

The elusive ^{229}Th isomer - characterization of its properties

Benedict Seiferle
LMU Munich

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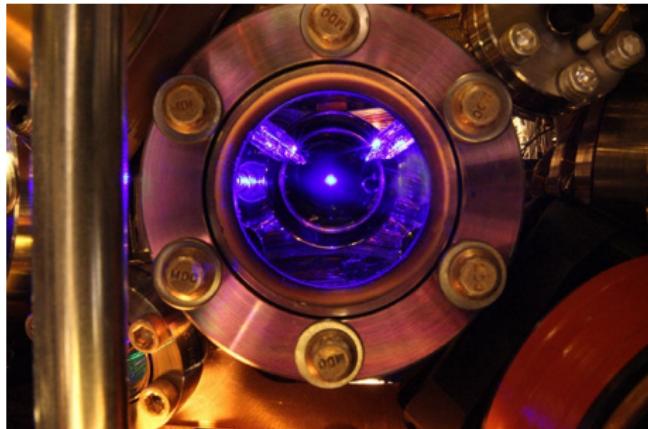
Outline

1. Motivation
2. Experimental Setup
3. Lifetime Measurements
4. Energy Measurements
5. Conclusion & Outlook

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Optical Atomic Clocks



- **ingredients:**
 - frequency reference: laser locked to a narrow atomic transition
 - counter or clockwork
- **best atomic clock:** Strontium lattice clock at NIST
 - frequency uncertainty [1]: 2.1×10^{-18}
 - precision is only limited by external electric and magnetic fields

Can one do better? - A nuclear optical clock

Idea: Use nuclear transition for time measurement [2]

- expected frequency uncertainty: 1.5×10^{-19} [3]

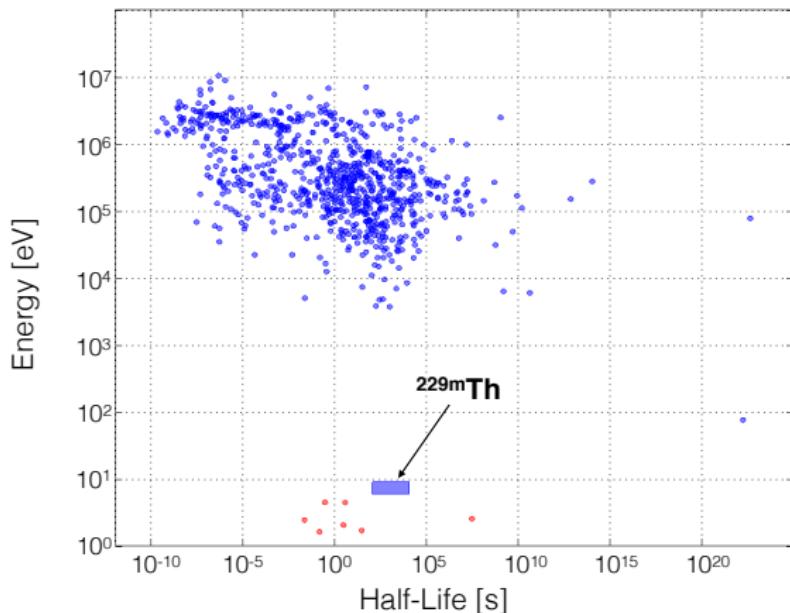
Expected advantages:

- nucleus is 5 orders of magnitudes smaller than the atom
 - highly resistant to external influences
- solid state clock feasible? [4]
 - 10^{19} atoms in crystal lattice vs. 10^4 in an optical lattice

[2] E. Peik & C. Tamm, Europhys. Lett. 61 (2003) 181. [3] C. Campbell et al., Phys. Rev. Lett. 108 (2012) 120802. [4] W.G. Rellergert et al., Phys. Rev. Lett. 104 (2010) 200802.

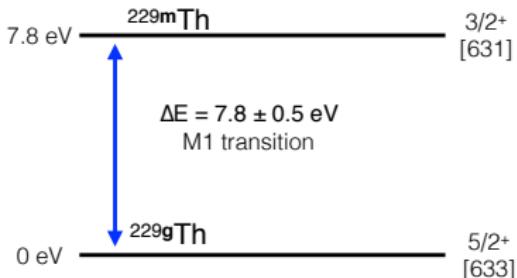
Requirements for a nuclear clock transition

- laser access
 - transition energy in the eV range
- small linewidth
 - lifetime in the range of at least some seconds



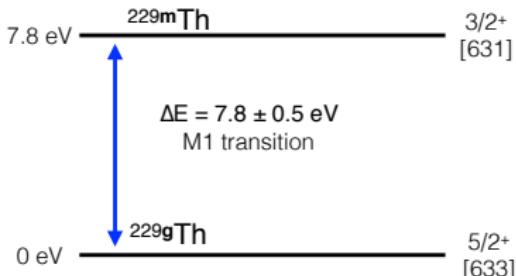
^{229}Th - What is known so far

- lowest excitation energy of all known nuclear states
 $\rightarrow E_I = 7.8 \pm 0.5 \text{ eV} (\approx 159 \text{ nm})$ [5]



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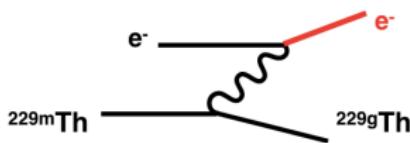


γ -decay



- emission of a photon
 $\rightarrow E_\gamma = E_I$
- $\tau \approx 10^4 \text{ s}$
 $\rightarrow \Delta E/E \approx 10^{-20}$

internal conversion decay



- emission of an electron
 $\rightarrow E_e = E_I - E_b$
- $0.6 \mu\text{s} < \tau < 14 \mu\text{s}$ [6]
- only possible in neutral $^{229\text{m}}\text{Th}$

Why don't you simply build a nuclear clock?

- direct nuclear laser excitation has not been achieved so far
- lifetime is unknown
- transition energy is not known precisely

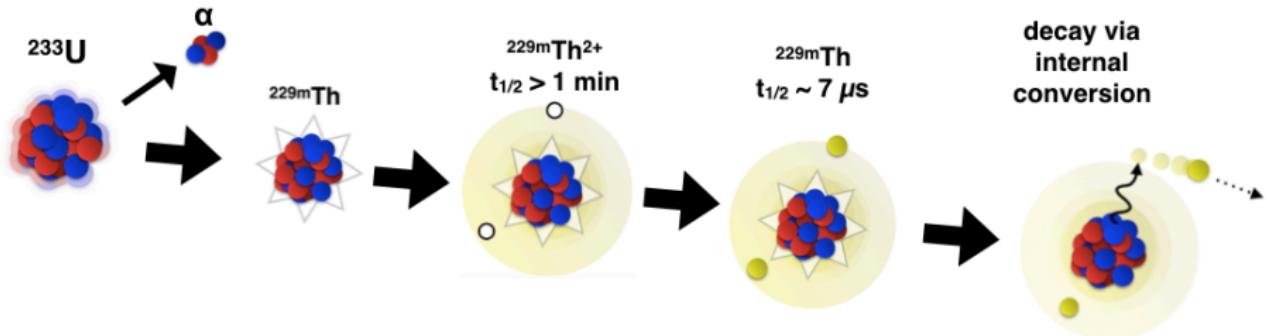
Experimental Objectives

- determine the lifetime
 - for internal conversion (section 3)
 - for γ -decay (outlook)
- improve the precision of the transition energy to better than 0.1 eV
(ongoing)

