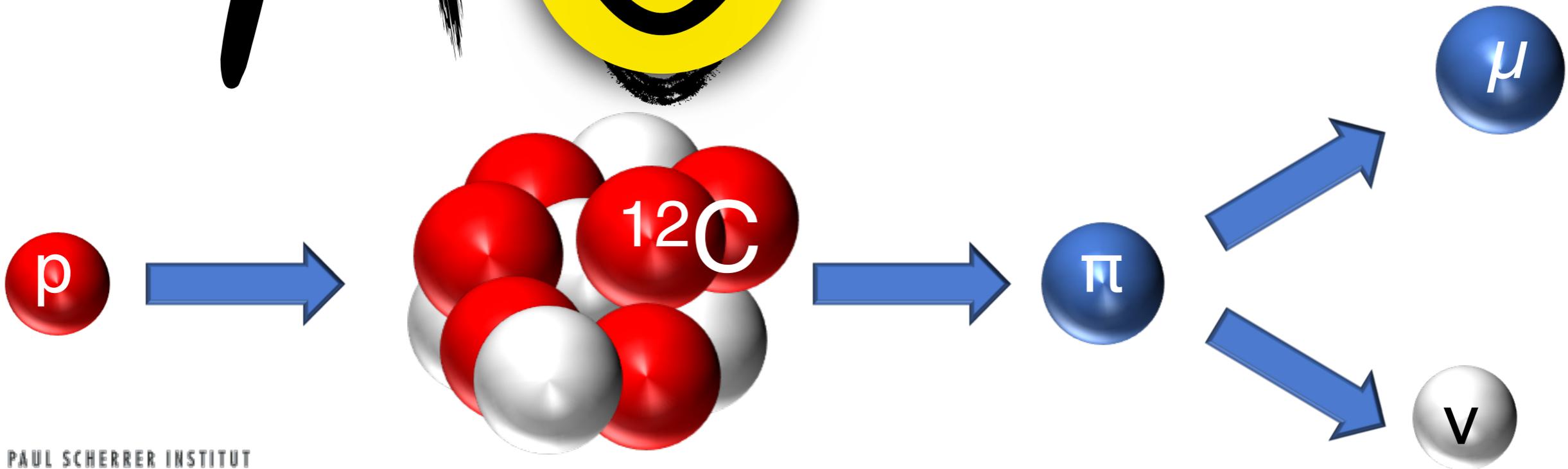


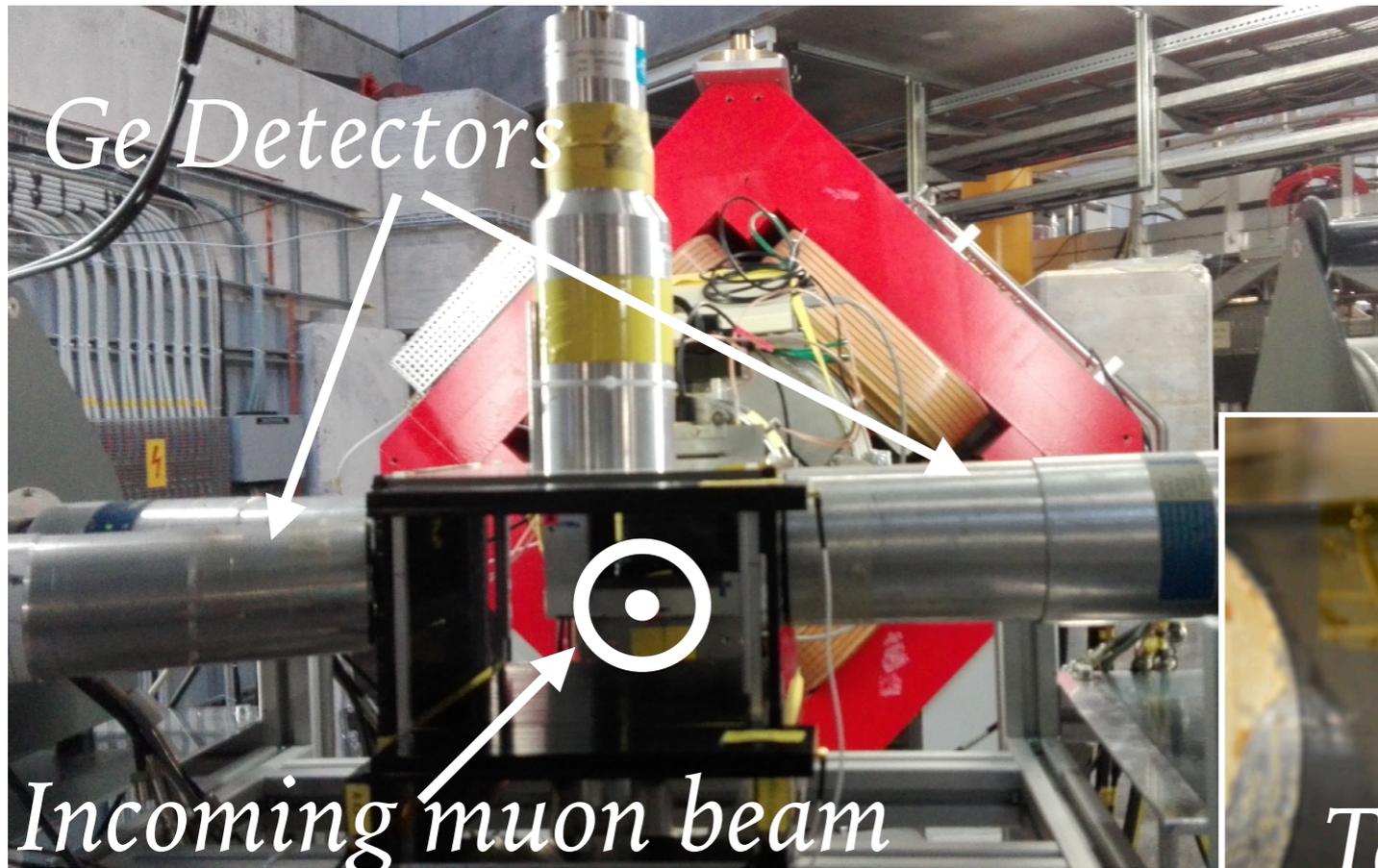
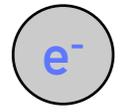
THE NUCLEAR CHARGE RADIUS OF RADIOACTIVE ISOTOPES BY MUONIC X-RAYS MEASUREMENTS

MUON

- “Big brother” of the electron
- 200 times heavier than an electron
- Lifetime of $2.2 \mu\text{s}$
- Same electric charge as electrons
- Pion decay is used for muon creation at PSI



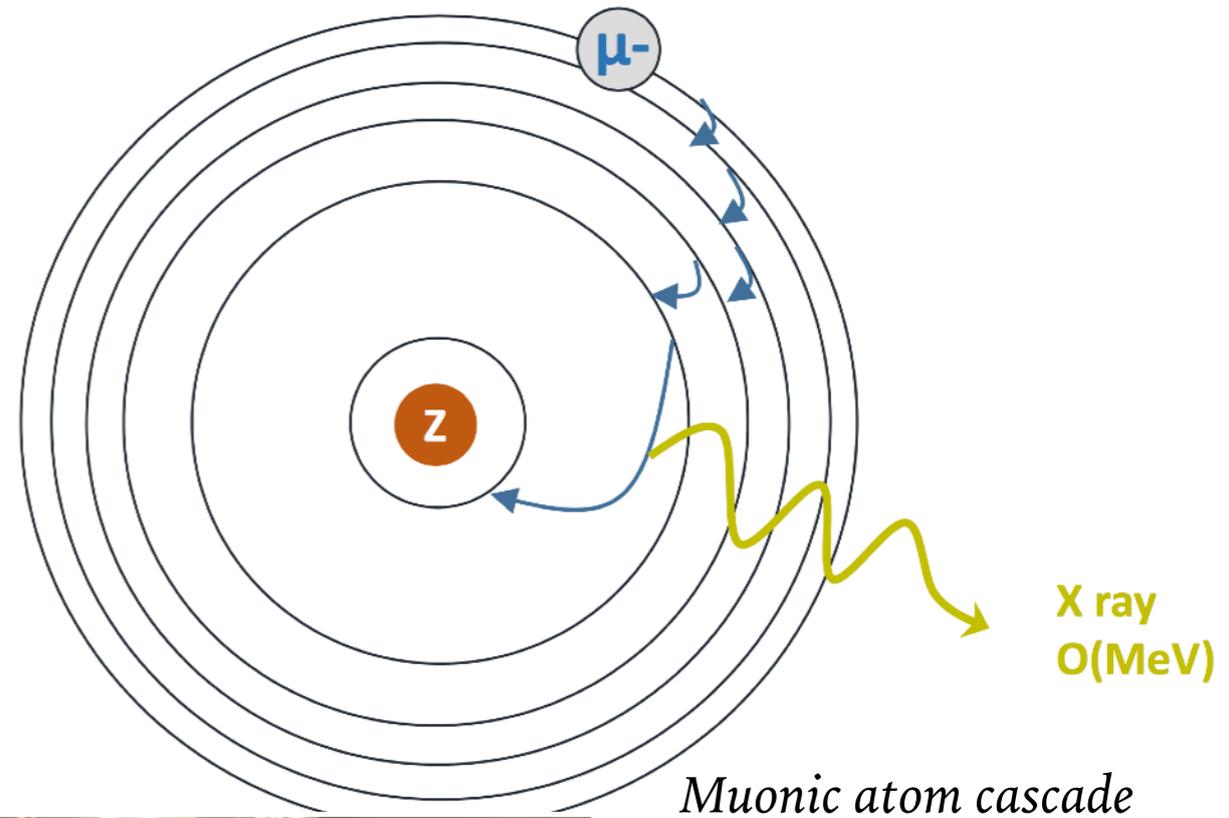
"TRADITIONAL" MUONIC ATOM SPECTROSCOPY



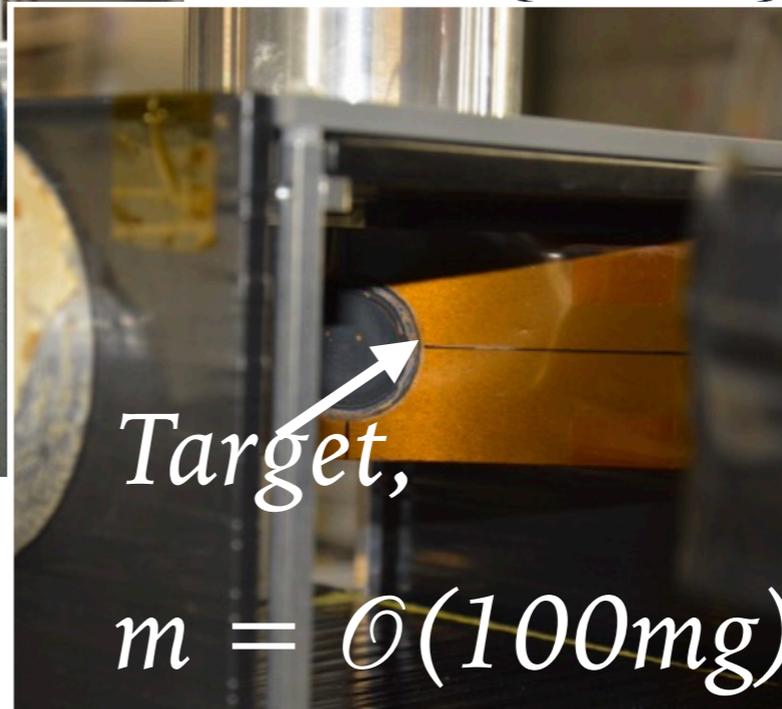
Ge Detectors

Incoming muon beam

Example of standard setup for muonic atom spectroscopy



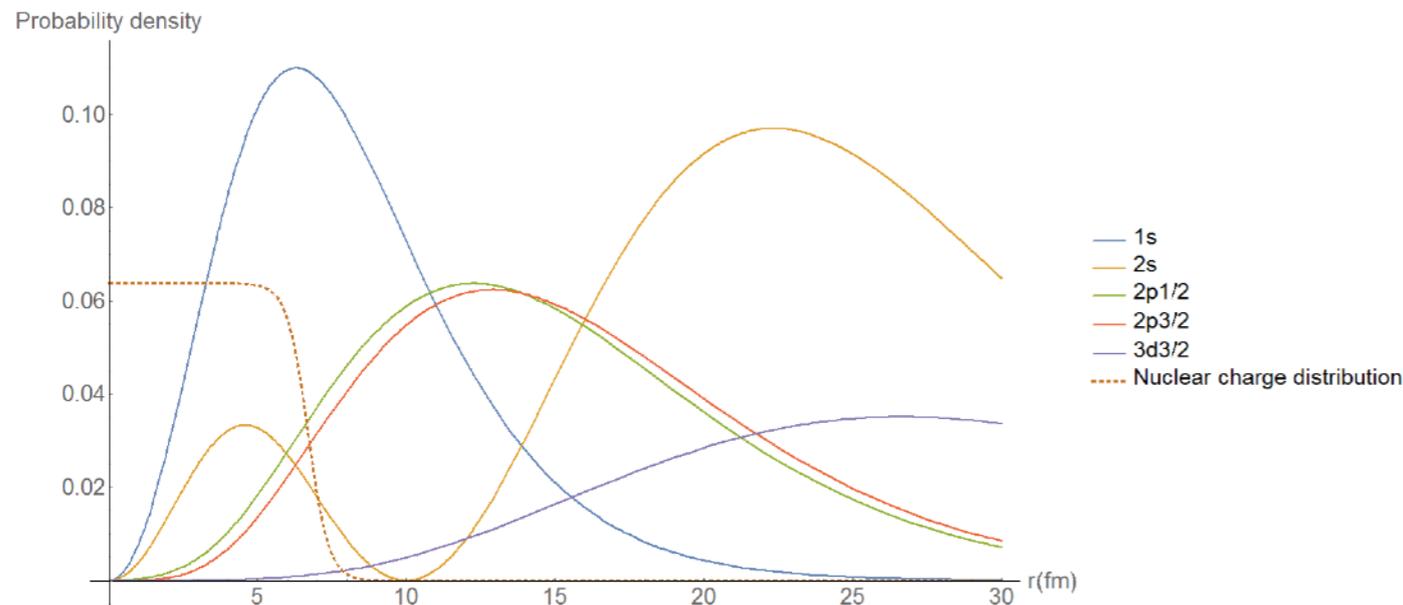
Muonic atom cascade



*Target,
 $m = \mathcal{O}(100\text{mg})$*

Simple target setup for unlimited target amount

NUCLEAR CHARGE RADII



Muon and nuclear charge distribution lead-208

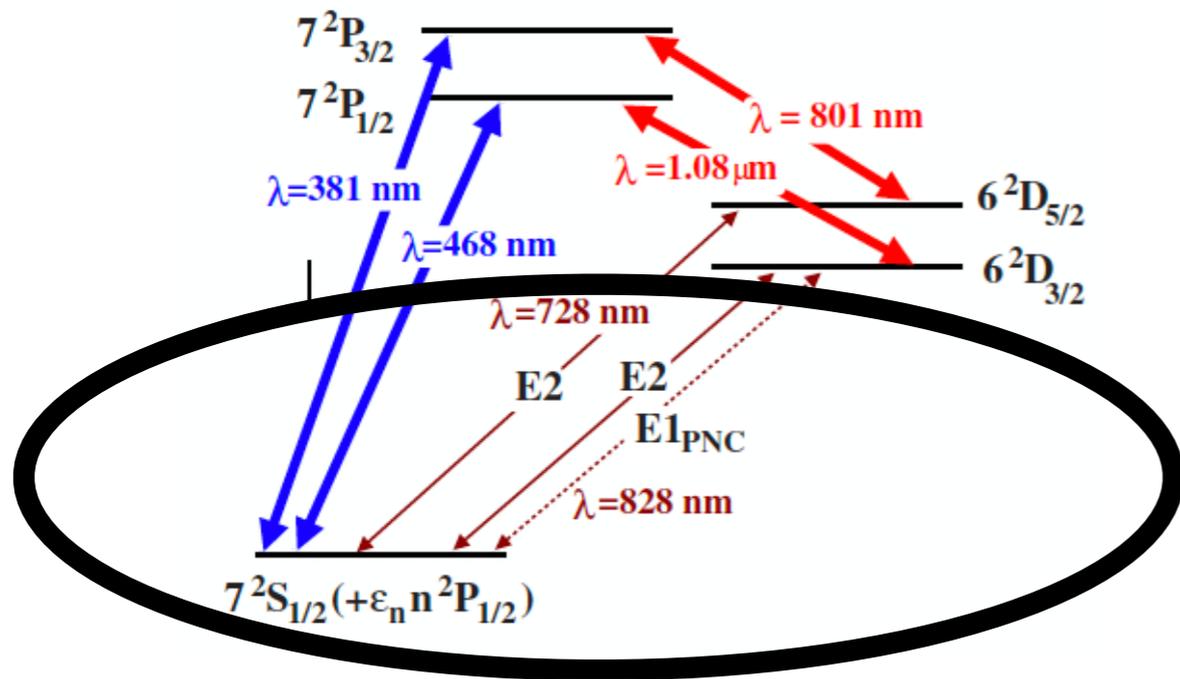
$E_{1s} (Z=82) \sim 19 \text{ MeV (point nucleus)}$

