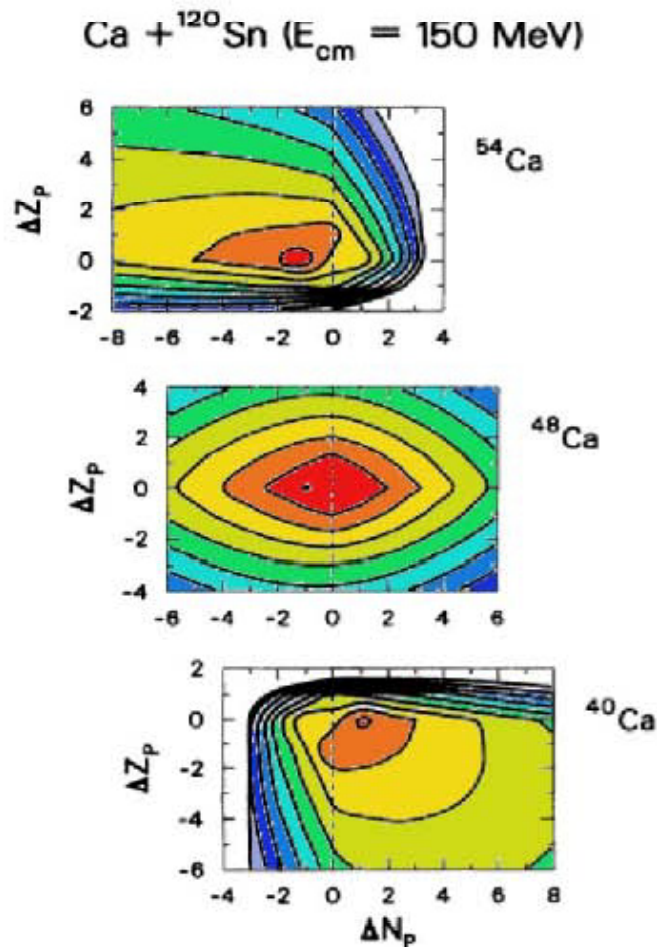


Pair Transfer

Heavy ion transfer reactions



C.H.Dasso, G. Pollaro, A. Winther
Phys. Rev. Lett. 73, 1907 (1994)

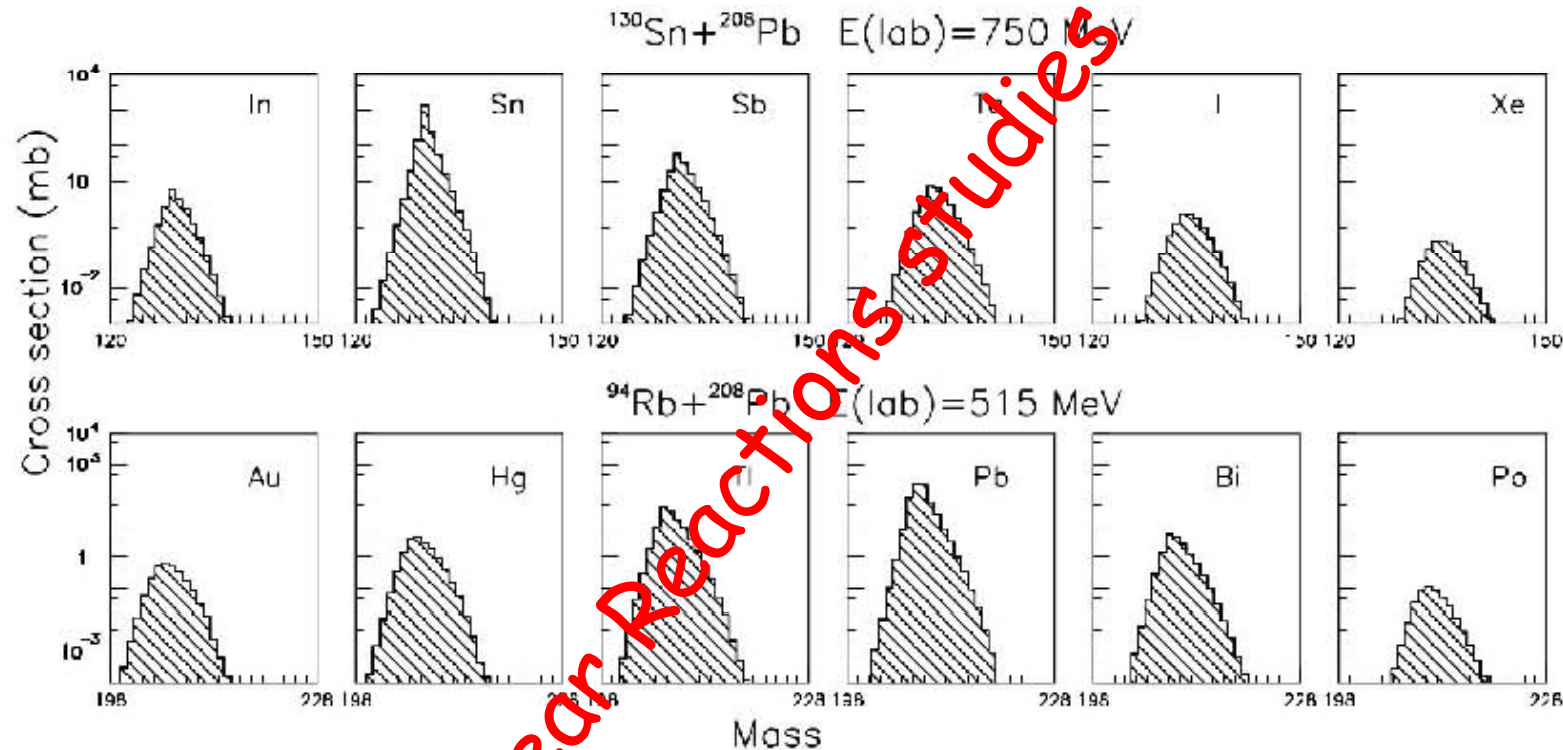
with neutron rich beams

- neutron stripping and proton pick-up
- **Heavy** partner to the **(neutron-rich side)** of the stability valley

with stable beams

- neutron pick-up and proton stripping
- **Heavy** partner to the **“left”** of the stability valley

Neutron-rich heavy nuclei explored via multinucleon transfers: Uni. Zagreb, IRES Strasbourg

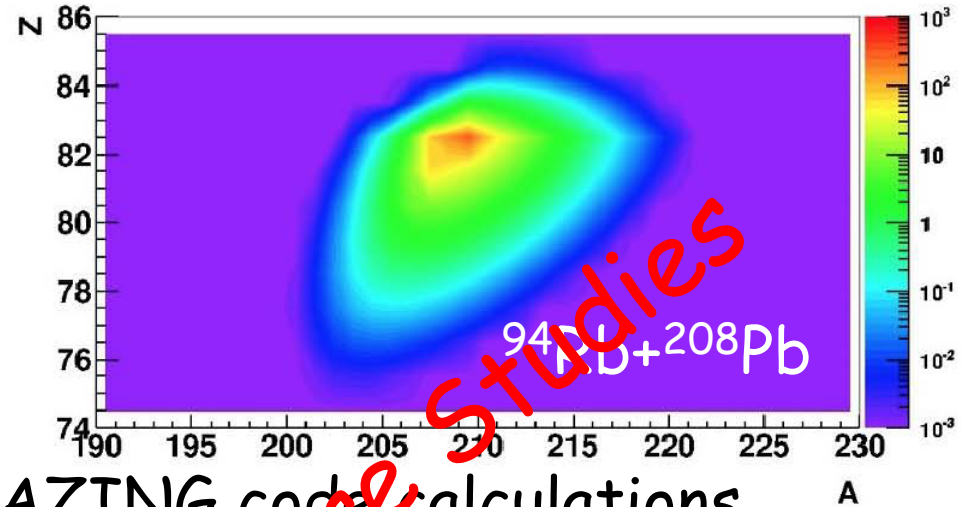
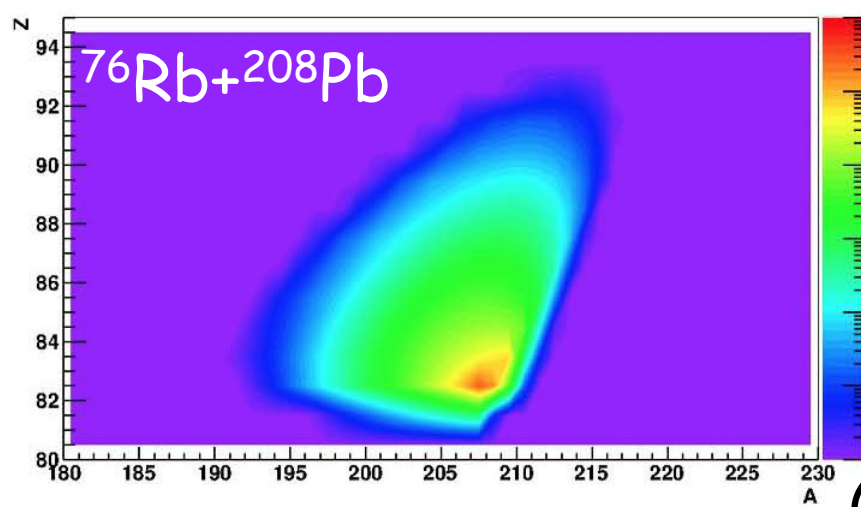


Calculation performed with GRAZING code: cross section

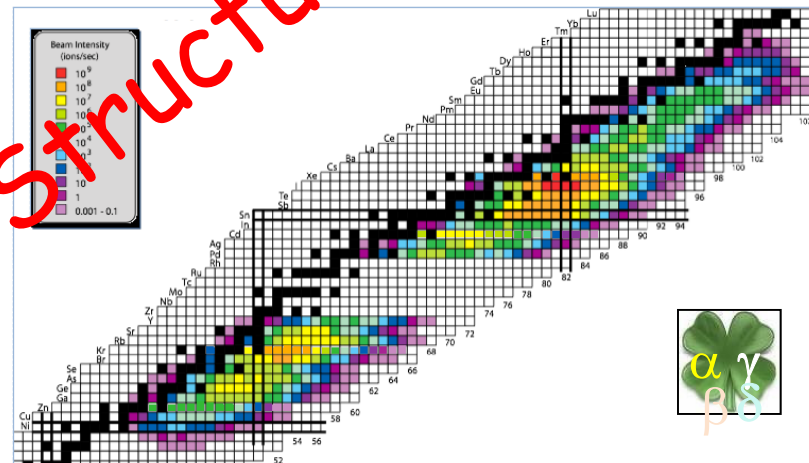
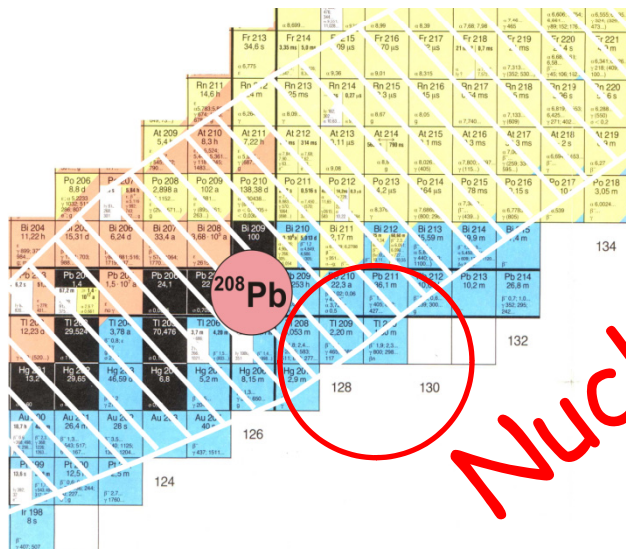
projectile-like reaction products in $^{130}\text{Sn} + ^{208}\text{Pb}$

target-like reaction products in $^{94}\text{Rb} + ^{208}\text{Pb}$

Deep Inelastic and Multinucleon Transfer Reactions with RIBS

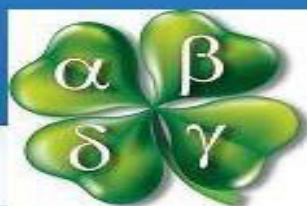


GRAZING code calculations



Nuclear Structure Studies

Super Heavy Elements



Second SPES International Workshop

26-28 May 2014 *INFN Laboratori Nazionali di Legnaro*
Europe/Rome timezone

The search of double magic superheavy nucleus in the region of neutron shell at $N=184$

We propose to perform the measurements of the cross sections for the production of new neutron rich nuclei located along the closed neutron shell $N=184$ using the beam of ^{140}Xe providing by SPES facility in collisions with ^{248}Cm target by the two arm detection system PRISMA or PRISMA+CORSET.

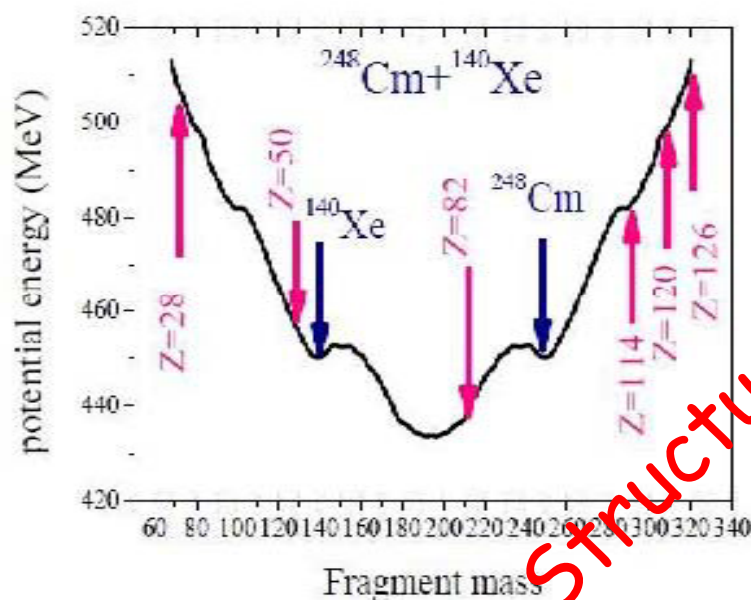


Figure 1. Potential energy at scission point as a function of the primary fragment mass in the reaction $^{248}\text{Cm} + ^{140}\text{Xe}$. The arrows indicate the positions of proton shells.

LOI LNL, Dubna (Ru)

Nuclear Astrophysics

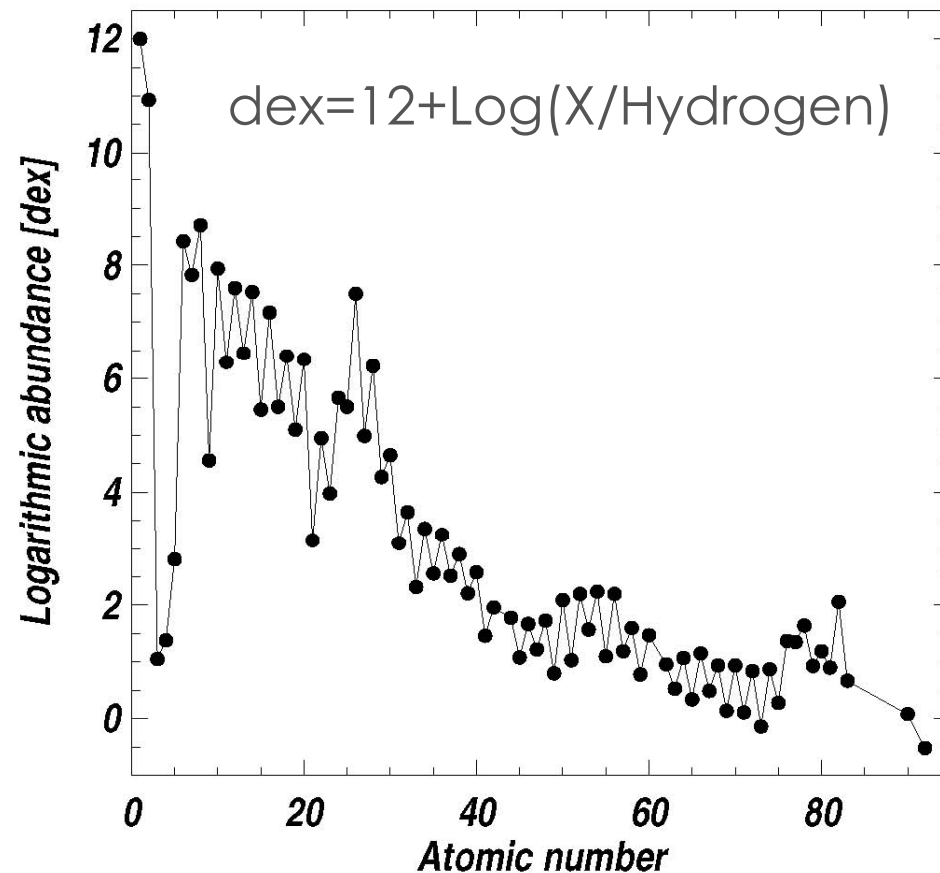
One of the challenges: Origin of the elemental abundances in the solar system

Stars are mostly made of hydrogen and helium, but each has a fairly unique pattern of other elements

The abundance of elements tells us about the history of events prior to the formation of our Sun

The plot shows the composition of the Sun Photosphere

How are these elements created prior to the formation of the Sun?



