

## 1. Introduction: a historical overview

## 2. Modern medical diagnostics

- CT, NMR, SPECT, PET
- $^{18}\text{F}$  production
- The SWAN project in Bern

## 3. Particle accelerators for medicine

## 4. Conventional radiation therapy

## 5. Basic principles of hadrontherapy

## 6. Present and future of hadrontherapy

## 7. A tour in a hadrontherapy centre

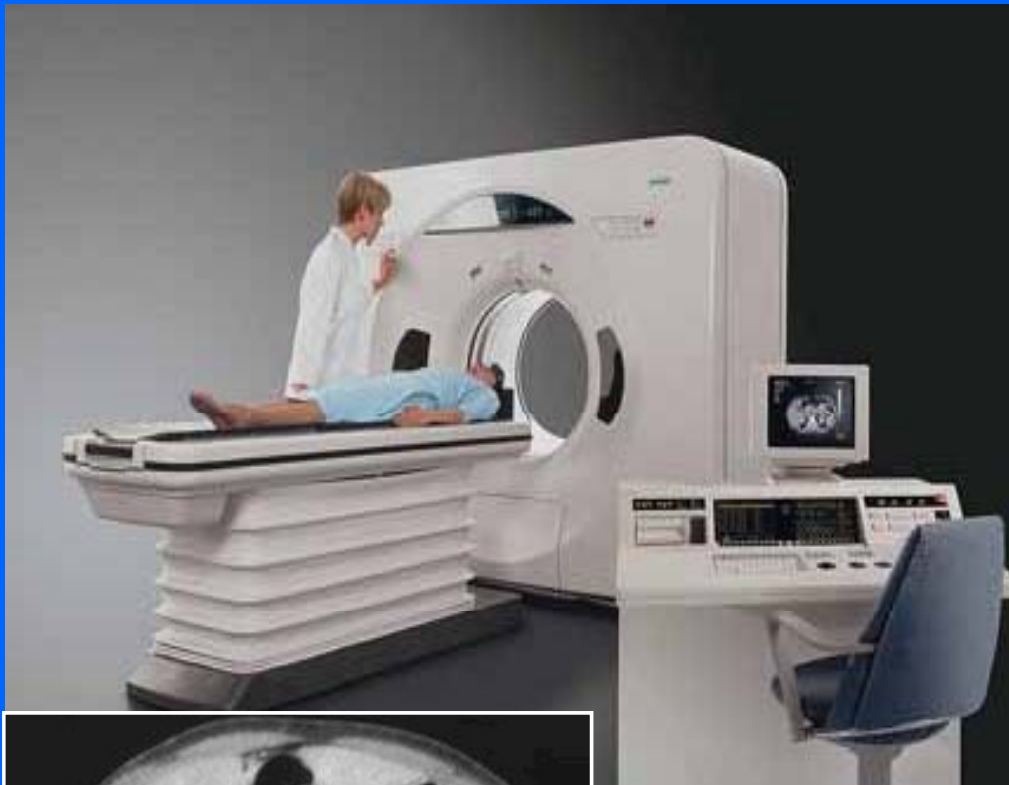
## 8. Specific topics in hadrontherapy

*Medical imaging is essential in modern medicine*

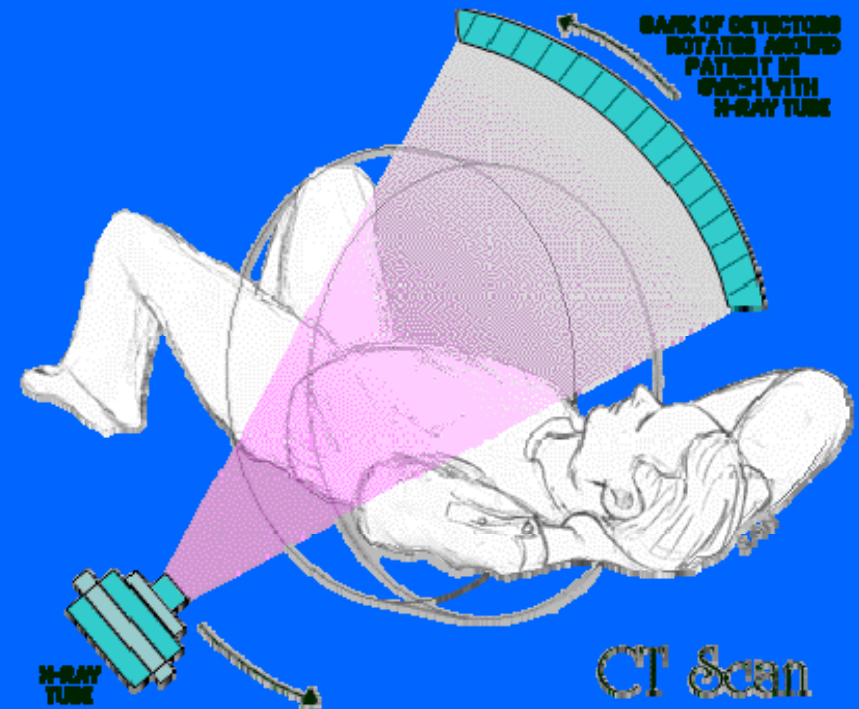
*...and the high precision of hadrontherapy would be useless without it*

# Diagnosics is essential!

## Computed Tomography (CT)



Abdomen



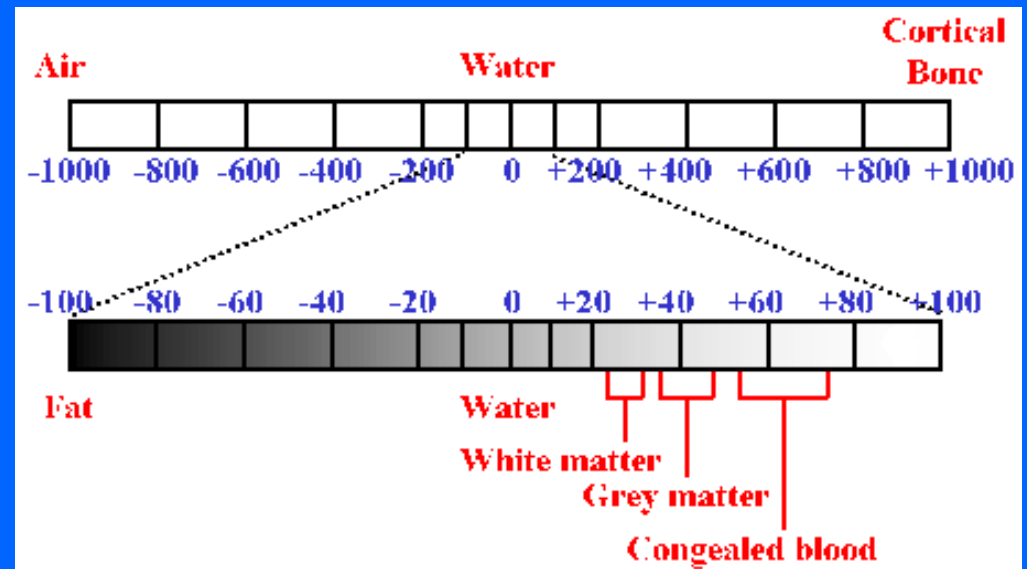
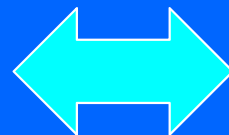
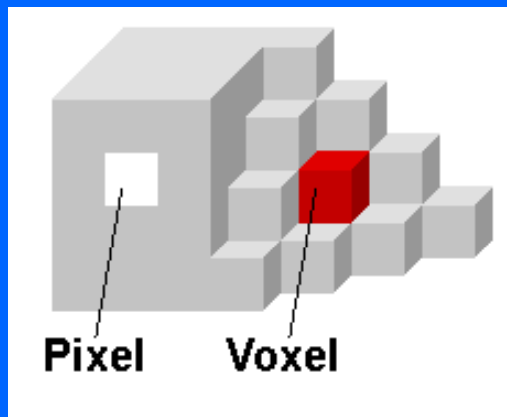
- Measurement of the electron density
- Information on the morphology

# CT and Hounsfield numbers



G. Hounsfield

1979 Nobel Prize for Physiology or Medicine

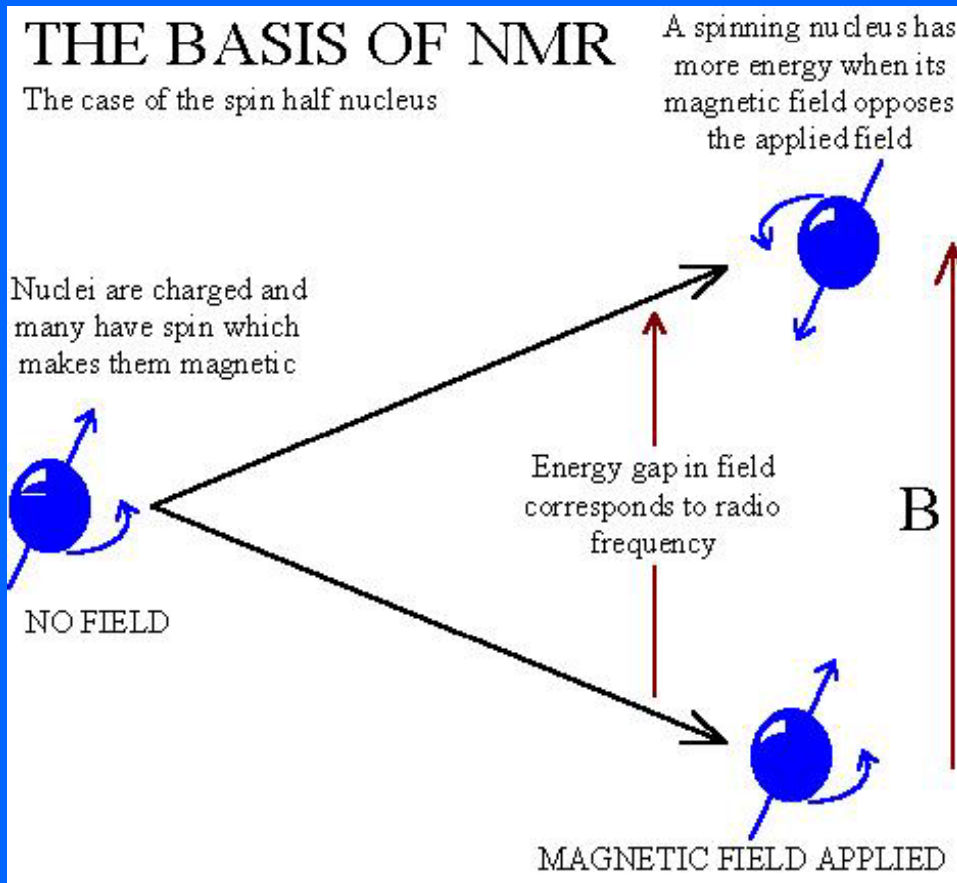


Through the measurement of the attenuation coefficient in many directions and slices (i.e. many radiographies) the Hounsfield numbers are calculated for all the Voxels (=VOLUME piXELS)

# Nuclear Magnetic Resonance

1938-1945

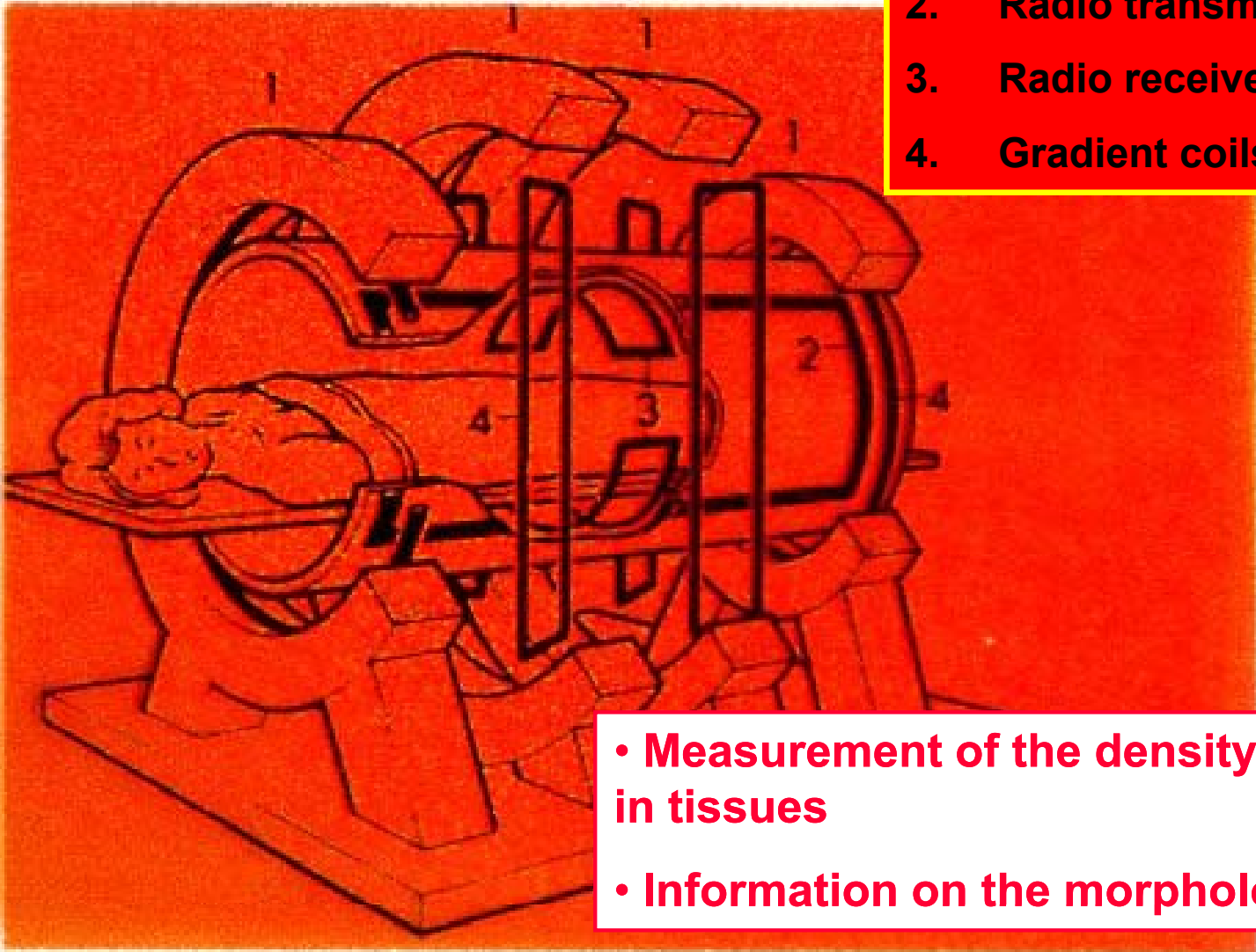
Felix Bloch and Edward Purcell  
discover and study  
NMR



In 1954 Felix Bloch became  
the first CERN Director General

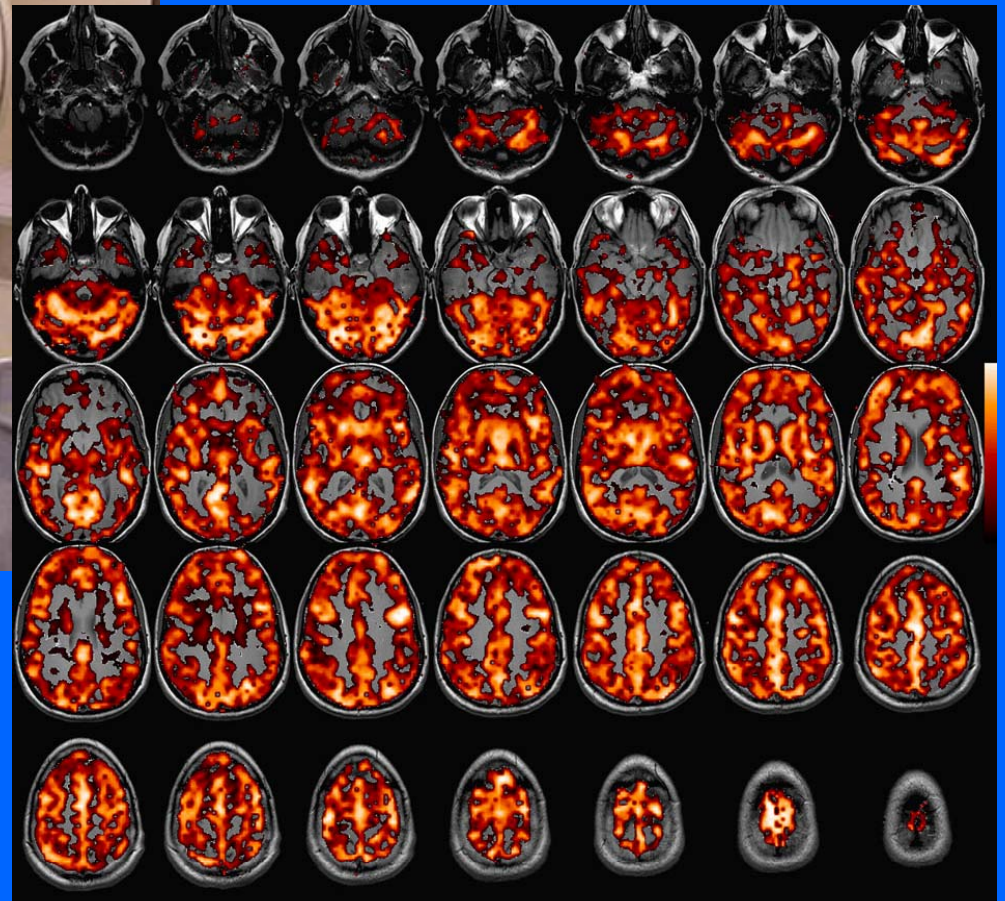
# **MRI = Magnetic Resonance Imaging**

- 1. Main magnet (0.5-1 T)**
- 2. Radio transmitter coil**
- 3. Radio receiver coil**
- 4. Gradient coils**