$10.6 \text{ MeV (finite size)}$

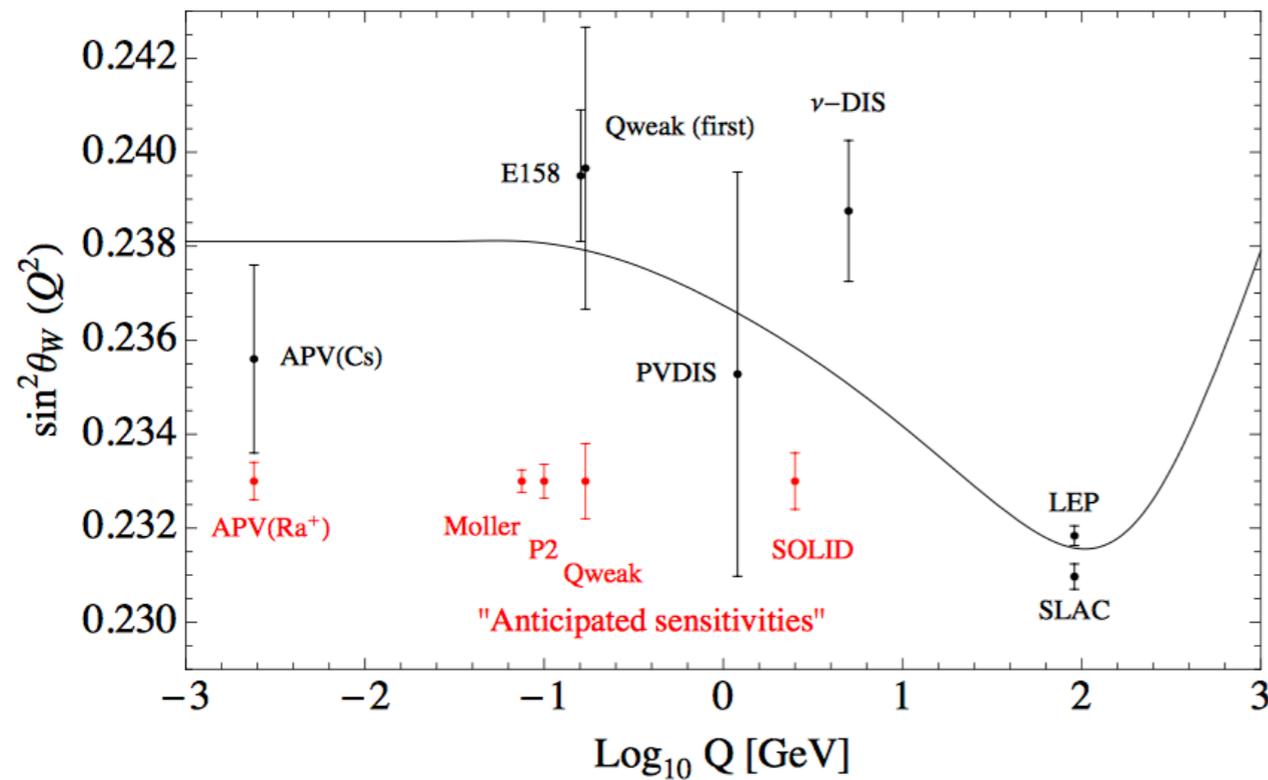
Kessler et al., PRC 11, 1719 (1975)

- Muonic atom energy spectrum is highly sensitive to nuclear charge distribution due to larger overlap
- Charge radius is extracted by QED calculations and model for nuclear charge distribution
- Precise radius extractions are possible
- Example: For ^{208}Pb was achieved a RMS radius of 5.5031(11) fm with 2×10^{-4} relative precision (Bergem et al., Physical Review C 37.6 (1988): 2821)

Ra^+



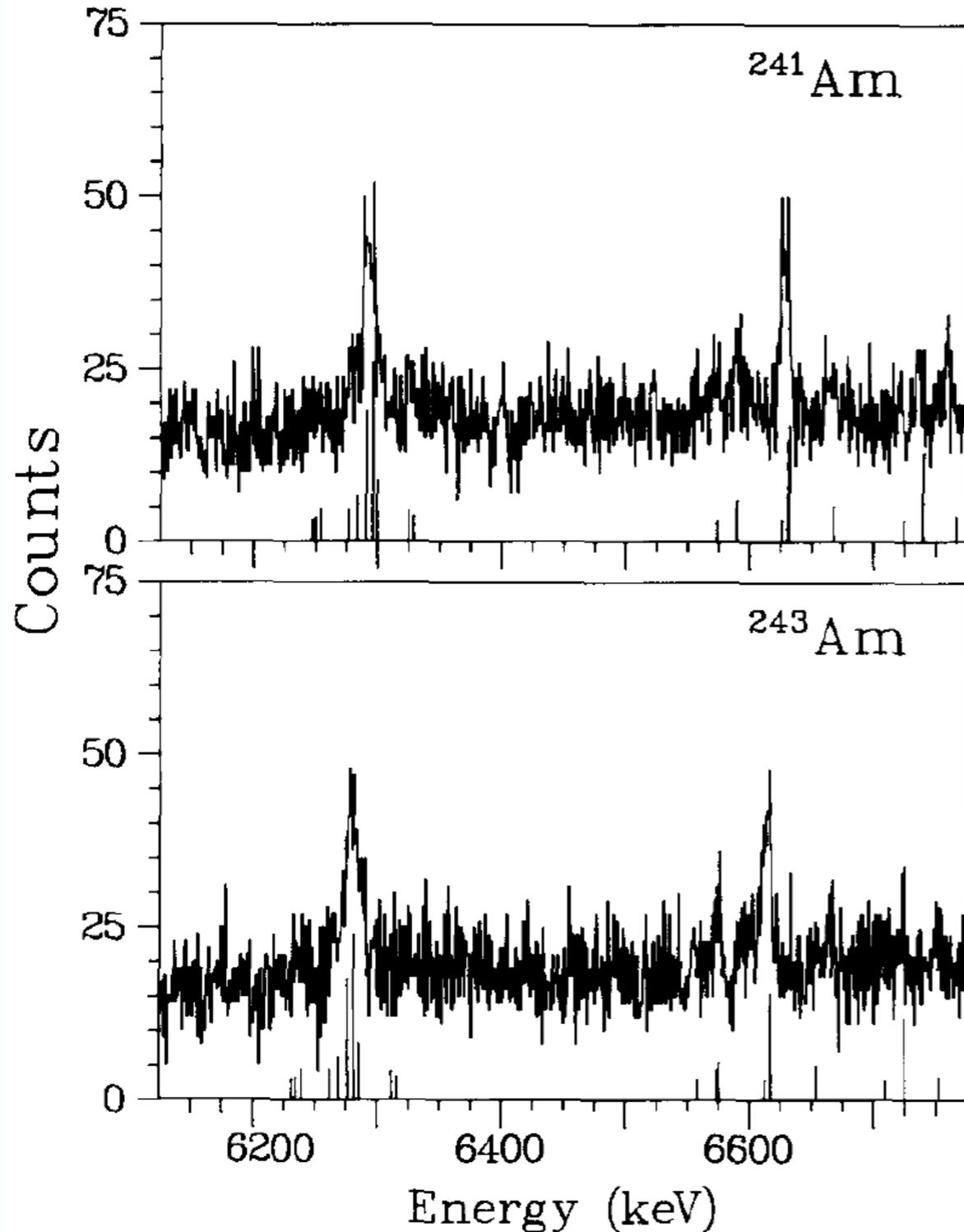
Parity non conserving transition in Ra-226



Wansbeek et al., PRA **78**, 050501 (2008)
 Wood et al., Science **275**, 1759 (1997)
 Lee, arXiv:1511.03783 (2015)

NUCLEAR CHARGE RADIUS OF ^{226}Ra

- A planned atom parity violation experiment requires the radium charge radius with 0.2% accuracy
- E1 transition between $6^2D_{3/2}$ and $7^2S_{1/2}$ is forbidden
- Due to weak interaction a small admixture of E1 in E2 is possible
- Using precise calculations the admixture can be used to extract weak charge

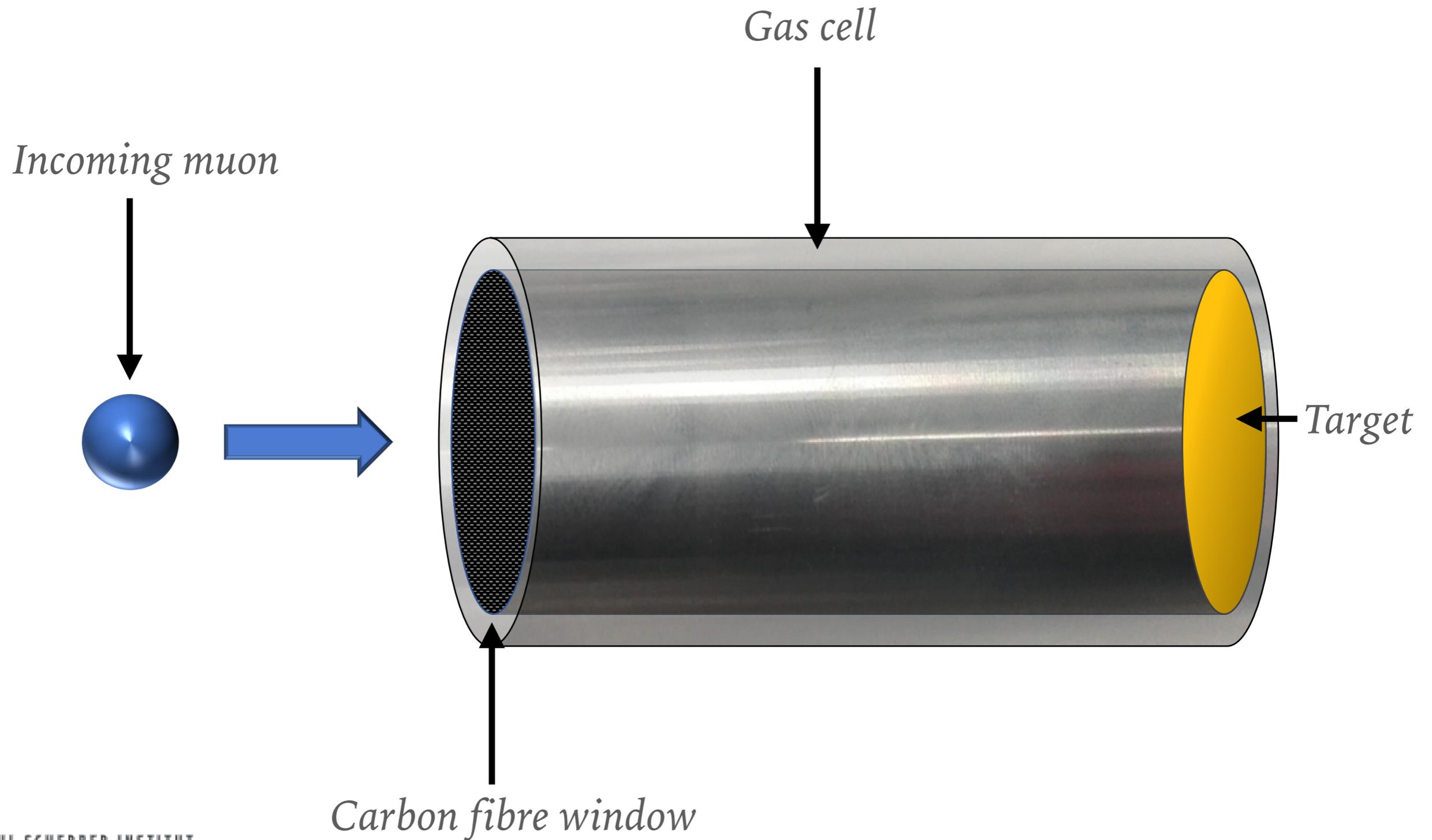


Energy spectrum of Americium-241 and -243

RADIUM TARGET

- So far only a few radioactive isotopes measured with muonic atom spectroscopy
- In the paper they describe the target weight as “modest weight of 1 g”
- The radium-226 target is allowed to have only several μg due to radioactivity safety regulations
- To stop enough muons $O(100 \text{ mg})$ of radium are required
- New target set up required to increase event rate

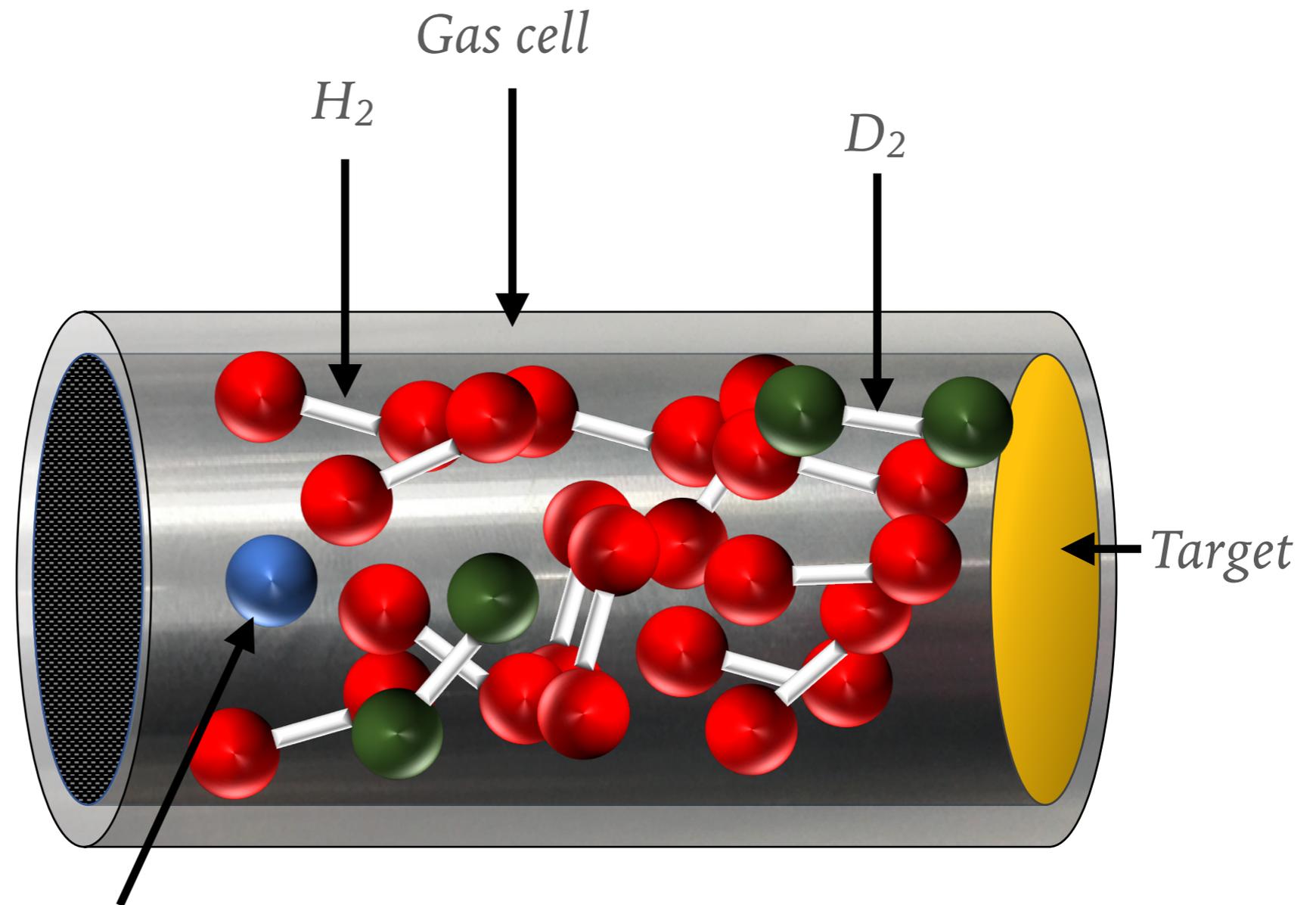
SKETCH GAS CELL TARGET



SKETCH GAS CELL TARGET

.....