Asplund M. et al.
Ann. Rev. Astr. 47, 481 (2009)

n-rich region of Pb and Ni

- β-decay half-lives
- neutron emission probabilities

different in **DEFORMED** and **SPHERICAL** nuclei

$$\tau_{\beta \text{ SPH}} \sim 10 \times \tau_{\beta \text{ DEF}}$$

$$P_{n \text{ SPH}} \sim 0.5 \times P_{n \text{ DEF}}$$

- Large uncertainty in r-process location
- difficulty to extrapolate to more exotic regions

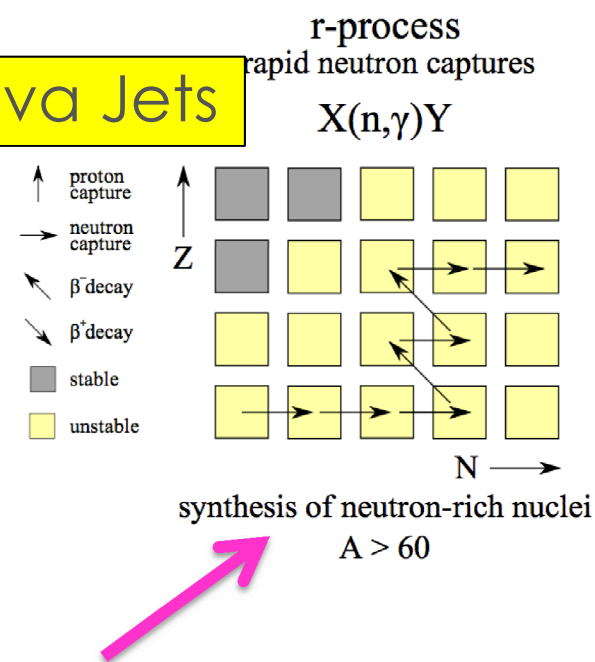
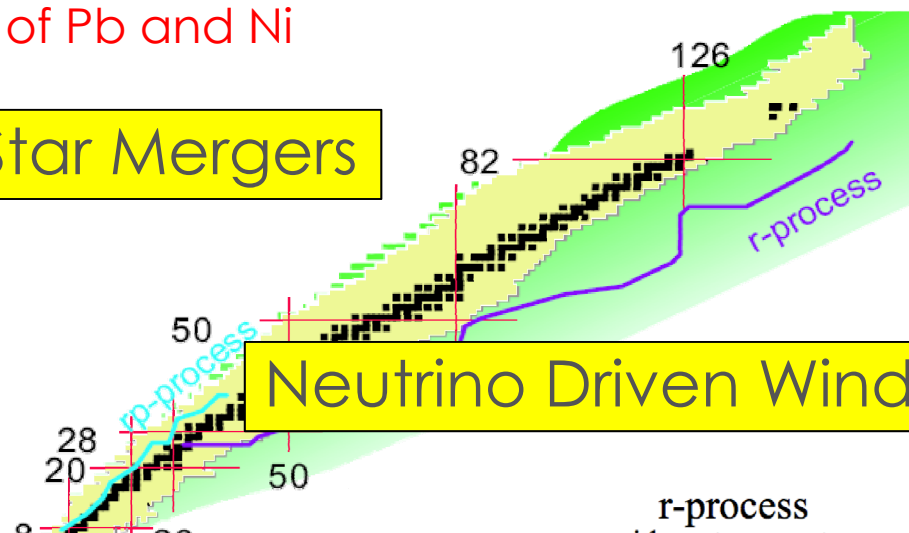
*Importance of Nuclear Structure studies
in the vicinity/towards r-process path:*

- ... first excited states
- ... Isomeric states
- ... β decay lifetimes

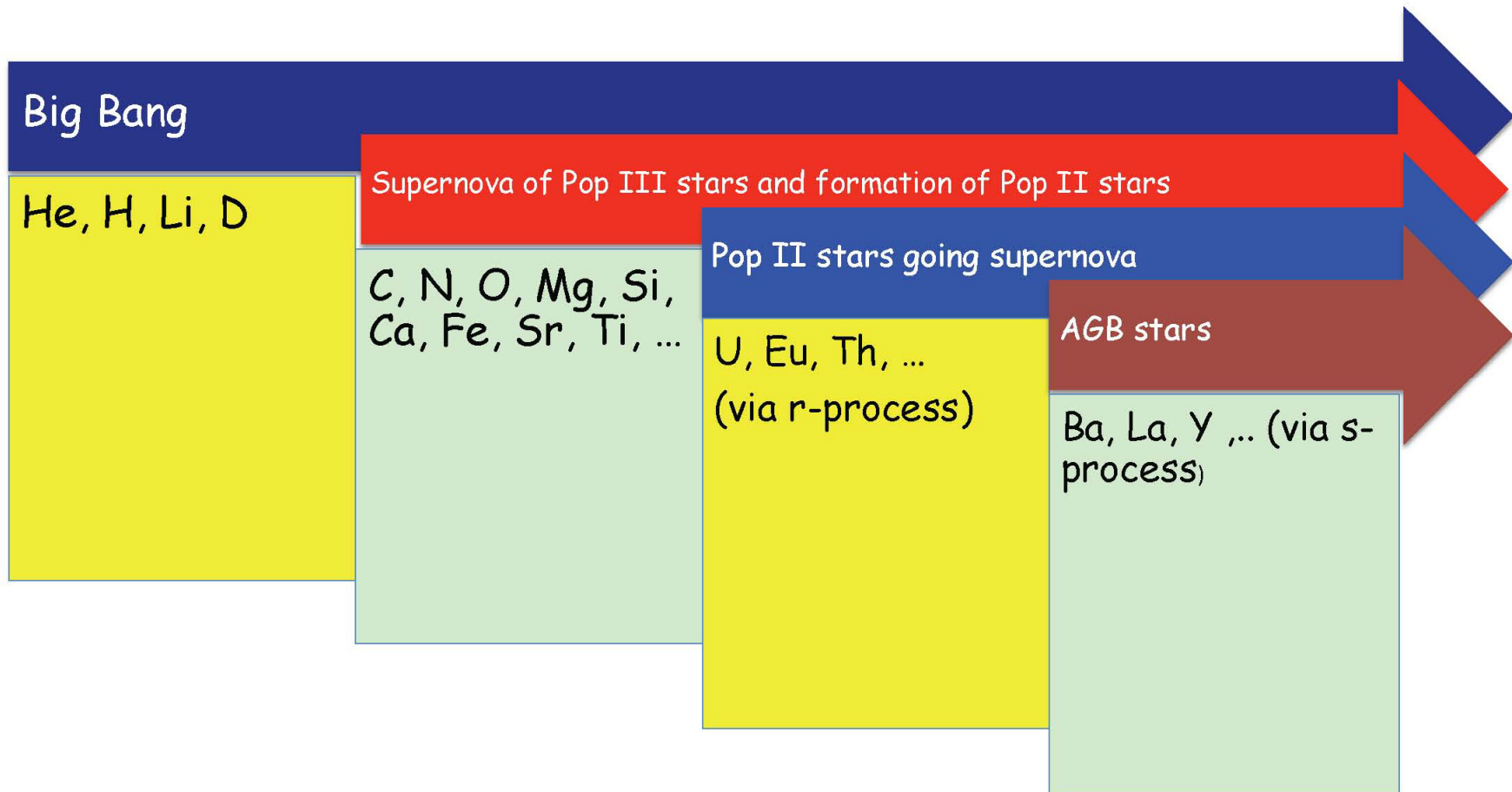
Star Mergers

Neutrino Driven Winds

MHD Supernova Jets

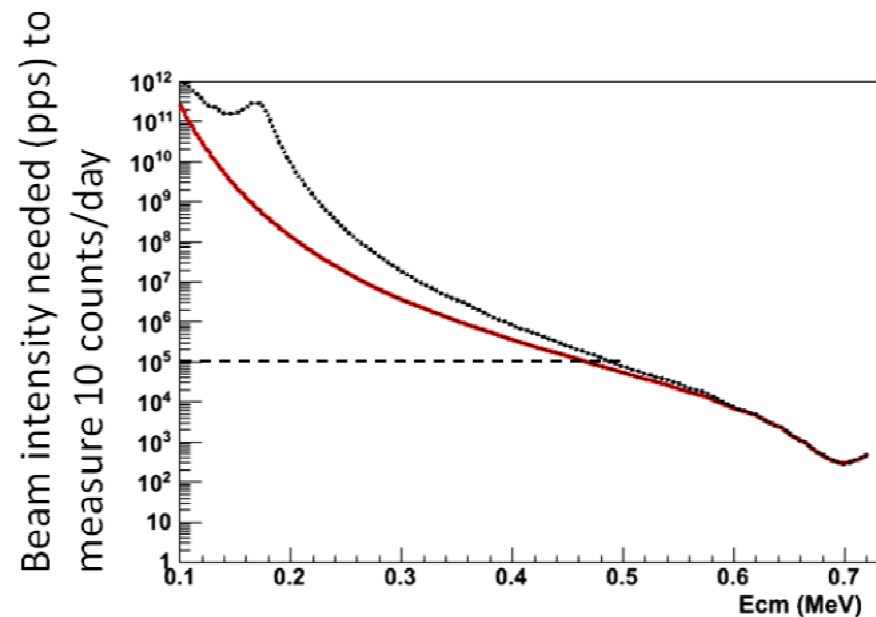
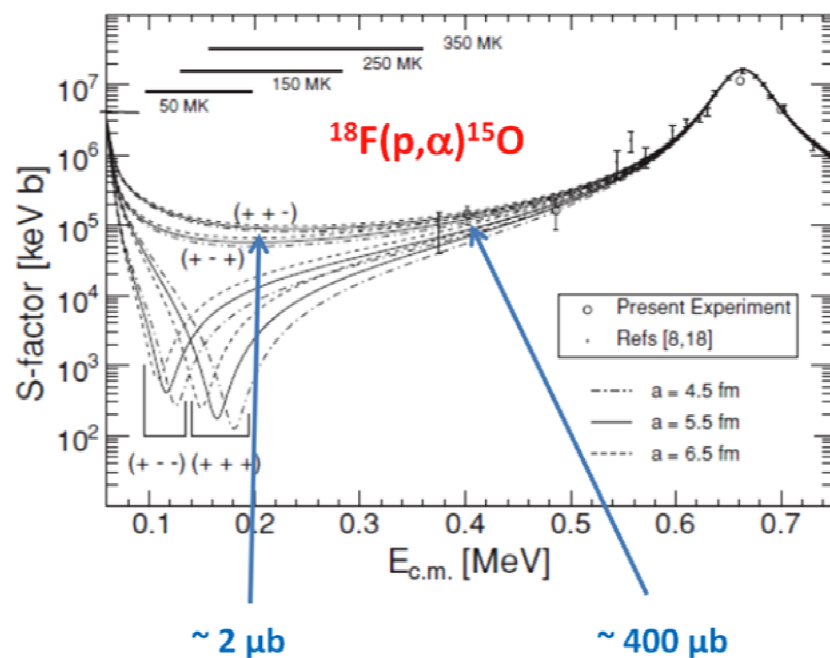
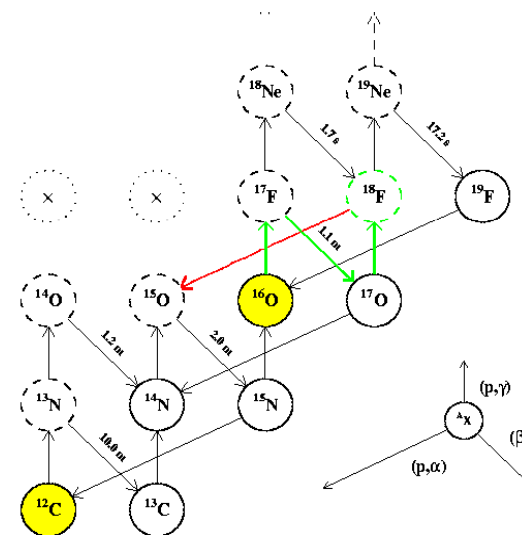


Where chemical elements are made

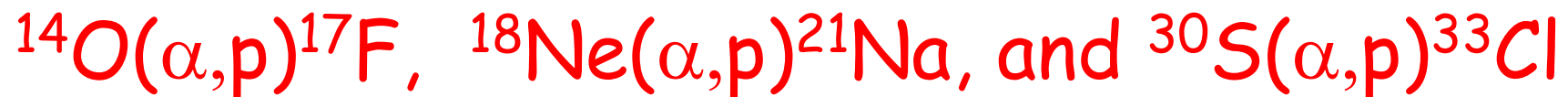
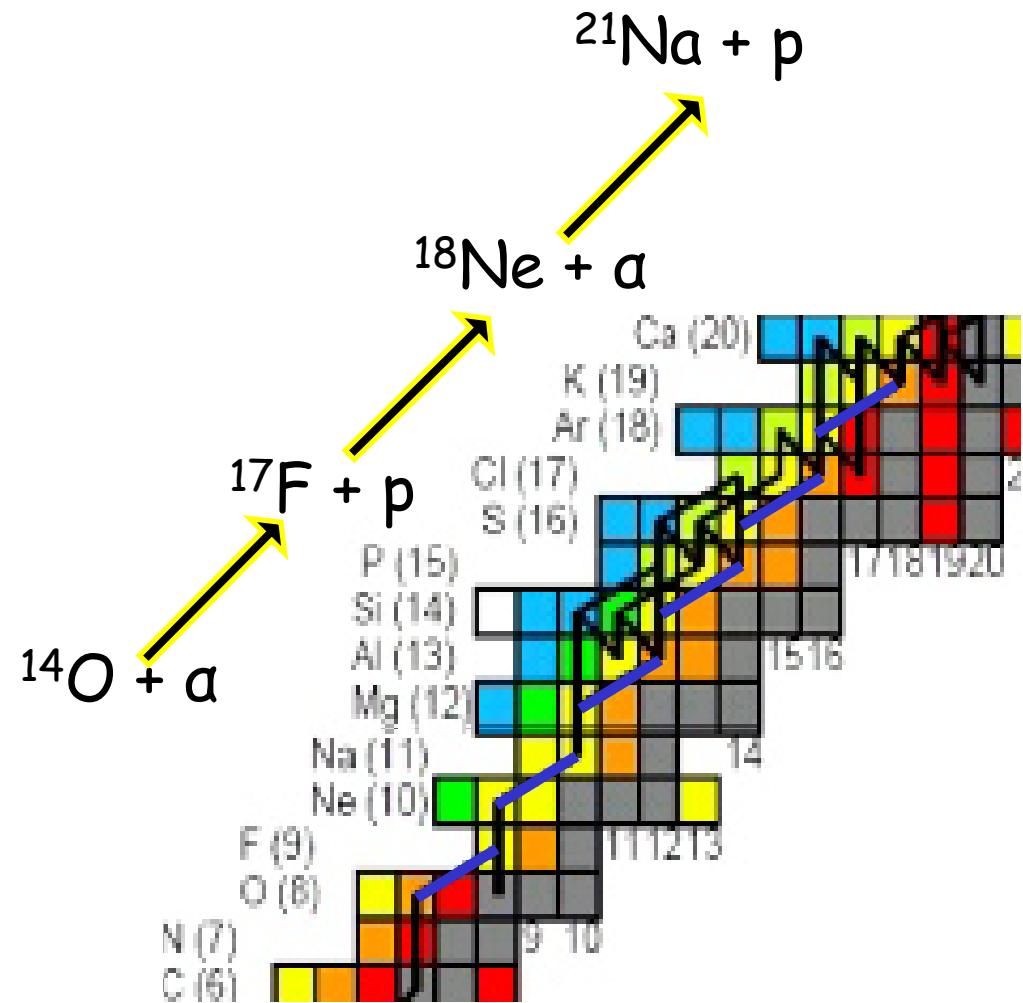
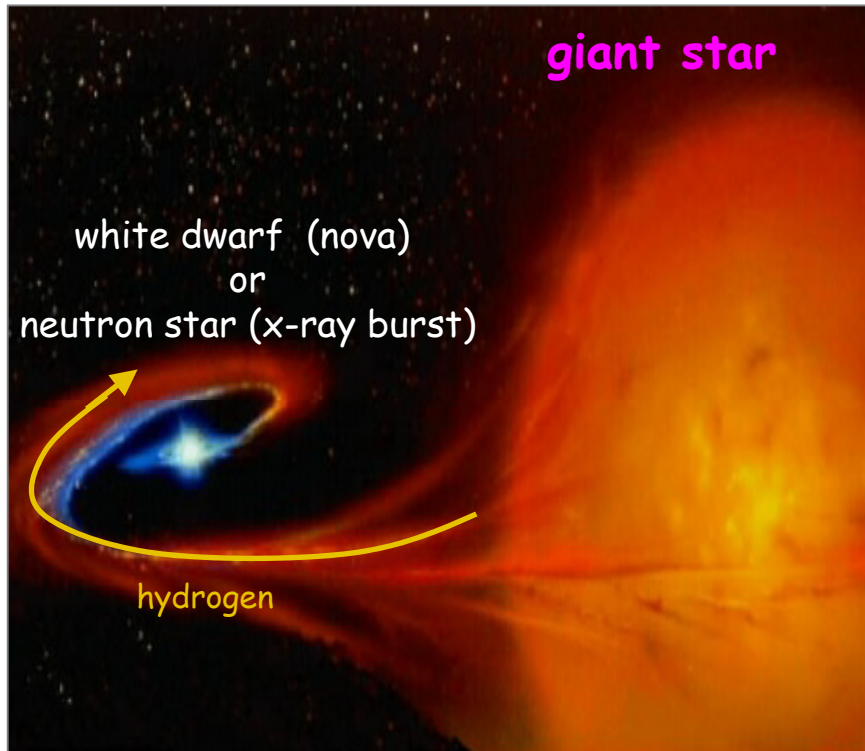