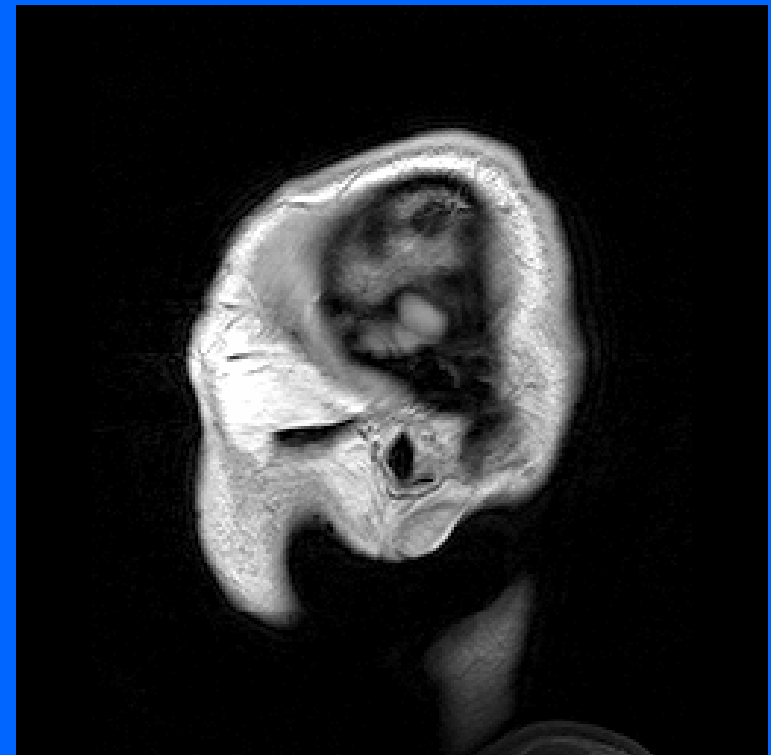


- Measurement of the density of the protons (water) in tissues**
- Information on the morphology**

# A MRI scanner

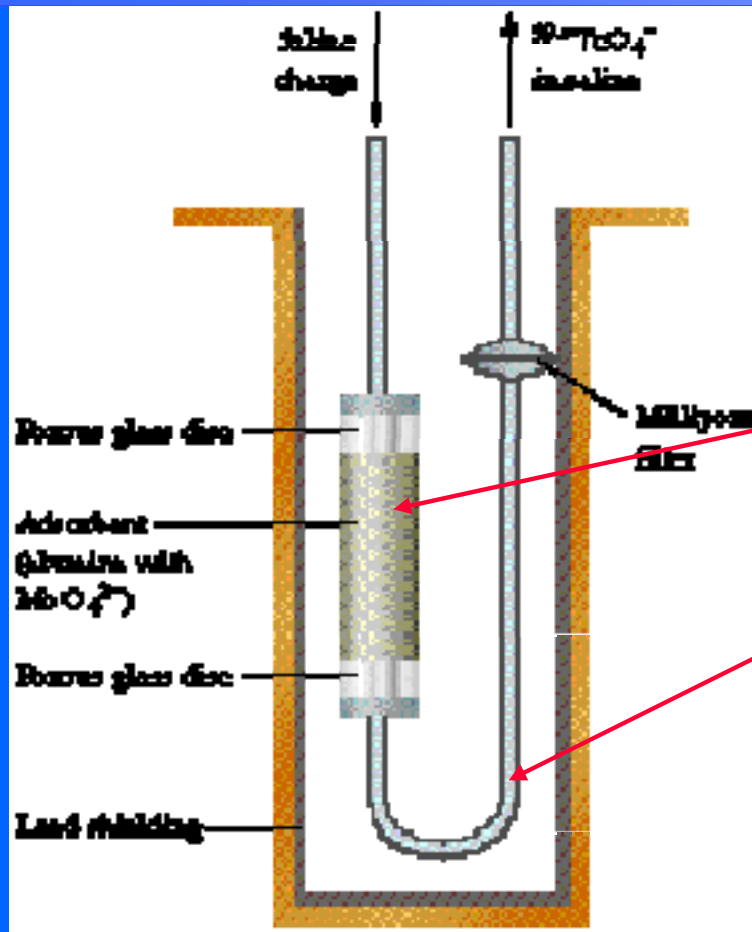


# MRI morphological imaging





# SPECT = Single Photon Emission Computer Tomography



In reactors slow neutrons produce

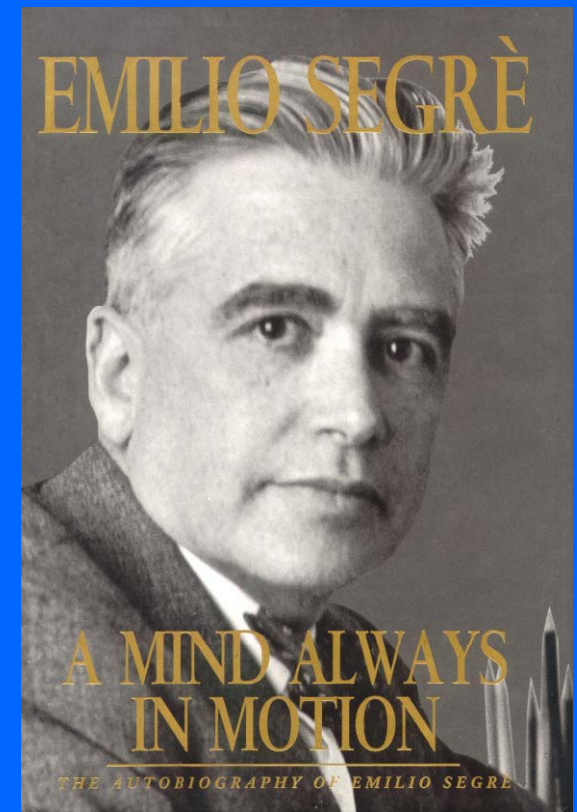


gamma of 0.14 MeV

Emilio Segrè

1937: Discovery of element 43 "Technetium"  $^{97}\text{Tc}$  (2.6 My)

1938: discovery of  $^{99\text{m}}\text{Tc}$   
with E. McMillan (m stands for metastable)



# The molybdenum technetium generator

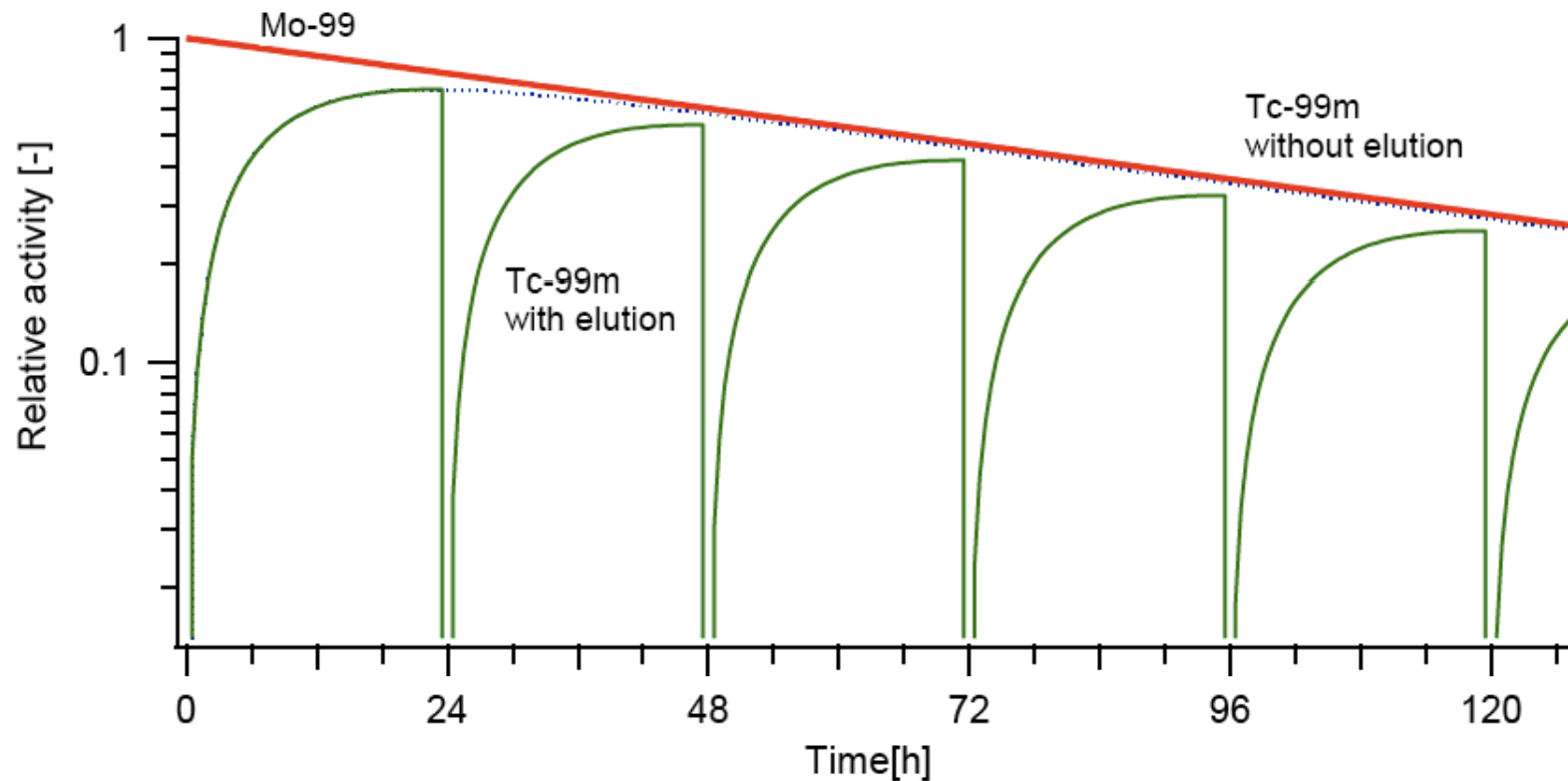


Figure 2.1: Functioning of an isotopic Tc-99m generator. The half-life of Mo-99 being greater than that of Tc-99m, the activity in Tc-99m increases, then decreases with the activity of the parent atoms. When a Tc-99m generator is functioning normally, we let its activity increase for 24 hours then we extract it (elution); the activity of the Tc-99m can then increase again in order to be used the next day.

**Periodic Table of Elements**

1	2											3	4	5	6	7	8	9	10
H	He											B	C	N	O	F	Ne		
3	4											13	14	15	16	17	18		
Li	Be											Al	Si	P	S	Cl	Ar		
11	12	13	14	15	16	17	18											35	36
Na	Mg	Al	Si	P	S	Cl	Ar											Br	Kr
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36		
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr		
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54		
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe		
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86		
Cs	Ba	*La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
87	88	89	104	105	106	107	108	109	110										
Fr	Ra	+Ac	Rf	Ha	106	107	108	109	110										

\* Lanthanide Series

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

+ Actinide Series

90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Legend - click to find out more...

H - gas

Li - solid

Br - liquid

Tc - synthetic

Non-Metals

Transition Metals

Rare Earth Metals

Halogens

Alkali Metals

Alkali Earth Metals

Other Metals

Inert Elements

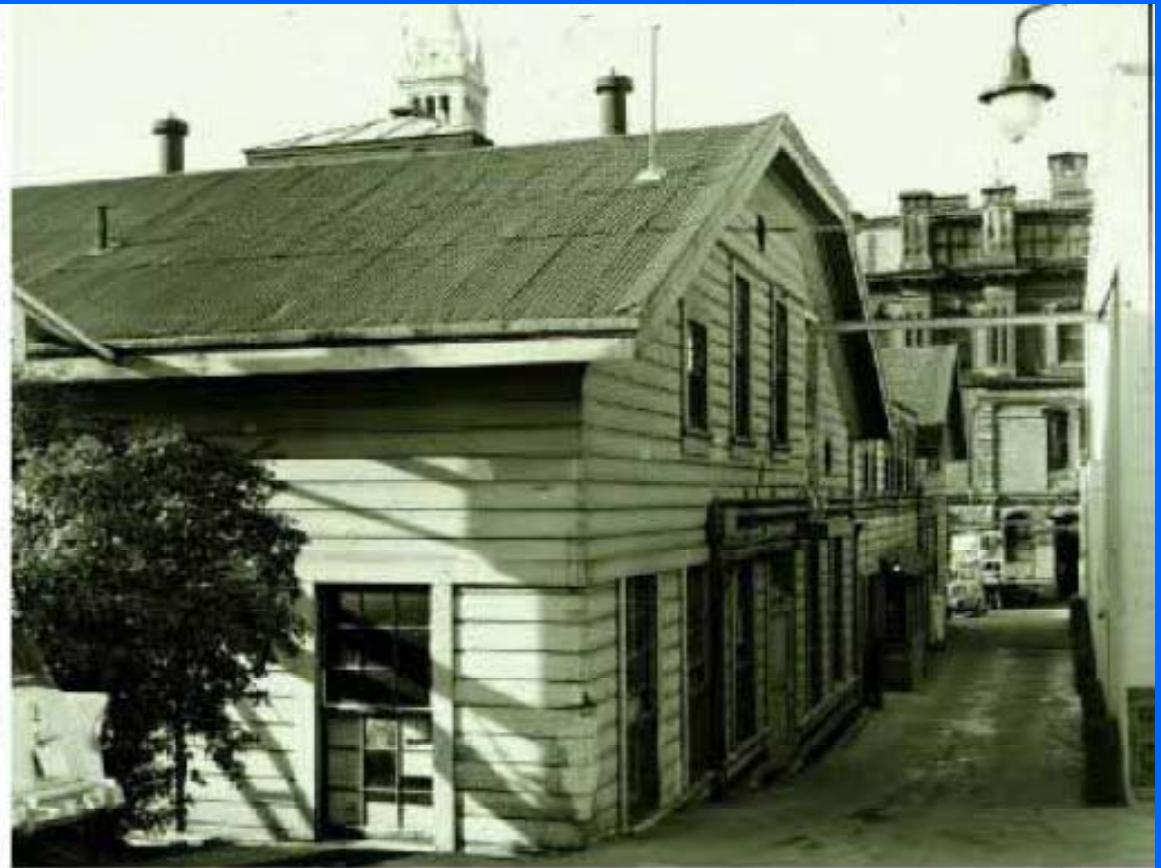
- The element 43 was missing

- In 1925 W. Noddack and I. Tacke announced the discovery of Rhenium (75) and Masurium (43)

- In 1934 Fermi and his group were bombarding "all" the elements with slow neutrons and Segrè was in charge of procuring the different elements

- ...but asking for a sample of Masurium he was answerd "Numquam vidi"...

## The discovery of technetium



The **Rad Lab** is officially established within the UC Physics Department with Lawrence as director; in Italy, Segrè examines an "invaluable gift" of material irradiated by the 27-inch cyclotron and discovers the first artificial element, later named **technetium**.

