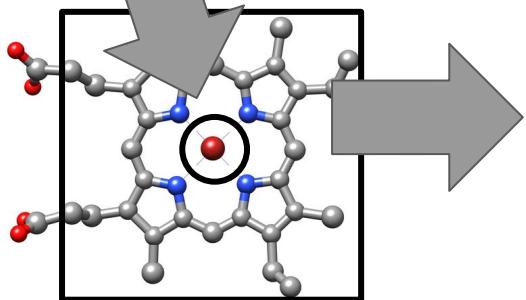
NIST

Douglas Bennett	Kelsey Morgan
W. Bertrand Doriese	Galen O'Neil
Joseph Fowler	Carl Reintsema
Johnathon Gard	Daniel Schmidt
Gene Hilton	Daniel Swetz
Young Il Joe	Joel Ullom

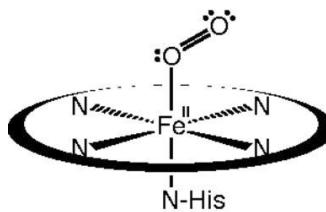
How does hemoglobin transport oxygen?



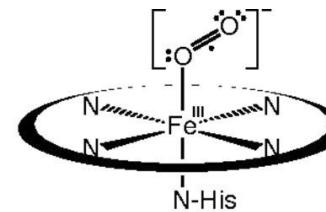
How does iron bind to oxygen in hemoglobin?
Pauling's first model published in **1936**. Can X-ray spectroscopy solve this **80-year old mystery**?



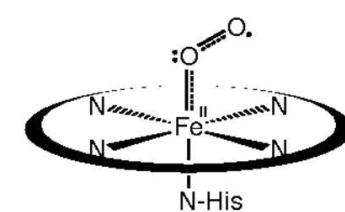
Controversy: whose model is best?



Pauling
 $\text{Fe}^{II} (S = 0) \cdot \text{O}_2 (S = 0)$



Weiss
 $\text{Fe}^{III} (S = 1/2) \cdot \text{O}_2^- (S = 1/2)$



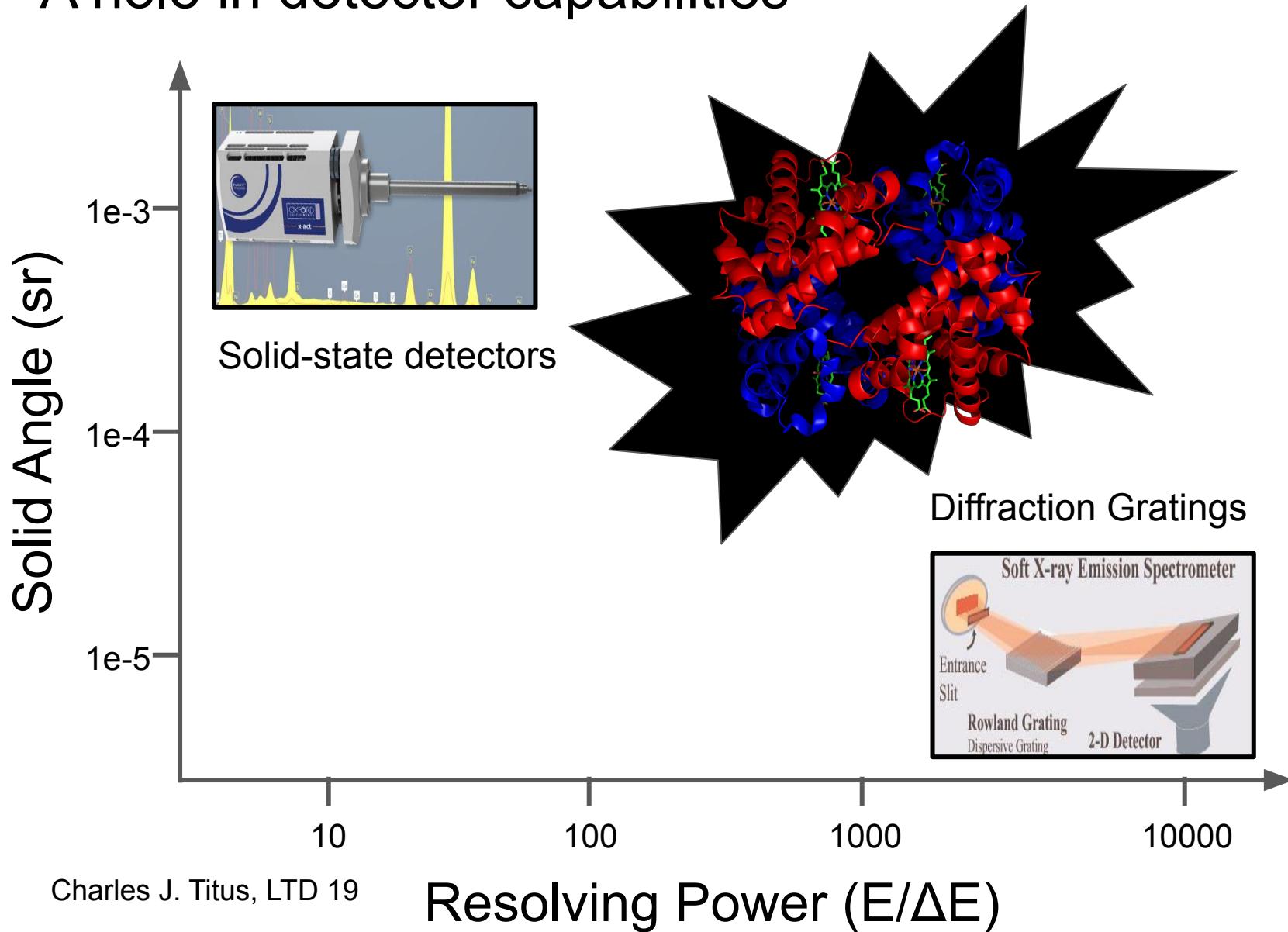
McClure-Goddard
 $\text{Fe}^{II} (S = 1) \cdot \text{O}_2 (S = 1)$

Wilson, S. A., et al (2013). PNAS, 110(41)

