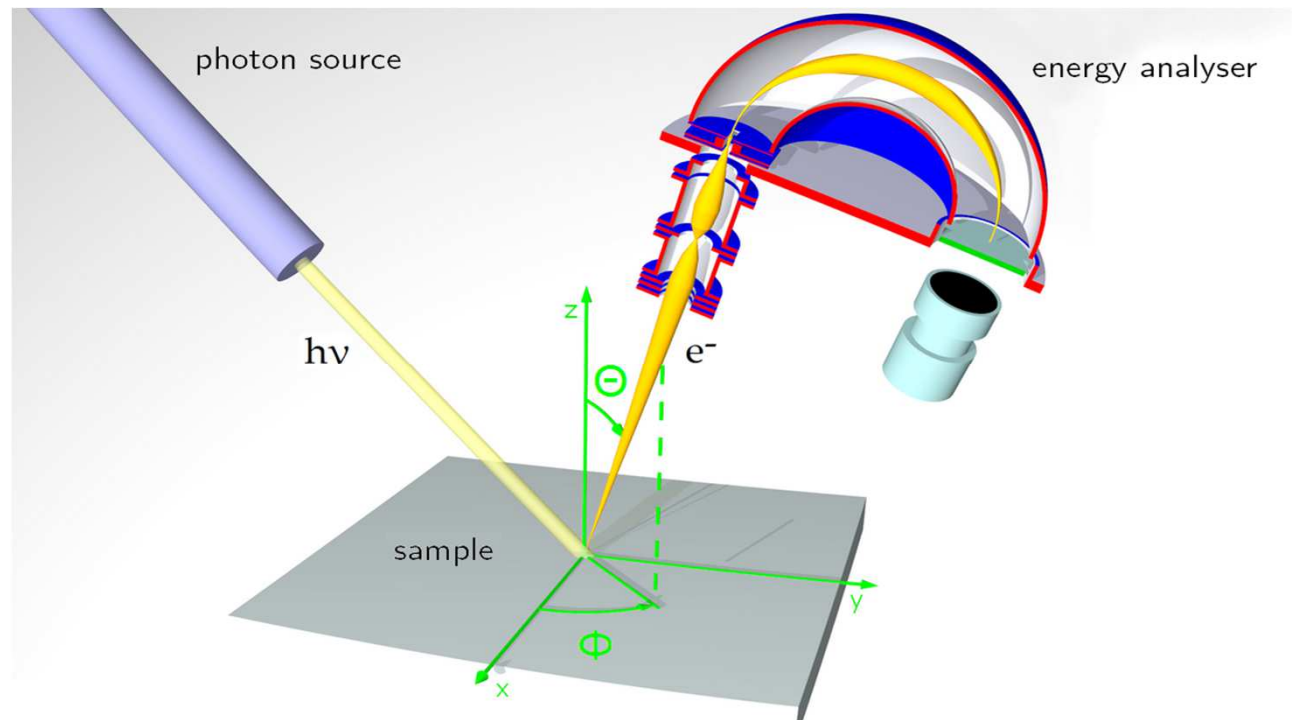
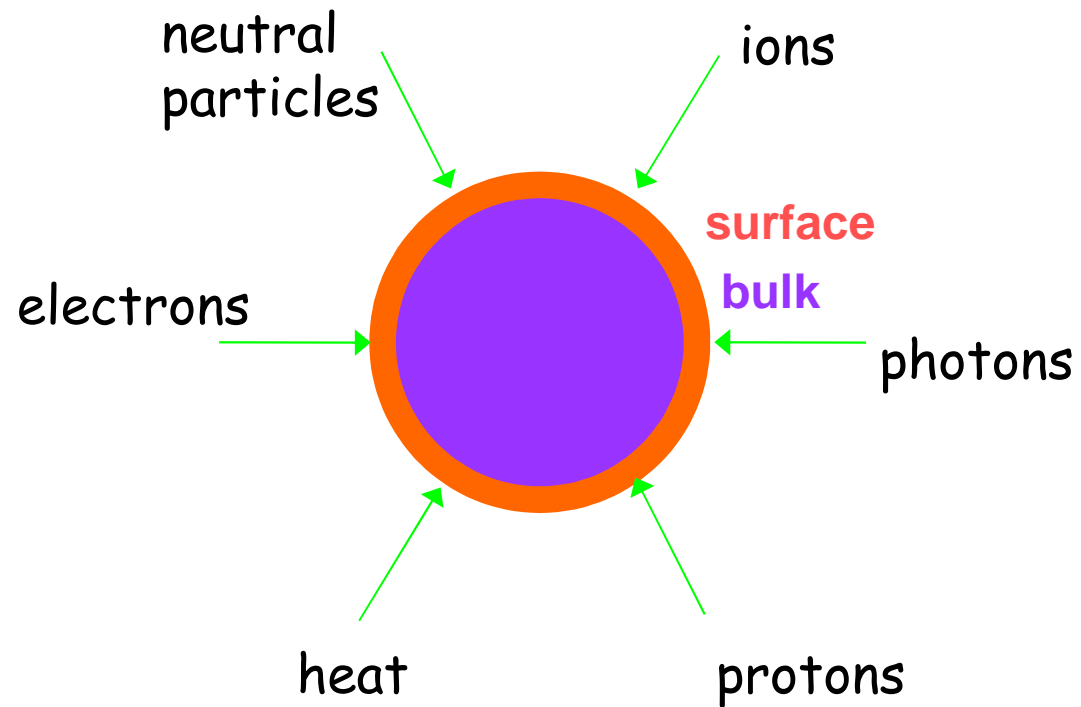


XPS SPECTROSCOPY



Rosanna Larciprete
CNR-ISC and LNF-INFN

Surface science



INFORMATION

Chemical

Electronic

Structural

Magnetic

Transport

Thermic

.....

Surface science

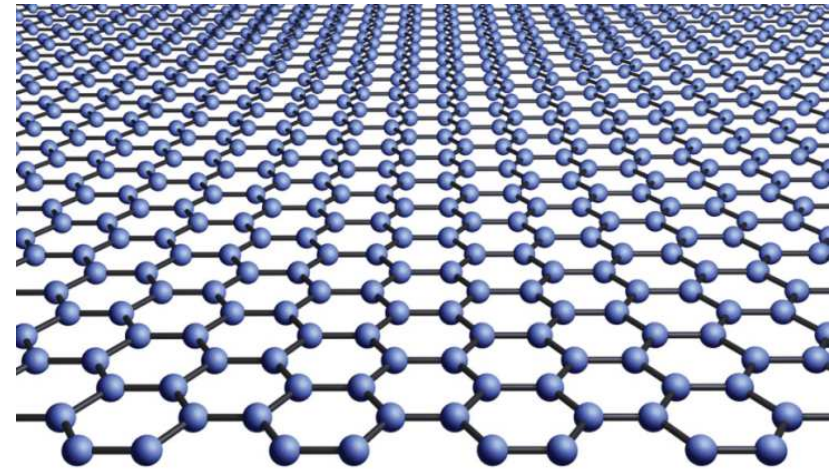
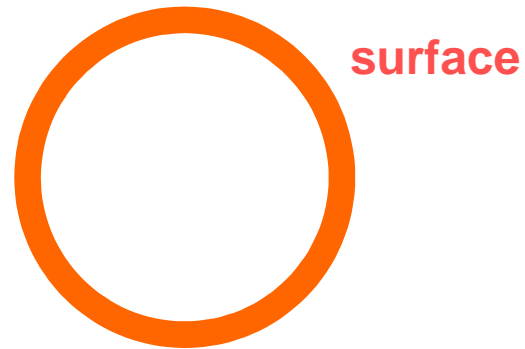
Chemical composition of a surface

Surface reactions (oxidation, catalysis)

Surface diffusion

Surface states

Epitaxy, thin film growth, self-assembly of nanostructures



graphene

Surface science

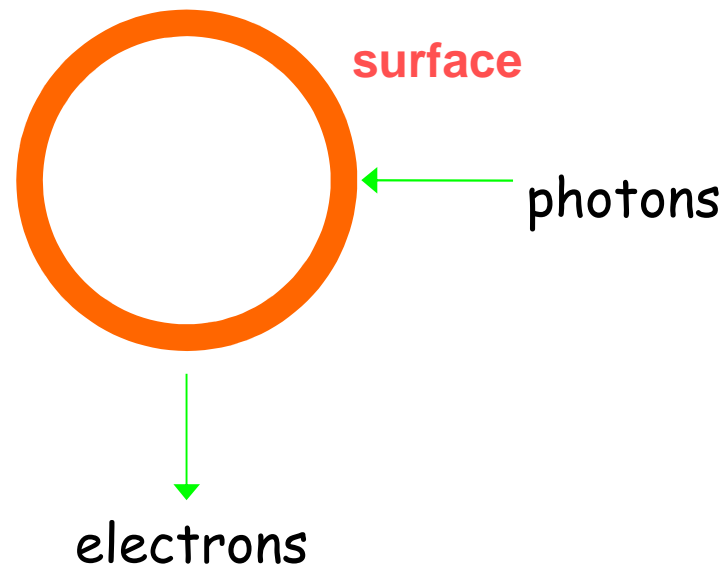
Chemical composition of a surface

Surface reactions (oxidation, catalysis)

