

QPTN-9 :: Phase transitions in atomic, molecular and other domains

An Overview of QPT and ESQPT in Molecular Systems

Facultad de Ciencias Experimentales
Centro de Estudios Avanzados en Física, Matemáticas y Computación
Universidad de Huelva

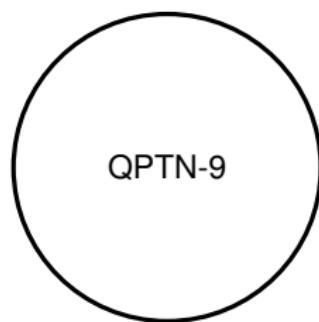
May 27, 2018

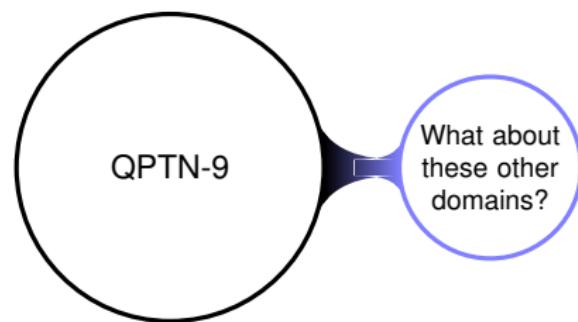


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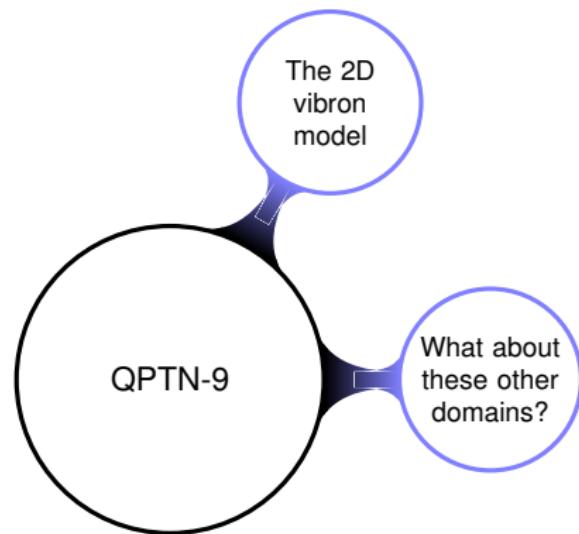


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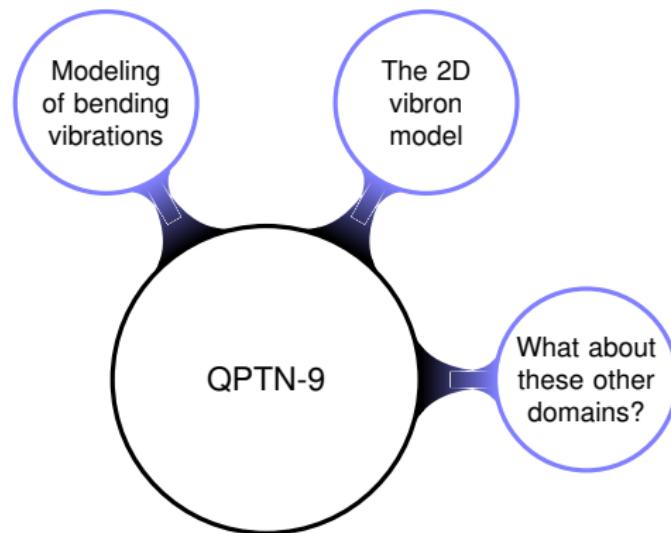




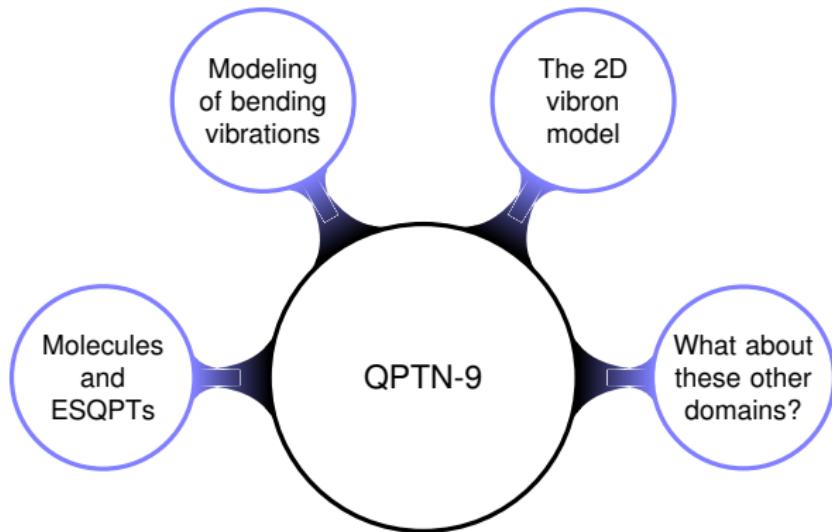
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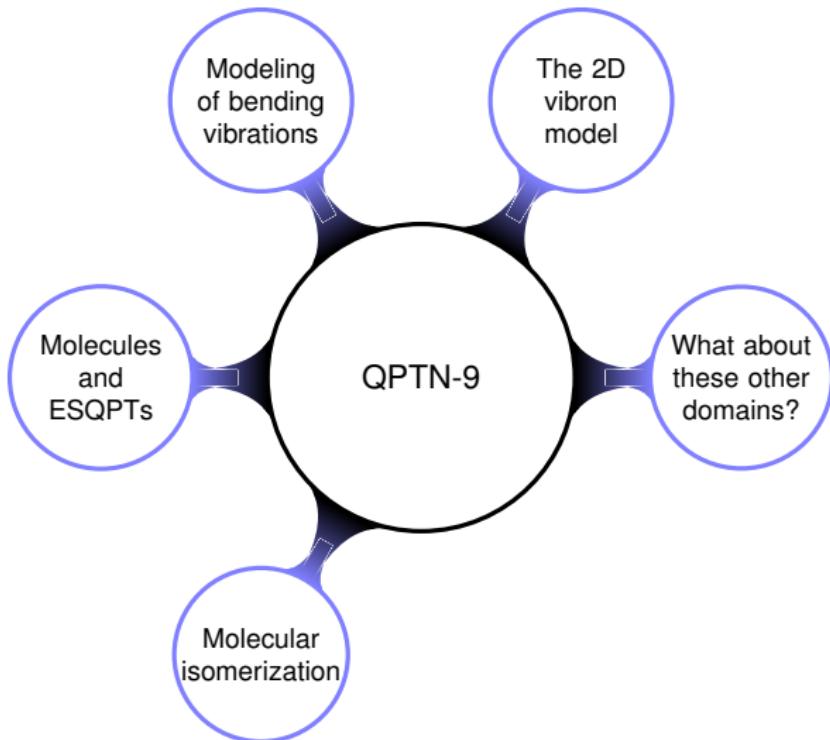
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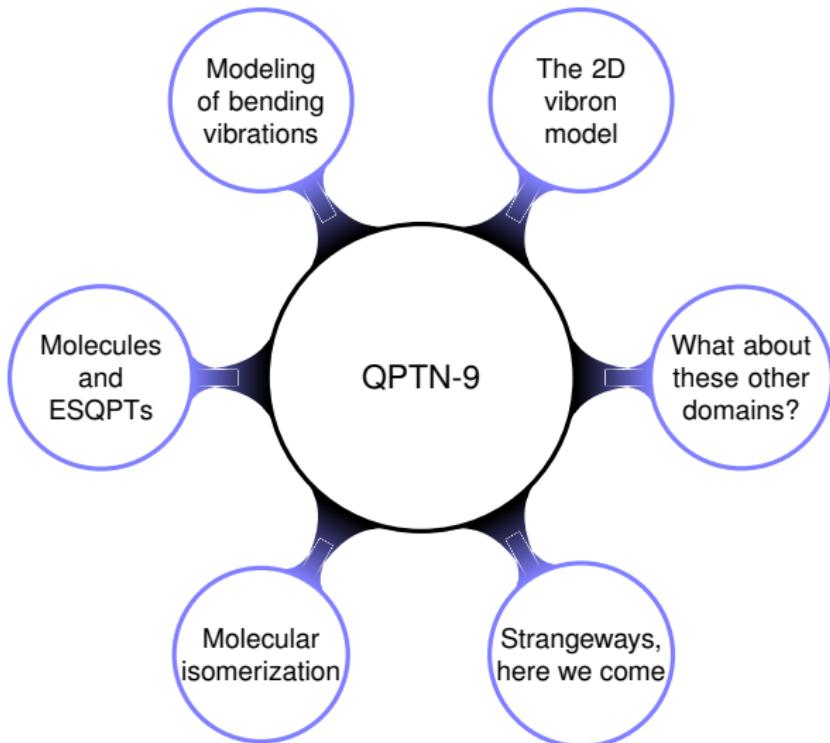
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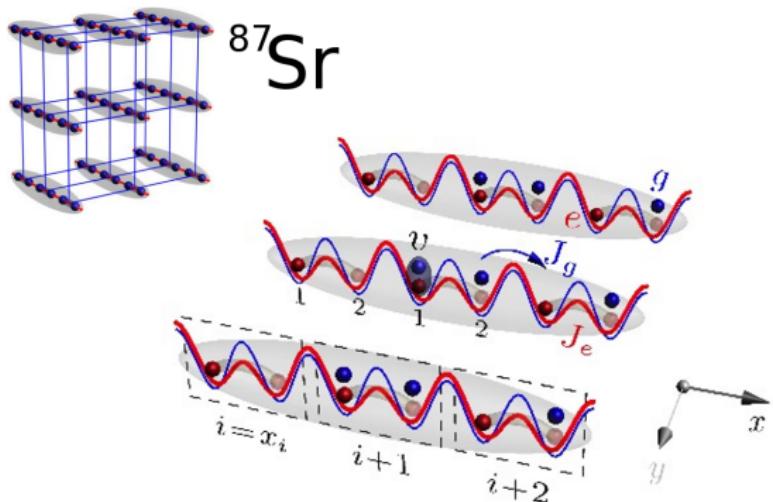