## *The discovery of technetium*

- Lawrence was using deflectors for the cyclotron made of Molybdenum (42)
- Segrè thought : Molybdenum + proton...  $42 + 1 = 43$  !
- In February 1937 Segrè received a letter from Lawrence with some Molybdenum coming from the deflectors and...
- ...the element 43 was identified with the help of a chemist (Carlo Perrier)
- The element 43 was called Technetium since it is the first element artificially produced (the most stable isotope has an half-life of  $4.2 \times 10^6$  years)

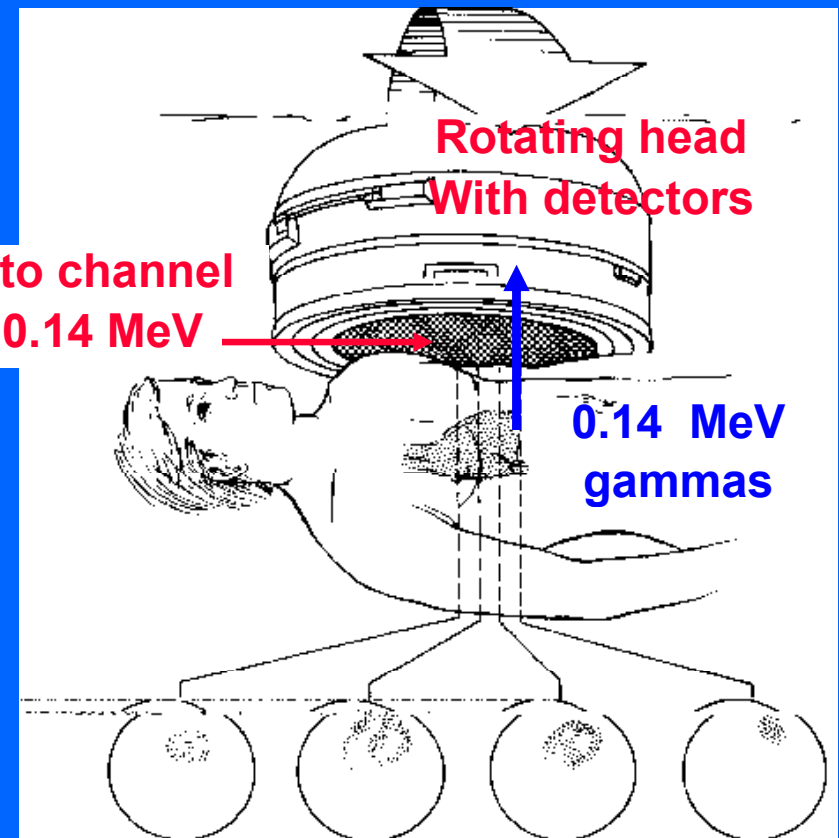
## *SPECT scanner*

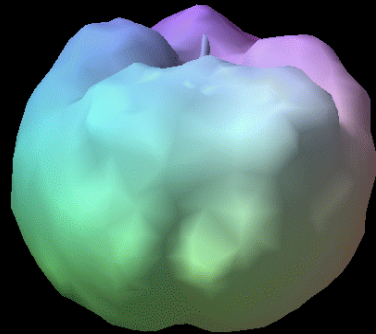
85% of all nuclear medicine examinations use technetium produced by slow neutrons in reactors

- Measurement of the density the molecules which contain technetium
- Information on morphology and/or metabolism

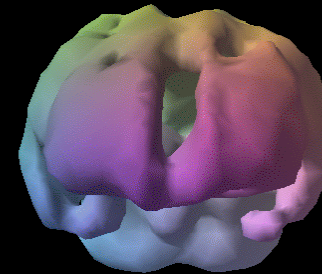
... liver  
lungs  
bones ...

Lead collimators to channel the gammas of 0.14 MeV

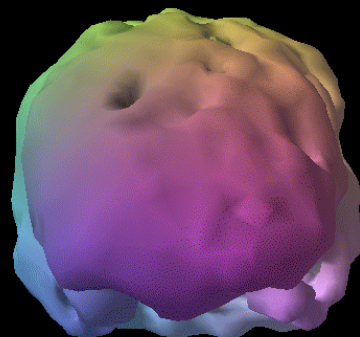




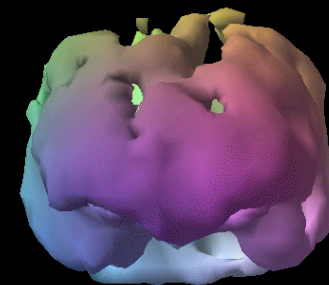
Healthy



Alcohol addict



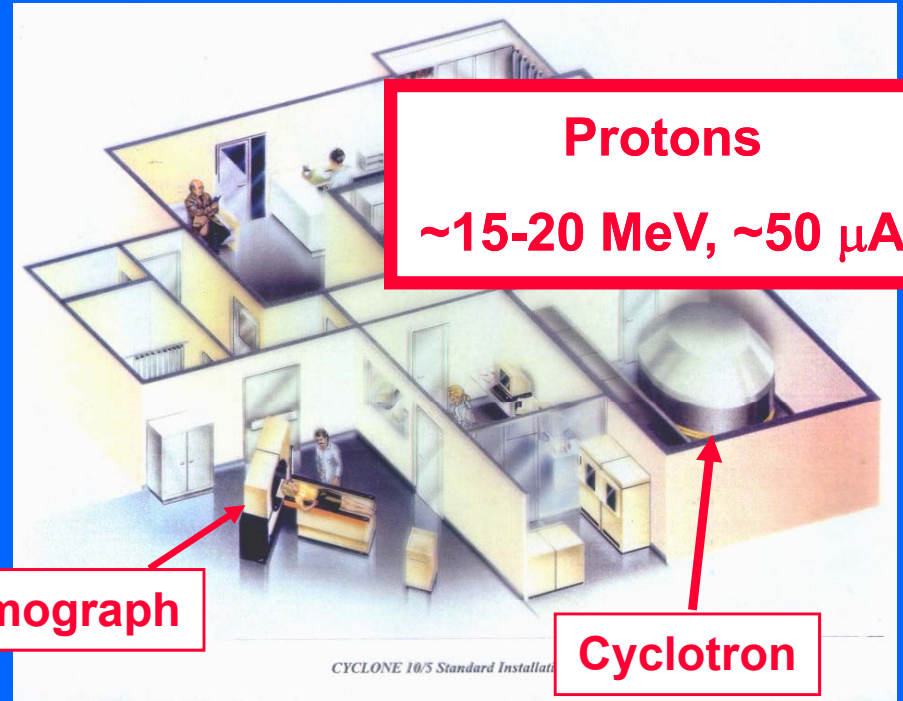
Drug addict



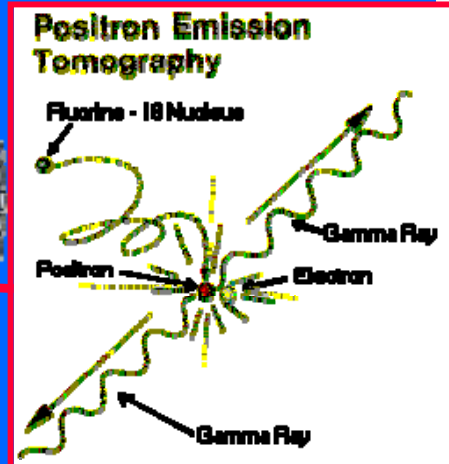
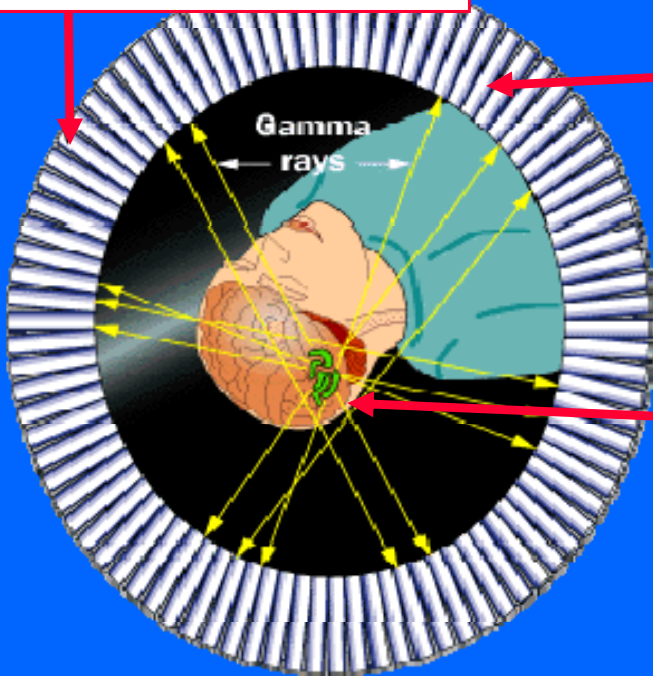
Encephalitis

# Positron Emission Tomography (PET)

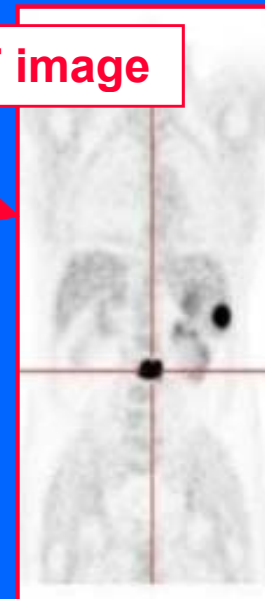
- FDG with  $^{18}\text{F}$  is the most used drug (half life 110 minutes)
- Measurement of the density of  $^{18}\text{F}$  through back-to-back gamma detection
- Information on metabolism



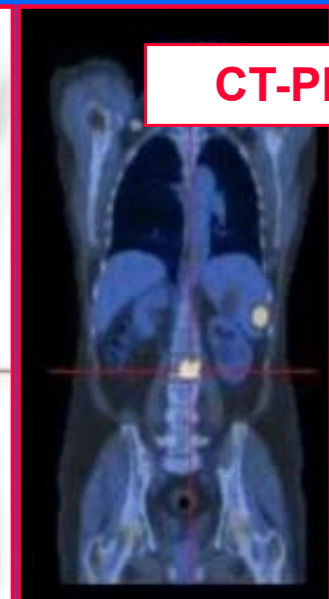
Gamma ray detectors  
(Ex. BGO crystals)



PET image



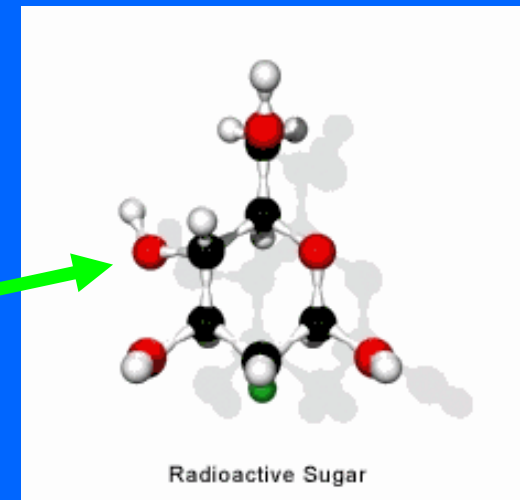
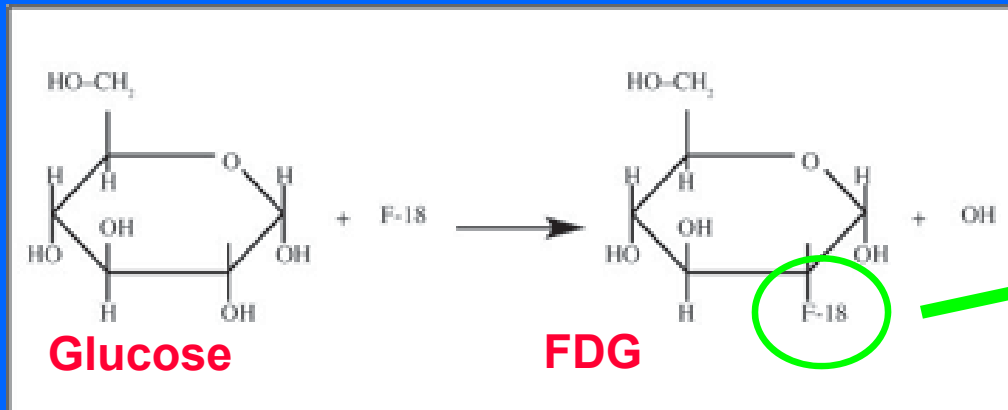
CT-PET





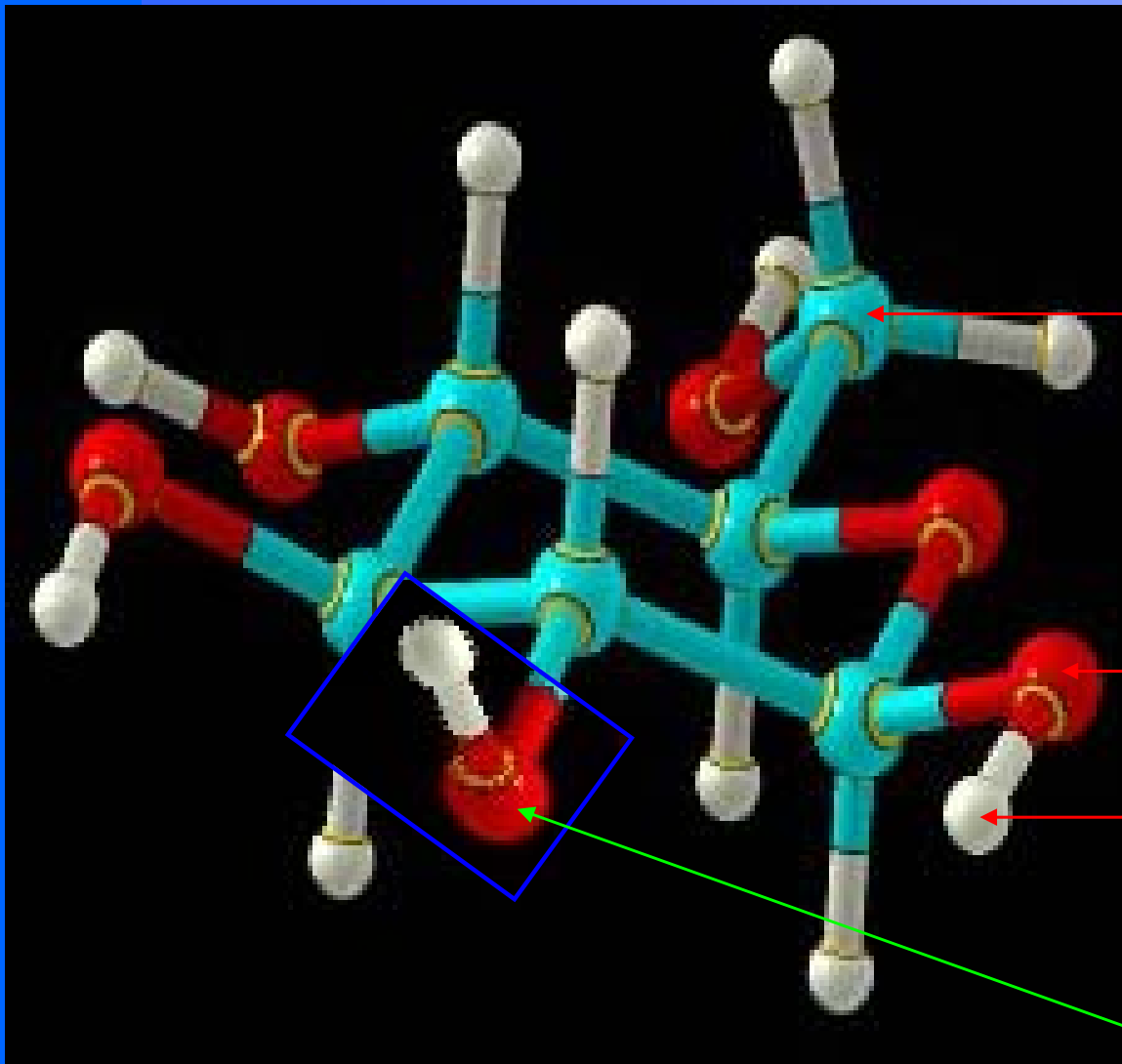
## How does it work?