Surface diffusion

Surface states

Epitaxy, thin film growth, self-assembly of nanostructures

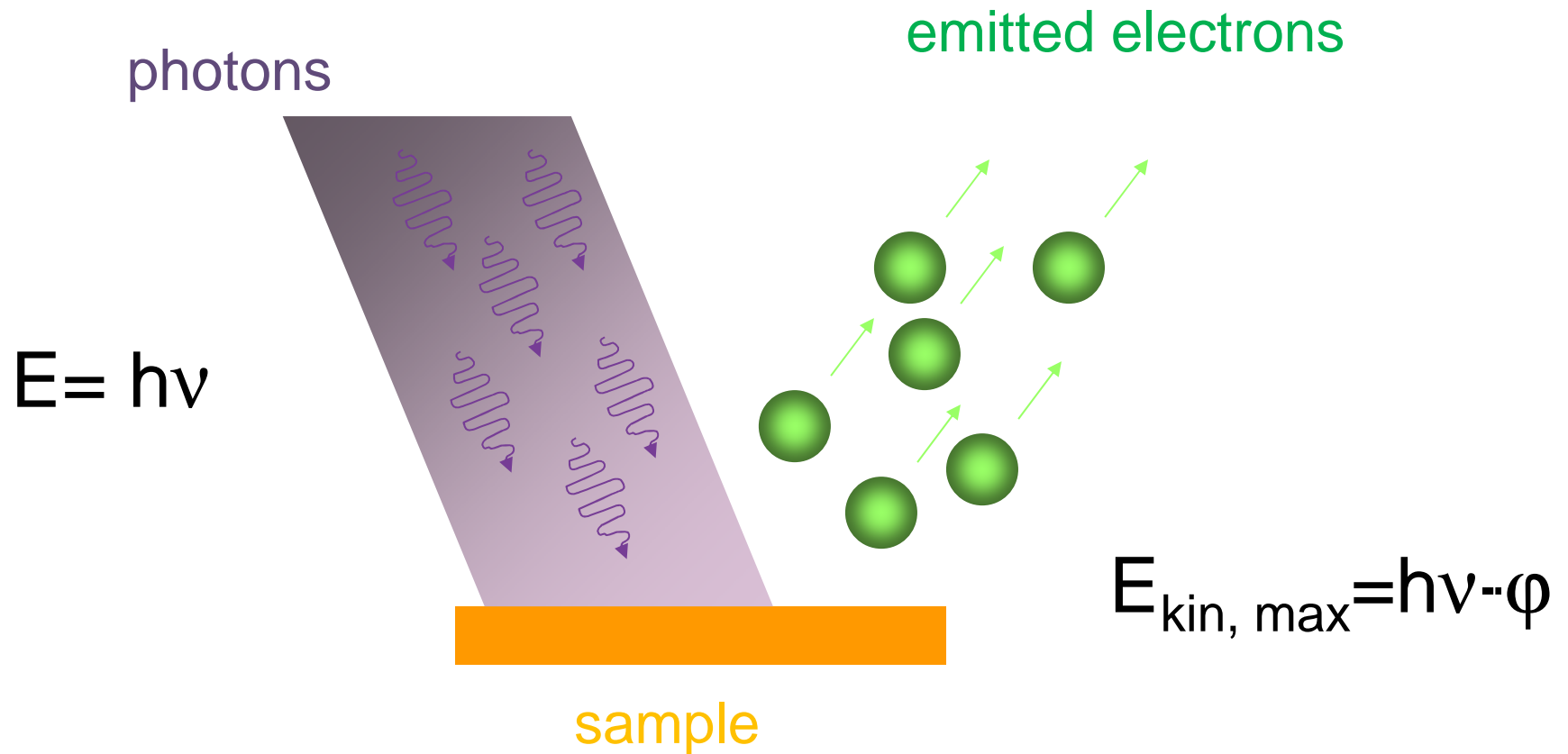


INFORMATION

Chemical

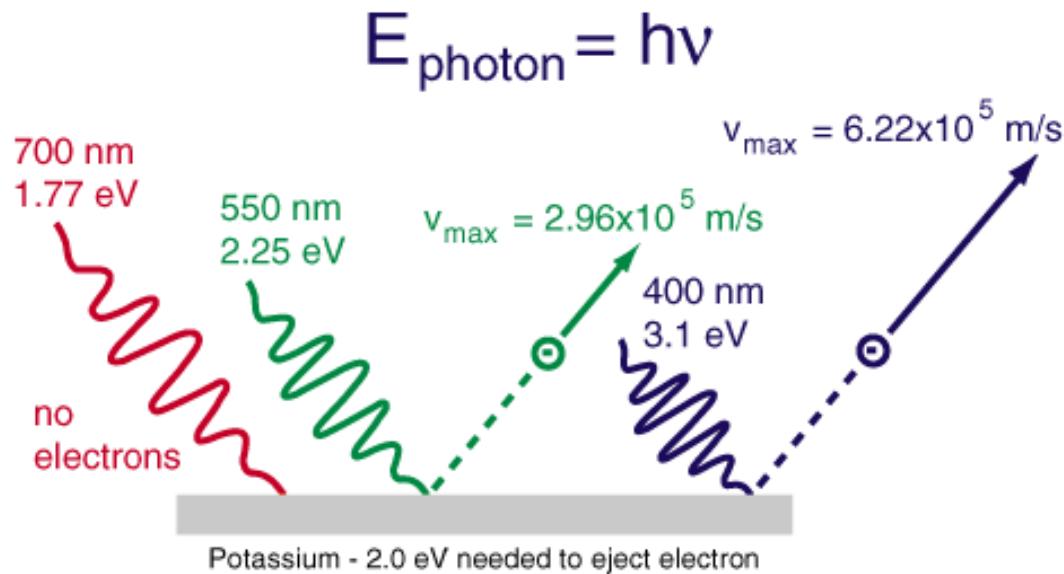
Electronic

Photoelectric effect



E_{kin} does not depend on the intensity of the photon beam

Photoelectric effect



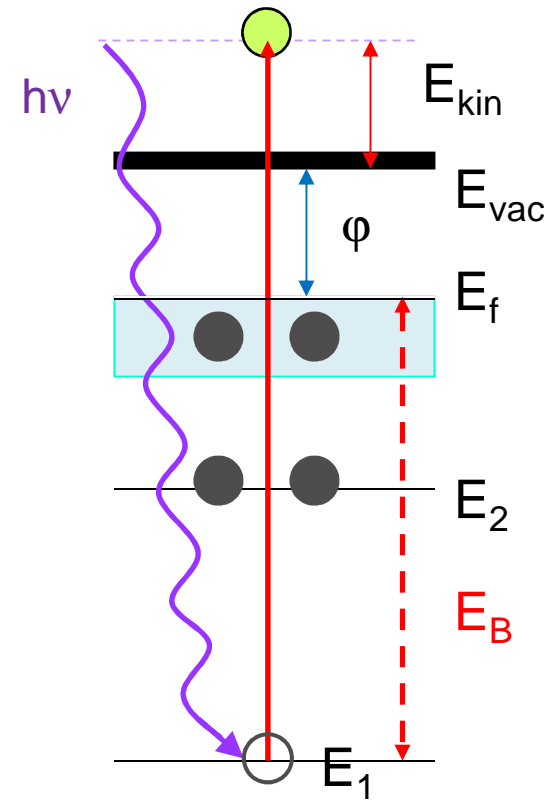
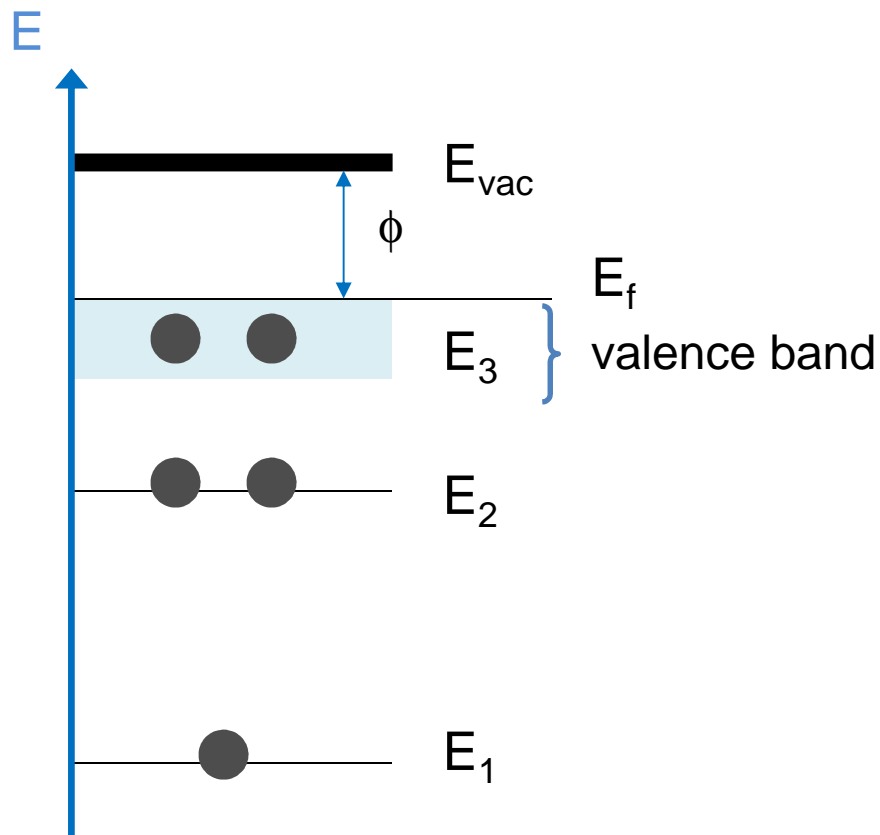
Photoelectric effect

The **work function ϕ** is the energy needed to to remove an electron from a solid to a point in the vacuum immediately outside the solid surface

Element Work Function(eV)

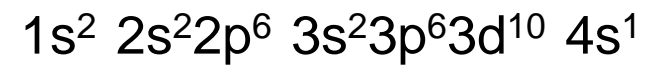
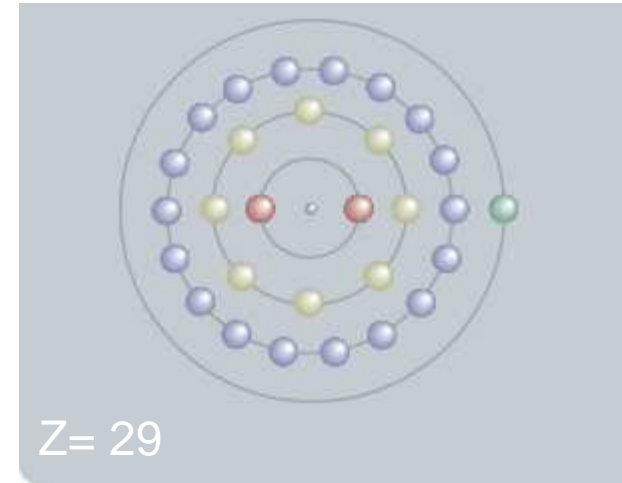
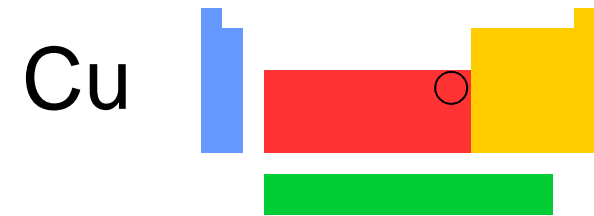
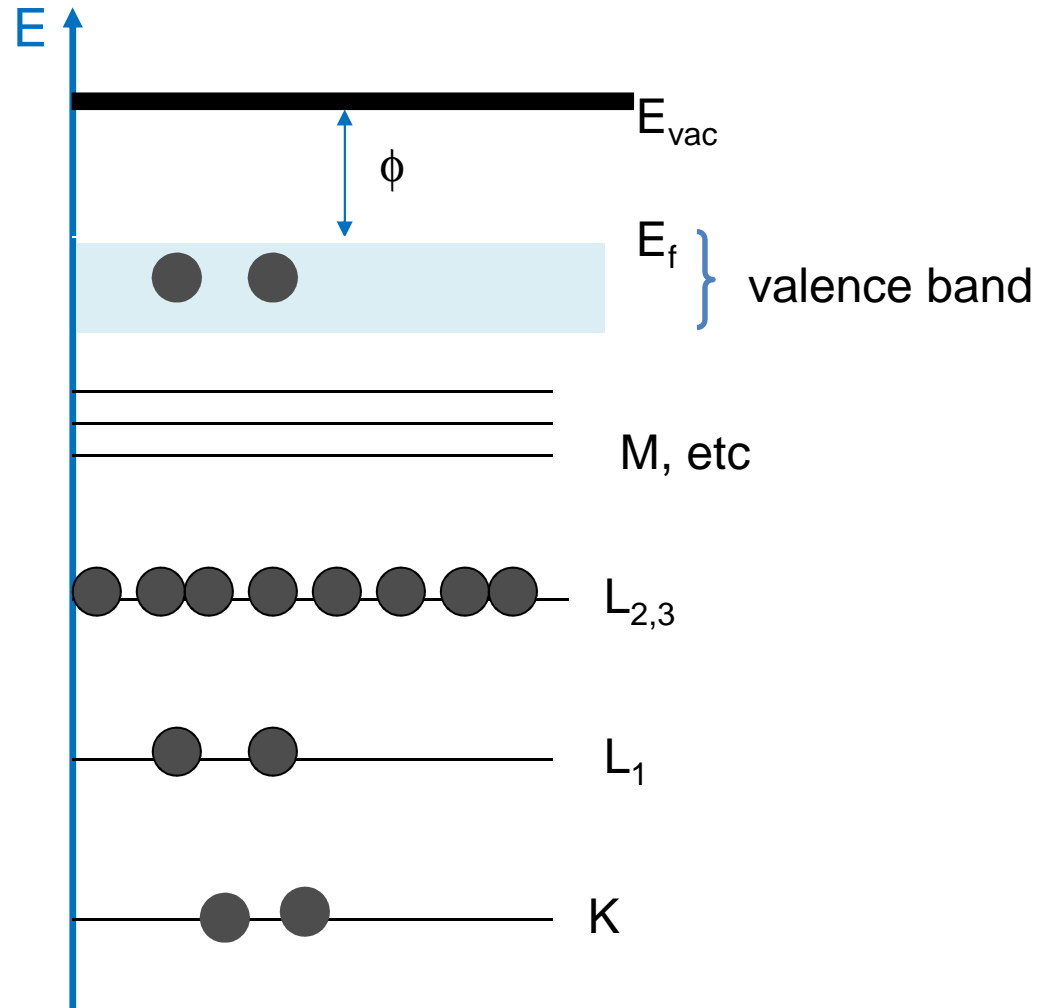
Aluminum	4.08	
Beryllium	5.0	
Cadmium	4.07	
Calcium	2.9	
Carbon	4.81	
Cesium	2.1	
Cobalt	5.0	
Copper	4.7	
Gold	5.1	
Iron	4.5	
Lead	4.14	
Magnesium		3.68
Mercury	4.5	
Nickel	5.01	
Niobium	4.3	
Potassium		2.3
Platinum	6.35	
Selenium	5.11	
Silver	4.73	
Sodium	2.28	
Uranium	3.6	
Zinc	4.3	

The photoemission process

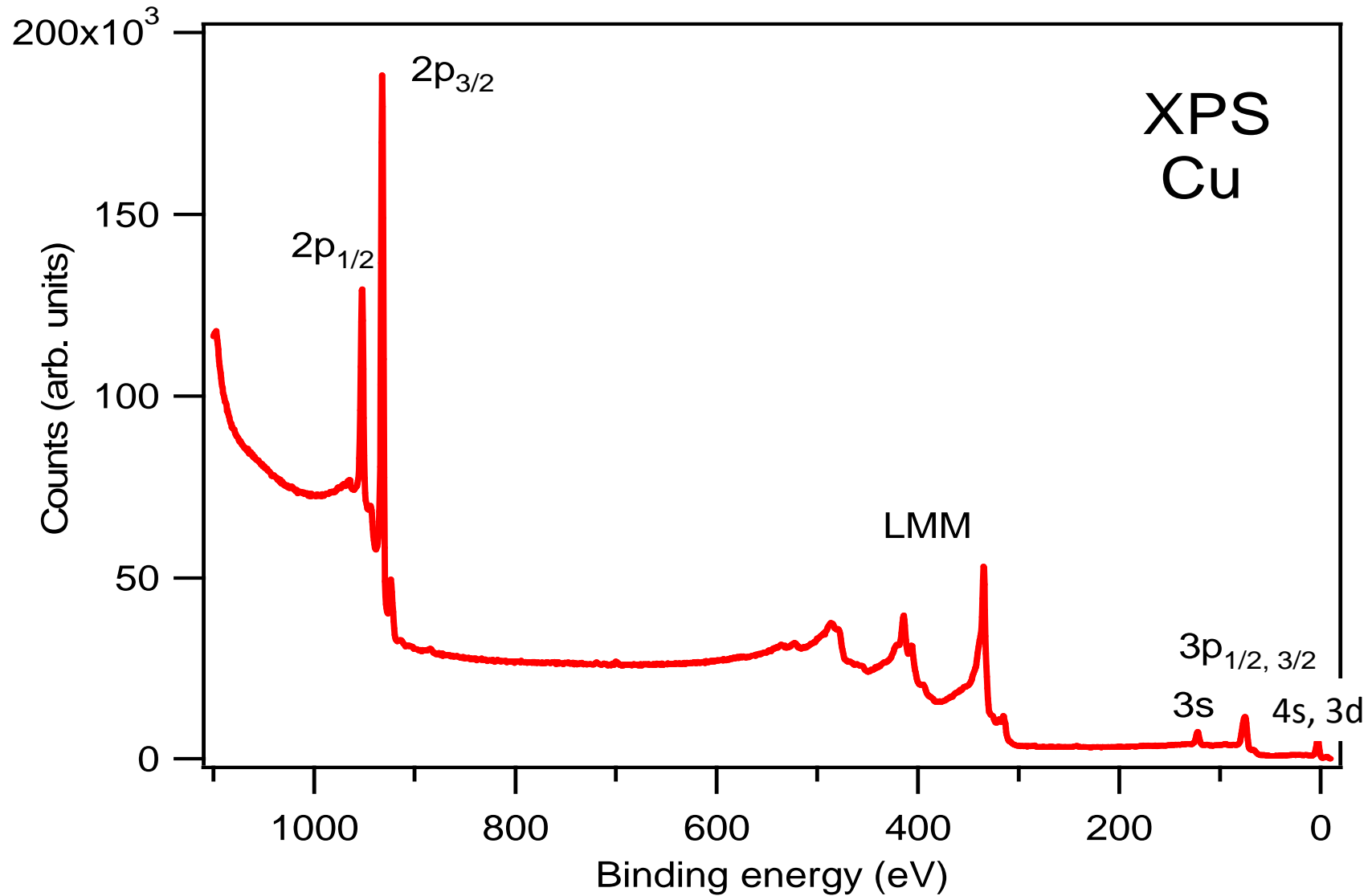


$$E_{Kin} = h\nu - E_B - \phi$$

Atomic energy levels

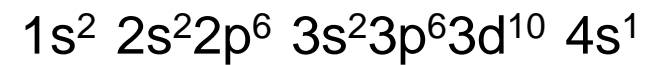
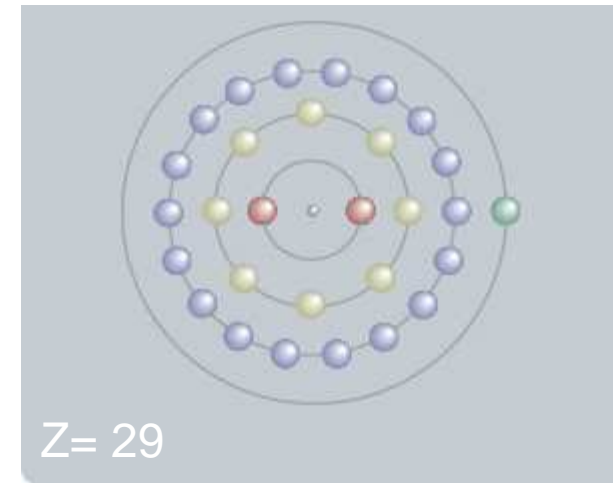
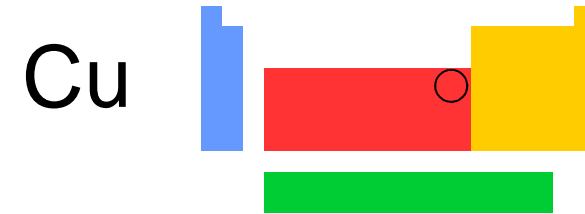
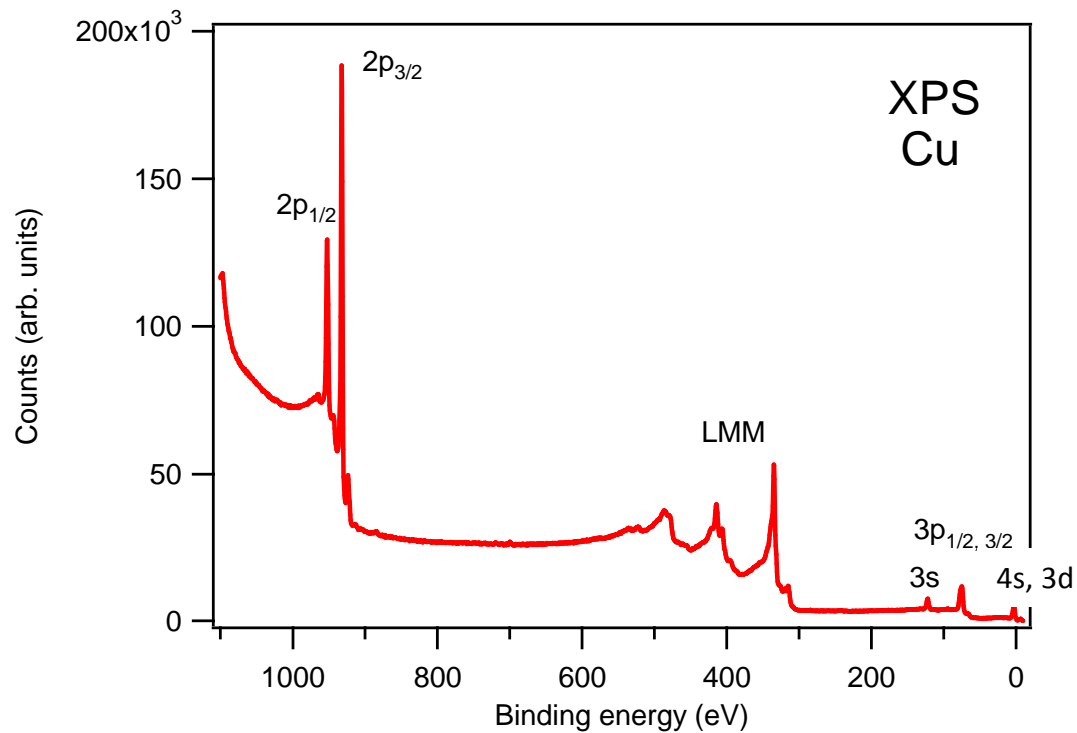