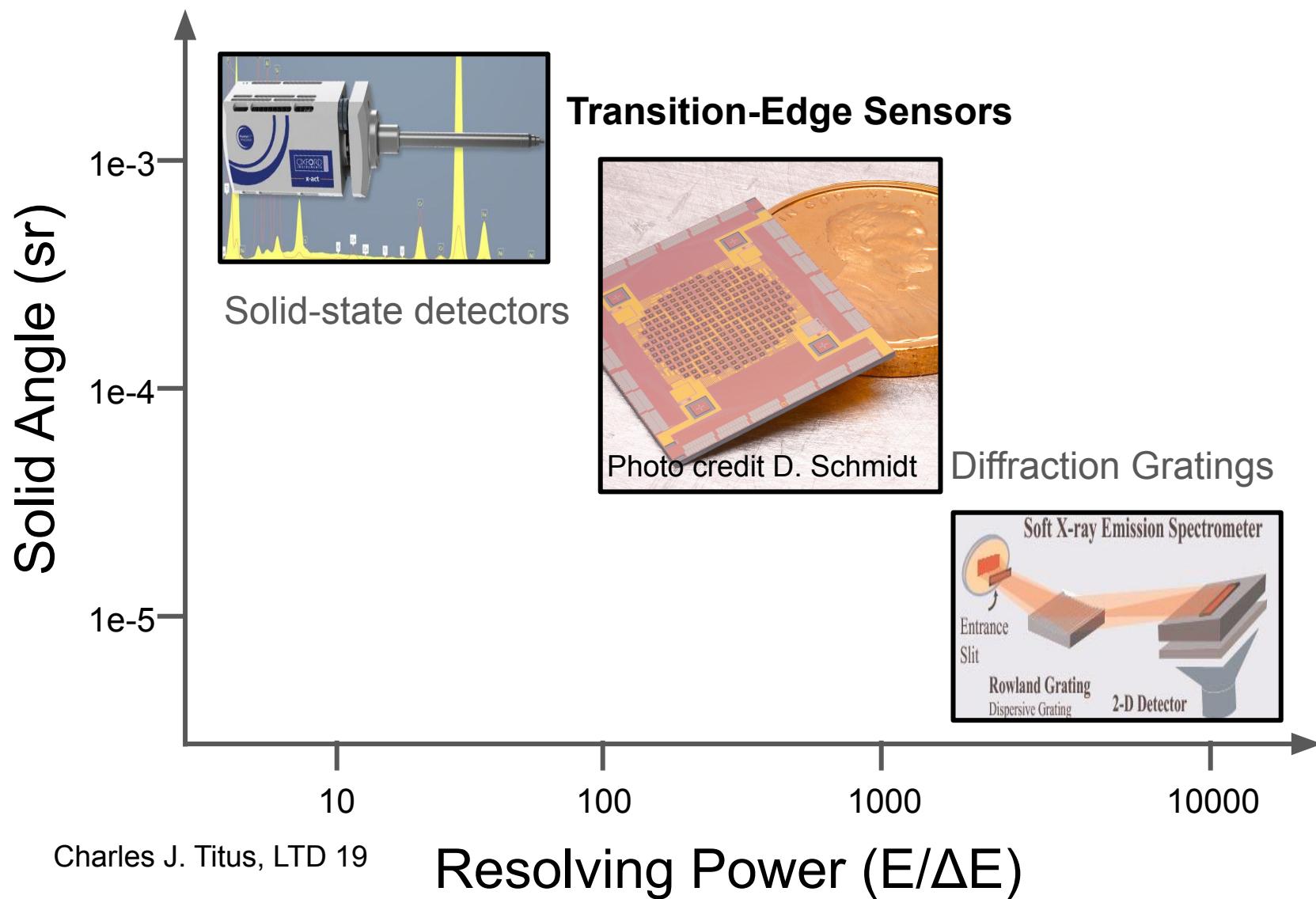
Outline

- **The TES advantage**
- Science at SSRL using a TES
 - Understanding battery degradation
 - Detecting contaminants in carbon nanotubes
 - Measuring the Fe-O bond in hemoglobin

A hole in detector capabilities



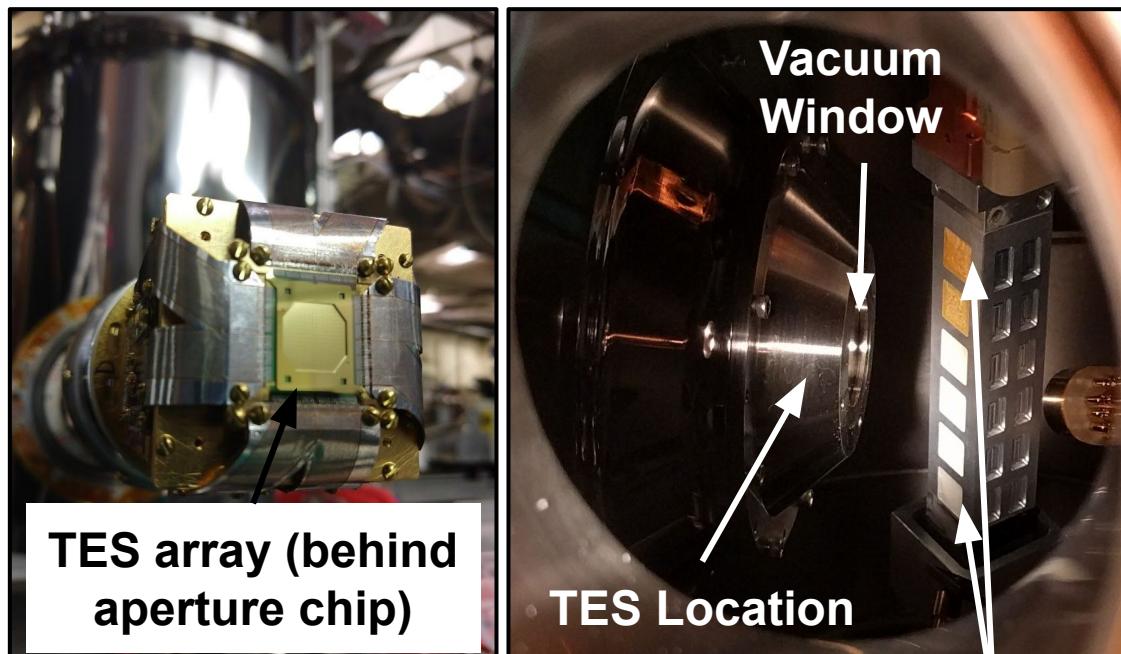
Can be filled by TES detectors!



TES at SSRL BL 10-1 available for users since 2017

Array Specs

Pixels	240
Solid Angle	$2e-3 \text{ sr}$
Energy range	200-2000 eV



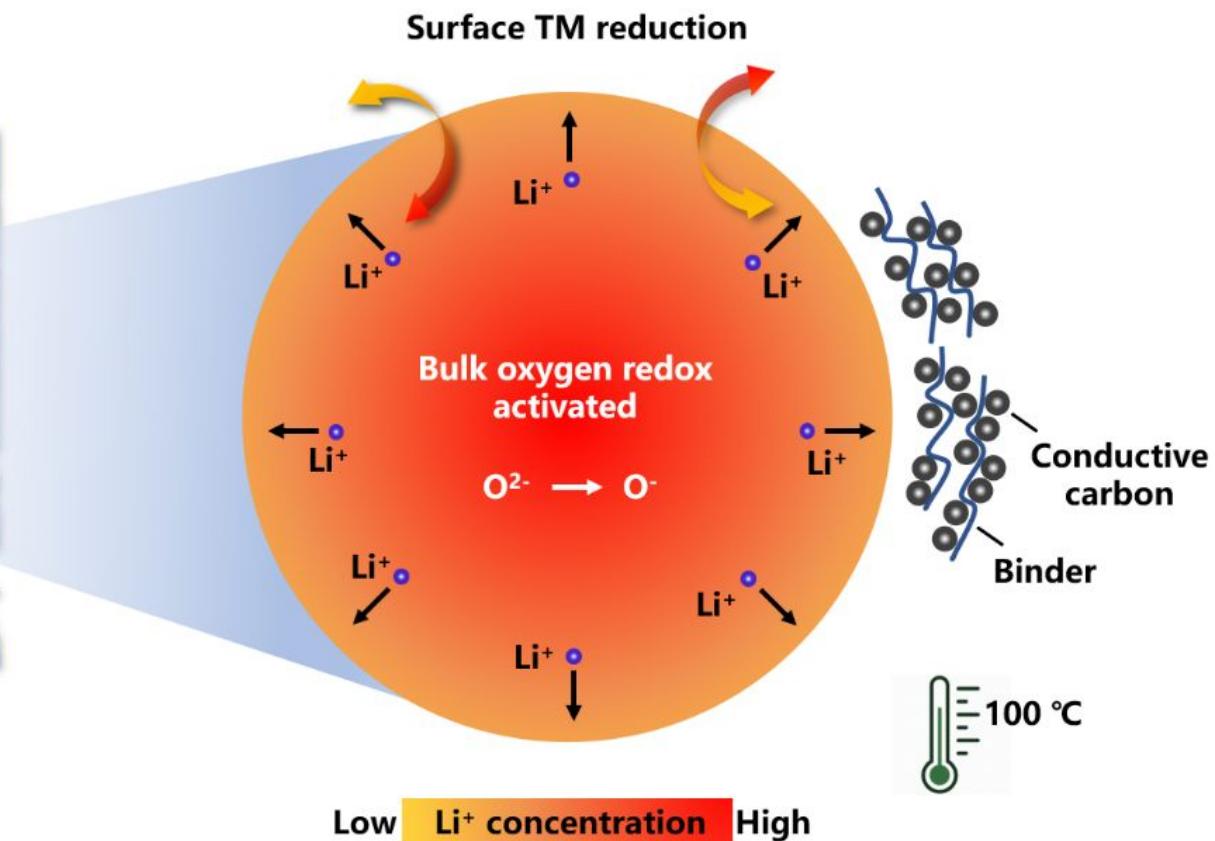
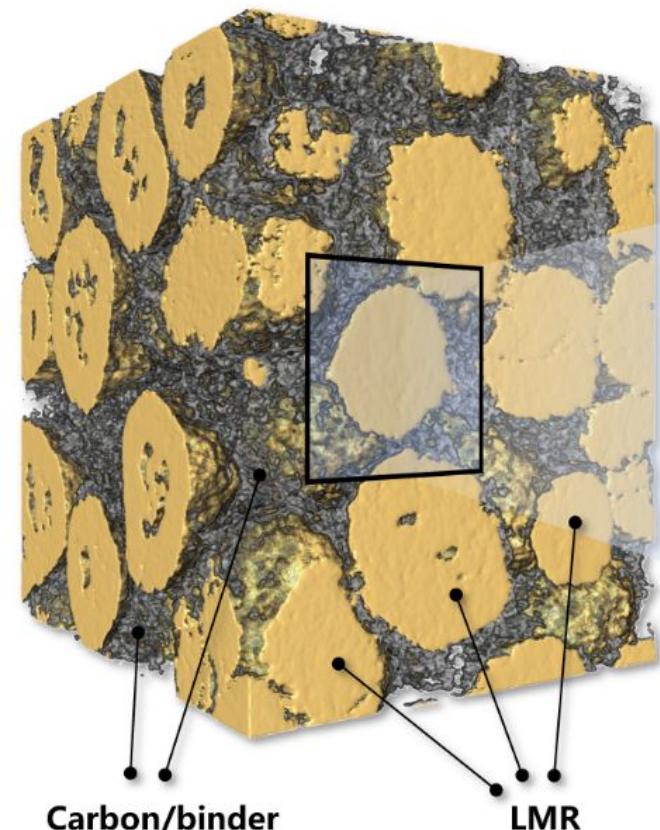
Pixel Specs