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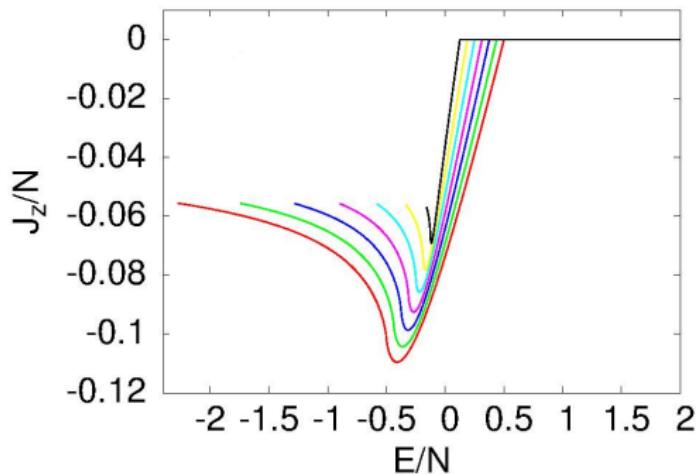
Topological superfluid phase with repulsive fermionic atoms

Gerardo Ortiz. Indiana University (USA)



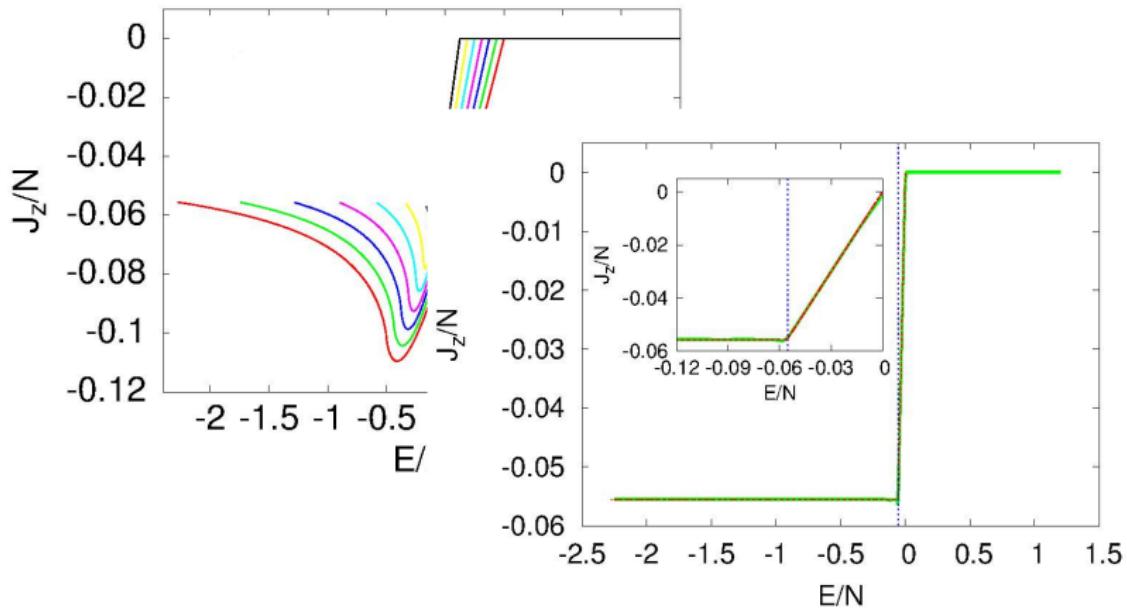
Relationship between the ESPQT and the thermal phase transition in the Dicke model

Pedro Pérez Fernández, Universidad de Sevilla (SPAIN)



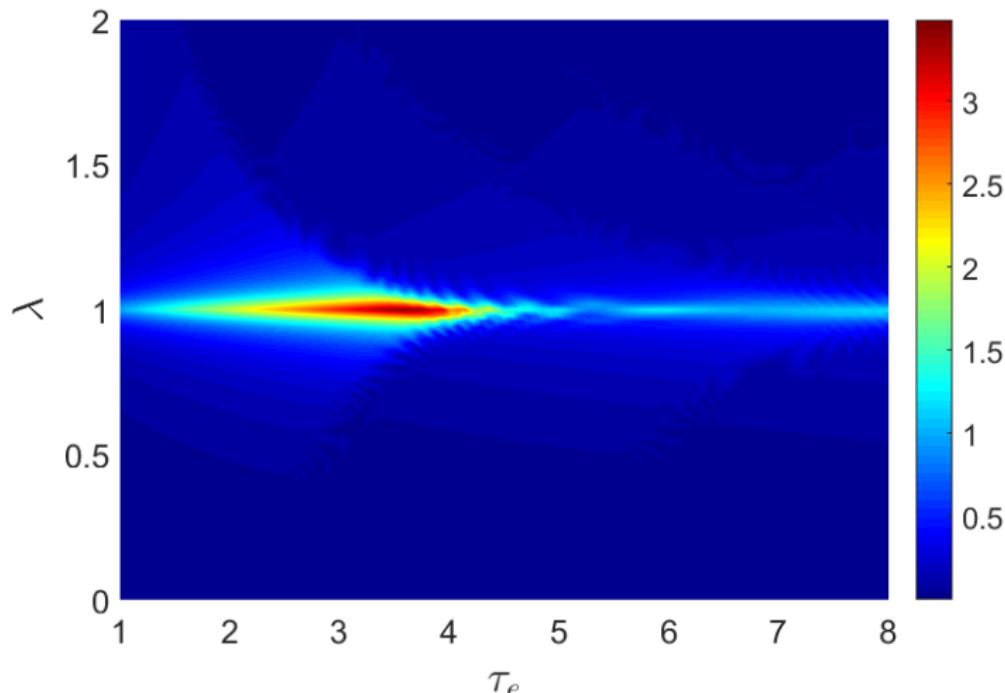
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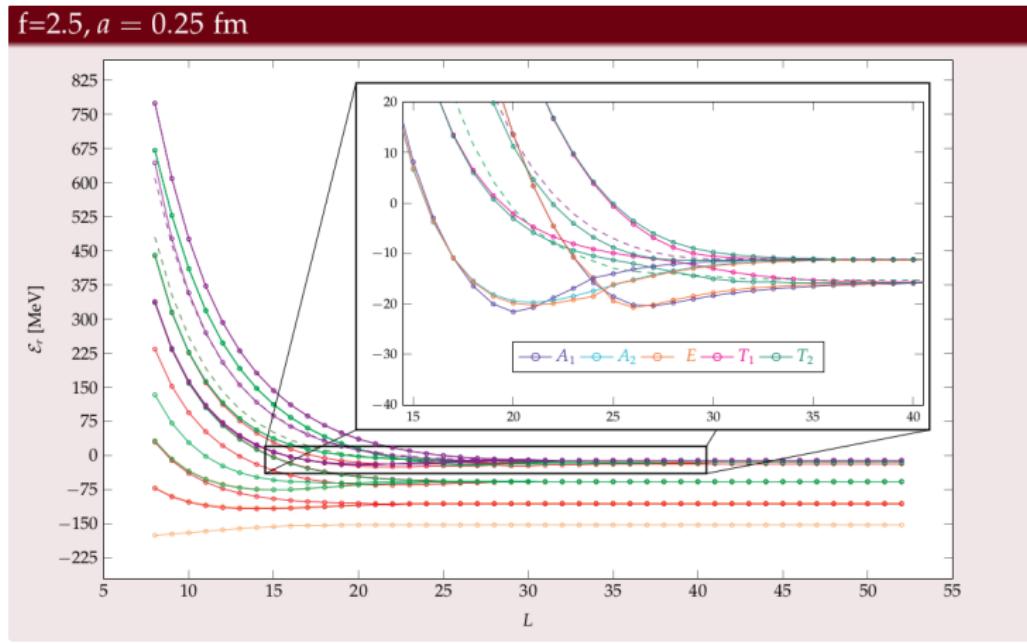
Excited-state quantum phase transition and quantum speed limit

Quin Wang. Zhejiang Normal University (CHINA)