Neutrinos play a crucial role in many nucleosynthesis scenarios.

Direct reaction in Novae $^{18}\text{F}(p,\alpha)^{15}\text{O}$, $^{30}\text{P}(p,\gamma)^{31}\text{S}$, and $^{25}\text{Al}(p,\gamma)^{26}\text{Si}$,

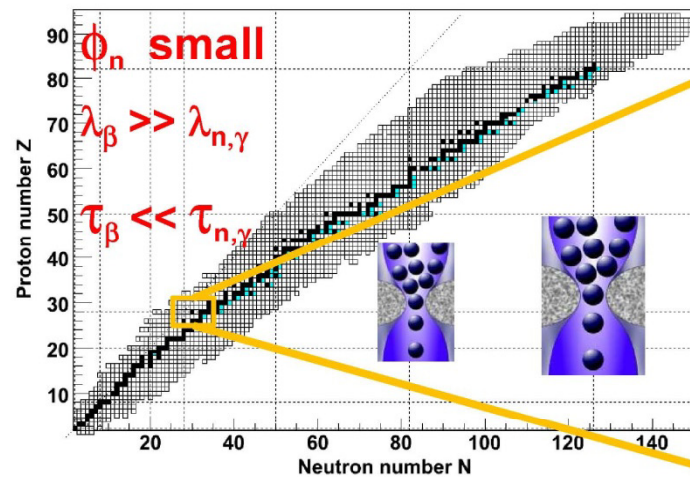


Direct reactions in x-ray bursts

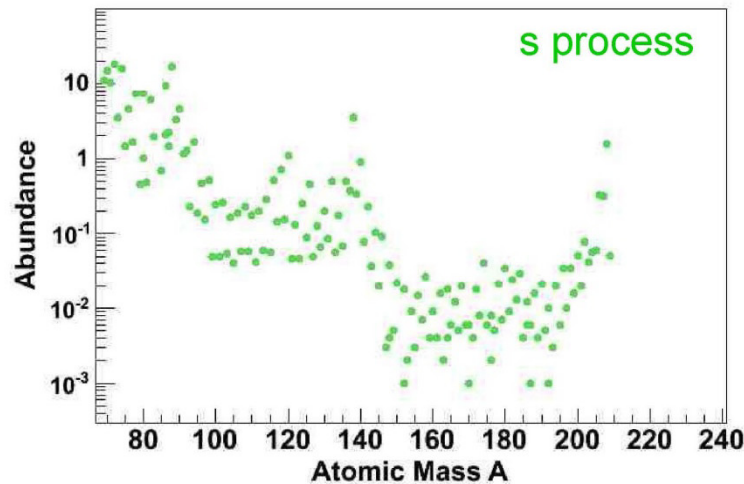
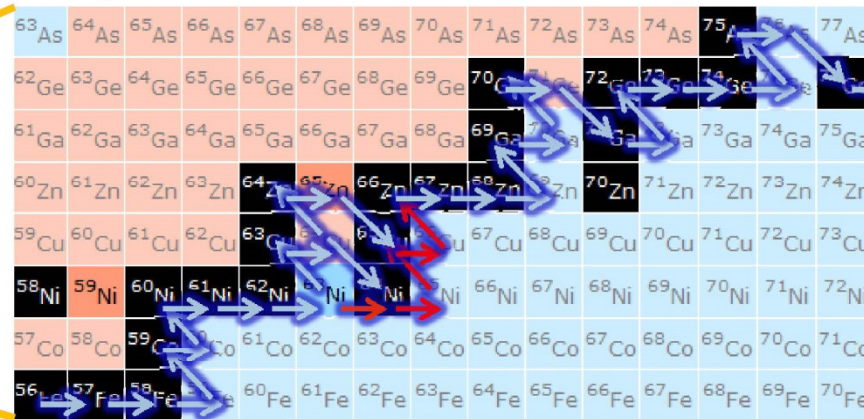


s-process nucleosynthesis and stellar n-flux

Stellar nucleosynthesis: the s process



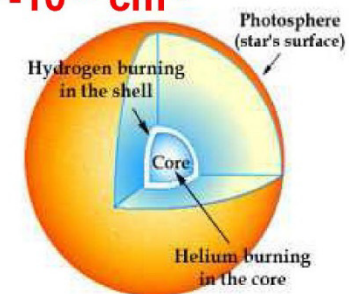
s process in AGB (Red Giant) Stars



core He-burning
 $3\text{--}3.5 \cdot 10^8 \text{ K}$
 $kT = 25 \text{ keV}$
 10^6 cm^{-3}

shell C-burning
 $\sim 1 \cdot 10^9 \text{ K}$
 $kT = 90 \text{ keV}$
 $10^{11}\text{--}10^{12} \text{ cm}^{-3}$

$^{22}\text{Ne}(\alpha, n)$



s-process nucleosynthesis and stellar n-flux

AGB- and Massive Stars: the neutron source of the S-PROCESS

s-process sites

Two components were identified and connected to stellar sites:

Main s-process $90 < A < 210$

TP-AGB stars $1-3 M_{\odot}$

shell H-burning

$0.9 \cdot 10^8 \text{ K}$

$kT = 8 \text{ keV}$

$10^7 - 10^8 \text{ cm}^{-3}$

$^{13}\text{C}(\alpha, n)^{16}\text{O}$

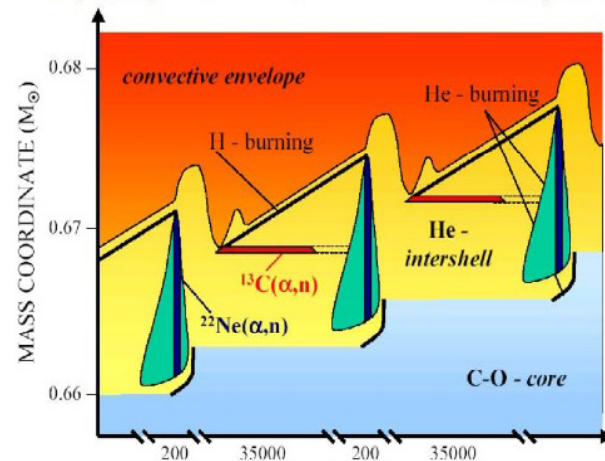
He-flash

$3 - 3.5 \cdot 10^8 \text{ K}$

$kT = 25 \text{ keV}$

$10^{10} - 10^{11} \text{ cm}^{-3}$

$^{22}\text{Ne}(\alpha, n)^{25}\text{Mg}$



Weak s-process $A < 90$

massive stars $> 8 M_{\odot}$

core He-burning

$3 - 3.5 \cdot 10^8 \text{ K}$

$kT = 25 \text{ keV}$

10^6 cm^{-3}

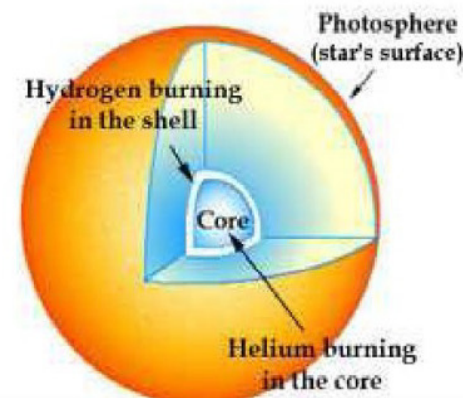
shell C-burning

$\sim 1 \cdot 10^9 \text{ K}$

$kT = 90 \text{ keV}$

$10^{11} - 10^{12} \text{ cm}^{-3}$

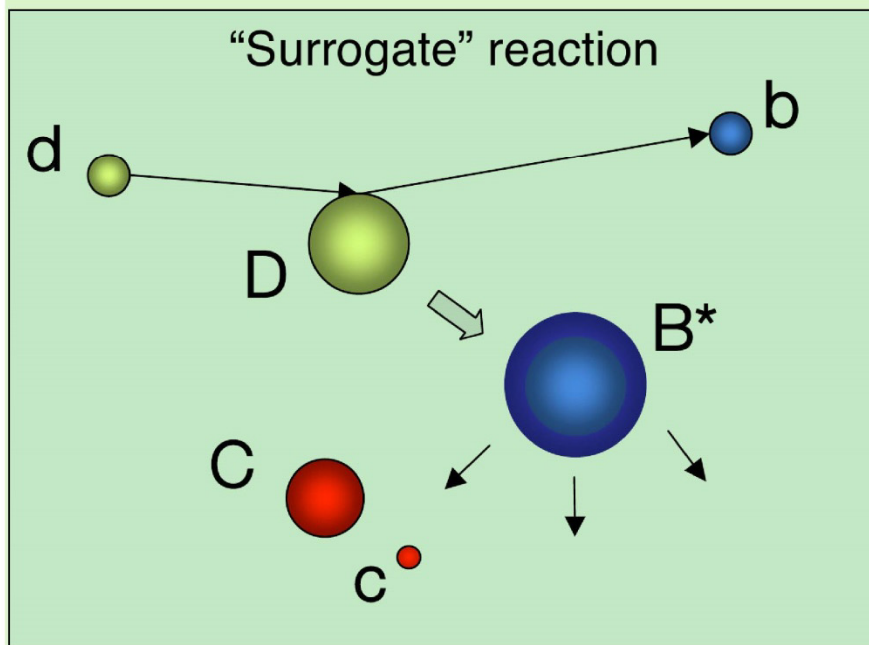
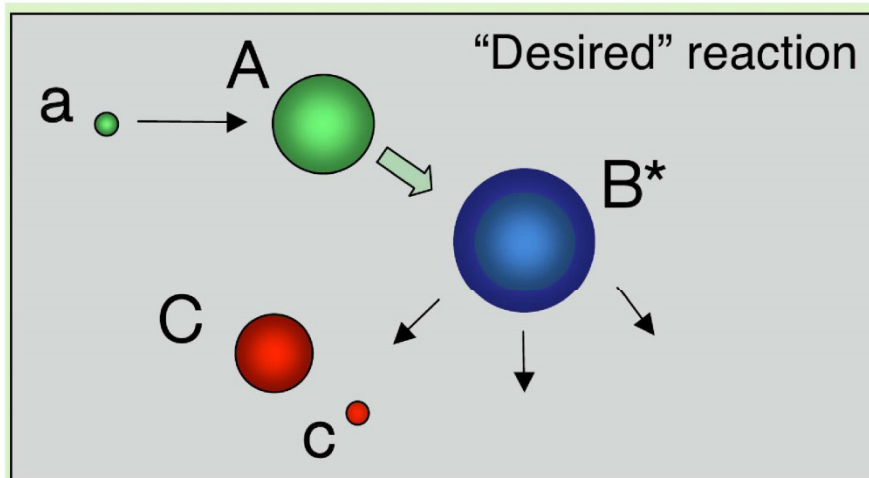
$^{22}\text{Ne}(\alpha, n)^{25}\text{Mg}$



SPES LOI: The $^{79}\text{Se}(n, \gamma)$ capture cross section via the surrogate $^{79}\text{Se}(d, p)^{80}\text{Se}$ reaction IFIC Spain

s-process nucleosynthesis and stellar n-flux

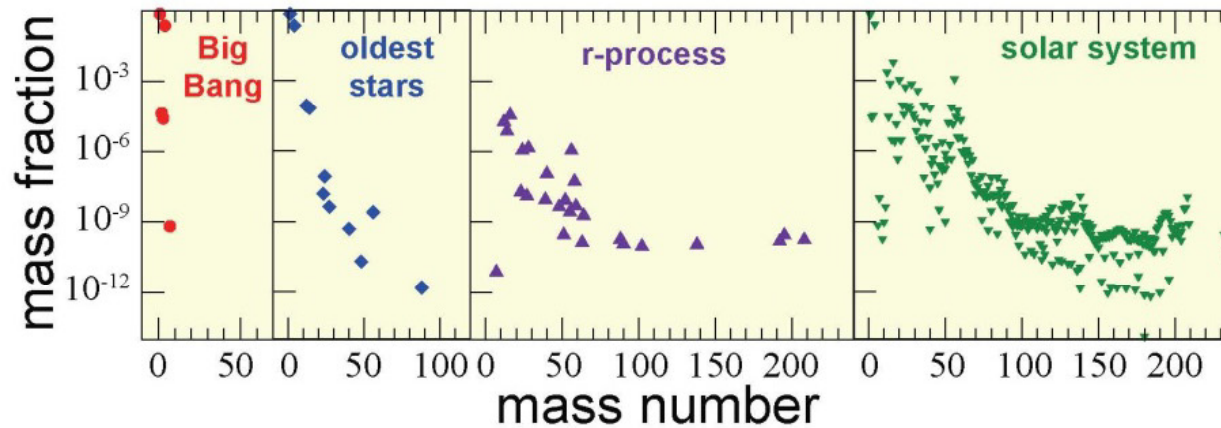
Indirect Determination of Cross Sections