XPS spectrum

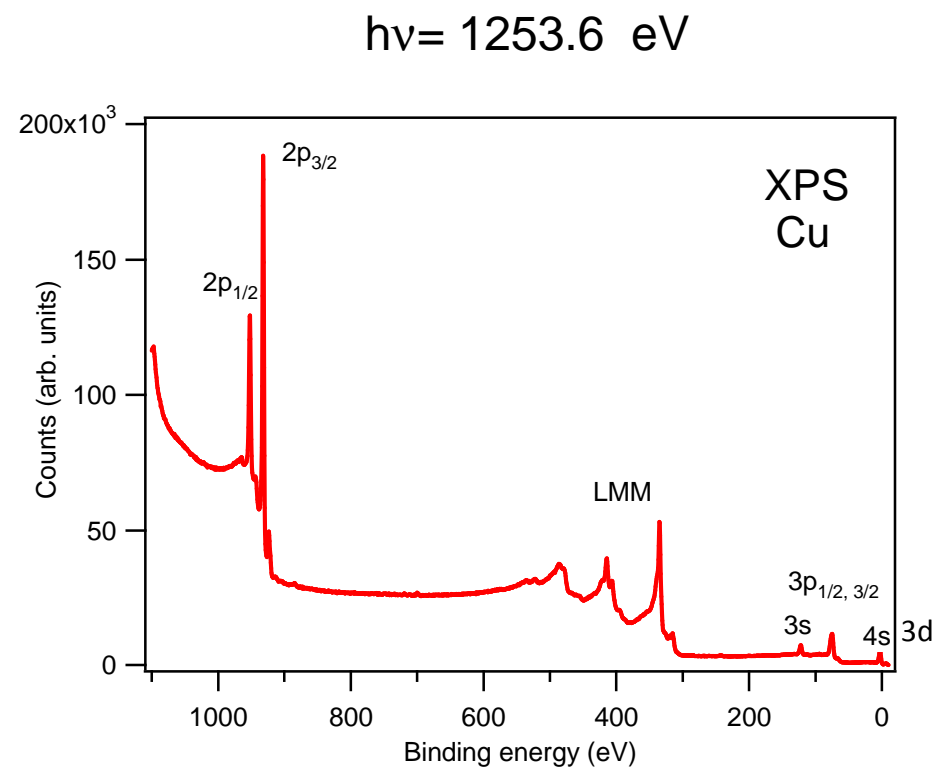
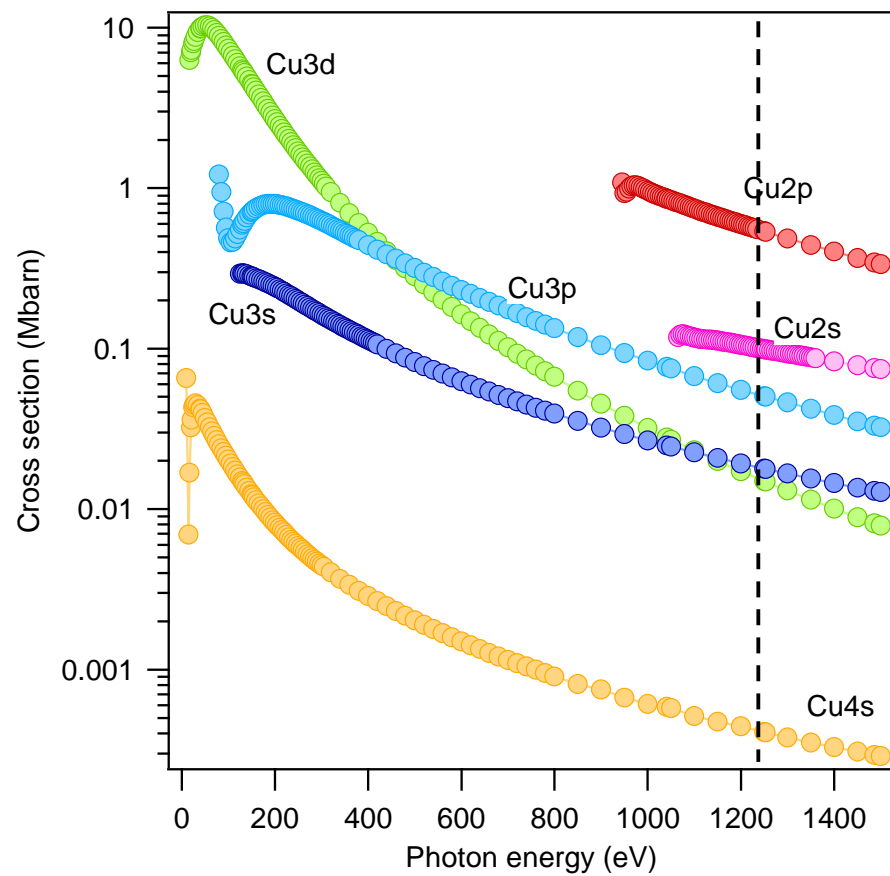


XPS spectrum

Label	Orbital	eV
K	<u>1s</u>	8979
L I	<u>2s</u>	1096.7
L II	<u>2p_{1/2}</u>	952.3
L III	<u>2p_{3/2}</u>	932.7
M I	<u>3s</u>	122.5
M II	<u>3p_{1/2}</u>	77.3
M III	<u>3p_{3/2}</u>	75.1



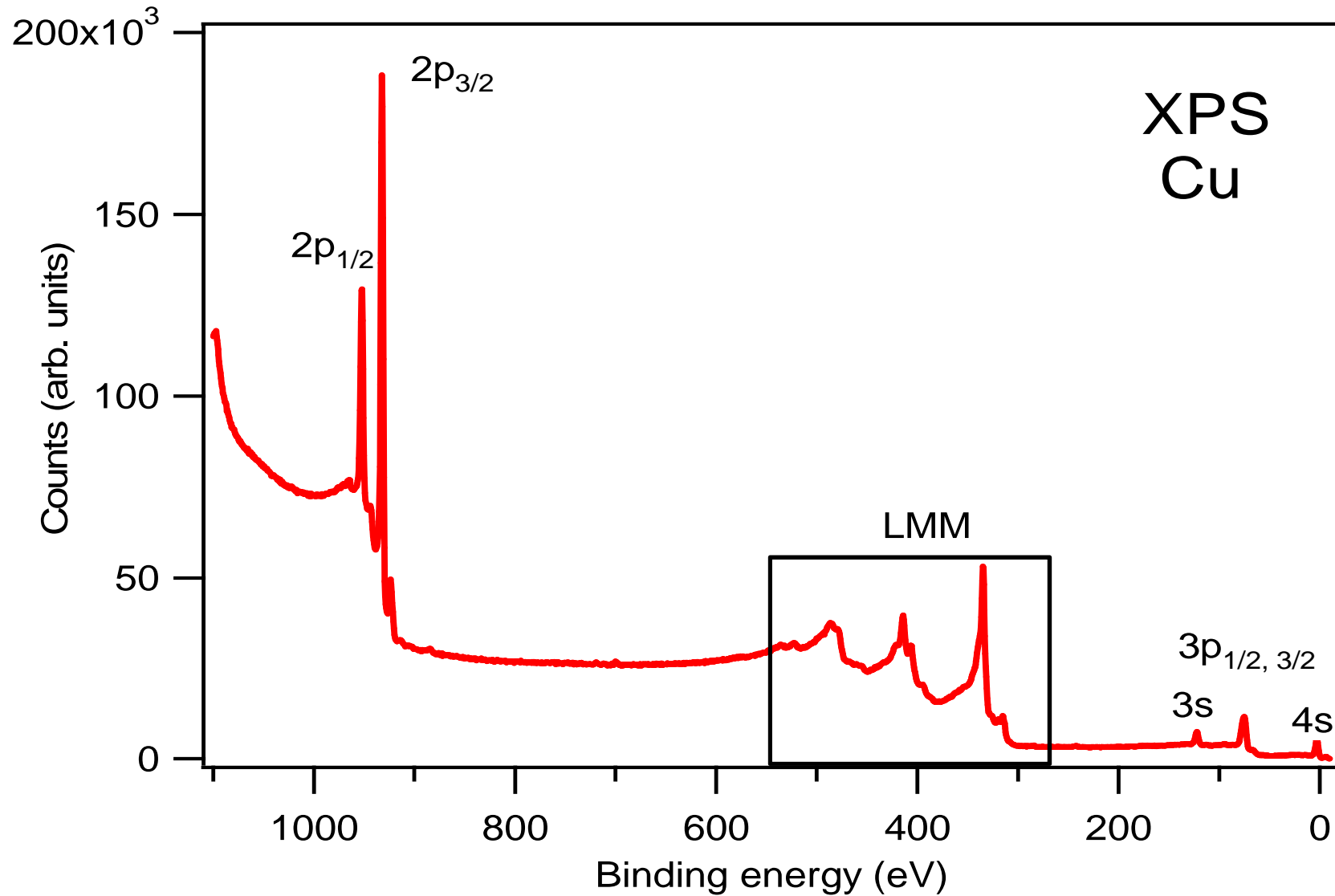
Photoionization cross section



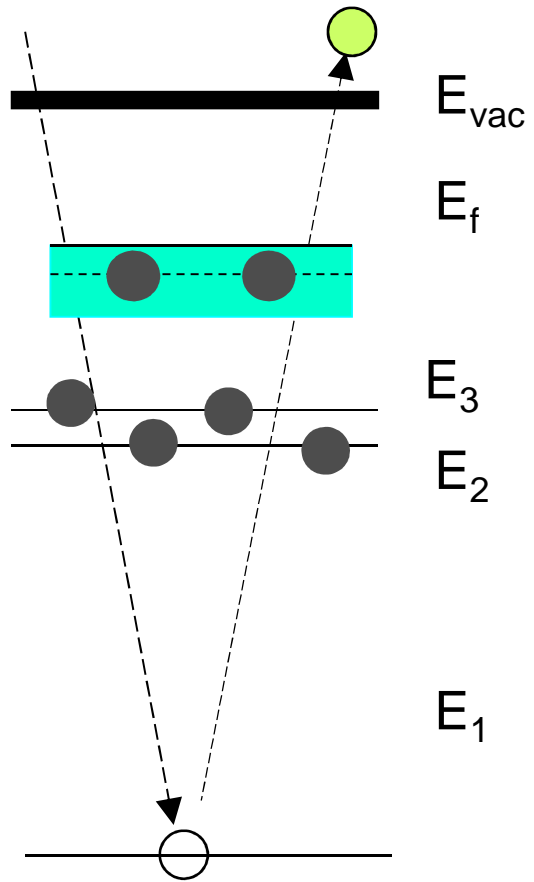
Core level binding energies

		K 1s	L-I 2s	L-II 2p1/2	L-III 2p3/2	M-I 3s	M-II 3p1/2	M-III 3p3/2	M-IV 3d3/2	M-V 3d5/2
1	H	13.6								
2	He	24.6*								
3	Li	54.7*								
4	Be	111.5*								
5	B	188*								
6	C	284.2*								
7	N	409.9*	37.3*							
8	O	543.1*	41.6*							
9	F	696.7*								
10	Ne	870.2*	48.5*	21.7*	21.6*					
11	Na	1070.8+	63.5+	30.4+	30.5*					
12	Mg	1303.0+	88.6*	49.6+	49.21					
13	Al	1559	117.8*	72.9*	72.5*					
14	Si	1839	149.7*b	99.8*	99.2*					
15	P	2145.5	189*	136*	135*					
16	S	2472	230.9	163.6*	162.5*					
17	Cl	2822	270*	202*	200*					
18	Ar	3205.9*	326.3*	250.6+	248.4*	29.3*	15.9*	15.7*		
19	K	3608.4*	378.6*	297.3*	294.6*	34.8*	18.3*	18.3*		
20	Ca	4038.5*	438.4+	349.7+	346.2+	44.3+	25.4+	25.4+		
21	Sc	4492	498.0*	403.6*	398.7*	51.1*	28.3*	28.3*		
22	Ti	4966	560.9+	460.2+	453.8+	58.7+	32.6+	32.6+		
23	V	5465	626.7+	519.8+	512.1+	66.3+	37.2+	37.2+		
24	Cr	5989	696.0+	583.8+	574.1+	74.1+	42.2+	42.2+		
25	Mn	6539	769.1+	649.9+	638.7+	82.3+	47.2+	47.2+		
26	Fe	7112	844.6+	719.9+	706.8+	91.3+	52.7+	52.7+		
27	Co	7709	925.1+	793.2+	778.1+	101.0+	58.9+	59.9+		
28	Ni	8333	1008.6+	870.0+	852.7+	110.8+	68.0+	66.2+		
29	Cu	8979	1096.7+	952.3+	932.7	122.5+	77.3+	75.1+		
30	Zn	9659	1196.2*	1044.9*	1021.8*	139.8*	91.4*	88.6*	10.2*	10.1*
31	Ga	10367	1299.0*b	1143.2+	1116.4+	159.51	103.5+	100.0+	18.7+	18.7+
32	Ge	11103	1414.6*b	1248.1*b	1217.0*b	180.1*	124.9*	120.8*	29.8*	29.2*
33	As	11867	1527.0*b	1359.1*b	1323.6*b	204.7*	146.2*	141.2*	41.7*	41.7*
34	Se	12658	1652.0*b	1474.3*b	1433.9*b	229.6*	166.5*	160.7*	55.5*	54.6*
35	Br	13474	1782*	1596*	1550*	257*	189*	182*	70*	69*