Outline

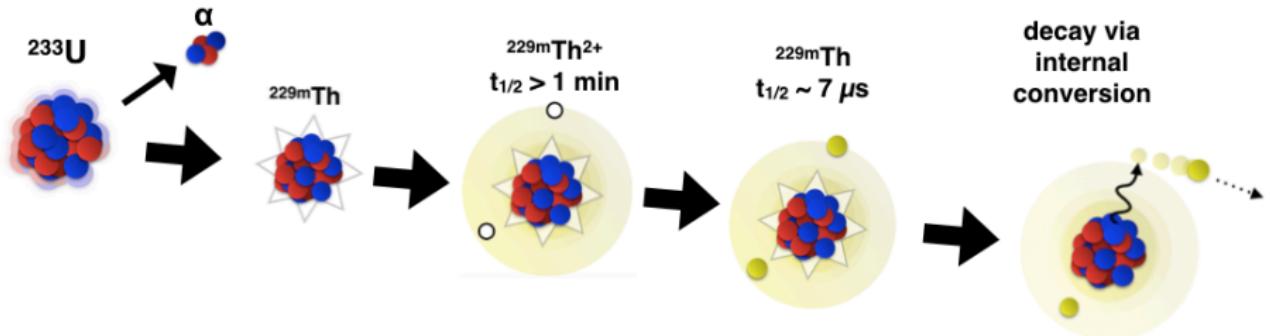
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Experimental Principle



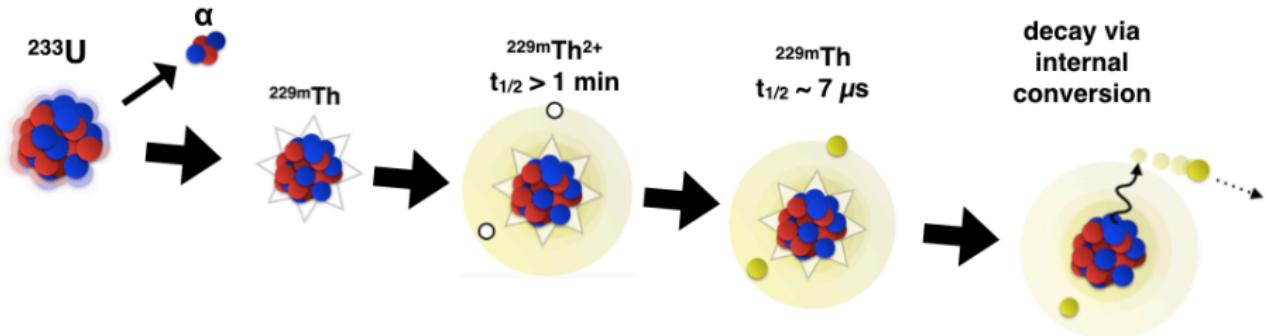
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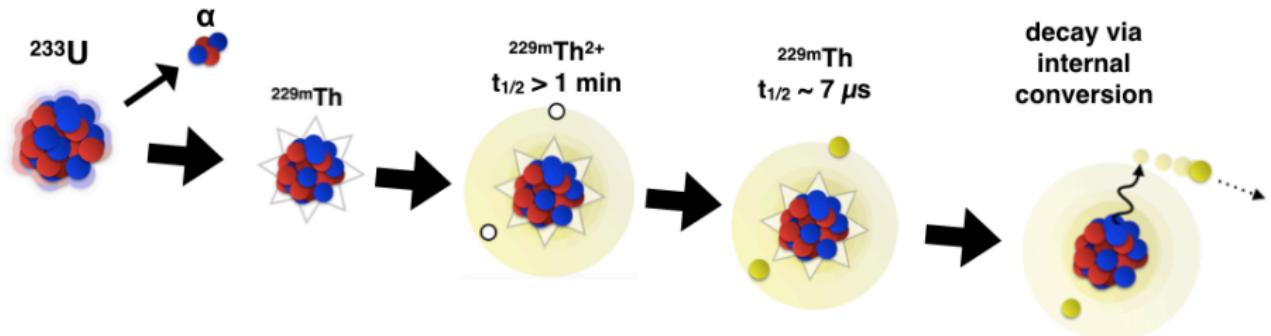
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- neutralize the ^{229m}Th ions

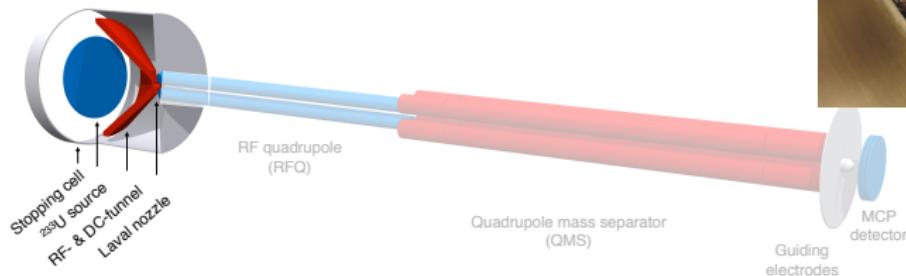
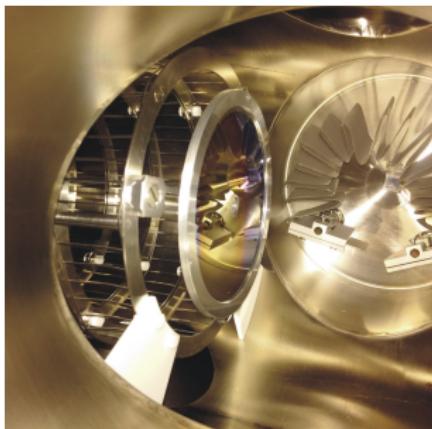
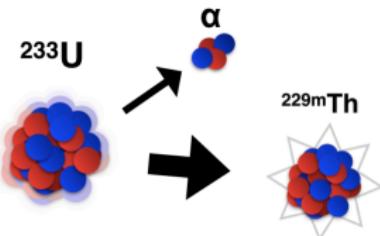
Experimental Principle



- populate ^{229m}Th via the 2% decay branch of the ^{233}U α -decay
- create a pure ^{229m}Th ion beam
- neutralize the ^{229m}Th ions
- detect the electron emitted during the internal conversion decay