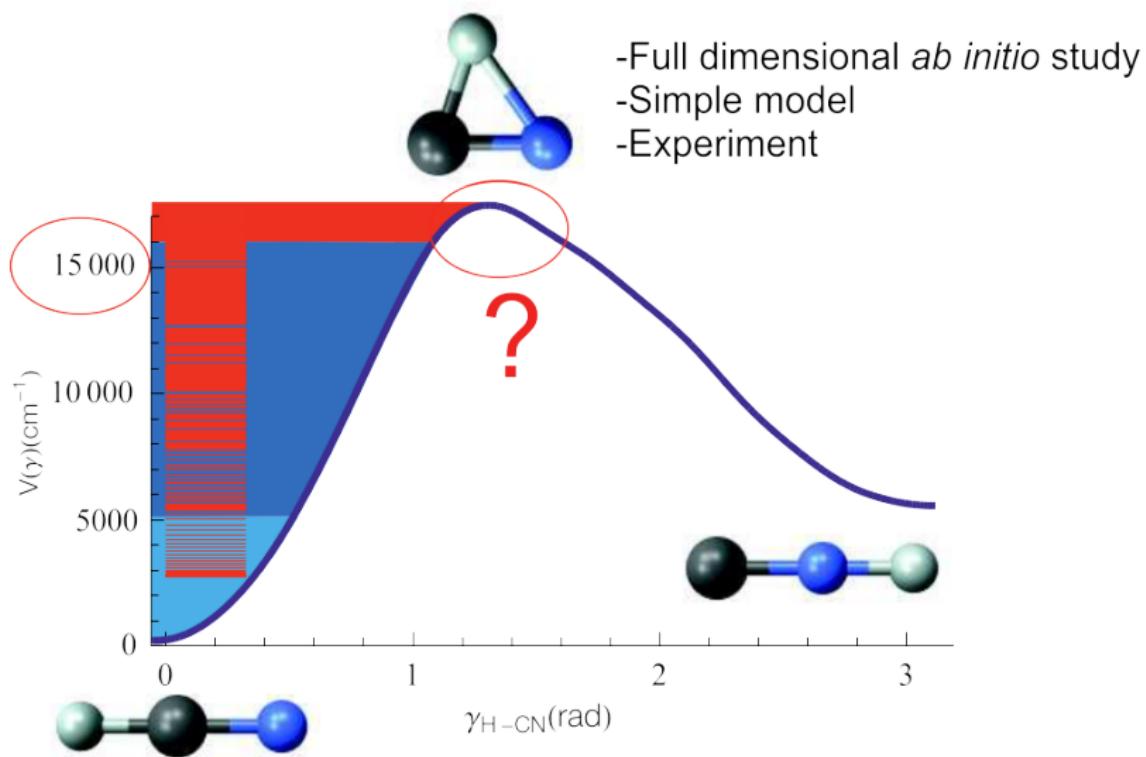
Breaking and restoration of rotational symmetry in the low-energy spectrum of light alpha-conjugate nuclei on the lattice

Gianluca Stellin, University of Bonn (GERMANY)



Saddle point localization of molecular wavefunctions

Georg Ch. Mellau, Justus-Liebig University Giessen (GERMANY)



The Vibron Model

An algebraic approach to molecular structure

F. Iachello, *Contemp. Math.* **160**, 151 (1994)

Study of N -dimensional quantum systems $\Rightarrow U(N + 1)$ SGA



- Dipolar interaction: $N = 3$
- SGA: $U(4)$ Lie algebra
- Vibron Model (VM)
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- 1D ($U(2)$) and 2D ($U(3)$) limits of the Vibron Model

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Algebraic Approach to 2D Systems

The 2D Vibron Model (2DVM)

- The 2D limit of the vibron model is the **simplest two-level model** which still retains a **non-trivial angular momentum** quantum number.
- F. Iachello and S. Oss. J. Chem. Phys. **104**, 6956 (1996).
- It has been successfully applied to the modeling of the bending vibrational dynamics of several molecular species with a single bender or with two coupled benders.
- Recent application to the study of vibrations in atomic lattices (Phys. Rev. B **91**, 214307 (2015)).

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Algebraic Approach to 2D Systems

The 2D Vibron Model Building Blocks

Boson Operators: $\{\tau_i^\dagger, \tau_i, \sigma^\dagger, \sigma\}; i = x, y$

$$[\tau_i, \tau_j^\dagger] = \delta_{i,j} ; \quad i, j = x, y \quad [\sigma, \sigma^\dagger] = 1$$

Circular Bosons:

$$\tau_\pm^\dagger = \mp \frac{\tau_x^\dagger \pm i\tau_y^\dagger}{\sqrt{2}} , \quad \tau_\pm = \mp \frac{\tau_x \mp i\tau_y}{\sqrt{2}}$$

Generators of the $U(3)$ Dynamical Algebra:

$$\{\hat{n}, \hat{n}_s, \hat{\ell}, \hat{Q}_\pm, \hat{R}_\pm, \hat{D}_\pm\}$$

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2DVM Dynamical Symmetries

Branching rules and general Hamiltonian up to 2-body operators.

$$U(3) \supset U(2) \supset SO(2) \quad \text{Dyn. Symmetry (I)}$$

$$N \qquad \qquad n \qquad \qquad \ell$$

$$U(3) \supset SO(3) \supset SO(2) \quad \text{Dyn. Symmetry (II)}$$

$$N \qquad \qquad w \qquad \qquad \ell$$

$$U(2) \quad \hat{C}_1[U(2)] = \hat{n} \quad \hat{C}_2[U(2)] = \hat{n}(\hat{n} + 1)$$

$$SO(3) \quad \hat{C}_2[SO(3)] = \hat{W}^2 = \frac{\hat{D}_+ \hat{D}_- + \hat{D}_- \hat{D}_+}{2} + \hat{\ell}^2$$

$$SO(2) \quad \hat{C}_1[SO(2)] = \hat{\ell} \quad \hat{C}_2[SO(2)] = \hat{\ell}^2$$

General one- and two-body Hamiltonian operator:

$$\hat{H} = \varepsilon \hat{n} + \alpha \hat{n}(\hat{n} + 1) + \beta \hat{\ell}^2 + A \hat{W}^2$$

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Cylindrical Oscillator Dynamical Symmetry

Chain I

$$U(3) \supset U(2) \supset SO(2)$$
$$[N] \qquad n \qquad \ell$$

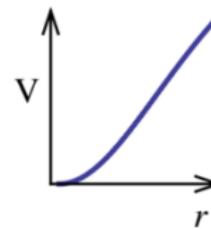
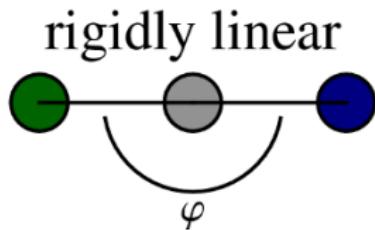
$$n = N, N-1, N-2, \dots, 0$$
$$\ell = \pm n, \pm(n-2), \dots, 1(\text{or } 0)$$

$$\begin{array}{c} \mathbf{n} \\ 3 \end{array} \qquad \begin{array}{c} \mathbf{l} \\ \pm 3 \Phi \\ \pm 1 \Pi \end{array}$$

$$\begin{array}{c} 2 \\ \hline \end{array} \qquad \begin{array}{c} \pm 2 \Delta \\ 0 \Sigma \end{array}$$

$$1 \qquad \pm 1 \Pi$$

$$0 \qquad 0 \Sigma$$



Displaced Oscillator Dynamical Symmetry

Chain II

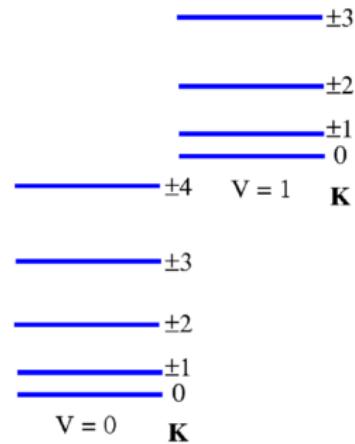
$$U(3) \supset SO(3) \supset SO(2)$$
$$\begin{matrix} N \\ \omega \\ \ell \end{matrix}$$

$$\omega = N, N-2, N-4, \dots, 1(\text{or } 0)$$

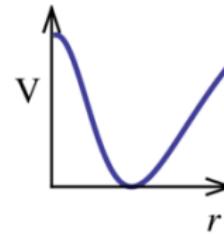
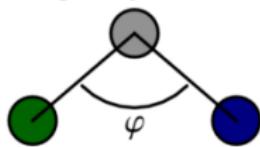
$$\ell = \pm\omega, \pm(\omega-1), \dots, 0$$

$$v = \frac{N-\omega}{2} = 0, 1, \dots, \frac{N-1}{2} (\text{or } \frac{N}{2})$$

$$\ell = 0, \pm 1, \pm 2, \dots, \pm(N-2v)$$



rigidly bent



Shape Phase Transitions: QPTs vs ESQPTs

Ground State Quantum Phase Transitions

Singularities in the evolution of the system's ground state properties (**shape phase transitions**) as a **control parameter** is varied (aka zero-temperature phase transitions).

P. Cejnar and J. Jolie. *Prog. Part. Nucl. Phys.* 62 210 (2009)

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Shape phase transitions strictly take place at the **thermodynamic limit** (large N): importance of **precursors** for mesoscopic systems and the **scaling behavior** of the relevant quantities.