- Gas cell is filled with 100 bar hydrogen and 0.25% deuterium admixture

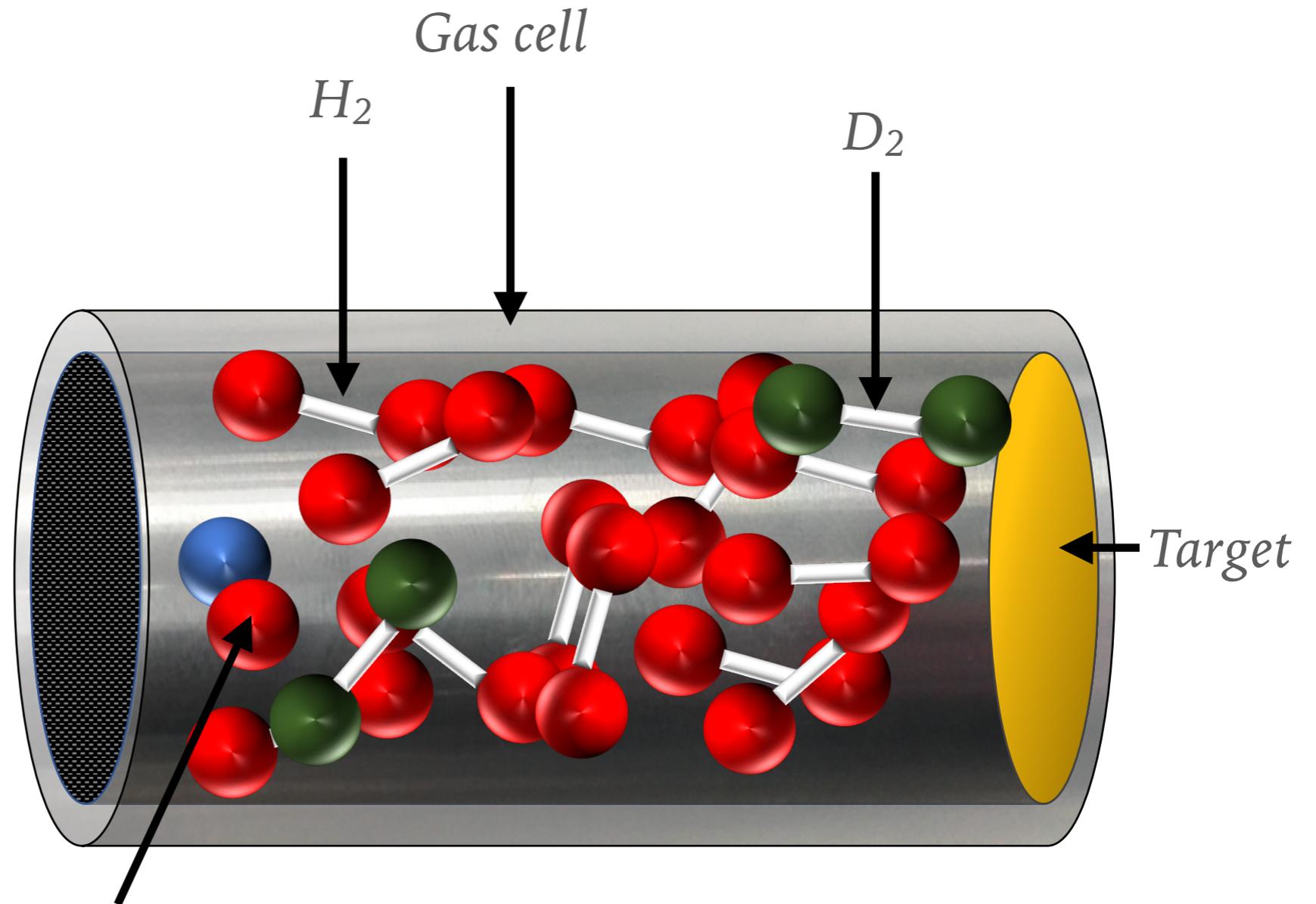


Muon collides with H_2 molecules in gas cell

SKETCH GAS CELL TARGET

.....

- Gas cell is filled with 100 bar hydrogen and 0.25% deuterium admixture

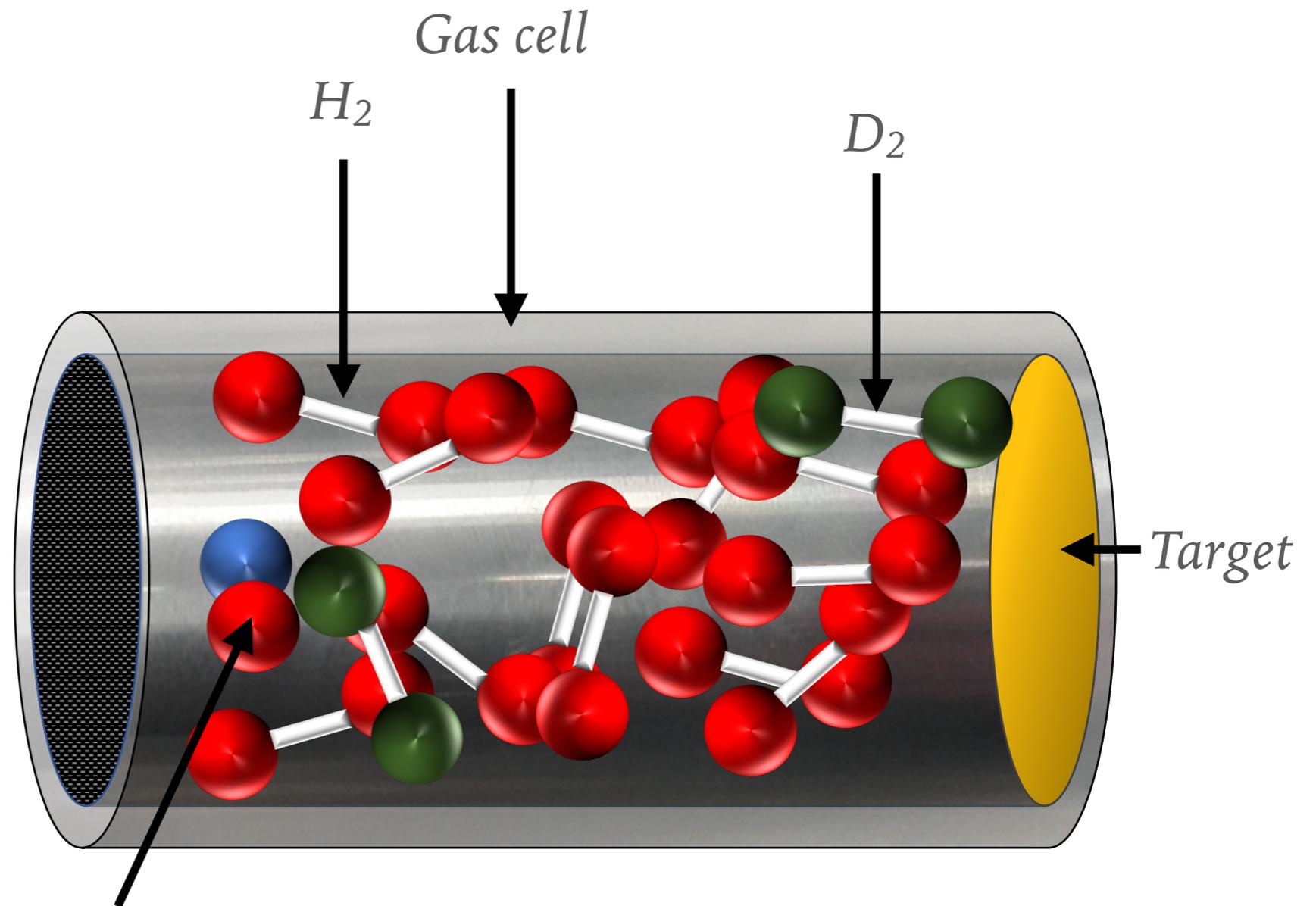


Muon in gas cell is captured by a proton of a H_2 molecule.

Muonic hydrogen μp is produced.

SKETCH GAS CELL TARGET

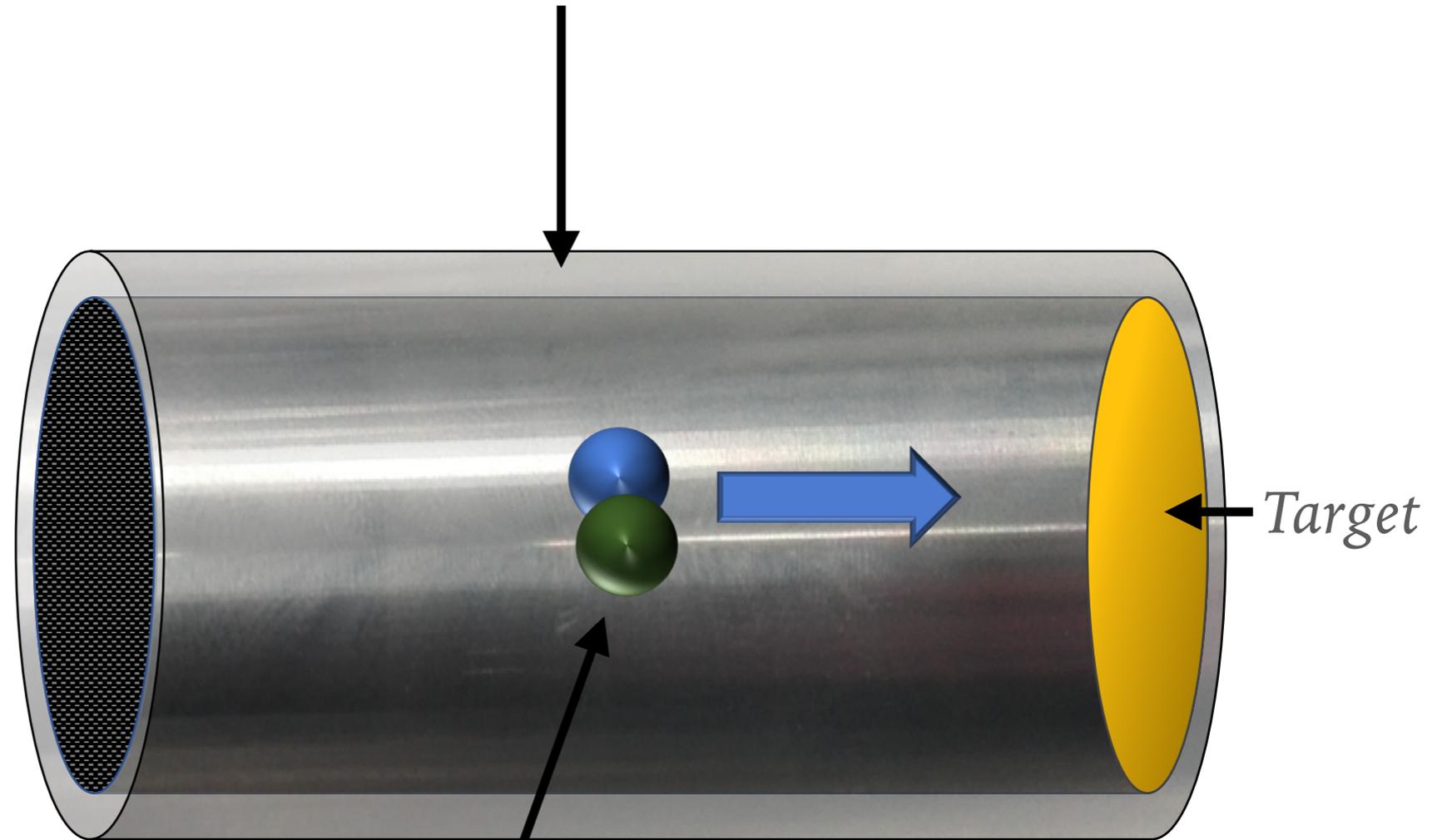
- Gas cell is filled with 100 bar hydrogen and 0.25% deuterium admixture



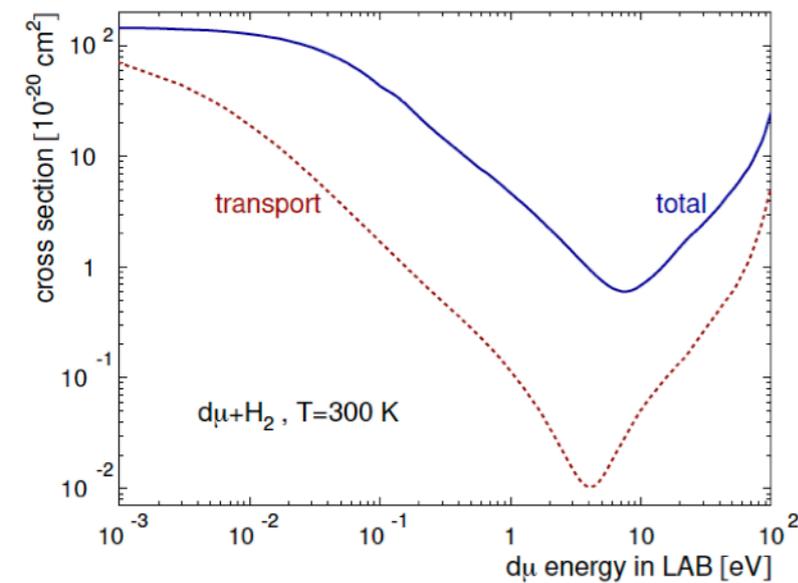
Muonic hydrogen collides with D₂. The muon is transferred to a deuteron. The transfer results in μ d with a kinetic energy boost of several eV.

SKETCH GAS CELL TARGET

Gas cell

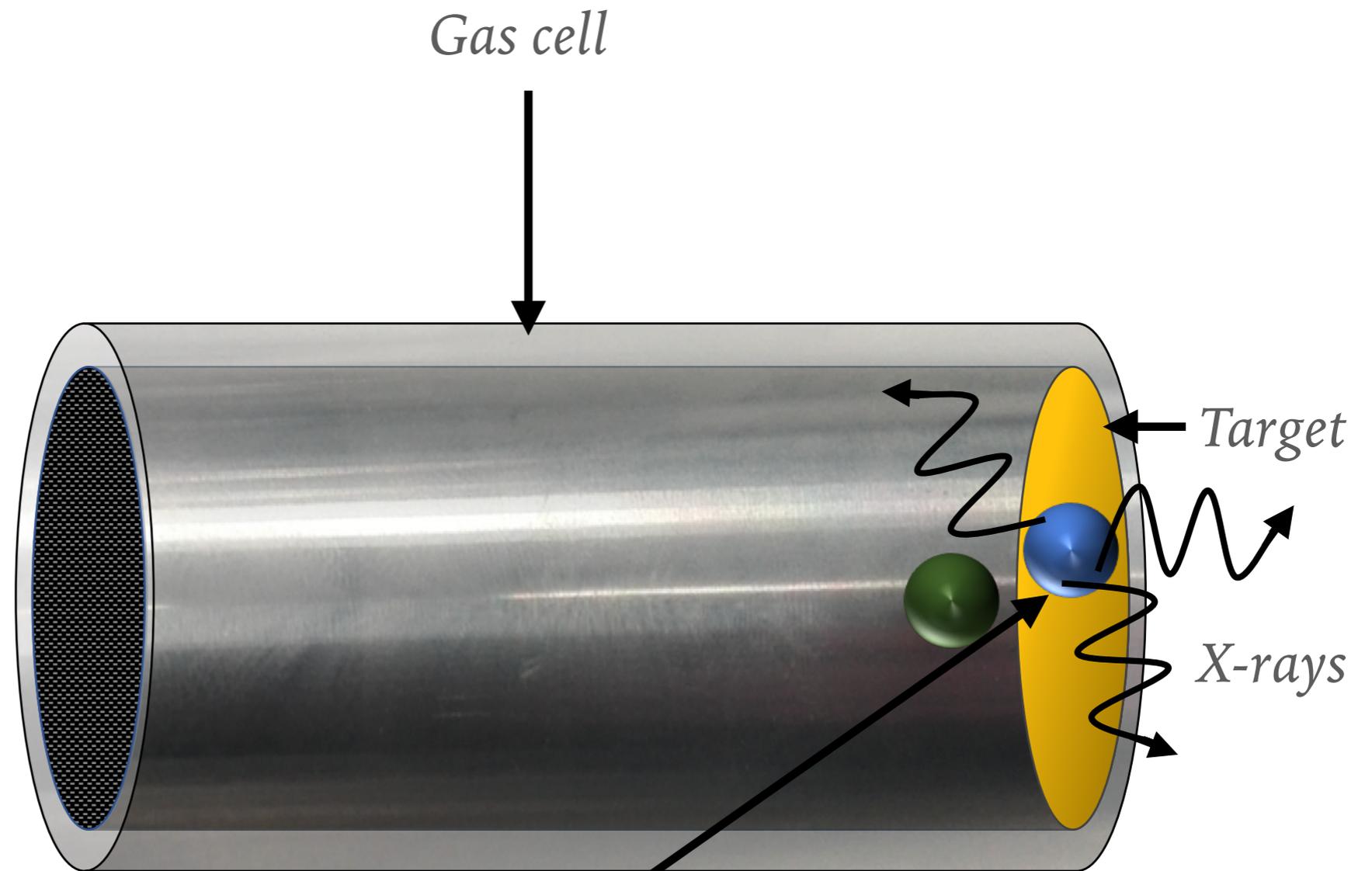


Target



The gas is almost transparent to the μd atom due to the Ramsauer-Townsend effect. μd can move to target.