Mo/Cu bilayer thickness	60 nm/200 nm
T _c	107 mK
Energy Resolution	1.5 eV @ 500 eV
Absorber thickness	2.8 um Bi

Samples

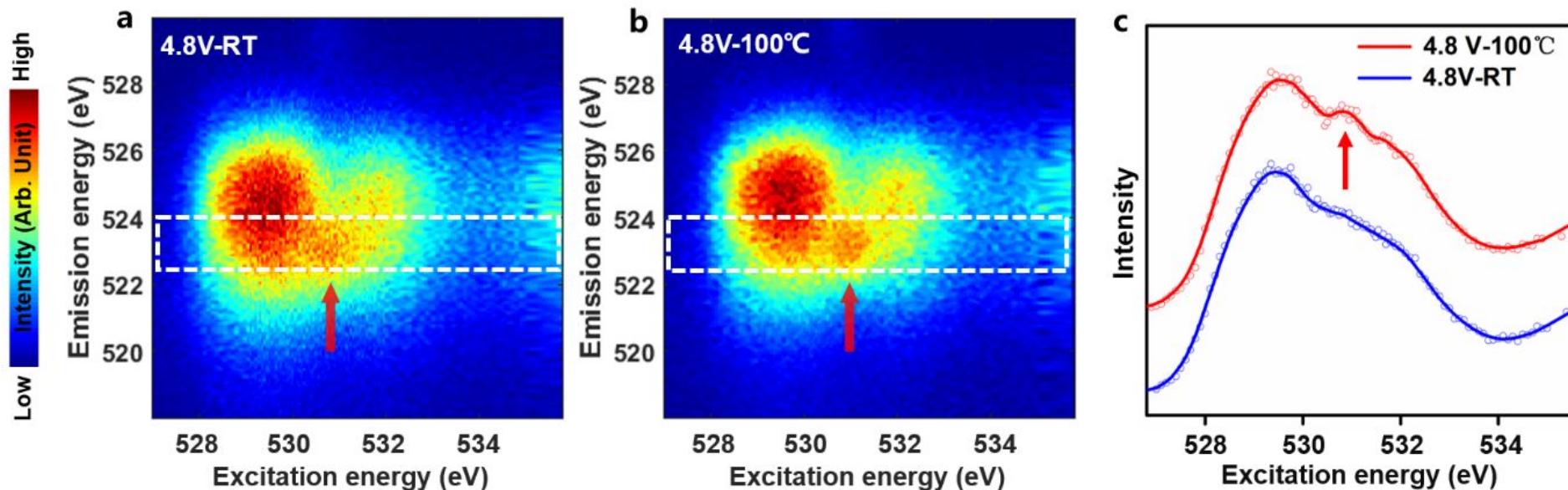
System details in
Doriese, et al (2017). Rev. Sci.
Instrum., 88(5), 53108.

What causes thermal degradation of batteries?



TES reveals that oxidation is the culprit

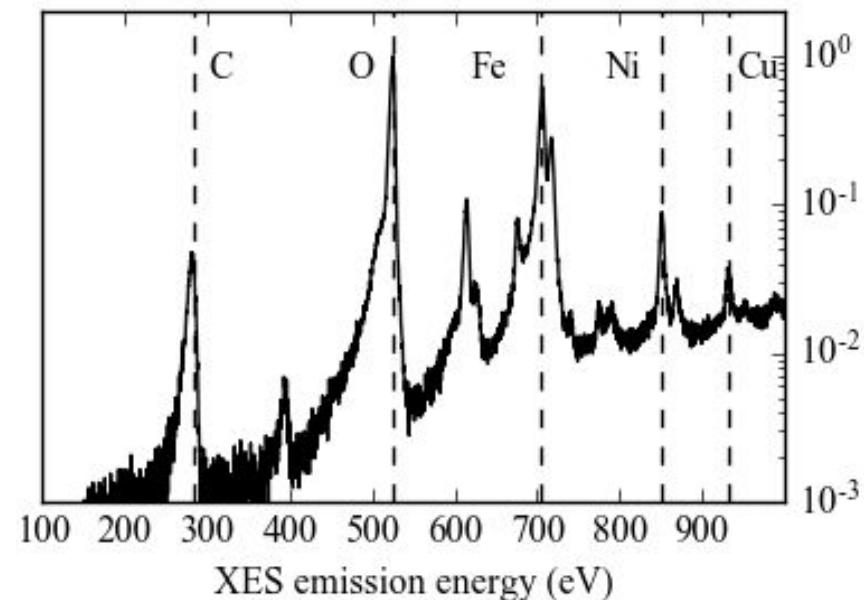
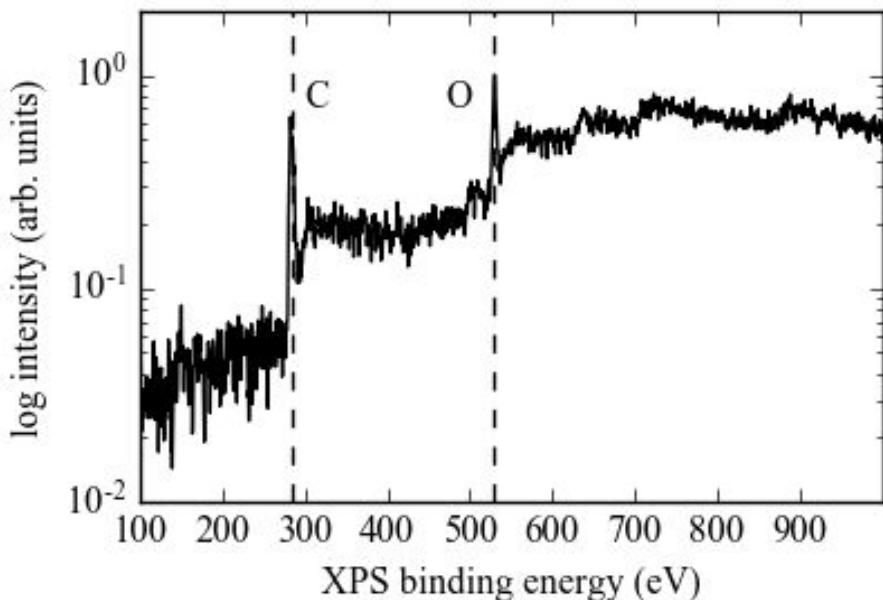
- Small changes in oxygen spectra indicate oxidation
- This work possible at SSRL for the first time, due to TES



What is contaminating carbon nanotube devices?