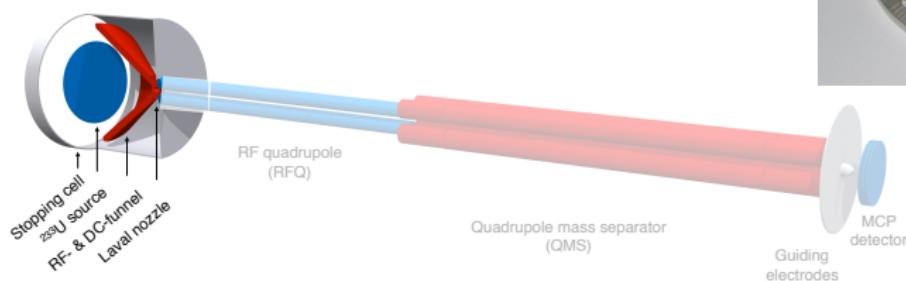
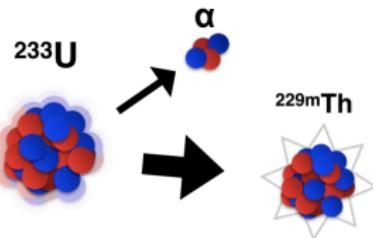
^{233}U source and buffer gas stopping cell

- ^{233}U source (260 kBq)
 - ^{229m}Th recoil ions (84 keV)
- buffer gas stopping cell (40 mbar He)
 - ^{229}Th recoil ions are stopped
- electric RF- and DC-funnel guides ions towards a Laval Nozzle ($\varnothing 0.6\text{ mm}$)
 - supersonic gas jet created
 - ions follow the jet into the next chamber



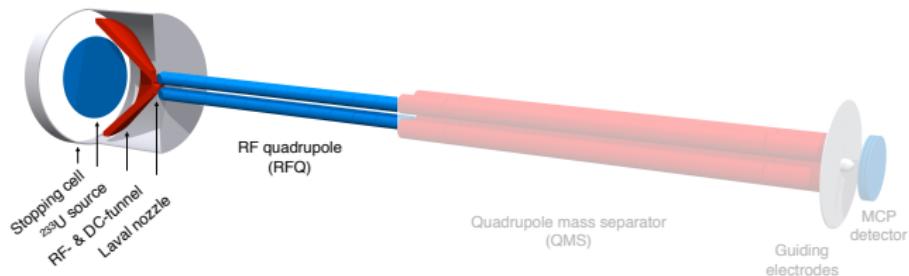
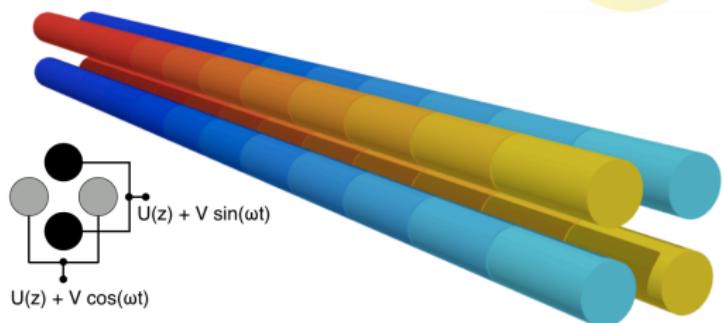
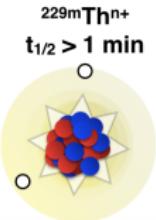
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Radio Frequency Quadrupole (RFQ)

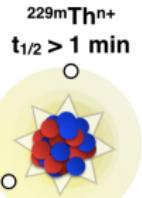
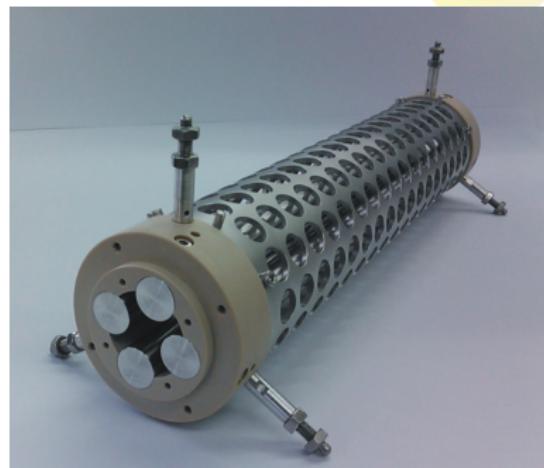
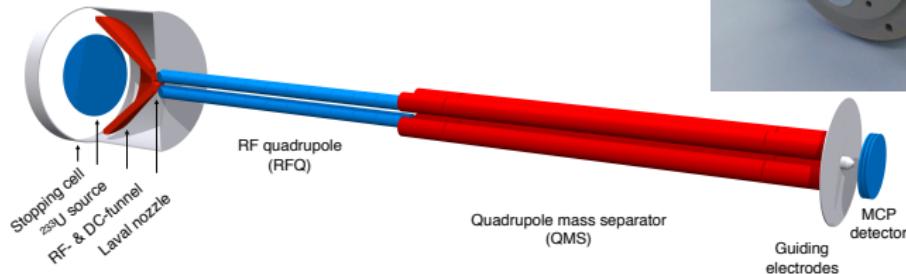
- segmented RF-quadrupoles
 - $f = 850 \text{ kHz}$
 - $V_{pp} \approx 150 \text{ V}$
- DC-gradient
 - guide the ions through the remaining buffer gas
 - cooled ion beam
 - ion bunches



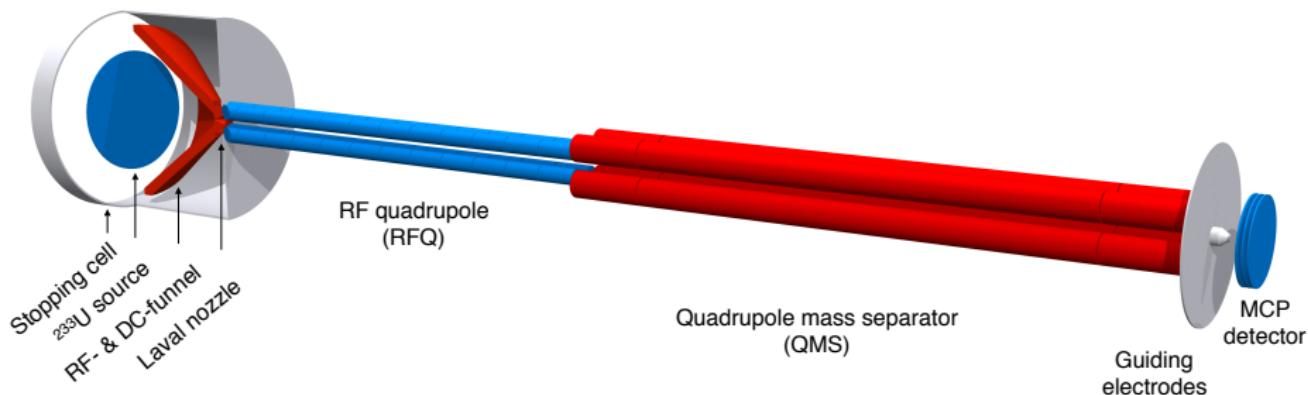
Quadrupole Mass Separator (QMS)

- ion beam contains ^{233}U daughter isotopes
- QMS purifies ion beam [7]
 $\rightarrow m/\Delta m = 150$ @ 80% efficiency
- behind the QMS: electrostatic lens

[7] Design values taken from: E. Haettner et al., Nuclear Inst. and Methods in Physics Research, A 880 (2018) 138-151.