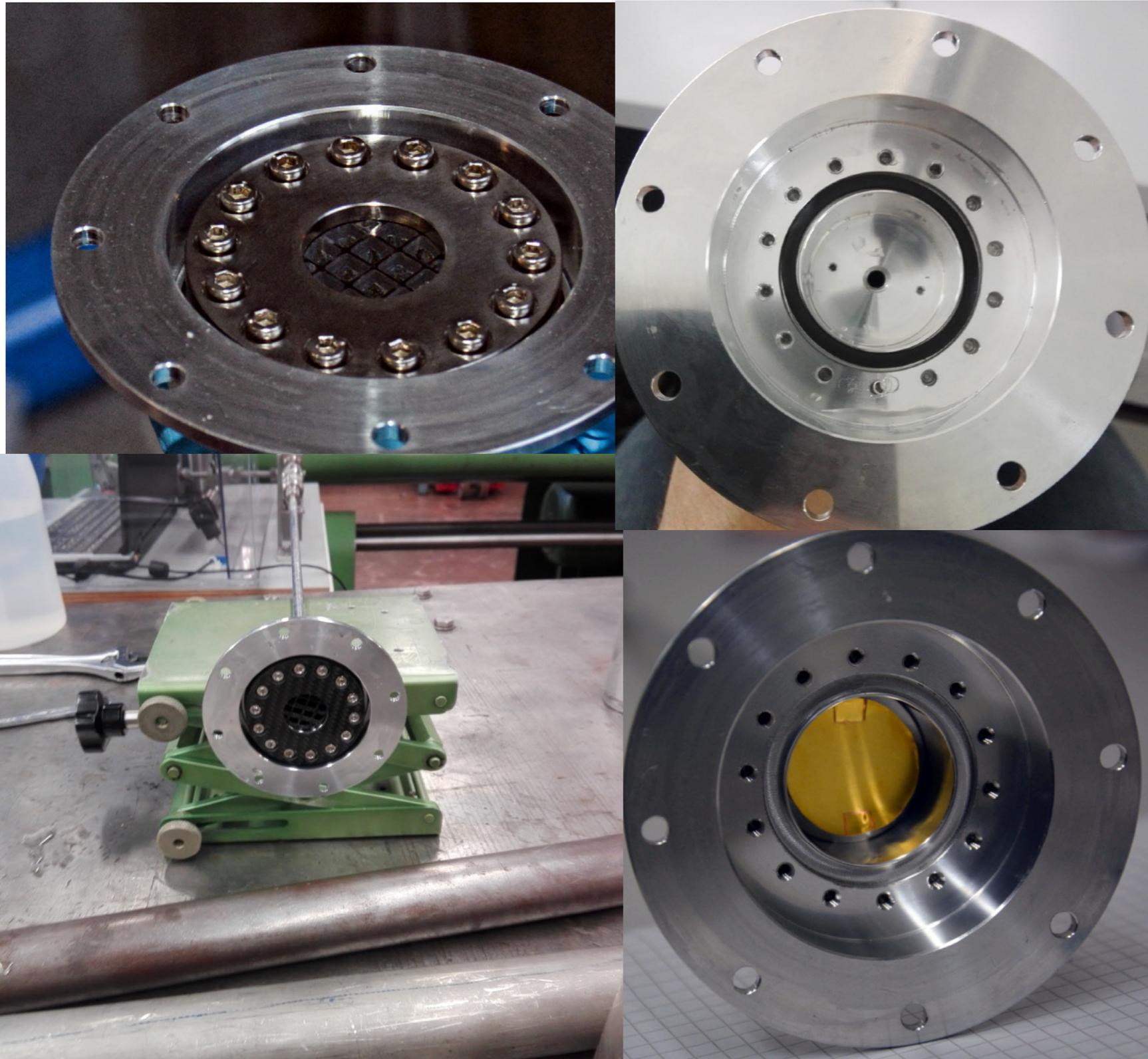
SKETCH GAS CELL TARGET



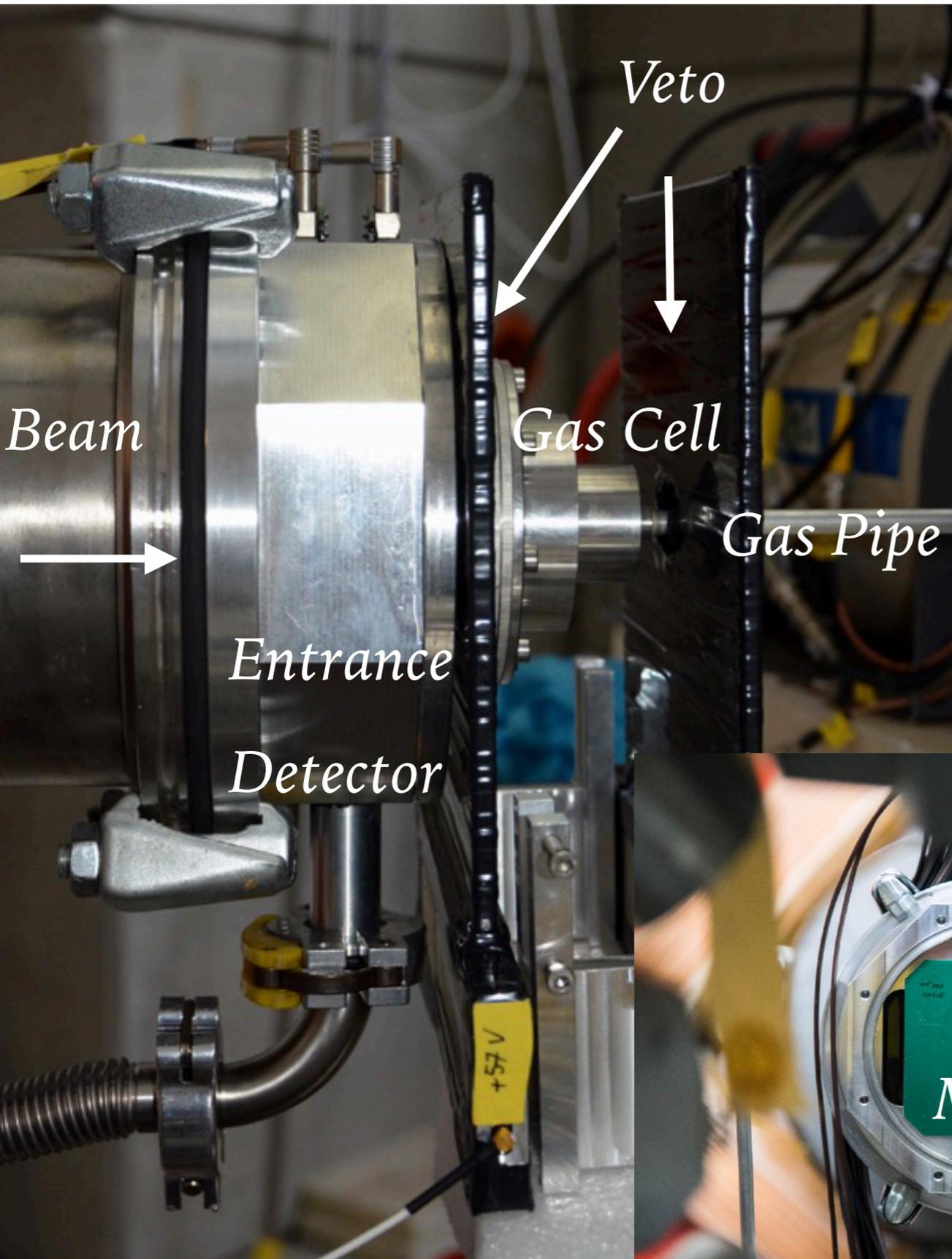
The μd atom transfers the muon to the target. Muonic X-rays are emitted.

PRESSURE TEST GAS CELL

.....



- ▶ Gas cell has to resist a pressure of 100 bar
- ▶ Tested different setups
- ▶ Final set up includes a carbon fibre window with two support grid layers containing carbon fibre and titanium
- ▶ Window withstands a pressure of more than 350 bar, some screw threads could not take the increasing pressure



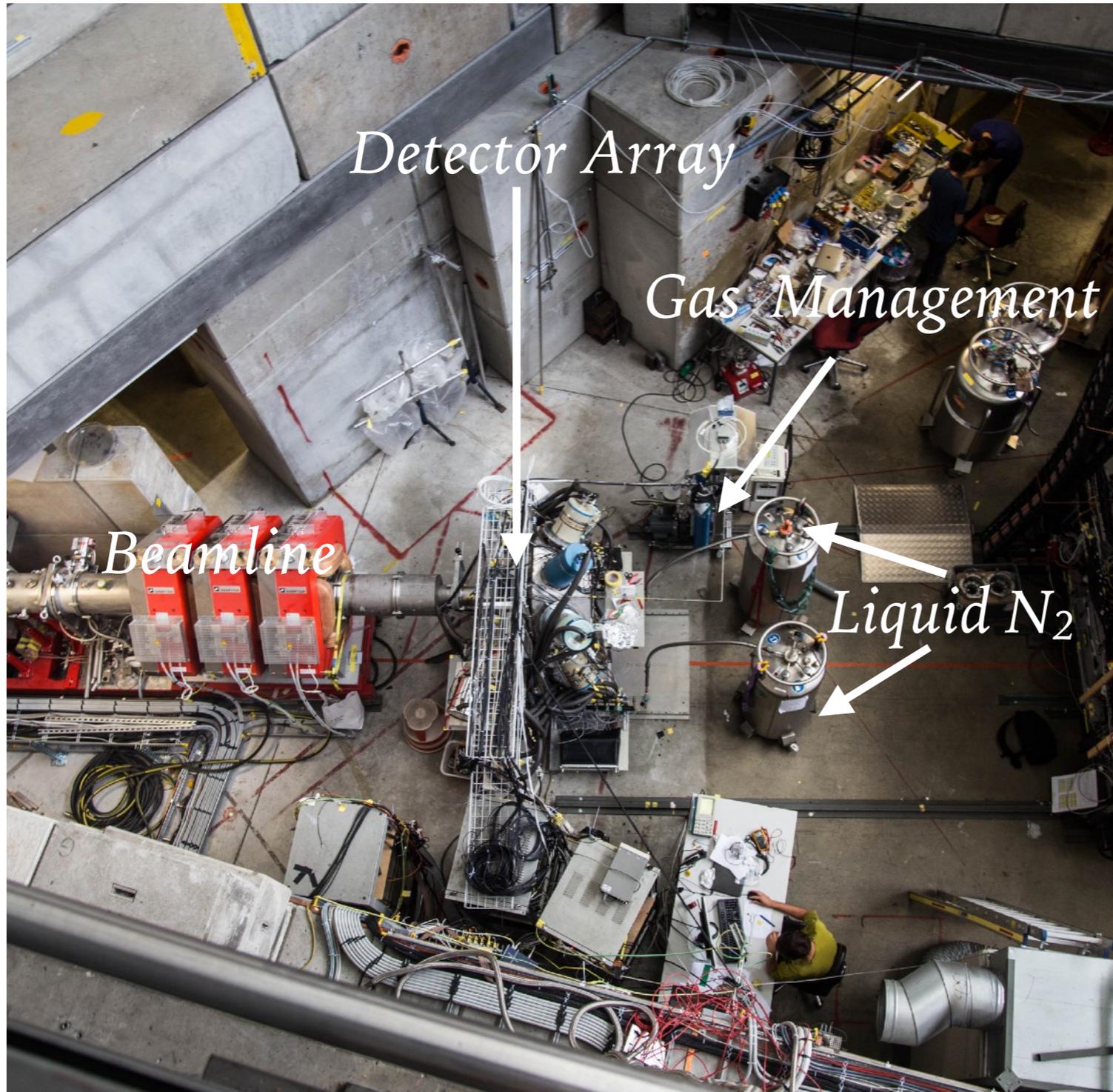
MUON COUNTER

.....

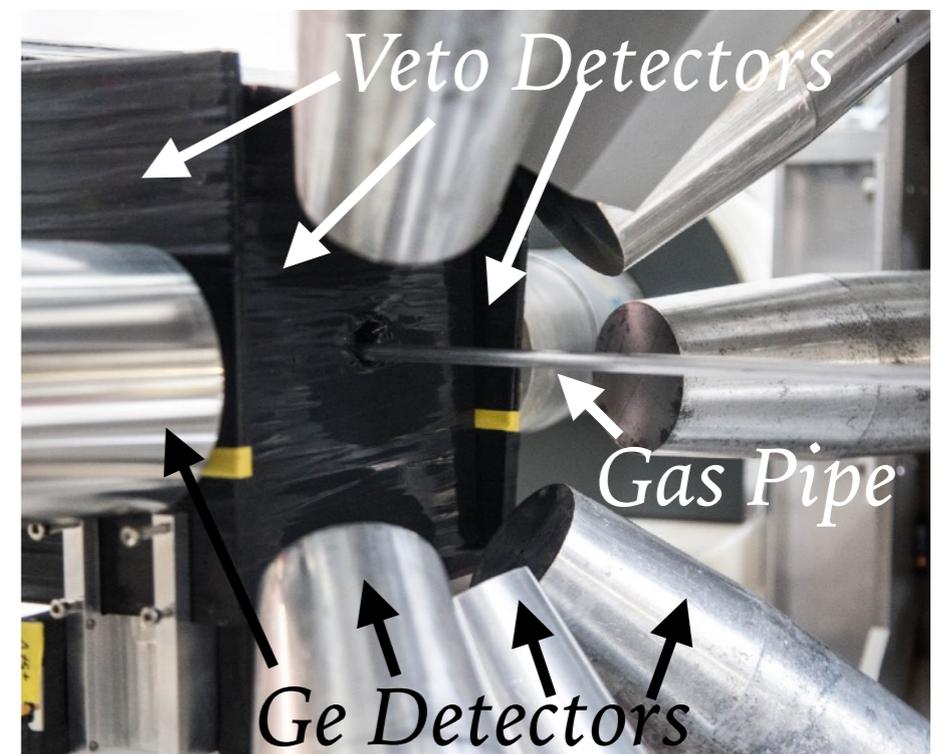
- To estimate the efficiency of the muon transfer to the target and to detect coincidences a thin scintillator is used to detect incoming muons.
- Other scintillators are used as veto detectors for anti-coincidence with decay electron.