- TES shows a bunch of weird metals!
- Contaminants (<0.1%) affect device function
- TES more sensitive than other common probes (XPS)

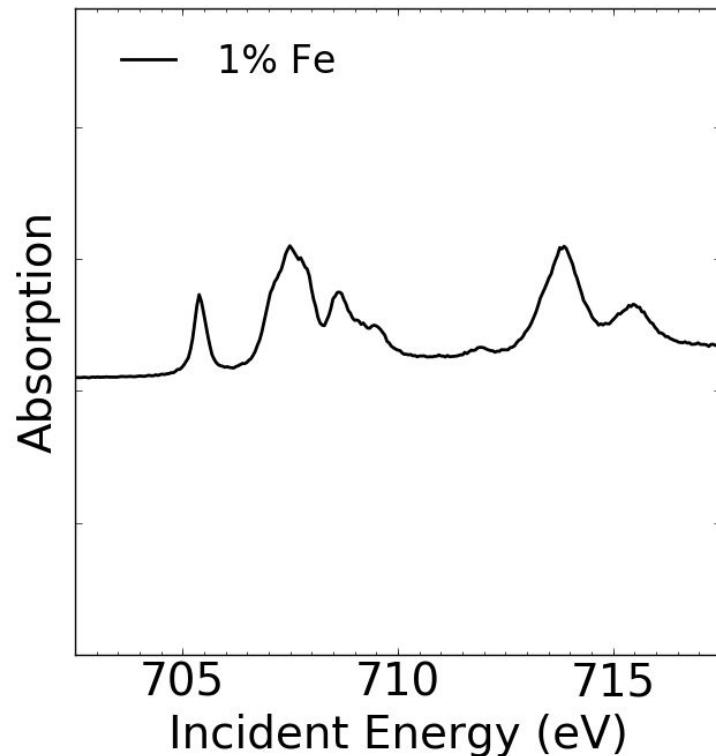
XPS Composition Analysis vs TES Composition Analysis



How well can we measure ultra-dilute samples?

XAS of diluted $\text{Fe}(\text{CN})_6$

Atomic %	Examples
>1% (concentrated)	Batteries, molecules, bulk solids

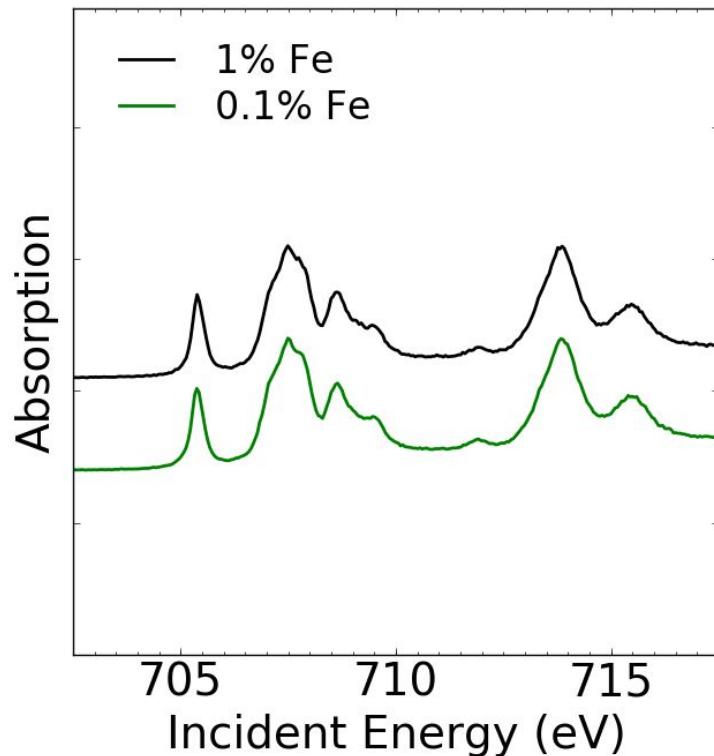


Titus et al, *J. Chem. Phys.* 147, 214201 (2017)

How well can we measure ultra-dilute samples?

Atomic %	Examples
>1% (concentrated)	Batteries, molecules, bulk solids
0.1% (dilute)	Contaminants, solutions

XAS of diluted $\text{Fe}(\text{CN})_6$

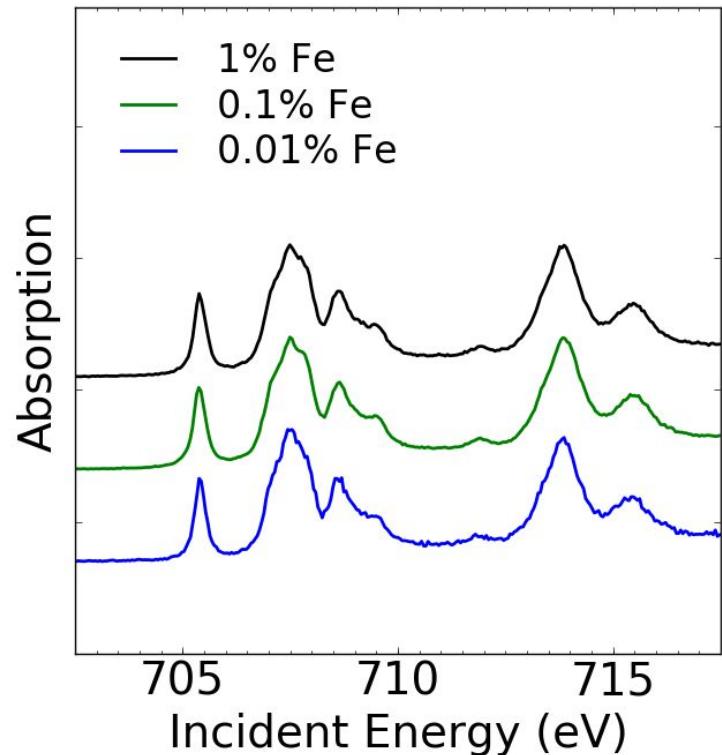


Titus et al, *J. Chem. Phys.* 147, 214201 (2017)

How well can we measure ultra-dilute samples?

XAS of diluted $\text{Fe}(\text{CN})_6$

Atomic %	Examples
>1% (concentrated)	Batteries, molecules, bulk solids
0.1% (dilute)	Contaminants, solutions
0.01% (very dilute)	Hemoglobin, many proteins, monolayers

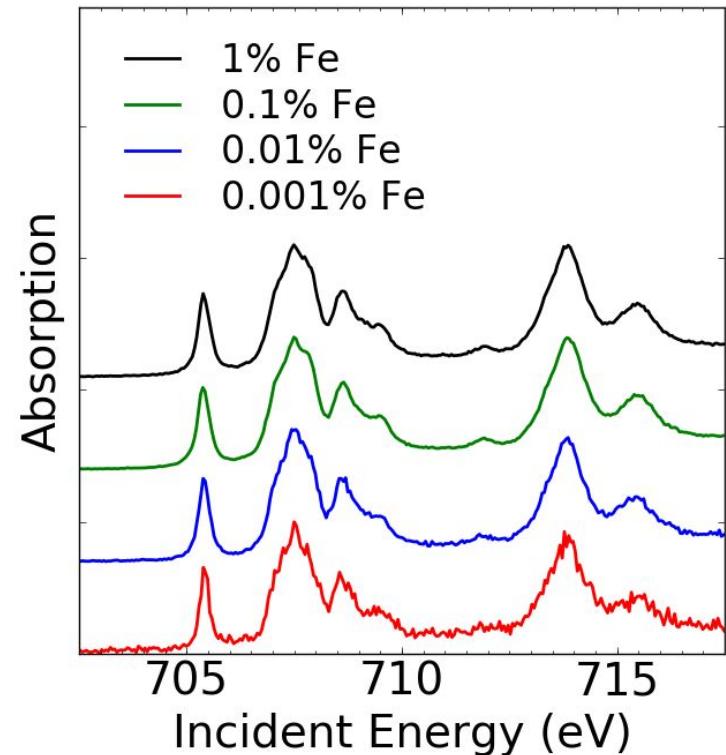


Titus et al, *J. Chem. Phys.* 147, 214201 (2017)