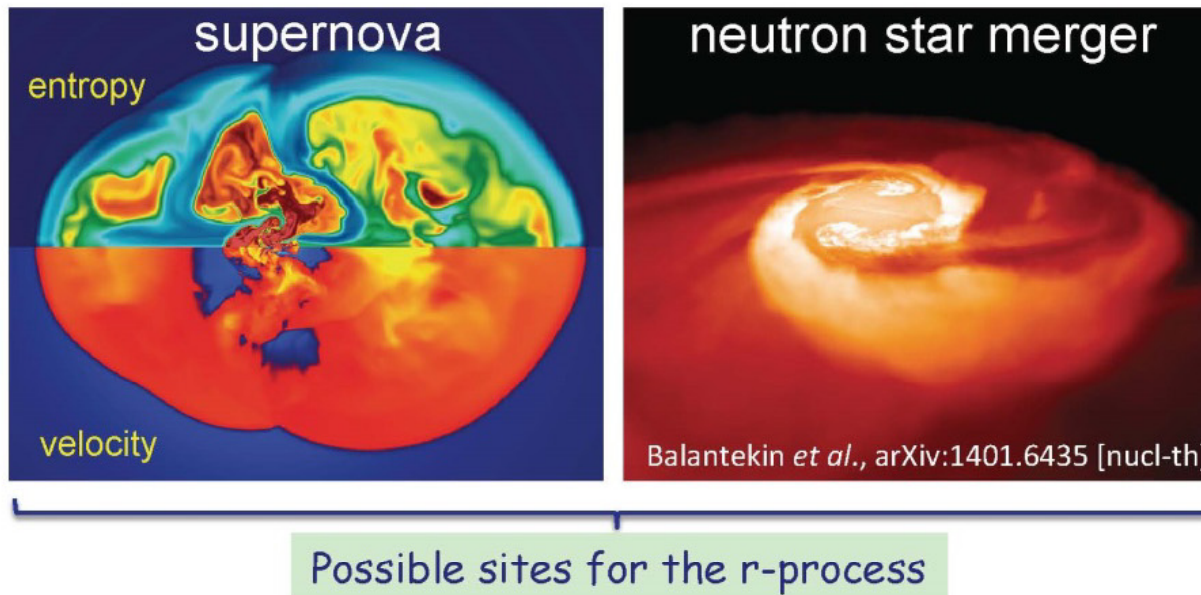
The Surrogate Nuclear Reactions approach is an indirect method for determining XS of CN reactions difficult to measure directly.

r-process nucleosynthesis

The origin of elements

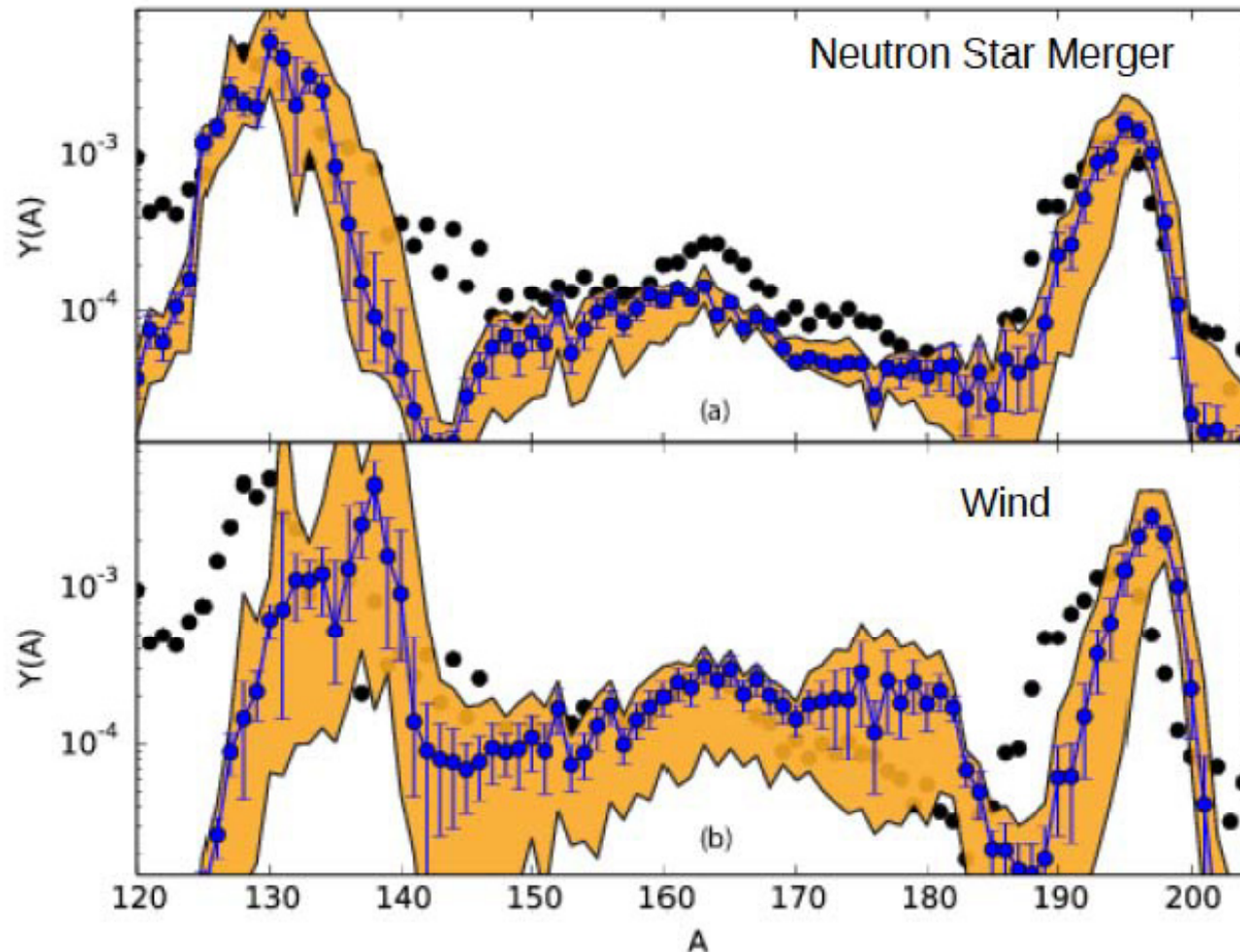


Neutrinos not only play a crucial role in the dynamics of these sites, but they also control the value of the electron fraction, the parameter determining the yields of the r-process.



r-process nucleosynthesis

Estimated Final Abundances With Uncertainties



Variations in masses of $N \sim 82$ and $N \sim 126$ nuclei of ± 1 MeV

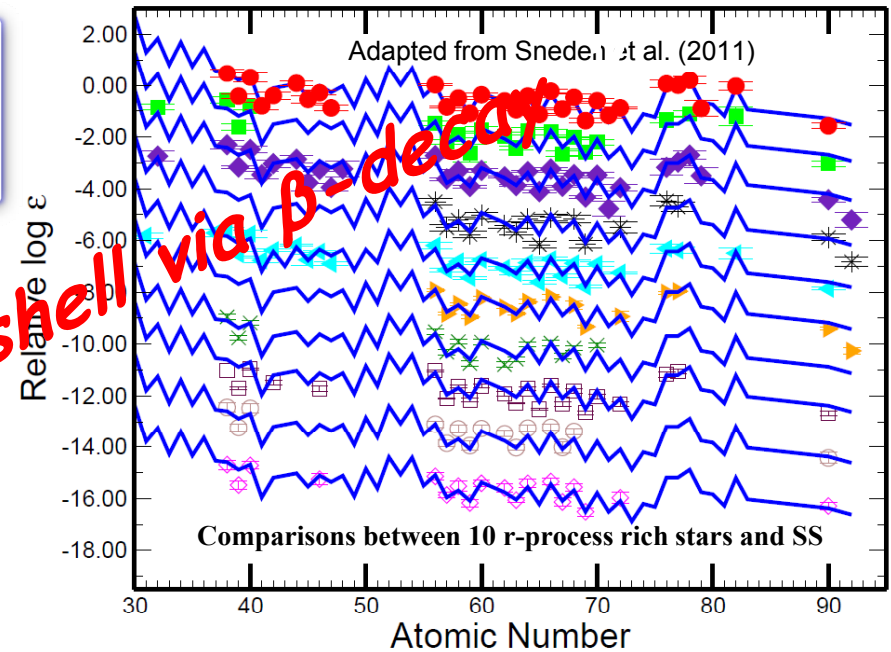
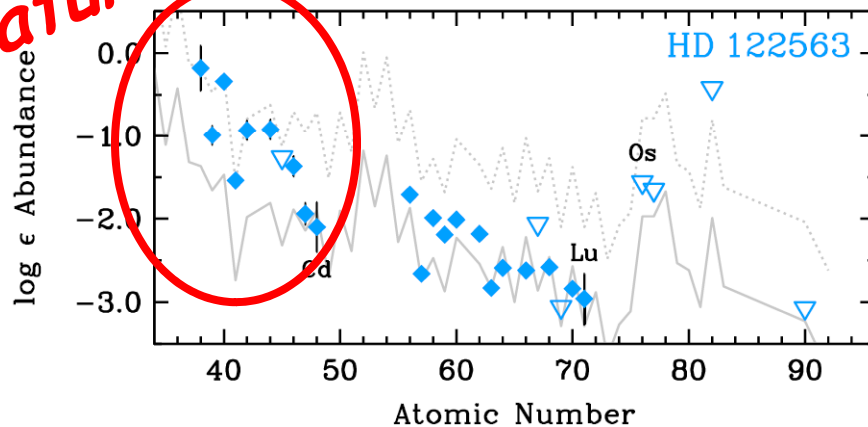
Mumpower et al. (in prep)

UMP giants stars provide crucial constraints to the stellar nucleosynthesis.

Its elemental abundances are consistent with the solar r-process elemental distribution.



Signatures of a $N = 56$ subshell via β -decay

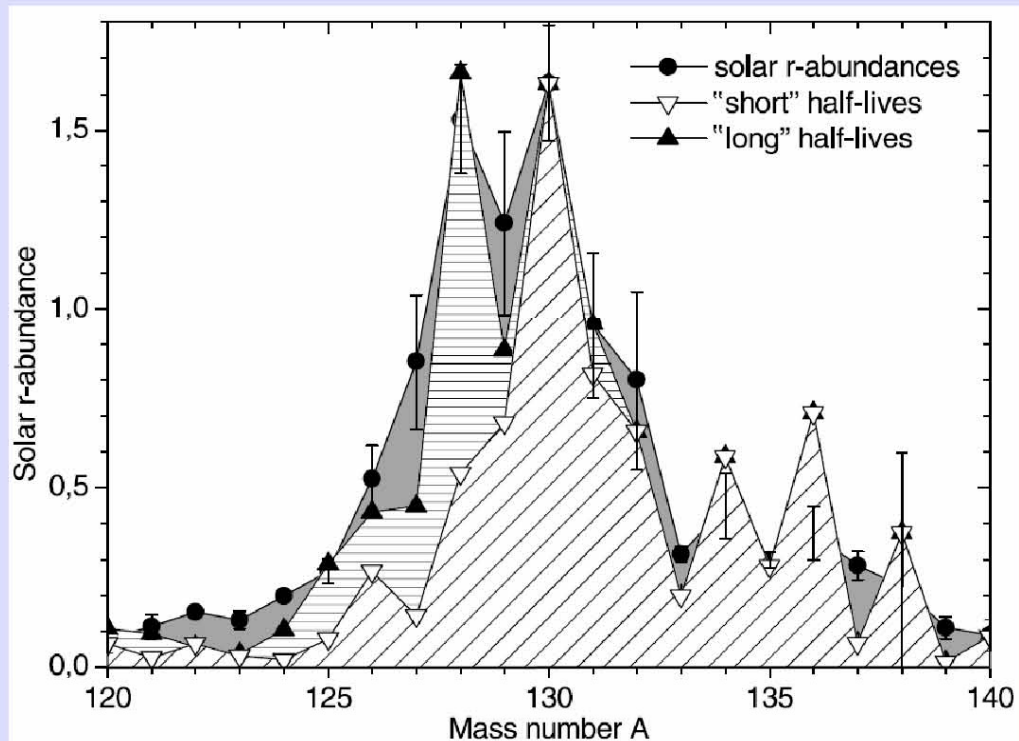


Overproduction of stable Sr, Y, and Zr isotopes in some UMP, compared to the SS r-process pattern.

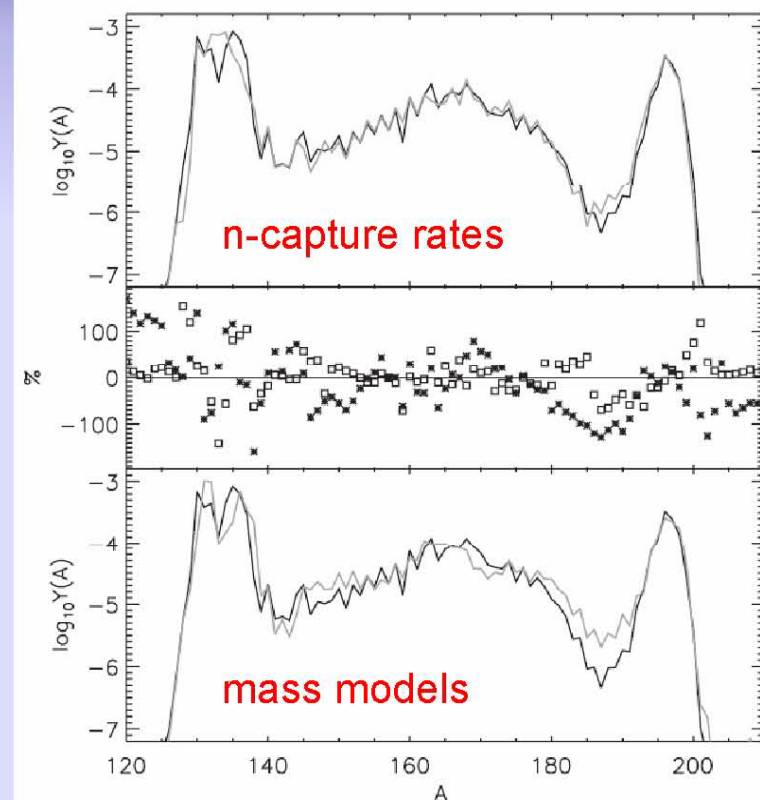
LOI ORNL (USA) r -process sensitivities

Sensitivities to global structure, and to individual n -capture rates

Adjustment of TBME to reproduce $1+$ state in ^{130}In



I. Dillman *et al.*, *Phys. Rev. Lett.* **91**, 162503 (2003)

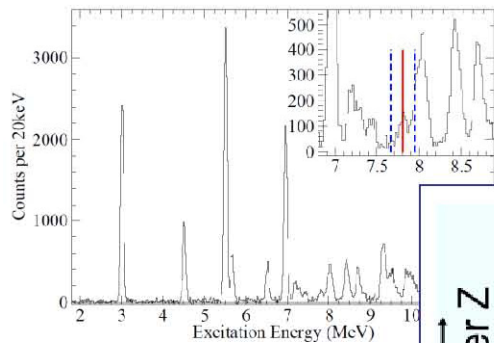


	52	^{130}Te	^{131}Te	^{132}Te	^{133}Te	^{134}Te	^{135}Te	^{136}Te	^{137}Te	^{138}Te	^{139}Te
	51	^{129}Sb	^{130}Sb	^{131}Sb	^{132}Sb	^{133}Sb	^{134}Sb	^{135}Sb	^{136}Sb	^{137}Sb	^{138}Sb
Z	50	^{128}Sn	^{129}Sn	^{130}Sn	^{131}Sn	^{132}Sn	^{133}Sn	^{134}Sn	^{135}Sn	^{136}Sn	^{137}Sn
	49	^{127}In	^{128}In	^{129}In	^{130}In	^{131}In	^{132}In	^{133}In	^{134}In	^{135}In	^{136}In
							^{131}Cd	^{132}Cd	^{133}Cd	^{134}Cd	^{135}Cd
							83	84	85	86	87
							N				