- $\text{H}_2^{18}\text{O}$  water is bombarded with protons to produce  $^{18}\text{F}$
- Fluoro-Deoxy-D-Glucose (FDG) is synthesized



- FDG is transported to the hospital
- FDG is injected into the patient
- FDG is trapped in the cells that try to metabolize it
- Concentration builds up in proportion to the rate of glucose metabolism
- Tumors have a high rate of glucose metabolism and appear as "hot spots" in PET images

# FDG synthesis



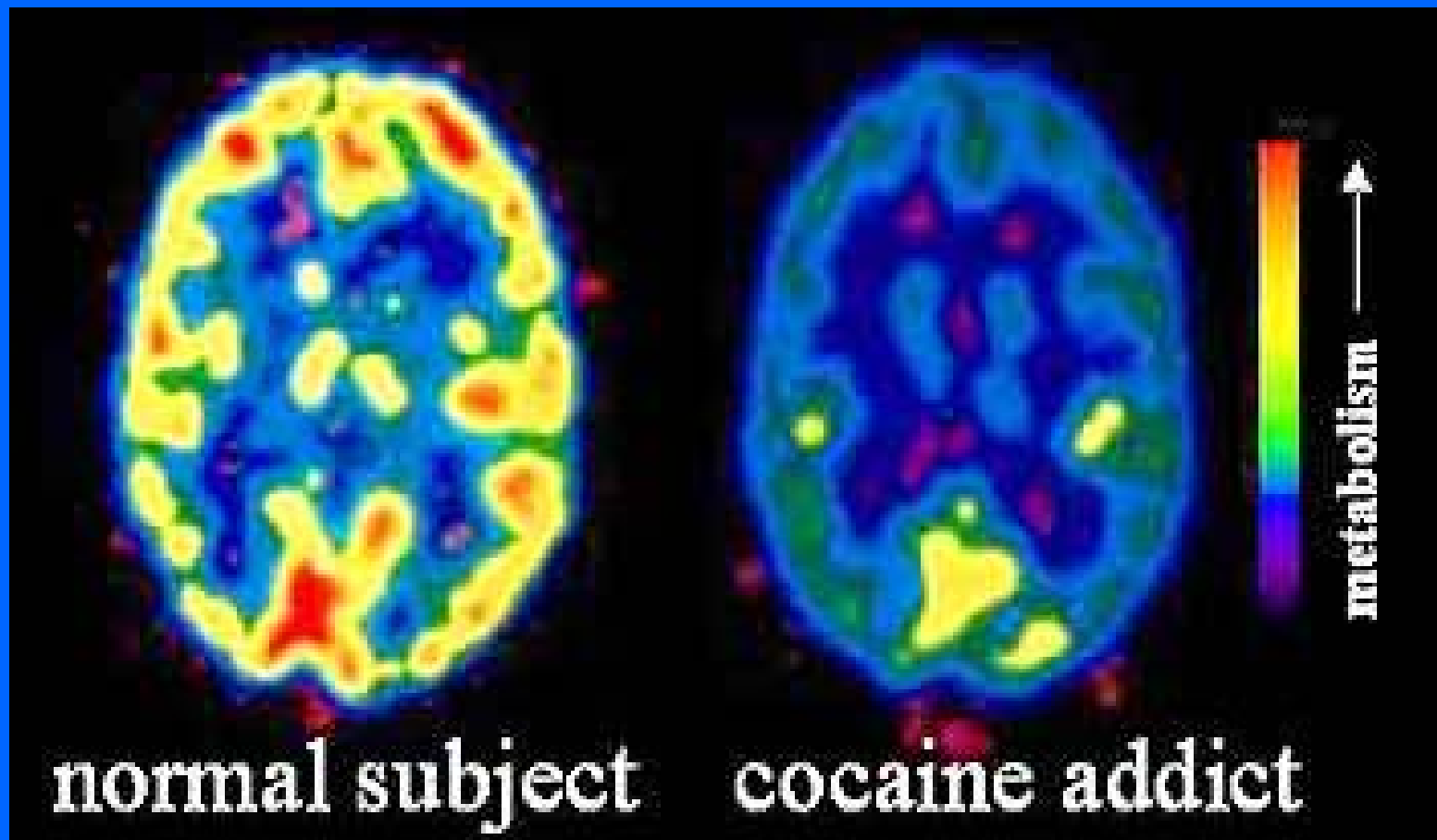
C

O

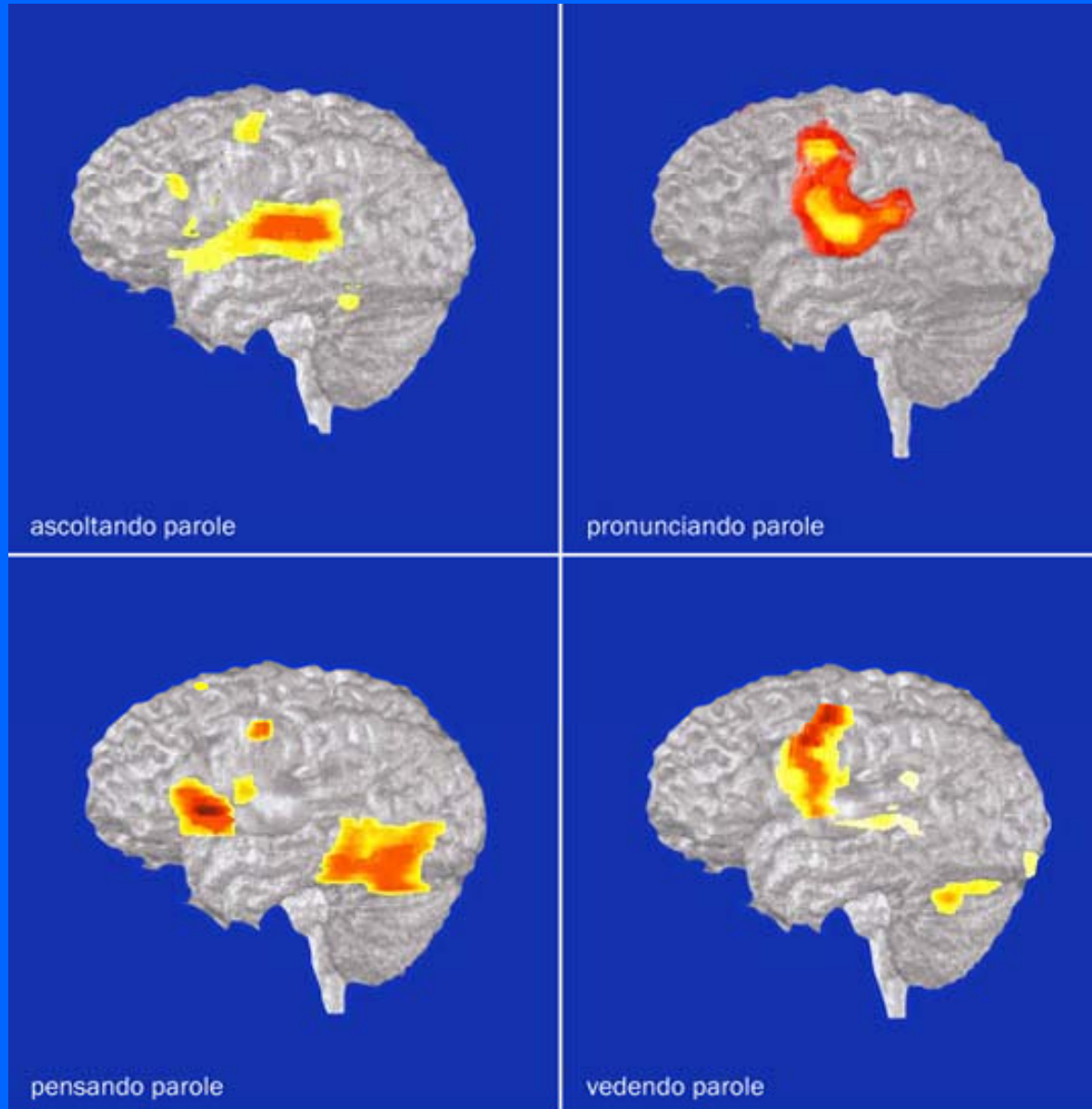
H

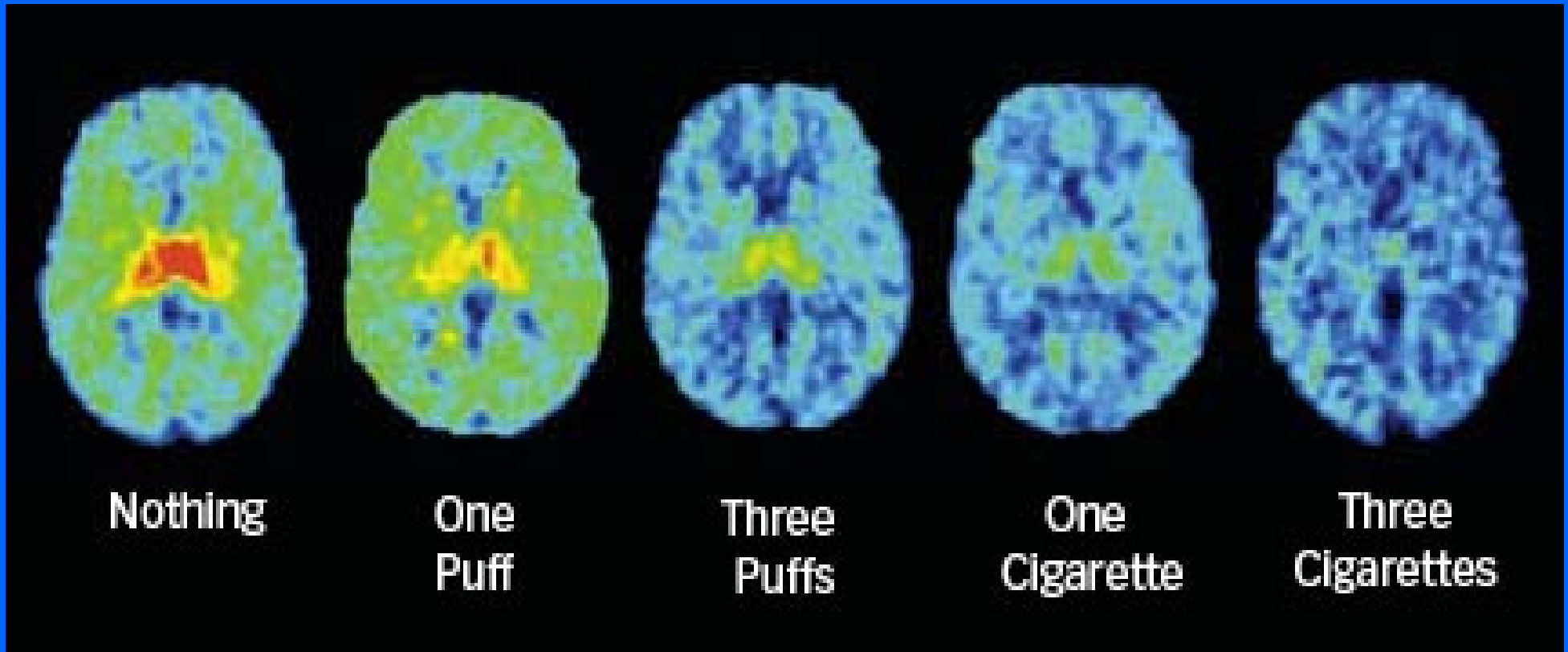
<sup>18</sup>F

D-glucose :  $\text{CH}_2\text{OH} (\text{CHOH})_4 \text{CHO}$



- **18FDG/PET images**
- **The cocaine addict has depressed metabolism !**

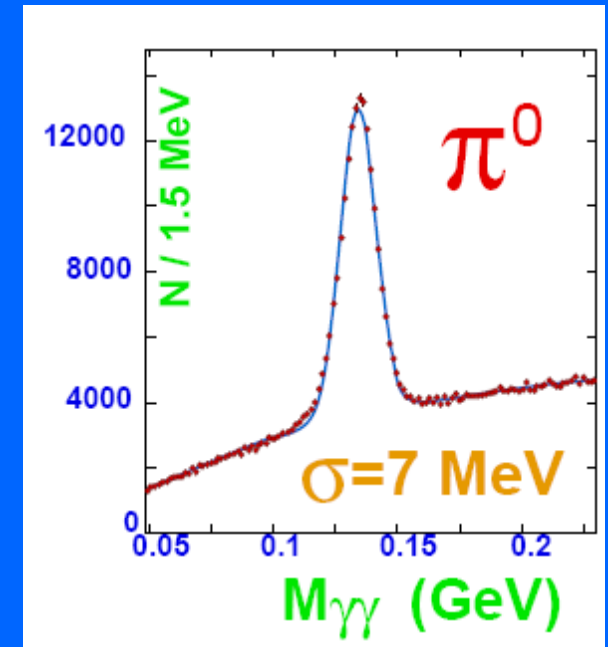
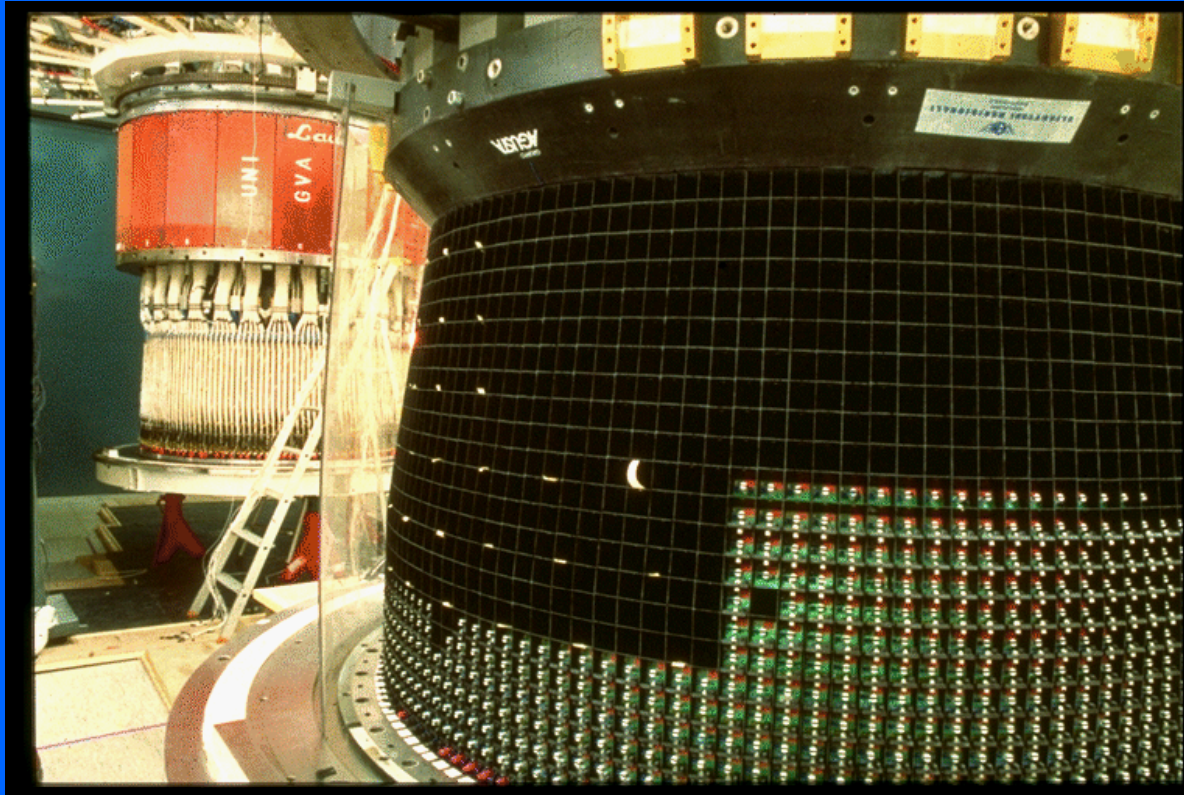




**SMOKING SATURATES RECEPTORS** As nicotine from a cigarette attaches to the  $\alpha 4 \beta 2^*$ -nACh nicotinic receptors in the brain, it displaces a radiolabeled tracer (red and yellow indicate high levels of the tracer, green indicates intermediate levels, and blue indicates low levels). The nicotine from three puffs displaced 75 percent of the tracer from study participants' receptors, and the nicotine from three cigarettes, nearly all.