Excellent measurements of ultra-dilute samples

XAS of diluted $\text{Fe}(\text{CN})_6$

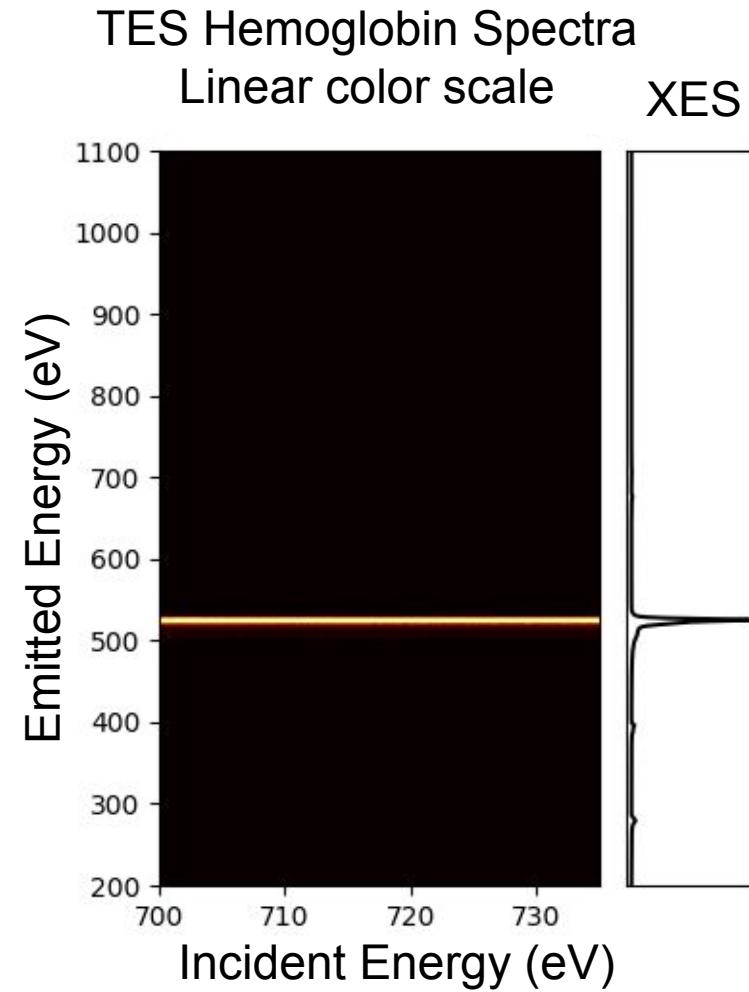
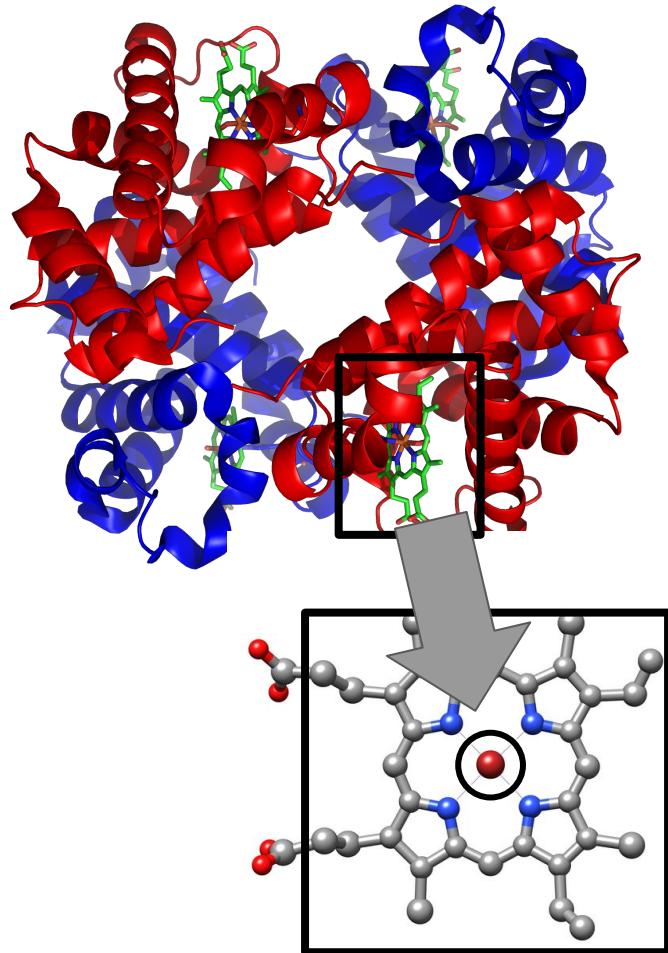
Atomic %	Examples
>1% (concentrated)	Batteries, molecules, bulk solids
0.1% (dilute)	Contaminants, solutions
0.01% (very dilute)	Hemoglobin, many proteins, monolayers
0.001% (ultra-dilute)	Photosystem-II, doped monolayers, trace impurities, color centers



New capability: 0.001% measurement opens many doors!

Titus et al, *J. Chem. Phys.* 147, 214201 (2017)

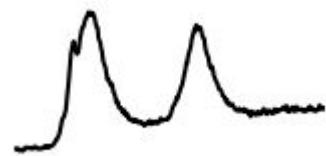
So can we measure hemoglobin?



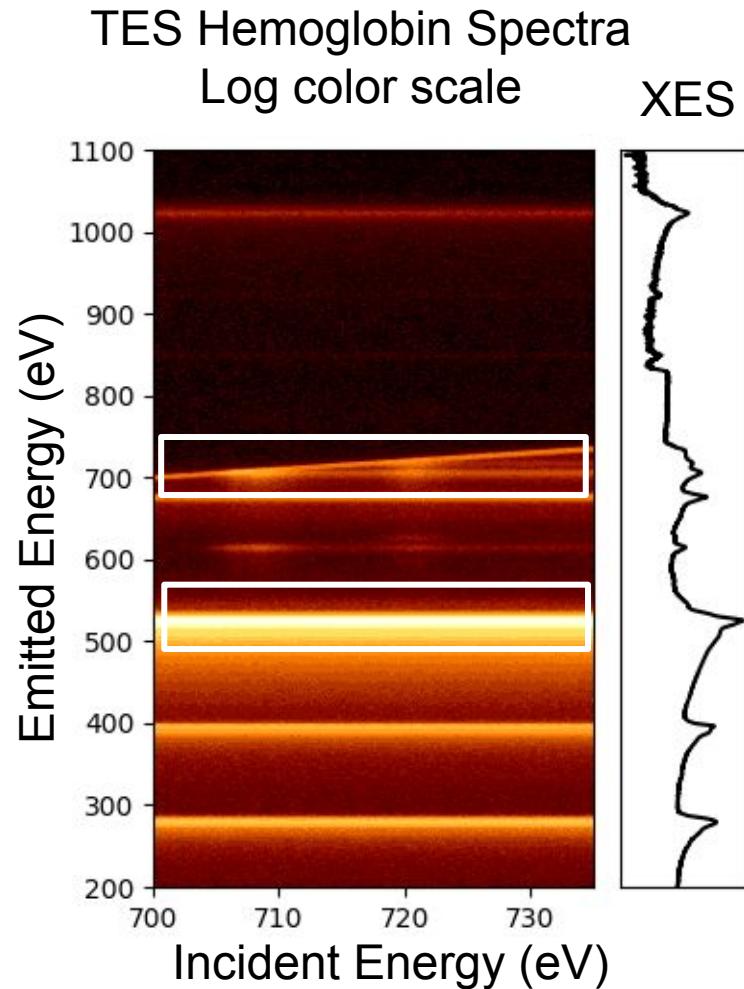
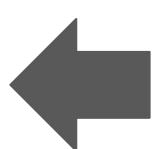
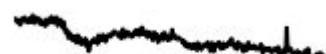
Yes!