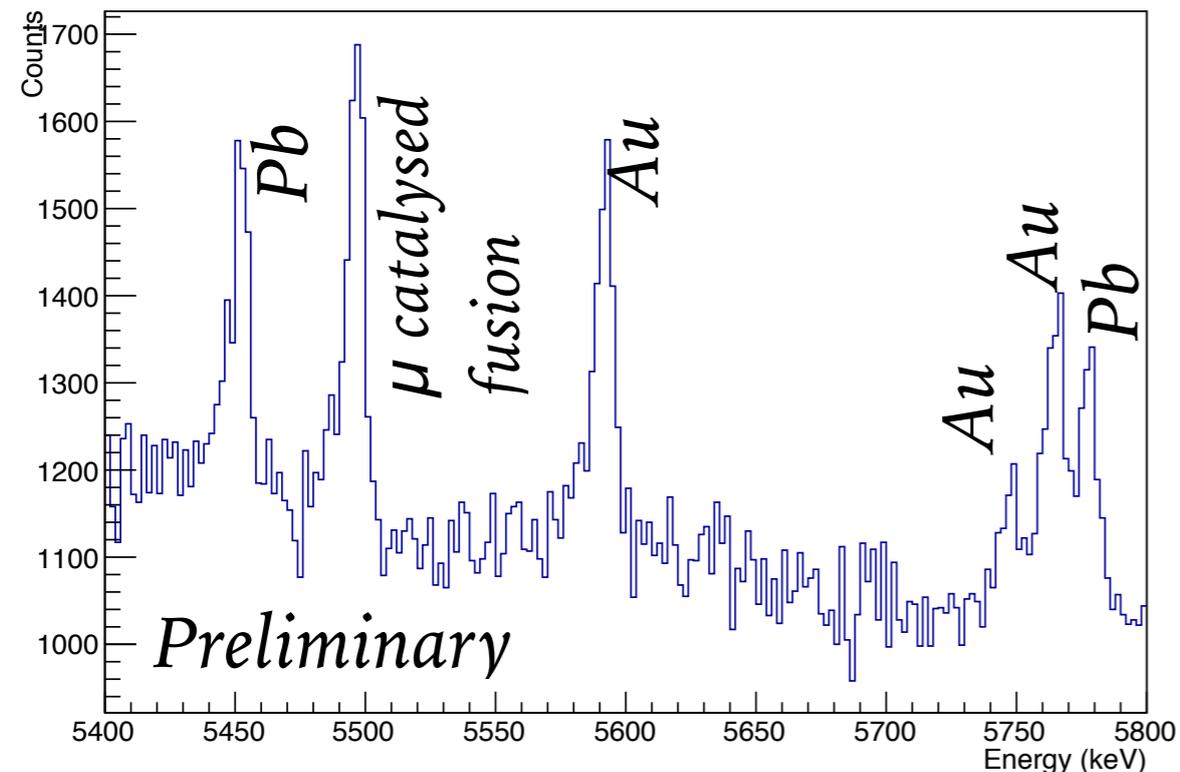
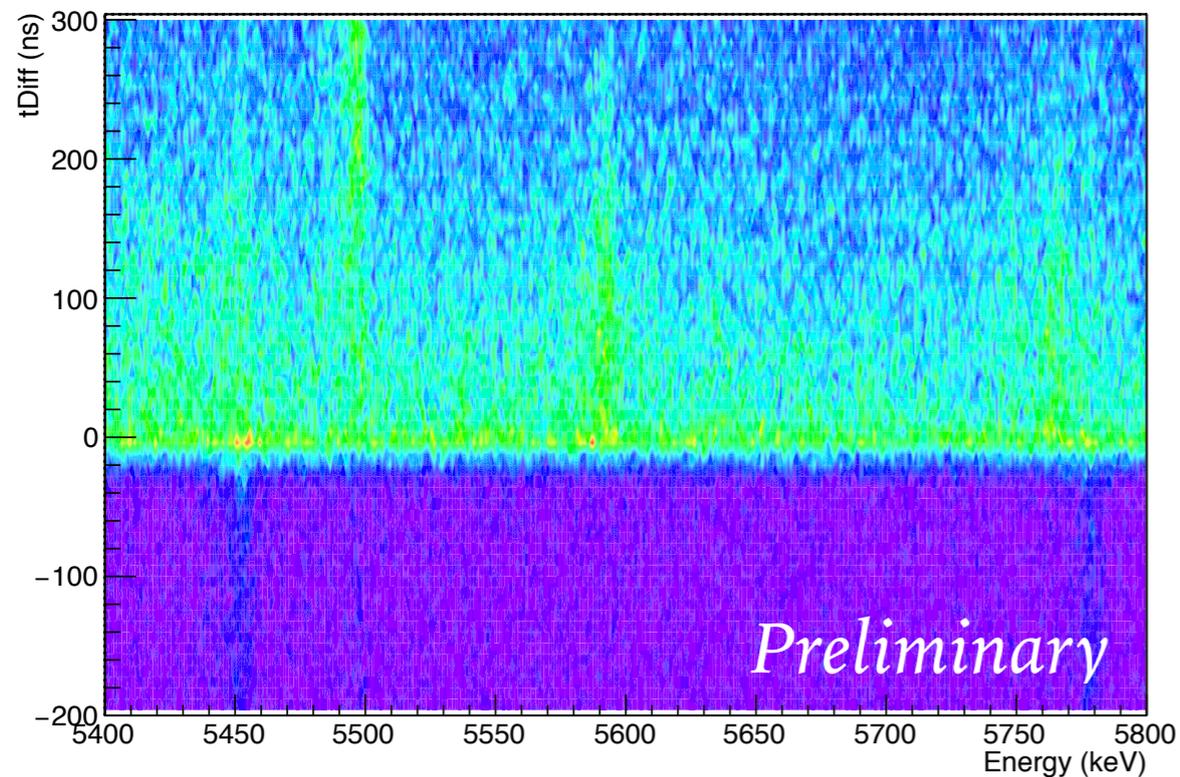


SET UP GOLD TARGET TEST



- During 2017 an array of 11 Ge detectors were used
- It was the first time that an array was used for muonic atom spectroscopy





Energy spectrum of 5 µg gold target

GOLD SPECTRUM

- Used gold to test muon transfer
- Observed gold spectrum with 5 µg target
- Used lead for energy calibration
- Observed also muon catalysed fusion

MUON TRANSFER EFFICIENCY

.....

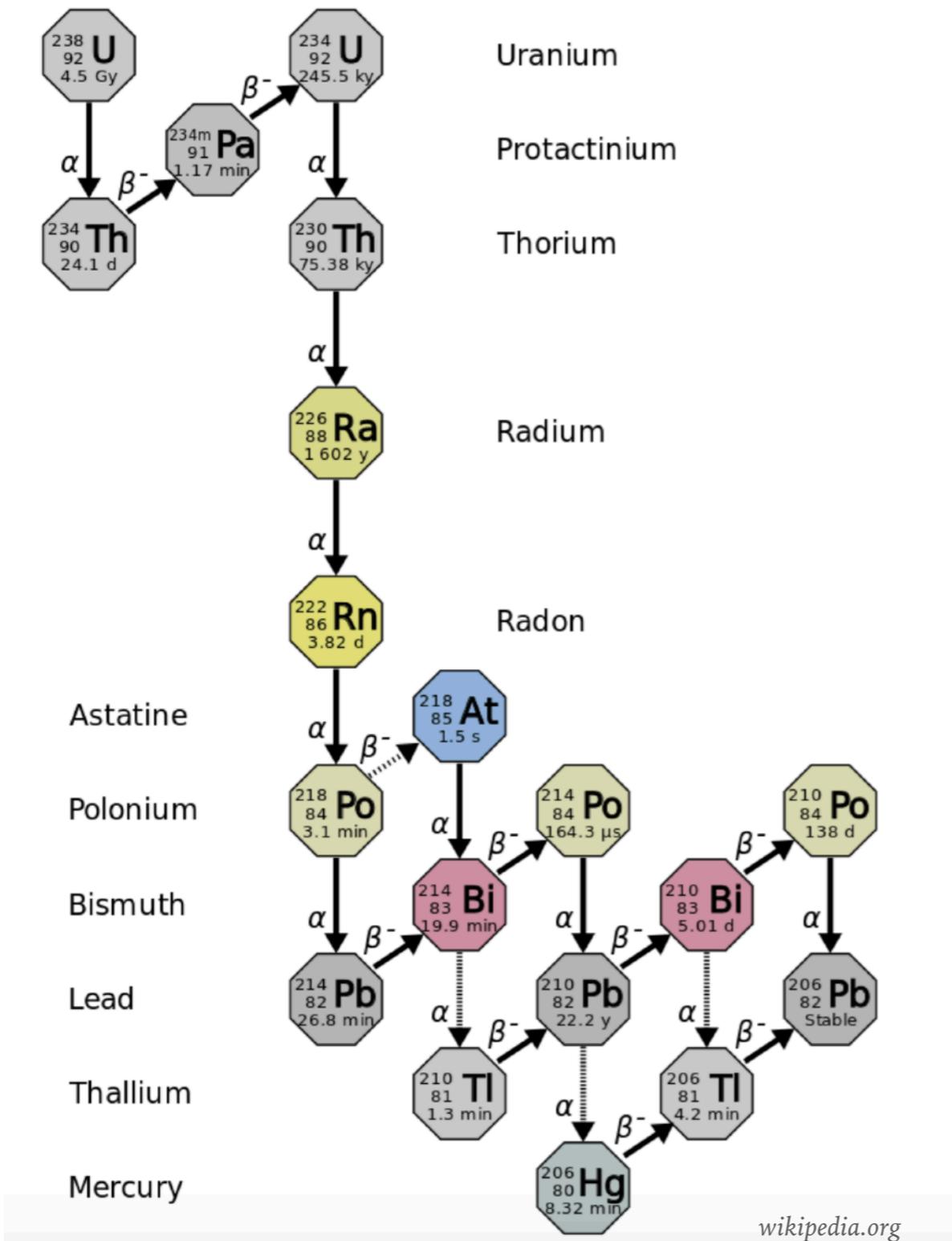
Target	Size	Backing	N_γ / N_μ	ϵ
50 nm Au	4.9 cm ²	Cu	$(10.9 \pm 0.3) \times 10^{-5}$	10.0%
10 nm Au	4.9 cm ²	Cu	$(6.9 \pm 0.2) \times 10^{-5}$	6.3%
3 nm Au	4.9 cm ²	Cu	$(3.6 \pm 0.1) \times 10^{-5}$	3.3%
3 nm Au	4.9 cm ²	kapton	$(3.2 \pm 0.1) \times 10^{-5}$	2.9%
3 nm Au	1 cm ²	Cu	$(1.3 \pm 0.1) \times 10^{-5}$	1.2%

Detected gammas per muon fraction in gold targets (preliminary)

- Detected 2p-1s gammas per incoming muon for various target sizes and amounts
- Type of backing layer seems to have negligible influence
- Even a sufficient number of photons is achieved in the 5 μg target

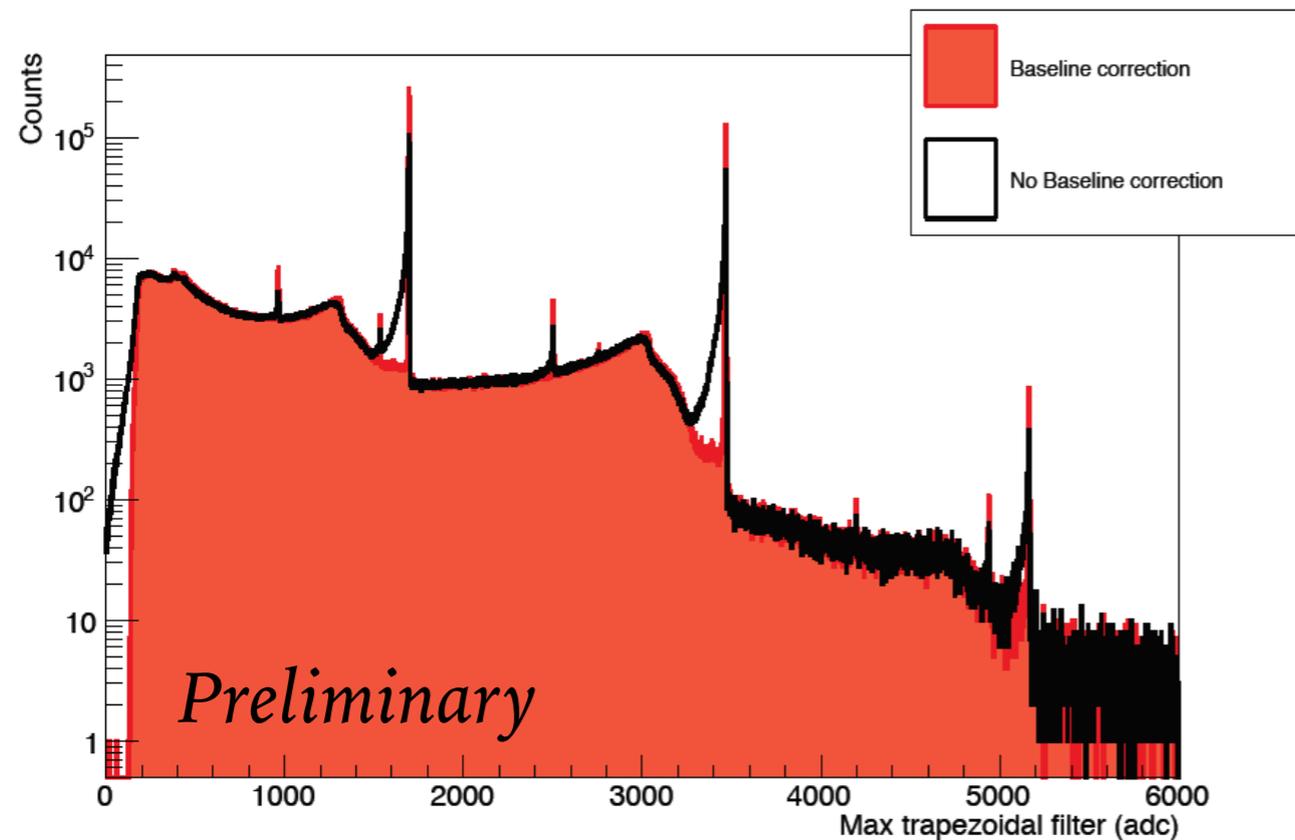
GAMMA, BETA BACKGROUND OF ^{226}Ra SOURCE

- A 5 μg radium-226 target leads to 200 kBq of all daughter
- Highest gamma emitters are lead-214 and bismuth-214
- The corresponding gamma rate is about 400 kHz



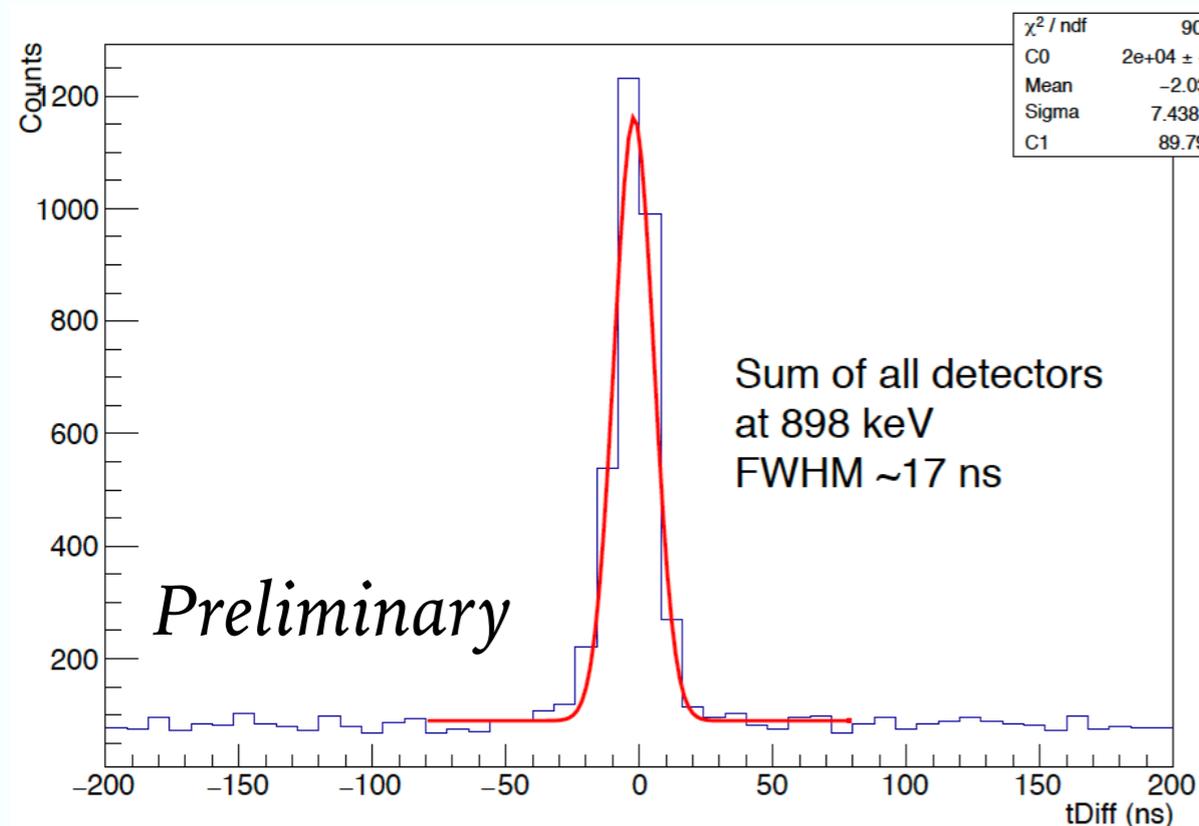
Decay chain of Radium-226

TIME AND ENERGY RESOLUTION RADIOACTIVE TARGETS



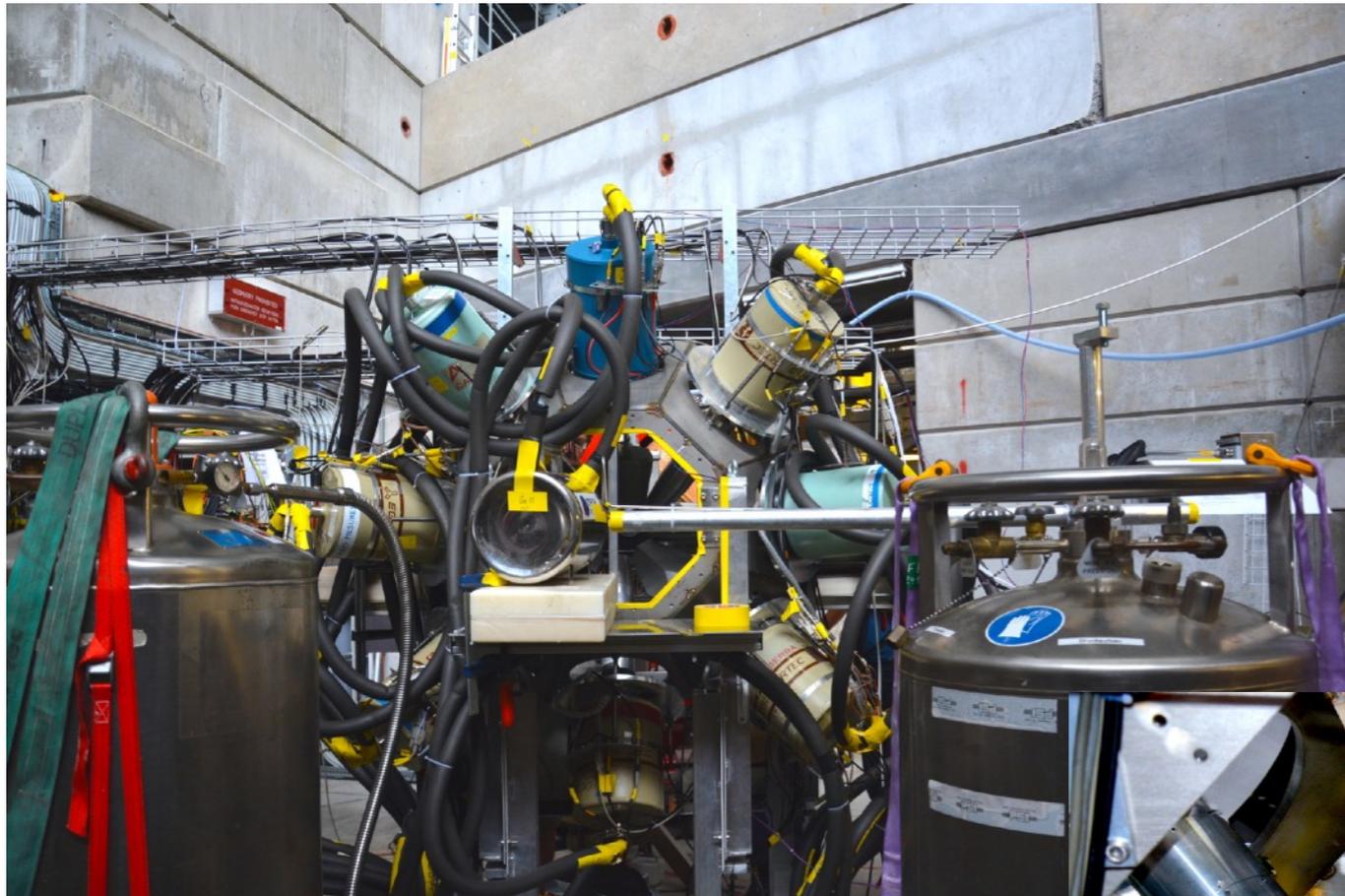
Effect of offline analysis on energy resolution