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Shape Phase Transitions: QPTs vs ESQPTs

Excited State Quantum Phase Transitions

Is this behavior extensible to states throughout the excitation spectrum? Yes

ESQPTs are universal to two-level pairing many-body models for both bosonic and fermionic constituents.

M.A. Caprio, P. Cejnar, F. Iachello. *Ann. Phys.* 323 1106 (2008).

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2DVM Phase Diagram

Single Bender Model Hamiltonian

$$\begin{array}{c} U(3) \supset U(2) \supset SO(2) \text{ Dynamical Symmetry (I)} \\ U(3) \supset SO(3) \supset SO(2) \text{ Dynamical Symmetry (II)} \end{array}$$

Single Bender Model Hamiltonian



$$\hat{\mathcal{H}} = \varepsilon \left[(1 - \xi) \hat{n} + \frac{\xi}{N-1} \hat{P} \right]$$

- ε : energy scale
- ξ : control parameter: $\xi \in [0, 1]$
 - $\xi = 0.0$ rigidly-linear
 - $0.0 < \xi \leq 0.2$ quasilinear
 - $0.2 < \xi < 1.0$ non-rigid
 - $\xi = 1.0$ rigidly-bent

The system undergoes a second order QPT in
 $\xi_c = 0.2$.



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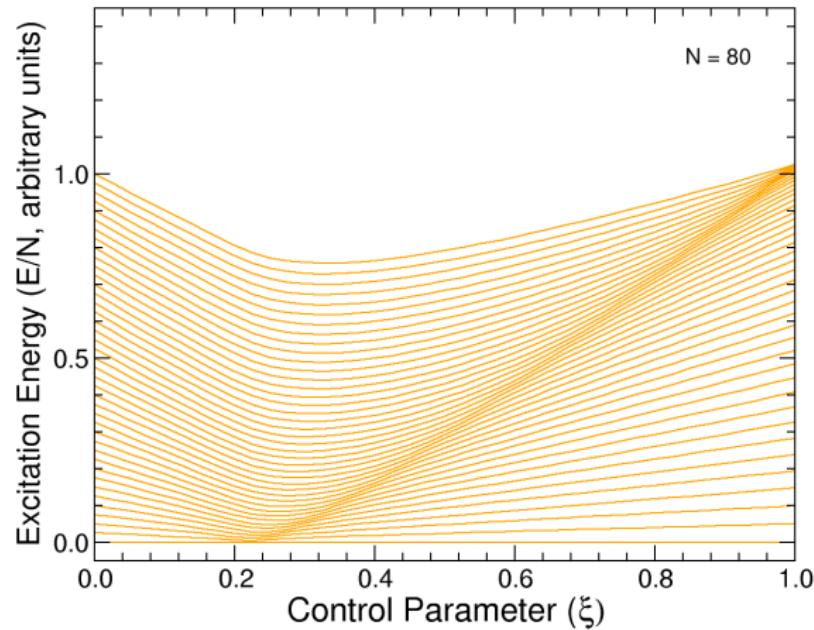


2DVM Excited State Quantum Phase Transition

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Infinite local level density (in the thermodynamic limit) at the critical point propagates to higher excitation energies: SEPARATRIX.

Possibility of crossing the ESQPT both horizontally and vertically.

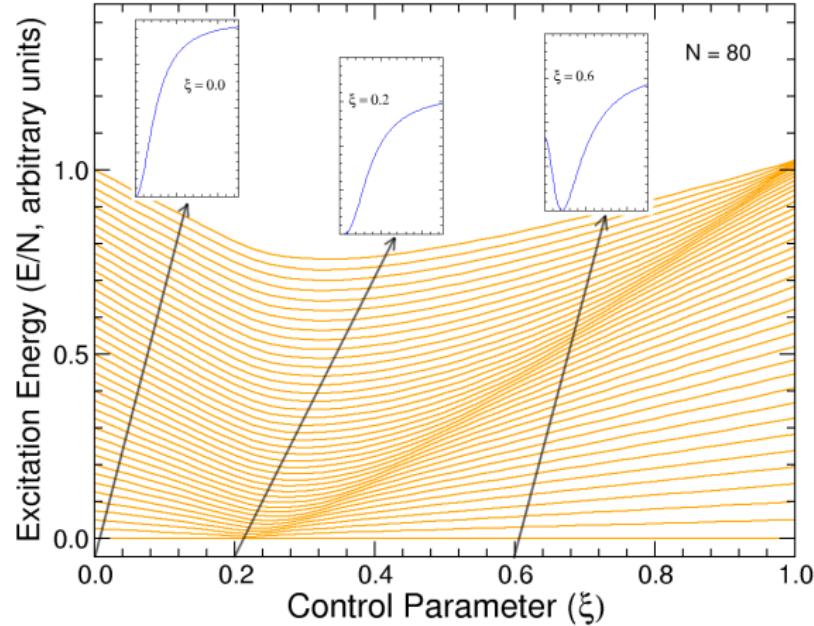


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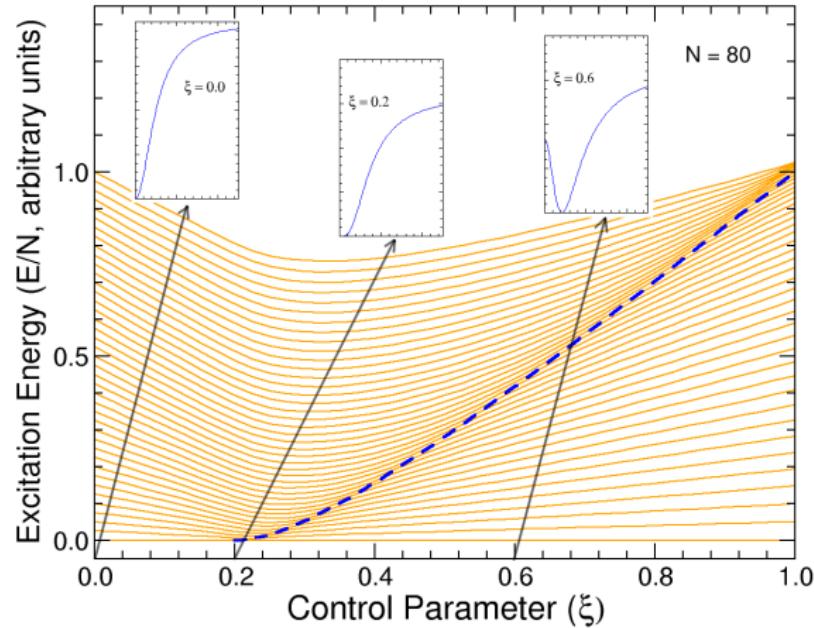


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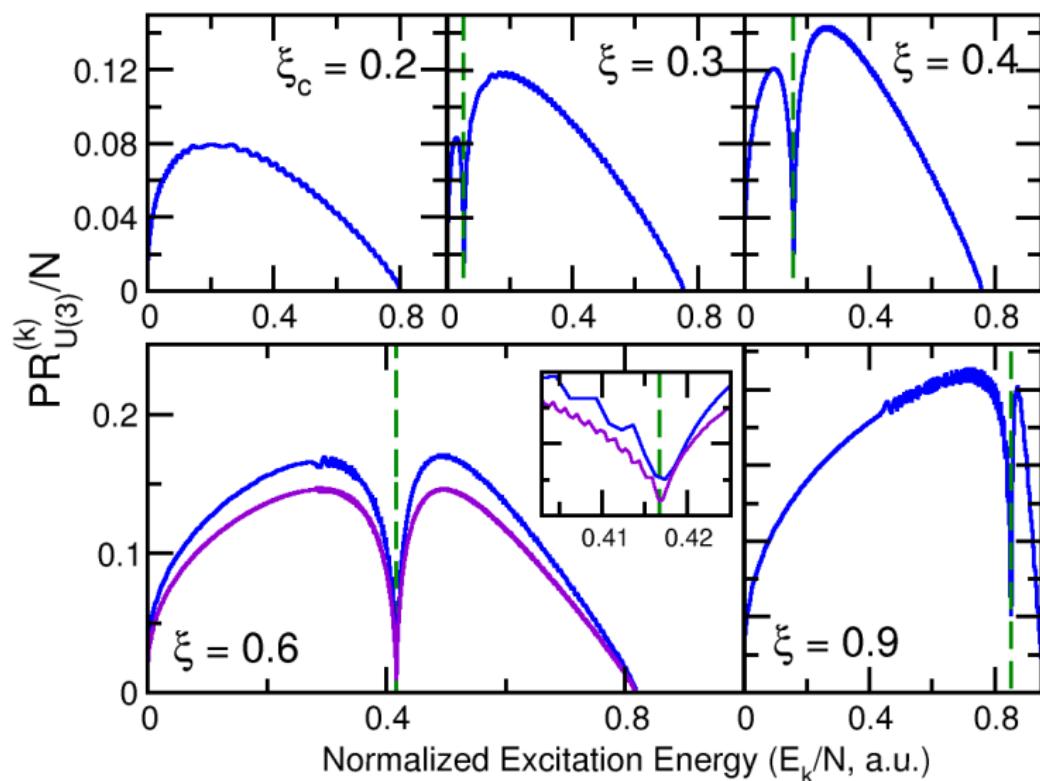
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Eigenstate Structure: Participation Ratio in an ESQPT