TES has the sensitivity to reveal the Fe L-edge of hemoglobin for the first time

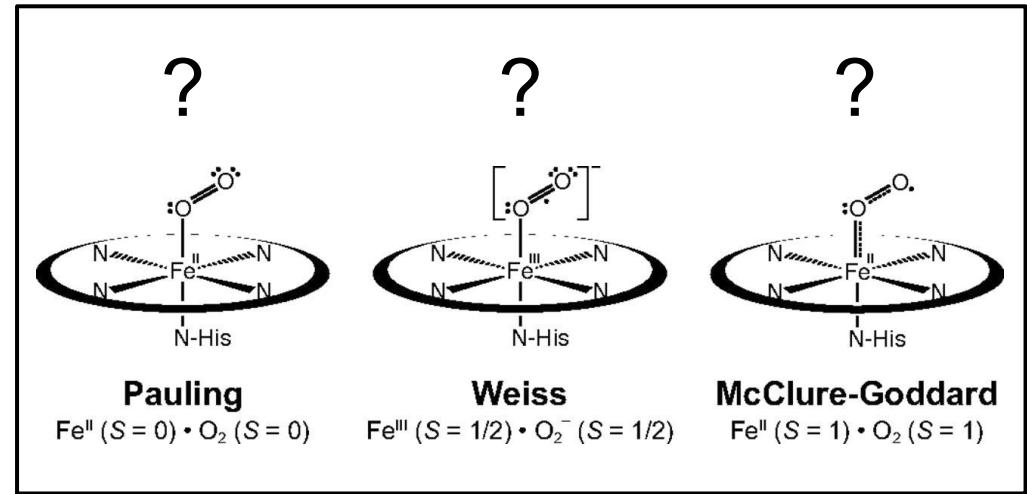
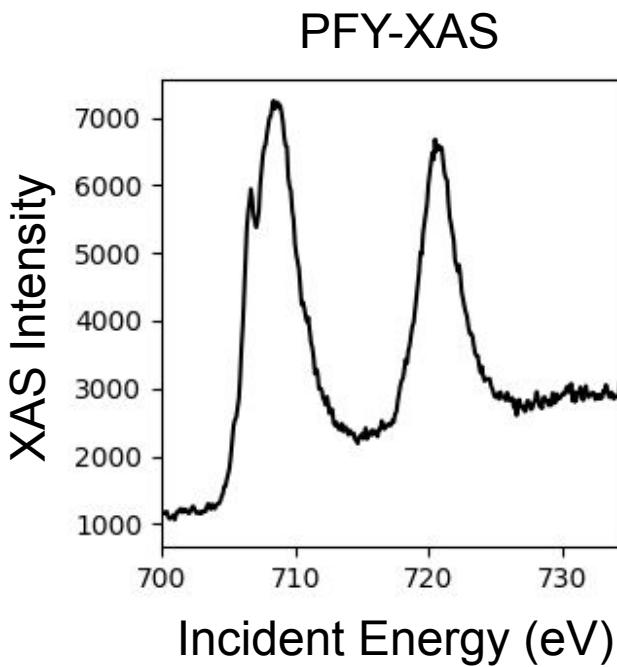
Signal!



Background!



First soft X-ray spectrum for oxyhemoglobin



Who is right? Analysis is ongoing!

Hemoglobin project in collaboration with:

Stanford University

Leland Gee

Augustus Braun

James Yan

Ed Solomon

University of Manchester

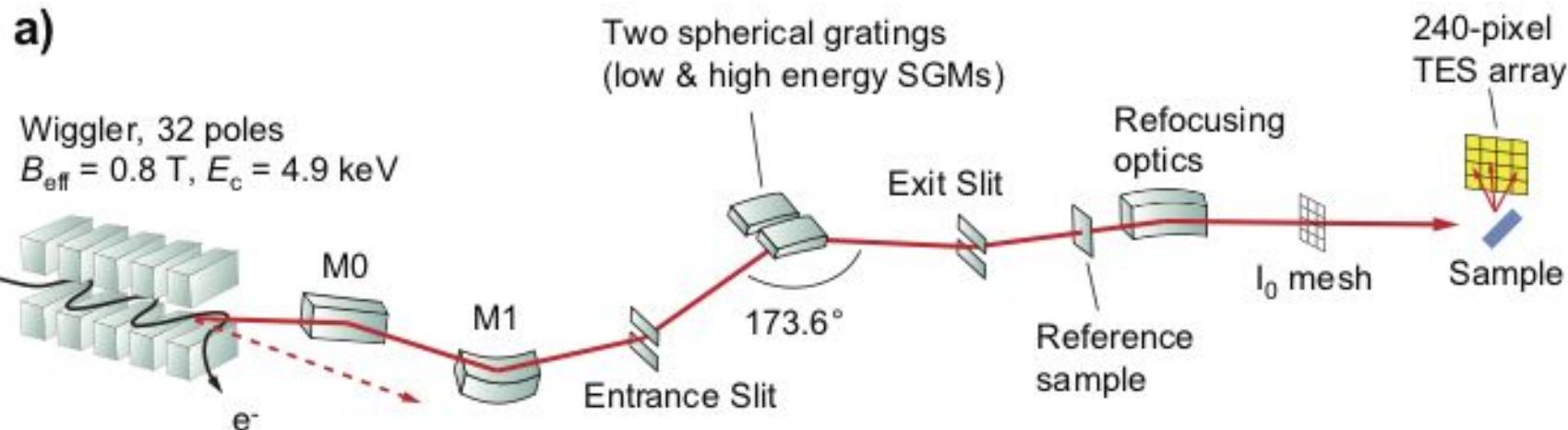
Michael Baker

Conclusions

- Robust beamline instrument with strong user demand
- TES enables battery measurements previously impossible at older light-sources
- A fast, non-destructive probe of composition
- TES enables new science in dilute compounds

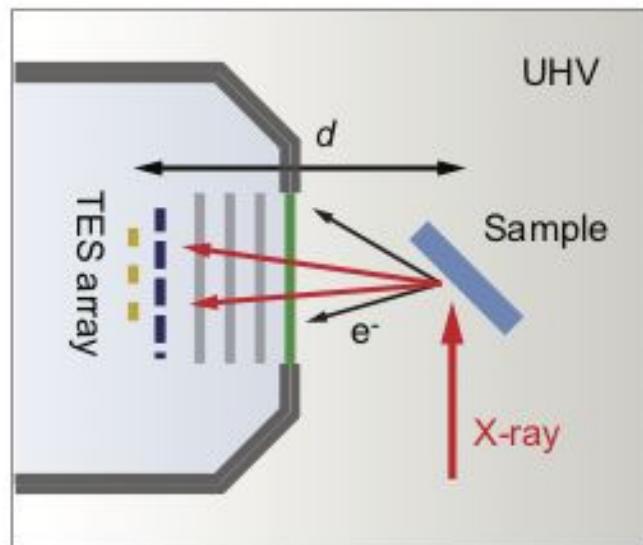
Extra Slides

TES at SSRL BL 10-1



Beamline 10-1 parameters

Photon Flux	$\sim 10^{11} \text{ phot/s}$
Energy Range	150-1400 eV
TES-sample distance	40 mm
Vacuum	10^{-8} torr



Soft X-ray Spectroscopy (150 eV - 1500 eV)

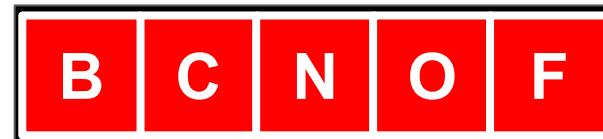
Element-specific probe of:

- Chemical composition
- Electronic structure
- Spin
- Symmetry

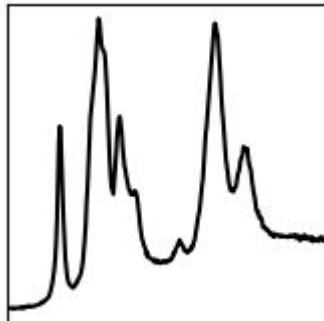
L-edges (2p)



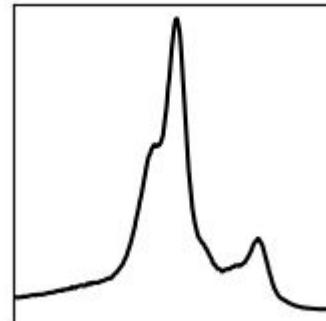
K-edge (1s)



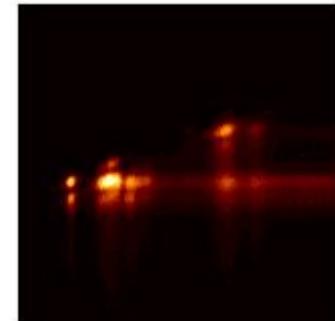
Absorption



Emission

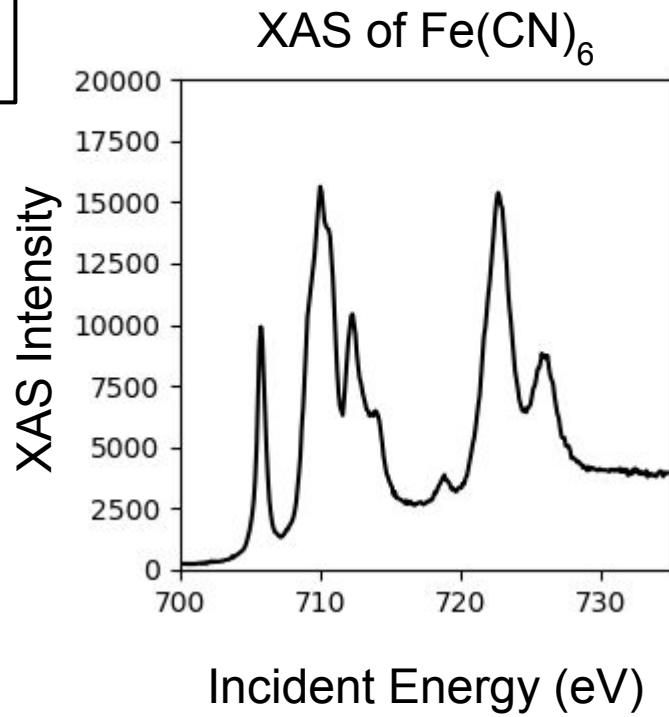
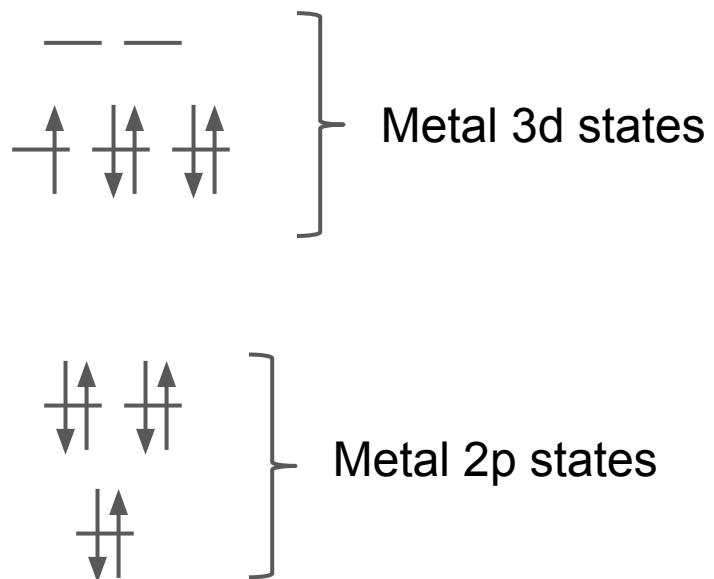


Resonant Scattering



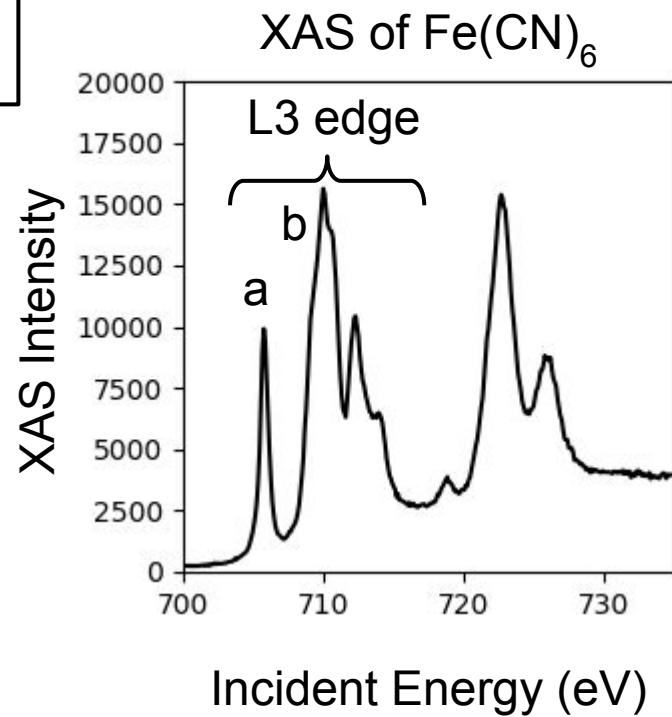
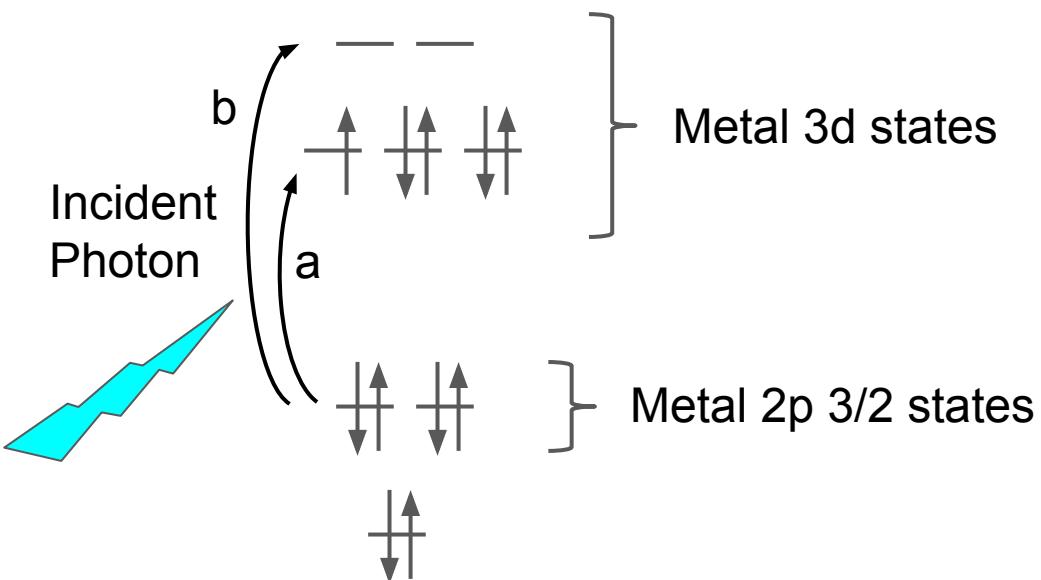
X-ray Absorption Spectroscopy (XAS)