Experimental Setup - Overview



Pure ^{229m}Th ion beam/bunches:

continuous extraction mode:

- $\approx 10\,000\, {}^{229}\text{Th}^{3+}$ ions per second
- $\approx 8\,000\, {}^{229}\text{Th}^{2+}$ ions per second

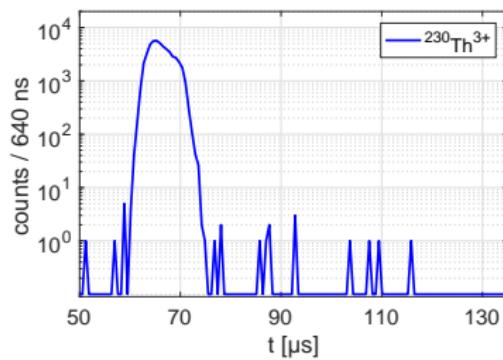
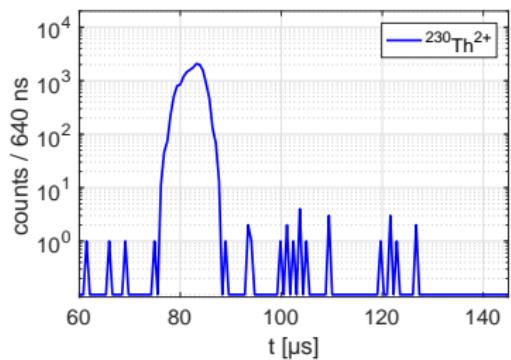
continuous extraction mode:
bunch mode:

- $\approx 400\, {}^{229}\text{Th}^{3+}/\approx 300\, {}^{229}\text{Th}^{2+}$ ions per bunch @ 10 Hz repetition rate
- $\Delta \text{TOF}_{\text{FWHM}} < 10\, \mu\text{s}$
 \rightarrow high (peak) flux: $4 \times 10^7\, 1/\text{s}$

Lifetime Measurements

Measurements with ^{230}Th :

- clear signal from ionic impact for $^{230}\text{Th}^{2+,3+}$



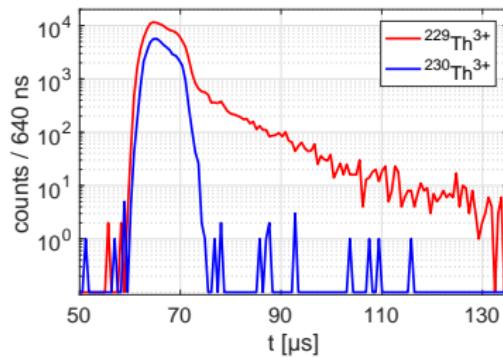
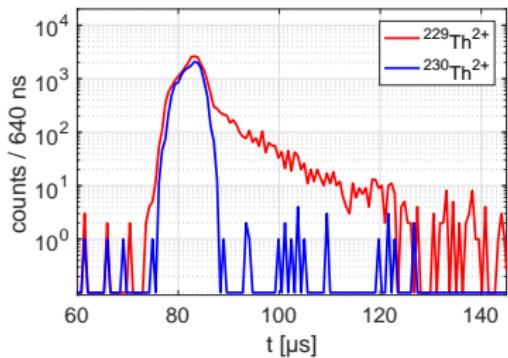
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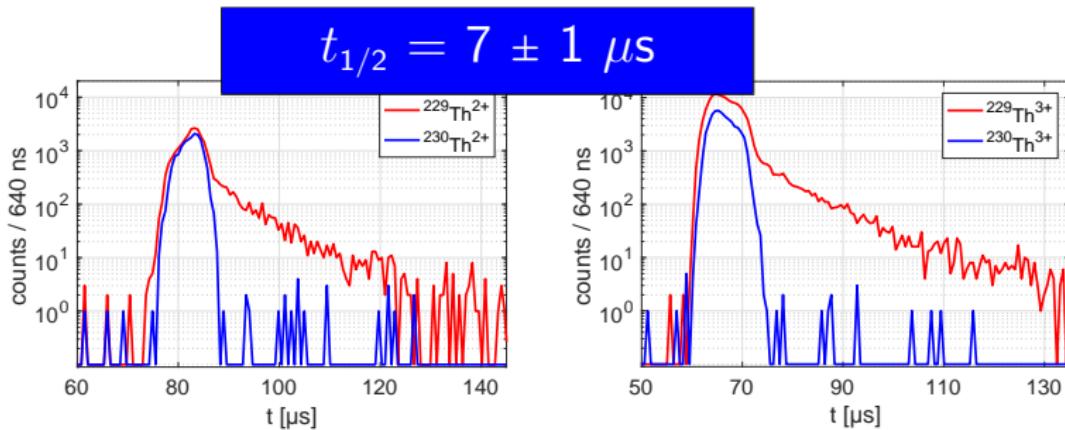
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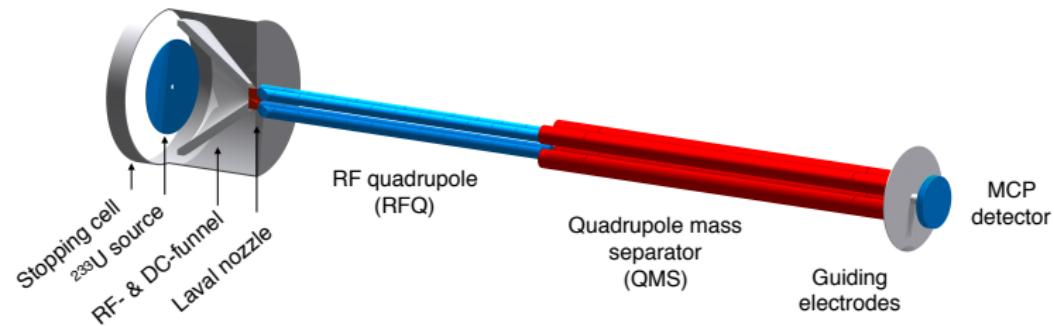
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Experimental Setup - Overview

measure the kinetic energy of the IC electrons!