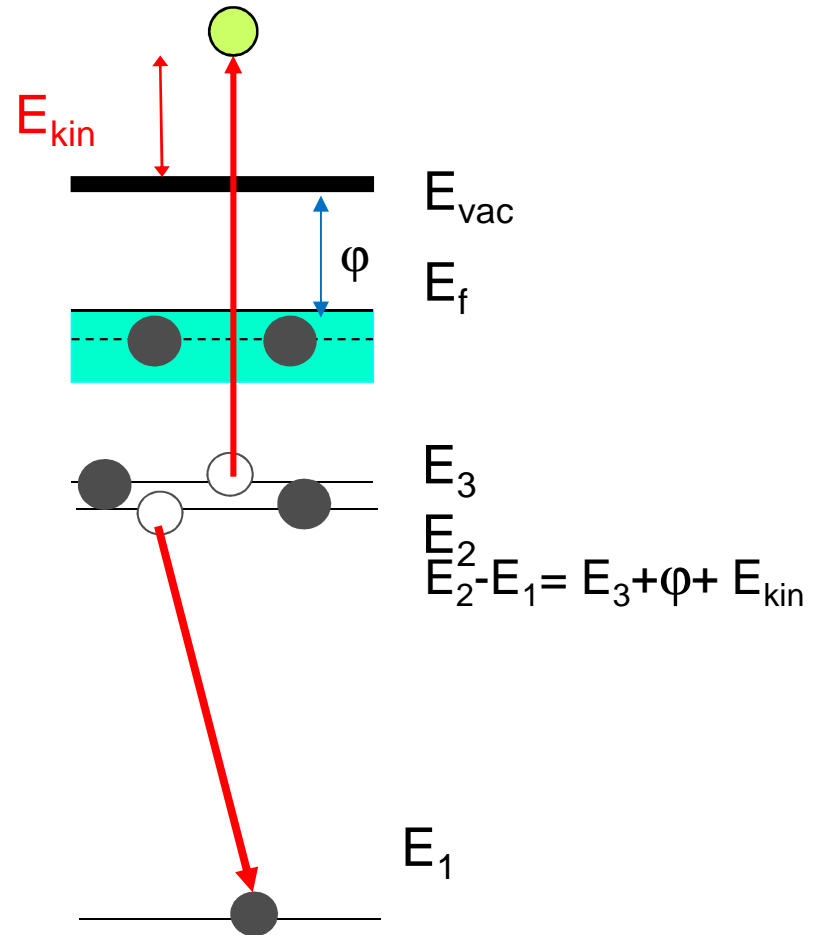
XPS spectrum



Auger process



Auger ($C_1C_2C_3$)

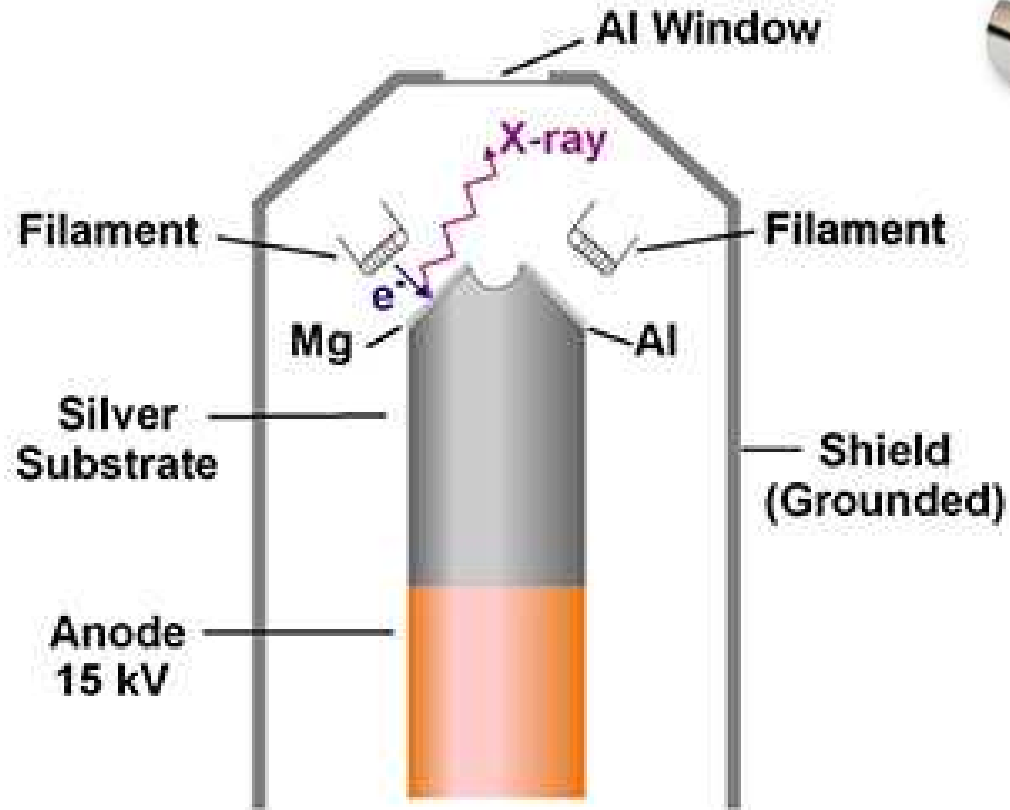


$$E_{kin} = E_2 - E_1 - E_3 - \phi$$

What do we need?

- **Ultra High Vacuum**
- **X-ray source**
- **Electron energy analyser**
- **Data acquisition system**

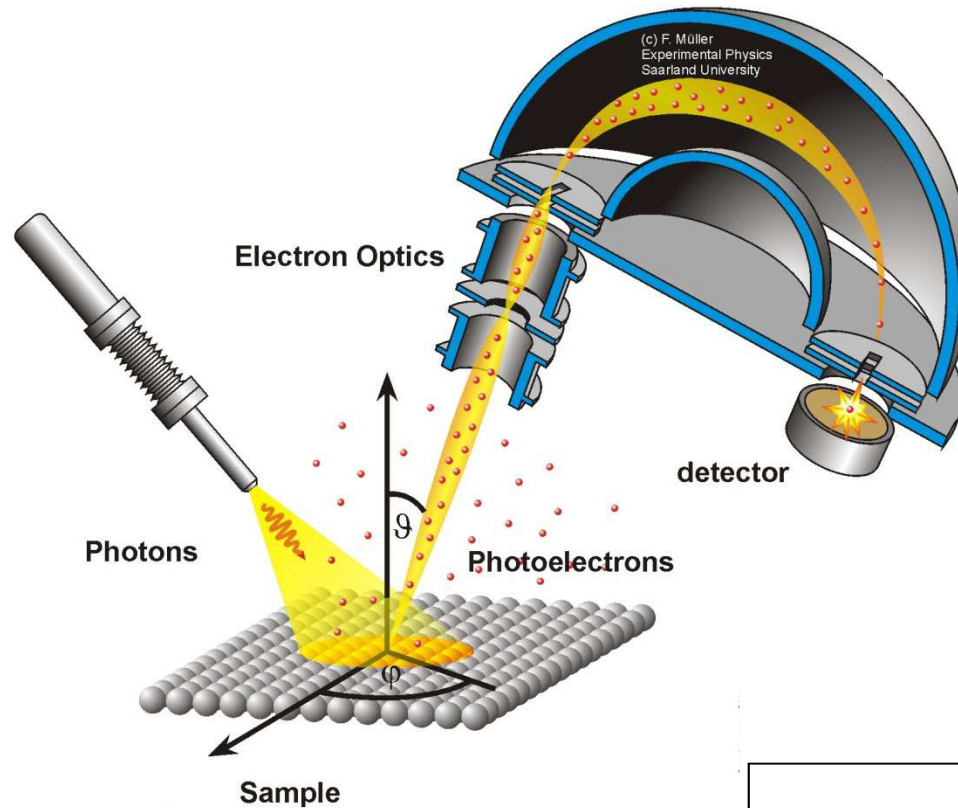
X-ray source



Mg $K\alpha$ 1253.6 eV

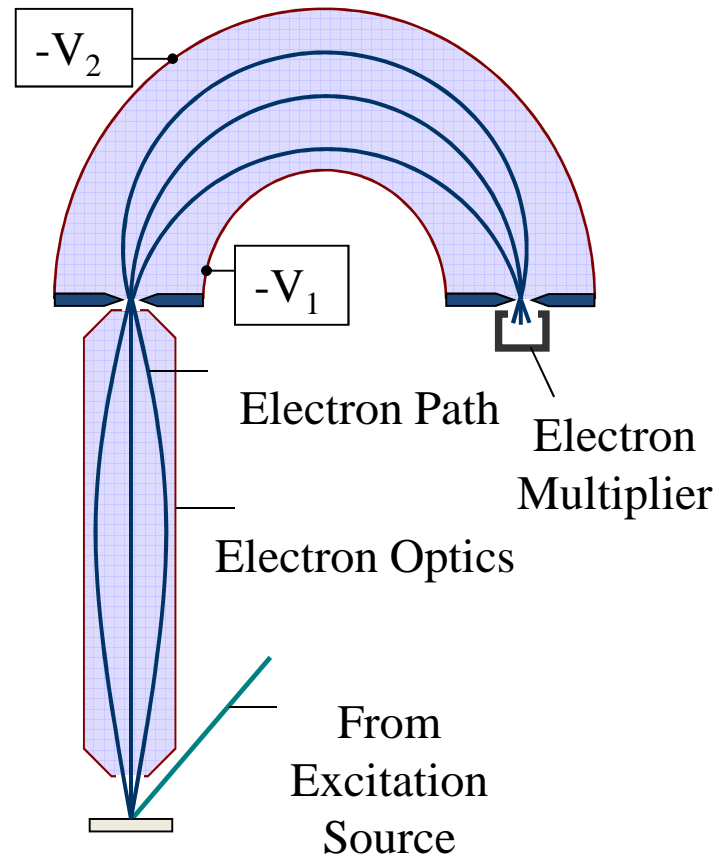
Al $K\alpha$ 1486.7 eV

Electron energy analyser

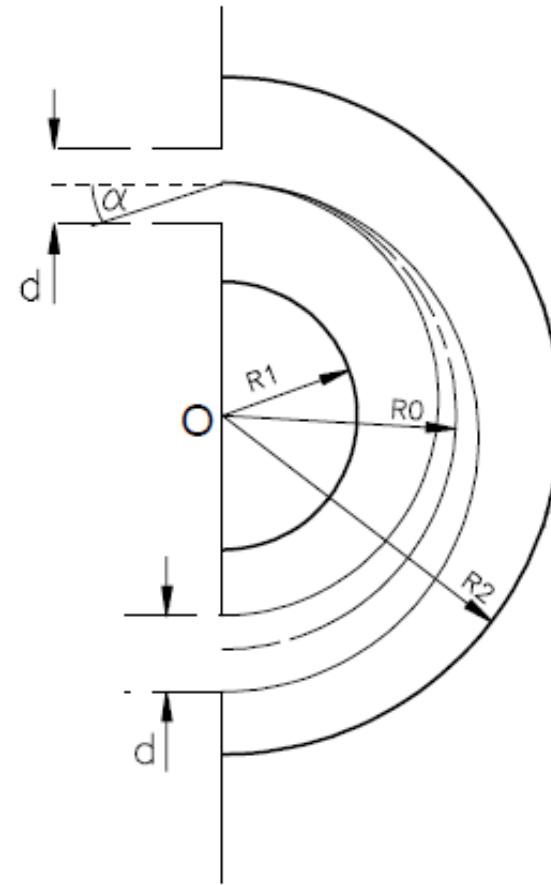
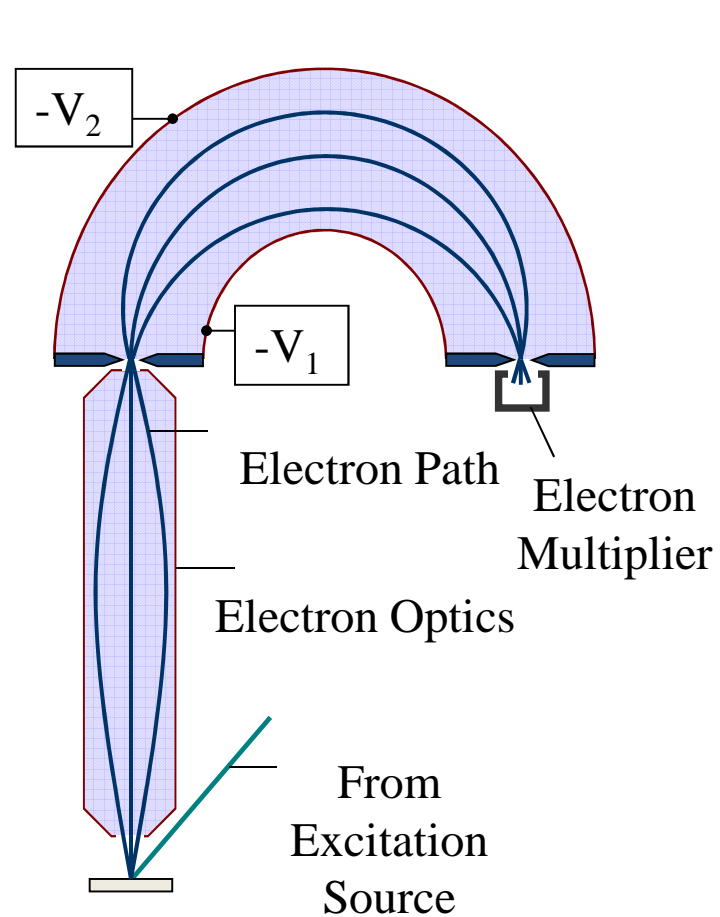


$$E_B = h\nu - E_{\text{Kin}} - \varphi$$

Electron energy analyser

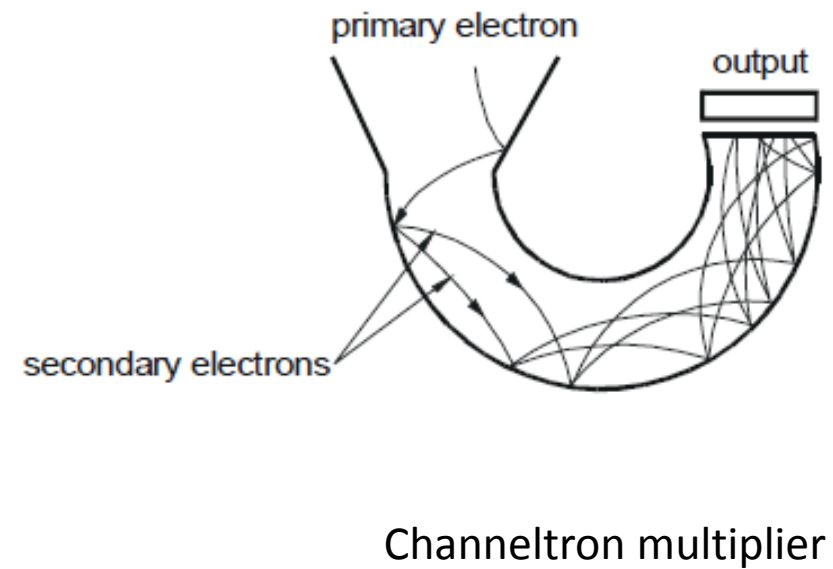
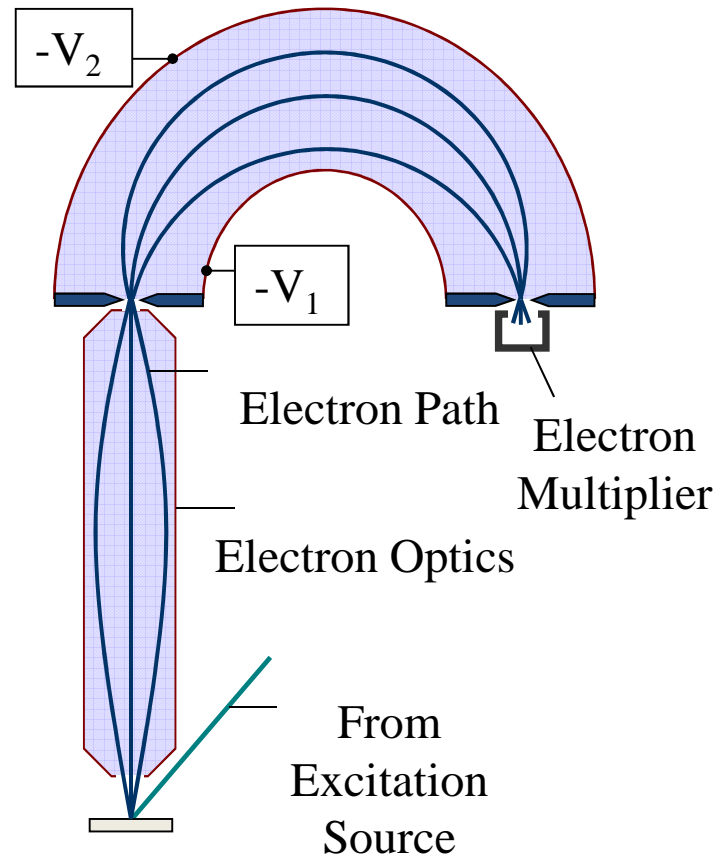