A photon is absorbed and kicks an electron to a higher energy level



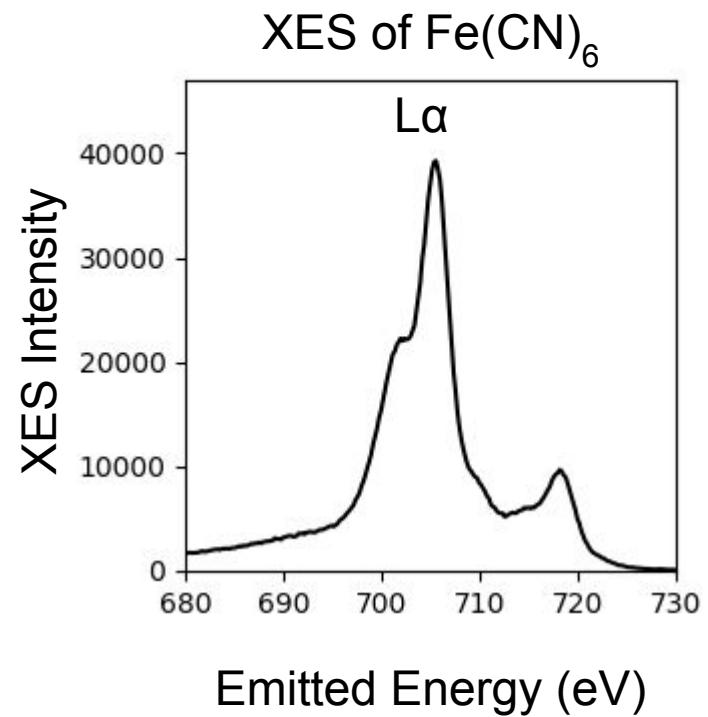
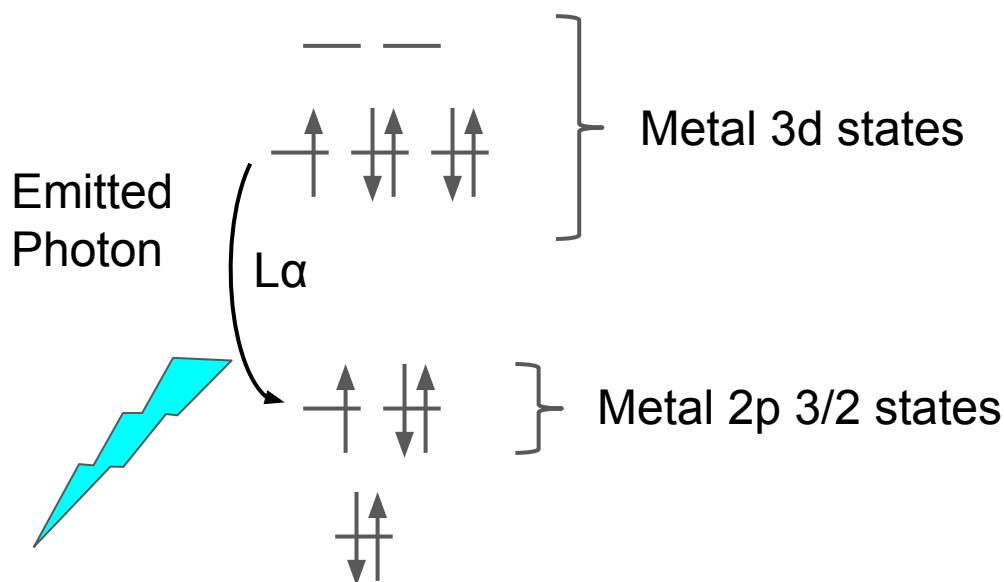
X-ray Absorption Spectroscopy (XAS)

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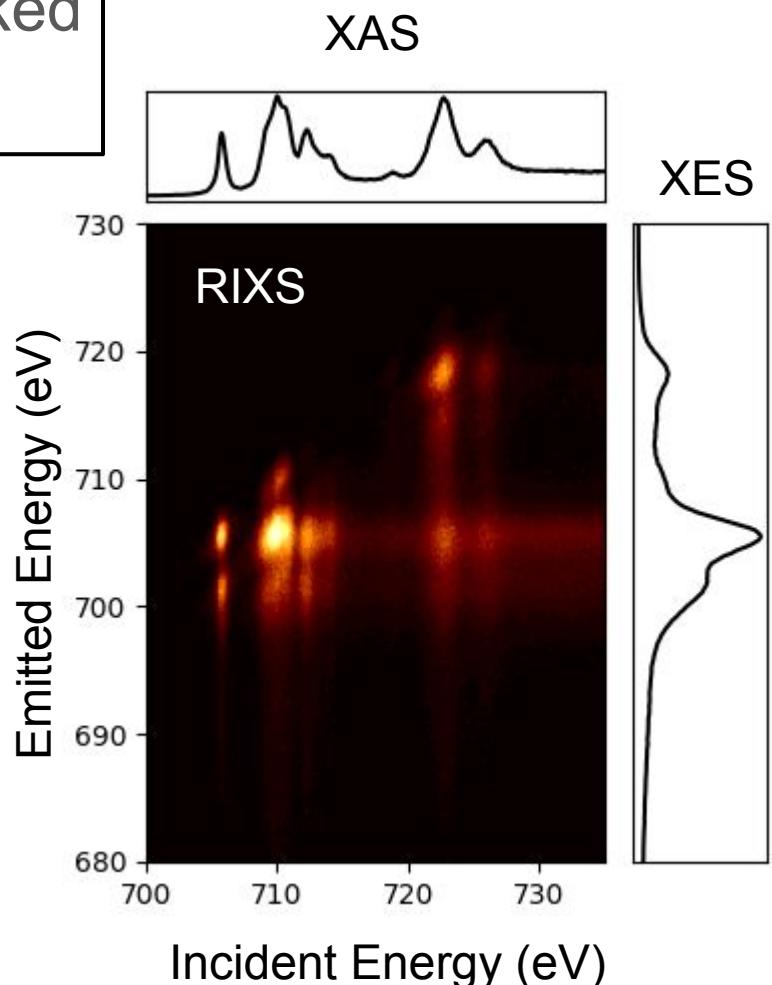
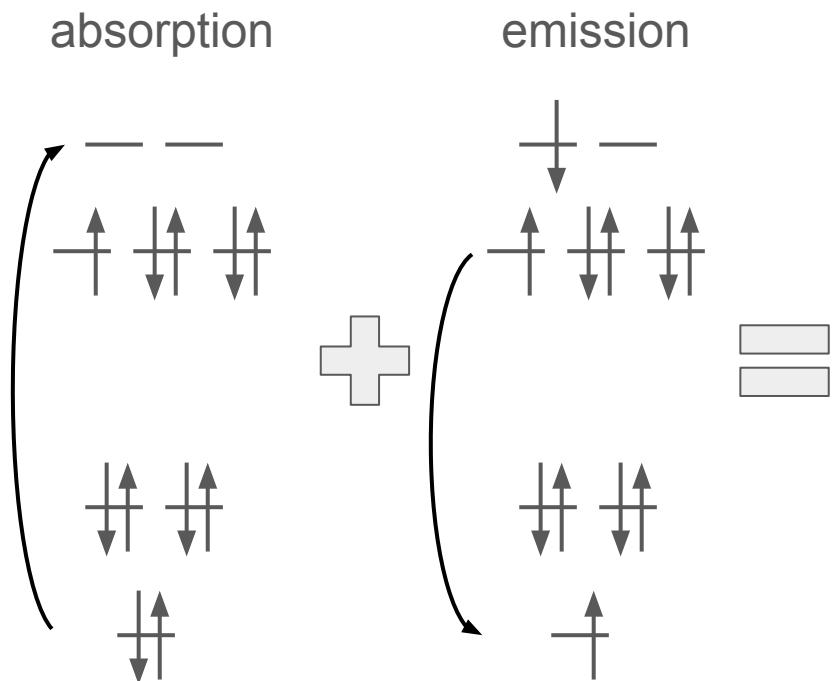
X-ray Emission Spectroscopy (XES)

A photon is emitted when an electron fills a core hole



Resonant Inelastic X-ray Spectroscopy (RIXS)

Absorption and emission are linked together in a resonant process



Routine Spectroscopy

Requirements

- Moderate solid angle
- Moderate energy resolution

Applications

- Everything