# The BGO calorimeter of the L3 experiment at LEP (CERN 1989-2000)

BGO crystals have been developed for detectors in particle physics

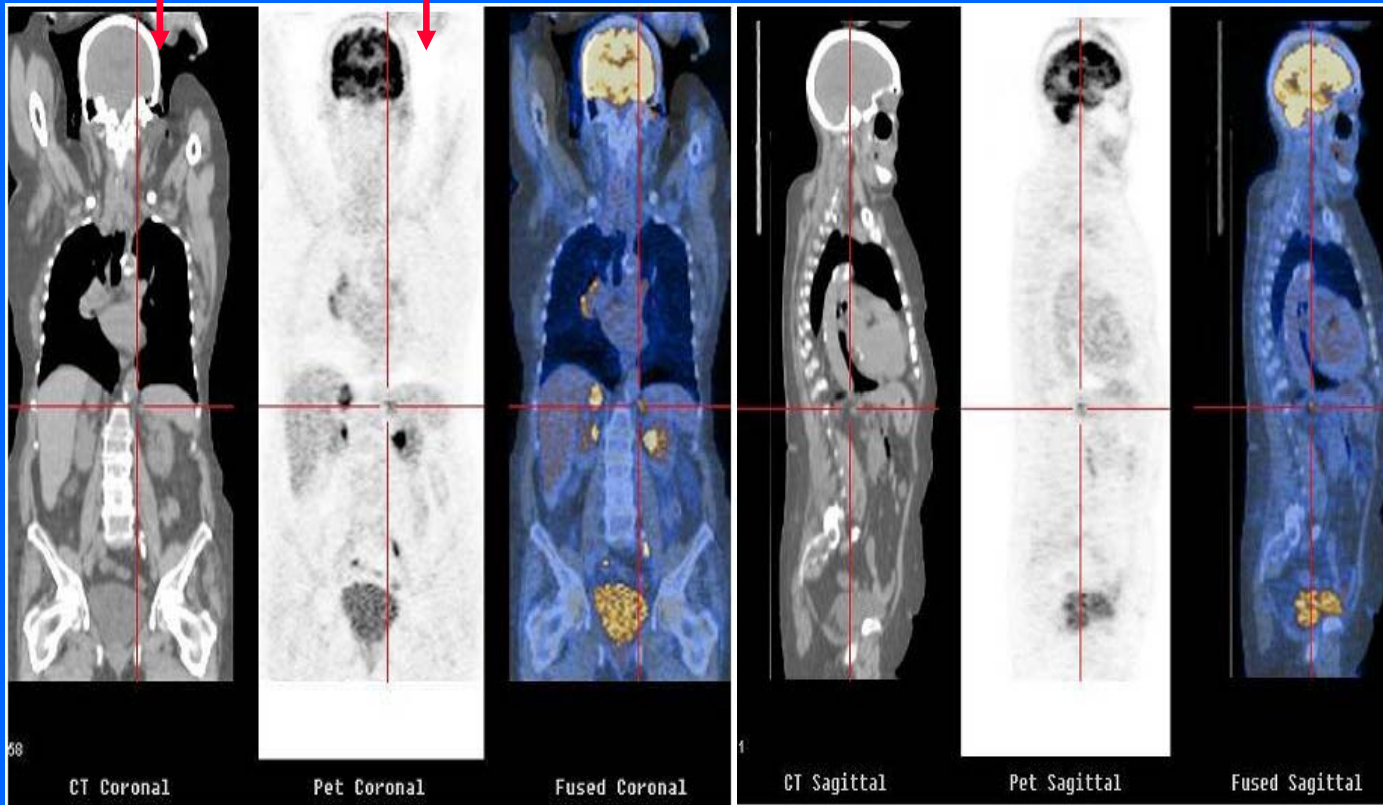


- 11000 BGO crystals
- Precise measurement of the energy deposited by the particles
- Almost  $4\pi$  coverage

# The new diagnostics: CT/PET

morphology

metabolism



David Townsend

CERN: 1970-78

Uni Ginevra

UPSM Pittsburgh

and

Ronald Nutt

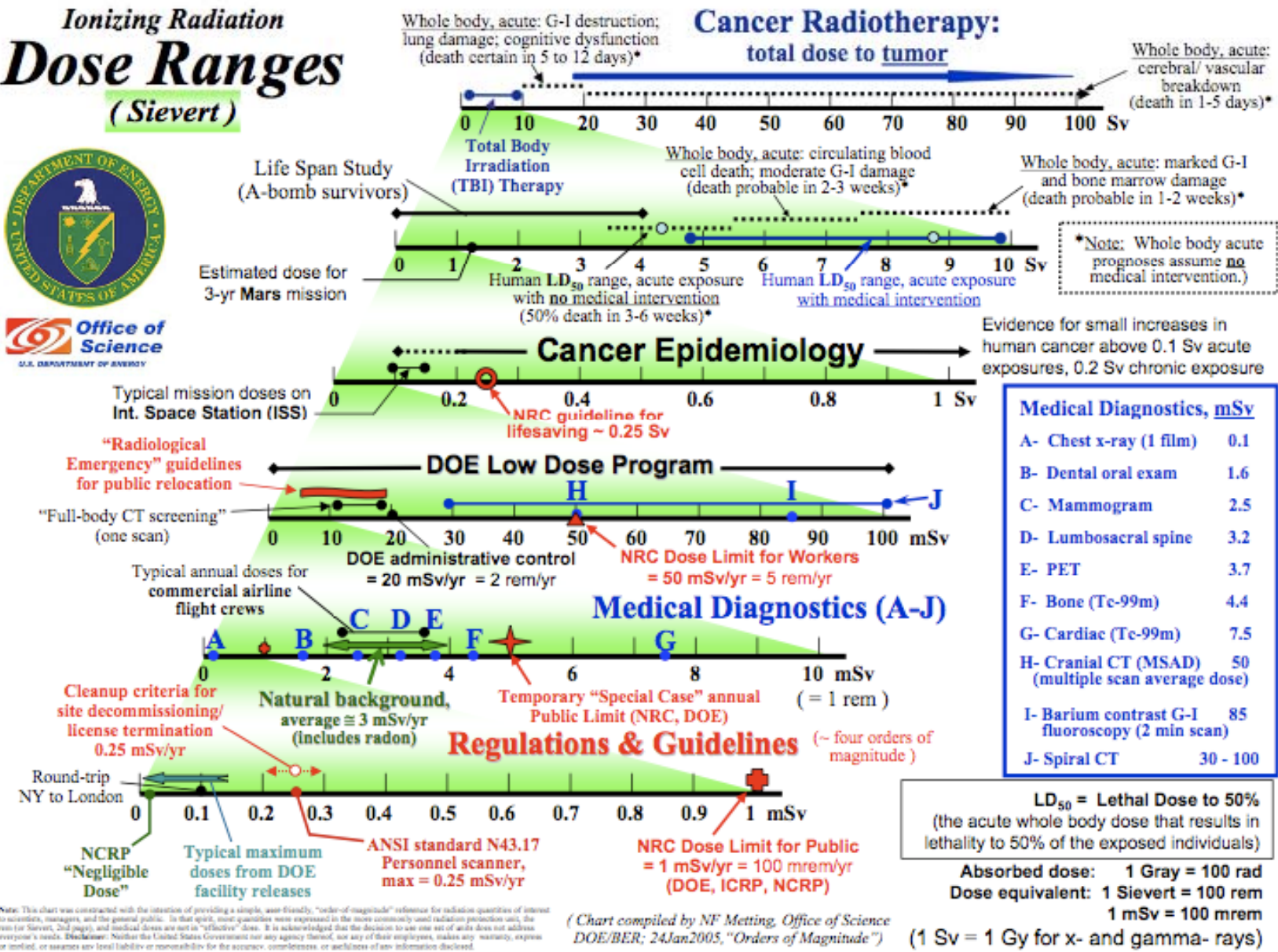
(CTS – CTI)

# Doses in medical diagnostics

## Ionizing Radiation Dose Ranges (Sievert)



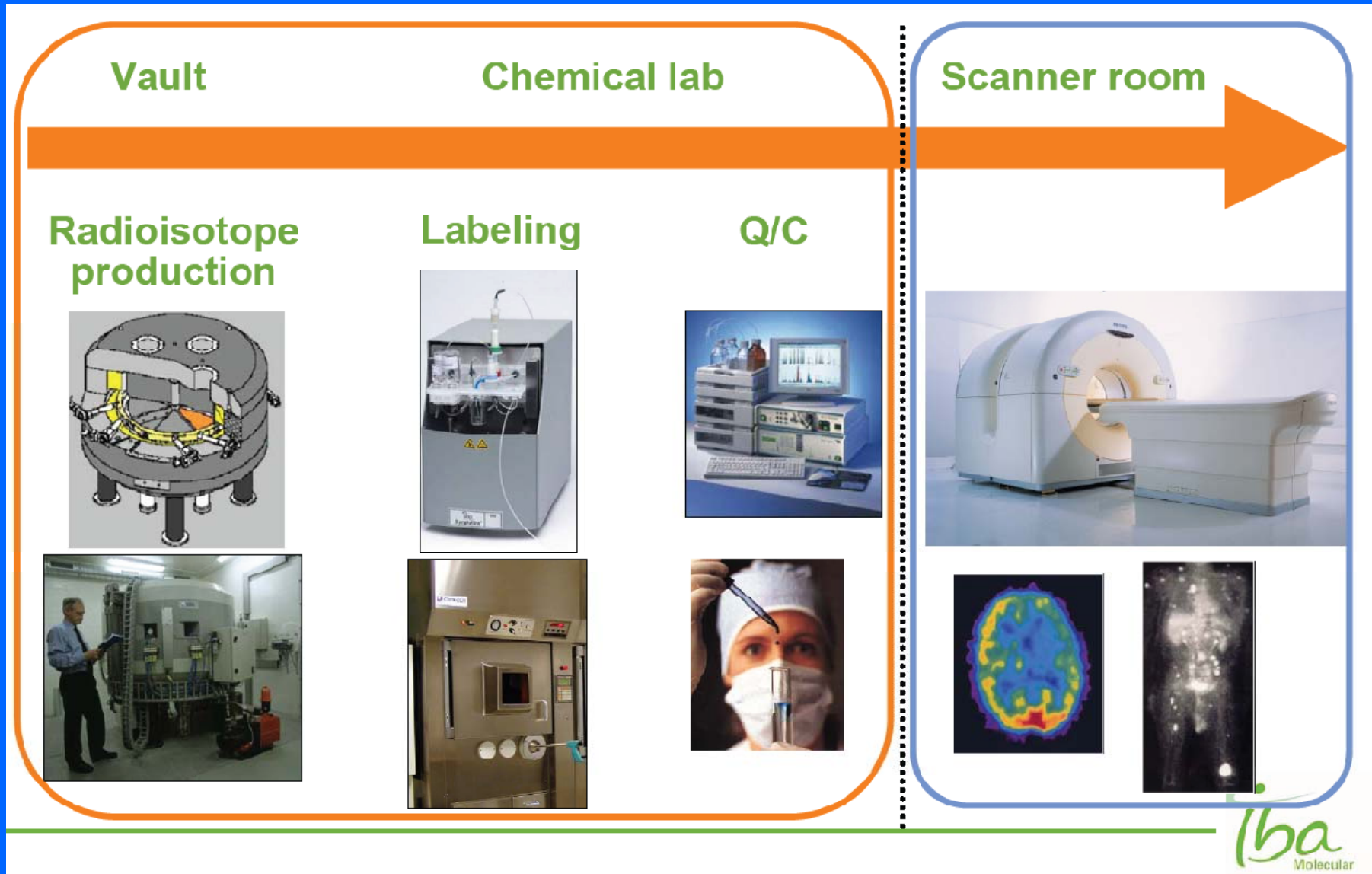
Office of Science  
U.S. DEPARTMENT OF ENERGY





***Exercise: the production of FDG for PET***

# The full FDG-PET chain



Courtesy IBA

- 20 MeV proton beam (cyclotron)
- Current : 50  $\mu\text{A}$
- FWHM : about 15 mm
- Target : 99%  $^{18}\text{O}$  enriched water
- Reaction :  $^{18}\text{O} (p,n) ^{18}\text{F}$
  
- Fluorine 18 : half-life  $t_{1/2}=110$  min.
- Irradiation time 60 min.

● What is the value of the  $^{18}\text{F}$  activity produced?



**One TR19 cyclotron by the company ACSI (Vancouver, Canada) is installed at the Policlinico Gemelli in Rome**  
**It is daily used for FDG production**

Nirta® Fluor |  
Patented Niobium insert.



Courtesy



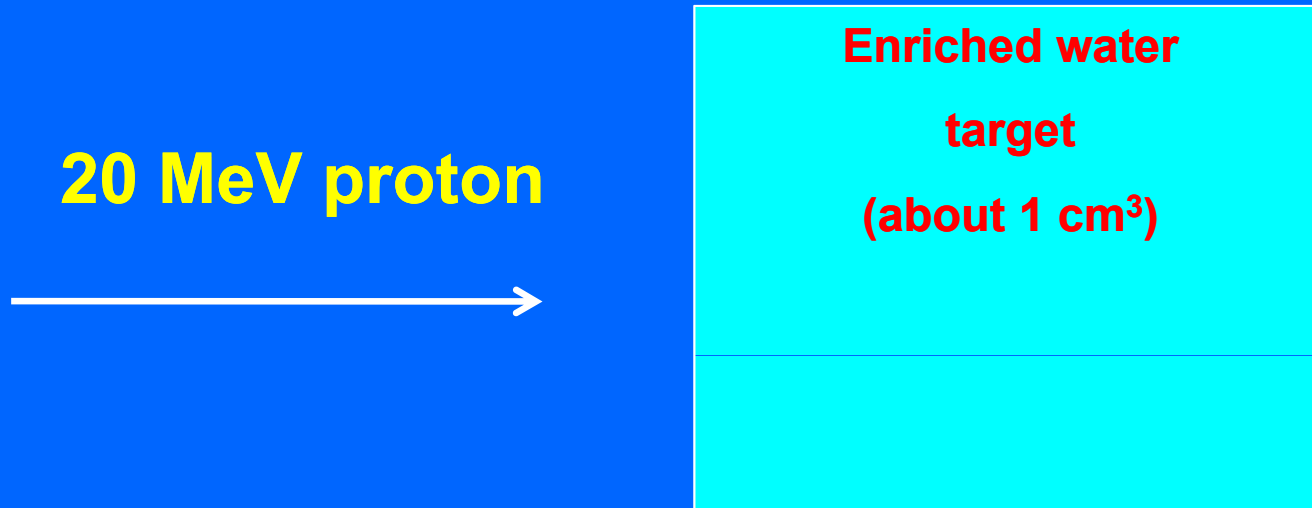
**Pipes for cooling. Why?**

**Let's suppose that the beam completely stops in the target:**

$$20 \text{ MeV} \times 50 \text{ } \mu\text{A} \times (1/e) = 1000 \text{ W}$$

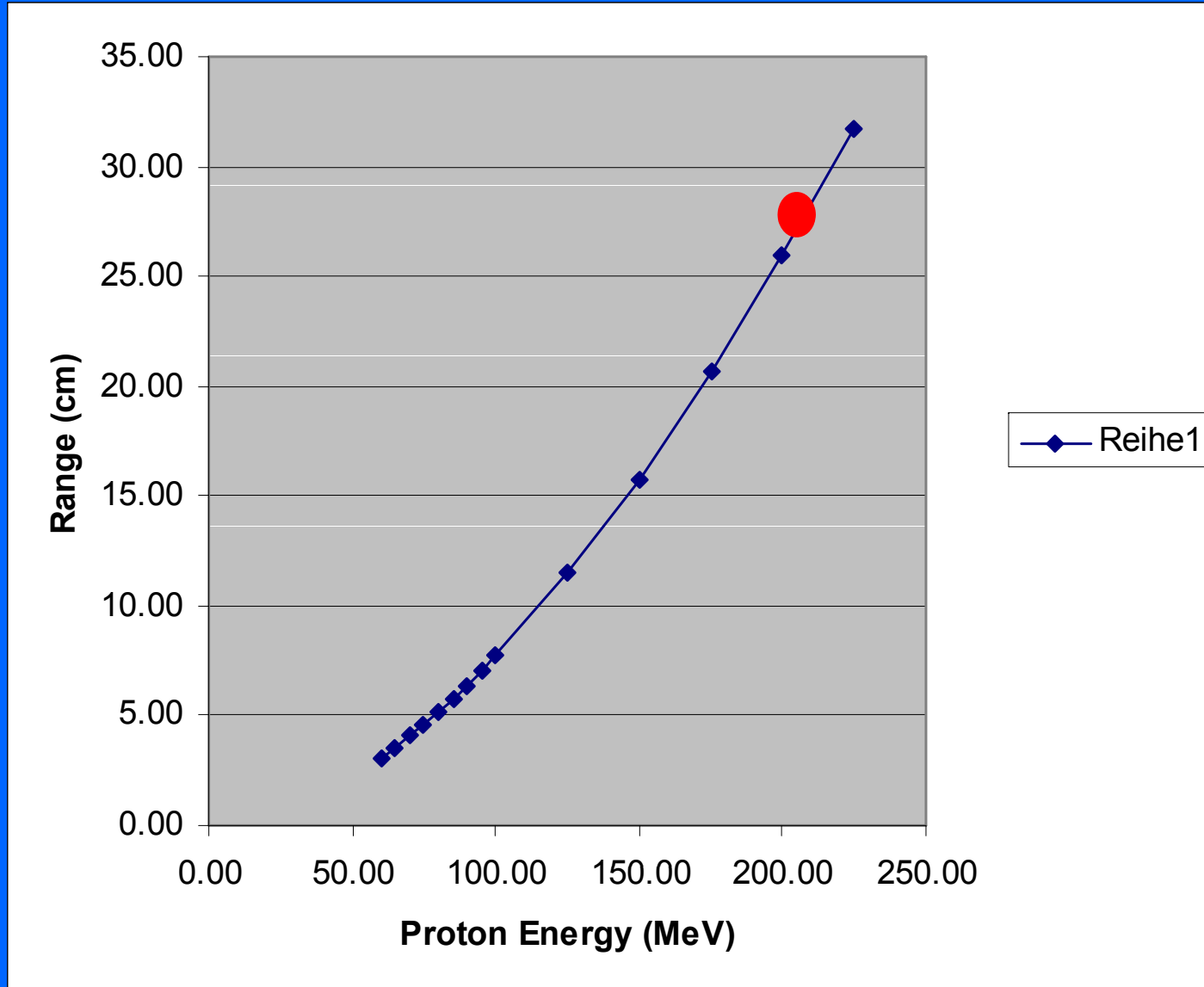
**1 cal = 4.18 J i.e. 1 cm<sup>3</sup> of water passes between 0 and 100 degrees**