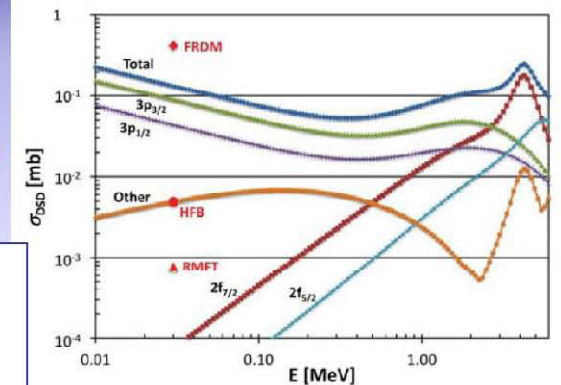
R. Surman, J. Beun, G. Ph

Transfer reactions to constrain
Capture cross sections

LOI ORNL (USA) Surrogate method

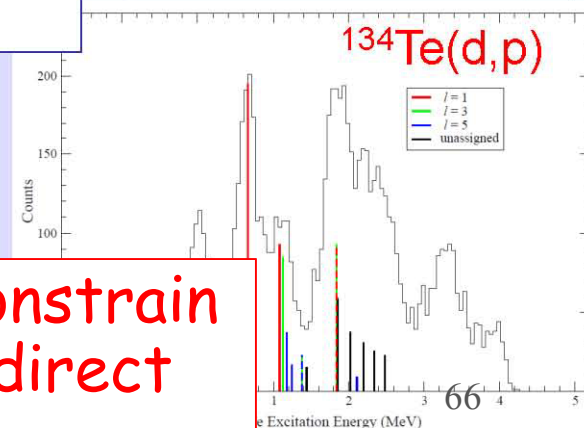
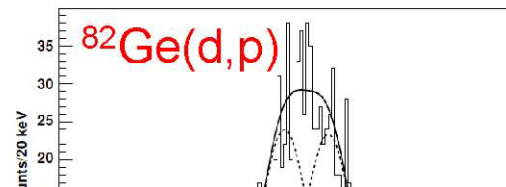
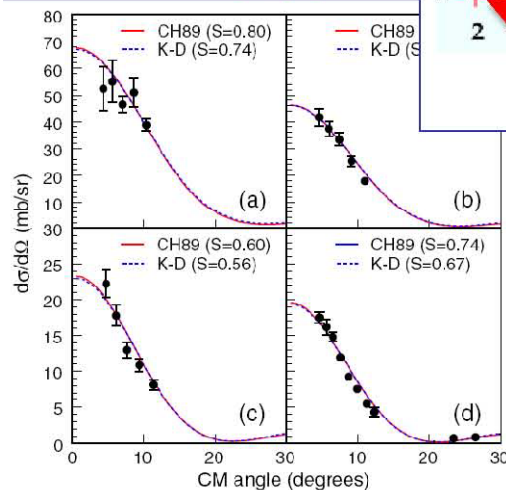
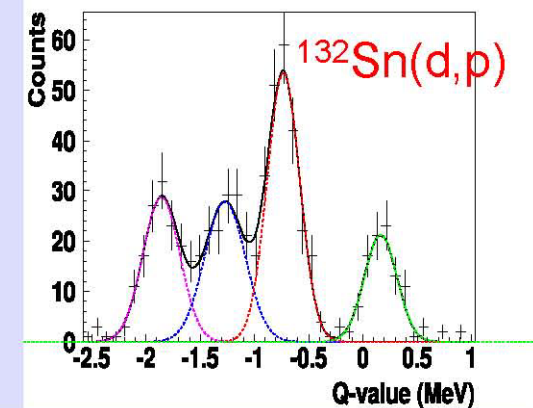
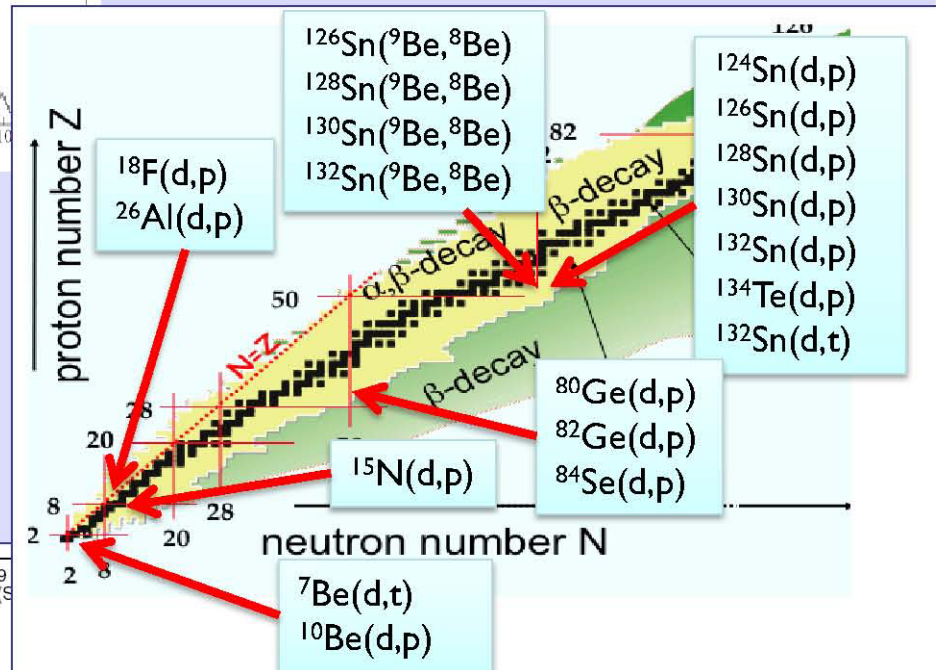


Constraint of DSD n-capture via $^{130}\text{Sn}(d,p)$



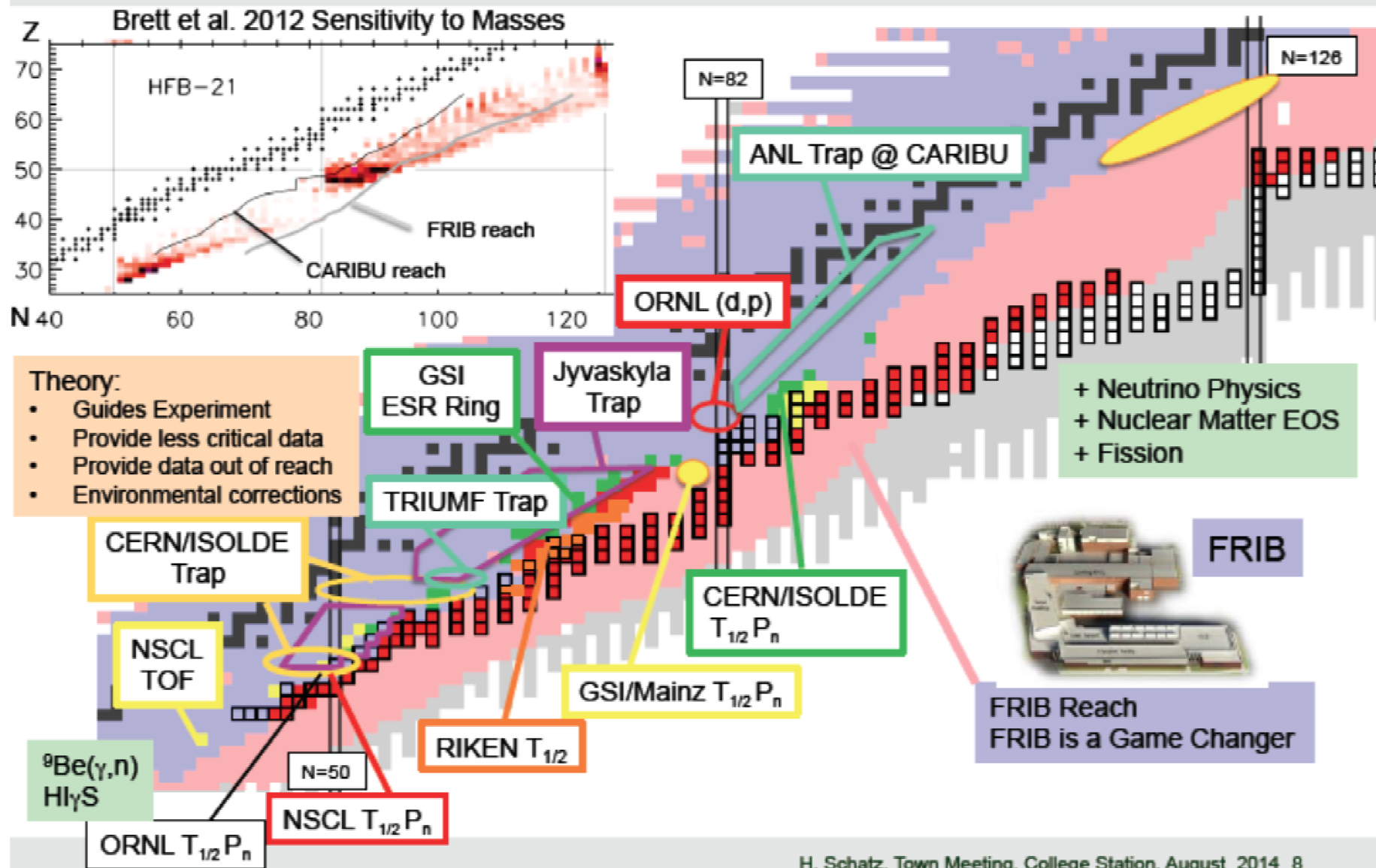
$^{26}\text{Al}(d,p)$
FWHM ~ 75 keV
(CoM)

$^{10}\text{Be}(d,p)$

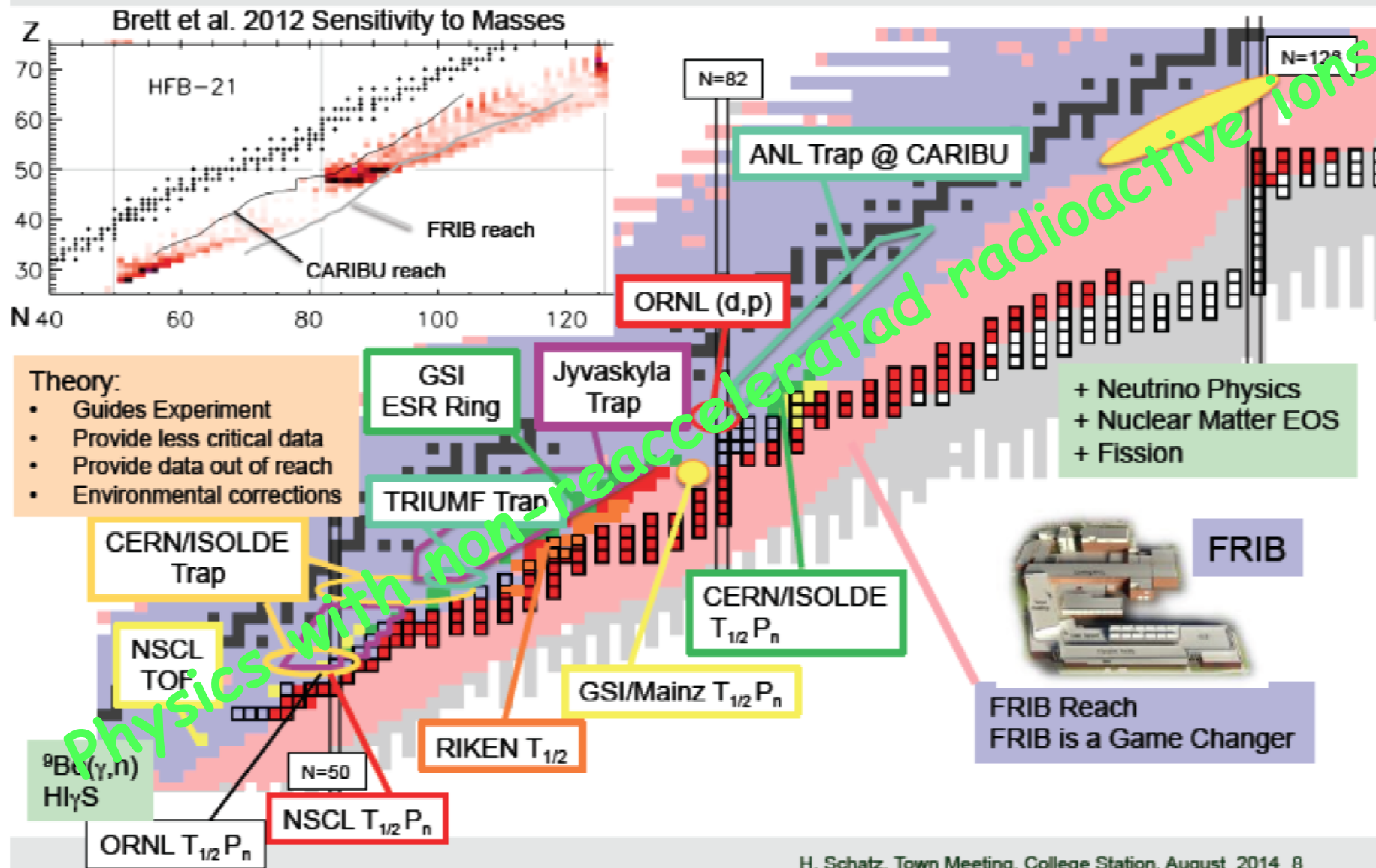


Transfer reactions to constrain capture cross sections (direct or statistical)

The Quest for r-process Nuclear Physics



The Quest for r-process Nuclear Physics



Fundamental Symmetries

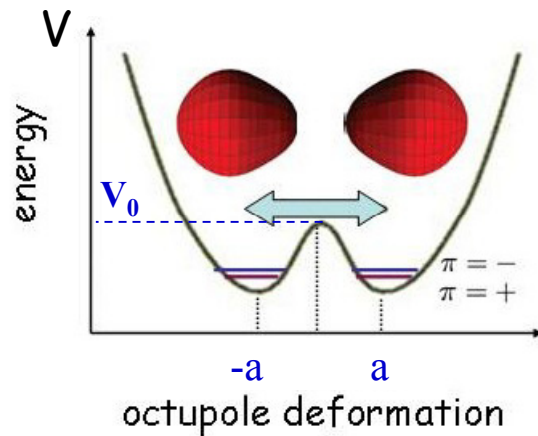
ONE OF THE CHALLENGES: REFLECTION ASYMMETRIC NUCLEI AND STATIC ELECTRIC DIPOLE MOMENT



The lopsided nuclei, described today (May 8) in the journal Nature, could be good candidates for researchers looking for new types of physics beyond the reigning explanation for the bits of matter that make up the universe (called the [Standard Model](#)), said study author [Peter Butler](#), a physicist at the University of Liverpool in the United Kingdom.

The findings could help scientists search for physics beyond the Standard model, said [Witold Nazarewicz](#). An electric dipole moment would provide a way to test extension theories to the Standard Model, such as supersymmetry, which could help explain why there is more matter than antimatter in the universe.