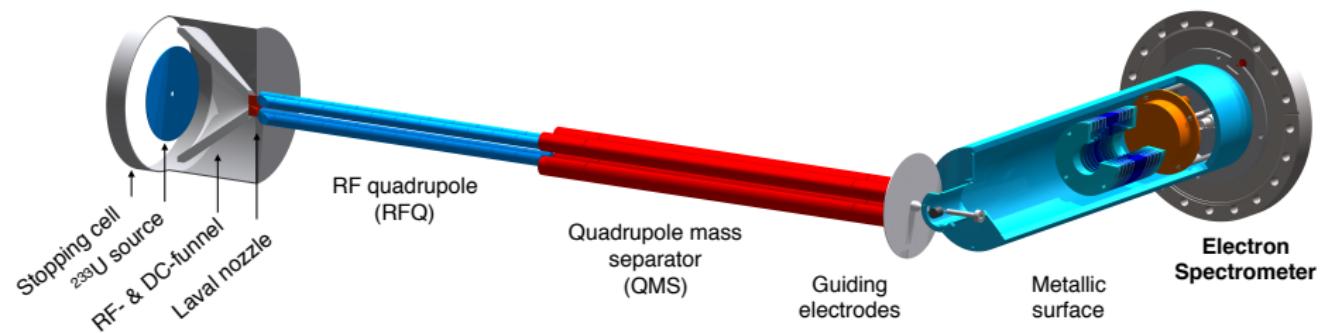
- replace MCP detector with metallic surface in an electron spectrometer



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IC Electron Spectroscopy

$$E_I = E_e + E_B$$

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Neutralization on a surface

- electrons are delocalized
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Surface cleanliness

- monolayer formation time @ 10^{-8} mbar: ≈ 300 s
- cleanliness affects the surface properties
 - work function reduction
 - binding energies of e^- (\approx eV)

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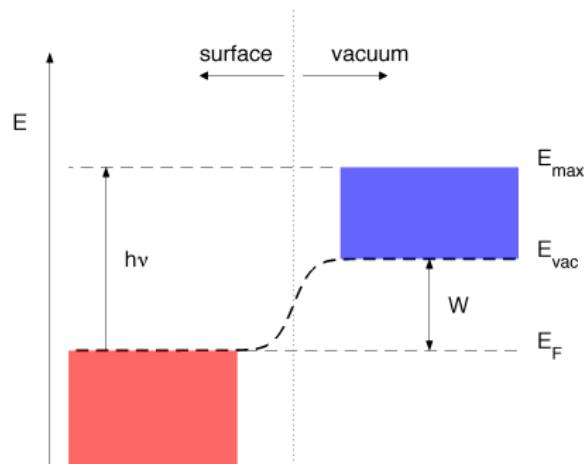
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We need surface-independent measurement principle!

From Photoelectron Spectroscopy to IC Electron Spectroscopy

Monochromatic light source $h\nu$ + metallic surface:

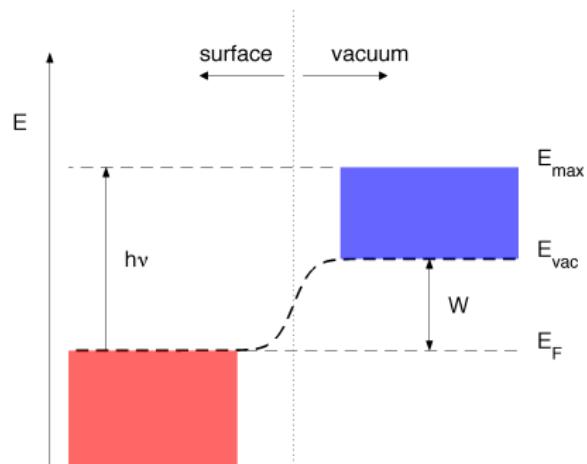
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- zero binding energy: fermi-edge
→ electron from fermi edge have maximum kinetic energy E_{\max}



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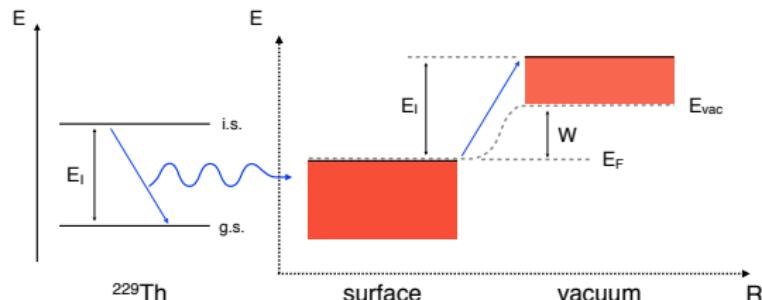
Measure kinetic energy with spectrometer:

- apply offset voltage to sample U
- spectrometer work-function W_S
 - $E_{\max} = h\nu - W_S - U$
 - independent from sample work-function

$$h\nu = E_{\max} + U + W_S$$

From Photoelectron Spectroscopy to IC Electron Spectroscopy

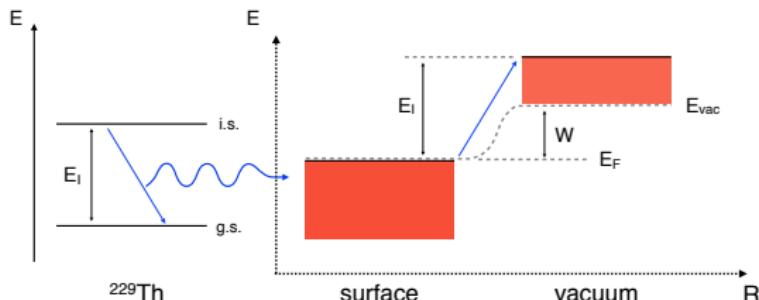
- collect ^{229m}Th on a metal surface
- electrons are delocalized
→ replace $h\nu$ with E_I



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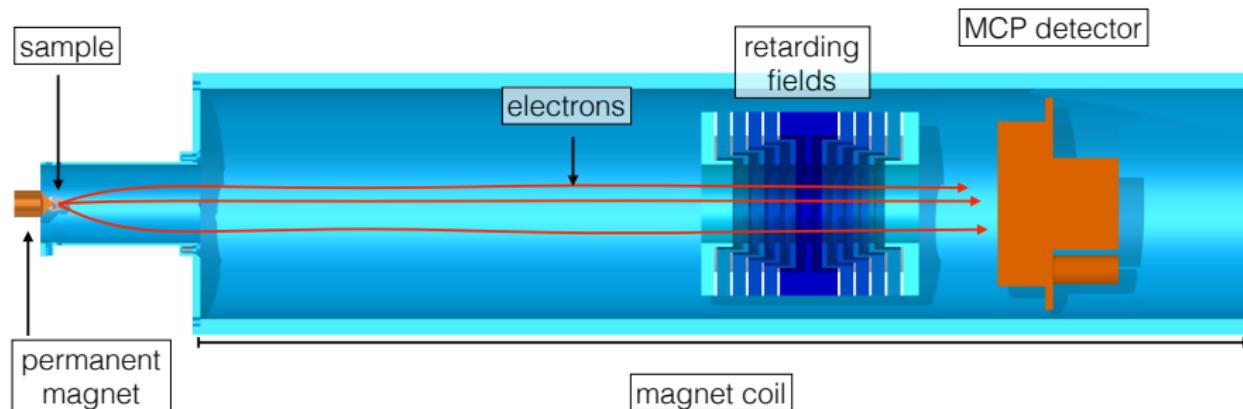