**in less than 0.5 seconds !**



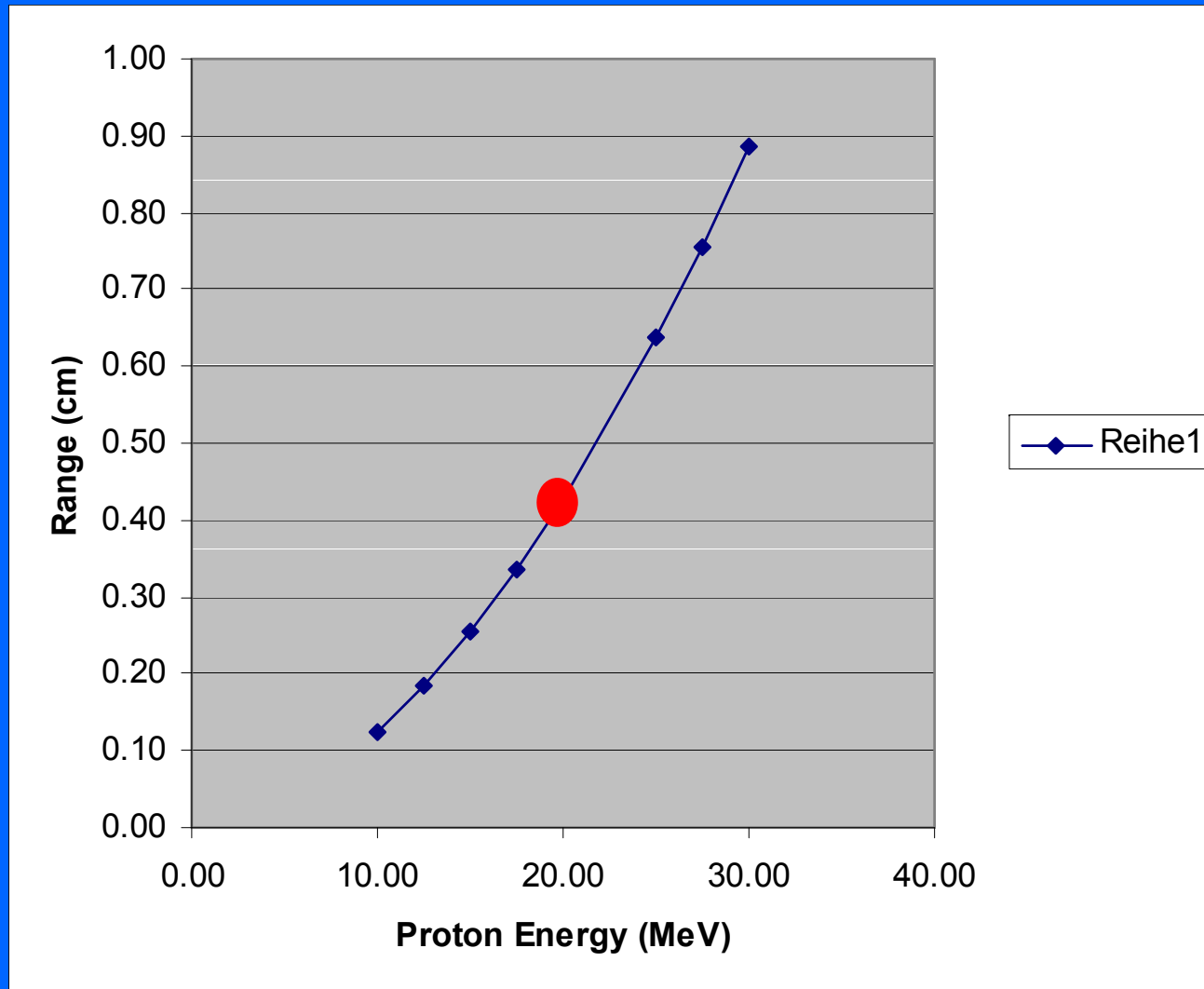
- Does the proton stop in water?
- “Sometimes” the reaction  $^{18}\text{O} (p,n) ^{18}\text{F}$  occurs. Probability?

## Range of the protons in water



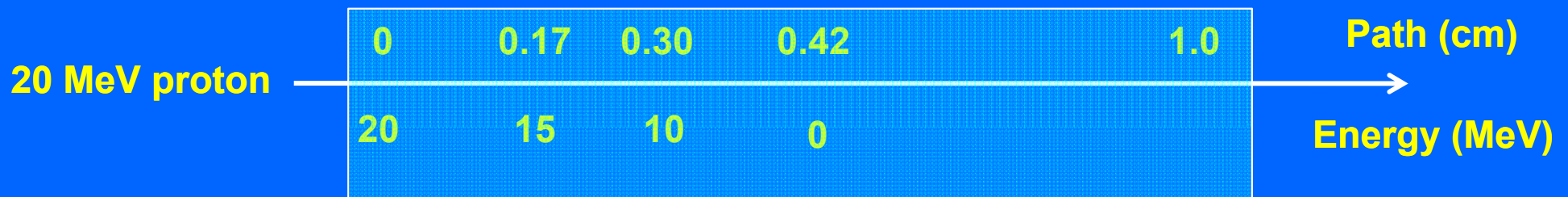
**Important to remember : 200 MeV → 27 cm**

## Range of protons in water



**20 MeV → 0.4 cm – All protons stop in the target**

Energy (MeV)	Range (cm)
20	0.42
15	0.25
10	0.12
5	0.036

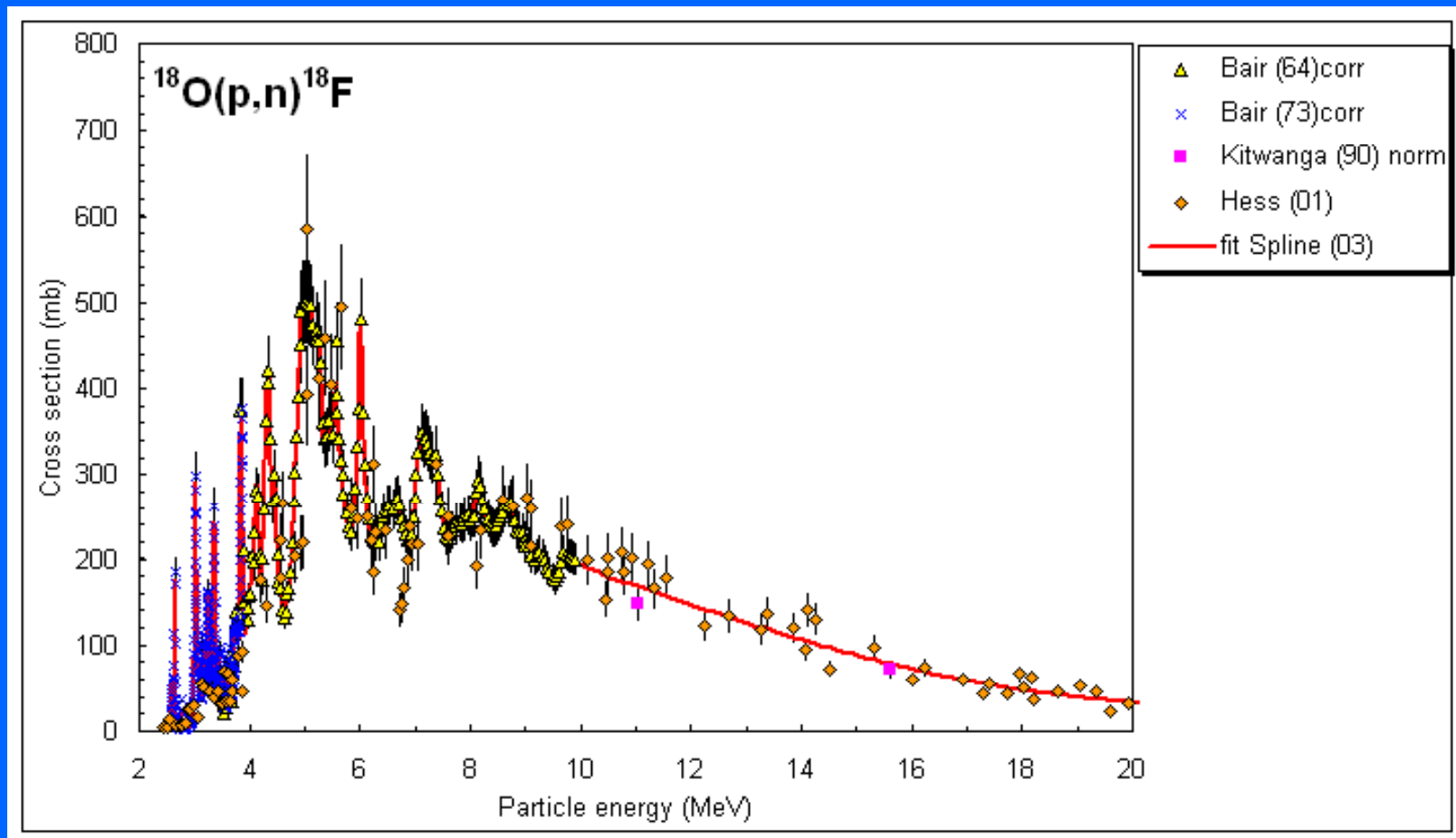


The reaction can take place in any point of the path  
i.e. at different energies !

[A useful link :](#)

[physics.nist.gov](http://physics.nist.gov) – NIST National Institute of Standards and technology





**For the exercise we will consider an average value  
of 100 mb for all the energies  
(1 barn =  $10^{-24}$  cm<sup>2</sup>)**

- How many  $^{18}\text{O}$  targets are there?

20 g (18+1+1) of enriched  $\text{H}_2\text{O}$  contain  $N_0$  molecules

$$\rightarrow 6.022 \times 10^{23} / 20 \rightarrow 3 \times 10^{22} \text{ } ^{18}\text{O} \text{ atoms/cm}^3$$

- How many “bullets” per second?

Current / charge of the proton

$$N_R = \sigma \times \frac{I}{e} \times \frac{N_t}{V} \times L_t \times \Delta t$$

$N_R$	number of reactions i.e. number of 18-F nuclides produced
$\sigma$	cross section
$I$	beam current
$e$	charge of the electron
$N_t$	number of target 18-O nuclei
$V$	volume of the target
$L_t$	<b>thickness of the target ?</b>
$\Delta t$	Irradiation time interval

- Thickness of the target  $\rightarrow$  range = 0.42 cm

## Result 1

- In 60 minutes :  $N_0 = 2 \times 10^{15}$  18-F nuclei are produced
- Which is the corresponding activity?
- $N(t) = N_0 \times \exp(-t/\tau)$
- At  $t=0$  the activity  $dN/dt$  is:  $N_0/\tau$
- F-18 :  $t_{1/2} = 110$  min  $\rightarrow \tau = t_{1/2} / \ln 2 = 158$  min = 9480 s
- Activity :  $A = 2 \times 10^{11}$  Bq (Bq  $\rightarrow$  Bequerel)
- 1 Ci =  $3.7 \times 10^{10}$  Bq (Ci  $\rightarrow$  Curie)