The double oscillator

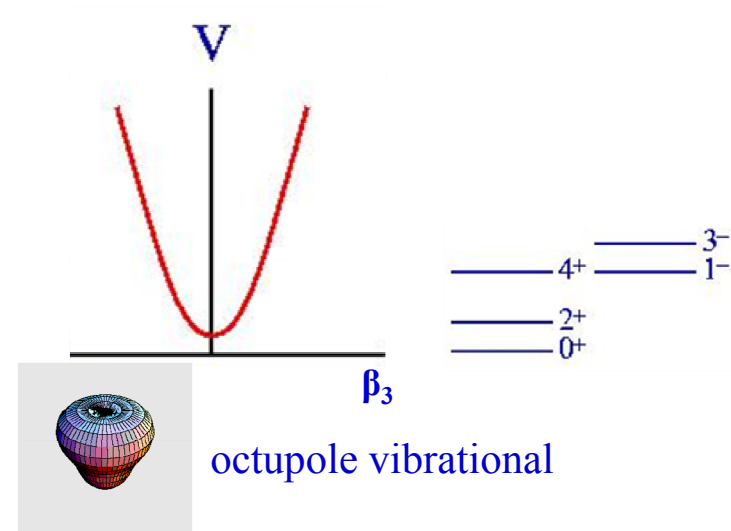
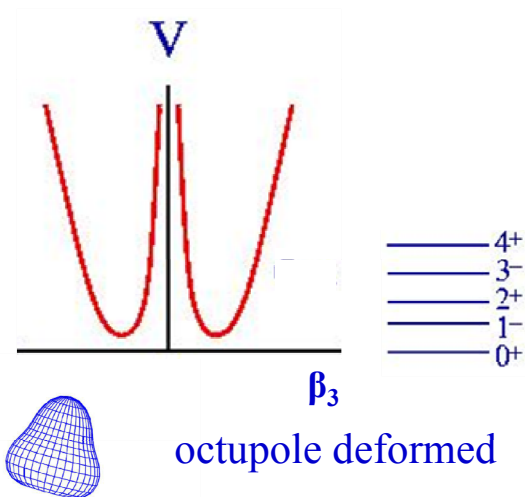


$$\begin{aligned}
 |\Psi\rangle &= |\text{prolate}\rangle \\
 P|\Psi\rangle &= |\text{oblate}\rangle \\
 P|\Psi\rangle &\neq |\Psi\rangle
 \end{aligned}$$

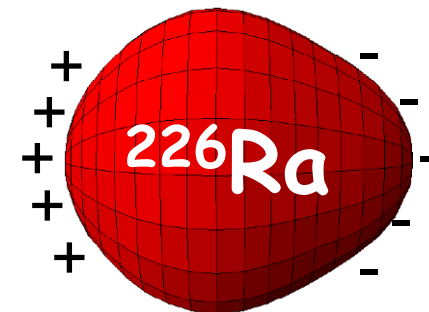
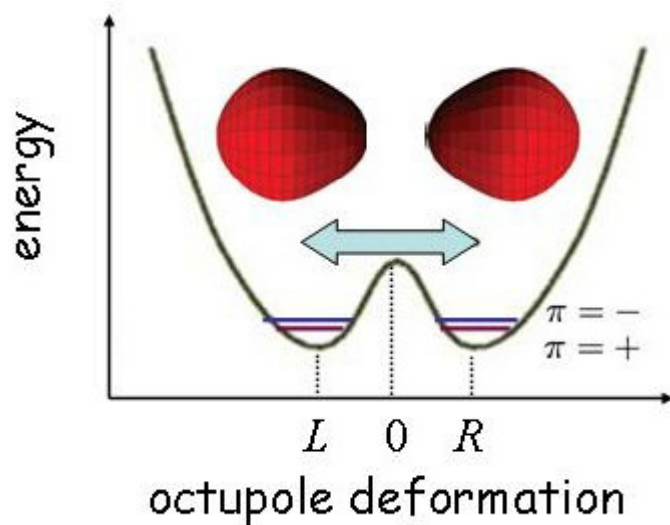
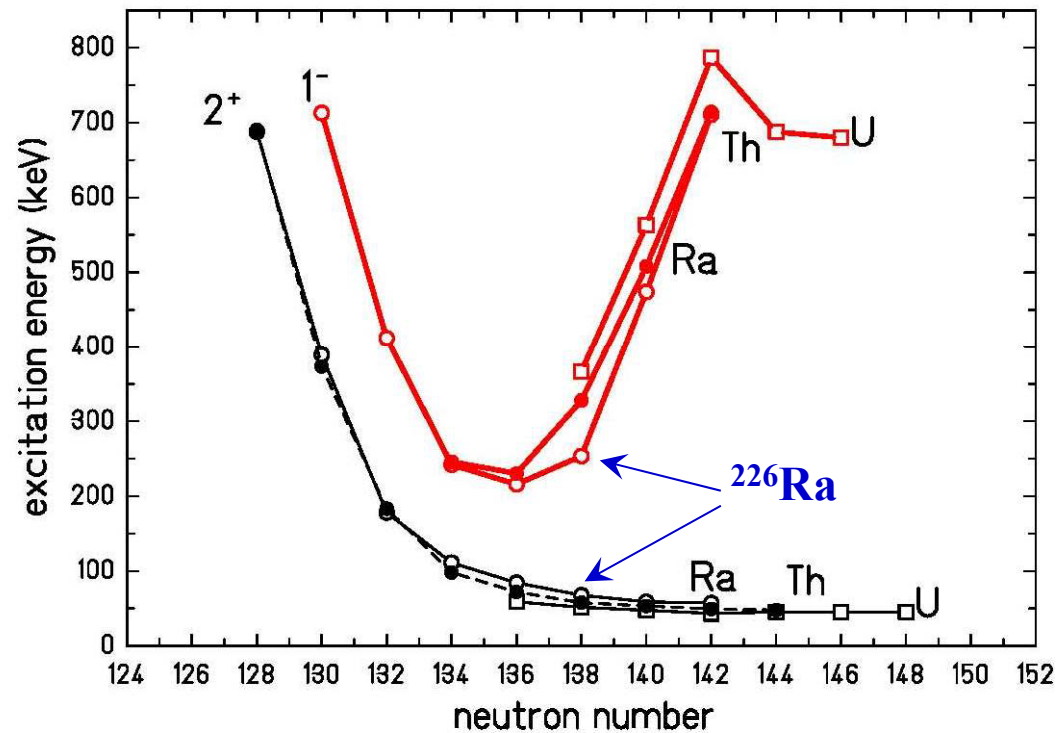
$$H \cdot \psi = -\frac{\hbar^2}{2 \cdot B} \frac{\partial^2 \psi}{\partial \beta_3^2} + \frac{V_0}{a^2} \cdot (|\beta_3| - a)^2 \cdot \psi = E \cdot \psi$$

$$E_{\text{even}} = \hbar\omega \cdot \left(\nu_{\text{even}} + \frac{1}{2} \right) = \hbar\omega \cdot \left(\frac{1}{2} - \sqrt{\frac{2 \cdot V_0}{\hbar\omega \cdot \pi}} \cdot e^{-\frac{2V_0}{\hbar\omega}} \right)$$

$$E_{\text{odd}} = \hbar\omega \cdot \left(\nu_{\text{odd}} + \frac{1}{2} \right) = \hbar\omega \cdot \left(\frac{1}{2} + \sqrt{\frac{2 \cdot V_0}{\hbar\omega \cdot \pi}} \cdot e^{-\frac{2V_0}{\hbar\omega}} \right)$$



Octupole collectivity



In an **octupole** deformed nucleus the center of mass and center of charge tend to separate, creating a non-zero **electric dipole moment**.

ONE OF THE CHALLENGES: REFLECTION ASYMMETRIC NUCLEI AND STATIC ELECTRIC DIPOLE MOMENT

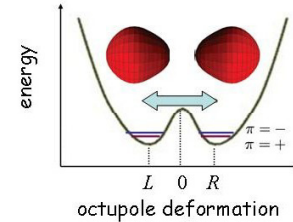
V Spevak, N Auerbach, and VV Flambaum
PR C **56** (1997) 1357

Schiff moment:

$$S = -2 \frac{J}{J+1} \frac{\langle \hat{S}_z \rangle \langle \hat{V}_{PT} \rangle}{\Delta E}$$

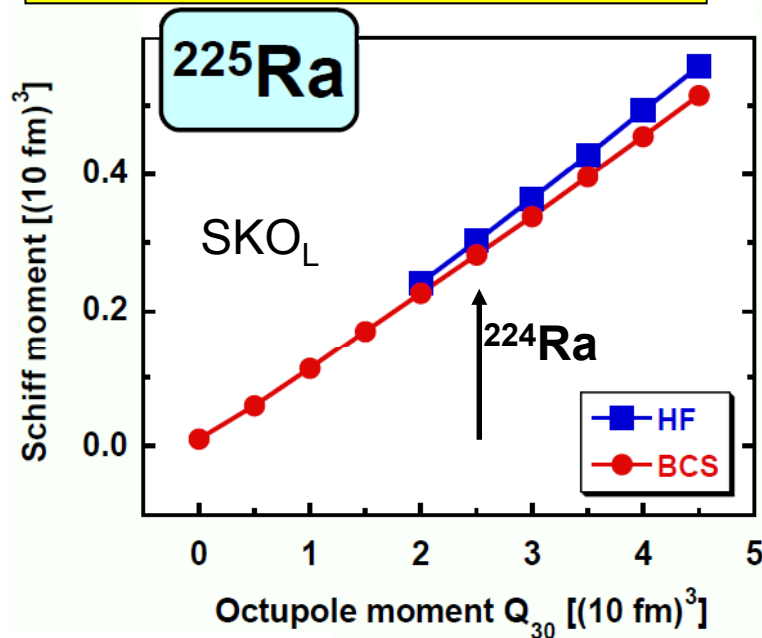
related to Q_3

P,T-violating n-n interaction



energy splitting of parity doublet

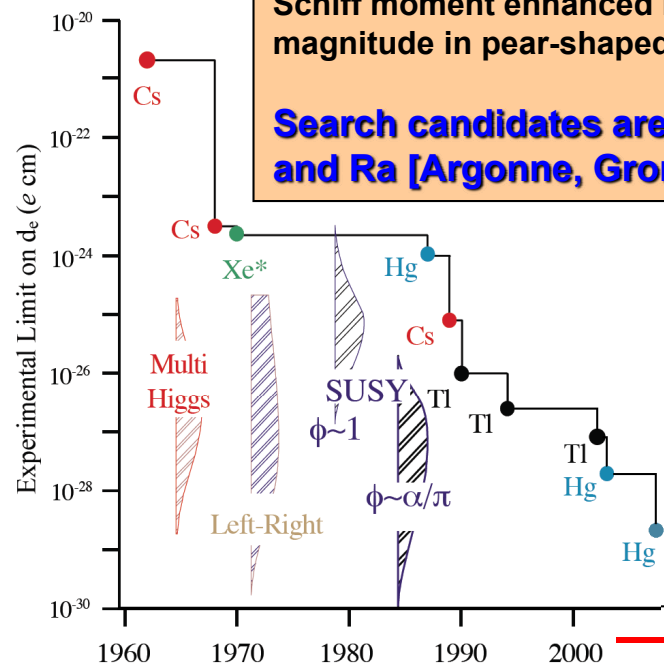
J Dobaczewski (Trento, 2010)



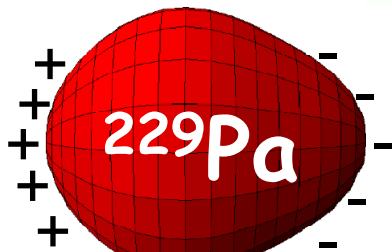
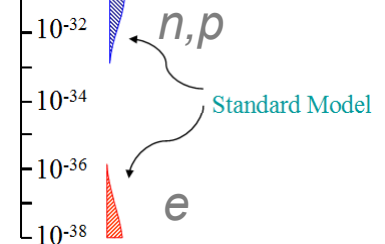
Experimental Limit on d_e (e cm)

Schiff moment enhanced by ~ 3 orders of magnitude in pear-shaped nuclei

Search candidates are odd-A Rn [TRIUMF] and Ra [Argonne, Groningen/ISOLDE]

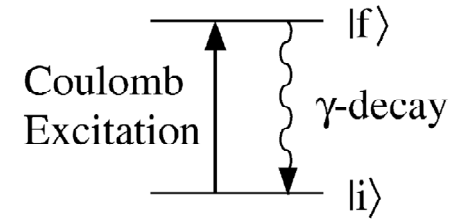
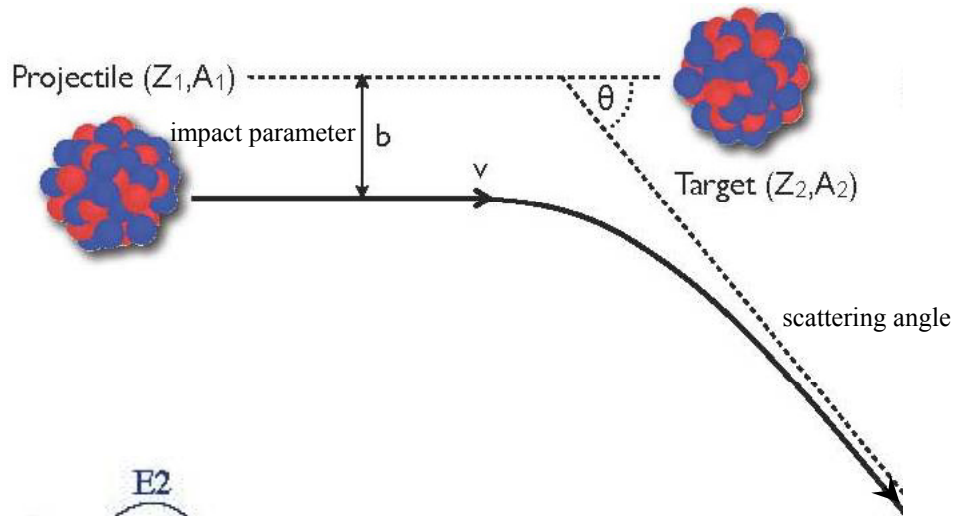


Rn, Ra



Measure: Q_3 in even-A Rn, Ra
 ΔE in odd-A Rn

Coulomb excitation

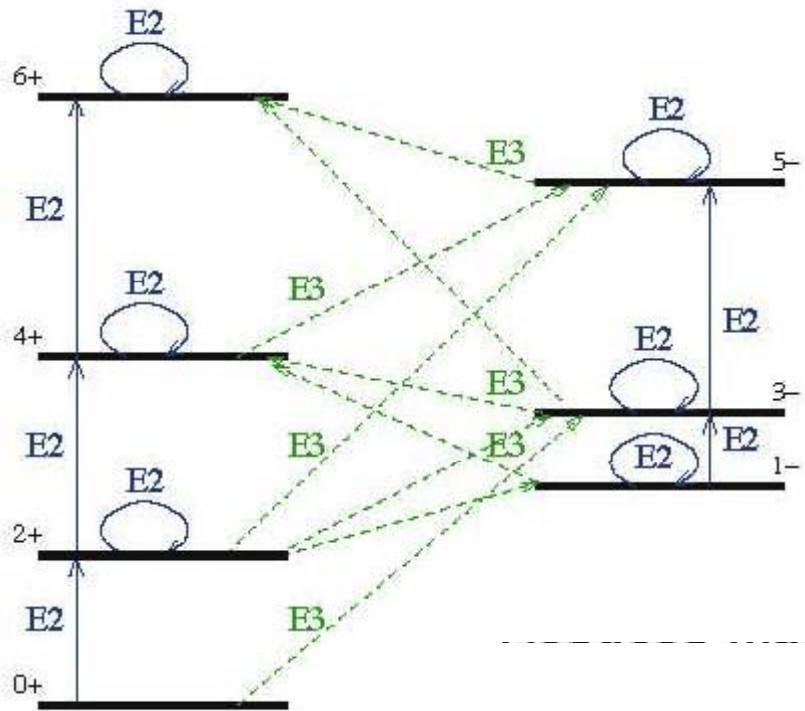


$$\frac{d\sigma_{i \rightarrow f}}{d\Omega_{cm}} = P_{i \rightarrow f} \cdot \frac{d\sigma_{Ruth}}{d\Omega_{cm}}$$

$$d\sigma_{E2} \cong 4.819 \cdot \left(1 + \frac{A_1}{A_2}\right)^{-2} \cdot \frac{A_1}{Z_2^2} \cdot E_{MeV} \cdot B(E2; I_i \rightarrow I_f) \cdot df_{E2}(\eta, \xi) [b]$$

Goal: Octupole deformation in ^{229}Pa

^{229}Pa by $^{232}\text{Th}(p,4n)$ reaction
Octupole strength measured
via Coulex