- Performed measurements of a high rate 420 kHz yttrium-88 gamma source
- The radium source has a similar activity
- Offline analysis improves time and energy resolution
- DAQ can handle high data rate



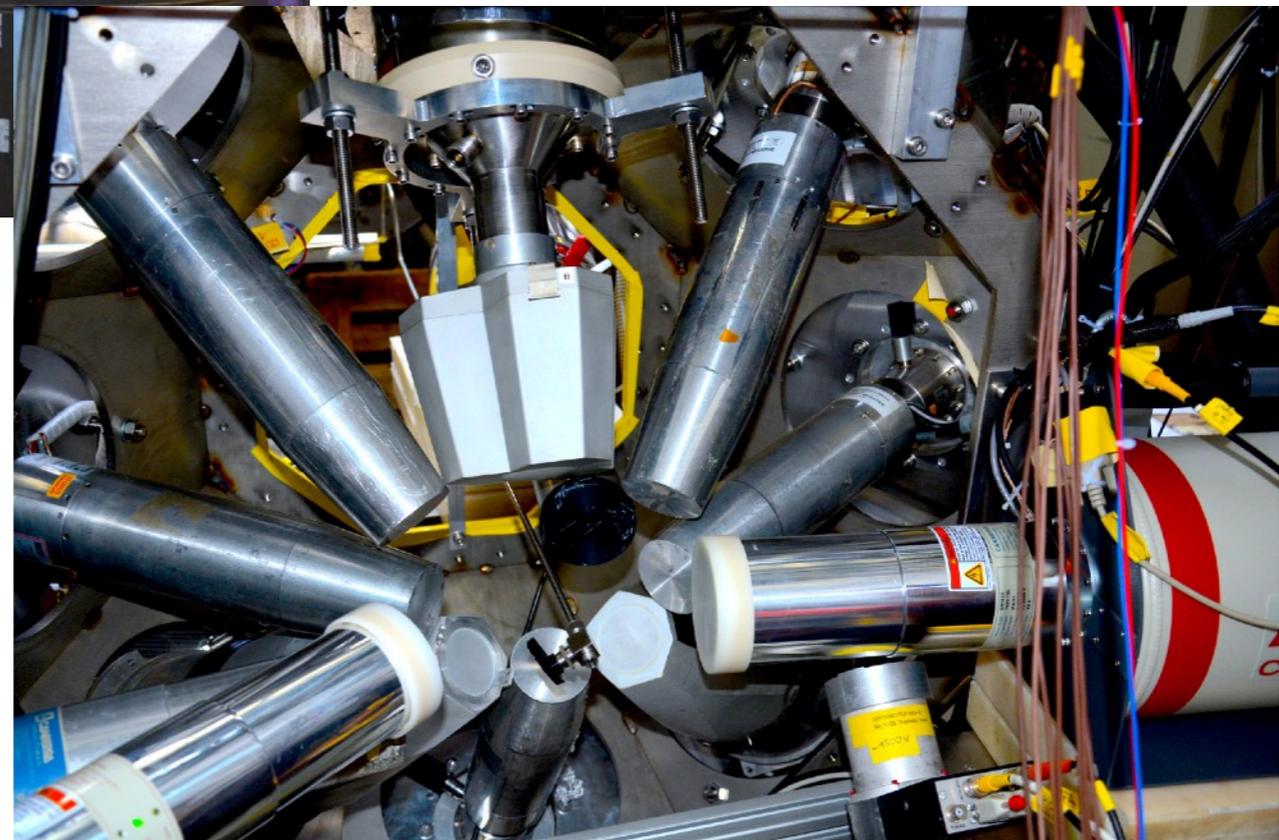
Time resolution of all Ge detectors at 898 keV

FIRST EXPERIENCES WITH CURIUM-248 AND RA-226 TARGETS



Detector setup radium-226 measurement

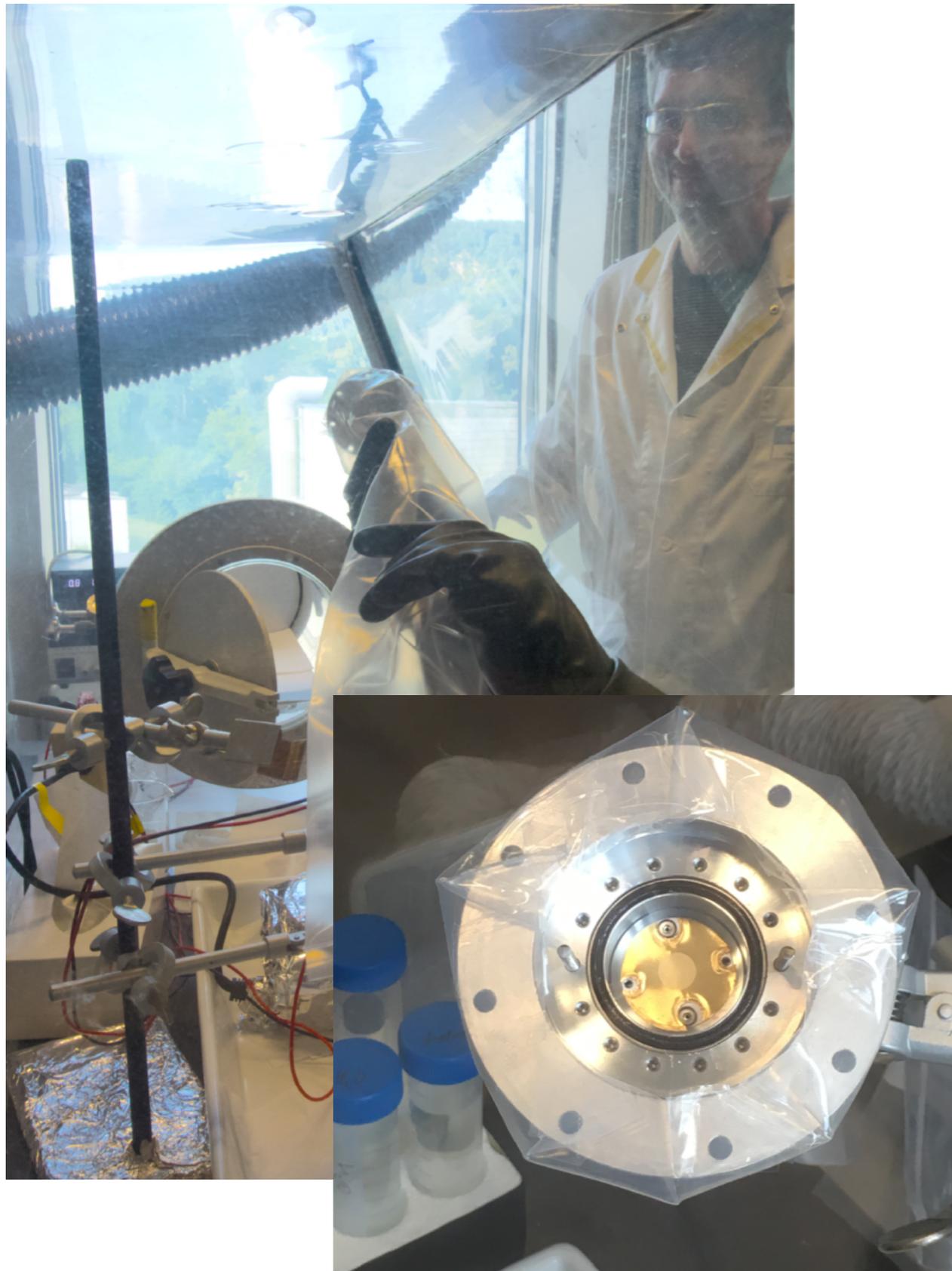
- During July 2018 the first attempt to observe muonic curium-248 and radium-226 happened
- The production of appropriate targets was unsuccessful



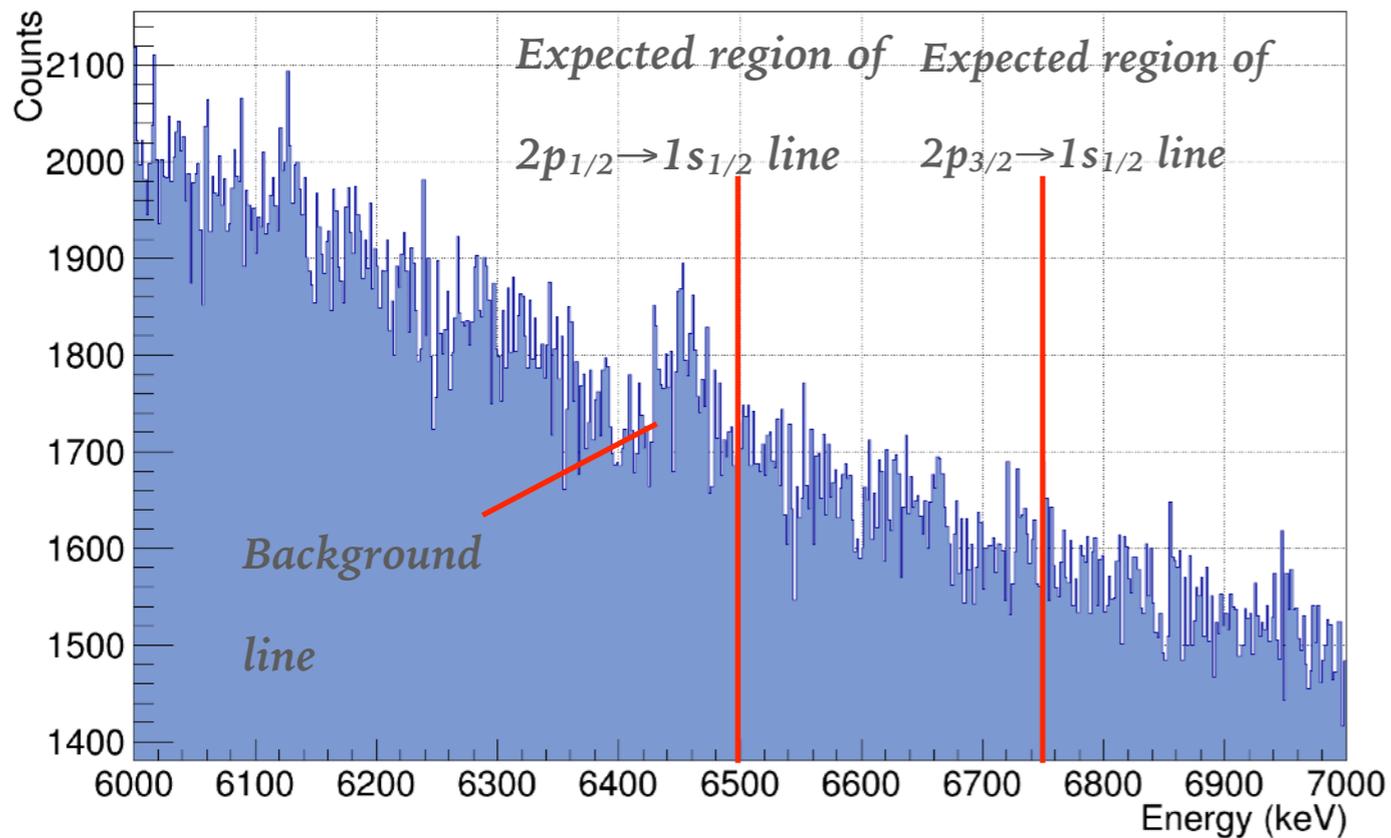
Detector array radium-226 measurement

FIRST TRY WITH CURIUM-248

- ▶ A $100\ \mu\text{m}$ copper plate is used as a substrate for curium-248. The copper is covered by a $50\ \text{nm}$ thin *gold* film to avoid unwanted oxidation
- ▶ Curium-248 was fixed on the gold layer by electrolysis
- ▶ The activity of the *curium-248* probe was $2,448\ \text{kBq}$ ($\sim 37\ \mu\text{g}$)
- ▶ The probe included an admixture of *curium-246* with an activity of $8,978\ \text{kBq}$ ($\sim 2\ \mu\text{g}$)



Curium gas cell prepared for sealing



Observed energy spectrum of ^{248}Cm measurement in the expected 2p-1s region



^{248}Cm and organic compounds on the gold plate

OBSERVED SPECTRUM CURIUM-248

-
- The estimated 2p → 1s transitions for Cm-248 are:
 - $2p_{1/2} \rightarrow 1s_{1/2} \sim 6500 \text{ keV}$
 - $2p_{3/2} \rightarrow 1s_{1/2} \sim 6754 \text{ keV}$
- No Cm lines were observed
- Cm was plated on the Au-Cu plate. Hence, a disturbing organic layer could cover the Cm target
- A flame treatment was applied to reduce the thickness of the organic layer. This resulted in no improvement of the gamma spectrum