Phys. Rev. A **92**, 050101 (2015), Phys. Rev. A **94**, 012113 (2016)

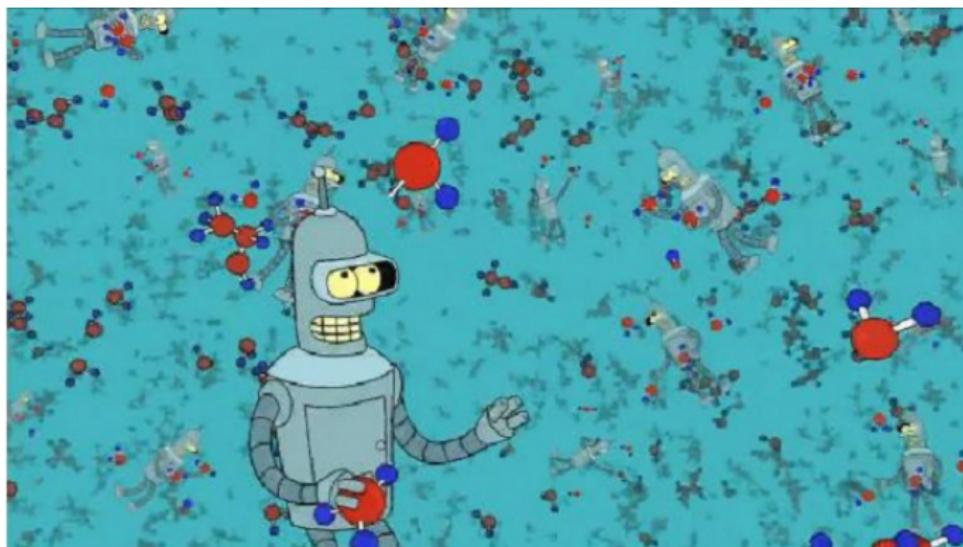
Chain I: Cylindrical oscillator case ($N = 600, 2000$)



Application to Single Bender Molecular Species

Vibrational bending excitation

- D. Larese and F. Iachello. *J. Mol. Struct.* **1006**, 611 (2011).
- D. Larese, FPB, and F. Iachello. *J. Mol. Struct.* **1051**, 310 (2013).



ESQPT/Quantum Monodromy in bending spectra

J. Mol. Struct. **1006**, 611 (2011), J. Mol. Struct. **1051**, 310 (2013)

Table 1

Best-fit values of Hamiltonian parameters. N (the integer vibron number) and ξ (control parameter value) are both dimensionless, while ε , α , β , A , and rms deviation are in cm^{-1} .

| Molecule | N | ε | α | β | A | rms | ξ |
|----------------------|-----|---------------|----------|---------|-----------|------|--------|
| HCNO | 24 | 716.715 | -13.0168 | 10.7685 | 7.4363 | 0.09 | 0.1927 |
| DCNO | 30 | 634.902 | -12.5727 | 8.4958 | 5.2775 | 0.25 | 0.1942 |
| BrCNO | 96 | 642.047 | -6.4154 | 3.1938 | 1.6980 | 0.57 | 0.2010 |
| ClCNO | 92 | 638.079 | -6.4925 | 3.2283 | 1.7766 | 0.71 | 0.2025 |
| CH ₃ NCS | 73 | 523.947 | -6.8652 | 7.8193 | 1.8524 | 0.48 | 0.2029 |
| GeH ₃ NCO | 90 | 385.319 | -4.1902 | 4.6941 | 1.1172 | 0.38 | 0.2069 |
| CH ₃ NCO | 78 | 453.756 | -5.5741 | 7.6712 | 1.6313 | 1.34 | 0.2168 |
| OCCCS | 30 | 77.5202 | -0.13767 | 0.37183 | 0.0013568 | 0.05 | 0.0004 |
| NCCNO | 28 | 149.141 | -1.8472 | 1.3019 | 0.97870 | 0.16 | 0.1507 |
| NCNCS | 70 | 203.572 | -2.5766 | 1.4957 | 0.81344 | 2.18 | 0.2154 |

| Molecule | N | ε | α | β | A | rms | ξ |
|--------------------------------|-----|---------------|----------|---------|-------|-----|--------|
| HCN | 50 | 749.88 | -3.26 | 4.25 | 0.36 | 0.7 | 0.0227 |
| HNC | 40 | 853.65 | -11.56 | 8.94 | 3.77 | 2.3 | 0.1471 |
| NiCN | 40 | 244 | -2.8 | 4 | 0.1 | 5 | 0.0154 |
| CaOH | 20 | 363.5 | -0.05 | 6.6 | 0.5 | 2.3 | 0.0261 |
| CaOD ^a | 27 | 273.7 | -3.0 | 4.6 | 0.1 | 0.5 | 0.0097 |
| MgOH ^b | 20 | 397.4 | -4.5 | | 4.7 | 2.2 | 0.1833 |
| MgOD ^{a,b} | 27 | 299.4 | -2.1 | | 2.7 | 4.2 | 0.1886 |
| OCS ^b | 40 | 728.3 | -5.1 | 1.4 | 2.2 | 0.2 | 0.1062 |
| HNCS | 25 | 810.1 | -22.96 | 16.97 | 10.77 | 3.3 | 0.2419 |
| H ₂ O ^c | 143 | 4815.0 | -32.15 | 15.44 | 8.7 | 9.1 | 0.2042 |
| D ₂ O ^c | 195 | 4163.0 | -20.70 | 8.48 | 5.5 | 4.0 | 0.2031 |
| H ₂ S ^d | 140 | | | 12.10 | 2.12 | 5.9 | 1.0000 |
| H ₂ Se ^d | 170 | | | 9.29 | 1.53 | 5.9 | 1.0000 |

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Table 1

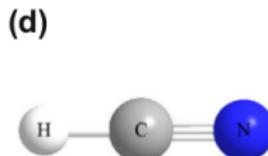
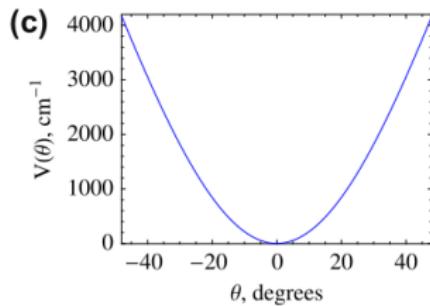
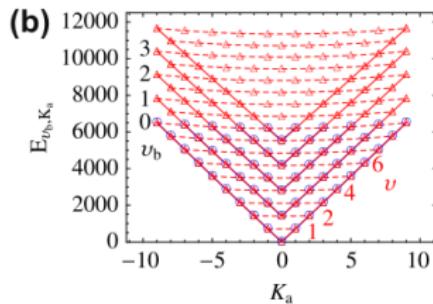
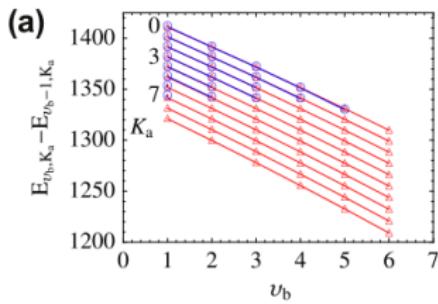
Best-fit values of Hamiltonian parameters. N (the integer vibron number) and ξ (control parameter value) are both dimensionless, while ε , α , β , A , and rms deviation are in cm^{-1} .

| Molecule | N | ε | α | β | A | rms | ξ |
|----------------------|-----|---------------|----------|---------|-----------|------|--------|
| HCNO | 24 | 716.715 | -13.0168 | 10.7685 | 7.4363 | 0.09 | 0.1927 |
| DCNO | 30 | 634.902 | -12.5727 | 8.4958 | 5.2775 | 0.25 | 0.1942 |
| BrCNO | 96 | 642.047 | -6.4154 | 3.1938 | 1.6980 | 0.57 | 0.2010 |
| ClCNO | 92 | 638.079 | -6.4925 | 3.2283 | 1.7766 | 0.71 | 0.2025 |
| CH ₃ NCS | 73 | 523.947 | -6.8652 | 7.8193 | 1.8524 | 0.48 | 0.2029 |
| GeH ₃ NCO | 90 | 385.319 | -4.1902 | 4.6941 | 1.1172 | 0.38 | 0.2069 |
| CH ₃ NCO | 78 | 453.756 | -5.5741 | 7.6712 | 1.6313 | 1.34 | 0.2168 |
| OCCCS | 30 | 77.5202 | -0.13767 | 0.37183 | 0.0013568 | 0.05 | 0.0004 |
| NCCNO | 28 | 149.141 | -1.8472 | 1.3019 | 0.97870 | 0.16 | 0.1507 |
| NCNCS | 70 | 203.572 | -2.5766 | 1.4957 | 0.81344 | 2.18 | 0.2154 |

| Molecule | N | ε | α | β | A | rms | ξ |
|--------------------------------|-----|---------------|----------|---------|-------|-----|--------|
| HCN | 50 | 749.88 | -3.26 | 4.25 | 0.36 | 0.7 | 0.0227 |
| HNC | 40 | 853.65 | -11.56 | 8.94 | 3.77 | 2.3 | 0.1471 |
| NiCN | 40 | 244 | -2.8 | 4 | 0.1 | 5 | 0.0154 |
| CaOH | 20 | 363.5 | -0.05 | 6.6 | 0.5 | 2.3 | 0.0261 |
| CaOD ^a | 27 | 273.7 | -3.0 | 4.6 | 0.1 | 0.5 | 0.0097 |
| MgOH ^b | 20 | 397.4 | -4.5 | | 4.7 | 2.2 | 0.1833 |
| MgOD ^{a,b} | 27 | 299.4 | -2.1 | | 2.7 | 4.2 | 0.1886 |
| OCS ^b | 40 | 728.3 | -5.1 | 1.4 | 2.2 | 0.2 | 0.1062 |
| HNCS | 25 | 810.1 | -22.96 | 16.97 | 10.77 | 3.3 | 0.2419 |
| H ₂ O ^c | 143 | 4815.0 | -32.15 | 15.44 | 8.7 | 9.1 | 0.2042 |
| D ₂ O ^c | 195 | 4163.0 | -20.70 | 8.48 | 5.5 | 4.0 | 0.2031 |
| H ₂ S ^d | 140 | | | 12.10 | 2.12 | 5.9 | 1.0000 |
| H ₂ Se ^d | 170 | | | 9.29 | 1.53 | 5.9 | 1.0000 |