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measured with isomer

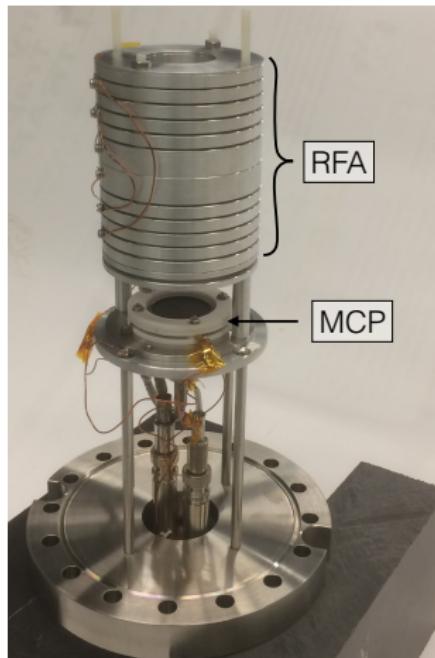
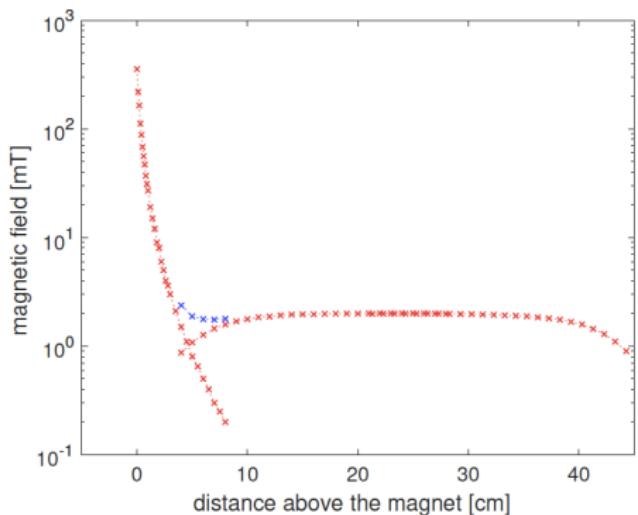
calibration measurements

Magnetic Bottle Spectrometer [9]



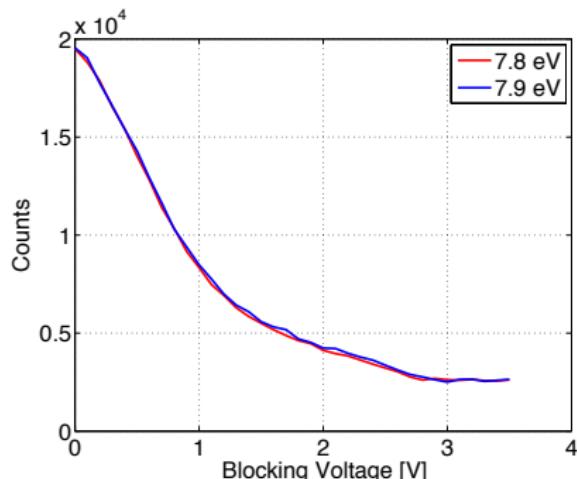
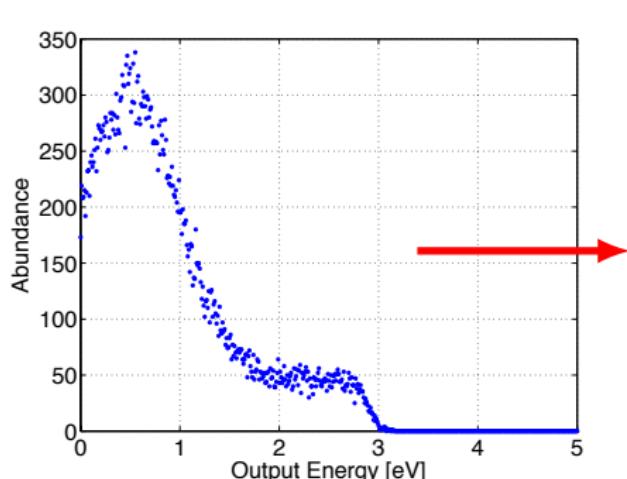
- (IC) electrons are emitted from a metallic sample
- **strong permanent magnet** → strongly inhomogeneous magnetic field
- **magnetic coil** → weak homogenous magnetic field
 - electrons are collected and guided towards weak magnetic field
 - collimated electron beam
- **retarding field analyzer**
 - electrons with energy larger than some blocking voltage reach detector

Magnetic Bottle Spectrometer [9]



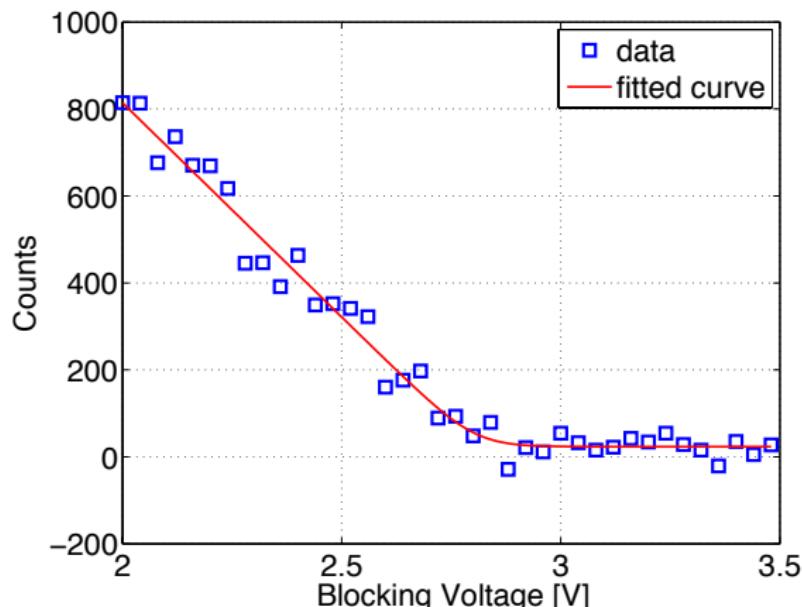
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- integrated spectrum
- how to obtain E_{max} ?
 - fit indefinite integral of the Fermi function to high-energy part of the spectrum