Electron energy analyser

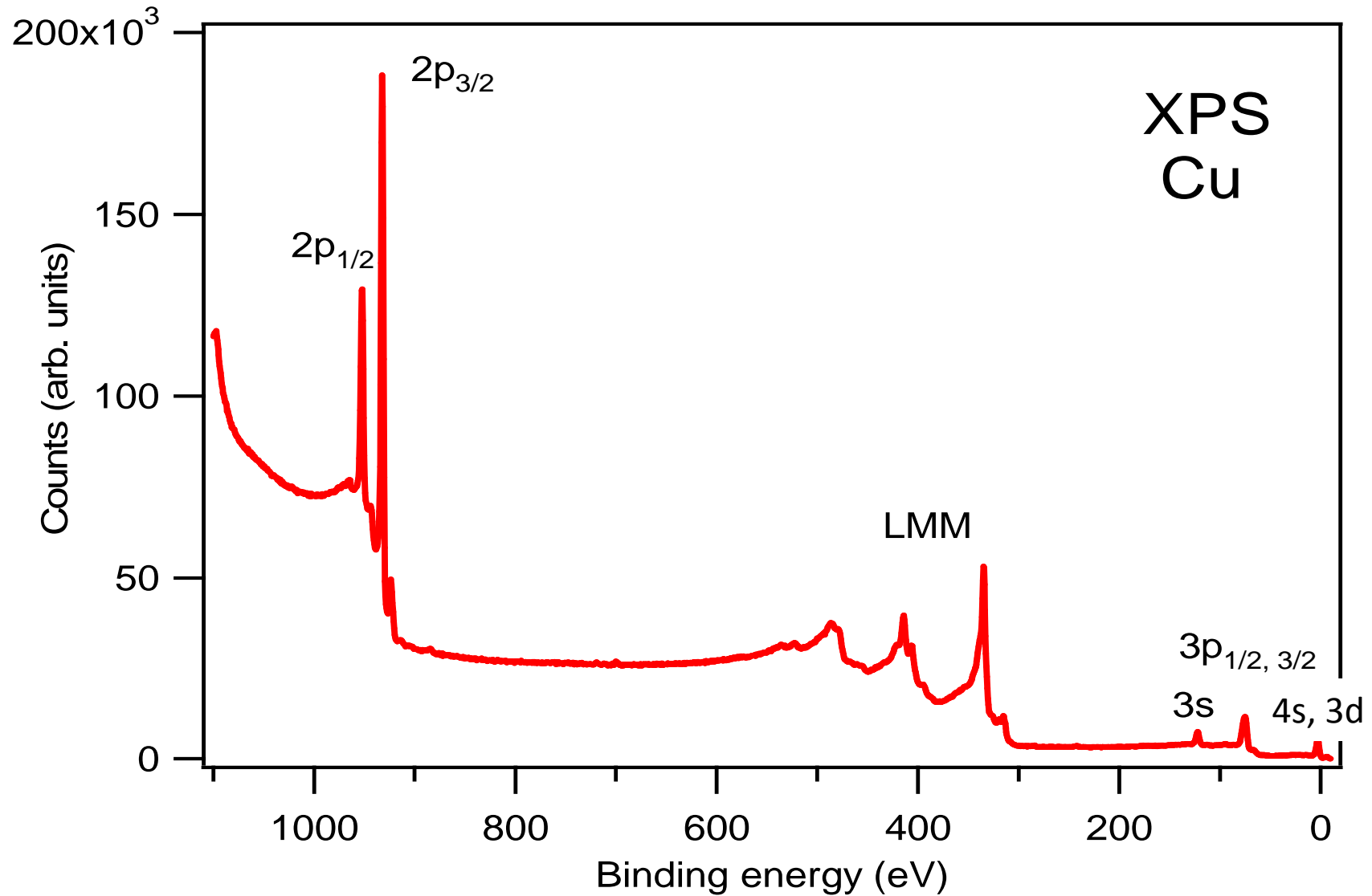


$$eV_0 = E_p \cdot \left(\frac{R_2}{R_1} - \frac{R_1}{R_2} \right)$$

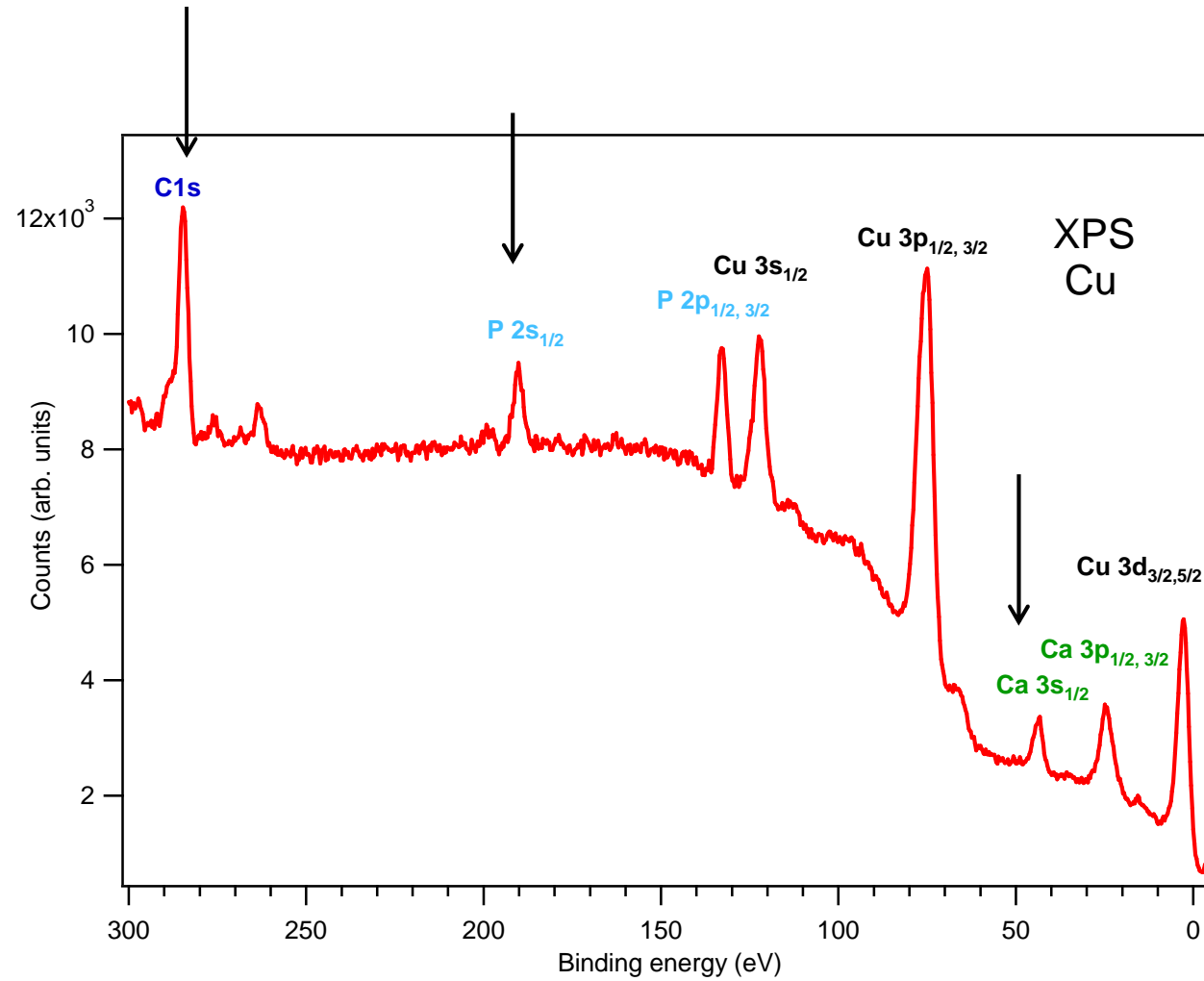
Electron energy analyser



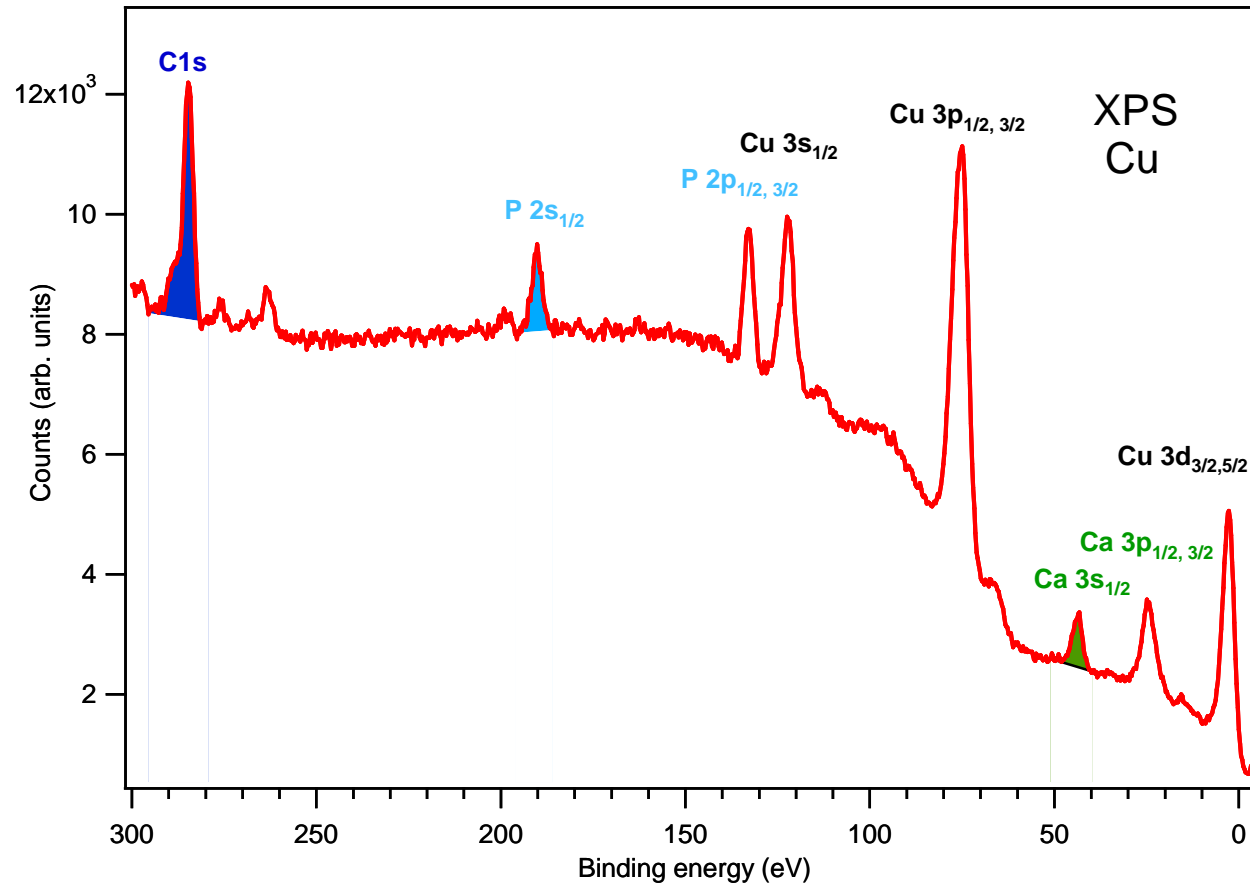
XPS spectrum



Quantitative analysis



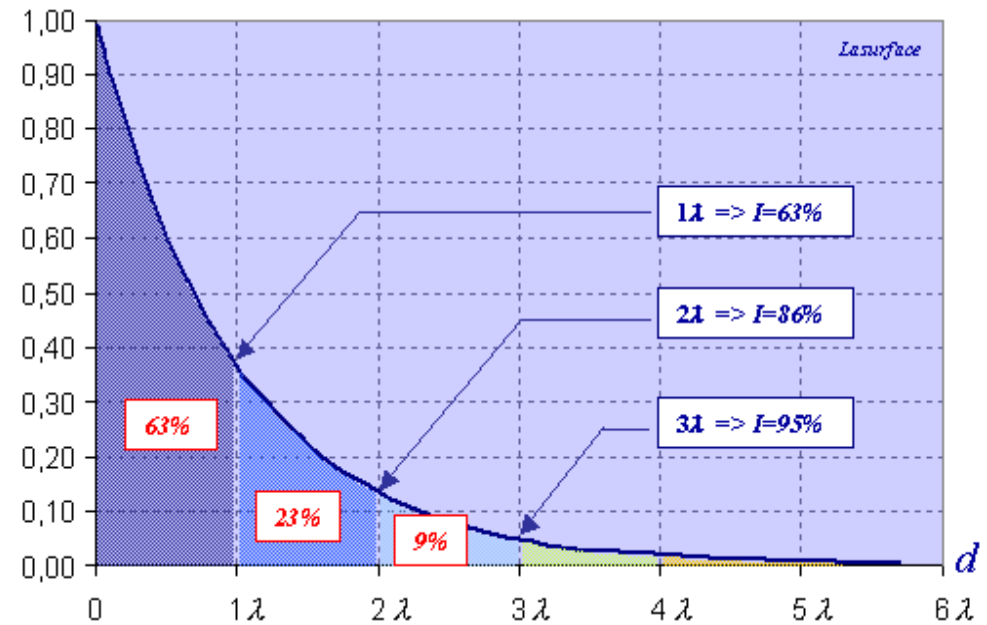
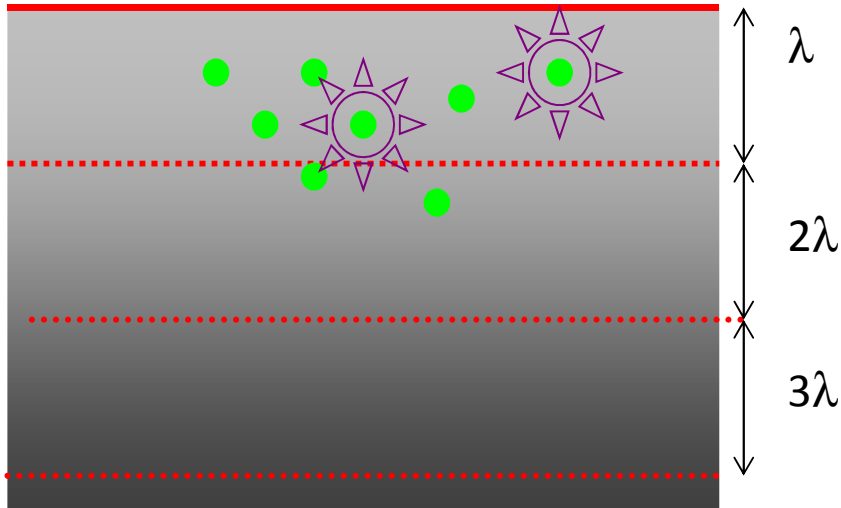
Quantitative analysis