Application of Isotopic Beams

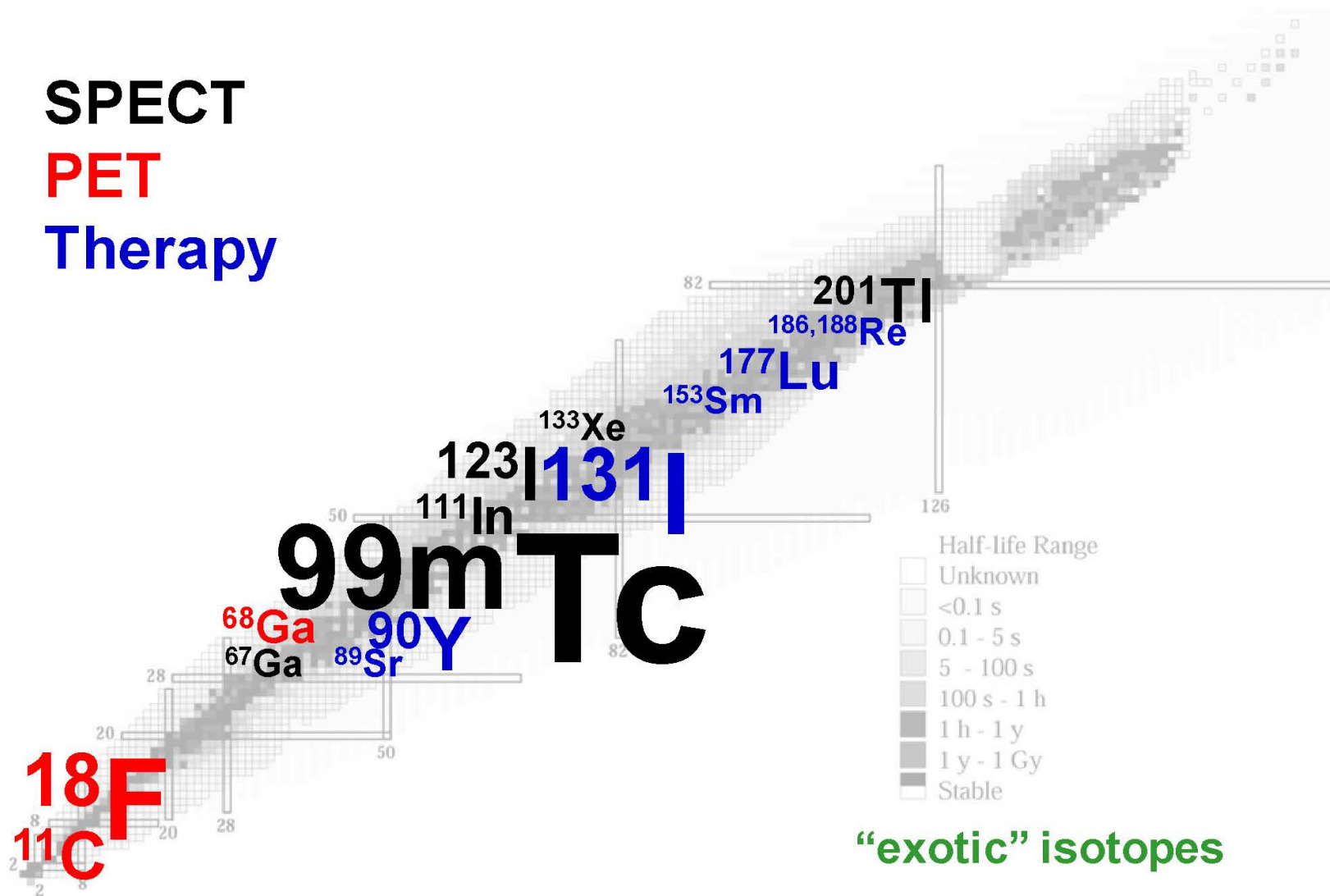
ONE OF THE CHALLENGES: RADIONUCLIDEI FOR MEDICINE

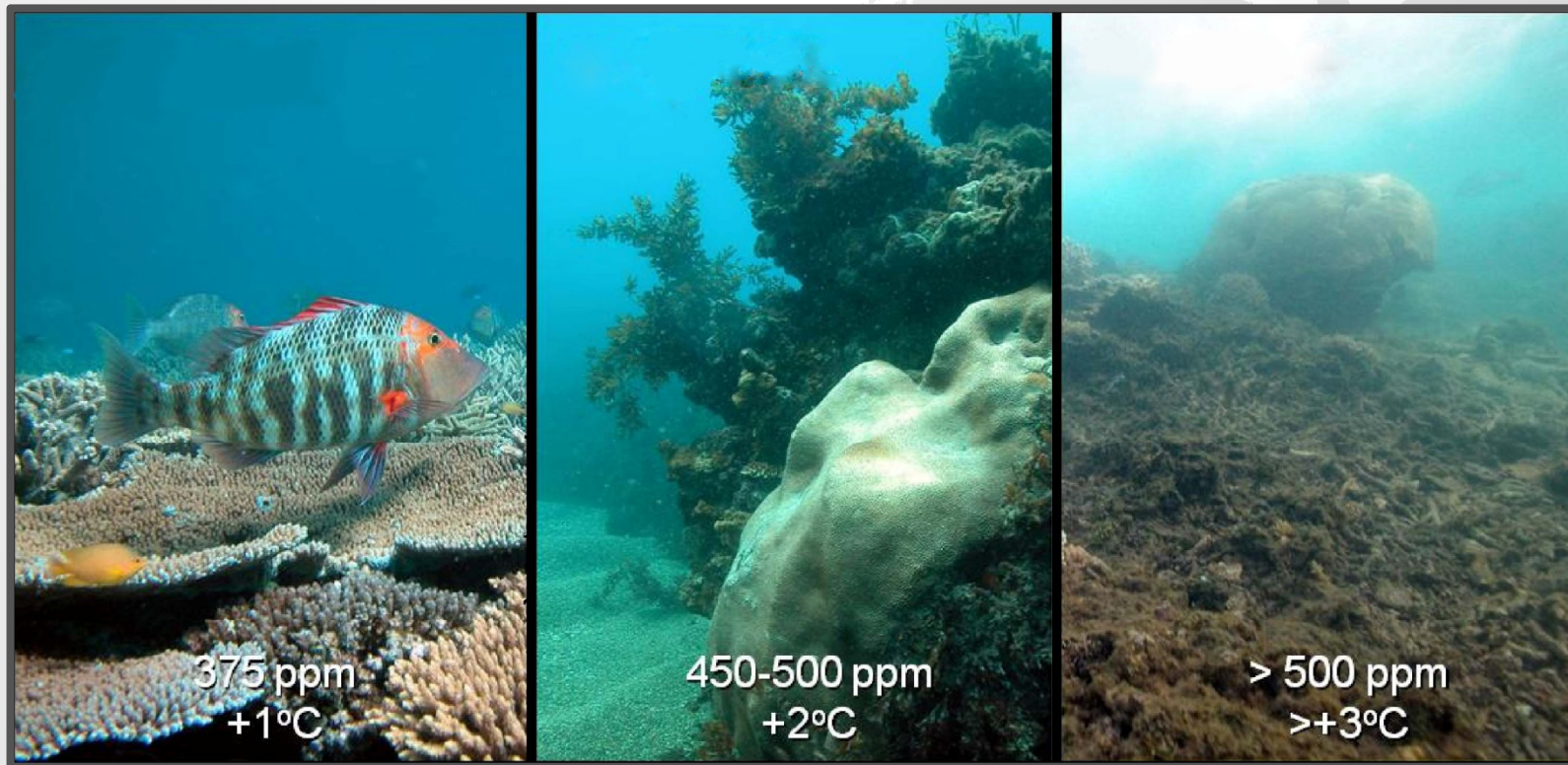
The chart of nuclides – nuclear medicine perspective

SPECT

PET

Therapy





<http://www.noaanews.noaa.gov>

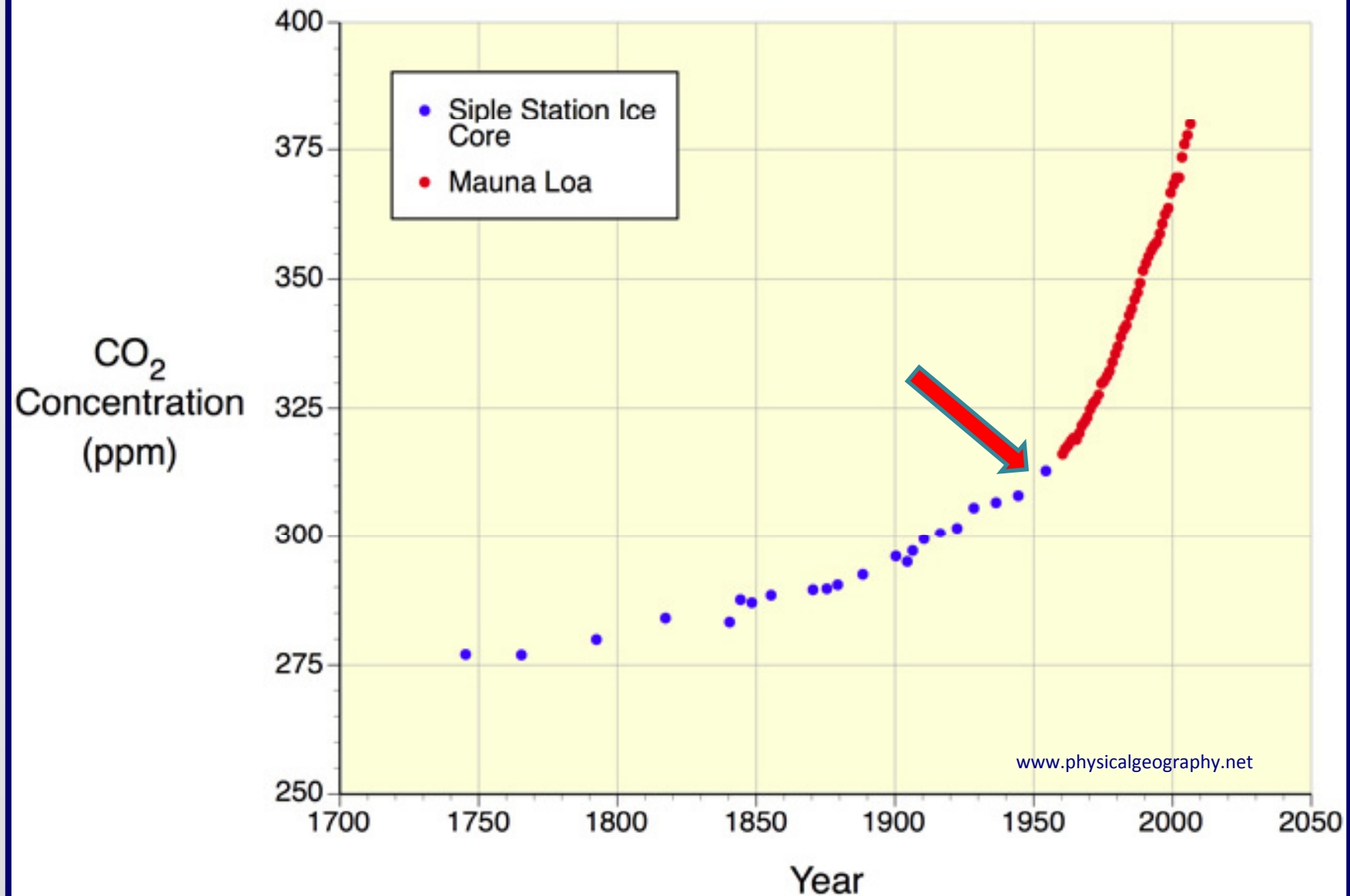
Ocean Acidification

Will the reef survive?

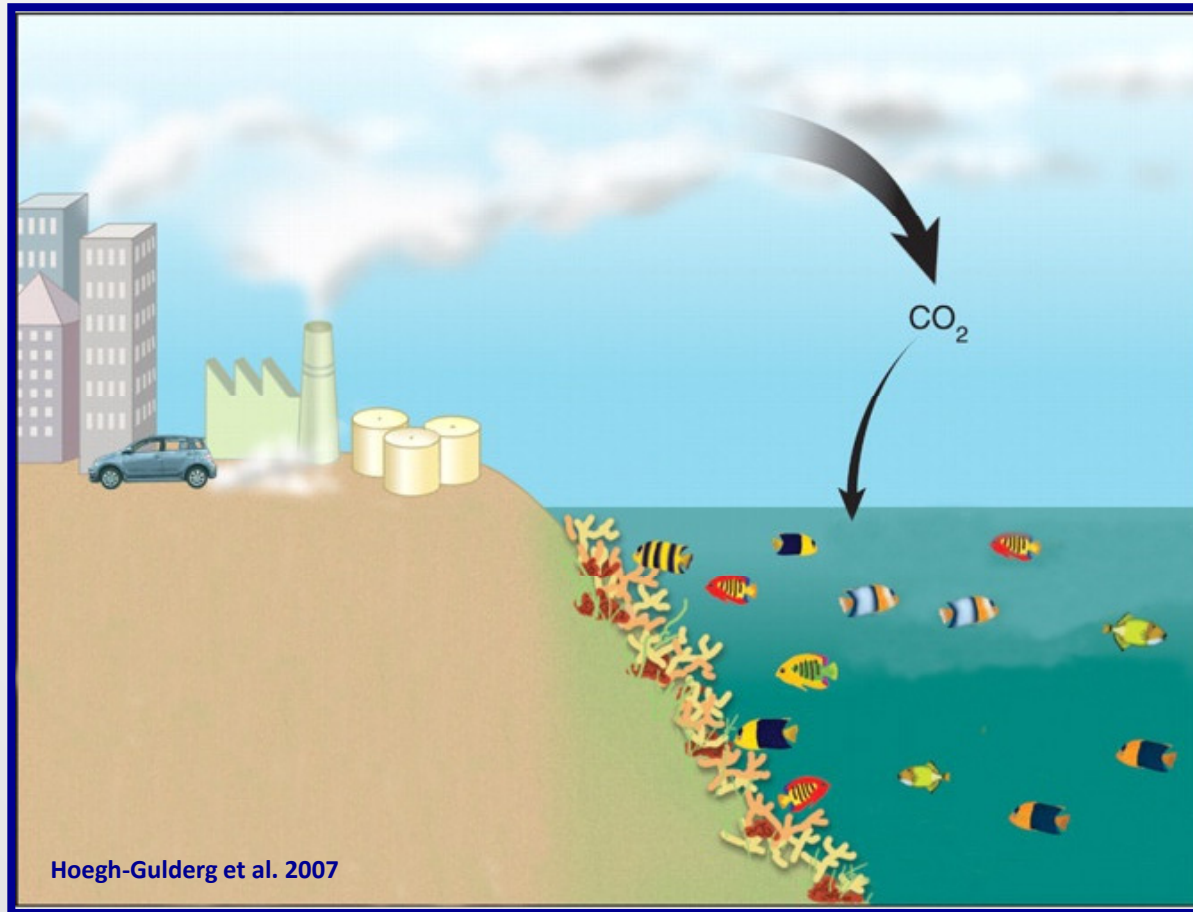
**The burning of fossil
fuels releases
11 BILLION TONS
of carbon dioxide into the
atmosphere every year.**



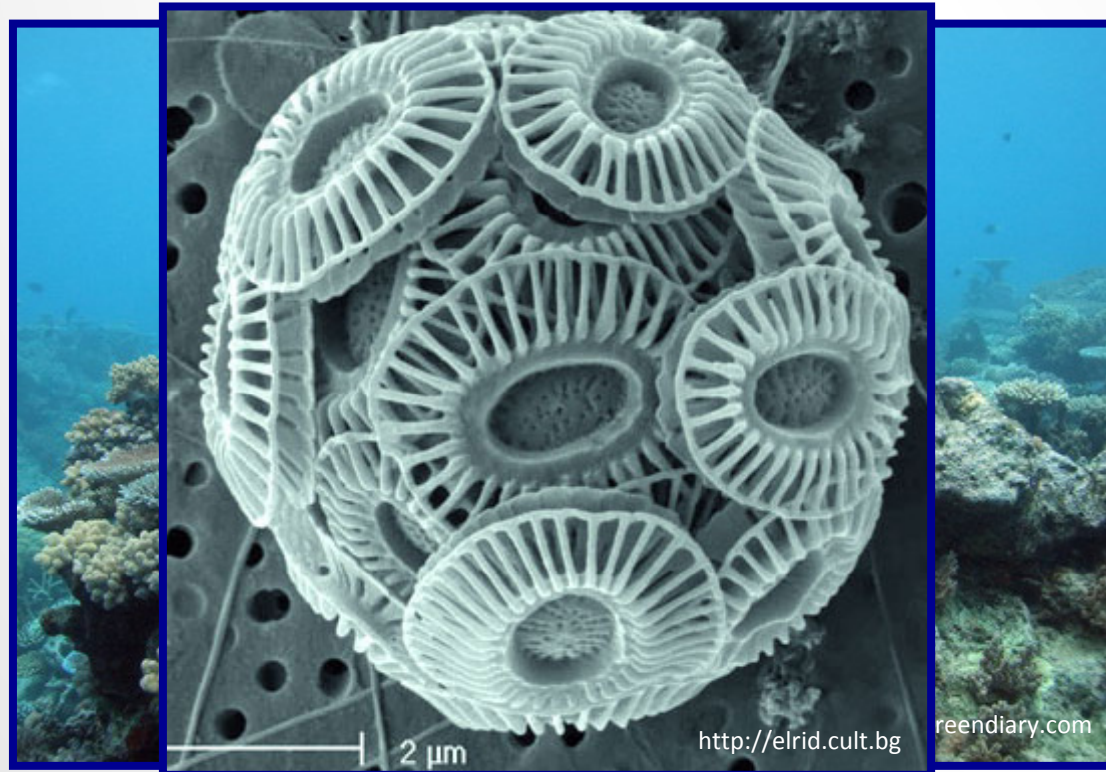
Atmospheric Concentration of Carbon Dioxide (1744-2005)



Carbon dioxide dissolves in the ocean, where it causes a potentially more serious problem → ocean acidification.