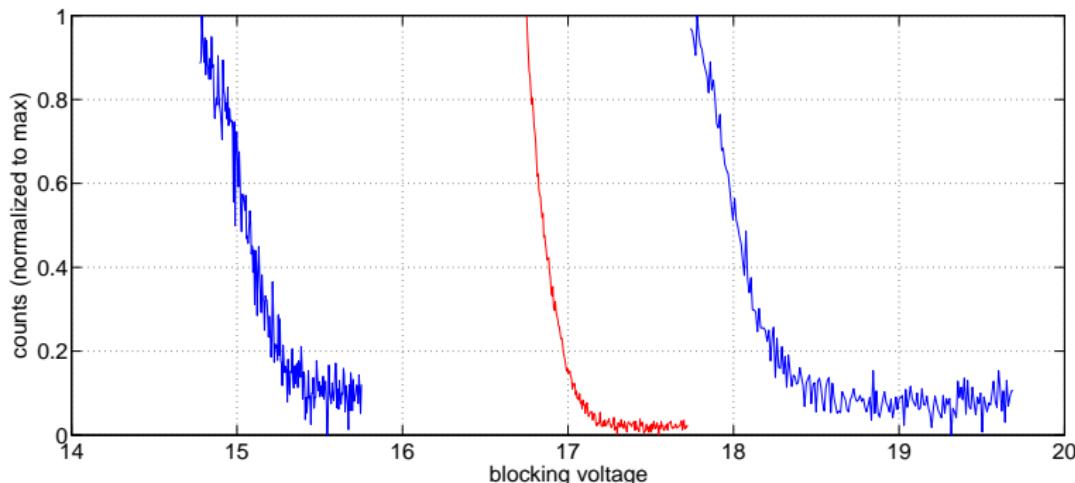
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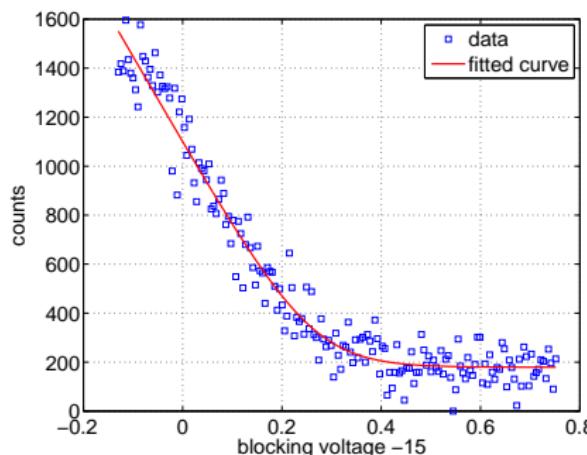
Calibration Measurements

- $h\nu = E_{\max} + U + W_S$
- calibration: measure $U + W_S$ (@ $U = -15$ V)
- spectrometer is calibrated with three lines
 - $h\nu = 4.888$ eV (Hg discharge 254 nm)
 - $h\nu = 6.705$ eV (Hg discharge 185 nm)
 - $h\nu = 7.897$ eV (F_2 excimer laser 157 nm)



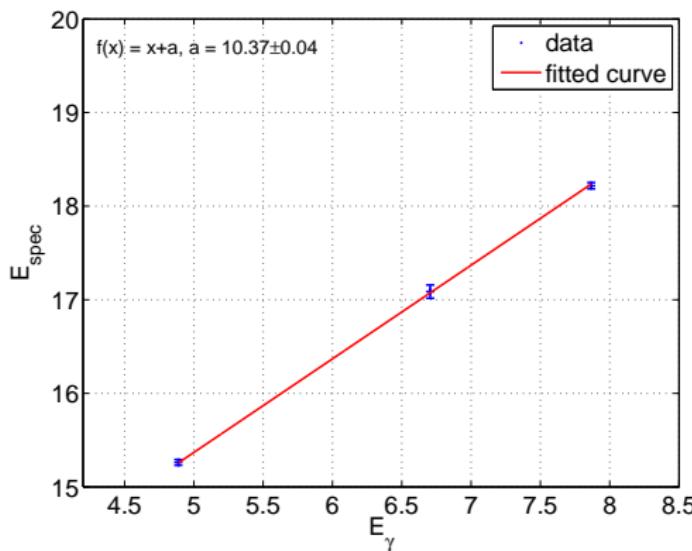
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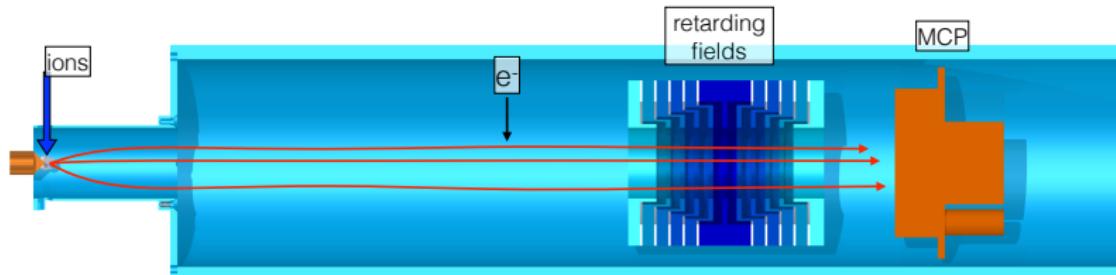
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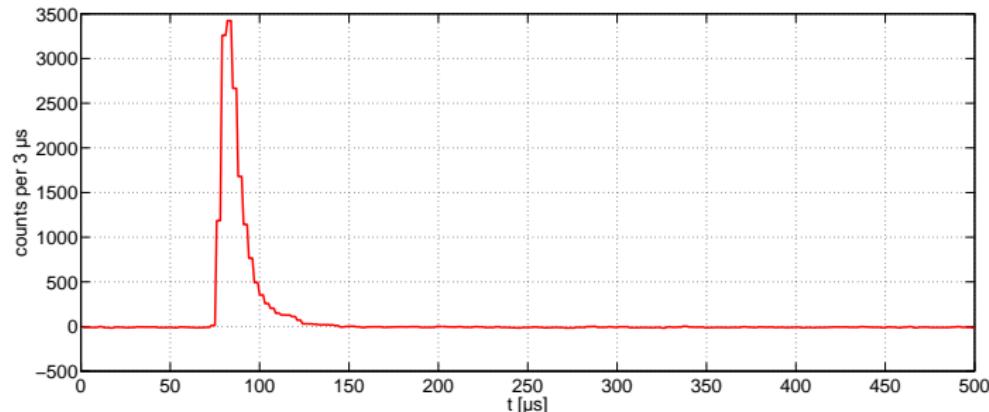
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- for the following measurements
 - $U + W_S = -10.59 \pm 0.05$ eV

IC Electron Measurements



- collect bunched $^{229m}\text{Th}^{2+}$ ion beam on a platinum surface (set to -15 V)
- -15 V blocking voltage
- clear (decay) signal
→ $E_I > 4.4 \text{ eV}$

