

## Charge exchange measurements with neutral Hydrogen using the X-ray Quantum Calorimeter (XQC)





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## 1. Introduction – Why study Charge Exchange?

- Highly-charged ions colliding with neutral atoms tend to accept an electron from the atom in a process known as **Charge Exchange (CX)**. The electron is accepted into an excited state which then de-excites back to ground state thus emitting an X-ray photon.
- CX is ubiquitous in astronomy. Wherever we have hot gas (with highly charged ions) interfacing with cold gas (with neutral atoms), we can have CX. Examples of such X-ray emission include







Courtesy : Chandra



- Different spectral line ratios are plotted for charge exchange of C<sup>6+</sup>. The heavy and non-trivial dependence
- Good news is charge exchange spectra with different neutral atoms (and thermal emission) are very different. So, line ratios can help identify the process.





• Modelling charge exchange emission is hard because the partial cross-sections and hence the Xray line strengths depends on the relative velocity of the ion and the atom. The relative velocity dictates the (n,l) quantum state after the collision.

Different angular momentum state can give rise to different spectral lines and line-ratios because of selection rules.





## 4. Results and Spectra

To distinguish the signal from background, we measure spectra both when the ion beam and neutral beam are in phase as well as when they are out of phase (to measure background from both sources).





## 5. Conclusions/Outlook

- The experiment to measure charge exchange
- There was a leak during the run which is why we had 10 mins of data which has been fixed, so we expect much better statistics for the next
- We plan on starting with the astrophysically relevant ions like Carbon, Nitrogen and Oxygen