Ocean acidification poses a threat to shell-forming organisms like corals and calcifying plankton.



Ocean acidification - numbers

- Ocean/atmosphere exchanges of CO₂ are very important
- Net uptake by oceans of 25-30% of man-made CO₂ emissions: **24 million tons** CO₂ per day!
- Key defence (buffer) against global warming, but drawback: causing an increase in acidity (decrease in pH).
- Acidity of the oceans has increased by **30%** since the onset of the industrial revolution.
- If CO₂ emissions continue at the current rate, acidity will increase by **150%** by 2100 (highest acidity experienced by marine ecosystems since at least **800 000 years**).
- The current rate of pH change is unprecedented for **300 millions years**.

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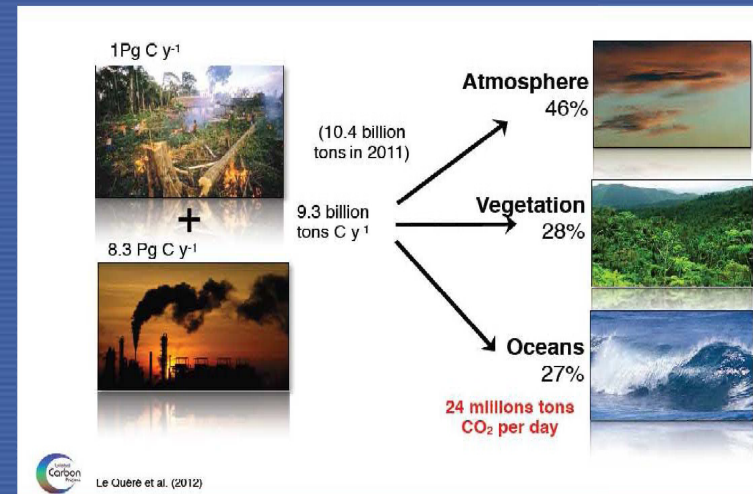


Image courtesy of J.-P. Gattuso

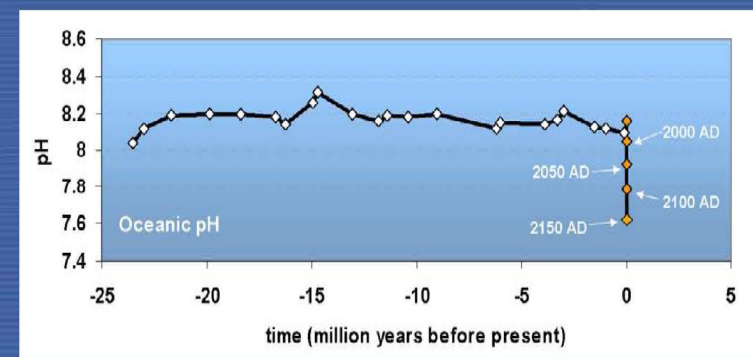


Image courtesy of C. Turley

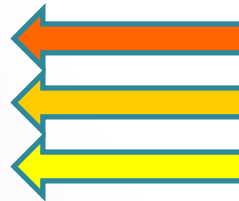
The pH Scale

ACIDS

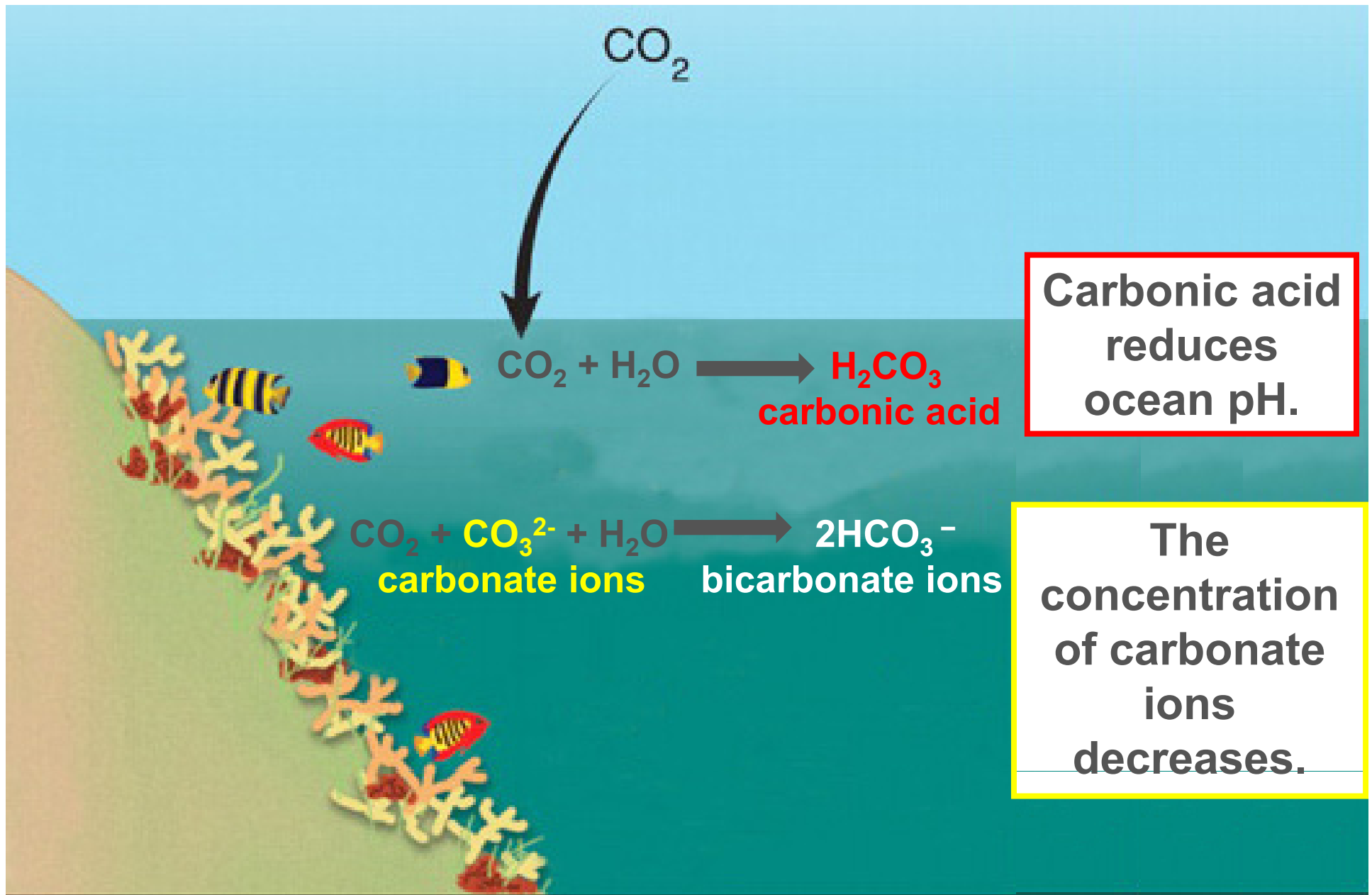
BASES

Concentration of Hydrogen ions compared to distilled water		Examples
10,000,000	pH 0	Battery acid
1,000,000	pH 1	Hydrochloric acid
100,000	pH 2	Lemon juice, vinegar
10,000	pH 3	Grapefruit, soft drink
1,000	pH 4	Tomato juice, acid rain
100	pH 5	Black coffee
10	pH 6	Urine, saliva
1	pH 7	"Pure" water
1/10	pH 8	Sea water
1/100	pH 9	Baking soda,
1/1,000	pH 10	Great Salt Lake
1/10,000	pH 11	Ammonia solution
1/100,000	pH 12	Soapy water
1/1,000,000	pH 13	Bleach
1/10,000,000	pH 14	Liquid drain cleaner

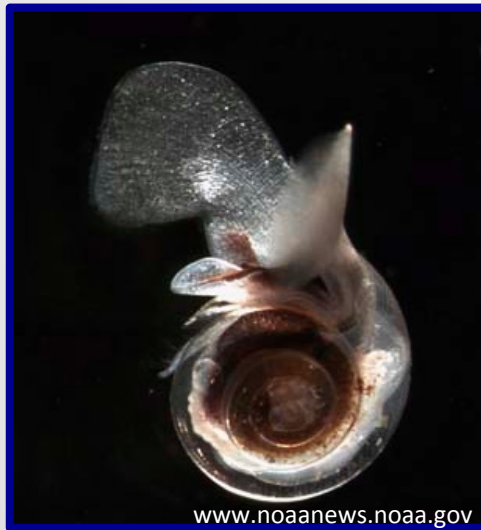
www.thegardenersresource.com



A small change in pH is equal to a LARGE change in acidity.



As the ocean acidifies, organisms such as corals, snails, and calcifying plankton will not be able to make their shells and grow.



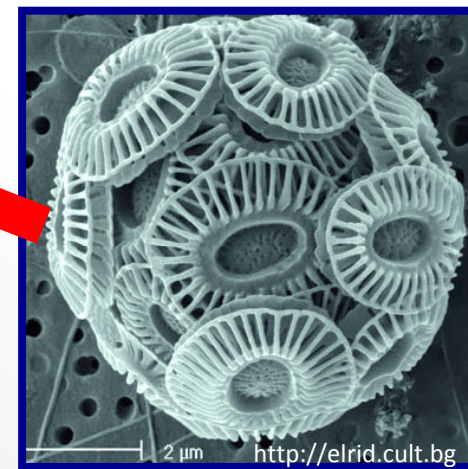
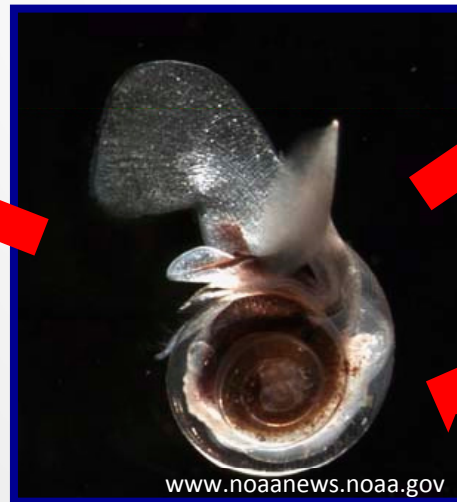
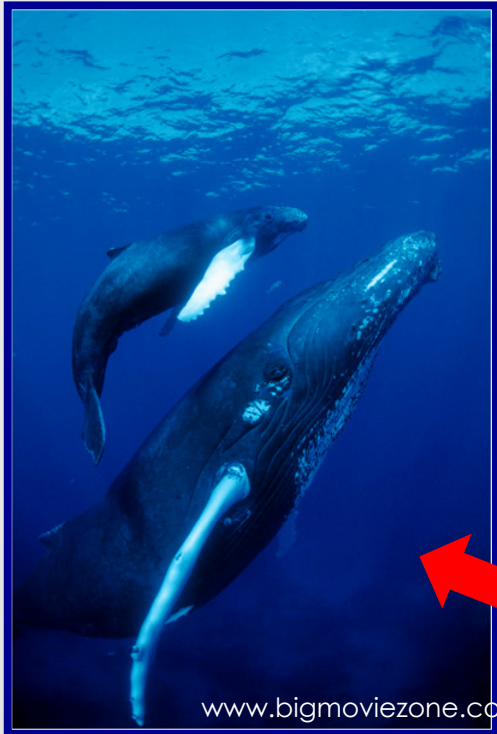
**Zooplankton
(Pteropod)**



Coral



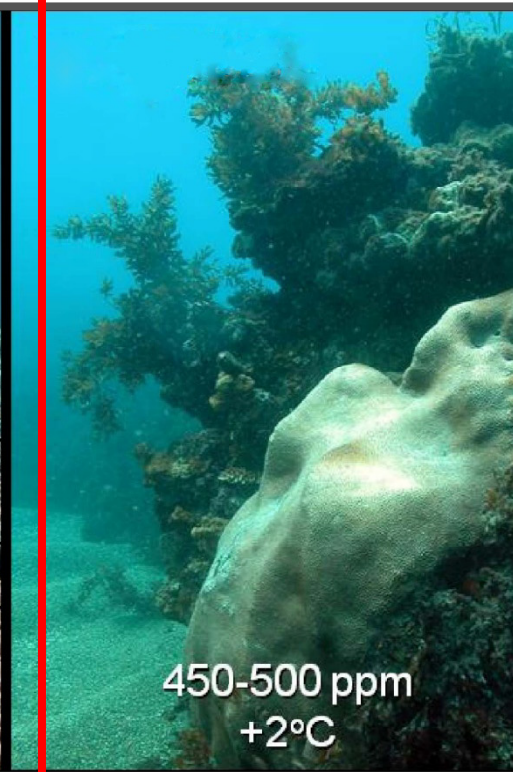
**Phytoplankton
(Coccolithophore)**





375 ppm
+1°C

<http://www.noaanews.noaa.gov>



450-500 ppm
+2°C



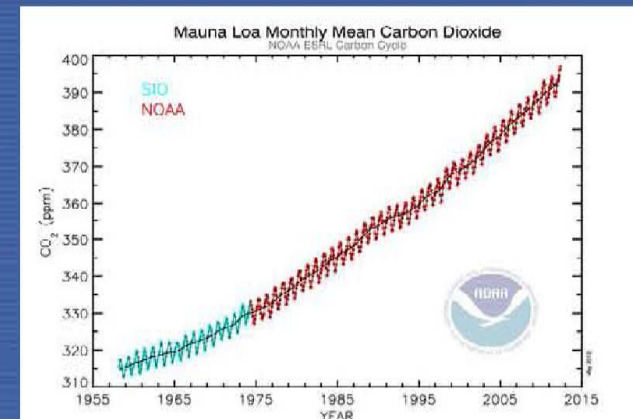
> 500 ppm
>+3°C

Ocean acidification

Ocean Acidification is a major, global environmental pressure due to increasing atmospheric CO₂ concentration in surface marine water.

It is predicted to have **major impact** on key marine ecosystems, including on biodiversity, safety and security of seafood resources and ecosystems services, especially in fragile ecosystems such as tropical coral reefs and polar regions.

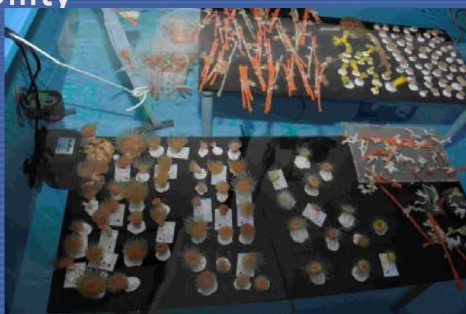
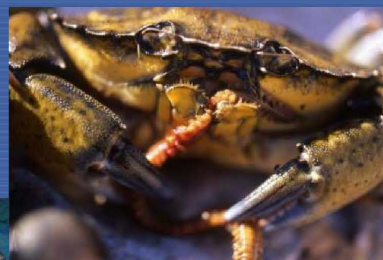
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Ocean acidification – Nuclear and isotopic applications

Unique tools to assess biological effects under projected $p\text{CO}_2$ scenarios, identify vulnerable organisms, evaluate potential coastal economic impacts (fisheries, aquaculture, ecosystem services), e.g.:

- Use of Ca-45 to assess growth and calcification rates
- Use of C-14 to assess primary production of marine phytoplankton
- Use of radio-tracers to assess change in pollutant availability



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