^{248}Cm

^{246}Cm

5034.89(25) keV

5343.5(10) keV

5078.41(25) keV

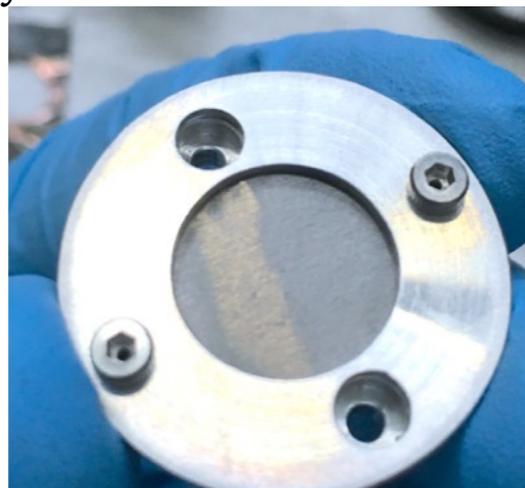
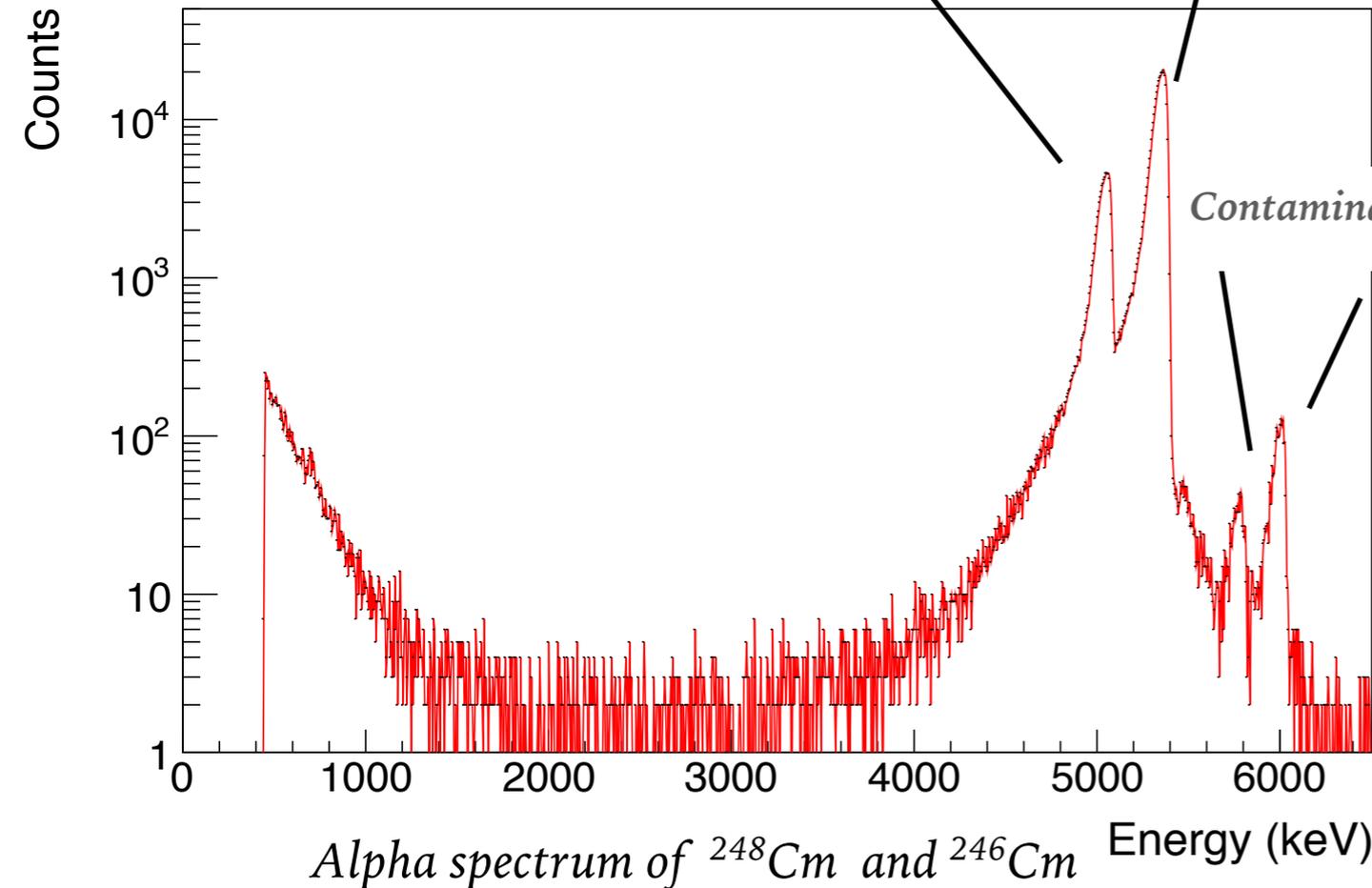
5386.5(10) keV

(Nuclear Data Sheets 146, 387 (2017))

Nuclear Data Sheets 96, 177 (2002)

ALPHA SPECTRUM CURIUM-248 TARGET

- The observed alpha spectrum of the target shows two lines corresponding to ^{248}Cm and ^{246}Cm
- Due to the organic layer:
 - The two lines of each isotope smear out
 - Peak shift of ~ 10 keV
 - Huge tails on the left side
- Tested effect of organic layer with carbon covered gold disks



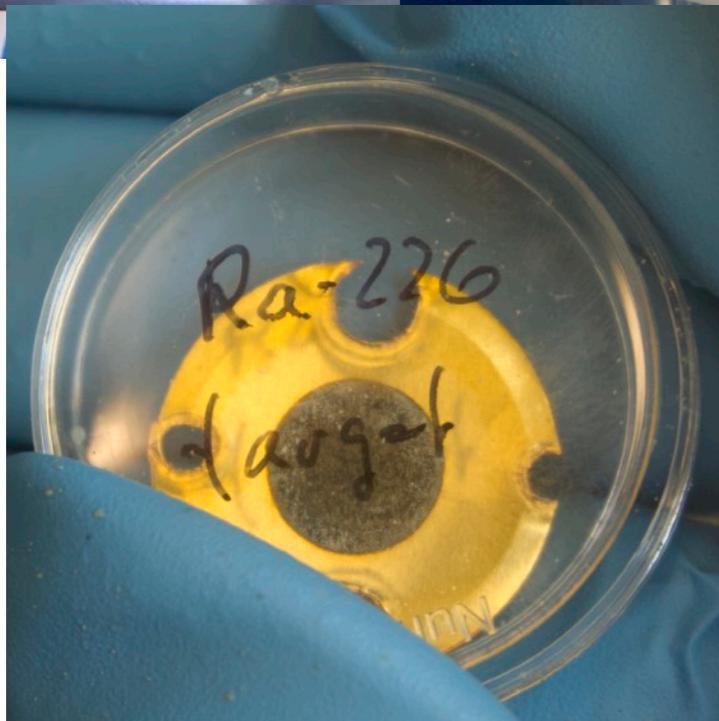
Carbon covered gold plate

FIRST TRY WITH RADIUM-226

- Ra-226 with an activity of 201.8 kBq is solved in acid
- During the first try it is plated on Cu-Au-Plate
- The plating destroyed the Au layer
- For a second plating the Ra-226 has to be removed from the Cu-Au-Plate
- During the separation occur many impurities in the solution
- The impurities have to be removed from the solution
- After all separations and the final plating only 1% of the original Ra-226 amount is left in the target
- A measurement in a reasonable time is not possible anymore



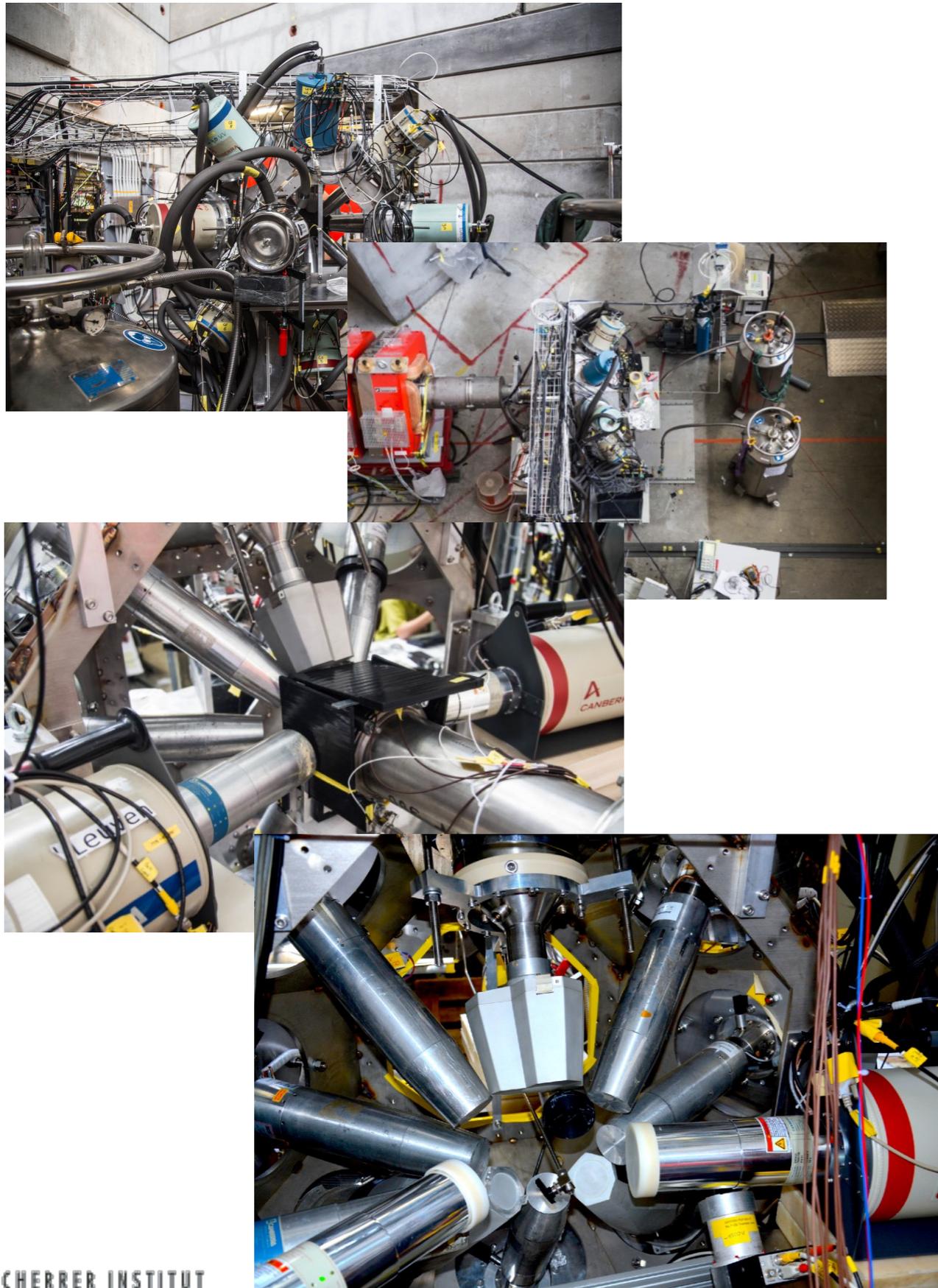
Container of radium-226



*First try of radium-226
plating failed*

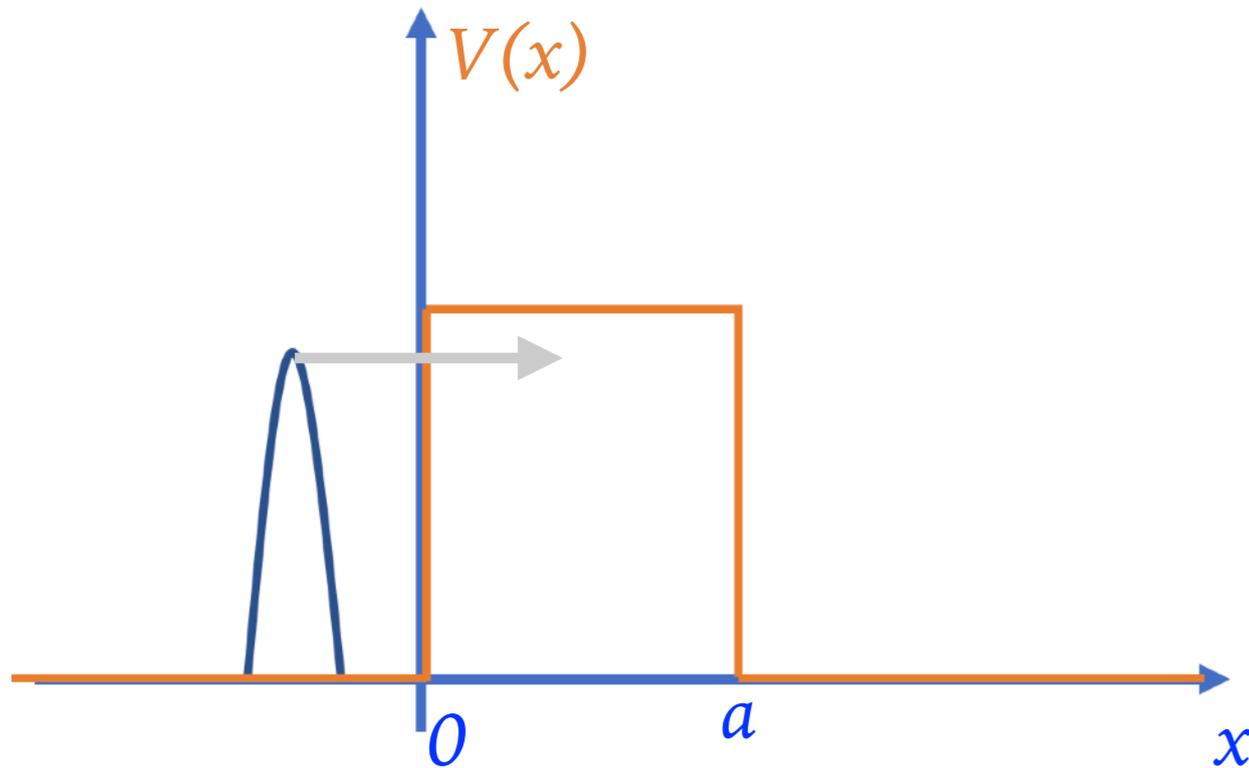
SUMMARY & OUTLOOK

- ▶ Muonic atom spectroscopy can be used for nuclear charge radius measurement
- ▶ Developed and tested a muon transfer method for tiny amount targets
- ▶ An improvement and quality assurance of the target production is required
- ▶ 2019 - Next try to observe muonic radium and curium X-rays