Dynamical Symmetry (I): HCN

J. Mol. Struct. 1051, 310 (2013)



- (a) Birge-Sponer Plot
(b) Monodromy Plot
(c) Bending Potential
(d) Molecule Model

$$N = 40$$

$$\varepsilon = 794.88$$

$$\alpha = -3.96$$

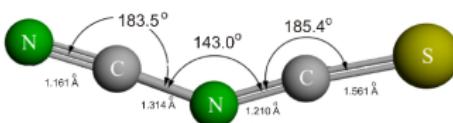
$$\beta = 4.25$$

$$A = 0.36$$

$$rms = 0.7 \text{ cm}^{-1}$$

2DVM and the NCNCS bending spectrum

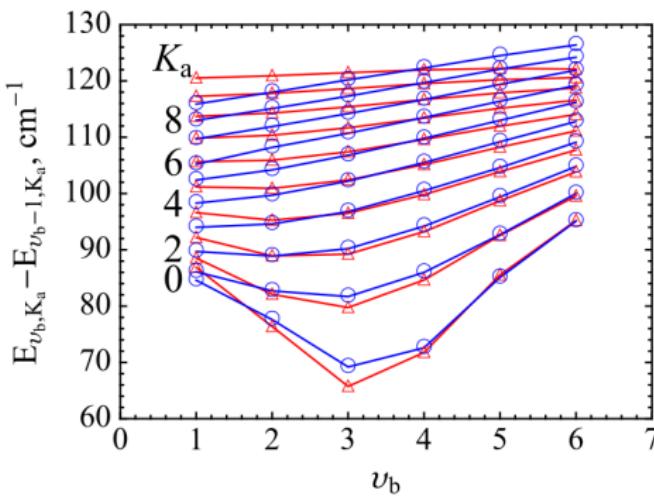
ESQPT in ν_7 normal mode of Cyanogen Isothiocyanate



General Hamiltonian

$$\hat{H} = \varepsilon \hat{n} + \alpha \hat{n}(\hat{n} + 1) + \beta \hat{\ell}^2 + A \hat{P}$$

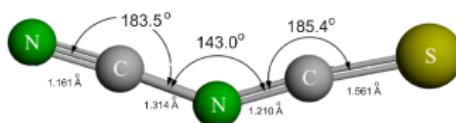
| | |
|-----------------------------|-----------------------------------|
| N | $r_{\text{rms}} (\text{cm}^{-1})$ |
| 50 | 2.2 |
| $\epsilon (\text{cm}^{-1})$ | $\alpha (\text{cm}^{-1})$ |
| 203.57 | -2.58 |
| $\beta (\text{cm}^{-1})$ | $A (\text{cm}^{-1})$ |
| 1.50 | 0.8134 |



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2DVM and the NCNCS bending spectrum

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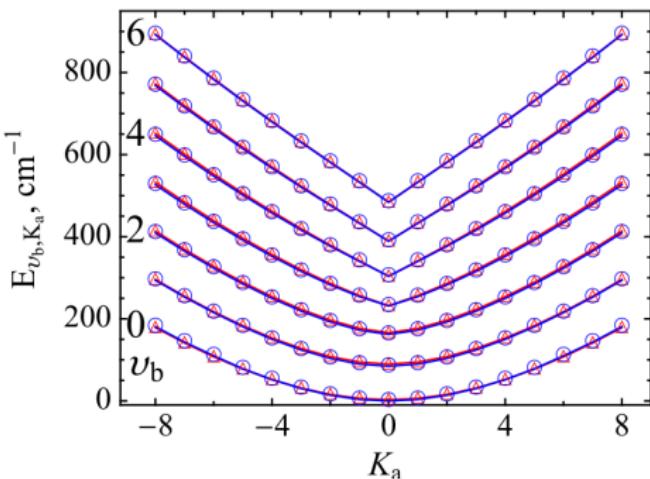


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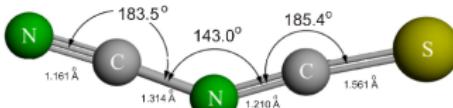
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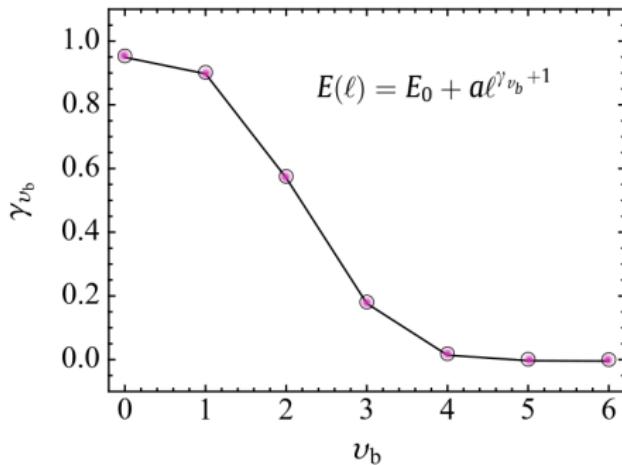


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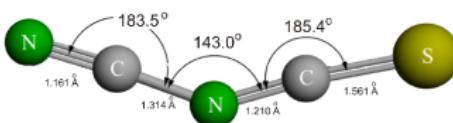
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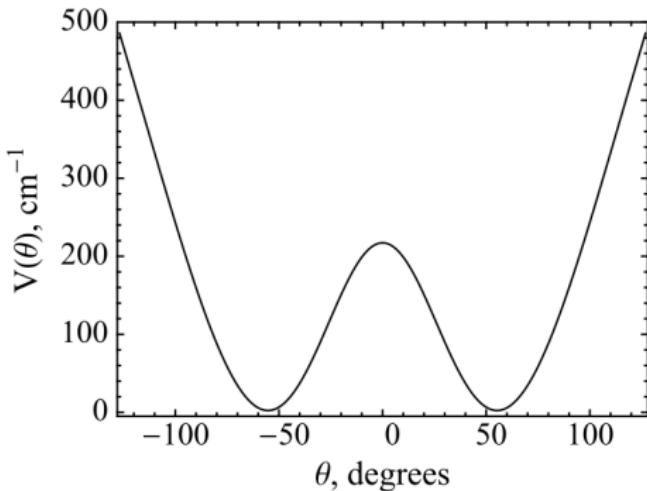


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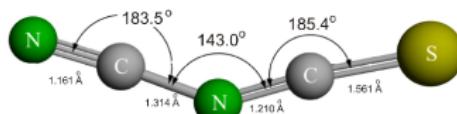
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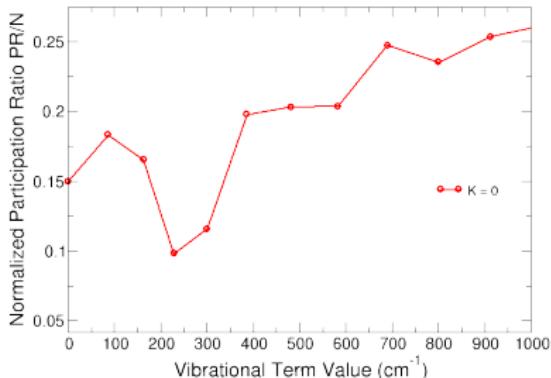


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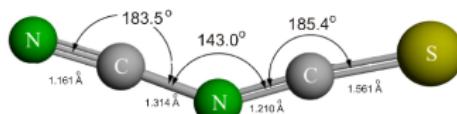
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J. Mol. Struct. **1006**, 611 (2011)