## Result 2

- The produced activity is about 6 Ci at the end of the irradiation

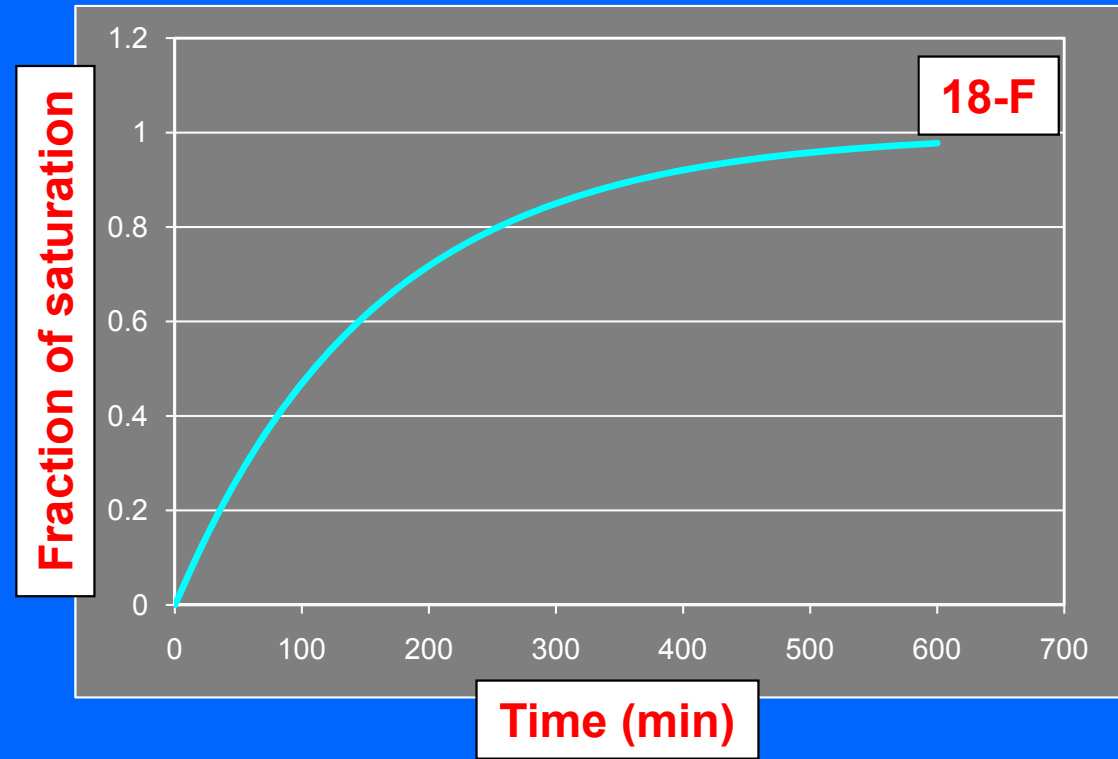
...but 18-F decays during irradiation

$$\Delta N = A \times \Delta t - N(t) \times \frac{\Delta t}{\tau}$$

Irradiation

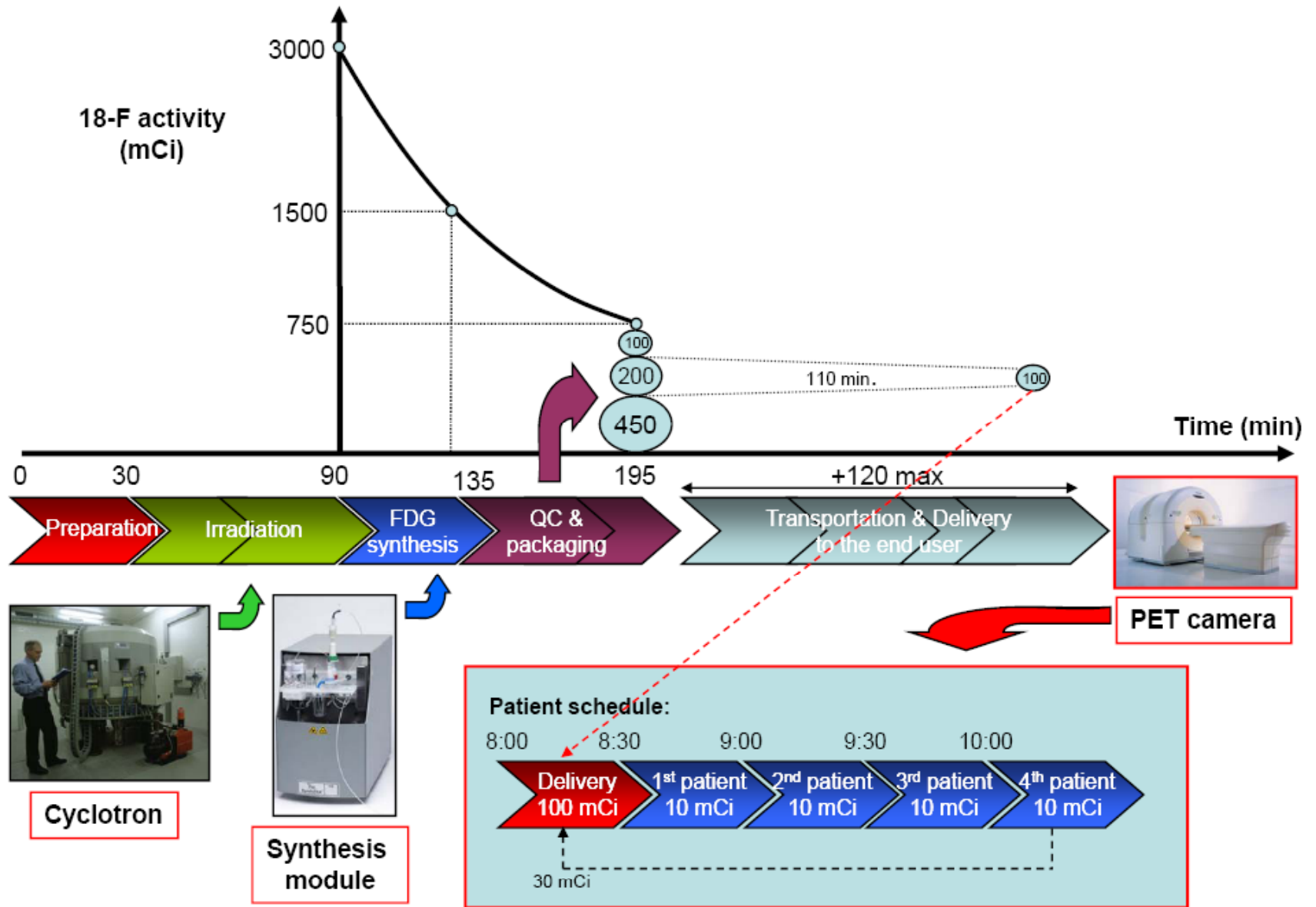
Decay

$$N(t) = A \tau \times (1 - e^{-t/\tau})$$

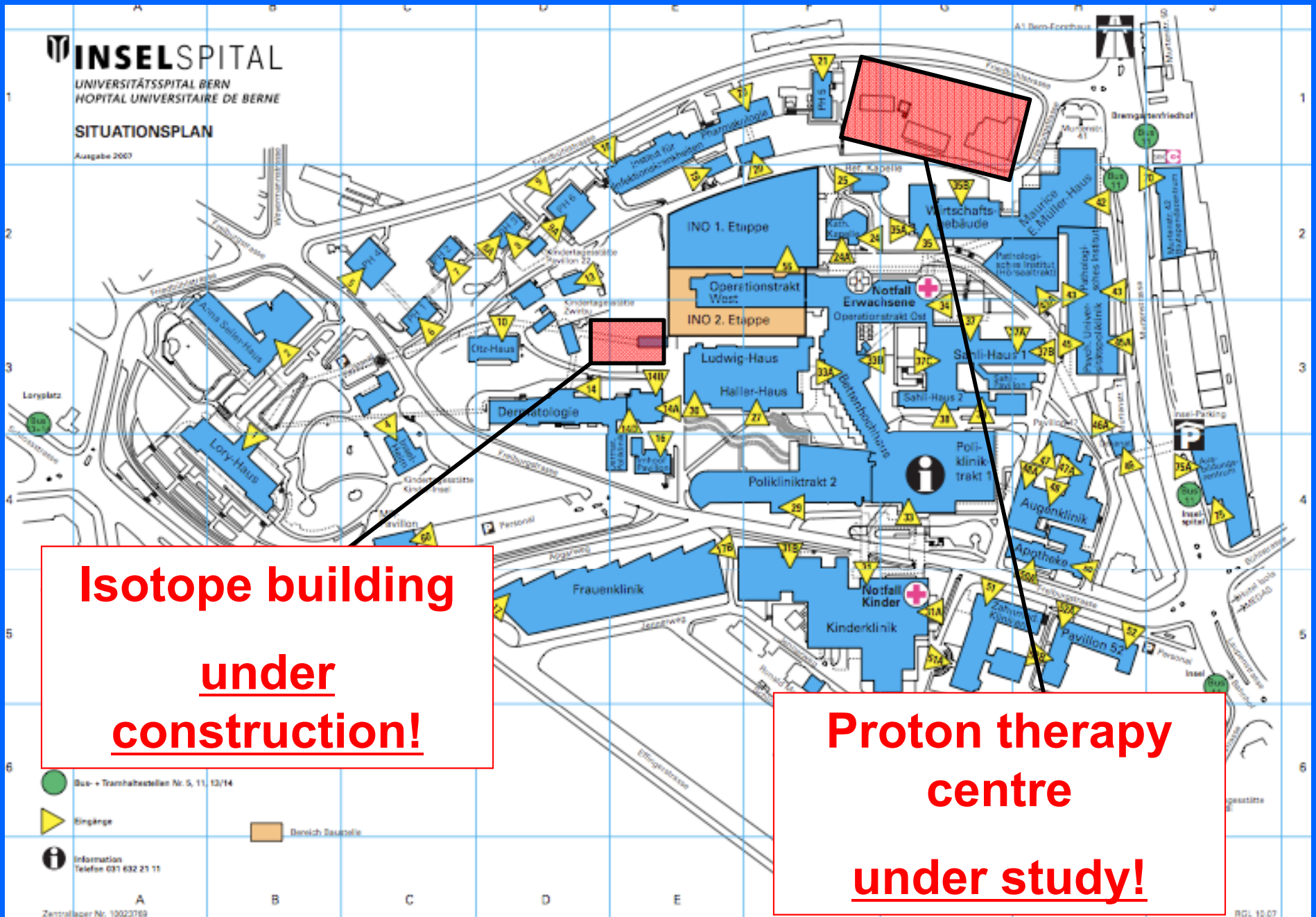


- If  $t \ll \tau$  the effect can be neglected
- If  $t \gg \tau$  saturation effect : production  $\sim$  decay
- For 18-F : the regime is far from saturation for  $t < 120$  min.
- Exercise – Taking this effect into account about 4.5 Ci of activity are produced in 60 min. irradiation

# A realistic supply chain



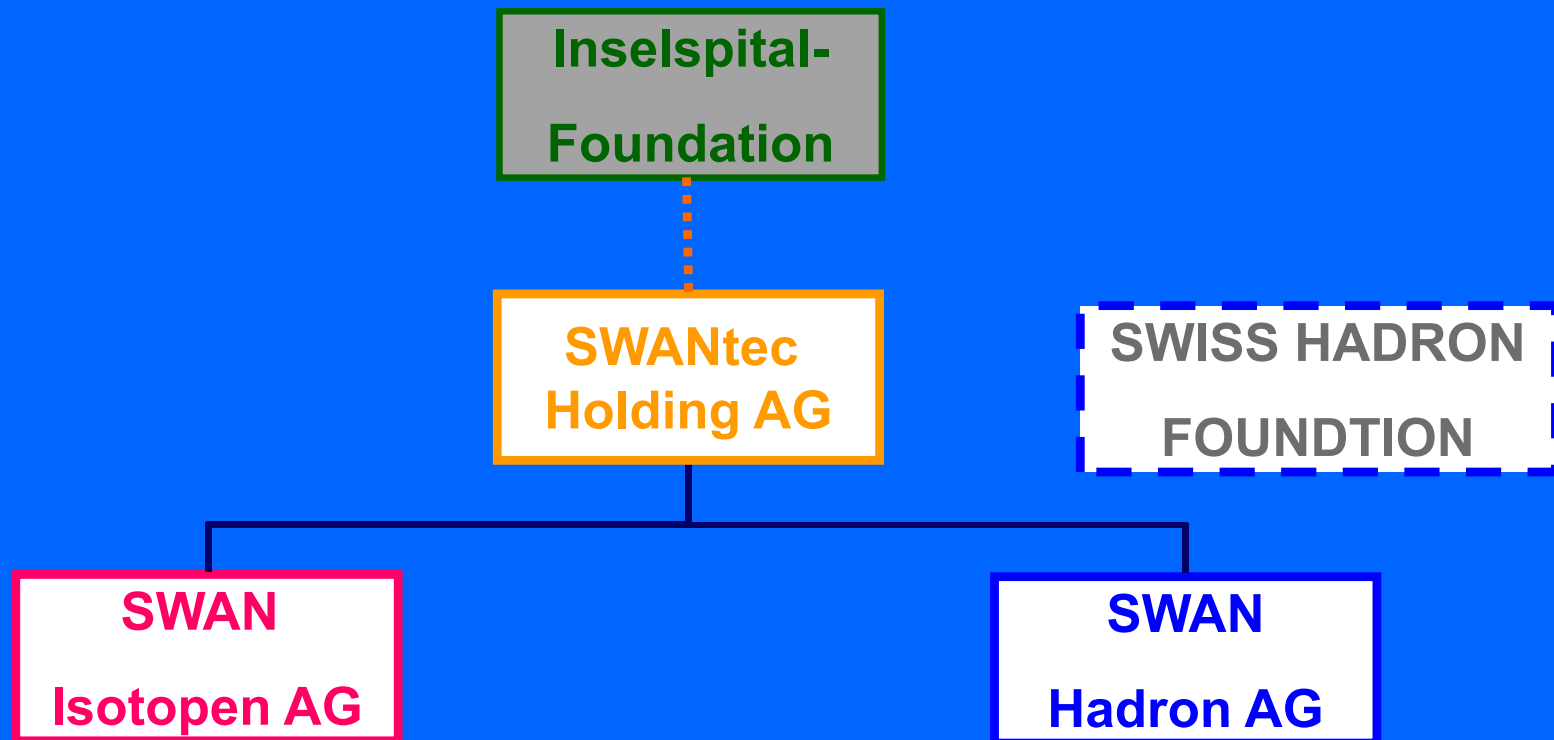
# The SWAN Project at the Inselspital in Bern



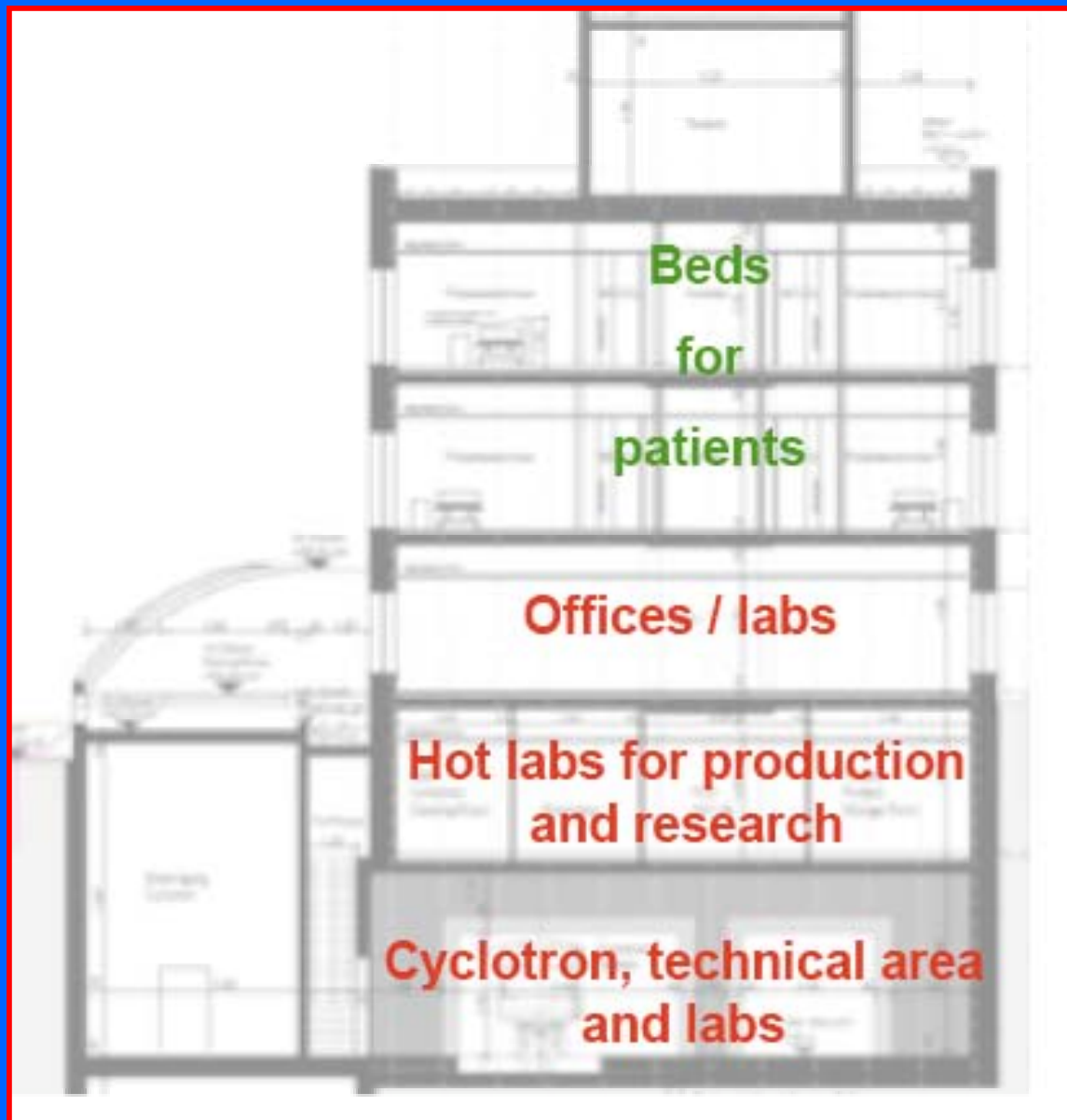
- Scope: constitute a combined centre for
  - Radioisotope production
  - Proton therapy
  - Research
- Short history
  - 2006/2007 first feasibility studies by Inselspital + Uni Bern
  - End 2007 approval and constitution of the main structure
  - 2008 detailed studies of the “Isotopen” part
  - 2009 start for the construction of the “Isotope building”
  - 2009 study for the implantation of the proton therapy centre
- Innovative structure involving public and private partners.
- Stakeholders:
  - Inselspital
  - University of Bern
  - Private investors



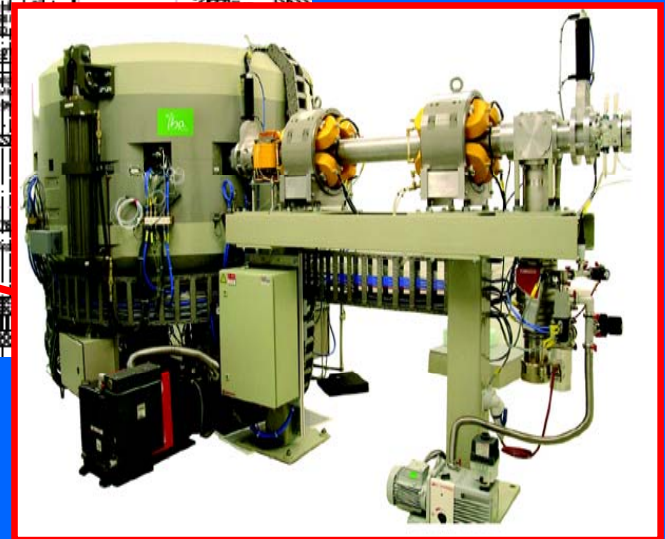
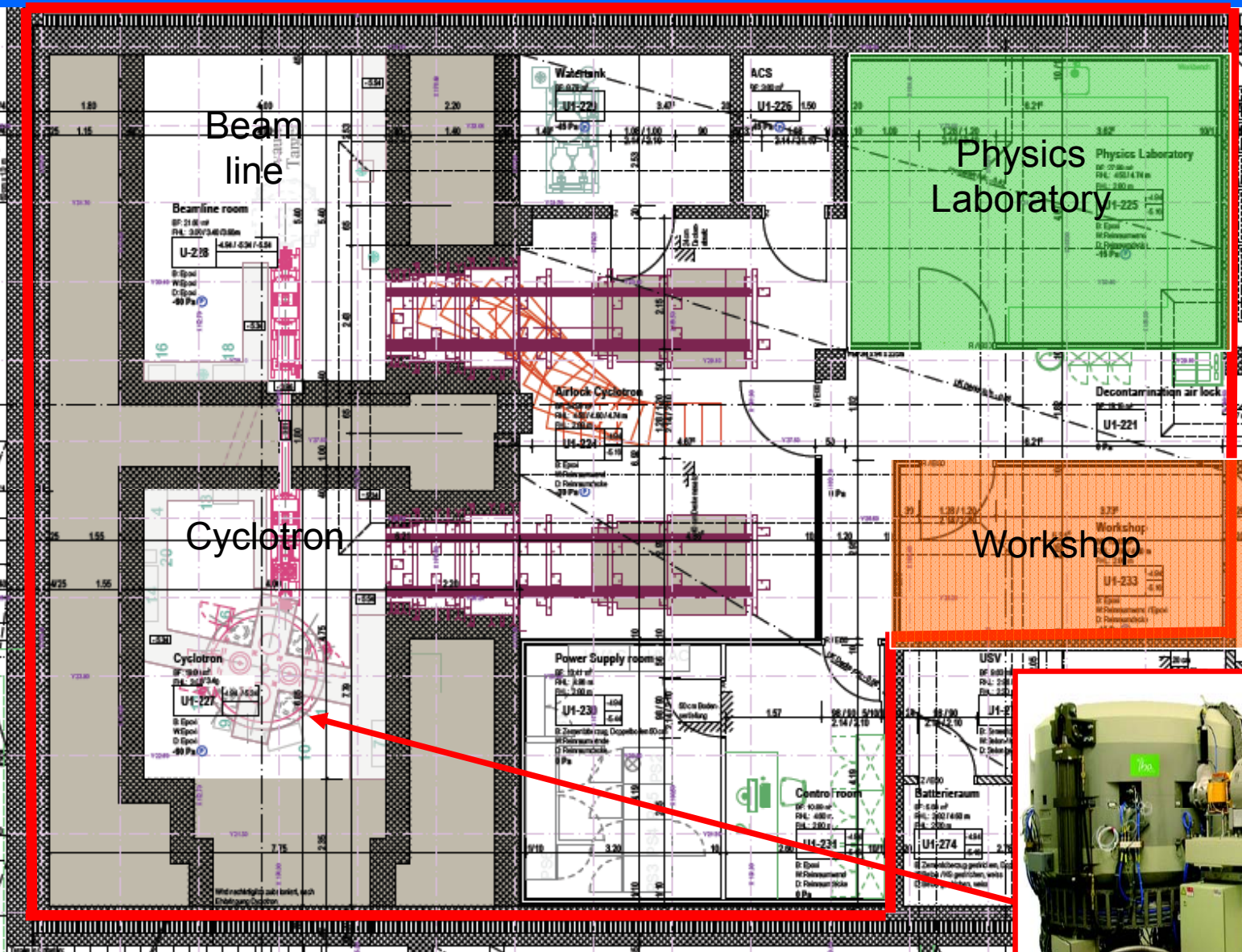
- **SWAN stands for SWiss hAdroN**
- **Founded at the end of 2007**
- **Shareholders: Inselspital, University of Bern, private investors**