Photoionization Cross sections

C 1s	0.02
Ca 3s	0.0068
P 2s	0.023

Electron escape depth

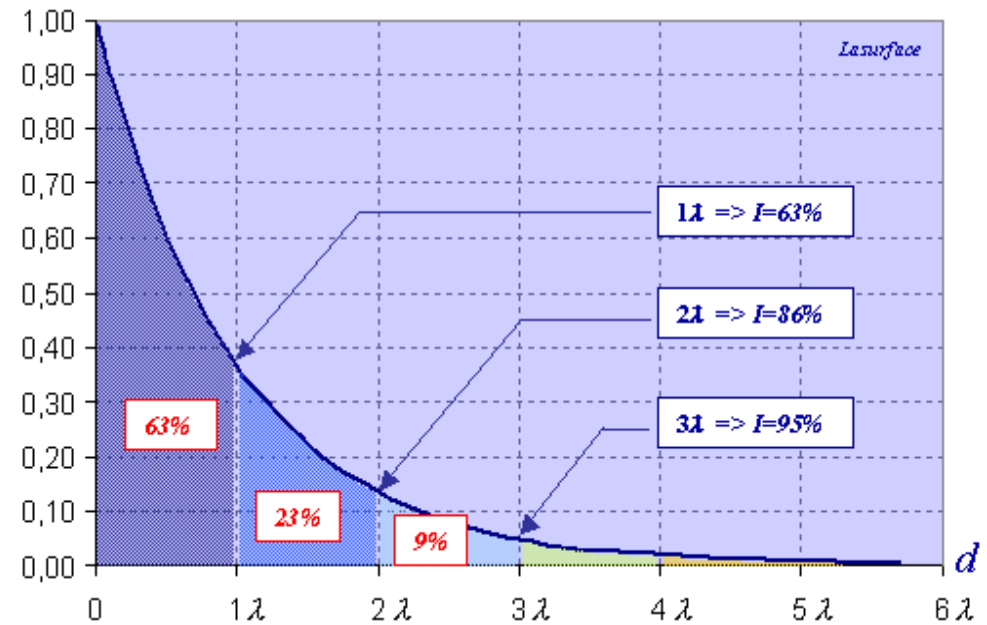
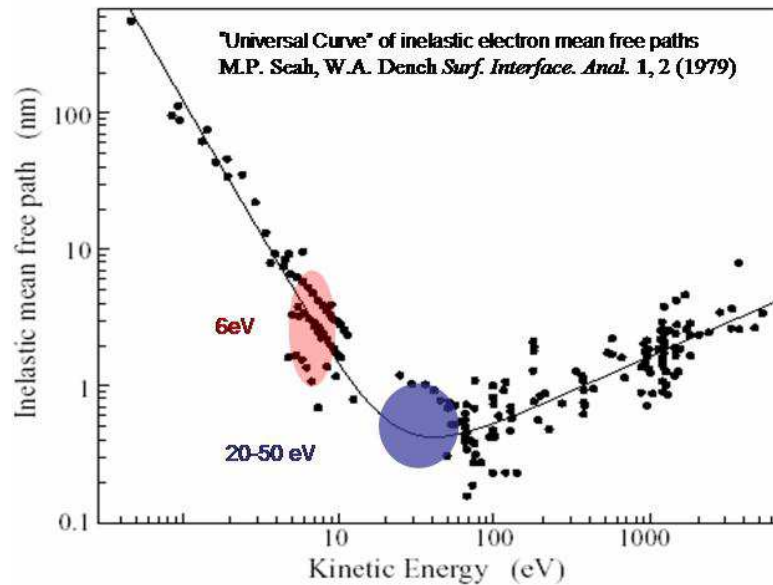


$$P(d) = \exp(-d/\lambda)$$

λ = inelastic electron mean free path

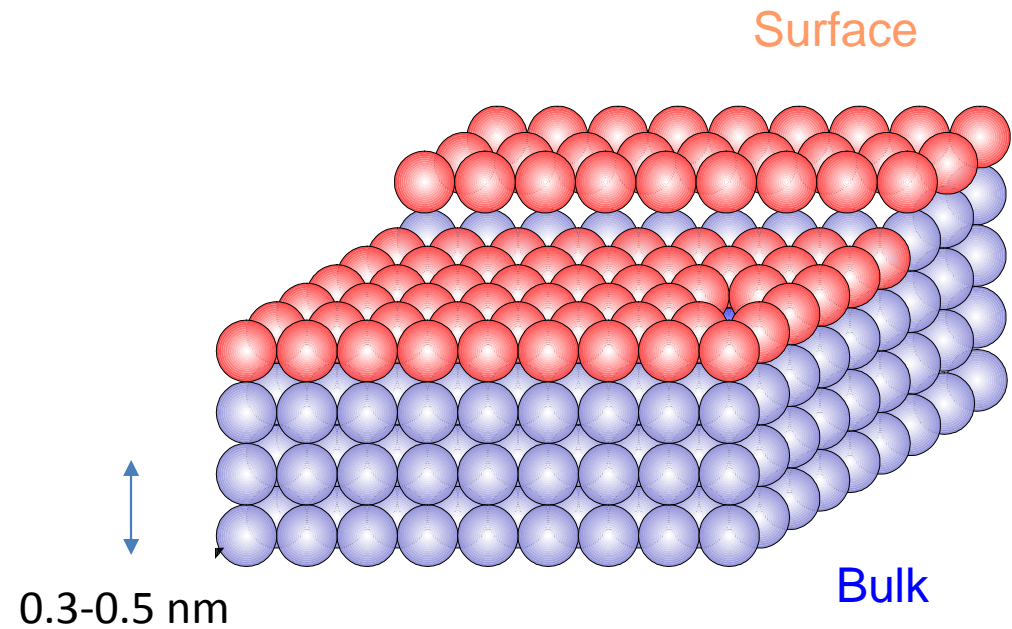
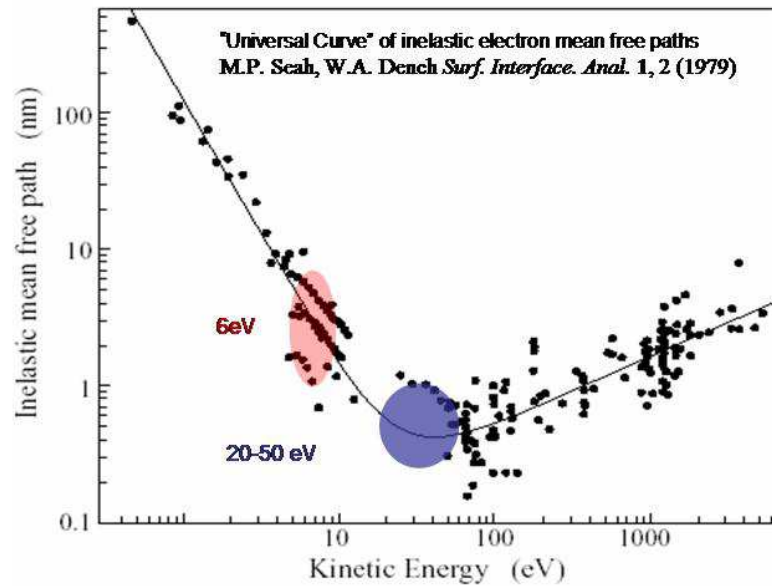
λ depends on the electron kinetic energy

Electron escape depth

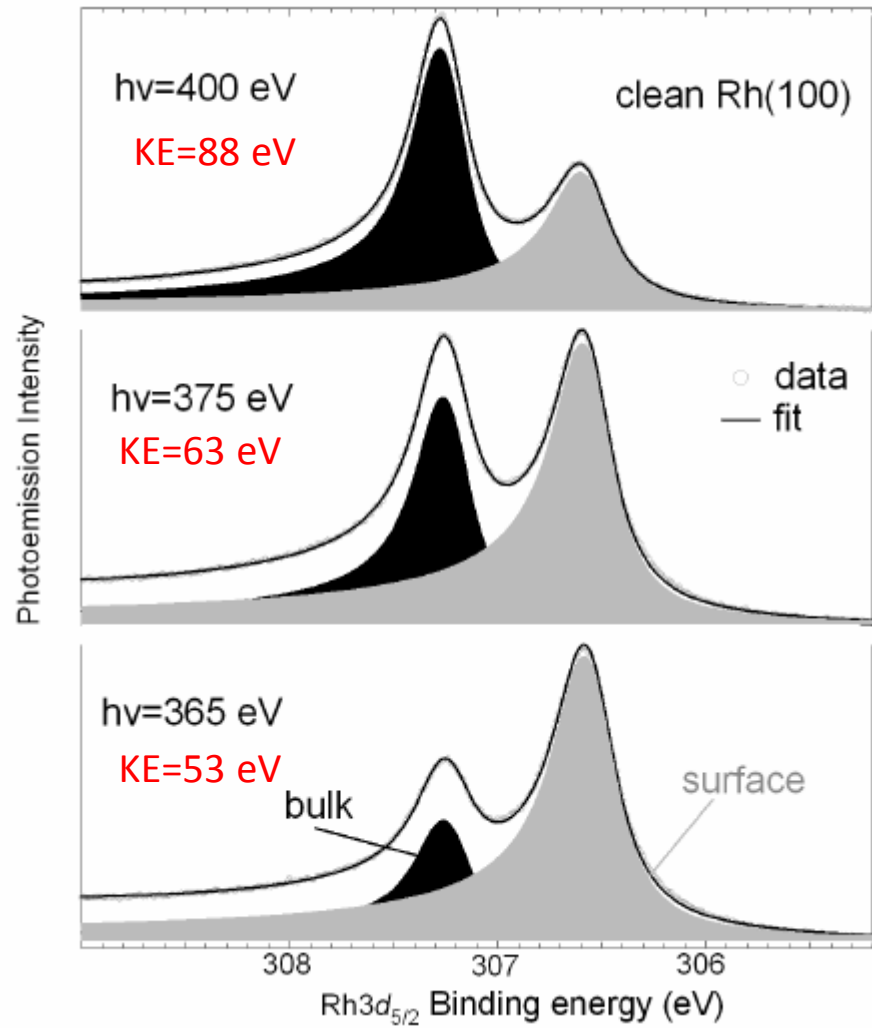
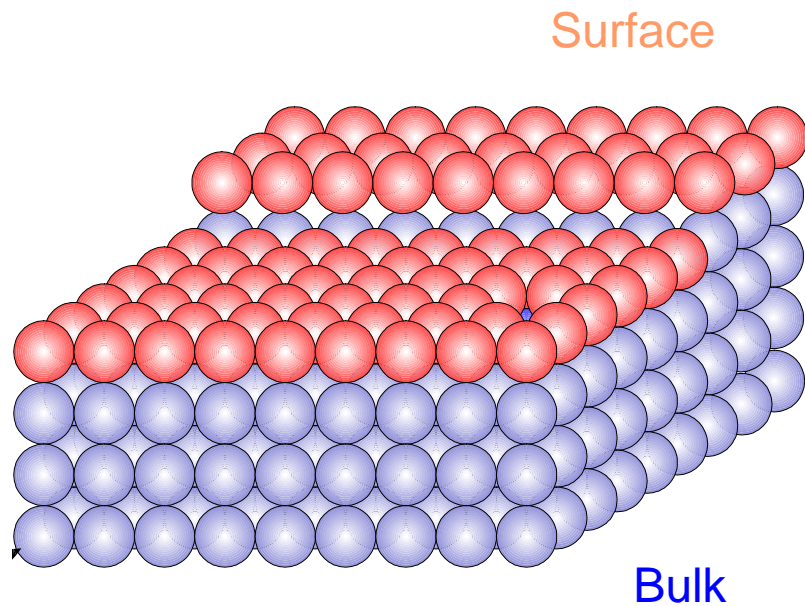


λ depends on the electron kinetic energy

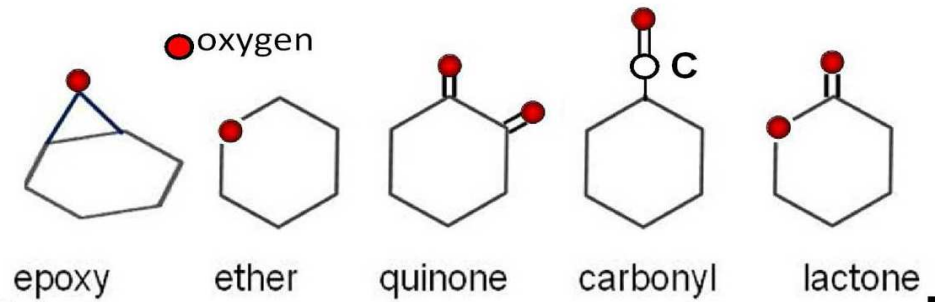
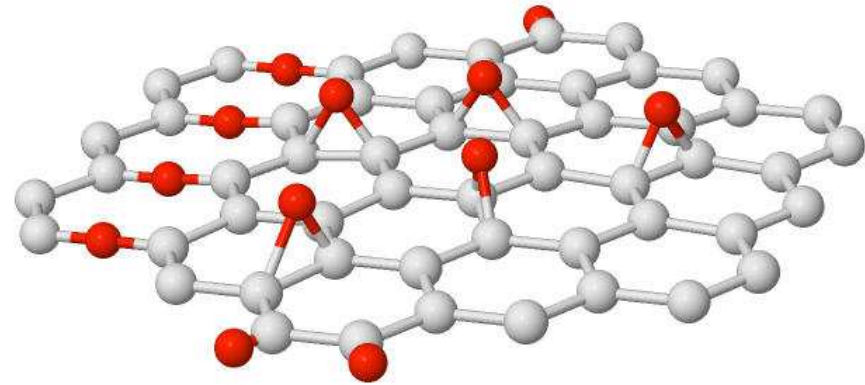
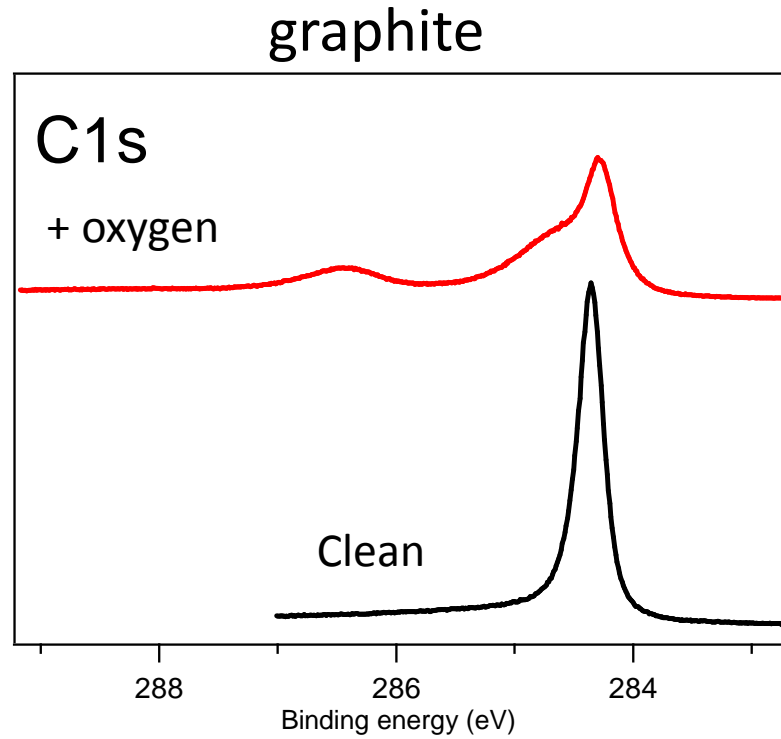
Electron escape depth