2DVM and the NCNCS bending spectrum

ESQPT in ν_7 normal mode of Cyanogen Isothiocyanate

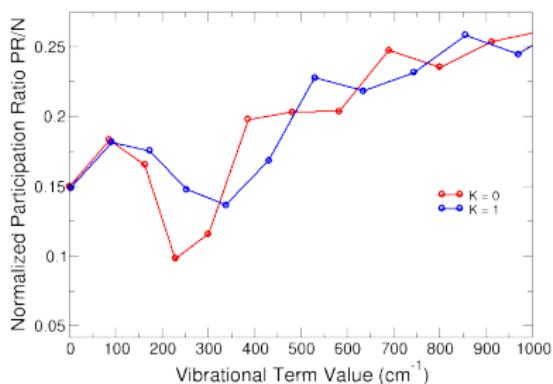


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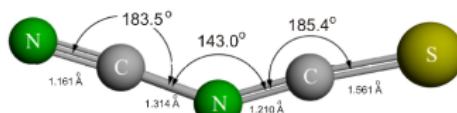
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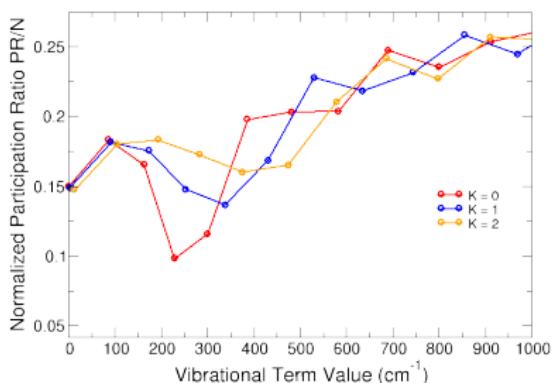


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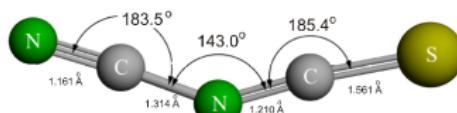
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J. Mol. Struct. 1006, 611 (2011)



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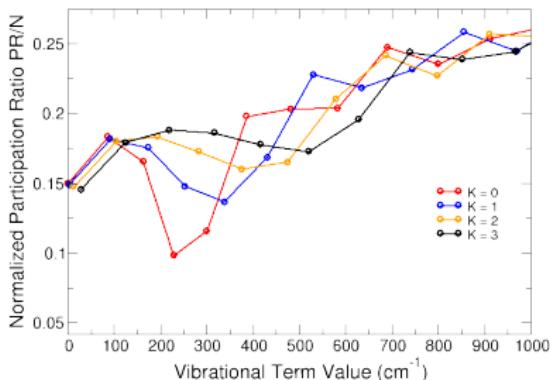


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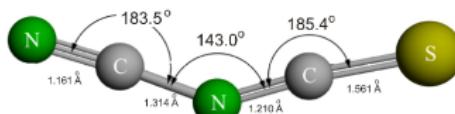
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J. Mol. Struct. **1006**, 611 (2011)



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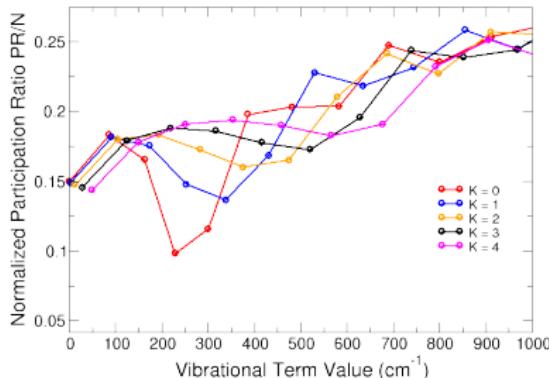


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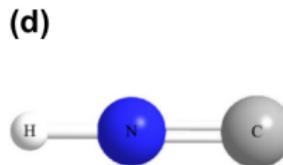
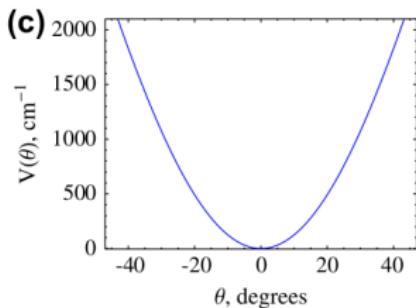
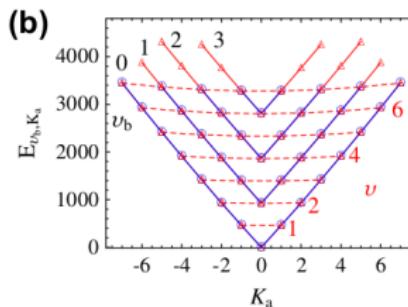
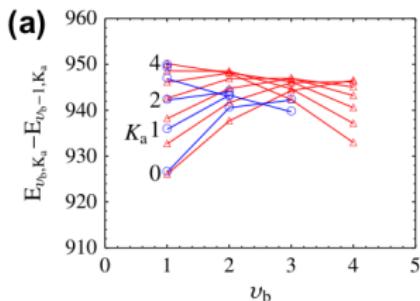
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J. Mol. Struct. **1006**, 611 (2011)



And... What's with HNC?

Figures from J. Mol. Struct. 1051, 310 (2013)



- (a) Birge-Sponer Plot
(b) Monodromy Plot
(c) Bending Potential
(d) Molecule Model

$$N = 40$$

$$\varepsilon = 853.65$$

$$\alpha = -11.56$$

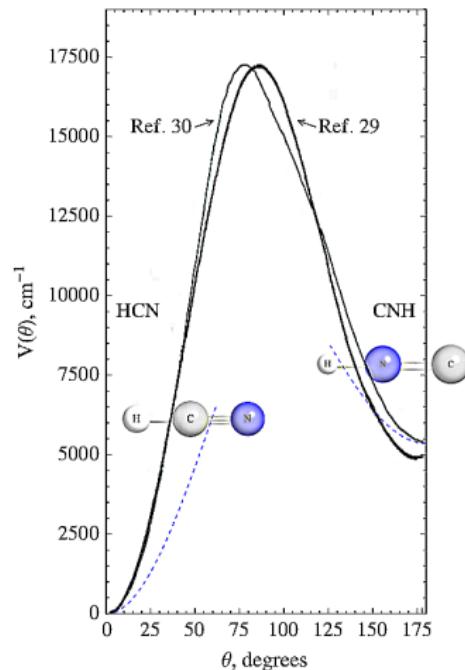
$$\beta = 8.94$$

$$A = 3.77$$

$$rms = 2.3 \text{ cm}^{-1}$$

And... What's with HNC?

Figure and excerpt from J. Mol. Struct. 1051, 310 (2013)



... Of particular interest is the comparison between hydrogen cyanide, HCN, and isocyanide, HNC. In a more complete description HCN and HNC are part of one potential surface. The qualitative and quantitative picture is well described in papers by Ross and Bunker [29] and Mellau, Winnewisser, and Winnewisser [30] ...

Refs. J. Mol. Spectrosc. 101, 199 (1983) and J. Mol. Spectrosc. 249, 23 (2008).

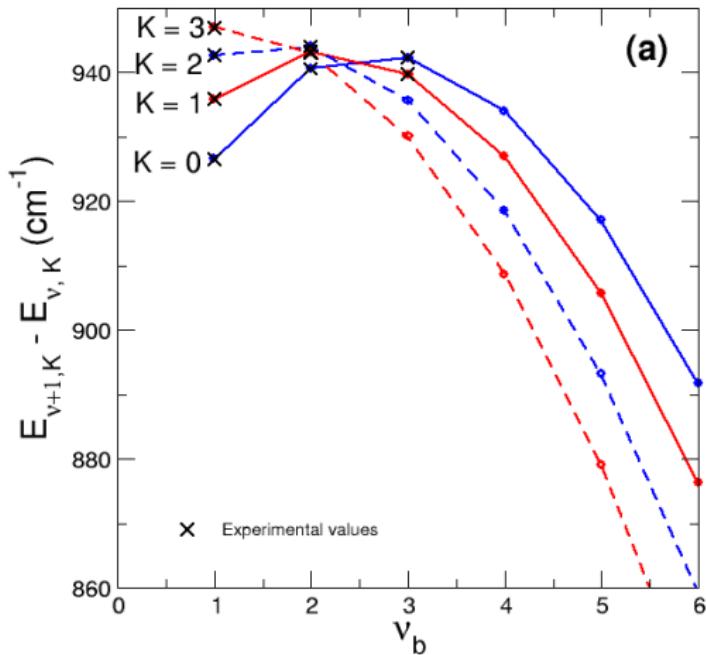
General Hamiltonian up to Four-Body Operators

Reanalysis of bending vibrational data in progress...

$$\begin{aligned}\hat{H} = & P_{11} \hat{n} \\ & + P_{21} \hat{n}^2 + P_{22} \hat{\ell}^2 + P_{23} \hat{W}^2 \\ & + P_{31} \hat{n}^3 + P_{32} \hat{n} \hat{\ell}^2 + P_{33} [\hat{n} \hat{W}^2 + \hat{W}^2 \hat{n}] \\ & + P_{41} \hat{n}^4 + P_{42} \hat{n}^2 \hat{\ell}^2 + P_{43} \hat{\ell}^4 + P_{44} \hat{\ell}^2 \hat{W}^2 \\ & + P_{45} [\hat{n}^2 \hat{W}^2 + \hat{W}^2 \hat{n}^2] + P_{46} \hat{W}^4 + P_{47} [\hat{W}^2 \hat{W}^2 + \hat{W}^2 \hat{W}^2] / 2\end{aligned}$$

HNC experimental data reanalysis

Warning: preliminary results!