BACKUP SLIDES

RAMSAUER-TOWNSEND EFFECT



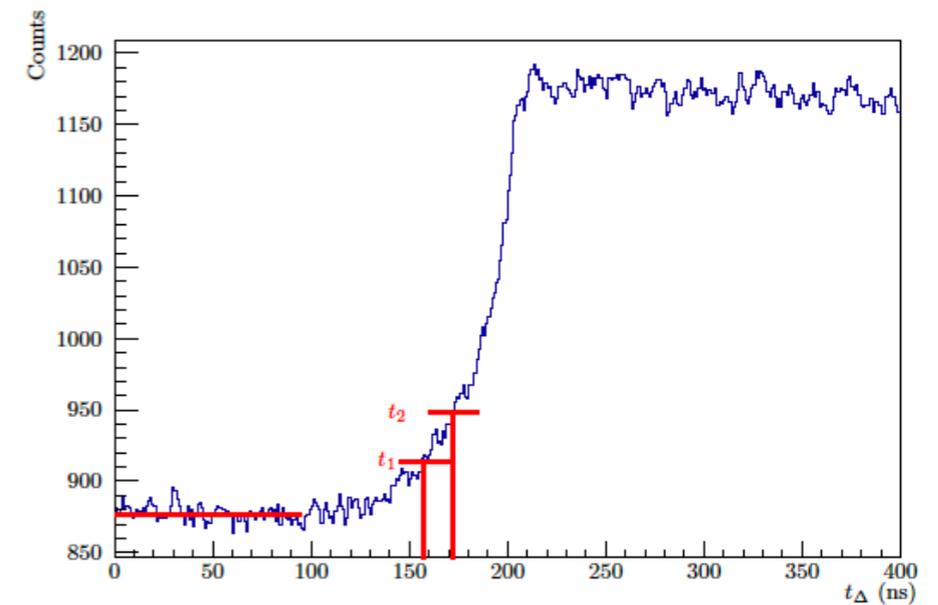
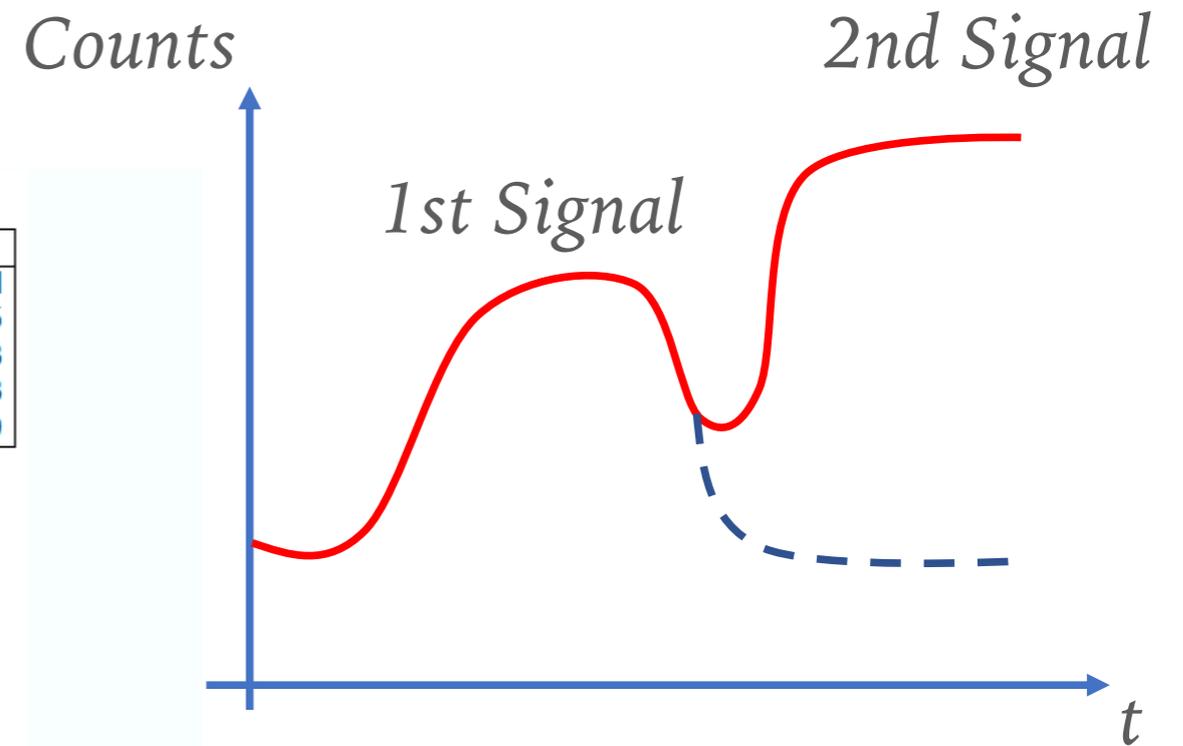
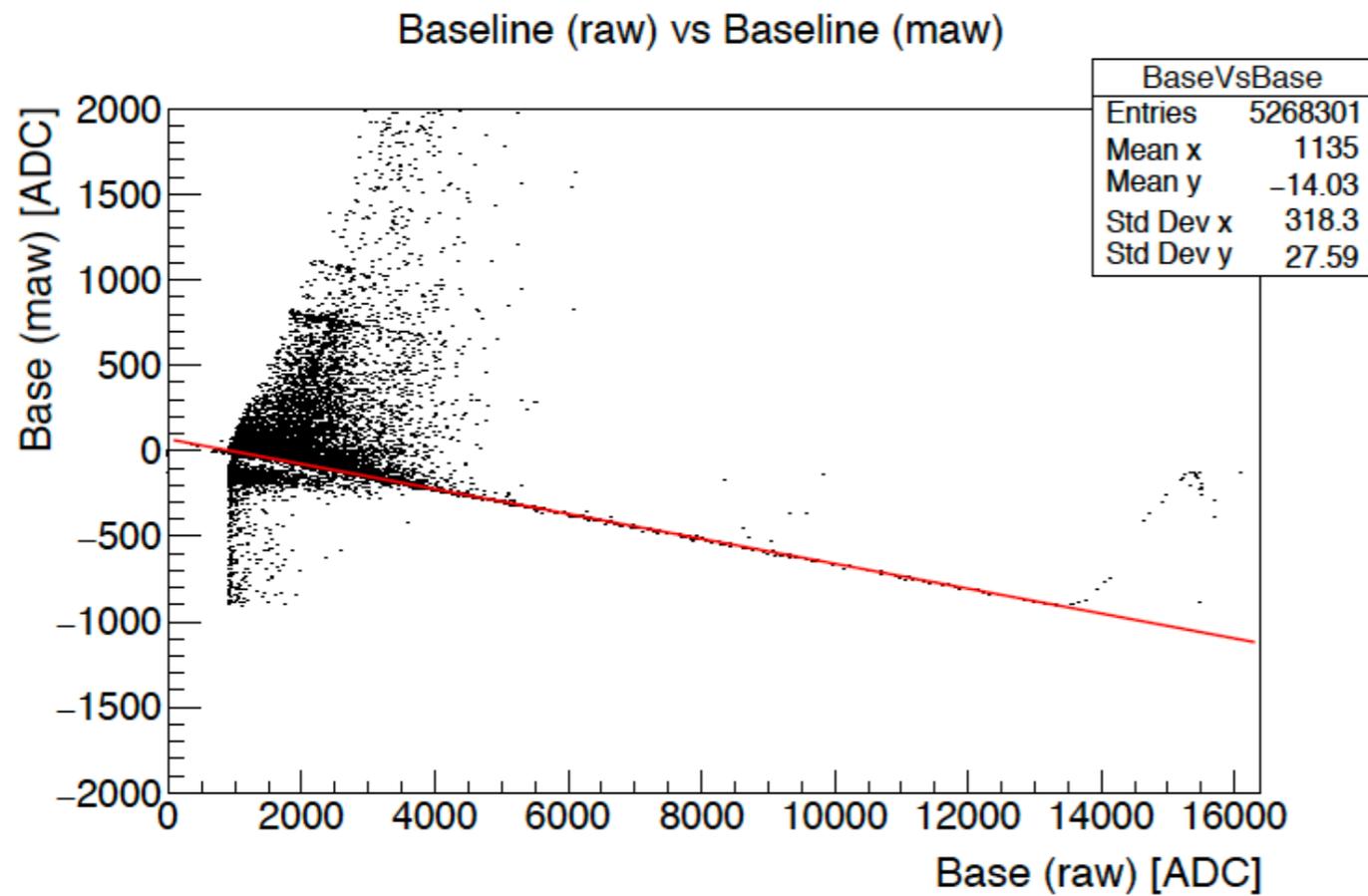
$$E\Psi = -\frac{1}{2m} \frac{\partial^2}{\partial x^2} \Psi + V\Psi$$

$$V(x) = \begin{cases} 0 & x < 0 \\ V_0 & 0 < x < a \\ 0 & x > a \end{cases} \quad \Psi(x) = \begin{cases} e^{ikx} + Re^{-ikx} & x < 0 \\ Ae^{iKx} + Be^{-iKx} & 0 < x < a \\ Te^{-ikx} & x > a \end{cases}$$

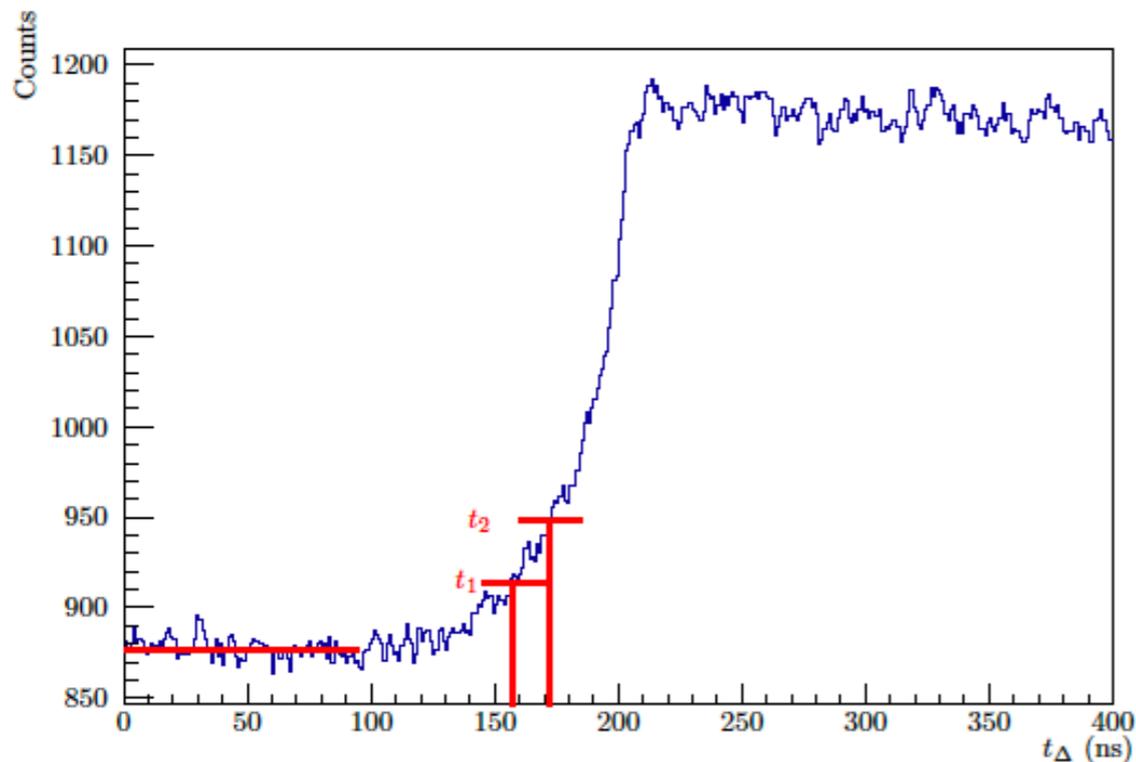
$$\mathcal{T} = |T|^2 = \frac{1}{1 + \frac{V_0^2}{4E(E-V_0)} \sin^2 Ka}$$

$$Ka = 2\pi \frac{a}{\lambda} = n\pi$$

BASELINE CORRECTION



ELET TIME CORRECTION



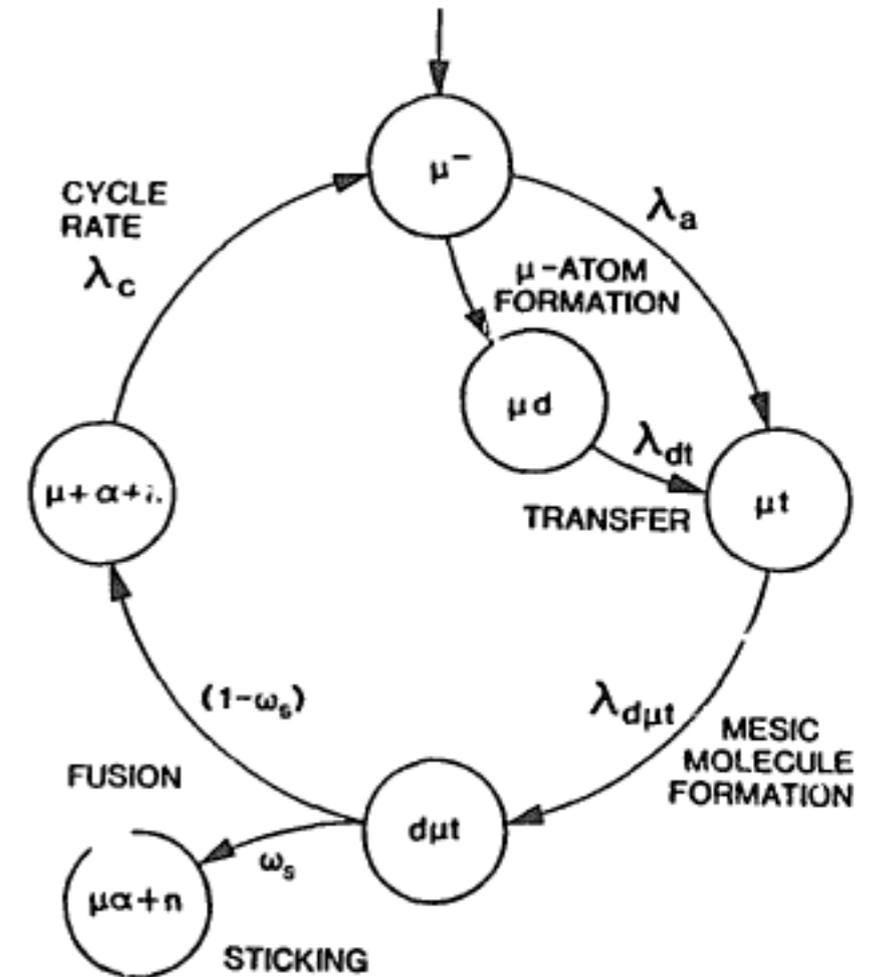
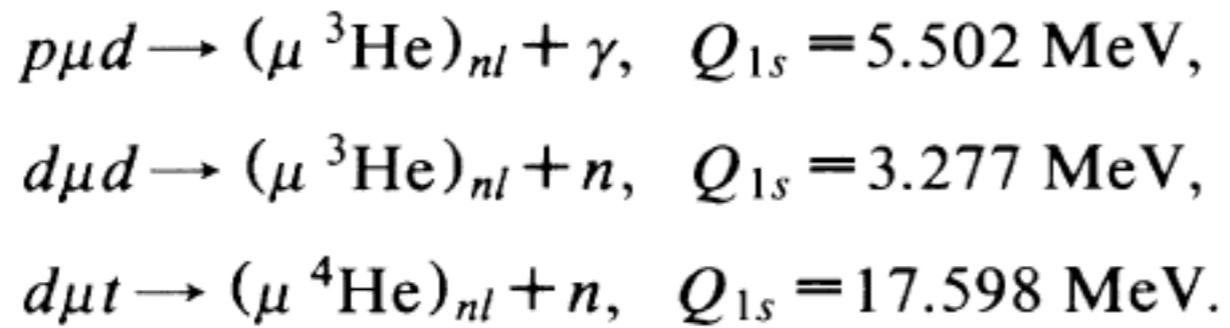
- Avoid threshold activation due to noise (jitter effect)
- Avoid walk effect due to varying time signal shape
- Assume that rising slope is almost linear at the beginning

$$\Delta t = t_1 - t_0$$

$$\Delta t = t_2 - t_1$$

$$t_0 = 2t_1 - t_2$$

MUON CATALYSED FUSION



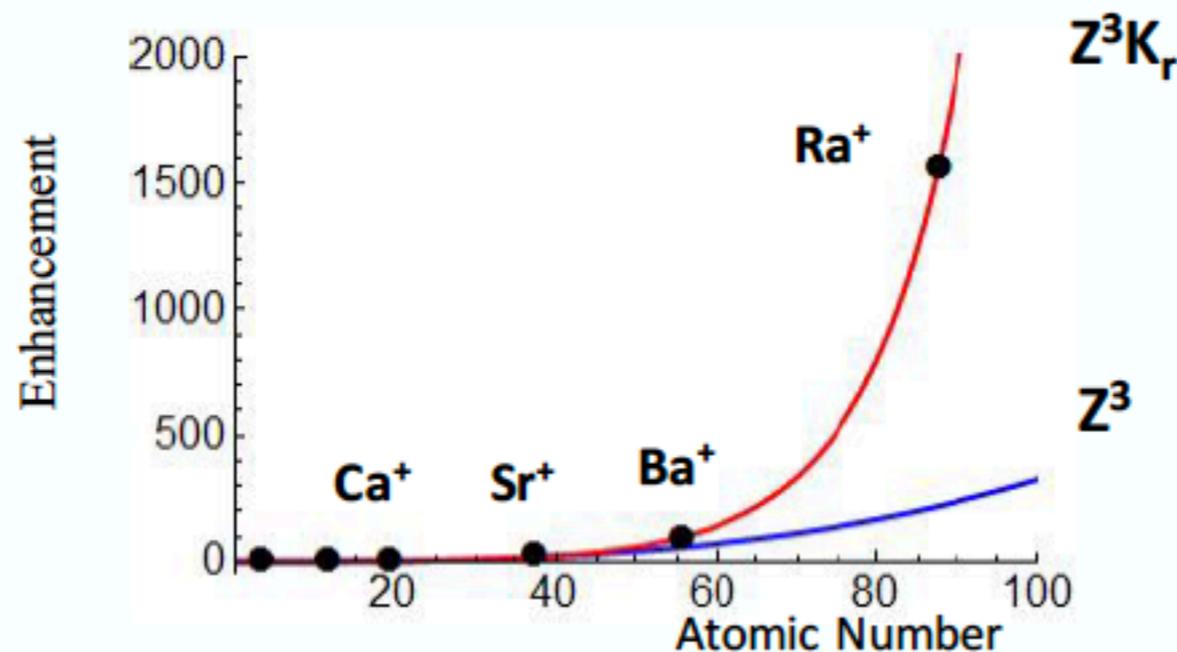
RADIUM APV

Scaling of the APV

increase faster than Z^3

(Bouchiat & Bouchiat, 1974)

$$\langle nS_{1/2} | H_W | nP_{1/2} \rangle \propto K_r Z^3 \quad K_r \text{ relativistic enhancement factor}$$



Ra^+ effects

larger by:

20 (Ba^+)

50 (Cs)

L. W. Wansbeek *et al.*,
Phys. Rev. A 78, 050501
(2008)

→ 5-fold improvement over Cs feasible in 1 day

Relativistic coupled-cluster (CC) calculation of $E1_{\text{APV}}$ in Ra^+

$$E1_{\text{APV}} = 46.4(1.4) \cdot 10^{-11} \text{iea}_0 (-Q_w/N) \quad (3\% \text{ accuracy})$$

Other results:

$$45.9 \cdot 10^{-11} \text{iea}_0 (-Q_w/N) \quad (\text{R. Pal } et al., \text{ Phys. Rev. A } 79, 062505 (2009), \text{ Dzuba } et al., \text{ Phys. Rev. A } 63, 062101 (2001).)$$

K. Jungmann, L. Willmann, Workshop
on Muonic Atom Spectroscopy (2016)

► Need reliable charge radius at <0.2% accuracy for atomic theory

ESTIMATION RADIUM PARAMETER

$$\rho(r) = \frac{\rho_0}{\left(1 + \exp\left(\frac{r-c}{a}\right)\right)} = \frac{\rho_0}{\left(1 + \exp\left(4\log(3)\frac{r-c}{t}\right)\right)}$$

$$r_{RMS} = \frac{\int \rho(r)r^2 dV}{\int \rho(r)dV} \quad t = 2.3 \text{ fm} \approx \text{const}$$

$$\text{rms} = 5.6841 \text{ fm}$$

I. Angeli/

Atomic Data and Nuclear Data Tables 87 (2004) 185\
[Dash]206