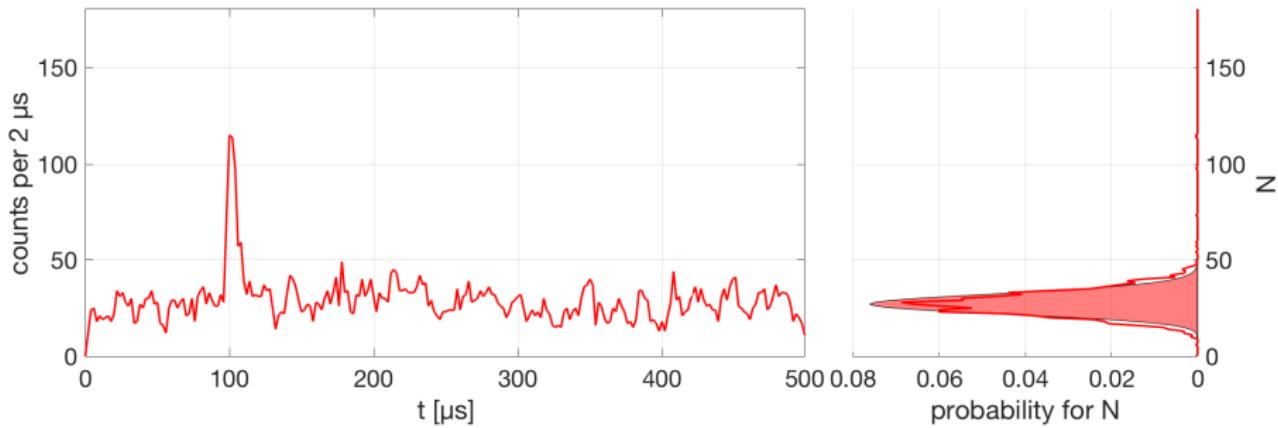
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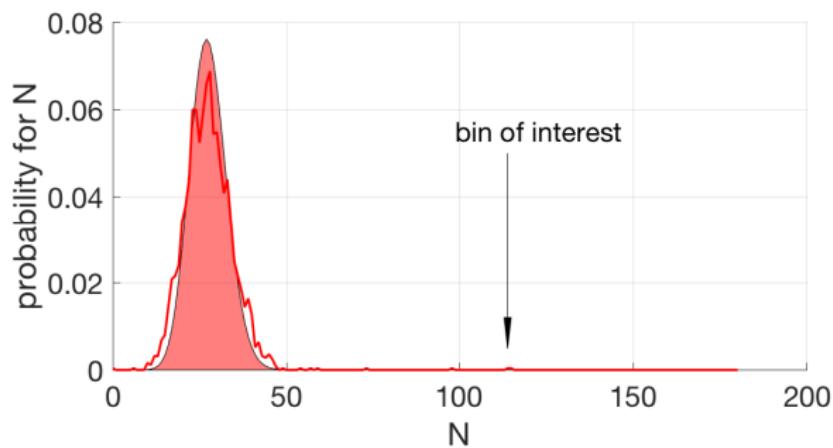
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- 19.0 V blocking voltage correspond to 8.41 ± 0.05 eV
- clear signal above Poisson distributed noise



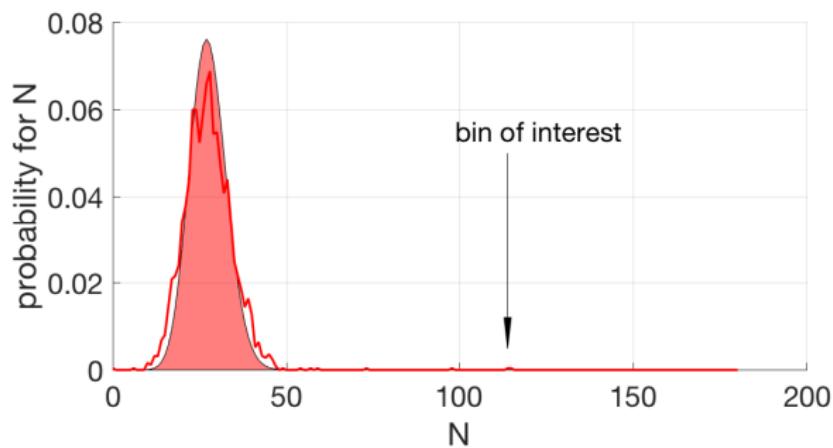
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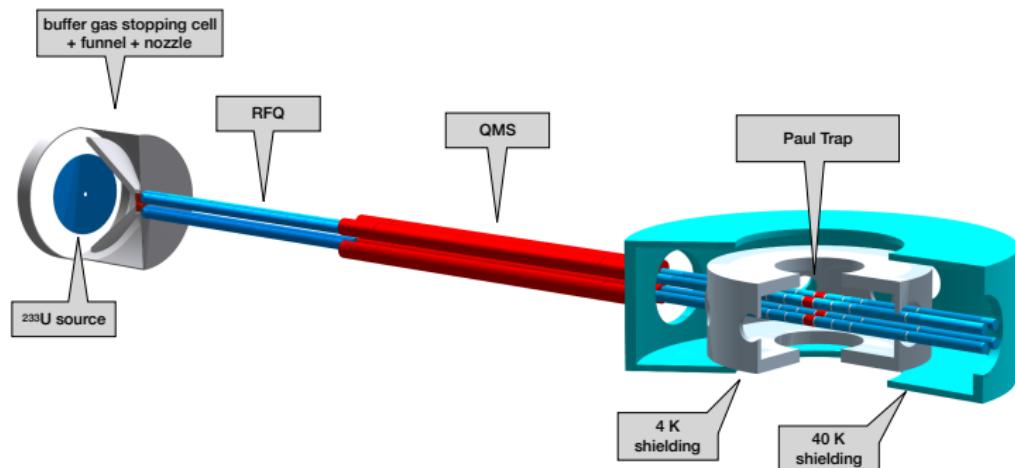
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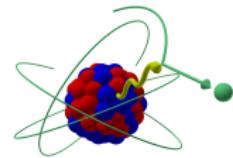


Lifetime Measurement of the γ decay channel (... and someday Th nuclear laser spectroscopy)

- requires long storage times (without any direct manipulation)
→ Cryogenic Paul Trap (CryPTEEx design, MPIK Heidelberg [10])
- Sympathetic laser cooling with Sr^+

Conclusion

- ^{229}Th is a special nucleus
 - lowest energy of all known nuclear levels
 - nuclear optical clock
- lifetime measurement of the internal conversion decay channel
 - $t_{1/2} = 7 \mu\text{s}$
 - how strong is the surface-dependence?
- energy measurement of IC electrons ongoing



Thank you for your attention!

Benedict Seiferle¹, Lars von der Wense¹, Nicolas Arlt¹, Benjamin Kotulski¹, Mustapha Laatiaoui^{2,3}, Jürgen B. Neumayr¹, Hans-Jörg Maier¹, Hans-Friedrich Wirth¹, Christoph Mokry^{3,4}, Jörg Runke^{2,4}, Klaus Eberhardt^{3,4}, Christoph E. Düllmann^{2,3,4}, Norbert G. Trautmann⁴ & Peter G. Thirolf¹

1 Ludwig-Maximilians-Universität München, **2** GSI Helmholtzzentrum für Schwerionenforschung GmbH, **3** Helmholtz-Institut Mainz **4** Johannes Gutenberg-Universität Mainz.

DFG Deutsche
Forschungsgemeinschaft



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