- (a) Birge-Sponer Plot
- (b) Monodromy Plot
- (c) Effective Frequency
- (d) Participation Ratio

$N = 40$

$$P_{11} = 1414(6)$$

$$P_{21} = -29.83(19)$$

$$P_{22} = 15.81(9)$$

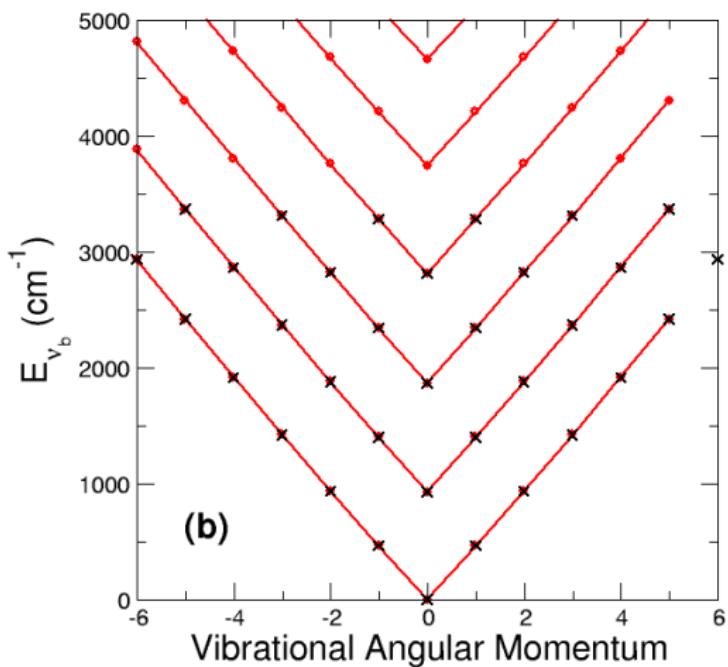
$$P_{23} = -8.05(5)$$

$$P_{32} = 0.0487(14)$$

$$rms = 7 \times 10^{-2} \text{ cm}^{-1}$$

HNC experimental data reanalysis

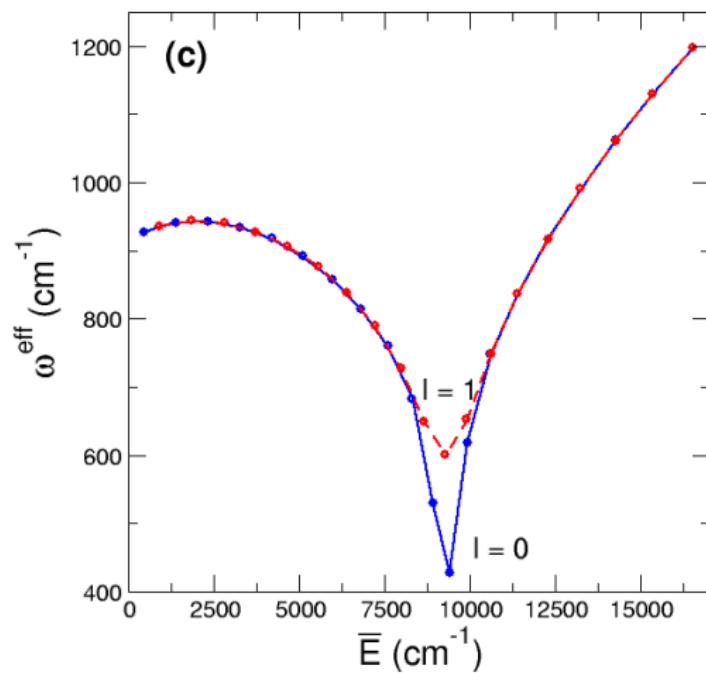
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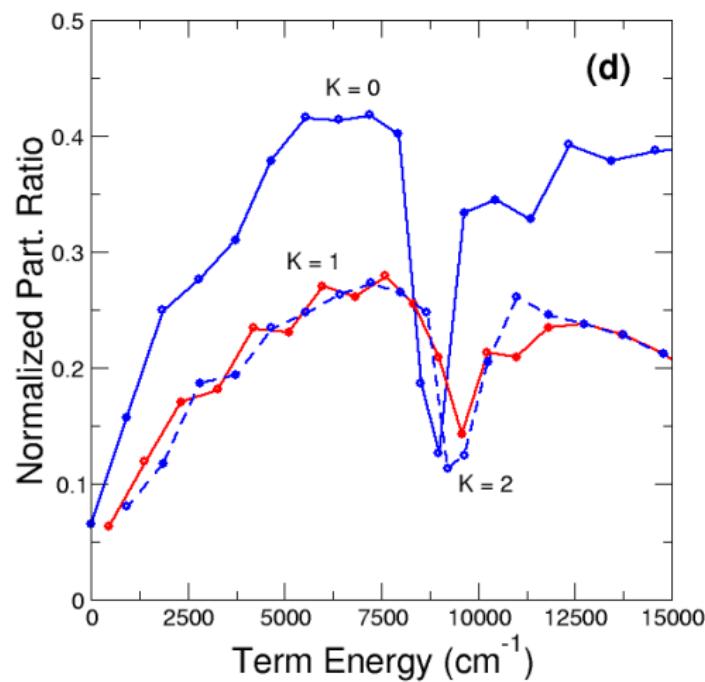
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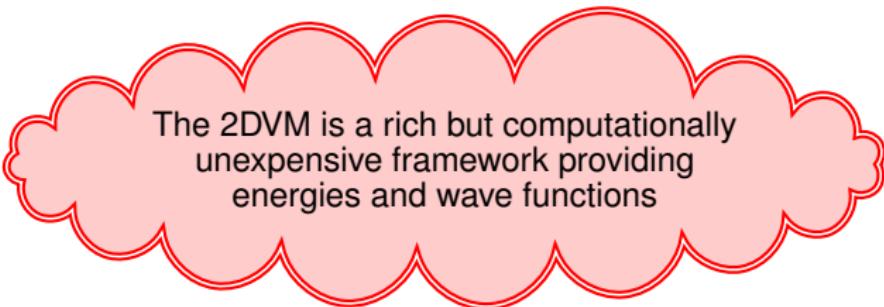
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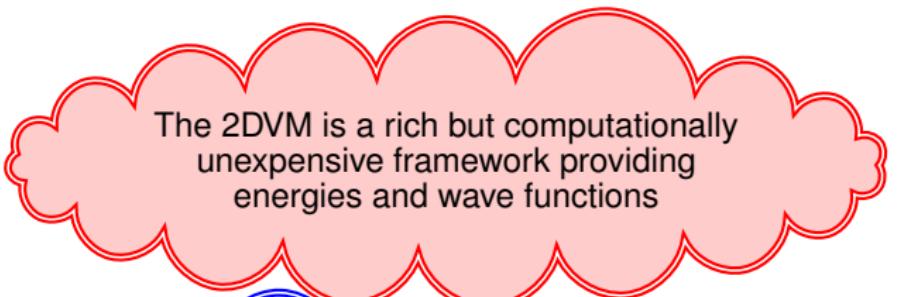
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Where are we heading now?...

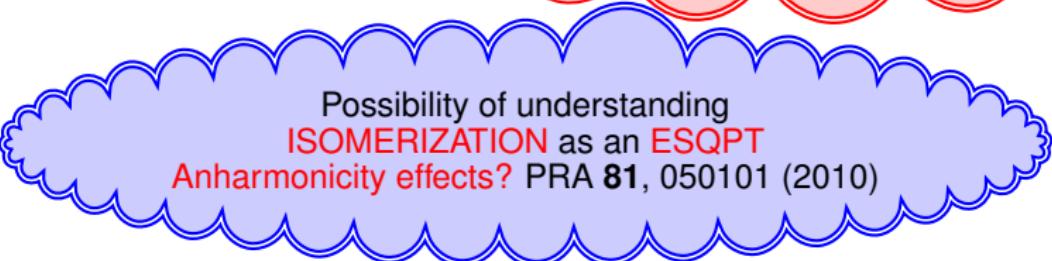


The 2DVM is a rich but computationally unexpensive framework providing energies and wave functions

Where are we heading now?...



The 2DVM is a rich but computationally unexpensive framework providing energies and wave functions



Possibility of understanding
ISOMERIZATION as an **ESQPT**
Anharmonicity effects? PRA **81**, 050101 (2010)

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ISOMERIZATION as an **ESQPT**
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Possibility of studying dynamics of the transition state

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ISOMERIZATION as an **ESQPT**
Anharmonicity effects? PRA **81**, 050101 (2010)

Possibility of studying dynamics of the transition state

Many cool questions still open.
E.g. application to coupled systems (C_2H_2) and relationship with shape coexistence formalism

Collaborators and Suggested References

Work done in collaboration with:
Miguel Carvajal-Zaera (UHU, ES)
Jamil Khalouf (UHU, ES)
Lea Santos (YU, USA)
Franco Iachello (Yale, USA)

2D Vibron Model
Phys. Rev. A **77**, 032115 (2008)
Phys. Rev. A **81**, 050101(R) (2010)
J. Chem. Phys. **140**, 014304 (2014)

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J. Mol. Struct. **1006**, 611 (2011)
J. Mol. Struct. **1051**, 310 (2013)

Molecular Isomerization
Science **350**, 1338 (2015)
Sci. Rep. **6**, 33068 (2016)

Thanks for your attention and
let's roll!