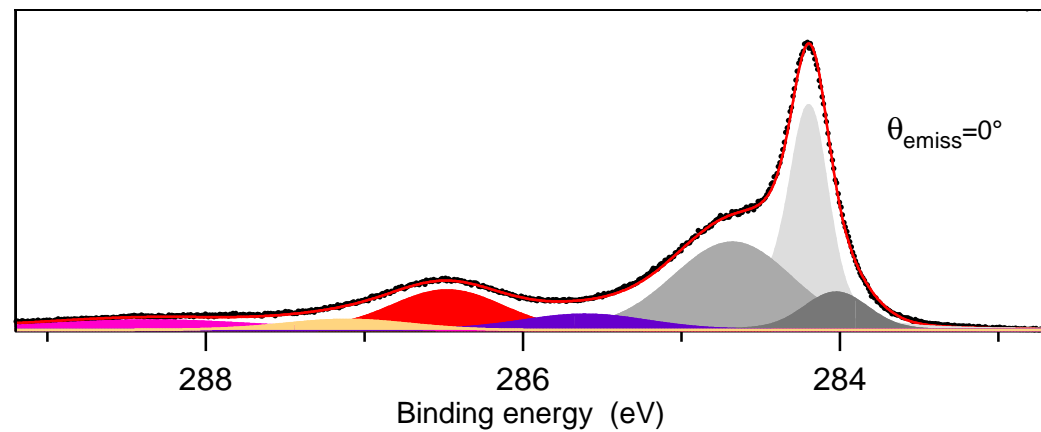
Surface sensitivity



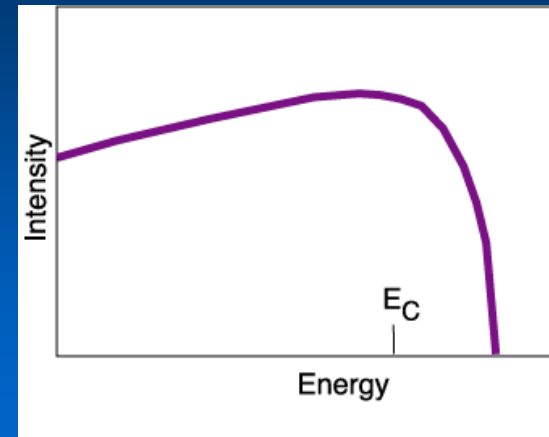
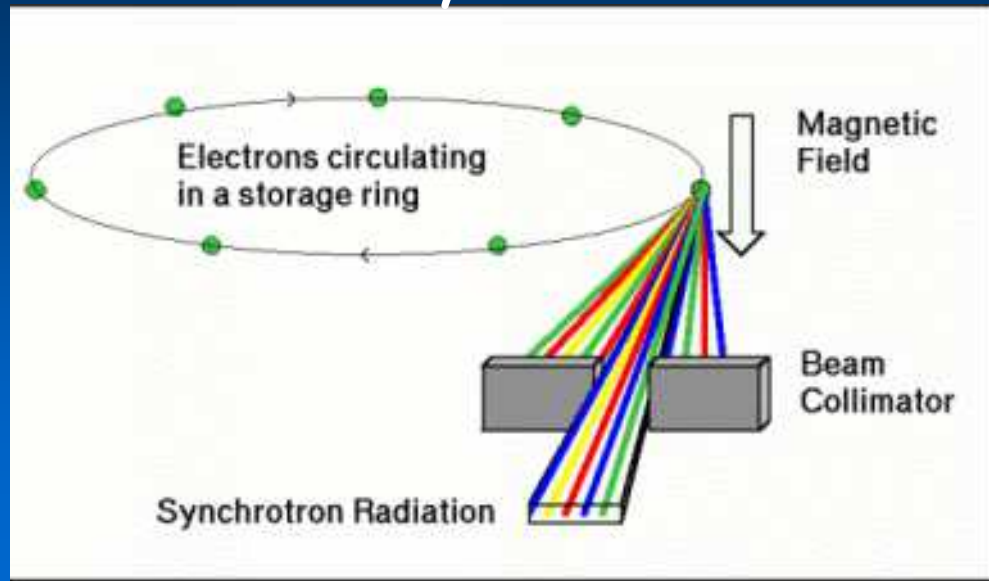
Chemical shift



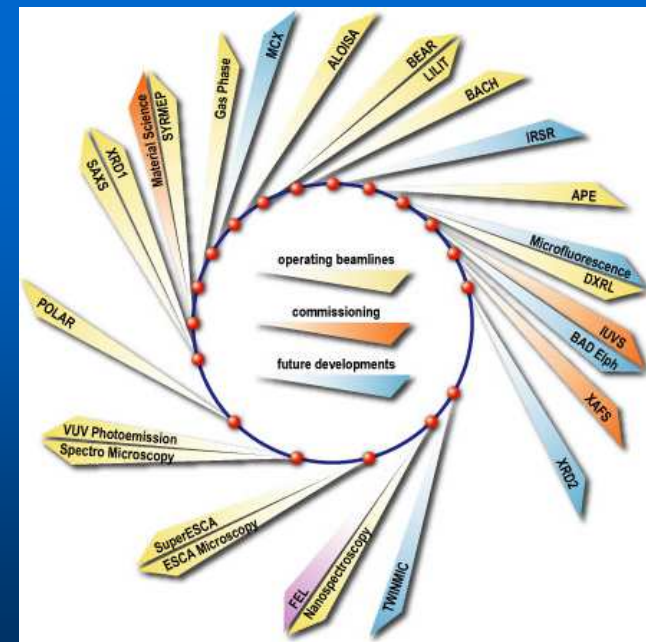
Chemical shift



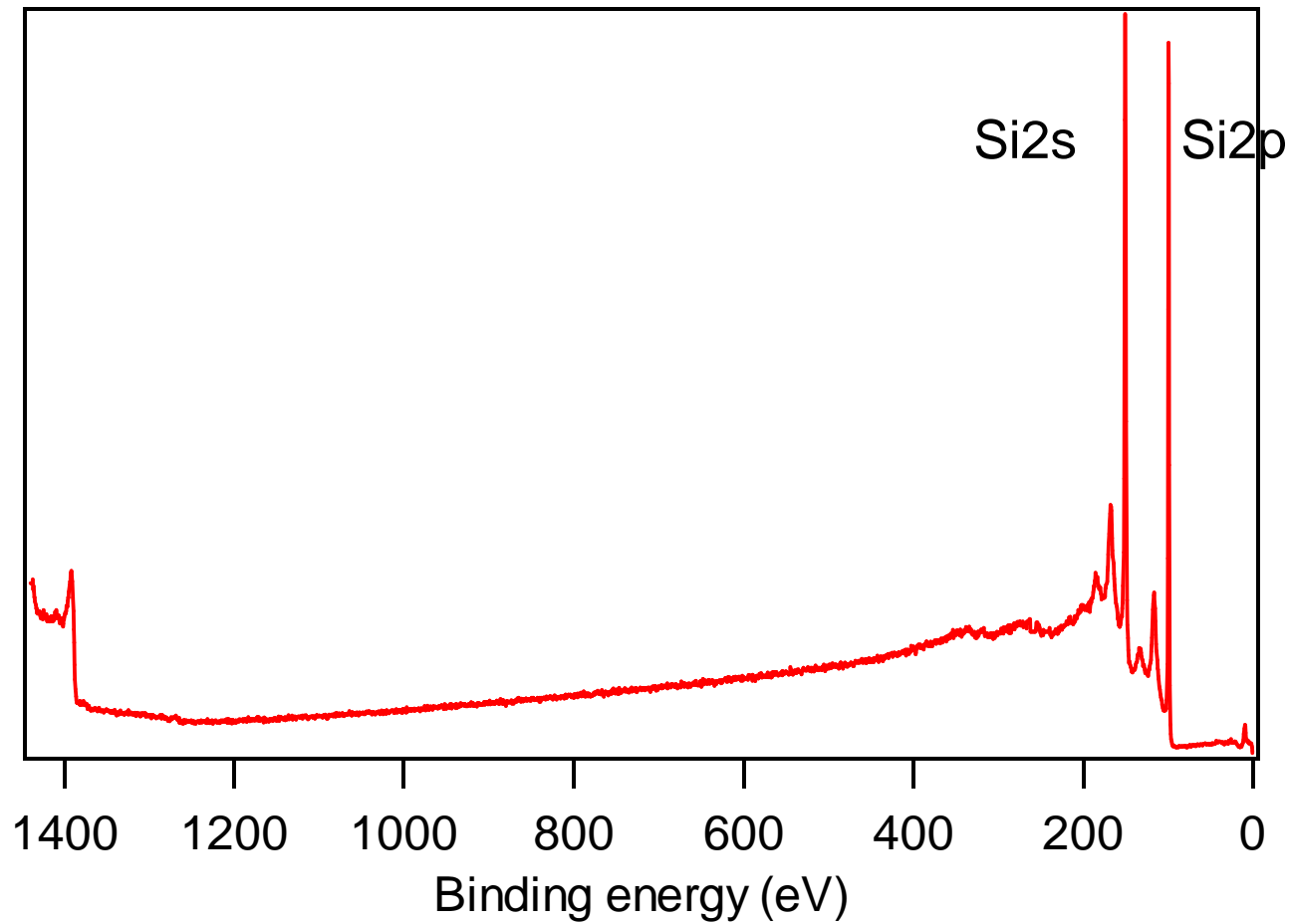
Synchrotron radiation



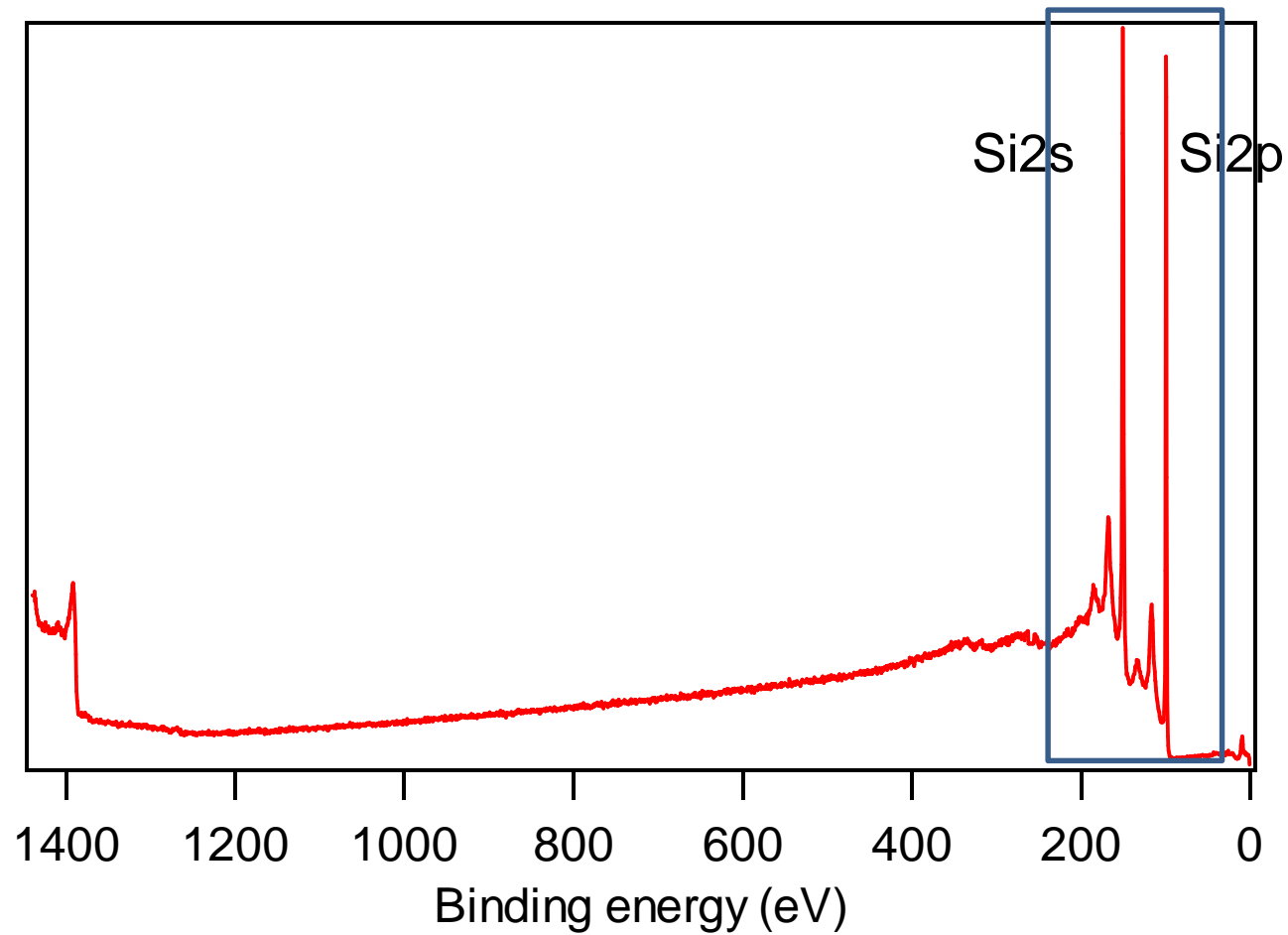
ELETTRA Trieste



Spectral resolution

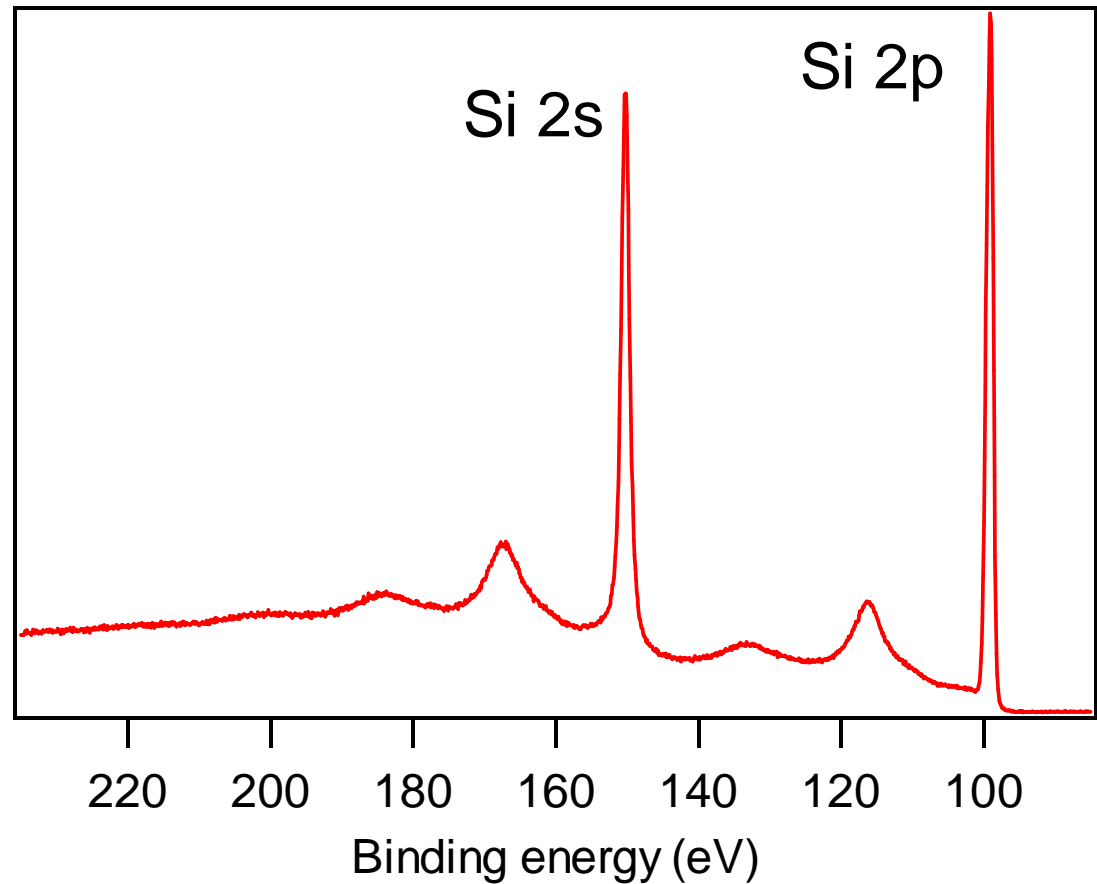
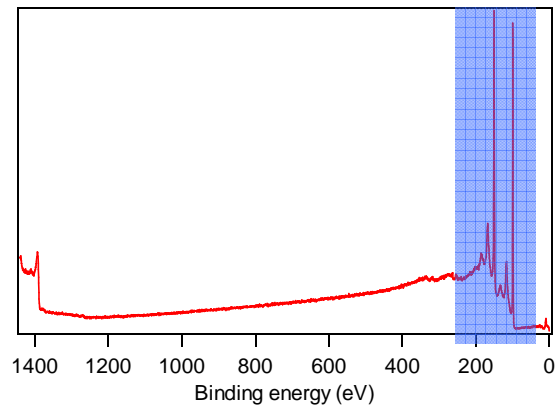