# Structure of the isotope building



# The 18 MeV cyclotron laboratory



## *Status of the construction (March 2010)*



## *Many possible research activities*

- **Fundamental physics**

- **Positronium**

- **Particle detectors**

- **Beam monitoring**

- **Gamma-ray Resonant Absorption (GRA) related activities**

- **Innovative detectors (tests, calibrations, etc.)**

- **Applied physics**

- **PIXE (Proton Induced X-ray Emission)**

- **PIGE (Proton Induced Gamma Emission)**

- **PALS (Positron Annihilation Lifetime Spectroscopy)**

- **TLA (Thin Layer Activation)**

- **Radiation bio-physics**

## *Many possible research activities*

- **Targets**
- **Radiochemistry**
- **Clinical research**
  - **Oncology**
  - **Neurology**
  - **Cardiology**



**Dedicated irradiation chamber in the  
beam line vault**

**(Picture from CNA, Seville, Spain)**

# Radiation protection issues

- To screen 511 keV photons:
  - lead (about 10 cm) in hot cells
  - Lead:  $\rho=11.3 \text{ g/cm}^3$  ;  $\mu\rho=1.81 \text{ cm}^{-1}$ ; 10 cm  $\rightarrow$  Reduction factor =  $1.4 \times 10^{-8}$
  - tungsten (about 5 cm) for transport containers



Hot cells host radiochemical modules    Transport container for the vials

- What about the cyclotron?
- The target is a very powerful neutron source!
  - p (18-O, 18-F) n
  - Neutrons are very dangerous (W in the range 10-20)

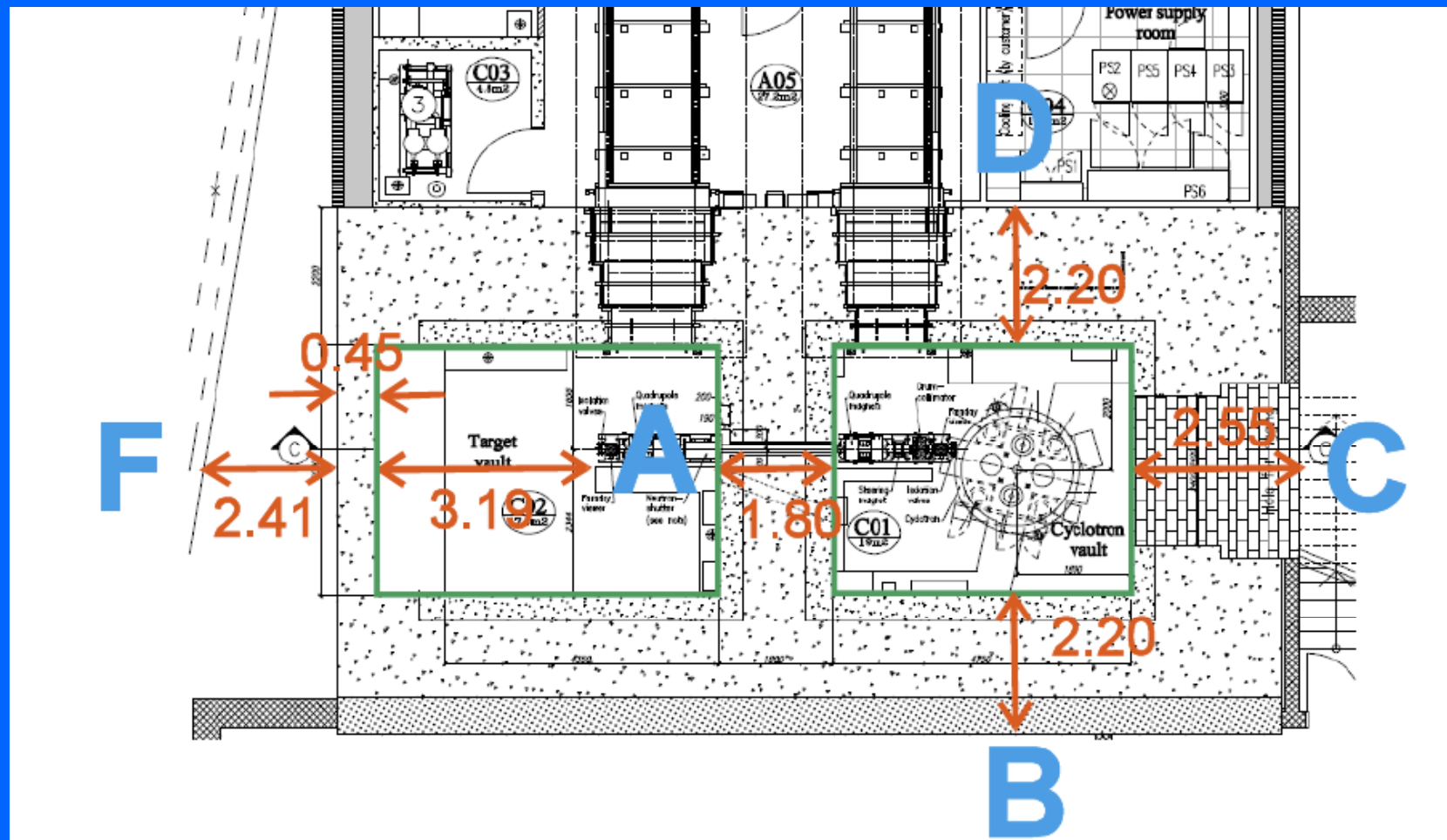
Fattori di ponderazione della radiazione incidente per ottenere dalla dose assorbita (in Gy) il valore di dose equivalente (in Sv)

Tipo di radiazione	Fattore di ponderazione della radiazione ( <b>Wr</b> )
Fotoni di tutte le energie ed elettroni	1
Neutroni di energia inferiore a 10 KeV	5
Neutroni tra 10 KeV e 100 KeV	10
Neutroni tra 100 KeV e 2 MeV	20
Neutroni tra 2 MeV e 20 MeV	10
Neutroni di energia maggiore di 10 MeV	5
Protoni	5
Particella alfa, nuclei pesanti	20

- Neutrons become “thermal neutrons” and produce activation!
  - The choice of the materials is important!



# Shielding against neutrons



## Shielding against neutrons

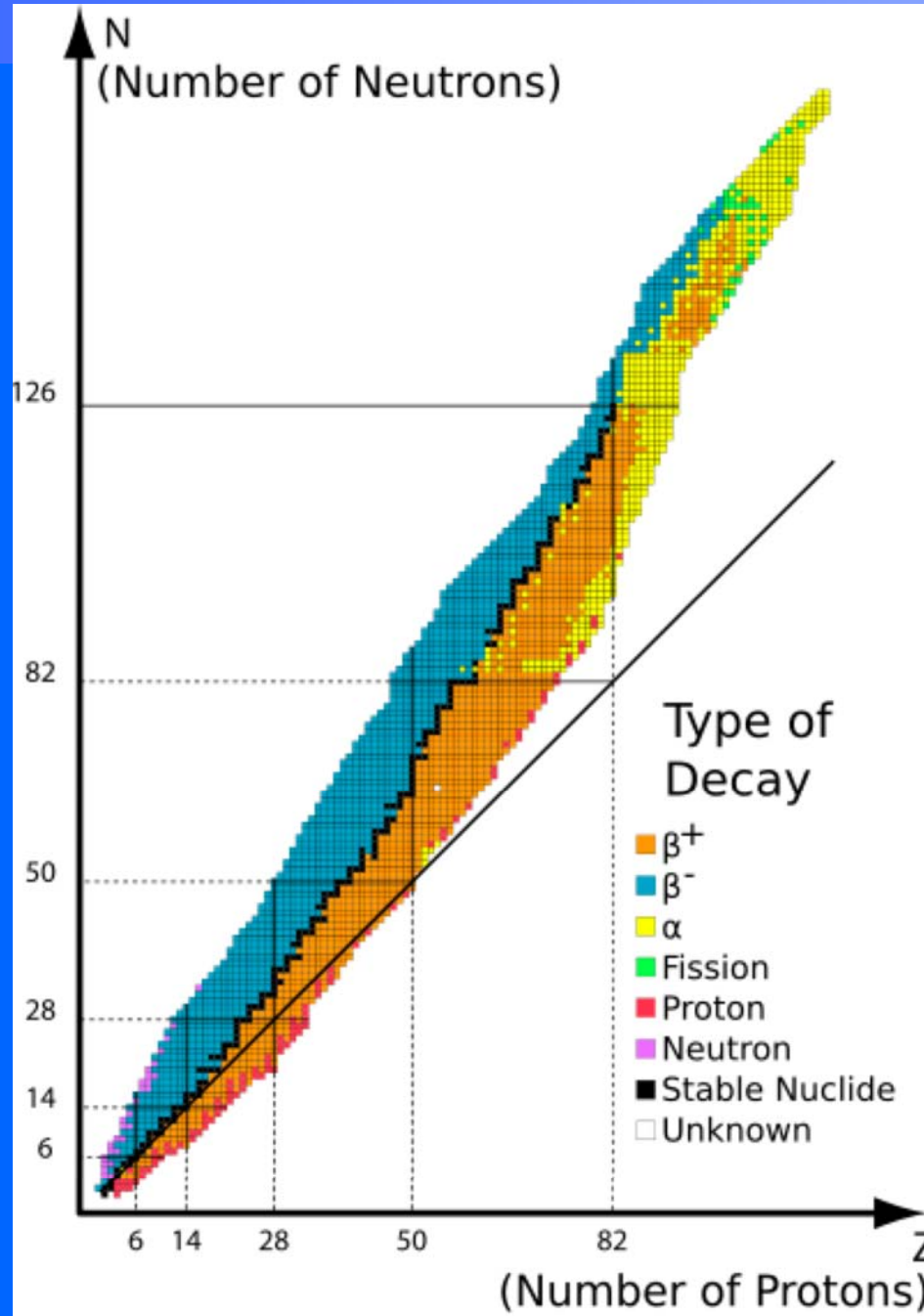
$$D = D_0 \left( \frac{d_0}{d_1} \right)^2 e^{-\frac{d_2}{\lambda}}$$

Tabelle 3: Schwächungslänge  $\lambda$  für Normalbeton für schnelle Neutronen.

$\lambda$ [cm]	Quelle
17.02	Argonne National Laboratory, LS Note 116, 1988
15.7	NCRP51 F10 (1977), ( $\rho = 2.35 \text{ gcm}^{-3}$ )
13.8 – 15.7	< 50 MeV, IAEA Technical Report Series 188 Seite 189 ( $\rho = 2.4 \text{ gcm}^{-3}$ )
15.2	Concrete Radiation Shielding, Tabelle 10.24, M.F. Kaplan ( $\rho = 2.2 \text{ gcm}^{-3}$ )
5.9	$E_n = 2 \text{ MeV}$ , [3]

- **Concrete** :  $\lambda=9.6 \text{ cm}^{-1}$ ; 200 cm  $\rightarrow$  Reduction factor =  $9.2 \times 10^{-10}$
- **Polyethylene** :  $\lambda=4.17 \text{ cm}^{-1}$ ; 87 cm  $\rightarrow$  Reduction factor =  $9.2 \times 10^{-10}$

*All this is possible thanks to artificially produced isotopes*



# The first table of isotopes comes from Rome!

## TABELLE RIASSUNTIVE E BIBLIOGRAFIA DELLE TRASMUTAZIONI ARTIFICIALI

Nota di **GIORGIO FEA** (\*)

Dato il grande sviluppo preso dallo studio delle trasmutazioni artificiali, soprattutto dopo l'impulso dato dalla scoperta della radioattività provocata, è parso utile, allo scrivente, riassumere in alcune tabelle sinottiche quanto è stato sin qui ottenuto dagli ormai moltissimi ricercatori, che si sono occupati dell'argomento.

Segue un'ampia bibliografia delle opere consultate per la compilazione delle tabelle stesse, e un quadro rappresentante, nel diagramma neutroni-protoni, quanto si conosce circa gli isotopi stabili e radioattivi.

### Disposizione delle tabelle.

§ 1. Esse sono quattro, in corrispondenza dei quattro tipi di corpuscoli bombardanti considerati, e cioè:

- 1) Trasmutazioni ottenute con bombardamento di protoni e deutoni.
- 2) Trasmutazioni ottenute con bombardamento di nuclei d'elio.
- 3) Trasmutazioni ottenute con bombardamento di neutroni.
- 4) Trasmutazioni ottenute con bombardamento di fotoni.

Ciascuna tabella è divisa in otto colonne, nelle quali è riportato, salvo nella seconda parte della terza tabella, quanto segue:

1) l'elemento bombardato, indicato col simbolo chimico e il numero atomico;

2)  $a$ : la energia del corpuscolo bombardante corrispondente ai dati energetici dei prodotti di trasmutazione riportati nella sesta colonna; questo dato ha importanza particolare per le trasmutazioni con nuclei d'elio, mentre può essere in generale trascurato nel caso di trasmutazioni con protoni e deutoni; perciò esso manca nella prima tabella.

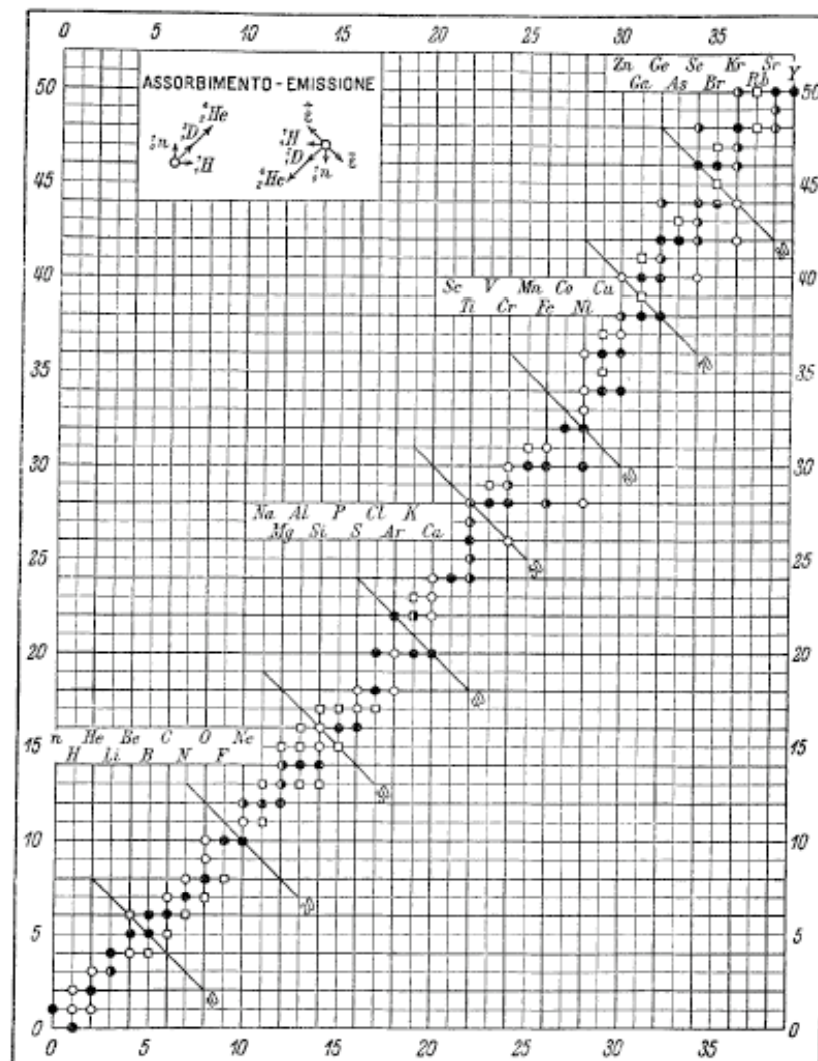
$b$ : in parentesi quadra e con segno  $>$  quella che da molti autori è data come energia minima per l'emissione dei corpuscoli indicati accanto in parentesi (se manca l'indicazione del corpuscolo il dato si riferisce alla

(\*) Si presentano la bibliografia e le tabelle riassuntive di tutte le trasmutazioni artificiali studiate sino al Maggio 1935.

SULLE TRASMUTAZIONI ARTIFICIALI

373

### QUADRO PROTONI-NEUTRONI - I.



*End of part II*