Spectral resolution



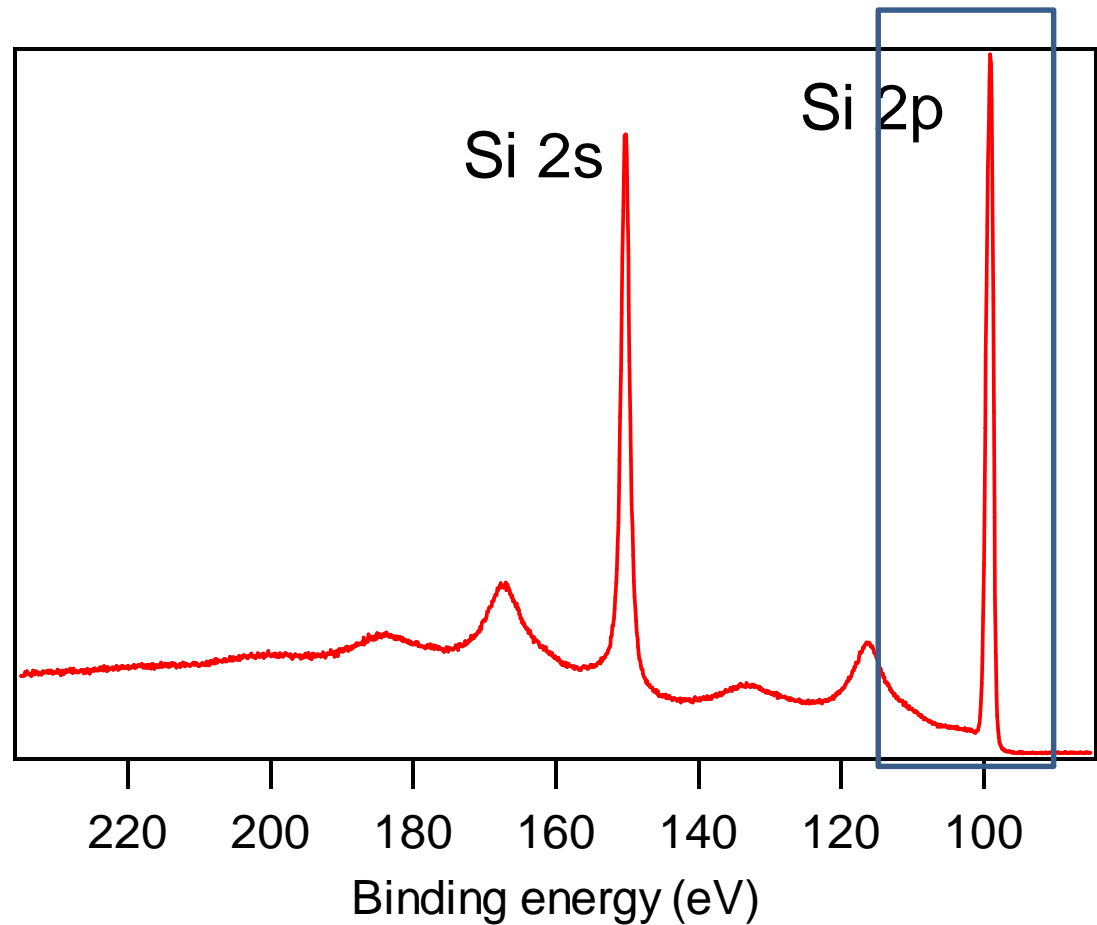
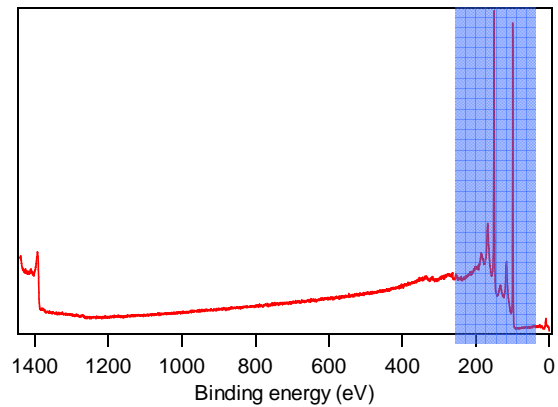
$\lambda = 1486.6 \text{ eV}$ $\Delta E = 0.9 \text{ meV}$

Spectral resolution



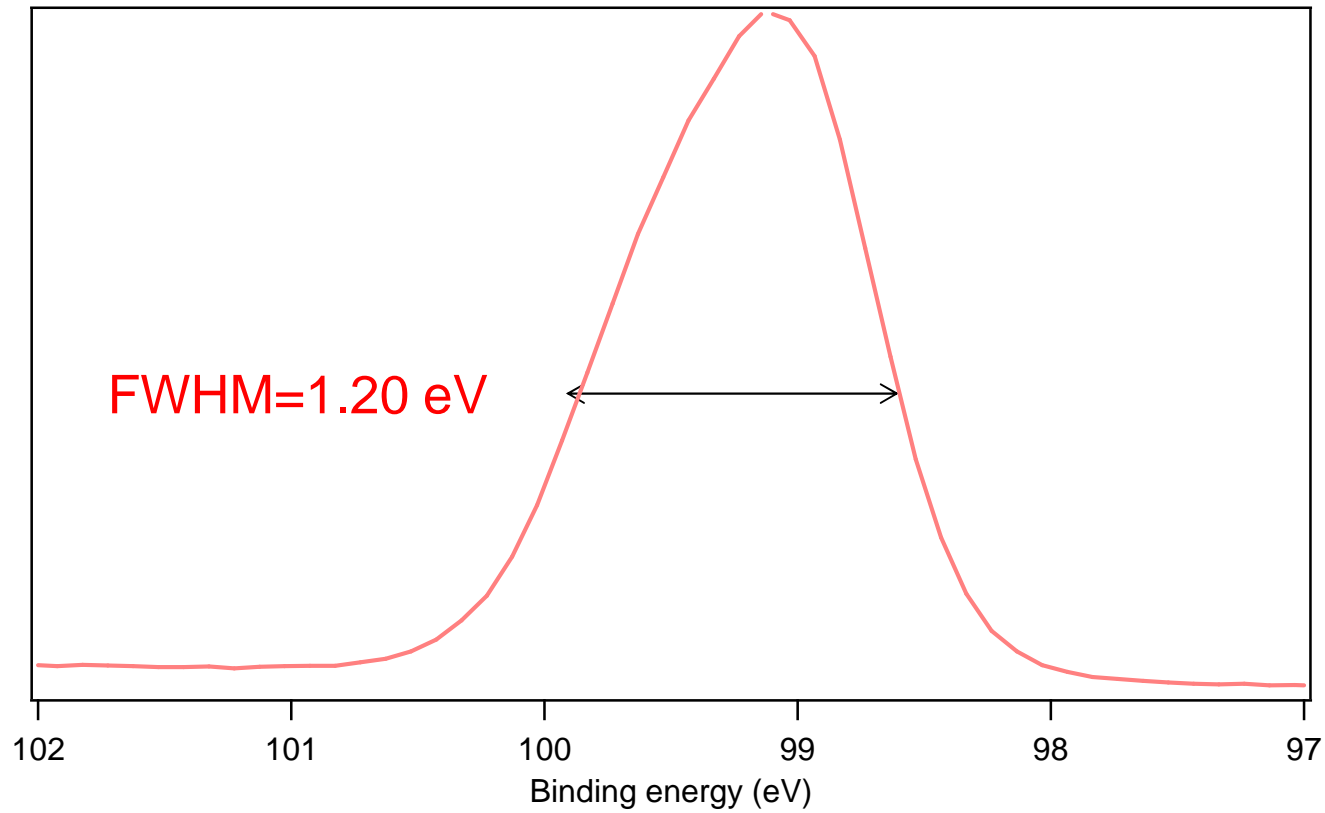
$$\lambda = 1486.6 \text{ eV} \quad \Delta E = 0.9 \text{ meV}$$

Spectral resolution



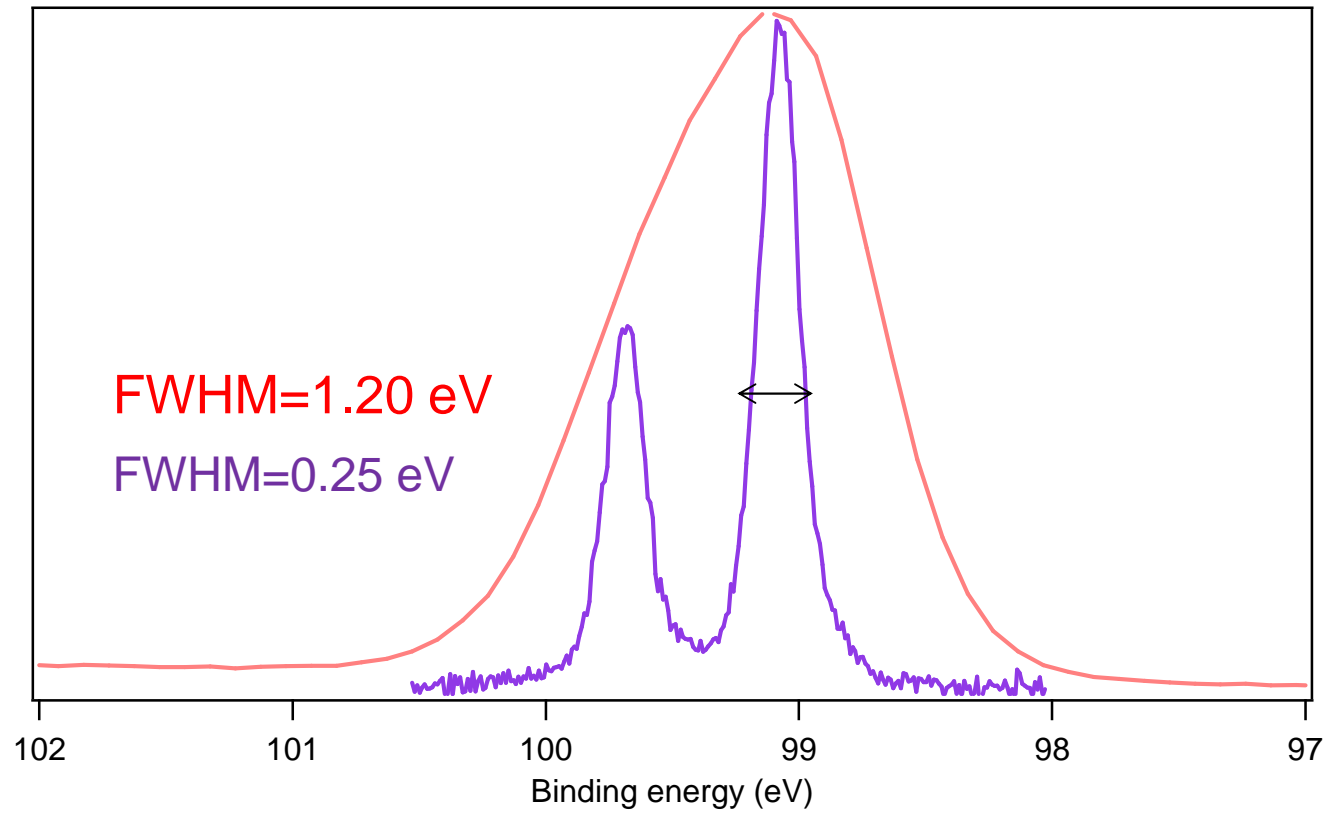
$$\lambda = 1486.6 \text{ eV} \quad \Delta E = 0.9 \text{ meV}$$

Spectral resolution



$$\lambda = 1486.6 \text{ eV} \quad \Delta E = 0.9 \text{ meV}$$

Spectral resolution



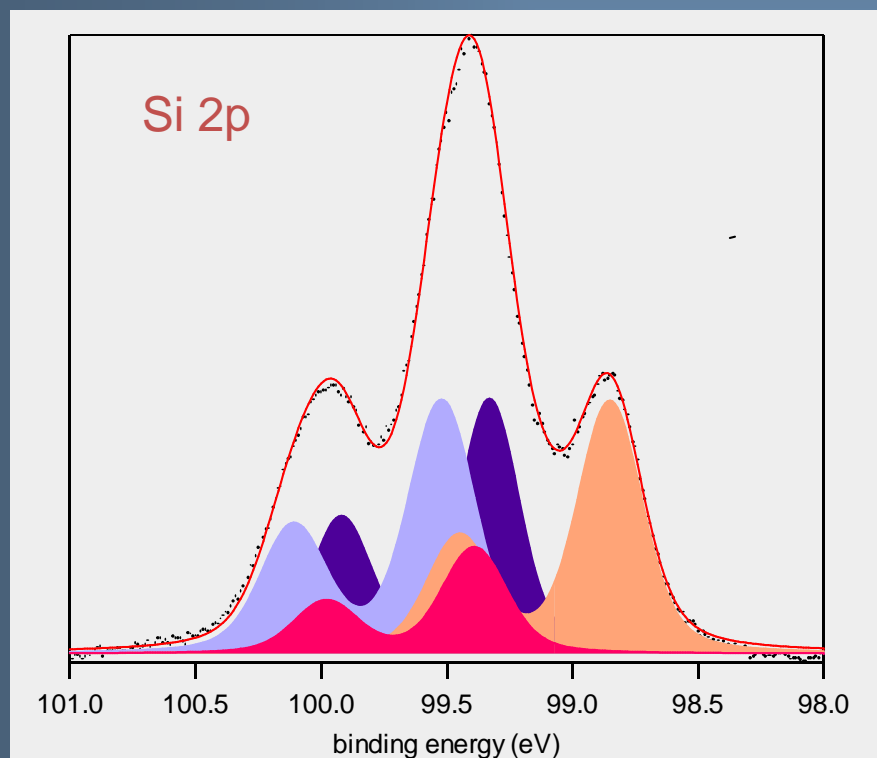
$\lambda = 1486.6 \text{ eV}$ $\Delta E = 0.9 \text{ meV}$

$\lambda = 130 \text{ eV}$ $\Delta E = 50 \text{ meV}$

Si(100)

(2x1) reconstruction

$h\nu=130$ eV



Si(001) (2x1)

